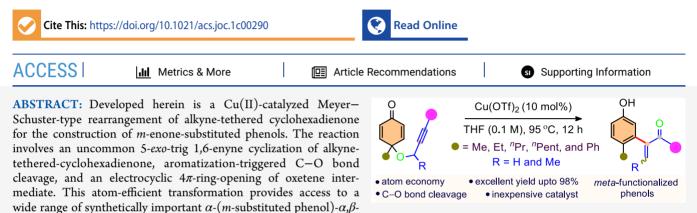
pubs.acs.org/joc

Lewis Acid-Driven Meyer–Schuster-Type Rearrangement of Yne-Dienone

Rajendra K. Mallick, Srinivas Vangara, Nagarjuna Kommu, Tirumaleswararao Guntreddi, and Akhila K. Sahoo*



unsaturated ketones, featuring a broad scope with labile functional group tolerance. The gram-scale demonstration makes this transformation synthetically viable. The synthetic application of α,β -unsaturated ketones is also showcased.

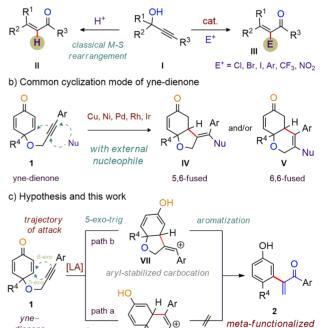
INTRODUCTION

Propargyl alcohols have been largely used for the development of novel synthetic transformations and the fabrication of complex molecular scaffolds.^{1,2} In this regard, the atom-efficient conversion of propargyl alcohol I to α_{β} -unsaturated carbonyl compound II is a trustworthy testimony of the classical Meyer-Schuster rearrangement (MSR; Scheme 1a).³ This transformation primarily involves an acid-catalyzed 1,3-hydroxy shift and tautomerization sequence of propargyl alcohol (Scheme 1a). Over the years, new variants of electrophileintercepted MSRs have been developed, and their synthetic potential has been significantly widened.^{4,5} Thus, transitionmetal (TM)/Lewis acid (LA)-catalyzed halo/aryl/CF₃/NO₂intercepted MSRs have led to diverse arrays of α -functionalized enones (Scheme 1a).⁵ Among all, the aryl-electrophile (available from the respective boronic acids, diazonium salts, hypervalent iodine complexes, etc.)-intercepted MSRs of propargyl alcohols are noteworthy, leading to highly functionalized α -arylated enones (Scheme 1a).^{5d,e} However, such transformations are possible in the presence of external electrophiles, expensive metal catalysts, and ligands.

Along this line, yne-dienones 1 are unequivocally important as these moieties have been largely used for the development of novel synthetic methods and the construction of unusual fused heteroaryls.⁶ In this line, the transition-metal (TM)-catalyzed arylative/borylative/silylative/reductive and acetate-triggered cyclizations of yne-dienones are most significant (Scheme 1b).^{7,8} These reactions primarily happen via nucleophile/radical-assisted 5-*exo*-trig/6-*exo*-trig cyclizations of yne-dienones to make [5,6]/[6,6]-fused heterocycles **IV/V**, respectively (Scheme 1b).

Scheme 1. Previous Work and Current Hypothesis





Received: February 5, 2021

6-exo-tria

dienone



phenols

Further to the recent demonstration on the thioarylative radical cyclization of yne-dienones,9 a Lewis acid-driven cyclization-aromatization-rearrangement cascade of 1 is envisaged (Scheme 1c). Inspired by the MSR of propargyl alcohols,^{3a} the O-bridged alkyne-tethered-cyclohexadienones 1 in the presence of Lewis acid could undergo an intramolecular 6exo-trig/5-exo-trig Michael attack by the tethered alkyne to respectively generate alkyl enabled vinyl carbocation VI and aryl stabilized vinyl carbocation VII (Scheme 1c). Next, aromatization of the stable intermediate VII with the concomitant cleavage of the C–O bond could form a strained oxetene species (see Scheme 6). Finally, rearrangement of oxetene could provide access to peripheral decorated *m*-enone-substituted phenols 2 [Baylis-Hillman-Basavaiah (BHB) type adduct],¹⁰ which are otherwise difficult to prepare in a conventional synthetic method (Scheme 1c). Despite the broad synthetic transformations uncovered in MSRs, a detailed investigation on the current findings α -arylative MSR is yet to be explored in detail. During the preparation of this manuscript, a silver(I)-catalyzed and TfOH-mediated cyclization-aromatization cascade of alkynetethered-cyclohexadienone has recently been appeared.¹

