

## Synthesis and Activity of Fluorinated Derivatives of Sulindac Sulphide and Sulindac Sulphone

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### Abstract

The synthesis of fluorinated derivatives of the sulphide and sulphone metabolites of sulindac, a non-steroidal anti-inflammatory agent with chemopreventative activity, is reported.

The key step in the synthesis is a Pummerer-rearrangement of the parent sulindac sulfoxide using (diethylamino)sulphur trifluoride as an activating agent and as a source of the fluoride nucleophile. The reaction leads to the formation of the 4-fluoromethylthio and 4-fluoromethylsulphonyl derivatives of sulindac (**4a** and **4b**, respectively). Sulindac sulphide is 2.5 times more potent as a COX-1 inhibitor compared with its fluorinated counterpart **4a**. The sulphones were inactive as COX-1 inhibitors, and none of the compounds inhibited COX-2 concentrations up to 0.1 mM. Cytotoxicity assays showed that **4a** and **4b** were as cytotoxic as sulindac sulphide and sulindac sulphone on 3719 colorectal carcinoma cell lines. Compound **4b** was the most potent compound on RCA cells with an IC<sub>50</sub> of 95 µM (sulindac sulphide 140 µM, sulindac sulphone 175 µM).

Fluorinated sulindac derivatives warrant further investigation since we have shown that cytotoxic activity can be retained or even increased independently from COX-inhibitory properties. This could help minimize the undesired side-effects associated with chronic sulindac administration.

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Inhibition of arachidonic acid pathways is one way to interfere with carcinogenesis. This has been studied extensively in numerous in-vitro and in-vivo models, and in the case of acetylsalicylic acid, activity has been documented in man (Kahn & Morrison 1997). One particularly interesting compound from the class of non-steroidal anti-inflammatory agents with respect to cancer prevention is sulindac (**1a**) (Kelloff et al 1985; Wattenberg 1997; Hakama 1998). This agent suppresses familial adenomatous polyposis (FAP), a premalignant disease that can lead to colon cancer (Giardiello et al 1993). The active form of sulindac in terms of anti-inflammatory activity is the sulphide **1b**, which is formed in the gastrointestinal tract. Sulphone **1c** accounts for a large percentage of the sulindac species in the blood (Duggan et al 1977; Elder & Paraskeva 1997). In contrast, **1b** is a cyclooxygenases (COX) inhibitor. Both agents are active in

suppressing chemically induced rat mammary carcinogenesis (Thompson et al 1995). Recent studies suggest that COX inhibition is not necessary for the anticancer activity of sulindac. The mechanisms by which these agents exert anticancer activity are not well understood. Some of the suggested mechanisms include increased production of ceramide (Chan et al 1998), induction of apoptosis (Piazza et al 1995; Shiff et al 1995), PPAR- $\alpha$  and - $\gamma$  agonistic activity (Lehmann et al 1997), suppression of PPAR- $\delta$  dependent processes (He et al 1999), alterations on mitochondrial membranes (Waddell 1998), inhibition of growth factor induced angiogenesis (Verheul et al 1999), and interaction with the ras/raf-pathway (Herrmann et al 1998).

As part of a programme for the development of new cancer chemoprevention drugs, we synthesized new derivatives and analogues of sulindac (**1a**). Since the total synthesis of **1a** is lengthy (Shuman et al 1977), we chose to modify the commercially available material, with a focus on alterations of the sulfoxide group. This molecular site apparently influences biological activity, given the differences

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in activity between sulphide **1b** and sulphone **1c**. It was thought that slight modifications at the sulphoxide position would be synthetically practical and would lead to new molecules of biological interest.

The Pummerer rearrangement is a powerful tool for the modification of sulphoxide groups, leading to  $\alpha$ -substituted sulphides that can subsequently be oxidized to form substituted sulphoxides and sulphones (De Lucchi et al 1991). We applied the Pummerer rearrangement reaction to **1a** using (diethylamino)sulphur trifluoride (DAST) (McCarthy et al 1985). This leads to the formation of  $\alpha$ -fluoro-substituted sulphoxides that can be further modified to form substituted sulphides and sulphones. It has been previously shown that substitution of hydrogen by fluorine can influence the biological activity of drugs without affecting the overall molecular size (Welch & Eswarakrishnan 1991; Filler et al 1993). In this study we describe the synthesis and biological properties of  $\alpha$ -fluoro-substituted derivatives of sulindac (**1a**).

