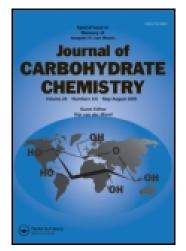
This article was downloaded by: [George Mason University]

On: 21 December 2014, At: 21:57

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Journal of Carbohydrate Chemistry

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/lcar20

Preparation of Primary and Secondary Azidosugars from Diols using the Dioxaphosphorane Methodology

Dominique Lafont & Paul Boullanger

Laboratoire de Chimie Organique II, UMR CNRS 5622
 Université Lyon I , Bât 308, 43 Bd du 11 Novembre 1918
 69622 Villeurbanne Cedex, France

b Laboratoire de Chimie Organique II, UMR CNRS 5622 Université Lyon I, Bât 308, 43 Bd du 11 Novembre 1918 69622 Villeurbanne Cedex, France Published online: 27 Feb 2008.

To cite this article: Dominique Lafont & Paul Boullanger (1999) Preparation of Primary and Secondary Azidosugars from Diols using the Dioxaphosphorane Methodology, Journal of Carbohydrate Chemistry, 18:6, 675-688, DOI: 10.1080/07328309908544029

To link to this article: http://dx.doi.org/10.1080/07328309908544029

PLEASE SCROLL DOWN FOR ARTICLE

Taylor & Francis makes every effort to ensure the accuracy of all the information (the "Content") contained in the publications on our platform. However, Taylor & Francis, our agents, and our licensors make no representations or warranties whatsoever as to the accuracy, completeness, or suitability for any purpose of the Content. Any opinions and views expressed in this publication are the opinions and views of the authors, and are not the views of or endorsed by Taylor & Francis. The accuracy of the Content should not be relied upon and should be independently verified with primary sources of information. Taylor and Francis shall not be liable for any losses, actions, claims, proceedings, demands, costs, expenses, damages, and other liabilities whatsoever or howsoever caused arising directly or indirectly in connection with, in relation to or arising out of the use of the Content.

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

Terms & Conditions of access and use can be found at http://www.tandfonline.com/page/terms-and-conditions

PREPARATION OF PRIMARY AND SECONDARY AZIDOSUGARS FROM DIOLS

USING THE DIOXAPHOSPHORANE METHODOLOGY

Dominique Lafont* and Paul Boullanger

Laboratoire de Chimie Organique II, UMR CNRS 5622 Université Lyon I, Bât 308, 43 Bd du 11 Novembre 1918 69622 Villeurbanne Cedex, France

Received January 14, 1999 - Final Form May 4, 1999

ABSTRACT

Treatment of methyl 2,3-di-O-benzyl- α -D-glucopyranoside (1), methyl 2,3-di-O-acetyl- α -D-glucopyranoside (4), 3-O-benzyl-1,2-O-(1-methylethylidene)- α -D-glucofuranose (6), 3-O-acetyl-1,2-O-(1-methylethylidene)- α -D-glucofuranose (9), 1,2-O-(1-methylethylidene)- α -D-xylofuranose (11) and methyl 2,3-di-O-acetyl- α -D-galactopyranoside (15) with diisopropylazodicarboxylate-triphenylphosphine in tetrahydrofuran led to the corresponding dioxaphosphoranes, which were opened by trimethylsilyl azide affording the silylated primary azidodeoxysugars. When the same reaction was performed on methyl 2,3-di-O-benzyl- α -D-galactopyranoside (20), an inversion of the regioselectivity of the dioxaphosphorane opening was observed, leading mainly to the 4-azido-4-deoxy- α -D-glucopyranoside derivative 27.

INTRODUCTION

Synthetic approaches to 6-azido-6-deoxy- and 4-azido-4-deoxysugars have been widely described, since these compounds are precursors of aminodeoxysugars which constitute part of antibiotics such as hybrimycin, ribostamycin, kanamycin^{1,2} or glyco-cyanomoylspermidines³ for example. More recently, 4-amino-4-deoxysugar derivatives have been used for the syntheses of 4-guanidinosugars⁴ and β -lactams.⁵ We have also

prepared a series of amphiphilic 6-aminocarbonyl derivatives from 6-azido-6-deoxy-D-glucose.⁶

The present paper deals with the syntheses of primary 5- or 6-azidodeoxysugars as well as a secondary 4-azido-4-deoxysugar, by reacting triphenylphosphine-diisopropylazodicarboxylate (DIAD) and trimethylsilyl azide with appropriate diols.

RESULTS AND DISCUSSION

Syntheses of primary azidodeoxysugars are well-documented in the literature. The 6-azido-6-deoxy derivatives are mostly obtained by nucleophilic displacement of leaving groups at C-6 (halogens⁷⁻¹⁰ or sulfonates¹¹⁻¹³) or by methods which involve an oxyphosphonium type of activation, such as Mitsunobu reaction with diphenylphosphoryl azide¹² or zinc azide/bis-pyridine complex¹⁴ as the nucleophile. Highly regioselective ring-opening of cyclic sulfites¹⁵ or sulfates^{12,16} with sodium azide have also been reported, affording 6-primary, as well as secondary azidodeoxysugar derivatives. On the other hand, 4-azido-4-deoxy-D-glucose derivatives are mainly obtained by displacement with sodium azide of 4-O-mesyl^{17,18} or 4-O-triflyl-D-galactose¹⁹ precursors.

E. Zbiral et al. have described the structural modifications of partly silylated carbohydrates, having two or three contiguous hydroxyl groups, with triphenylphosphine-DIAD and a nucleophile.^{20,21} When the reaction was performed on methyl 2,6-bis-*O-tert*-butyldimethylsilyl-β-D-glucopyranoside, methyl 3-deoxy-3-halogeno-D-allopyranosides were obtained using triphenylphosphine hydrobromide or methyl iodide as nucleophiles whereas, in the corresponding α-series, methyl 4-deoxy-4-halogeno-D-galactopyranosides were formed. Methyl 3-azido-3-deoxy-D-allopyranoside was also obtained from methyl 6-*O-tert*-butyldimethylsilyl-β-D-glucopyranoside, using triphenylphosphine-DIAD and hydrazoic acid. These reactions proceed via a dioxaphosphorane intermediate (A) in equilibrium with two zwitterionic open forms (B and C). The regio- and stereoselectivity of the nucleophilic attack on the latter are governed by stereoelectronic factors (Scheme 1).

To our knowledge, this reaction was neither extended to the 4,6-diols in the hexopyranose series nor to the 5,6- or 3,5-diols in the hexo- or pentofuranoside series.