RESULTS AND DISCUSSION

To make the envisioned plan workable, 4-methyl-4-((3phenylprop-2-yn-1-yl)oxy)cyclohexa-2,5-dienone (1a) was exposed to various Lewis acids (Table 1). At first, subjecting 1a with Sc(OTf)₃ (10 mol %) in tetrahydrofuran (THF) at 85 °C for 12 h led to $\alpha_{,\beta}$ -unsaturated enone rearranged product 2a in 63% yield (entry 1). Screening other Lewis acids, Yb(OTf)₃, Fe(OTf)₃, and In(OTf)₃, provided 2a in 73%, 54%, and 66%

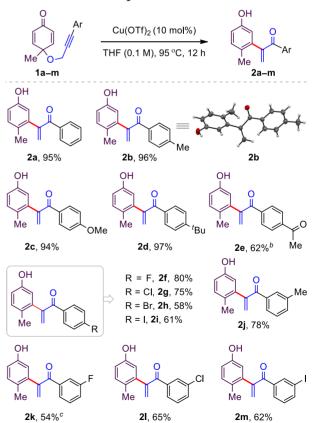
Table 1. Optimization of Reaction Conditions ^{a,b}					
		Ph Me 1a	catalyst (10 mol%) solvent, temp.	OH Me 2a	Ph
	entry	catalyst	solvent	temp (°C)	yield
	1	Sc(OTf) ₃	THF	85	63
	2	Yb(OTf) ₃	THF	85	73
	3	Fe(OTf) ₃	THF	85	54
	4	In(OTf) ₃	THF	85	66
	5	$Cu(OTf)_2$	THF	85	75
	6 ^c		THF	85	no reaction
	7 ^c	$Cu(OTf)_2$	ClCH ₂ CH ₂ Cl	85	trace
	8	$Cu(OTf)_2$	1,4-dioxane	85	42
	9 ^c	$Cu(OTf)_2$	toluene	85	no reaction
	10 ^c	$Cu(OTf)_2$	CHCl ₃	85	trace
	11 ^d	$Cu(OTf)_2$	THF	85	30
	12 ^c	$Cu(OTf)_2$	THF	rt	no reaction
	13 ^c	$Cu(OTf)_2$	THF	60	trace
	14	$Cu(OTf)_2$	THF	95	98
	15 ^e	CuOTf-toluene	THF	95	64
	16 ^c	Cu(MeCN) ₄ PF ₆	THF	95	trace
	17 ^c	CuCl	THF	95	trace
	18 ^{e,f}	TfOH	THF	95	72
	19 ^{e,f}	Tf ₂ NH	THF	95	74

^aReactions were carried out using 1a (0.06 mmol), Lewis acid (10 mol %), and solvent (0.1 M) for 12 h. ^bCrude NMR yield. ^cObserved by TLC. ^dReaction stopped in 5 h. ^eReaction was carried out using 1a (0.3 mmol) and isolated yield. ^JBrønsted acid (20 mol %) was used.