## Materials and Methods

### Chemistry

Chloroform was dried using molecular sieves (3Å). Compound **1a** is commercially available from Sigma. Melting points are uncorrected. The following instruments were used: Perkin Elmer CHN-Analyser 240 for elemental analyses, Shimadzu 470 for infrared spectroscopy, Varian Gemini 200 (200 MHz) for  $^1\text{H}$  NMR, Varian Gemini 200 (50.29 MHz) for  $^{13}\text{C}$  NMR, Varian MAT 44S and Finnigan MAT 312 for mass spectrometry. Silica gel 60, 230–400 mesh (Merck), was used for flash chromatography.

### Methyl 5-fluoro-2-methyl-1-Z-(4-methylsulphinyl-benzylidene)-3-indenylacetate (**2**)

Compound **1a** (5 g) was dissolved in 200 mL methanol and 1 mL concentrated  $\text{H}_2\text{SO}_4$  was added. The mixture was heated to reflux for 6 h. After cooling to room temperature the mixture was evaporated and the resulting solid was recrystallized from  $\text{CH}_2\text{Cl}_2$ /hexane. The identity and purity of **2** as previously described by Jones (1975) was determined using IR, MS,  $^1\text{H}$  and  $^{13}\text{C}$  NMR.

### Methyl 5-fluoro-1-Z-(4-fluoromethylthio-benzylidene)-2-methyl-3-indenylacetate (**3a**)

To a solution of **2** (1.85 g, 5 mmol) in 25 mL of  $\text{CHCl}_3$  in an oven dried flask purged with  $\text{N}_2$ , DAST (1.32 mL, 1.61 g, 10 mmol) and  $\text{ZnBr}_2$

(34 mg, 0.15 mmol) were added. After stirring at room temperature for 24 h, 20 mL 5 % aqueous  $\text{NaHCO}_3$  was added. The organic phase was separated, dried over  $\text{Na}_2\text{SO}_4$  and evaporated. The desired product (**3a**) was obtained as yellow crystals after flash chromatography (ethyl acetate/hexane 1:3) and recrystallization from hexane/ $\text{CH}_2\text{Cl}_2$  (1.34 g, 72%): mp 102°C;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  = 2.18 (s, 3H, 2- $\text{CH}_3$ ), 3.56 (s, 2H,  $\text{CH}_2\text{COOR}$ ), 3.70 (s, 3H,  $\text{OCH}_3$ ), 5.80 (d,  $^2J_{\text{H,F}}$  = 52.6 Hz,  $\text{CH}_2\text{F}$ ), 6.52 (ddd,  $^3J$  = 9.3 Hz,  $^3J$  = 9.1 Hz,  $^4J_{\text{H,H}}$  = 2.3 Hz, 1H, 6-H), 6.88 (dd,  $^3J_{\text{H,F}}$  = 8.9 Hz,  $^4J_{\text{H,H}}$  = 2.3 Hz, 1H, 4-H), 7.13 (s, 1H,  $\text{Ar}-\text{CH}=\text{C}$ ), 7.23–7.31 (m, 1H, 7-H), 7.47–7.55 (m, 4H,  $\text{CH}_3\text{SOAr}-\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  = 10.54 (2- $\text{CH}_3$ ), 31.58 ( $\text{CH}_2\text{COO}$ ), 52.22 ( $\text{OCH}_3$ ), 87.98 (d,  $^1J_{\text{H,F}}$  = 216.6 Hz,  $\text{CH}_2\text{F}$ ), 105.91 (d,  $^2J_{\text{C,F}}$  = 23.8 Hz, C-4), 110.66 ( $^2J_{\text{C,F}}$  = 22.7 Hz, C-6), 123.68 (d,  $^3J_{\text{C,F}}$  = 9.0 Hz, C-7), 129.12 ( $\text{Ar}-\text{CH}=\text{C}$ ), 129.66 (d,  $^4J_{\text{C,F}}$  = 3.1 Hz, C-7a), 129.95 and 130.25 (C-2'/C-6', C-3'/C-5'), 131.16 (C-3), 134.58 (C-2), 135.89 (C-1'), 138.32 (C-1), 140.80 (C-4'), 146.58 (d,  $^3J_{\text{C,F}}$  = 9.1 Hz, C-3a), 163.24 (d,  $^1J_{\text{C,F}}$  = 246.1 Hz, C-F), 170.80 (CO); MS:  $m/z$  = 372,  $[\text{M}^+]$ , 233 (100); IR (KBr):  $\nu$  = 1720 (CO), 1590 (C=C)  $\text{cm}^{-1}$ ; Anal. ( $\text{C}_{21}\text{H}_{18}\text{F}_2\text{O}_2\text{S}$ ) 372.44; calc. C 67.72 H 4.88; found C 67.62 H 5.04.

