



## Original article

An efficient and facile synthesis of novel 1,2,3-triazolyl-*N*-acylpyrazoline hybridsPoovan Shanmugavelan, Murugan Sathishkumar, Sangaraiah Nagarajan,  
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## ABSTRACT

An efficient and facile synthesis of a library of hitherto novel 1,2,3-triazolyl-*N*-acetyl/N-propionylpyrazoline hybrids (16 examples) in excellent yields (90%–96%) has been accomplished from easily accessible 1,2,3-triazolyl chalcone precursors.

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Facile

Efficient

1,2,3-Triazole

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1,2,3-Triazolyl-*N*-acylpyrazolines

## 1. Introduction

Recently, considerable attention has been focused on pyrazolines and substituted pyrazolines due to their interesting biological activities. They possess anti-microbial [1], anti-fungal [2], anti-depressant [3], anti-convulsant [4], anti-inflammatory [5], etc. properties. Moreover, *N*-acetyl pyrazoline derivatives exhibit important pharmaceutical profiles [6–8]. Hence, the pyrazoline framework represents an interesting template for combinatorial [9] as well as medicinal chemistry [10,11]. On the other hand, 1,2,3-triazoles have received much attention due to their interesting bioactivity profile such as anti-biotic, anti-fungal [12], anti-cancer [13], anti-HIV [14], anti-microbial [15], etc. properties. Also, they serve as potential chemotherapeutic agents for various diseases [16].

In connection with the above and in continuation of our earlier work on environmentally benign, green synthesis of 1,2,3-triazolyl chalcone hybrids [17], we decided to attempt the synthesis of molecules containing both of the two pharmacologically active moieties mentioned above in a single frame. Thus, we disclose the synthesis of 1,2,3-triazolyl-*N*-acylpyrazoline hybrids from easily accessible 1,2,3-triazolyl chalcones, the details of which are presented *vide infra*.

Chalcones, belonging to the flavonoids family, are convenient synthons for the synthesis of five [18], six [19] and seven membered [20] heterocyclic compounds. With regard to pyrazoline derivatives, several methods have been employed for their synthesis, including the condensation of chalcones with hydrazine, hydrazine derivatives [21–24] and thiosemicarbazide [25] under acidic [21,22] or basic [25] conditions, and the cycloaddition of nitrilimines, generated *in situ* from the corresponding hydrazoneoyl halides by the action of a suitable base, to carbon-carbon double bonds of a dipolarophile [26–29]. Hence, considerable interest has been focused on the synthesis of pyrazolines from chalcones.

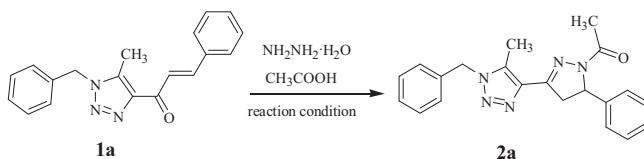
## 2. Experimental

Typical procedure: To a solution of (*E*)-1-(1-benzyl-5-methyl-1*H*-1,2,3-triazol-4-yl)-3-(4-methylphenyl)prop-2-en-1-one (**1e**, 1.0 equiv.) and hydrazine hydrate (2.0 equiv.) in acetic acid (5 mL) was refluxed for 3 h. Then, the reaction mixture was poured onto ice-water to afford the 1,2,3-triazolyl-*N*-acetylpyrazoline hybrid (**2e**) in 96% yield as a white solid, which was filtered and recrystallized from ethanol.

1-(3-(1-Benzyl-5-methyl-1*H*-1,2,3-triazol-4-yl)-5-(4-methyl-phenyl)-4,5-dihydro-1*H*-pyrazol-1-yl)ethanone (**2e**): Mp: 150–151 °C; <sup>1</sup>H NMR (300 MHz, CDCl<sub>3</sub>): δ 7.11–7.37 (m, 9H, ArH), 5.54 (s, 2H, NCH<sub>2</sub>), 5.50 (dd, 1H, J = 4.8 & 12.0 Hz, CH), 3.87 (dd, 1H, J = 12.0 & 18.6 Hz, CH<sub>2</sub>), 3.44 (dd, 1H, J = 4.8 & 18.6 Hz, CH<sub>2</sub>), 2.51 (s, 3H, COCH<sub>3</sub>), 2.32 (s, 3H, CH<sub>3</sub>), 2.29 (s, 3H, ArCH<sub>3</sub>); <sup>13</sup>C NMR