vields, respectively (entries 2, 3, and 4). To our delight, the compound 2a was obtained in 75% yield when the reaction was conducted in $Cu(OTf)_2$ (10 mol %) at 85 °C for 12 h (entry 5). In the absence of catalyst, 2a was not even formed (entry 6). The product yield was not enhanced when the reaction was carried out in ClCH₂CH₂Cl, 1,4-dioxane, toluene, and/or CHCl₂ solvents (entries 7-10). However, the reaction under the identical condition of entry 5 in 5 h affected product formation affording 30% **2a** (entry 11). The reaction at room temperature did not proceed (entry 12), whereas, at 60 °C, a trace of 2a was detected (entry 13). To our delight, the reaction was clean at 95 °C providing access to 2a in 98% yield (entry 14). Thus, rearrangement of 1a was very smooth when conducted in the presence of Cu(OTf)₂ (10 mol %) in THF at 95 °C for 12 h (entry 14). We next scrutinized various copper salts and Brønsted acid catalysts. The reaction in the presence of 10 mol % CuOTf-toluene delivered the desired product in 64% yield (entry 15). Disappointingly, Cu(MeCN)₄PF₆ and CuCl catalysts were not suitable (entries 16 and 17). However, the reaction in the presence of Brønsted acids (TfOH and Tf₂NH) successfully delivered the product 2a in 72% and 74% yields, respectively (entries 18 and 19).

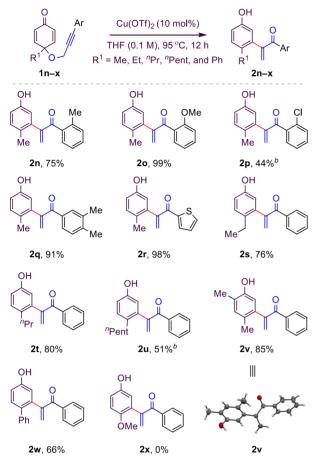
The scope and limitation of the $Cu(OTf)_2$ -catalyzed rearrangement of alkyne-tethered-cyclohexadienones 1 were explored, and the results are shown in Schemes 2 and 3. The rearrangement of 1a (0.3 mmol) was smooth to afford 95% of 2a. The electron-rich arenes (having substituents p-Me/p- $OMe/p^{-t}Bu$) on alkyne terminus of yne-dienones 1b-d





^aReactions were carried out using 1 (0.3 mmol), Cu(OTf)₂ (10 mol %), in THF (0.1 M; 3 mL) at 95 °C for 12 h. ^bStirred for 36 h. ^cUnreacted starting precursor recovered.

Scheme 3. Substrate Scope II^a



^{*a*}Reactions were carried out using 1 (0.3 mmol), Cu(OTf)₂ (10 mol %), in THF (0.1 M; 3 mL) at 95 °C for 12 h. ^{*b*}Unreacted starting precursor recovered.

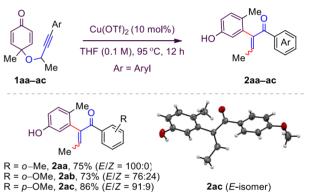
successfully underwent the cationic rearrangement to deliver the respective products 2b-d in excellent yields (94-97%; Scheme 2). Although the reaction took a longer time, 36 h for completion, the desired product 2e (62%) was isolated in a moderate yield from the electron-deficient p-COMe areneenabled 1e. Presumably, the substituent -I effect affects the stability of vinyl carbocation intermediates and the reaction productivity. Likewise, Cu(OTf)2-catalyzed rearrangement of yne-dienones 1f-i, having halo-substitution -F/Cl/Br/I on the *p*-position of arenes in alkyne terminus, provided 2f-i in 58-80% yields (Scheme 2); the labile halo groups survived under the catalytic conditions, and hence, further functionalization of these groups is possible. The yne-dienones 1j-m exhibiting *m*-Me or the transformable halo (m-F/m-Cl/m-I) at the arene moiety also reacted well to furnish the corresponding metasubstituted phenols 2j-m in 54-78% yields (Scheme 2).

To further examine the effect of steric bulkiness in this cyclization-aromatization-rearrangement cascade, the reaction of alkyne-tethered-cyclohexadienones with *ortho*-substituted arene [*o*-Me (1n) and *o*-OMe (1o)] on the alkyne terminus was carried out under the optimized conditions that lead to 2n (75%) and 2o (99%), respectively (Scheme 3), whereas the transformable *o*-Cl arene-substituted yne-dienone 1p delivered 2p, albeit in a moderate yield. An excellent product yield of 2q was isolated from 1q (the *m*,*p*-dimethyl arene-substituted yne-dienone). The reaction was compatible with the

