



Contents lists available at ScienceDirect

Tetrahedron Letters

journal homepage: www.elsevier.com/locate/tetlet

New catalytic route for the synthesis of an optically active tetralone-derived amine for rotigotine

Christopher J. Cobley*, George Evans, Tamara Fanjul, Shaun Simmonds, Amy Woods

Chirotech Technology Centre, Dr. Reddy's Laboratories, Unit 410 Cambridge Science Park, Milton Road, Cambridge CB4 0PE, UK

ARTICLE INFO

Article history:

Received 10 December 2015

Revised 11 January 2016

Accepted 18 January 2016

Available online xxx

Keywords:

Rotigotine

Chiral amine

Asymmetric hydrogenation

ABSTRACT

Rotigotine is a launched drug for the treatment of Parkinson's disease and restless legs syndrome. The key steps of an alternative route for the synthesis of rotigotine have been demonstrated. Formation of a prochiral enamide, asymmetric hydrogenation of the enamide with high enantioselectivity, and reduction of the resulting amide to an amine have been proved to work successfully. The best conditions screened to date for the asymmetric hydrogenation of enamide **9** to amide **10** were with [(RuCl((*R*)-T-BINAP))₂(μ-Cl)₃][NH₂Me₂] at 25 bar H₂ and 30 °C (500:1 S/C ratio, 99% conversion, 91% ee *S*). Reduction of amide **10** to amine **5** was best achieved with Red-Al giving 95% conversion.

© 2016 Elsevier Ltd. All rights reserved.

Introduction

Rotigotine is a launched drug of the non-ergoline class of medications that acts as a dopamine agonist for the treatment of Parkinson's disease and restless legs syndrome. It was developed by Aderis Pharmaceuticals and licensed by UCB S.A. and it has been in use in Europe since 2006.¹ Currently, the majority of reported syntheses of the chiral amine intermediates used in the preparation of this drug involve the classical resolution of racemic amines^{1,2} (and hence a maximum theoretical yield of 50%) as shown in [Scheme 1](#).

The development of an efficient asymmetric synthesis would lead to obvious cost and yield benefits. One approach, reported in a recent publication, is the synthesis of a chiral tetralone primary amine as a key intermediate via a multi-enzymatic approach of ene-reductases (ERs) and alcohol dehydrogenases (ADHs).³ A different strategy that has been patented is a chiral auxiliary based route using α-methylbenzylamine.⁴ Herein, we report the preparation of a prochiral enamide (**9**), from the appropriately substituted tetralone (**3**) and propionamide, followed by asymmetric hydrogenation which gives access to a chiral amide (**10**) in high yield and enantioselectivity. Subsequent reduction of this chiral amide yields the necessary secondary amine intermediate (**5**) in a highly efficient and selective fashion ([Scheme 2](#)). The remaining downstream steps to rotigotine, as shown in [Scheme 1](#), have been previously reported.⁵

The key elements of the overall alternative process are a highly selective asymmetric hydrogenation reaction that uses a directing group of relevance to the downstream product of interest, a concise overall route where each step is high yielding with an efficient amide reduction as the final step, minimising any loss of enantioselectivity.

Results and discussion

The reaction between a substituted β-tetralone and a primary amide to yield the corresponding enamide has been widely reported.⁶ In this case, the specific combination shown in [Scheme 3](#) is required.^{2b,7}