Treatment of methyl 2,3-di-O-benzyl- α -D-glucopyranoside²² (1) with a slight excess of DIAD and triphenylphosphine in tetrahydrofuran, followed by addition of trimethylsilyl azide (2.5 eq) at 0 °C, afforded methyl 6-azido-2,3-di-O-benzyl-6-deoxy-4-O-trimethylsilyl- α -D-glucopyranoside (2) in 88% yield. The reaction was assumed to proceed by the intermediate of a cyclic phosphorane²³ which was opened by attack of the azide at the less hindered C-6 with simultaneous silylation at O-4. Such a result was not unexpected, even though the same reaction performed on 1,2-propanediol affords the 2-azido-1-trimethyl-

silyloxypropane with a very high regioselectivity (>99%).²⁴ The attack of the nucleophile at the secondary position, in this latter example, is due to inductive effects which largely overcome steric hindrance. Desilylation of product 2 with potassium carbonate in methanol gave rise to methyl 6-azido-2,3-di-*O*-benzyl-6-deoxy-α-D-glucopyranoside²⁵ (3) in high yield. Similar results were obtained from methyl 2,3-di-*O*-acetyl-α-D-glucopyranoside²⁶ (4) which was transformed into the 6-azido-4-*O*-silylated product 5 by the intermediate of a dioxaphophorane (³¹P NMR δ -57.5 ppm in THF). 3-*O*-Benzyl-1,2-*O*-(1-methylethylidene)-α-D-glucofuranose²⁷ (6) or its 3-*O*-acetyl analogue²⁸ 9, treated in the same conditions, afforded 6-azido-6-deoxy-5-*O*-trimethylsilyl products 7 and 10 with the same regioselectivity in 78% and 81% yields respectively. As observed with compound 1, the reaction was regiospecific and only one compound was oberved by ¹H NMR of the crude reaction mixture. The regiospecificity of the reaction was again directed towards the less hindered primary position. Further desilylation of 7 afforded the azido-alcohol 8 already described in the literature.²⁹

$$HO \longrightarrow R$$

$$(CH_2)_n$$

$$PPh_3P^+O \longrightarrow R$$

$$(CH_2)_n$$

$$O \longrightarrow R'$$

$$Ph_3P \longrightarrow (CH_2)_n$$

$$O \longrightarrow R'$$

$$Ph_3P^+O \longrightarrow R'$$

$$O \longrightarrow R$$

Scheme 1

Treatment of 1,2-O-(1-methylethylidene)- α -D-xylofuranose³⁰ (11) with triphenylphosphine-DIAD and trimethylsilyl azide led to 5-azido-5-deoxy-1,2-O-(1-methylethylidene)-3-O-trimethylsilyl- α -D-xylofuranose (12) accompanied with 10% of 1,2-O-(1-methylethylidene)-3,5-di-O-trimethylsilyl- α -D-xylofuranose (13). The latter could be the result of the silylation of the diol 11, before the formation of the dioxaphosphorane adduct. The structure of 13 was ascertained by direct silylation of the diol 11 with trimethylsilyl triflate and triethylamine in dichloromethane. Desilylation of the mixture 12/13 afforded 5-azido-5-deoxy-1,2-O-(1-methylethylidene)- α -D-xylofuranose³¹ (14) in 73% overall yield.

The azidosilylation of methyl 2,3-di-O-acetyl- α -D-galactopyranoside³² (15) afforded a mixture of three compounds, i. e., the expected 6-azido-6-deoxy-4-O-silylated derivative 16 (67%), the 6-azido-6-deoxy-3-O-silylated derivative 17 (7%) and the 4,6-di-Osilylated product 18 (9%). The formation of compound 17 results of the migration of the 3-O-acetyl group whereas that of 18 probably results of the silvlation of diol 15 (scheme 2). Azidosilylation of methyl 2,3-di-O-benzyl-α-D-galactopyranoside¹⁸ (20), using a lower excess of trimethylsilyl azide (1.5 eq) proceeded quite differently and afforded a mixture from which the components (20-23) were separated only after desilylation (K₂CO₃ in methanol). The major compound (66% overall yield from 20) was identified as methyl 4-azido-2,3-di-O-benzyl-4-deoxy-α-D-glucopyranoside (23)18 and resulted of a nucleophilic attack with inversion at the secondary position C-4. The minor derivatives were shown to be methyl 4,6-diazido-2,3-di-O-benzyl-4,6-dideoxy-α-D-glucopyranoside (21) (3%), methyl 6-azido-2,3-di-O-benzyl-6-deoxy-α-D-galactopyranoside (22) (3%) and methyl 2,3-di-O-benzyl-α-D-galactopyranoside (20) (3%). Compound 22 is the hydrolysed form of the azidosilyl adduct due to the attack of the nucleophile at C-6, whereas compound 20 is the hydrolyzed form of the 4,6-di-O-silyl intermediate 24. The formation of the diazido derivative 21 could result from the slight excess of PPh₃-DIAD still present in the reaction mixture after a first nucleophilic attack. Silylation of compounds 20, 22 and 23 to 24, 26 and 27, respectively, confirmed the formation of the three products during the reaction by comparison of their ¹³C NMR spectra with that of the crude azidosilylation mixture.

The most striking feature concerns the regionselectivity of the nucleophilic attack in pyranose rings. The latter is directed towards the primary position in compounds 1, 4 and 15, whereas it is directed towards the secondary position in compound 20. The explanation could lay in the geometry of the dioxaphosphorane rings (e.g., 19) or in the equilibrium between cyclic dioxaphosphorane and open phosphonium intermediates (e.g., 19) or 19b). Nevertheless, 31P NMR of reaction mixtures did not reveal any difference

in the azidosilylation of 20 and the other pyranose 4,6-diols. In our opinion, that discrepancy is more probably due to the steric hindrance governing the approach of the nucleophile. The disfavoured axial attack of leaving groups at the C-4 position of D-gluco derivatives is well examplified and could explain the substitution at C-6 in compounds 1 and 4. The equatorial substitution at C-4 of the dioxaphosphorane 25 (or phosphonium 25a) is also in agreement with the usual equatorial attack at C-4 of D-galacto derivatives. The reverse regioselectivity observed in the 3-O-acetyl derivative 15 could, in our opinion, be explained by the participation of the acetyl group that shifts the phosphorane (19)/phosphonium (19a and 19b) equilibrium to 19c. Then, the substitution at C-6 results from a nucleophilic attack on the latter (scheme 2).

CONCLUSION

In conclusion, the use of triphenylphosphine-DIAD-trimethylsilylazide allows the preparation of primary azidodeoxysugars from primary-secondary diols, in good yields and with a high regioselectivity, in the D-xylo and D-gluco series. In the D-galacto series, the regioselectivity depends on the protecting groups at the neighbouring position. This one-pot methodology complements the other methods described in the literature and avoids the use of protection/deprotection steps for the regioselective azidations of unprotected diols.