heteroaryl-enabled yne-dienone 1r, leading to 2-thienyl-bearing enones 2r in a quantitative yield (Scheme 3). The alkynecyclohexadienone with various alkyl substituents at the angular position [for example, ethyl (1s), *n*-propyl (1t), and *n*-pentyl (1u)] smoothly underwent cyclization-aromatization-rearrangement cascade to furnish the desired BHB-type adducts 2s**u** in good yields (Scheme 3). A tetrasubstituted phenol derivative, 2v, with a m-enone motif was successfully made in 85% yield. Likewise, yne-dienone 1w with the bulky-phenyl group on the angular position successfully participated in the rearrangement to provide the desired product 2w in 66% yield. To our disappointment, compound **1x** (having a methoxy group in the angular position of yne-dienone) decomposed during the course of the reaction. Overall, the present demonstration is comparatively productive and effective over the complementary method.¹¹ In addition, the electron-rich alkyne-containing ynedienone delivers the best result; this information supports the involvement of a cationic intermediate in this transformation.

To further study the scope and limitations of α -aryl- α , β unsaturated ketones **2**, the reaction was performed with ynedienone **1** having substituents at the α -position (Scheme 4). In

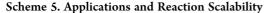


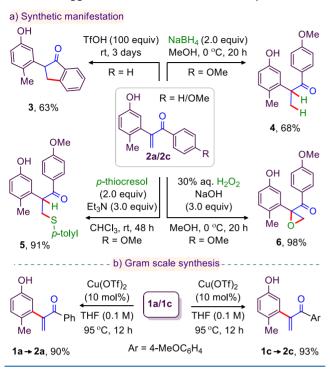


^{*a*}Reactions were carried out using 1aa–ac (0.3 mmol), Cu(OTf)₂ (10 mol %), in THF (0.1 M; 3 mL) at 95 °C for 12 h. ^{*b*}The diasteriomeric ratio was calculated by ¹H NMR analysis.

this context, the reaction of a sterically encumbered *ortho*-Mesubstituted yne-dienone **1aa** under the optimized conditions delivered the desired product **2aa** in 75% yield as a single *E*isomer (Scheme 4). Similarly, the yne-dienone **1ab** having *ortho*-OMe on the arene terminus provided **2ab** in 73% yield as an inseparable mixture of diastereomers (E/Z 76:24). The respective major adduct **2ac** (*E*-isomer) was obtained from **1ac** (Scheme 4). The structure of **2ac** was further confirmed by single-crystal X-ray analysis (Scheme 4).

To expand the synthetic versatility of α -aryl- α , β -unsaturated ketones, common organic transformations were performed on 2a/2c (Scheme 5a). For instance, the BrØnsted acid (TfOH)mediated intramolecular electrophilic cyclization of enone 2a delivered 2-arylated-2,3-dihydro-indenone 3 in 63% yield (Scheme 5a).¹² Next, NaBH₄-mediated reduction of olefin moiety of 2c in MeOH led to aryl-alkyl ketone 4 in 68% yield (Scheme 5a).¹³ Moreover, thioether 5 (91%) was constructed from the Michael addition of *p*-tolylthiol to 2c (Scheme 5a).¹⁴ Finally, peroxide-mediated epoxidation of electron-deficient olefin in 2c furnished the desired product 6 in 98% yield (Scheme 5a).¹⁵ The catalytic system developed for the rearrangement of 1 to 2 was robust; the gram-scale conversion





of 1a (1.0 g) \rightarrow 2a (0.9 g, 90%) and 1c (1.0 g) \rightarrow 2c (0.93 g, 93%) under standard conditions justified the synthetic potential of the current strategy (Scheme 5b).

To understand the major Van der Waals interactions of α -aryl- α,β -unsaturated ketones, we have carried out Hirshfeld surface analysis¹⁶ and 2D-fingerprint plots of compounds 2b and 2v, the images (a, e) in Figure 1; the deep red circles indicate strong intermolecular interactions (O···H/H···O), and other spots are due to H…H and C…H/H…C interactions. The 2D-fingerprint plots (b, f) in Figure 1 denote a pair of spikes at the bottom left, which indicates strong O…H and H…O interactions. The H…H and C…H/H…C interactions are scattered in the middle of the 2D-fingerprint plots. The 2D-fingerprint images (c, g) show H···· H (2b, 53.8%; 2v, 56.2%) contacts and contribute to the total Hirshfeld surface. The presence of an additional methyl group in compound 2v increases H···H interactions by 2.4% than compound 2b. The individual atomic contact distributions of 2b and 2v are shown in images d and h in Figure 1. Hence, these studies disclose the stabilization of crystal lattices 2b and 2v by three main interactions: H...H, O...H/H...O, and C...H/H...C.