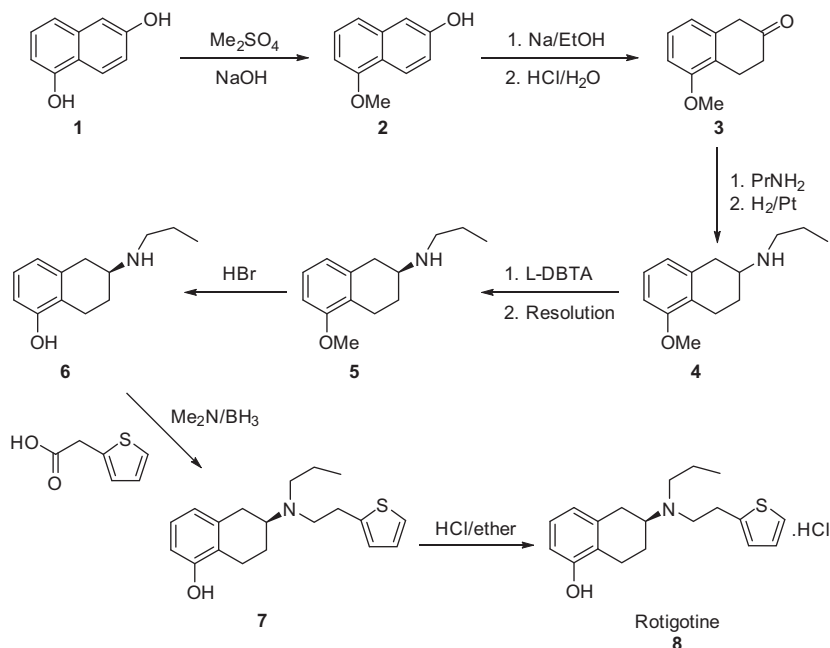
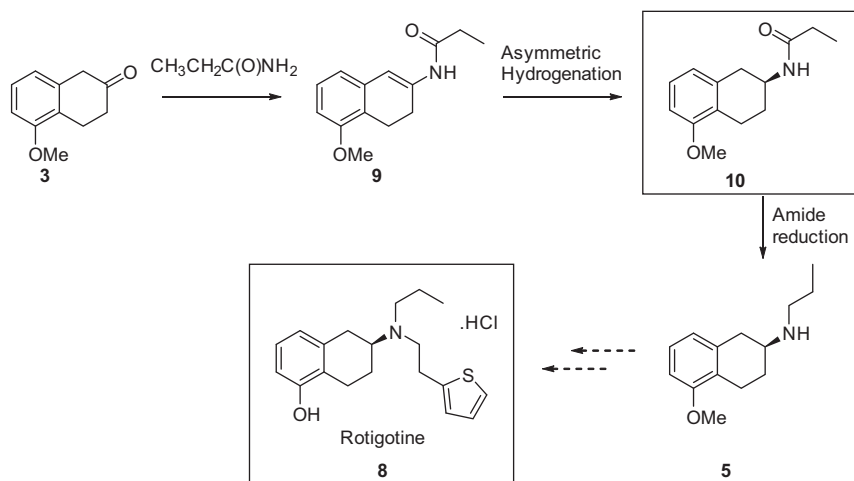
Enamide **9** was prepared on a 16 g scale showing quantitative conversion by ¹H NMR spectroscopy. Water was removed under Dean–Stark conditions using pTSA as an acidic catalyst. Recrystallization was achieved from the reaction solvent in an unoptimized 80% isolated yield.

The asymmetric hydrogenation of tetralone-derived enamides has been reported in recent years.⁸ However, the reduction of the preferred enamide **9** leading to rotigotine (**8**) had not been reported prior to this work⁷ ([Scheme 4](#)).

A screen of Rh and Ru catalyst was undertaken ([Table 1](#) shows representative results of a total of 4 Rh and 26 Ru precatalysts). Hydrogenation of **9** with Rh complexes such as [((*R,R*)-Ph-BPE)Rh(cod)]BF₄ and [((*R*)-PhanePhos)Rh(cod)]BF₄ showed good conversion, albeit at a low enantioselectivity (entries 1 and 2). On the other hand, the hydrogenation carried out with Ru dimer species showed a low conversion at 10 bar hydrogen pressure, but a very

* Corresponding author. Tel.: +44 1223 728010; fax: +44 1223 506701.

E-mail address: ccobley@drreddys.com (C. J. Cobley).

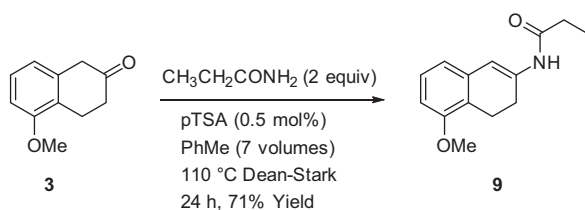
Scheme 1. Synthesis of rotigotine via classical resolution.^{1,2}

Scheme 2. Catalytic, enantioselective synthesis of rotigotine.

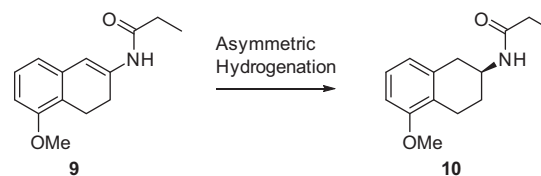
high enantioselectivity; leaving promising scope for process development and optimization (entry 5). Increasing both pressure and temperature for precatalyst $[\text{RuCl}_2((R)\text{-BINAP})_2]\text{-NET}_3$ had a beneficial effect in terms of conversion but with a concomitant reduction in enantioselectivity (cf. entries 5 and 6 and entries 6 and 7).