EXPERIMENTAL

General methods. Tetrahydrofuran was dried by refluxing under argon with sodium-benzophenone prior to distillation. Methanol was refluxed with sodium methylate before distillation. Dichloromethane was washed with H₂SO₄ and water prior to distillation. Melting points were determined on a Büchi apparatus and were uncorrected. TLC analyses were performed on aluminium sheets coated with silica gel 60 F 254 Merck. Compounds were visualized by spraying the TLC plates with dilute 15 % aqueous sulfuric acid, followed by charring at 150 °C for a few minutes. Column chromatographies were performed on silica gel Geduran Si 60 Merck. Optical rotations were recorded on a Perkin-Elmer 241 polarimeter in a 1 dm cell at 21 °C. ¹H and ¹³C NMR spectra were recorded with Bruker AC-200 or AM-300 spectrometers operating at 200 or 300 MHz and 50 or 75.5 MHz respectively with tetramethylsilane as internal standard. Mass spectra

were recorded on Finnigan MAT 95 XL spectrometer. Elemental analyses were carried out by the "Laboratoire Central d'Analyses du CNRS" (Vernaison, France).

General procedure for the azidosilylation reaction. To a suspension of the sugar (1.0 mmol) in dry THF (5 mL) were added successively diisopropylazodicarboxylate (203 µL, 1.05 mmol) and triphenylphosphine (0.275 g, 1.05 mmol). The mixture was stirred at rt for one hour, then cooled to 0 °C before addition of trimethylsilyl azide (328 µL, 2.5 mmol, unless otherwise stated). The solution was allowed to reach rt and stirring was maintained overnight. The reaction mixture was then poured into sat aq NaHCO₃ (20 mL) and the aqueous layer was extracted twice with CH₂Cl₂ (2x20 mL). The combined organic layers were dried (Na₂SO₄) and concentrated under reduced pressure to afford the crude product which was purified by column chromatography.

General procedure for the desilylation reaction. To a solution of the silylated sugar (0.5-0.8 mmol) in dry MeOH (20 mL) was added anhydrous K₂CO₃ (0.5 g, 1.51 mmol). The mixture was stirred overnight, concentrated and the products were then extracted twice with CH₂Cl₂ (2x30 mL). The combined organic layers were washed with water (10 mL), dried (Na₂SO₄) and concentrated under reduced pressure to afford the crude product which was purified by column chromatography.

General procedure for the silylation reaction. To a solution of the sugar (0.20 mmol) in CH₂Cl₂ (5 mL), was added successively at -20 °C triethylamine (3 eq/OH) and trimethylsilyltrifluoromethane sulfonate (1.5 eq/OH). The solution was stirred for 16 h, then allowed to reach to rt and concentrated. The residue was applied at the top on a short column of silica-gel; elution with EtOAc/petroleum ether (1:4 v/v) afforded the pure product as an oil.

Methyl 6-Azido-2,3-di-O-benzyl-6-deoxy-4-O-trimethylsilyl-α-D-glucopyranoside (2). Obtained by the azidosilylation procedure described above from methyl 2,3-di-O-benzyl-α-D-glucopyranoside²² (1). Purification by column chromatography (EtOAc/petroleum ether 1:5 v/v) afforded pure product 2 (0.410 g, 88% yield). Compound 2: liquid; R_f 0.78 (EtOAc/petroleum ether 1:5 v/v); $[\alpha]_D$ +56.4° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.10 (s, 9 H, 3 CH₃Si), 3.34 (dd, 1 H, J_{5,6b} = 5.3 Hz, J_{6a,6b} = 12.8 Hz, H-6b), 3.40 (s, 3H, OCH₃), 3.41 (dd, 1 H, J_{5,6a} = 2.5 Hz, H-6a), 3.50 (dd, 1 H, J_{1,2} = 3.6 Hz, J_{2,3} = 9.6 Hz, H-2), 3.54 (dd, 1 H, J_{3,4} = J_{4,5} = 9.6 Hz, H-4), 3.74 (m, 2 H, H-3,5), 4.49 (d, 1 H, H-1), 4.60 and 4.72 (2d, 2 H, J = 12.0 Hz, CH₂Ph), 4.74 and 5.01 (2d, 2 H, J = 11.3 Hz, CH₂Ph), 7.29-7.35 (m, 10 H, 2 C₆H₅); ¹³C NMR (CDCl₃) δ 0.51 (3 C, 3 CH₃Si), 51.33 (C-6), 55.20 (OCH₃), 71.03, 71.71 (C-4,5), 73.25, 75.26 (2 C, 2 CH₂Ph), 81.20, 81.39 (C-2,3), 97.91 (C-1), 127.15-128.36, 137.96, 138.91 (12 C, 2 C₆H₅).

Anal. Calcd for $C_{24}H_{33}N_3O_9Si$ (471.61): C, 61.11; H, 7.05; N, 8.91. Found: C, 60.90; H, 7.22; N, 8.67.

Methyl 6-Azido-2,3-di-*O*-benzyl-6-deoxy-α-D-glucopyranoside (3). Obtained as described above from methyl 6-azido-2,3-di-*O*-benzyl-6-deoxy-4-*O*-trimethylsilyl-α-D-glucopyranoside (2) (0.377 g, 0.80 mmol). Purification by column chromatography (EtOAc/petroleum ether 1:2 v/v) afforded pure product 3 (0.303 g, 95% yield). Compound 3^{25} : liquid; R_f 0.85 (EtOAc/petroleum ether 1:2 v/v); [α]_D +13.8° (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 2.23 (d, 1 H, J_{4,OH} = 2.3 Hz, OH), 3.35-3.45 (m, 5 H, H-6a,6b, OCH₃), 3.47 (dd, 1 H, J_{3,4} = 9.0 Hz, J_{4,5} = 9.5 Hz, H-4), 3.54 (dd, 1 H, J_{1,2} = 3.5 Hz, J_{2,3} = 9.5 Hz, H-2), 3.72 (m, 1 H, H-5), 4.66 (d, 1 H, H-1), 4.68 and 4.79 (2d, 2 H, J = 12.0 Hz, CH₂Ph), 4.69 and 5.05 (2d, 2 H, J = 11.5 Hz, CH₂Ph), 7.29-7.35 (m, 10 H, 2 C₆H₅); ¹³C NMR (CDCl₃) δ 51.59 (C-6), 55.46 (OCH₃), 70.41, 70.76 (C-4,5), 73.17, 75.43 (2 C, 2 CH₂Ph), 79.90, 81.19 (C-2,3), 98.14 (C-1), 128.03-128.72, 138.00, 138.73 (12 C, 2 C₆H₅).