### Methyl 5-fluoro-1-Z-(4-fluoromethylsulphonyl-benzylidene)-2-methyl-3-indenylacetate (**3b**)

Compound **3a** (740 mg, 2 mmol) and meta-chlorobenzoic acid (MCPBA) (990 mg, 4 mmol) were dissolved in 10 mL THF and stirred for 3 h at room temperature. After evaporation, flash chromatography (ethyl acetate/hexane 2:3) and recrystallization (hexane/ $\text{CH}_2\text{Cl}_2$ ), yellow crystals of **3b** were obtained (610 mg, 75%): mp 141°C;  $^1\text{H}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  = 2.21 (s, 3H, 2- $\text{CH}_3$ ), 3.57 (s, 2H,  $\text{CH}_2\text{COOR}$ ), 3.72 (s, 3H,  $\text{OCH}_3$ ), 5.21 (d,  $^2J_{\text{H,F}}$  = 47.1 Hz,  $\text{CH}_2\text{F}$ ), 6.58 (ddd,  $^3J_{\text{H,F}}$  = 8.9 Hz,  $^3J_{\text{H,H}}$  = 8.5 Hz,  $^4J_{\text{H,H}}$  = 2.4 Hz, 1H, 6-H), 6.88 (dd,  $^3J_{\text{H,F}}$  = 8.9 Hz,  $^4J_{\text{H,H}}$  = 2.4 Hz, 1H, 4-H), 7.28 (dd,  $^3J_{\text{H,F}}$  = 5.1 Hz,  $^3J_{\text{H,H}}$  = 8.5 Hz, 1H, 7-H), 7.13 (s, 1H,  $\text{Ar}-\text{CH}=\text{C}$ ), 7.72–8.05 (m, 4H,  $\text{CH}_3\text{SOAr}-\text{H}$ );  $^{13}\text{C}$  NMR ( $\text{CDCl}_3$ ):  $\delta$  = 10.46 (2- $\text{CH}_3$ ), 31.61 ( $\text{CH}_2\text{COO}$ ), 52.23 ( $\text{OCH}_3$ ), 92.07 (d,  $^1J_{\text{H,F}}$  = 220.4 Hz,  $\text{CH}_2\text{F}$ ), 106.42 (d,  $^2J_{\text{C,F}}$  = 24.0 Hz, C-4), 110.01 ( $^2J_{\text{C,F}}$  = 22.8 Hz, C-6), 123.71 (d,  $^3J_{\text{C,F}}$  = 9.1 Hz, C-7), 126.78 ( $\text{Ar}-\text{CH}=\text{C}$ ), 129.23 (C-7a), 129.29 and 130.35 (C-2'/C-6', C-3'/C-5'), 132.63 (C-3), 135.31 (C-2), 138.05 (C-1'), 142.77 (C-1), 143.83 (C-4'), 146.94 (d,  $^3J_{\text{C,F}}$  = 9.1 Hz, C-3a), 163.64 (d,  $^1J_{\text{C,F}}$  = 247.4 Hz, C-F), 170.49 (CO); MS (70 eV);  $m/z$  (%) = 404  $[\text{M}^+]$ , 233 (100 %); IR (KBr):  $\nu$  = 1720 (CO), 1590 (C=C), 1310s ( $\text{SO}_2$ ),

1150s (SO<sub>2</sub>); Anal. (C<sub>21</sub>H<sub>18</sub>F<sub>2</sub>O<sub>4</sub>S) 404.44; calc. C 62.36 H 4.45; found C 62.20 H 4.43.