Different Ru precursors and BINAP derivatives were tried giving good conversions and ees of up to 99%.

Similar results were obtained with BINAP derived ligands (entries 8, 9, 12, and 13) and ligands such as DuPhos and BIPHEP (Fig. 1). This, in addition to the economic attractiveness of BINAP derivatives relative to these other ligand families, precatalyst



Scheme 3.



Scheme 4.

Table 1
Asymmetric hydrogenation^a screening results

Entry	Precatalyst	P (bar)	T (°C)	Conv. (%)	ee (%)
1	[(<i>R,R</i>)-Ph-BPE Rh(cod)]BF ₄	10	20	89	28 (S)
2	[(<i>R</i>)-PhanePhos Rh(cod)]BF ₄	10	20	98	42 (S)
3	[(<i>R,R</i>)-Et-DuPhos Rh(cod)]BF ₄	10	20	17	8 (S)
4	[(<i>S,S</i>)-Et-FerroTane Rh(cod)]BF ₄	10	20	43	4 (S)
5	[RuCl ₂ ((<i>R</i>)-BINAP) ₂ ·NEt ₃]	10	30	13	>99 (S)
6	[RuCl ₂ ((<i>R</i>)-BINAP) ₂ ·NEt ₃]	25	30	51	>99 (S)
7	[RuCl ₂ ((<i>R</i>)-BINAP) ₂ ·NEt ₃]	25	40	>99	87 (S)
8	[RuCl((<i>R</i>)-BINAP) ₂ (μ-Cl) ₃] [NH ₂ Me ₂]	25	30	92	87 (S)
9	[RuCl((<i>R</i>)-T-BINAP) ₂ (μ-Cl) ₃] [NH ₂ Me ₂]	25	30	99	89 (S)
10	[RuCl((<i>R</i>)-DM-BINAP) ₂ (μ-Cl) ₃] [NH ₂ Me ₂]	25	50	99	85 (S)
11	[Ru((<i>R</i>)-DM-BINAP)(OAc) ₂]	25	30	93	84 (S)
12	[Ru((<i>R,R</i>)-Et-DuPhos)(TFA) ₂]	10	30	99	90 (S)
13	[Ru((<i>R</i>)-MeO-BIPHEP)(TFA) ₂]	10	30	99	88 (S)

^a Reaction conditions: 0.14 M in MeOH, S/C ratio 200:1, 18 h.

[RuCl((*R*)-T-BINAP)₂(μ-Cl)₃][NH₂Me₂] was chosen to conduct a further study increasing the pressure and molar substrate to catalyst (S/C) ratio (Table 2). Increasing pressure had a positive effect on the rate of hydrogenation of enamide **9**; the reaction at 25 bar at 500 S/C ratio reached completion in 41 h, whereas the reaction at 100 bar using the same catalyst loading proceeded to 98% conversion in only 18 h. The reaction proceeded to full conversion with a high ee at 1000 S/C and 50 °C. At 5000 S/C, the conversion only reached 69%.

A hot recrystallization of **10** from ethyl acetate increased the enantioselectivity from 91% to 98% with an unoptimized recovery of 28%.

The downstream reduction of amide to amine (Scheme 5) has been published for a range of tetrahydronaphthalen-2-amides with a variety of reducing agents (including BH₃·THF, BH₃·SMe₂, Zn/silane, LiAlH₄, DIBAL-H, and Red-Al).⁹

Reduction of amide **10** (R = Et) with 4 different reducing agents was demonstrated. Reaction of **10** with Red-Al, LiAlH₄, BH₃·THF, and DIBAL-H resulted in the formation of **5** (R = Et) in 67–95% conversion. The reaction with BH₃·NEt₃ was not successful, with only starting material observed (Table 3).

Amine **5** was isolated as the hydrochloric salt **5**·HCl in 98.56% ee concluding that the loss of enantioselectivity did not occur in the downstream chemistry.