Methyl 2,3-Di-O-acetyl-6-azido-6-deoxy-4-O-trimethylsilyl-α-D-glucopyranoside (5). Obtained as described above from methyl 2,3-di-O-acetyl-α-D-glucopyranoside²⁶ (4) (0.278 g, 1.00.mmol). Purification by column chromatography (EtOAc/petroleum ether 1:3 v/v) afforded pure product 5 (0.311 g, 83% yield). Compound 5: liquid; R_f 0.76 (EtOAc/petroleum ether 1:3 v/v); $[\alpha]_D$ +116.5° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.10 (s, 9 H, 3 CH₃Si), 2.07 (s, 6 H, 2 CH₃COO), 3.37 (dd, 1 H, J_{5,6b} = 4.8 Hz, J_{6a,6b} = 13.1 Hz, H-6b), 3.42 (s, 3H, OCH₃), 3.52 (dd, 1 H, J_{5,6a} = 2.3 Hz, H-6a), 3.73 (dd, 1 H, J_{3,4} = 8.5 Hz, J_{4,5} = 9.5 Hz, H-4), 3.83 (ddd, 1 H, H-5), 4.81 (dd, 1 H, J_{1,2} = 3.7 Hz, J_{2,3} = 10.1 Hz, H-2), 4.90 (d, 1 H, H-1), 5.37 (dd, 1 H, H-3); ¹³C NMR (CDCl₃) δ 0.27 (3 C, 3 CH₃Si), 20.64, 21.01 (2 C, 2 CH₃COO), 50.88 (C-6), 55.22 (CH₃O), 70.14, 70.94, 71.20, 72.40 (4 C, C-2,3,4,5), 96.85 (C-1), 169.59, 170.20 (2 C, 2 CH₃CO).

Anal. Calcd for $C_{14}H_{25}N_3O_7Si$ (375.45): C, 44.78; H, 6.71; N, 11.19. Found: C, 44.85; H, 6.94; N, 10.83.

6-Azido-3-*O*-benzyl-6-deoxy-1,2-*O*-(1-methylethylidene)-5-*O*-trimethylsilyl-α-D-glucofuranose (7). Obtained as described above from 3-*O*-benzyl-1,2-*O*-(1-methylethylidene)-α-D-glucofuranose²⁷ (6) (0.310 g, 1.0 mmol). Purification by column chromatography (EtOAc/petroleum ether 1:4 v/v) afforded pure product 7 (0.318 g, 78% yield). Compound 7: liquid; R_f 0.70 (EtOAc/petroleum ether 1:4 v/v); [α]_D -43.1° (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.13 (s, 9 H, 3 CH₃Si), 1.32, 1.52 (2s, 6 H, 2 CH₃C), 3.39 (ddd, 1 H, J_{4,6b} = J_{5,6b} = 2.1 Hz, J_{6a,6b} = 12.1 Hz, H-6b), 3.59 (bd, 1 H, J_{5,6a} < 1.0 Hz, H-6a), 4.05 (bs, 1 H, H-3), 4.19 (bs, 2 H, H-4,5), 4.60 (d, 1 H, J_{1,2}

= 3.7 Hz, H-2), 4.57 and 4.69 (2d, 2 H, J = 11.6 Hz, CH_2Ph), 5.87 (d, 1 H, H-1), 7.30-7.40 (m, 5 H, C_6H_5); ¹³C NMR (CDCl₃) δ 0.49 (3 C, 3 CH_3Si), 26.40, 26.88 (2 C, 2 CH_3C), 55.23 (C-6), 68.62 (C-5), 71.70 (CH_2Ph), 80.70, 81.35, 81.76 (C-2,3,4), 105.12 (C-1), 111.97 ($C(CH_3)_2$), 127.38-128.53, 137.55 (C_6H_5).

Anal. Calcd for $C_{19}H_{29}N_3O_5Si$ (407.53): C, 56.00; H, 7.17; N, 10.31. Found: C, 56.32; H, 6.91; N, 10.28.

6-Azido-3-*O*-benzyl-6-deoxy-1,2-*O*-(1-methylethylidene)-α-D-glucofuranose (8). Obtained as described above from 6-azido-3-*O*-benzyl-6-deoxy-1,2-*O*-(1-methylethylidene)-5-*O*-trimethylsilyl-α-D-glucofuranose (7) (0.265 g, 0.65 mmol). Purification by column chromatography (EtOAc/petroleum ether 1:2 v/v) afforded pure product 8^{29} (0.207 g, 95% yield). Compound 8: liquid; R_f 0.35 (EtOAc/ petroleum ether 1:4 v/v); [α]_D -61.0° (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 1.34, 1.50 (2s, 6 H, 2 CH₃C), 2.32 (d, 1 H, J_{5,OH} = 3.3 Hz, H-5), 3.42 (m, 1 H, H-6b), 3.56 (dd, 1 H, J_{5,6a} = 2.0 Hz, J_{6a,6b} = 12.0 Hz, H-6a), 4.11 (bs, 3 H, H-3,4,5), 4.64 (d, 1 H, J_{1,2} = 3.7 Hz, H-2), 4.54 and 4.75 (2d, 2 H, J = 11.8 Hz, CH₂Ph), 5.93 (d, 1 H, H-1), 7.32-7.42 (m, 5 H, C₆H₅); ¹³C NMR (CDCl₃) δ 26.34, 26.87 (2 C, 2 CH₃C), 54.65 (C-6), 68.40 (C-5), 72.13 (CH₂Ph), 80.11, 81.67, 82.15 (C-2,3,4), 105.26 (C-1), 112.02 (C(CH₃)₂), 127.93-128.80, 137.21 (*C*₆H₅).

3-*O*-Acetyl-6-azido-6-deoxy-1,2-*O*-(1-methylethylidene)-5-*O*-trimethylsilyl-α-D-glucofuranose (10). Obtained as described above from 3-*O*-acetyl-1,2-*O*-(1-methylethylidene)-α-D-glucofuranose²⁸ (9) (0.262 g, 1.0 mmol). Purification by column chromatography (EtOAc/petroleum ether 1:4 v/v) afforded pure product 10 (0.318 g, 81% yield). Compound 10: liquid; R_f 0.70 (EtOAc/petroleum ether 1:4 v/v); [α]_D-37.2° (*c* 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.13 (s, 9 H, 3 CH₃Si), 1.30, 1.52 (2s, 6 H, 2 CH₃C), 2.11 (s, 3 H, CH₃COO), 3.35 (ddd, 1 H, J_{5,6a} = 2.6 Hz, J_{5,6b} = 5.4 Hz, J_{6a,6b} = 12.6 Hz, H-6b), 3.59 (dd, 1 H, H-6a), 4.05 (ddd, 1 H, J_{4,5} = 8.8 Hz, H-5), 4.05 (ddd, 1 H, J_{4,5} = 8.8 Hz, H-5), 4.24 (dd, 1 H, J_{3,4} = 2.7 Hz, H-3), 4.48 (d, 1 H, J_{1,2} = 3.7 Hz, H-2), 5.17 (d, 1 H, H-3), 5.84 (d, 1 H, H-1); ¹³C NMR (CDCl₃) δ 0.12 (3 C, 3 CH₃Si), 20.90 (CH₃COO), 26.16, 26.57 (2 C, 2 CH₃C), 54.92 (C-6), 68.53 (C-5), 76.00 (C-3), 78.72 (C-4), 82.88 (C-2), 104.71 (C-1), 112.22 (*C*(CH₃)₂), 169.40 (CH₃CO).