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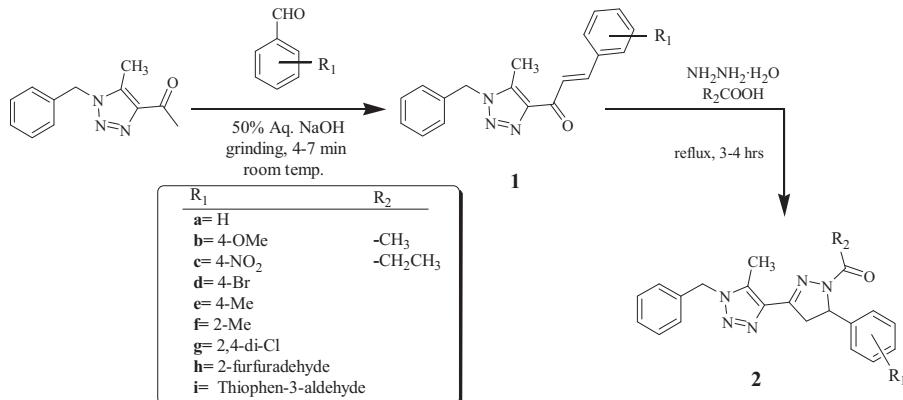


**Scheme 1.** Optimization for the synthesis of 1,2,3-triazole-N-acetylpyrazoline hybrid (**3a**).

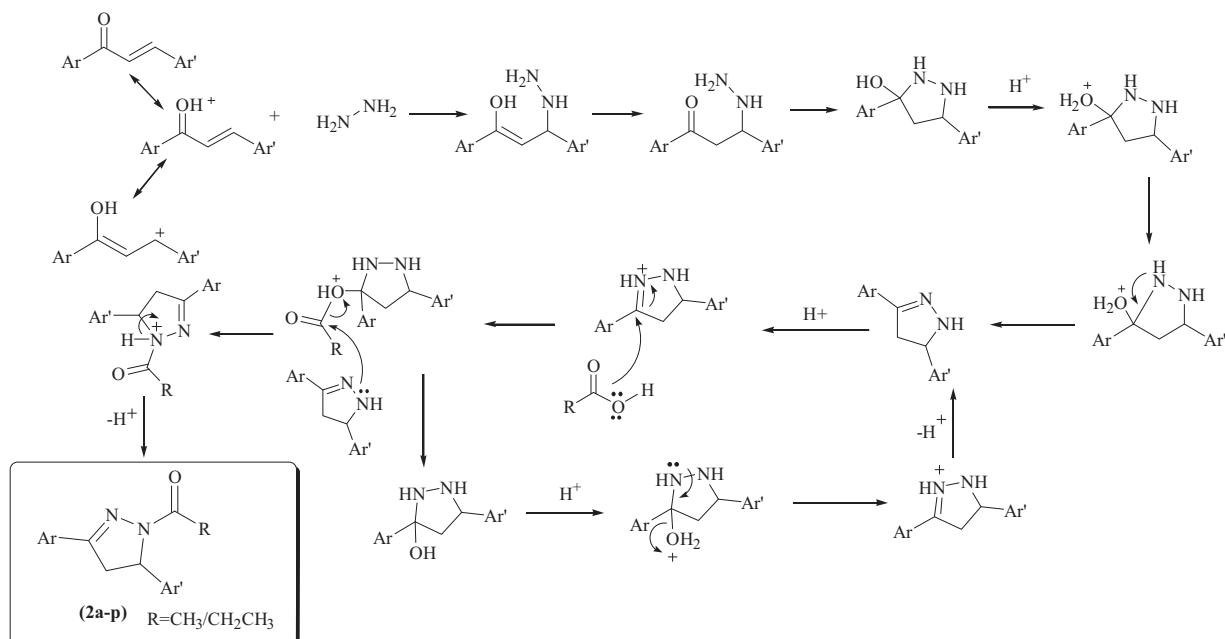
(75 MHz, CDCl<sub>3</sub>): δ 168.46, 149.47, 138.68, 138.24, 137.19, 134.25, 132.80, 129.41, 129.08, 128.53, 127.25, 125.65, 58.69, 51.95, 43.25, 21.79, 21.00, 9.41; Mass (ES/MS): *m/z* 396.58 [M+Na]<sup>+</sup>; Anal. Calcd. for C<sub>22</sub>H<sub>23</sub>N<sub>5</sub>O: C 70.76, H 6.21, N 18.75. Found: C 70.70, H 6.22, N 18.78.