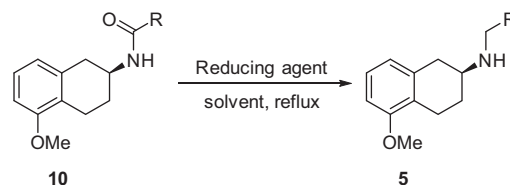
Table 2
Asymmetric hydrogenation results at different catalyst loadings and conditions

Precatalyst	S/C	P (bar)	T (°C)	Conv. (%)	ee (%)
[RuCl((<i>R</i>)-T-BINAP) ₂ (μ-Cl) ₃] [NH ₂ Me ₂]	500 ^a	25	30	99	91 (S)
	500 ^b	100	30	98	91 (S)
	1000 ^b	25	50	99	85 (S)
	5000 ^c	100	50	69	88 (S)

^a Reaction condition: 0.43 M in MeOH, 41 h.

^b Reaction condition: 0.22 M in MeOH, 22 h.

^c Reaction condition: 0.22 M in MeOH, 18 h.

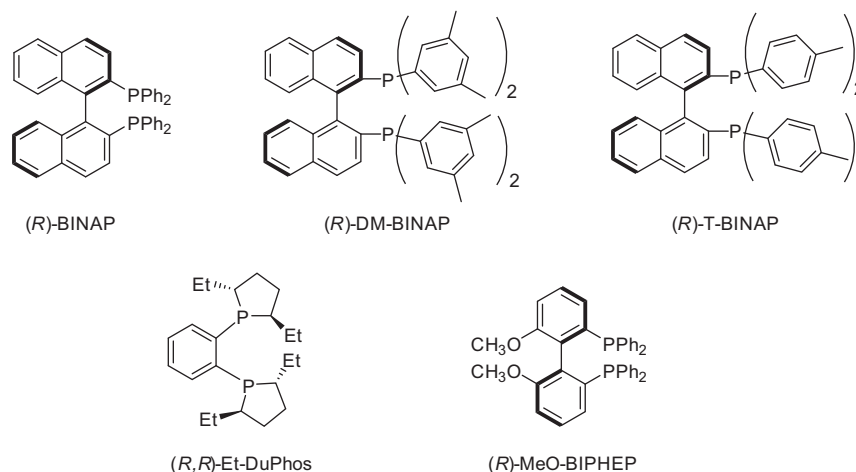
**Scheme 5.****Table 3**
Conversion results for the reduction of amide **10** to amine **5**

Reducing agent	Conversion (%)
Red-Al	95
LiAlH ₄	89
BH ₃ ·THF	67
DIBAL-H	78
BH ₃ ·NEt ₃	No reaction

Reaction conditions: see ESI details

Conclusions

The key steps of an alternative route for the synthesis of an achiral amine intermediate for rotigotine have been demonstrated. The formation of a prochiral enamide, its asymmetric hydrogenation with good enantioselectivity, and the reduction of the resulting amide to an amine have been proved to work successfully. The best conditions screened to date for the asymmetric hydrogenation of enamide **9** to amide **10** were with [RuCl((*R*)-T-BINAP)₂(μ-Cl)₃][NH₂Me₂] at 25 bar H₂ and 30 °C (500:1 S/C ratio, 99% conversion, 91% ee S). When the S/C ratio was lowered to

**Figure 1.**

1000:1 at 25 bar H₂ and 50 °C, the conversion remained high but the enantioselectivity dropped to 85%. Reduction of amide 10 to amine 5 was best achieved with Red-Al giving 95% conversion.

Experimental

Please refer to E.S.I. for experimental details.

Acknowledgement

This work has been funded by Dr. Reddy's Laboratories.

Supplementary data

Supplementary data associated with this article can be found, in the online version, at <http://dx.doi.org/10.1016/j.tetlet.2016.01.060>.