#### General procedure for ester cleavage

The ester was dissolved in tetrahydrofuran (THF) and two equivalents of LiOH were added as a 1 M aqueous solution. After stirring for 5 h at room temperature, 20 mL diethyl ether and 20 mL H<sub>2</sub>O were added. The aqueous phase was separated and acidified with 2 M HCl (pH 2–3). The acidic solution was extracted with 3 × 30 mL ethyl acetate. The organic layer was dried over Na<sub>2</sub>SO<sub>4</sub> and evaporated.

#### 5-Fluoro-1-Z-(4-fluoromethylthio-benzylidene)-2-methyl-3-indenylacetic acid (**4a**)

Ester cleavage as described above was performed on **3a** (370 mg, 1 mmol). The crude product was purified by flash chromatography (ethyl acetate/CH<sub>3</sub>OH 50 : 1, 0.5% HOAc). Recrystallization from CH<sub>2</sub>CH<sub>3</sub>OH/H<sub>2</sub>O gave the desired product **4a** as yellow-orange powder (80 mg, 22%); mp 148°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ = 2.20 (s, 3H, 2-CH<sub>3</sub>), 3.58 (s, 2H, CH<sub>2</sub>COOR), 5.80 (d, <sup>2</sup>J<sub>H,F</sub> = 52.8 Hz, CH<sub>2</sub>F), 6.58 (ddd, <sup>3</sup>J<sub>H,F</sub> = 9.2 Hz, <sup>3</sup>J<sub>H,H</sub> = 8.7 Hz, <sup>4</sup>J<sub>H,H</sub> = 2.4 Hz, 1H, 6-H), 6.88 (dd, <sup>3</sup>J<sub>H,F</sub> = 9.1 Hz, <sup>4</sup>J<sub>H,H</sub> = 2.5 Hz, 1H, 4-H), 7.26 (s, 1H, Ar-CH=C), 7.28 (dd, <sup>3</sup>J<sub>H,F</sub> = 5.2 Hz, <sup>3</sup>J<sub>H,H</sub> = 8.7 Hz, 1H, 7-H), 7.45–7.56 (m, 4H, CH<sub>3</sub>SOAr-H); <sup>13</sup>C NMR (DMSO-d<sub>6</sub>): δ = 10.53 (2-CH<sub>3</sub>), 31.28 (CH<sub>2</sub>COO), 87.17 (d, <sup>1</sup>J<sub>H,F</sub> = 216.9 Hz, CH<sub>2</sub>F), 105.93 (d, <sup>2</sup>J<sub>C,F</sub> = 24.0 Hz, C-4), 110.83 (<sup>2</sup>J<sub>C,F</sub> = 22.7 Hz, C-6), 123.83 (d, <sup>3</sup>J<sub>C,F</sub> = 8.9 Hz, C-7), 129.32 (Ar-CH=C), 129.69 (d, <sup>4</sup>J<sub>C,F</sub> = 3.1 Hz, C-7a), 130.09 and 130.16 (C-2'/C-6', C-3'/C-5'), 130.62 (C-3), 134.70 (C-2), 135.97 (C-1'), 138.75 (C-1), 140.81 (C-4'), 146.49 (d, <sup>3</sup>J<sub>C,F</sub> = 9.0 Hz, C-3a), 163.33 (d, <sup>1</sup>J<sub>C,F</sub> = 236.4 Hz, C-F), 175.11 (CO); MS: *m/z* = 358 [M<sup>+</sup>], 233 (100); IR (KBr): ν = 1690 (CO), 1590 (C=C) cm<sup>-1</sup>; Anal. (C<sub>20</sub>H<sub>16</sub>F<sub>2</sub>O<sub>2</sub>S) 358.42; Calc. C 67.02 H 4.51; Found C 67.02 H 4.62.

