

**Electrochemical Synthesis of Sobrerol O-Derivatives**

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( $\pm$ )- $\alpha$ -Pinene (**1**) has been converted electrochemically to ( $\pm$ )-*cis/trans*-sobrerol [(5-(1-hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-ol] dimethyl ether (**2a/2b**) and ( $\pm$ )-*trans*-sobrerol diacetate (**3**), which can be hydrolyzed to ( $\pm$ )-*trans*-sobrerol (**5**) or its monoacetate (**4**).

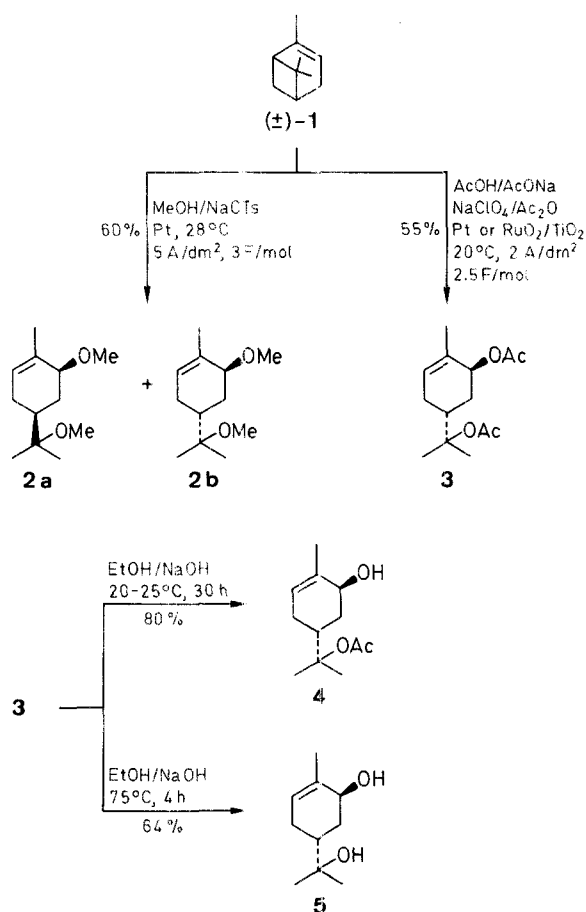
Sobrerol (5-(1-hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-ol), (**5**) is a well-known mucolytic agent.<sup>1</sup> Some of its derivatives also reveal pharmacological activities.<sup>2</sup> Recently, a mention has been made of the application of sobrerol (**5**) as an intermediate in pheromone synthesis.<sup>3</sup>

The synthesis of sobrerol is usually a multi-step process.<sup>4-7</sup> Hence, research aimed at finding more efficient methods for the synthesis of this compound have been carried out.<sup>8</sup> Recent literature describes a possible direct electrochemical transformation of ( $\pm$ )- $\alpha$ -pinene (**1**) into ( $\pm$ )-*trans*-sobrerol (**5**),<sup>9</sup> however, the yield is low (21 %).

In continuation of our investigations on the electrocatalytical transformations of terpenoids,<sup>10</sup> we wish to present a simple and

efficient method for the synthesis of sobrerol dimethyl ether (**2**) and sobrerol diacetate (**3**) from ( $\pm$ )- $\alpha$ -pinene (**1**). The diacetate **3** can be easily hydrolyzed to ( $\pm$ )-sobrerol (**5**) or its monoacetate (**4**). Previously, Shono et al.<sup>11</sup> had obtained *O*-derivatives of sobrerol in poor yield from  $\alpha$ -pinene (**1**) by an electrochemical method using carbon electrodes.

The synthesis of sobrerol dimethyl ethers (**2a, b**) was carried out by electrochemical methoxylation of  $\alpha$ -pinene (**1**) with methanol on a platinum electrode in the presence of sodium *p*-toluenesulfonate as the supporting electrolyte. The main reaction product (60%) was a mixture of dimethyl ethers of ( $\pm$ )-*cis*- and ( $\pm$ )-*trans*-sobrerol (**2a** and **2b**) in a ratio of 25:75. These isomers were separated by column chromatography and their structures were determined using <sup>1</sup>H-NMR data (Table).



It was found that if  $\alpha$ -pinene (**1**) is electrolyzed on a platinum or titanium electrode covered with titanium dioxide/ruthenium dioxide in acetic acid in the presence of sodium perchlorate and sodium acetate, 55% yield of ( $\pm$ )-*trans*-sobrerol diacetate (**3**) with ca 3–4% of the *cis*-isomer is obtained.

Sobrerol acetate (**3**) obtained electrochemically is a convenient substrate for the synthesis of sobrerol (**5**, alkaline hydrolysis at elevated temperature) or 8-acetoxydihydrocarbeole (**4**, alkaline hydrolysis at room temperature). The selective elimination of acetic acid, which takes place due to prolonged heating with acetic anhydride and sodium acetate, leads to carveyl acetate.

**Apparatus:**<sup>10</sup> Typical three-electrode electrolyzer. The anode is placed centrally between the Pt cathodes at a distance 0.3 cm.

**Electrolyte:** a) For the ether **2**: Pt anode and sodium *p*-toluenesulfonate (0.5 g) in MeOH (28 mL, 0.7 mol). b) For the ester **3**: Pt or RuO<sub>2</sub>/TiO<sub>2</sub> anode and NaOAc (0.25 mol), and NaClO<sub>4</sub> (0.1 mol) in a mixture of AcOH (4 mL) and Ac<sub>2</sub>O (0.1 mol).