References and notes

- Integrity—Prous Science.
- (a) Han, X.; Zhang, X.; Fan, Q. *Xiandai Yaowu Yu Linchuang* **2011**, *5*, 375–377; (b) Ates, C.; Schule, A.; Palacio, M.; Deutsch, P.; Vasselin, D.; Carly, N.; Phadtare, G.; Delatinne, J.; Escudero Hernandez, M. L.; Pinilla, V. WO 2011161255; (c) Wang, K.; Wang, M. CN 103058985 (2013).
- Brenna, E.; Gatti, F. G.; Malpezzi, L.; Monti, D.; Parmeggiani, F.; Sacchetti, A. *J. Org. Chem.* **2013**, *78*, 4811–4822.
- Huang, Q.; Huang, Q.; Lou, M. US2014046095.
- (a) Webster, R.; Boyer, A.; Fleming, M. J.; Lautens, M. *Org. Lett.* **2010**, *12*, 5418–5421; (b) Banfi, A.; Belogi, G.; Fuganti, C.; Pizzocaro, R. WO 2010035111; (c) Marquillas Olondriz, F.; Pomares Marco, M. WO 2010043571; (d) Kandula, M. WO 2013168033; (e) Meng, Q.; Yang, M.; Wang, T.; Wang, Q.; Li, J.; Ruan, Z. WO 2013000273; (f) Vellanki, S. R. P.; Rane, D. R.; Patil, S. N.; Vantikomm, J.; Datta, D. IN 2009CH00795; (g) Banfi, A.; Belogi, G.; Fuganti, C.; Pizzocaro, R. IT 1392387; (h) Minaskanian, G.; Rippel, K. WO 2001038321; (i) Pomares, M.; Marquillas, F. WO 2010066755; (j) Avdagic, A. WO 2009056791; (k) Biswas, S.; Zhang, S.; Fernandez, F.; Ghosh, B.; Zhen, J.; Kuzhikandathil, E.; Reith, M. E. A.; Dutta, A. K. *J. Med. Chem.* **2008**, *51*, 101–117; (l) Weichert, D.; Banerjee, A.; Hiller, C.; Kling, R. C.; Huebner, H.; Gmeiner, P. *J. Med. Chem.* **2015**, *58*, 2703–2717; (m) Seiler, M. P.; Stoll, A. P.; Closse, A.; Frick, W.; Jatton, A.; Vigouret, J. M. *J. Med. Chem.* **1986**, *29*, 912–917; (n) Zuo, H.; Du, H.; Li, M.; Liu, X.; Du, Z.; Wang, S. *Beijing Huagong Daxue Xuebao, Ziran Kexueban* **2007**, *34*, 575–578; (o) Sonesson, C.; Barf, T.; Nilsson, J.; Dijkstra, D.; Carlsson, A.; Svensson, K.; Smith, M. W.; Martin, I. J.; Duncan, J. N. *J. Med. Chem.* **1995**, *38*, 1319–1329; (p) Hacksell, U.; Svensson, U.; Nilsson, J. L. G.; Hjorth, S.; Carlsson, A.; Wikstroem, H.; Lindberg, P.; Sanchez, D. *J. Med. Chem.* **1979**, *22*, 1469–1475; (q) Yao, B.; Ji, H.; Cao, Y.; Zhou, Y.; Zhu, J.; Lu, J.; Li, Y.; Chen, J.; Zheng, C.; Jiang, Y.; Liang, R.; Tang, H. *J. Med. Chem.* **2007**, *50*, 5293–5300.
- (a) Guthrie, D. B.; Damodaran, K.; Curran, D. P.; Wilson, P.; Clark, A. J. *J. Org. Chem.* **2009**, *74*, 4262–4266; (b) Brown, D. A.; Kharkar, P. S.; Parrington, I.; Reith, M. E. A.; Dutta, A. K. *J. Med. Chem.* **2008**, *51*, 7806–7819; (c) Imanishi, M.; Nakajima, Y.; Tomishima, Y.; Hamashima, H.; Washizuka, K.; Sakurai, M.; Matsui, S.; Imamura, E.; Ueshima, K.; Yamamoto, T.; Yamamoto, N.; Ishikawa, H.; Nakano, K.; Unami, N.; Hamada, K.; Matsumura, Y.; Takamura, F.; Hattori, K. *J. Med. Chem.* **2008**, *51*, 4804–4822; (d) Lucarini, S.; Bedini, A.; Spadoni, G.; Piersanti, G. *Org. Biomol. Chem.* **2008**, *6*, 147–150; (e) Guastavino, J. F.; Barolo, S. M.; Rossi, R. A. *Eur. J. Org. Chem.* **2006**, *17*, 3898–3902; (f) Lapars, A.; Campos, K. R.; Chen, C.; Volante, R. P. *Org. Lett.* **2005**, *7*, 1185–1188; (g) Renaud, J. L.; Dupau, P.; Hay, A.