#### 5-Fluoro-1-Z-(4-fluoromethylsulphonylbenzylidene)-2-methyl-3-indenylacetic acid (**4b**)

Ester cleavage as described above was performed on **3b** (202 mg, 0.5 mmol). Recrystallization from CH<sub>2</sub>CH<sub>3</sub>OH/H<sub>2</sub>O gave the desired product as a yellow powder (120 mg, 62%); mp 176°C; <sup>1</sup>H NMR (DMSO-d<sub>6</sub>): δ = 2.21 (s, 3H, 2-CH<sub>3</sub>), 3.60 (s, 2H, CH<sub>2</sub>COOR), 5.21 (d, <sup>2</sup>J<sub>H,F</sub> = 47.2 Hz, CH<sub>2</sub>F), 6.58 (ddd, <sup>3</sup>J<sub>H,F</sub> = 9.1 Hz, <sup>3</sup>J<sub>H,H</sub> = 8.4 Hz, <sup>4</sup>J<sub>H,H</sub> = 2.4 Hz, 1H, 6-H), 6.89 (dd, <sup>3</sup>J<sub>H,F</sub> =

8.8 Hz, <sup>4</sup>J<sub>H,H</sub> = 2.4 Hz, 1H, 4-H), 7.07 (dd, <sup>3</sup>J<sub>H,F</sub> = 5.1 Hz, <sup>3</sup>J<sub>H,H</sub> = 8.4 Hz, 1H, 7-H), 7.14 (s, 1H, Ar-CH=C), 7.71–7.77 (m, 4H, CH<sub>3</sub>SOAr-H); <sup>13</sup>C NMR (CH<sub>3</sub>OH-d<sub>4</sub>): δ = 10.37 (2-CH<sub>3</sub>), 32.12 (CH<sub>2</sub>COO), 93.52 (d, <sup>1</sup>J<sub>H,F</sub> = 215.6 Hz, CH<sub>2</sub>F), 107.13 (d, <sup>2</sup>J<sub>C,F</sub> = 24.4 Hz, C-4), 111.48 (<sup>2</sup>J<sub>C,F</sub> = 23.9 Hz, C-6), 124.72 (d, <sup>3</sup>J<sub>C,F</sub> = 9.2 Hz, C-7), 128.55 (Ar-CH=C), 129.82 (C-7a), 130.32 and 131.45 (C-2'/C-6', C-3'/C-5'), 136.93 (C-2), 139.31 (C-1'), 143.69 (C-1), 145.16 (C-4'), 148.65 (d, <sup>3</sup>J<sub>C,F</sub> = 8.0 Hz, C-3a), 164.84 (d, <sup>1</sup>J<sub>C,F</sub> = 245.4 Hz, C-F), 174.00 (CO); MS: *m/z* = 390 [M<sup>+</sup>], 233 (100); IR (KBr): ν = 1690 (CO), 1590 (C=C), 1310 (SO<sub>2</sub>), 1150 (SO<sub>2</sub>) cm<sup>-1</sup>; Anal. (C<sub>20</sub>H<sub>16</sub>F<sub>2</sub>O<sub>4</sub>S) 390.41; Calc. C 61.52 H 4.14; Found C 61.79 H 4.47.

#### Inhibition of cyclooxygenases

Inhibition of cyclooxygenase activity was studied using a whole blood assay as described previously (Panara et al 1995; Neupert et al 1997). COX-1 inhibition was analysed by measuring immuno-reactive thromboxan B<sub>2</sub> (irTXB<sub>2</sub>) and COX-2 (after induction with LPS) by monitoring immuno-reactive prostaglandin E<sub>2</sub> (irPGE<sub>2</sub>), both via enzyme immunoassay. For COX-2 assessment aspirin was added to suppress COX-1 contribution.