Anal. Calcd for $C_{14}H_{25}N_3O_6Si$ (359.45): C, 46.78; H, 7.01; N, 11.69. Found: C, 46.69; H, 6.89; N, 11.49.

5-Azido-5-deoxy-1,2-O-(1-methylethylidene)-3-O-trimethylsilyl- α -D-xylofuranose (12) and 1,2-O-(1-methylethylidene)-3,5-di-O-trimethylsilyl- α -D-xylofuranose (13). Obtained as described above from 1,2-O-(1-methylethyl-

idene)- α -D-xylofuranose³⁰ (11) (0.190 g, 1.0 mmol). The crude product was purified by column chromatography (EtOAc/petroleum ether 1:5 v/v) to afford products 12 and 13 as an inseparable mixture (0.274 g). Compound 12 : liquid; R_f 0.75 (EtOAc/petroleum ether 1:5 v/v); ¹H NMR (CDCl₃) δ 0.15 (s, 9 H, 3 CH₃Si), 1.30, 1.48 (2s, 6 H, 2 CH₃C), 3.35 (dd, 1 H, J_{4,5b} = 2.6 Hz, J_{5,6b} = 6.5 Hz, J_{6a,6b} = 12.3 Hz, H-5b), 3.55 (dd, 1 H, J_{4,5a} = 6.5 Hz, H-5a), 4.15 (d, 1 H, J_{3,4} = 2.7 Hz, H-3), 4.23 (ddd, 1 H, H-4), 4.36 (dd, 1 H, J_{1,2} = 3.6 Hz, H-2), 5.90 (d, 1 H, H-1); ¹³C NMR (CDCl₃) δ -0.21 (3 C, 3 CH₃Si), 26.26, 26.80 (2 C, 2 CH₃C), 49.29 (C-5), 75.13 (C-3), 79.11 (C-4), 85.41 (C-2), 105.00 (C-1), 111.77 (C(CH₃)₂).

Compound 13 was also synthesized by direct silylation of the diol 11: liquid; $[\alpha]_D$ -40.0° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.12, 0.15 (2s, 18 H, 6 CH₃Si), 1.32, 1.51 (2s, 6 H, 2 CH₃C), 3.73 (d, 2 H, J_{5,6a} = J_{5,6b} = 6.3 Hz, H-5a,5b), 4.20 (m, 2 H, H-3,4), 4.36 (dd, 1 H, J_{1,2} = 3.6 Hz, H-2), 5.91 (d, 1 H, H-1); ¹³C NMR (CDCl₃) δ -0.62, -0.21 (6 C, 2 CH₃Si), 26.35, 26.85 (2 C, 2 CH₃C), 59.49 (C-5), 75.29 (C-3), 81.09 (C-4), 85.56 (C-2), 105.00 (C-1), 111.51 (C(CH₃)₂).

Anal. Calcd for C₁₄H₃₀O₅Si₂ (334.55): C, 50.26; 9.04. Found: C, 50.73; H, 8.89.

5-Azido-5-deoxy-1,2-O-(1-methylethylidene)- α -D-xylofuranose (14). Obtained by the desilylation procedure from the mixture of 12 and 13 (0.274 g). The crude product was purified by column chromatography (EtOAc/petroleum ether 1:1 v/v) to afford the pure product 14 (0.158 g) in 73% yield [calculated from 1,2-O-(1-methylethylidene)- α -D-xylofuranose]. Compound 14: mp 61-62 °C (lit.³¹ mp 58.5-60 °C); R_f 0.72 (EtOAc/petroleum ether 1:1 v/v); [α]_D -34.2° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 1.32, 1.50 (2s, δ H, 2 CH₃C), 2.39 (d, 1 H, J_{3,OH} = 4.3 Hz, OH), 3.58 (dd, 1 H, J_{4,5b} = 2.8 Hz, J_{5,6b} = 6.3 Hz, J_{6a,6b} = 12.3 Hz, H-5b), 3.65 (dd, 1 H, J_{4,5a} = 6.3 Hz, H-5a), 4.26 (dd, 1 H, J_{3,4} = 2.8 Hz, H-3), 4.29 (ddd, 1 H, H-4), 4.52 (dd, 1 H, J_{1,2} = 3.7 Hz, H-2), 5.96 (d, 1 H, H-1); ¹³C NMR (CDCl₃) δ 26.16, 26.71 (2 C, 2 CH₃C), 49.19 (C-5), 74.90 (C-3), 78.63 (C-4), 85.36 (C-2), 104.81 (C-1), 112.05 (C(CH₃)₂).

Methyl 2,3-di-O-acetyl-6-azido-6-deoxy-4-O-trimethylsilyl- α -D-galactopyranoside (16), methyl 2,4-di-O-acetyl-6-azido-6-deoxy-3-O-trimethylsilyl- α -D-galactopyranoside (17) and methyl 2,3-di-O-acetyl-4,6-di-O-trimethylsilyl- α -D-galactopyranoside (18). Obtained by the azidosilylation procedure from methyl 2,3-di-O-acetyl- α -D-galactopyranoside³² (15) (0.278 g, 1.0 mmol). Purification by column chromatography (EtOAc/petroleum ether 1:3 v/v) afforded a fraction of two compounds (Rf 0.74), plus pure product 16 (Rf 0.64).

16: 0.252 g (67% yield); white solid, mp 92-93 °C; R_f 0.54 (EtOAc/petroleum ether 1:3 v/v); $[\alpha]_D$ +112.5° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.15 (s, 9 H, 3 CH₃Si), 2.08 (s,

6 H, 2 C H_3 COO), 3.16 (dd, 1 H, J_{5,6a} = 5.3 Hz, J_{6a,6b} = 12.4 Hz, H-6b), 3.43 (s, 3 H, C H_3 O), 3.53 (dd, 1 H, J_{5,6a} = 7.9 Hz, H-6a), 3.94 (ddd, 1 H, J_{4,5} = 0.5 Hz, H-5), 4.09 (dd, 1 H, J_{3,4} = 2.3 Hz, H-4), 4.98 (dd, 1 H, J_{1,2} = 2.8 Hz, H-1), 5.20 (dd, 1 H, J_{2,3} = 10.8 Hz, H-2), 5.31 (dd, 1 H, H-3); ¹³C NMR (CDCl₃) δ 0.22 (3 C, 3 CH_3 Si), 20.75, 20.90 (2 C, 2 CH_3 COO), 51.10 (C-6), 55.36 (CH_3 O), 68.21, 69.47, 69.83, 69.85 (4 C, C-2,3,4,5), 97.28 (C-1), 169.87, 170.21 (2 C, 2 CH_3 CO).