### 3. Results and discussion

Perusal of literature indicates a lack of reports on the synthesis of *N*-acylpyrazolines linked with a 1,2,3-triazole core. In this regard, we herein report synthesis of novel 1,2,3-triazolyl-*N*-acylpyrazoline hybrids in a facile and efficient manner starting from chalcones [17].



**Scheme 2.** Synthesis of 1,2,3-triazolyl-*N*-acetyl/*N*-propionylpyrazolines (**2**).



**Scheme 3.** A feasible mechanism for *N*-acylatedpyrazolines.

**Table 1**  
Optimization for the synthesis of 1,2,3-triazole-*N*-acetylpyrazoline hybrid (**3a**).

Chalcone	Reaction condition				
	Conventional heating	MW-irradiation	Time (min)	Temp. (°C)	Yield (%)
<b>1a</b>	1	55	10	100	55
<b>1a</b>	2	75	20	100	60
<b>1a</b>	2.5	88	30	100	60
<b>1a</b>	3	95	20	130	60
<b>1a</b>	3.5	95			

For the purpose of optimization, synthesis of *N*-acetylpyrazoline (**3a**) was attempted (Scheme 1) under different conditions viz. (i) reaction through conventional heating (ii) reaction under microwave irradiation. The details are presented below.

A mixture of 1,2,3-triazolyl chalcone (**1a**, 1.0 equiv.) and hydrazine hydrate (2.0 equiv.) in acetic acid was conventionally heated to reflux. It was noted that the yield of *N*-acetylpyrazoline (**2a**) was increased and reached 95% in three hours. However, by changing the source of heating to microwave irradiation, a comparable yield of the product could not be obtained

**Table 2**Synthesis of 1,2,3-triazolyl-N-acetyl/N-propionylpyrazoline hybrids (**2a–p**).

Entry	Chalcones ( <b>1a–i</b> )	R <sub>2</sub>	Time <sup>a</sup> (h)	Product	Yield (%)
1	<b>1a</b>	CH <sub>3</sub>	3	<b>2a</b>	95
2	<b>1b</b>	CH <sub>3</sub>	3	<b>2b</b>	96
3	<b>1c</b>	CH <sub>3</sub>	3.5	<b>2c</b>	94
4	<b>1d</b>	CH <sub>3</sub>	3	<b>2d</b>	95
5	<b>1e</b>	CH <sub>3</sub>	3	<b>2e</b>	96
6	<b>1f</b>	CH <sub>3</sub>	4	<b>2f</b>	93
7	<b>1g</b>	CH <sub>3</sub>	4	<b>2g</b>	93
8	<b>1h</b>	CH <sub>3</sub>	3.5	<b>2h</b>	94
9	<b>1i</b>	CH <sub>3</sub>	3.5	<b>2i</b>	95
10	<b>1a</b>	CH <sub>3</sub> CH <sub>2</sub>	4	<b>2j</b>	92
11	<b>1c</b>	CH <sub>3</sub> CH <sub>2</sub>	4	<b>2k</b>	92
12	<b>1d</b>	CH <sub>3</sub> CH <sub>2</sub>	3.5	<b>2l</b>	92
13	<b>1e</b>	CH <sub>3</sub> CH <sub>2</sub>	4	<b>2m</b>	93
14	<b>1g</b>	CH <sub>3</sub> CH <sub>2</sub>	4	<b>2n</b>	90
15	<b>1h</b>	CH <sub>3</sub> CH <sub>2</sub>	3	<b>2o</b>	91
16	<b>1i</b>	CH <sub>3</sub> CH <sub>2</sub>	3	<b>2p</b>	90

<sup>a</sup> Reaction times are reported based on TLC monitoring.

(Table 1). Thus, conventional heating was found to be the best methodology for the target compounds.

Following optimization, the broad scope of the present protocol was established via the synthesis of a library of novel 1,2,3-triazolyl-N-acetyl/N-propionylpyrazoline hybrids (**2a–p**, Scheme 2) in excellent yields (90%–96%, Table 2), and it was noted that the reaction involving acids with long alkyl chains such as butyric acid, pentanoic acid, etc. was not satisfactory.

All the synthesized compounds were completely characterized by IR, NMR (1D & 2D) and mass spectral techniques. The feasible mechanism is presented vide Scheme 3.

#### 4. Conclusion

In conclusion, we have achieved an efficient and facile synthesis of a library of hitherto novel 1,2,3-triazolyl-N-acetyl/N-propionylpyrazoline hybrids in excellent yields via easily accessible chalcones.

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