-E.; Guingouain, M.; Dixneuf, P. H.; Bruneau, C. *Adv. Synth. Catal.* **2003**, *345*, 230–238; (h) Dupau, P.; Bruneau, C.; Dixneuf, P. H. *Adv. Synth. Catal.* **2001**, *343*, 331–334; (i) Dupau, P.; Le Gendre, P.; Bruneau, C.; Dixneuf, P. H. *Synlett* **1999**, 1832–1834; (j) Devocelle, M.; Mortreux, A.; Agbossou, F.; Dormoy, J.-R. *Tetrahedron Lett.* **1999**, *40*, 4551–4554; (k) Tschäen, D. M.; Abramson, L.; Cai, D.; Desmond, R.; Dolling, U.-H.; Frey, L.; Karady, S.; Shi, Y.-J.; Verhoeven, T. R. *J. Org. Chem.* **1995**, *60*, 4324–4330; (l) Bochu, C.; Couture, A.; Lablache-Combier, A. *Tetrahedron* **1988**, *44*, 1959–1970.
- While this work was ongoing, a similar approach was being independently investigated by chemists at UCB. Vasselin, D.; Carly, N.; Ates, C.; WO2011095539.
- (a) Liu, G.; Liu, X.; Cai, Z.; Jiao, G.; Xu, G.; Tang, W. *Angew. Chem., Int. Ed.* **2013**, *52*, 4235–4238; (b) Patureau, F. W.; de Boer, S.; Kuil, M.; Meeuwissen, J.; Breuil, P.-A. R.; Siegler, M. A.; Spek, A. L.; Sandee, A. J.; de Bruin, B.; Reek, J. N. H. *J. Am. Chem. Soc.* **2009**, *131*, 6683–6685; (c) Sala, X.; Serrano, I.; Rodriguez, M.; Romero, I.; Llobet, A.; van Leeuwen, P. W. N. M. *Catal. Commun.* **2008**, *9*, 117–119; (d) Sandee, A. J.; Van der Burg, A. M.; Reek, J. N. H. *Chem. Commun.* **2007**, 864–866; (e) Jiang, X.-B.; Lefort, L.; Goudriaan, P. E.; de Vries, A. H. M.; van Leeuwen, P. W. N. M.; de Vries, J. G.; Reek, J. N. H. *Angew. Chem., Int. Ed.* **2006**, *45*, 1223–1227; (f) Bernsmann, H.; van den Berg, M.; Hoen, R.; Minnaard, A. J.; Mehler, G.; Reetz, M. T.; de Vries, J. G.; Feringa, B. L. *J. Org. Chem.* **2005**, *70*, 943–951; (g) Hoen, R.; Van den Berg, M.; Bernsmann, H.; Minnaard, A. J.; de Vries, J. G.; Feringa, B. L. *Org. Lett.* **2004**, *6*, 1433–1436; (h) Duprat De Paule, S.; Champion, N.; Vidal, V.; Genet, J. P.; Dellis, P. Fr. Demande 2003, FR 2830254 A1; (i) Guillen, F.; Rivard, M.; Toffano, M.; Legros, J.-Y.; Daran, J.-C.; Fiaud, J.-C. *Tetrahedron* **2002**, *58*, 5895–5904; (j) Tang, W.; Chi, Y.; Zhang, X. *Org. Lett.* **2002**, *4*, 1695–1698; (k) Burk, M. J.; Malan, C. G. WO 2001094359; (l) Argouarch, G.; Samuel, O.; Riant, O.; Daran, J.-C.; Kagan, H. B. *Eur. J. Org. Chem.* **2000**, *16*, 2893–2899; (m) Argouarch, G.; Samuel, O.; Kagan, H. B. *Eur. J. Org. Chem.* **2000**, *16*, 2885–2891; (n) Zhang, Z.; Zhu, G.; Jiang, Q.; Xiao, D.; Zhang, X. *J. Org. Chem.* **1999**, *64*, 1774–1775.
- BH₃·THF: (a) DeMarinis, R. M.; Shah, D. H.; Hall, R. F.; Hiebla, J. P.; Pendleton, R. G. *J. Med. Chem.* **1982**, *25*, 136–141; LiAlH₄: (b) Indra, B.; Matsunaga, K.; Hoshino, O.; Suzuki, M.; Ogasawara, H.; Ohizumi, Y. *Eur. J. Pharmacol.* **2002**, *437*, 173–178; Red-Al: (c) Cannon, J. G.; Walker, K. A.; Montanari, A.; Long, J. P.; Flynn, J. R. *J. Med. Chem.* **1990**, *33*, 2000–2006; Zn/silane: (d) Das, S.; Addis, D.; Zhou, S.; Junge, K.; Beller, M. *J. Am. Chem. Soc.* **2010**, *132*, 1770–1771.