Anal. Calcd for $C_{14}H_{25}N_3O_7Si$ (375.45): C, 44.78; H, 6.71; N, 11.19. Found: C, 44.54; H, 6.75; N, 11.14.

A new chromatographic separation of the first fraction (R_f 0.74) using ether/petroleum ether (3:4 v/v) as the eluent allowed the isolation of compounds 17 and 18. 17: 0.028 g (7%); liquid; R_f 0.69 (ether/petroleum ether 1:1 v/v); $[\alpha]_D$ +84.3° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.11, 0.13 (2s, 18 H, 6 CH₃Si), 2.07, 2.08 (2s, 6 H, 2 CH₃COO), 3.38 (s, 3 H, CH₃O), 3.61 (d, 2 H, J_{5,6a} = J_{5,6b} = 6.9 Hz, H-6a,6b), 3.79 (dt, 1 H, J_{4,5} = 1.1 Hz, H-5), 4.20 (dd, 1 H, J_{3,4} = 2.2 Hz, H-4), 4.95 (d, 1 H, J_{1,2} = 2.8 Hz, H-1), 5.21 (dd, 1 H, J_{2,3} = 10.8 Hz, H-2), 5.28 (dd, 1 H, H-3); ¹³C NMR (CDCl₃) δ -0.10 (3 C, 3 CH₃Si), 20.76, 20.91 (2 C, 2 CH₃COO), 51.15 (C-6), 55.59 (CH₃O), 66.63, 68.21, 71.35, 71.53 (4 C, C-2,3,4,5), 97.42 (C-1), 170.50, 170.55 (2 C, 2 CH₃CO).

Anal. Calcd for $C_{14}H_{25}N_{3}O_{7}Si$ (375.45): C, 44.78; H, 6.71; N, 11.19. Found: C, 44.57; H, 6.51; N, 10.70. 18: 0.040 g (9%); liquid; R_{f} 0.60 (ether/petroleum ether 1:1 v/v); $[\alpha]_{D}$ +113.1° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.11 (s, 9 H, 3 CH₃Si), 2.11, 2.14 (2s, 6 H, 2 CH₃COO), 3.15 (dd, 1 H, $J_{5,6a}$ = 3.9 Hz, $J_{6a,6b}$ = 12.9 Hz, H-6b), 3.40 (s, 3 H, CH₃O), 3.45 (dd, 1 H, $J_{5,6a}$ = 8.6 Hz, H-6a), 4.04 (ddd, 1 H, $J_{4,5}$ = 1.1 Hz, H-5), 4.13 (dd, 1 H, $J_{2,3}$ = 9.9 Hz, $J_{3,4}$ = 3.7 Hz, H-3), 4.92 (dd, 1 H, $J_{1,2}$ = 3.6 Hz, H-2), 5.00 (dd, 1 H, H-1), 5.20 (dd, 1 H, H-4).

Anal. Calcd for $C_{17}H_{34}O_8Si_2$ (422.57): C, 48.32; H, 8.11. Found: C, 48.29; H, 8.03.

Methyl 4,6-diazido-2,3-di-O-benzyl-4,6-dideoxy- α -D-glucopyranoside (21), methyl 6-azido-2,3-di-O-benzyl-6-dideoxy- α -D-galactopyranoside (22) and methyl 4-azido-2,3-di-O-benzyl-4-deoxy- α -D-glucopyranoside (23). Methyl 2,3-di-O-benzyl- α -D-galactopyranoside ¹⁸ (20) (0.562 g, 1.50 mmol) was reacted as described above, using a smaller excess of trimethylsilyl azide (1.5 eq instead of 2.5 eq). After usual treatment, the crude mixture was applied at the top of a small silica gel column (EtOAc/petroleum ether 1:4 v/v) in order to separate the reaction products (R_f 0.62-0.75) from triphenylphosphine oxide and diisopropyloxycarbonyl-

hydrazine. The fraction containing the mixture of sugars was concentrated and treated overnight with potassium carbonate (0.500 g) in MeOH (25 mL). MeOH was then evaporated and the residue was dissolved in CH₂Cl₂ (50 mL), and washed with water (10 mL). After drying (Na₂SO₄) and evaporation the organic extract afforded a mixture of four compounds which were purified by column chromatography (EtOAc/petroleum ether 1:4 v/v, then 1:1 v/v, then 2:1 v/v). Compounds 21, 22, 23 and 20 (0.015 g, 3%) were successively eluted.

21 (0.020 g 3% yield) : liquid; R_f 0.62 (EtOAc/petroleum ether 1:4 v/v); $[\alpha]_D$ +109.1° (c 1.0, CHCl₃); 1 H NMR (CDCl₃) δ 3.41 (s, 3 H, CH₃O), 3.39-3.46 (m, 3 H, H-4,6a,6b), 3.57 (dd, 1 H, $J_{1,2}$ = 3.5 Hz, $J_{2,3}$ = 9.5 Hz, H-2) 3.64 (ddd, 1 H, $J_{4,5}$ = 10.2 Hz, $J_{5,6a}$ = 2.5 Hz, $J_{5,6b}$ = 5.3 Hz, H-5), 3.90 (dd, 1 H, $J_{3,4}$ = 9.2 Hz, H-3), 4.64 (d, 1 H, H-1), 4.66 and 4.80 (2d, 2 H, J = 12.0 Hz, CH₂Ph), 4.83 and 4.99 (2d, 2 H, J = 10.5 Hz, CH₂Ph), 7.31-7.40 (m, 10 H, 2 C₆H₅); 13 C NMR (CDCl₃) δ 51.83 (C-6), 55.68 (OCH₃), 62.50 (C-4), 69.20 (C-5), 73.43, 75.83 (2 C, 2 CH₂Ph), 79.85, 79.94 (C-2,3), 98.21 (C-1), 128.15-128.64, 137.85, 137.95 (12 C, 2 C₆H₅). HRMS Calcd for C₂₁H₂₅N₄O₄ (MH-N₂): 397.18758. Found: 397.18764.

Anal. Calcd for $C_{21}H_{24}N_6O_4$ (424.44): C, 59.42; H, 5.70; N, 19.80. Found: C, 59.11; H, 5.89; N, 19.07.