#### Growth inhibition of RCA colon cancer cells

The RCA (Ricinus communis agglutinin) human colon cancer cells was provided by M. Brattain, University of Texas, San Antonio. The 3719 human colorectal carcinoma cell line was established from a freshly surgically removed colorectal carcinoma tumour. The cell lines were plated at approximately 5000 cells/well into 96 well tissue culture plates in 150 μL of RPMI medium containing 10% foetal bovine serum, 60 μg mL<sup>-1</sup> penicillin and 0.1 mg mL<sup>-1</sup> streptomycin. After adhering overnight at 37°C, the cells were exposed to varying concentrations of the drugs for 72 h. XTT tetrazolium dye was added to the wells, and after 6 h, dye reduction was measured in a plate reader.

## Results and Discussion

Sulindac methyl ester **2** was treated with DAST and catalytic amounts of zinc bromide. The desired α-fluoro sulphide **3a** was isolated after chromatography in 72% yield. Oxidation of **3a** with meta-chloroperbenzoic acid led to a 75% yield of the α-fluoro sulphone **3b**. Hydroxide mediated cleavage of the esters **3a** and **3b** gave the respective acids **4a**

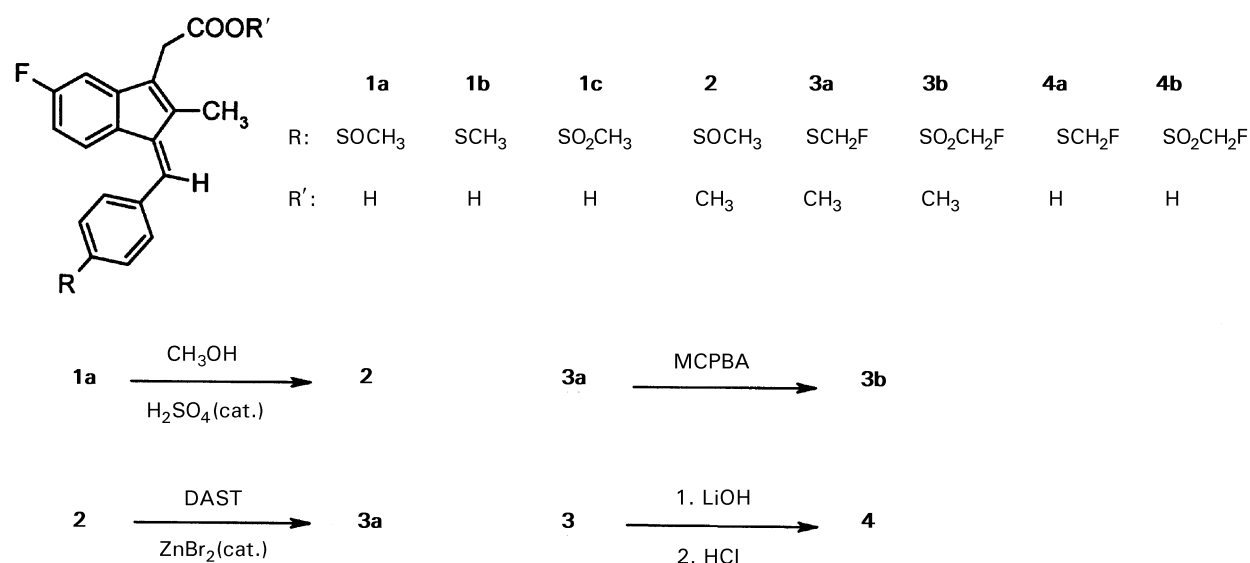


Figure 1. Structure and synthesis of the fluorinated derivatives of sulindac metabolites.

and **4b**. All compounds were obtained as crystalline solids that were analytically pure (Figure 1).

The compounds were then investigated for their ability to inhibit COX-1 and 2 using enzyme immunoassays for the quantification of respective metabolites (irTXB<sub>2</sub> for COX-1 and irPGE<sub>2</sub> for COX-2) as described previously (Panara et al 1995; Neupert et al 1997). None of the compounds displayed significant inhibition of COX-2. Sulindac sulphide (**1b**) and its fluorinated derivative **4a** were both active as inhibitors of COX-1, with **1b** (IC<sub>50</sub> 2.6  $\mu\text{M}$ ) being 2.5 times more potent than **4a** (IC<sub>50</sub> 6.4  $\mu\text{M}$ ). Neither of the sulphones **1c** or **4b** inhibited COX-1 at the highest concentrations tested (Table 1).