#### 5-(1-Hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-ol *cis*- and *trans*-Dimethyl Ether (**2a, b**):

( $\pm$ )- $\alpha$ -Pinene (**1**; 3.41 g, 0.025 mol) is added to the cell, and the mixture is electrolyzed at 28°C until 3F/mol has been delivered (current density 5.0 A/dm<sup>2</sup>). The excess MeOH is evaporated *in vacuo*, and the product is extracted with Et<sub>2</sub>O (3 × 20 mL). The extract is washed with water (20 mL), dried (MgSO<sub>4</sub>), and the solvent is evaporated. The residue is fractionally distilled under reduced pressure to give the ether **2** as a mixture of racemic *cis/trans*-isomers **2a, b**; yield: 2.98 g (60%). The *cis/trans*-mixture is separated by column chromatography on silica gel [petroleum ether (bp 36–40°)/ether, 96:4].

#### Racemic 5-(1-Hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-ol Diacetate (**3**):

( $\pm$ )- $\alpha$ -Pinene (**1**; 115.8 g, 0.85 mol) is added to the cell, and the mixture is electrolyzed at 20°C until 2.5 F/mol has been delivered (current density 2.0 A/dm<sup>2</sup>). Excess of AcOH is evaporated (68–70°C) *in vacuo* (27 mbar), and the product is extracted with Et<sub>2</sub>O (3 × 200 mL). The extract is washed with 5% NaHCO<sub>3</sub> solution, dried (MgSO<sub>4</sub>), and the solvent is evaporated. The residue is fractionally distilled in the presence of a small amount of K<sub>2</sub>CO<sub>3</sub> to give the racemic diacetate **3**; yield: 119 g (55%); bp 116–121°C/0.5 mbar.

#### Racemic 5-(1-Hydroxy-1-methylethyl)-2-methyl-2-cyclohexen-1-ol 1-Monoacetate (**4**):

A mixture of diacetate **3** (25.44 g, 0.1 mol) and NaOH (8.4 g, 0.21 mol) in EtOH (100 mL) is allowed to stand for 30 h at room temperature. The solvent is distilled out under vacuum, and the product is extracted with Et<sub>2</sub>O (3 × 100 mL). The combined ether layer is washed with water (80 mL), (MgSO<sub>4</sub>), and the solvent is evaporated. Distillation of the residue affords pure **4**; yield: 16.99 g (80%); bp 112–117°C/0.3 mbar.

#### ( $\pm$ )-*trans*-Sobrerol (**5**):

A mixture of diacetate **3** (25.44 g, 0.1 mol) and NaOH (12 g, 0.3 mol) in EtOH (200 mL) is refluxed for 4 h. Then EtOH is evaporated *in vacuo*,

**Table.** Sobrerol *O*-Derivatives **2–4** Prepared

Product	Yield (%)	bp (°C)/mbar	$n_D^{20}$	Molecular Formula <sup>a</sup>	IR (neat) <sup>b</sup> $\nu$ (cm <sup>-1</sup> )	<sup>1</sup> H-NMR (CDCl <sub>3</sub> /TMS) <sup>c</sup> $\delta$
<b>2a</b> ( <i>cis</i> )	60 <sup>d</sup>	130–133/27 <sup>d</sup>	1.4669	C <sub>12</sub> H <sub>22</sub> O <sub>2</sub> (198.3)	1140, 1115, 1095, 1075, 910, 805	1.12 (s, 6H); 1.25–2.40 (m, 5H); 1.68 (br s, 3H); 3.17 (s, 3H); 3.33 (s, 3H); 3.77 (m, 1H, H-1); 5.43 (m, 1H)
<b>2b</b> ( <i>trans</i> )	60 <sup>d</sup>	130–133/27 <sup>d</sup>	1.4684	C <sub>12</sub> H <sub>22</sub> O <sub>2</sub> (198.3)	1140, 1095, 1075, 910, 805	1.12 (s, 6H); 1.20–2.30 (m, 5H); 1.77 (br s, 3H); 3.17 (s, 3H); 3.37 (s, 3H); 3.43 (m, 1H, H-1); 5.50 (m, 1H)
<b>3</b> ( <i>trans</i> )	55	116–121/0.5	1.4660	C <sub>14</sub> H <sub>22</sub> O <sub>4</sub> (254.3)	1735, 1240, 1135, 1030, 1015, 790	1.41 (s, 6H); 1.50–2.45 (m, 5H); 1.67 (m, 3H); 1.90 (s, 3H); 2.00 (s, 3H); 5.10 (m, 1H); 5.65 (m, 1H)
<b>4</b> ( <i>trans</i> )	80	112–117/0.3	1.4865	C <sub>12</sub> H <sub>20</sub> O <sub>3</sub> (212.3)	3400, 1735, 1600, 1255, 1135, 1050, 1035	1.40 (s, 6H); 1.55–2.50 (m, 5H); 1.73 (m, 3H); 1.90 (s, 3H); 3.87 (m, 1H, H-1); 5.45 (m, 1H)

<sup>a</sup> Satisfactory microanalyses obtained: C  $\pm$  0.23, H  $\pm$  0.03.

<sup>b</sup> Recorded on a Specord 71 IR spectrophotometer.

<sup>c</sup> Obtained on a Varian EM 360 spectrometer (60 MHz).

<sup>d</sup> These values refer to *cis/trans* mixture of **2a/2b**.

and the product is extracted with EtOAc ( $3 \times 150$  mL). The combined extract is washed with brine (50 mL), and the solvent is evaporated. The residue is recrystallized from  $\text{CHCl}_3$  (40 mL); yield: 11 g (64 %); mp  $128-130^\circ\text{C}$  (Lit.<sup>4</sup> mp  $131-131.5^\circ\text{C}$ ); IR and  $^1\text{H-NMR}$  data correspond to literature data.<sup>3</sup>

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