22 (0.020 g, 3% yield): mp 65-66 °C; R_f 0.81 (EtOAc/petroleum ether 1:1 v/v); $[\alpha]_D$ +20.0° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 3.27 (dd, 1 H, $J_{5,6b}$ = 4.5 Hz, $J_{6a,6b}$ = 12.7 Hz, H-6b), 3.41 (s, 3 H, CH_3O), 3.61 (dd, 1 H, $J_{5,6a}$ = 6.2 Hz, H-6a), 3.84 (dd, 1 H, $J_{1,2}$ = 3.3 Hz, $J_{2,3}$ = 9.3 Hz, H-2), 3.87 (dd, 1 H, $J_{3,4}$ = 3.0 Hz, H-3) 3.88 (ddd, 1 H, $J_{4,5}$ = 1.0 Hz, H-5), 3.92 (dd, 1 H, H-4), 4.70 (d, 1 H, H-1), 4.68 and 4.83 (2d, 2 H, $J_{5,5a}$ = 12.3 Hz, CH_2Ph), 4.71 and 4.84 (2d, 2 H, $J_{5,5a}$ = 11.4 Hz, CH_2Ph), 7.29-7.39 (m, 10 H, 2 C_6H_5); ¹³C NMR (CDCl₃) δ 51.31 (C-6), 55.55 (OCH₃), 68.25, 68.87 (C-4,5), 73.14, 75.58 (2 C, 2 CH_2Ph), 75.61, 77.36 (C-2,3), 98.61 (C-1), 127.94-128.63, 138.02, 138.30 (12 C, 2 C_6H_5).

Anal. Calcd for $C_{21}H_{25}N_3O_5$ (399.43): C, 63.14; H, 6.31; N, 10.52. Found: C, 63.46; H, 6.44; N, 10.34.

23¹⁸ (0.395 g, 66% yield): liquid; R_f 0.56 (EtOAc/petroleum ether 1:1 v/v); $[\alpha]_D$ +97.8° (c 1.0, CHCl₃) (lit.¹⁸ $[\alpha]_D$ +90°); ¹H NMR (CDCl₃) δ 1.78 (t, 1 H, J_{6a,OH} = J_{6b, OH}, = 5.9 Hz, 3.36 (s, 3 H, CH₃O), 3.48 (m, 1 H, H-5), 3.51 (dd, 1 H, J_{1,2} = 3.5 Hz, J_{2,3} = 9.5 Hz, H-2), 3.52 (dd, 1 H, J_{3,4} = 9.5 Hz, H-3), 3.71 (ddd, 1 H, J_{5,6b} = 5.1 Hz, J_{6a,6b} = 11.0 Hz, H-6b), 3.80 (ddd, 1 H, J_{5,6a} = 2.4 Hz, H-6a), 3.95 (dd, 1 H, J_{4,5} = 9.5 Hz, H-4), 4.58 (d, 1 H, H-1), 4.65 and 4.80 (2d, 2 H, J = 10.6 Hz, CH₂Ph), 4.83 and 4.97 (2d, 2 H, J = 10.6 Hz, CH₂Ph), 7.29-7.41 (m, 10 H, 2 C₆H₅); ¹³C NMR (CDCl₃) δ

55.49 (OCH₃), 61.57 (C-4), 61.96 (C-6), 70.70 (C-5), 73.39, 75.77 (2 C, 2 CH₂Ph), 79.96, 80.00 (C-2,3), 98.31 (C-1), 127.92-128.61, 137.94, 138.09 (12 C, 2 C₆H₅).

Methyl 2,3-di-O-benzyl-4,6-bis-O-trimethylsilyl-α-D-galactopyranoside (24). The general silylation procedure was applied to 20^{18} (0.075 g, 0.20 mmol). Compound 23 (0.093 g, 90% yield) was recovered as a colorless liquid: R_f 0.72 (EtOAc/petroleum ether 1:4 v/v); [α]_D +30.8° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.10 and 0.13 (2s, 18 H, 6 CH₃Si), 3.40 (s, 3 H, CH₃O), 3.57 (dd, 1 H, J_{5,6b} = 5.9 Hz, J_{6a,6b} = 8.9 Hz, H-6b), 3.66 (bd, 1 H, H-6a), 3.73 (bd, 1 H, H-5), 3.81 (dd, 1 H, J_{2,3} = 10.1 Hz, J_{3,4} = 2.6 Hz, H-3), 3.93 (dd, 1 H, J_{1,2} = 3.4 Hz, H-2), 4.18 (dd, 1 H, J_{4,5} = 0.7 Hz, H-4), 4.71 (d, 1 H, H-1), 4.69 and 4.82 (2d, 2 H, J = 12.3 Hz, CH₂Ph), 4.74 and 4.86 (2d, 2 H, J = 11.4 Hz, CH₂Ph), 7.28-7.37 (m, 10 H, 2 C₆H₅); ¹³C NMR (CDCl₃) δ -0.46 and 0.60 (2s, 6 C, 6 CH₃Si), 55.33 (OCH₃), 61.10 (C-6), 69.61, 71.14 (C-4,5), 73.37, 73.73 (2 C, 2 CH₂Ph), 75.98, 76.47 (C-2,3), 98.94 (C-1), 127.45-128.40, 138.60, 138.76 (12 C, 2 C₆H₅). HRMS Calcd for C₂₄H₄₂O₆Si₂: 518.25199. Found: 518.25.019.

Anal. Calcd for C₂₇H₄₂O₆Si₂ (518.78): C, 62.51; H, 8.16. Found: C, 61.84; H, 8.07.

Methyl 6-azido-2,3-di-O-benzyl-6-deoxy-4-O-trimethylsilyl- α -D-galactopyranoside (26). Obtained, as described above, from 22 (0.080 g, 0.20 mmol) as a colorless liquid (0.086 g, 91% yield): R_f 0.66 (EtOAc/petroleum ether 1:4 v/v); $[\alpha]_D$ +30.8° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.14 (s, 9 H, 3 CH₃Si), 3.15 (dd, 1 H, J_{5,6b} = 5.3 Hz, J_{6a,6b} = 12.4 Hz, H-6b), 3.48 (s, 3 H, CH₃O), 3.50 (ddd, 1 H, J_{5,6a} = 7.9 Hz, H-6a), 3.81 (dd, 1 H, J_{2,3} = 10.1 Hz, J_{3,4} = 2.5 Hz, H-3), 3.82 (m, 1 H, H-5), 3.91 (dd, 1 H, J_{1,2} = 3.3 Hz, H-2), 4.02 (m, 1 H, H-4), 4.71 (d, 1 H, H-1), 4.67 and 4.83 (2d, 2 H, J = 12.1 Hz, CH₂Ph), 4.73 and 4.87 (2d, 2 H, J = 11.3 Hz, CH₂Ph), 7.30-7.35 (m, 10 H, 2 C₆H₅); ¹³C NMR (CDCl₃) δ - 0.60 (s, 3 C, 3 CH₃Si), 51.49 (C-6), 55.56 (OCH₃), 70.15 (C-4), 70.79 (C-5), 73.42, 74.02 (2 C, 2 CH₂Ph), 75.63, 77.58 (C-2,3), 98.94 (C-1), 127.61-128.47, 138.39, 138.44 (12 C, 2 C₆H₅).