Based on the experimental facts and the literature reports, the electronic factor controls the reaction dynamics. The plausible mechanism is depicted in Scheme 6. The transformation begins with the activation of enone by $Cu(OTf)_2$ to form the activated complex VIII.¹⁷ Next, intramolecular 5-*exo*-trig Michael attack of alkyne to the activated-olefin of VIII gives aryl-stabilized vinyl-carbocation intermediate IX. The aromatization triggered C–O bond cleavage, and intramolecular charged recombination generates an unstable oxa-cyclobutene intermediate X. Finally, electrocyclic 4π -ring-opening of X produces the desired product 2 and regenerates the active catalyst for the next cycle.

CONCLUSION

In summary, a Cu-catalyzed α -arylative MSR of yne-dienones for the synthesis of *m*-enone-substituted phenol derivatives has been developed. The labile halo groups, heteroaryl systems, and

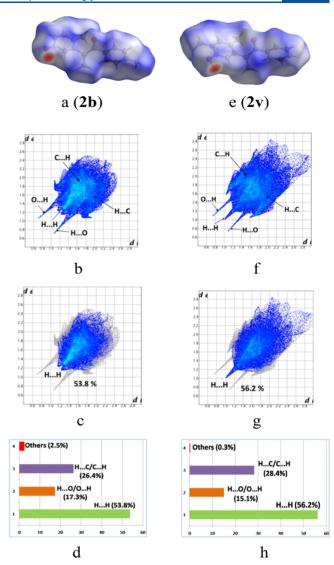
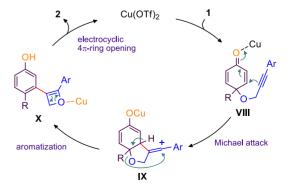


Figure 1. Hirshfeld surface plots for compounds 2b and 2v in the crystal structures. Images a and e are the Hirshfeld surface graphs with proximity of close contacts around 2b and 2v molecules. Images b and f are 2D-fingerprint plots in crystal packing found in 2b and 2v. Images c and g are the H···H atomic contact percentage contribution to the Hirshfeld surface for 2b and 2v molecules. Images d and h are the percentage contributions of the individual atomic contacts to the Hirshfeld surface for compounds 2b and 2v.

Scheme 6. Plausible Mechanism



the aliphatic chains did not affect the reaction outcome and were well tolerated. The overall process is atom-efficient, providing

The Journal of Organic Chemistry

access to a wide range of novel peripheral decorated *m*-phenol derivatives in excellent yields. The transformation is also scalable. Common synthetic transformations of enones make this strategy synthetically useful.

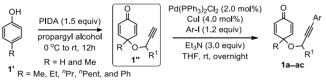
EXPERIMENTAL SECTION

General Information. All reactions were performed in oven-dried reaction vials under a nitrogen atmosphere. Commercial grade solvents were distilled prior to use. Column chromatography was performed using silica gel (100-200 Mesh) with a hexane and ethyl acetate mixture. Thin-layer chromatography (TLC) was performed on silica gel GF 254 plates. Visualization of spots on the TLC plate was accomplished with UV light (254 nm) and staining over an I₂ chamber. Proton and carbon nuclear magnetic resonance spectra (¹H NMR, ¹³C NMR, and ¹⁹F NMR) were recorded on a 400 MHz (¹H NMR, 400 MHz; ¹³C NMR, 101 MHz; ¹⁹F NMR, 376 MHz) spectrometer, 500 MHz (¹H NMR, 500 MHz; ¹³C NMR, 126 MHz; ¹⁹F NMR, 471 MHz) spectrometer, and a 600 MHz (¹H NMR, 600 MHz; ¹³C NMR, 151 MHz) spectrometer, having solvent resonance as an internal standard (¹H NMR, CHCl₃ at 7.26 ppm; ¹³C NMR, CDCl₃ at 77.0 ppm). In a few cases, tetramethylsilane (TMS) at 0.00 ppm was used as an reference standard. Data for ¹H NMR are reported as follows: chemical shift (ppm), multiplicity (s = singlet; br s = broad singlet; d = doublet; bd = broad doublet, dd= doublet of doublet, dt = doublet of triplet, tt = triplet of triplet, t = triplet; bt = broad triplet; q = quartet; pent = pentet, m = multiplet), coupling constants J, in (Hz), and integration.