Cytotoxicity assays were performed using the RCA and 3719 human colorectal carcinoma cell lines. The cells were exposed to the drugs for 72 h, and cell viability was determined using the tetrazolium stain XTT, which gives a readout of mitochondrial membrane integrity. The IC<sub>50</sub> values of the sulindac derivatives **1b**, **1c**, **4a** and **4b** were approximately the same on the 3719 cell line. On

the RCA cells the fluoro sulphone **4b** (IC<sub>50</sub> of 95  $\mu\text{M}$ ) was more active than the other agents. The observed IC<sub>50</sub> values are in the range of sulindac concentrations that have previously been shown to induce apoptosis in colon cancer cells (120  $\mu\text{M}$  in HAT-29 cells; Piazza et al 1995) (Table 1).

Several studies have shown that long term chronic use of NSAIDs can be effective in reducing the incidence and death rate from colorectal carcinoma (Elder & Paraskeva 1997). Sulindac (**1a**) is one such agent that has been shown to reduce polyp formation in patients with familial adenomatous polyposis (Shiff et al 1995). While many of the chemopreventive agents actively inhibit COX-1 and/or COX-2, there is now evidence that COX inhibition is not required for the activity of **1a** and its metabolites. We (Table 1), and others (Thompson et al 1997) have shown that sulindac sulphone (**1c**), an agent that is chemopreventive in rat models of breast and colorectal carcinoma, does not inhibit either of the COX enzymes. COX-1 inhibition by sulindac sulphide (**1b**) may actually be detrimental for long term administration, since the enzyme is constitutively expressed, and inhibitors of COX-1 lead to significant gastrointestinal toxicity. While the use of selective COX-2 inhibitors might be one possibility to avoid side effects (Oshima et al 1996), we are aiming to establish compounds with increased proapoptotic properties that are independent of inhibition of prostaglandin synthesis. We have demonstrated that the fluoro derivative of sulindac sulphide **4a** retains, or even has slightly increased cytotoxic activity on colorectal cell lines, but displays 2.5-fold less inhibitory activity on COX-1 than sulindac sulphide **1b**. Additionally, the fluorinated sulphone **4b** shows an increased cyto-

Table 1. Cox-inhibition and cytotoxic activity of sulindac metabolites and derivatives.

| Assay                  | IC <sub>50</sub> ( $\mu\text{M}$ ) |                    |                    |                    |
|------------------------|------------------------------------|--------------------|--------------------|--------------------|
|                        | <b>1b</b>                          | <b>1c</b>          | <b>4a</b>          | <b>4b</b>          |
| COX-1 inhibition       | 2.6                                | > 100 <sup>a</sup> | 6.4                | > 100 <sup>a</sup> |
| COX-2 inhibition       | > 100 <sup>a</sup>                 | > 100 <sup>a</sup> | > 100 <sup>a</sup> | > 100 <sup>a</sup> |
| 3719 growth inhibition | 125                                | 125                | 125                | 125                |
| RCA growth inhibition  | 140                                | 175                | 140                | 95                 |

<sup>a</sup>Less than 20% inhibition at 100  $\mu\text{M}$ .

toxicity on RCA cells and retained activity on the 3719 cells without displaying COX-inhibitory properties. Therefore, fluorinated derivatives of sulindac metabolites constitute interesting prototypes for further structure–activity studies. These may lead to new agents for cancer chemoprevention with reduced undesired side-effects compared with the clinically approved agent from which they are derived.

#### Acknowledgements

Manfred Jung thanks B. Unterhalt for his generous support and also the Fonds der Chemischen Industrie and the DPhG-Stiftung für den Wissenschaftlichen Nachwuchs im Stifterverband für die Deutsche Wissenschaft for funding. We thank Dr. G. Winde, Department of General Surgery, University of Münster, for the generous gift of compound **1a**.

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