Anal. Calcd for $C_{24}H_{33}N_{3}O_{5}Si$ (471.61): C, 61.12; H, 7.05; N, 8.91. Found: C, 61.16; H, 7.23; N, 8.67.

Methyl 4-azido-2,3-di-O-benzyl-4-deoxy-6-O-trimethylsilyl- α -D-glucopyranoside (27). Obtained, as described above, from 23 (0.080 g, 0.20 mmol) as a colorless liquid (0.088 g, 93 % yield): R_f 0.66 (EtOAc/petroleum ether 1:4 v/v); [α]_D +77.1° (c 1.0, CHCl₃); ¹H NMR (CDCl₃) δ 0.14 (s, 9 H, 3 CH₃Si), 3.37 (s, 3 H, CH₃O), 3.44 (ddd, 1 H, J_{4,5} = 10.2 Hz, J_{5,6a} = 3.2 Hz, J_{5,6b} = 1.8 Hz, H-5), 3.50 (dd, 1 H, J_{1,2} = 3.5 Hz, J_{2,3} = 9.5 Hz, H-2), 3.61 (dd, 1 H, J_{3,4} = 9.5 Hz, H-4), 3.75 (dd, 1

H, $J_{6a,6b} = 11.5$ Hz, H-6b), 3.84 (ddd, 1 H, H-6a), 3.90 (dd, 1 H, H-3), 4.64 (d, 1 H, H-1), 4.65 and 4.80 (2d, 2 H, J = 12.1 Hz, CH_2Ph), 4.83 and 4.99 (2d, 2 H, J = 10.5 Hz, CH_2Ph), 7.31-7.44 (m, 10 H, 2 C_6H_5); ¹³C NMR (CDCl₃) δ -0.47 (s, 3 C, 3 CH_3Si), 55.37 (OCH₃), 61.29 (C-4), 61.65 (C-6), 70.28 (C-5), 73.38, 75.86 (2 C, 2 CH_2Ph), 79.90, 80.22 (C-2,3), 98.33 (C-1), 127.93-128.53, 138.02, 138.14 (12 C, 2 C_6H_5).

Anal. Calcd for C₂₄H₃₃N₃O₅Si (471.61): C, 61.12; H, 7.05; N, 8.91. Found: C, 61;13; H, 6.89; N, 8.77.

ACKNOWLEDGEMENT

The authors thank Dr Denis Bouchu (Université Lyon I) for performing mass spectrometry.

REFERENCES

- 1. N. R. Williams and J. D. Wander in *The Carbohydrates*; W. Pigman and D. Horton Eds, Academic Press: New York, 1980; Vol IB, p 761.
- A. K. Mallans in Carbohydrate Chemistry, J. F. Kennedy Ed, Clarendon Press: Oxford, 1988, p 73.
- 3. K. Doboshi, K. Nagaoka, Y. Watanabe, M. Nishida, H. Naganawa, T. Takita, T. Takenshi, and H. Umezawa, J. Antibiot., 38, 1185 (1986).
- 4. J. Lehmann, B. Rob, and H.-A. A. Wagenknecht, *Carbohydr. Res.*, 278, 176 (1995).
- 5. P. Vogel and P. Gries, J. Carbohydr. Chem., 13, 37 (1994).
- V. Maunier, P. Boullanger, D. Lafont, and Y. Chevalier, Carbohydr. Res., 299, 49 (1997).
- A. Mezher, L Hough, and A. C. Richardson, Carbohydr. Res., 216, 271 (1991).
- 8. A. Fürstner, J. Baumgartner and D. N. Jumbam, J. Chem. Soc., Perkin Trans. 1, 131 (1993).
- J. L. J. Blanco, J. M. G. Fernández, A. Gadelle, and J. Defaye, Carbohydr. Res., 303, 367 (1997).
- 10. M. V. Reddington, J. Chem. Soc., Perkin Trans. 1, 143 (1998).
- W. R. Kobertz, C. R. Bertozzi, and M. D. Bednarski, J. Org. Chem., 61, 1894 (1996).
- 12. F. Morís-Varas, X-H Qian, and C-H Wong, J. Am. Chem. Soc., 118, 7647 (1996).
- H. P. Wessel, D. Banner, K. Gubernator, H. Hilpert, K. Müller, and T. Tschopp, Angew. Chem. Int. Ed. Engl., 36, 751 (1997).
- 14. M. C. Viaud, and P. Rollin, Synthesis, 130 (1990).
- 15. A. Guiller, C. H. Gagnieu, and H. Pacheco, Tetrahedron Lett., 26, 6343 (1985).
- 16. J. Fuentes, M. Angulo, and M. Angeles Pradera, *Tetrahedron Lett.*, 39, 7149 (1998).

- 17. J. A. Secrist and K. D. Barnes, J. Org. Chem., 45, 4526 (1980).
- 18. T. Ogawa, K. Katano, and M. Matsui, *Tetrahedron*, 36, 2727 (1980).
- 19. S. Knapp, Y. H. Choe, and E. Reilly, *Tetrahedron Lett.*, 34, 4443 (1993).
- 20. H. H. Brandstetter and E. Zbiral, Helv. Chim. Acta, 61, 1832 (1978).
- 21. H. H. Brandstetter and E. Zbiral, Helv. Chim. Acta, 63, 328 (1980).
- 22. D. J. Bell and J. Lorber, J. Chem. Soc., 453 (1940).
- 23. M. Von Itzstein and I. D. Jenkins, J. Chem. Soc., Perkin Trans. 1, 437 (1986).
- 24. I. Mathieu-Pelta and S. A. Evans, Jr., J. Org. Chem., 57, 3409 (1992).
- M. Hendrix, P. B. Alper, E. S. Priestley, and C. H. Wong, Angew. Chem., Int. Ed. Engl., 36, 95 (1997).
- 26. R. L. Whistler and S. J. Kazeniac, J. Am. Chem. Soc., 76, 3044 (1954).
- C. H. Heathcock, C. T. White, J. J. Morrison, and D. Van Derveer, J. Org. Chem., 46, 1296 (1981).
- 28. K. Josephson, Liebigs Ann. Chem., 472, 217 (1929).
- 29. H. Saeki and E. Ohki, Chem. Pharm. Bull., 16, 2471 (1968).
- 30. B. R. Baker and R. E. Schaub, J. Am. Chem. Soc., 77, 5900 (1957).
- 31. W. A. Szarek and J. K. N. Jones, Can. J. Chem., 43, 2345 (1965).
- 32. D. J. Bell and G. D. Greville, J. Chem. Soc., 1136 (1955).