¹³C NMR and ¹⁹F NMR were reported in terms of chemical shift (ppm). IR spectra were recorded on an FT/IR spectrometer and are reported in cm⁻¹. High-resolution mass spectra (HRMS) were obtained in ESI mode by using a TOF analyzer. Melting points were determined by electrothermal heating and are uncorrected. X-ray data was collected at 298 and 296 K on a Bruker D8 Quest CCD and Bruker D8 VENTURE Photon III detector diffractometer using Mo K α radiation (0.71073 Å).

Materials. Unless otherwise noted, all reagents and intermediates were obtained commercially and used without purification. Toluene, 1,2-dichloroethane, 1,4-dioxane, chloroform, ethyl acetate, and hexane were distilled over CaH₂. THF was freshly distilled over sodium/ benzophenone ketyl under dry nitrogen. Cu(OTf)₂, Sc(OTf)₃, Fe(OTf)₃, In(OTf)₃, Yb(OTf)₃, triflic acid (TfOH), sodium borohydride (NaBH₄), *p*-thiocresol, triethylamine (Et₃N), 30% aq H₂O₂, sodium hydroxide (NaOH) pellets, methanol (CH₃OH), *p*-cresol, propargyl alcohol, iodosobenzene diacetate, CuI, and PdCl₂(PPh₃)₂ were commercially available and used as received. Aryl iodides were purchased and used as such. Following the known procedure, ^{8c,9,18,19} compounds **1a**–**ac** were synthesized. Analytical and spectral data of all those known compounds exactly match the reported values.

General Method for the Preparation of Starting Materials.



General Procedure for the Synthesis of 1'' (GP-1). To a stirred solution of 4-substituted phenol 1' (10 mmol) in 10 mL of propargyl alcohol was added phenyliodo(III)diacetate (15 mmol) in several portions at 0 °C. The resulting reaction mixture was warmed to room temperature and stirred overnight. Then the reaction was quenched with saturated aqueous sodium bicarbonate (30 mL) and extracted with ethyl acetate (3 × 20 mL). The combined organic layer was washed with brine (20 mL), dried over Na₂SO₄, filtered, and concentrated in vacuo. The crude reaction mixture was purified by silica gel (100–200 mesh) column chromatography (EtOAc/hexane) to give the respective terminal cyclohexadienones 1".

General Procedure for the Synthesis of Alkynyl Cyclohexadienone 1a-ac (GP-2):.^{8C,9,18,19} To a solution of 1" (1.0 mmol), Pd(PPh₃)₂Cl₂ (2.0 mol %, 14 mg), and CuI (4.0 mol %, 7.6 mg) in THF (2.0 mL) was pubs.acs.org/joc

added aryliodide (1.2 mmol) under a nitrogen atmosphere. Finally, triethylamine (3.0 mmol, 0.42 mL) was added to the above mixture, and the resulting mixture was stirred at room temperature overnight. After completion, the reaction mixture was filtered, concentrated in vacuo, and purified by silica gel (100–200 mesh) column chromatography to give the desired alkynyl cyclohexadienones 1a-ac. The analytical and spectral data for compounds 1a-c, 1e-h, 1j-k, 1m, 1n, 1p, 1r-t, 1v, and 1w exactly match the reported values.

4-((3-(4-(tert-Butyl)phenyl)prop-2-yn-1-yl)oxy)-4-methylcyclohexa-2,5-dienone (1d). Compound 1d (153 mg) was obtained in 52% yield: brown solid; mp = 93–95 °C; $R_f = 0.40$ (9:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.36–7.30 (m, 4H), 6.89 (dt, J = 10, 3.0 Hz, 2H), 6.32 (dt, J = 10, 3.5 Hz, 2H), 4.22 (s, 2H), 1.49 (s, 3H), 1.29 (s, 9H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 185.0, 151.8, 151.0, 131.4, 130.3, 125.2, 119.3, 87.0, 85.1, 73.2, 54.6, 34.7, 31.1, 26.4; IR (neat) ν_{max} 2964, 1661, 1604, 1512, 1452, 1369, 1274, 1194 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₂₃O₂+295.1693, found 295.1703.

4-((3-(4-lodophenyl)prop-2-yn-1-yl)oxy)-4-methylcyclohexa-2,5dienone (1i). Compound 1i (149 mg) was obtained in 41% yield: colorless solid; mp = 117–119 °C; $R_f = 0.31$ (9:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCI₃) δ 7.63 (dt, J = 8.5, 2.0 Hz, 2H), 7.12 (dt, J = 8.5, 2.0 Hz, 2H), 6.86 (dt, J = 10.5, 3.0 Hz, 2H), 6.32 (dt, J = 10, 3.5 Hz, 2H), 4.19 (s, 2H), 1.49 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCI₃) δ 184.9, 150.7, 137.5, 133.1, 130.4, 121.8, 94.6, 87.1, 85.8, 73.2, 54.4, 26.3; IR (neat) ν_{max} 2986, 2230, 1657, 1627, 1477, 1452, 1452, 1385, 1299, 1264, 1188 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄HO₂⁺ 365.0033, found 365.0038.

4-((3-(3-Chlorophenyl)prop-2-yn-1-yl)oxy)-4-methylcyclohexa-2,5-dienone (11). Compound 11 (196 mg) was obtained in 72% yield: brown solid; mp = 76–78 °C; R_f = 0.35 (9:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCI₃) δ 7.38 (s, 1H), 7.30–7.27 (m, 2H), 7.23 (d, *J* = 8.0 Hz, 1H), 6.86 (dt, *J* = 10.5, 3.0 Hz, 2H), 6.32 (dt, *J* = 10, 3.5 Hz, 2H), 4.20 (s, 2H), 1.49 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCI₃) δ 184.9, 150.7, 134.1, 131.5, 130.4, 129.8, 129.5, 128.9, 124.0, 87.0, 85.3, 73.2, 54.3, 26.3; IR (neat) ν_{max} 2983, 1661, 1621, 1595, 1484, 1398, 1372, 1299, 1198 cm⁻¹; HRMS (ESI) *m*/*z* [M + H]⁺ calcd for C₁₆H₁₄ClO₂⁺ 273.0677, found 273.0683.

4-((3-(2-Methoxyphenyl)prop-2-yn-1-yl)oxy)-4-methylcyclohexa-2,5-dienone (10). Compound 1o (169 mg) was obtained in 63% yield: colorless solid; mp = 124–126 °C; $R_f = 0.20$ (9:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.37 (dd, J = 7.5, 2.0 Hz, 1H), 7.31–7.27 (m, 1H), 6.94–6.88 (m, 3H), 6.86 (d, J = 8.5 Hz, 1H), 6.32 (dt, J = 10, 3.0 Hz, 2H), 4.28 (s, 2H), 3.86 (s, 3H), 1.50 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 185.1, 160.2, 151.2, 133.7, 130.3, 130.0, 120.4, 111.6, 110.6, 90.0, 83.3, 73.3, 55.7, 54.8, 26.4; IR (neat) ν_{max} 2919, 2227, 1668, 1624, 1595, 1499, 1452, 1267, 1245, 1185 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₃⁺ 269.1172, found 269.1179.

4-((3-(3,4-Dimethylphenyl)prop-2-yn-1-yl)oxy)-4-methylcyclohexa-2,5-dienone (1q). Compound 1q (120 mg) was obtained in 45% yield: gray solid; mp = 142–144 °C; R_f = 0.38 (9:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.19 (s, 1H), 7.15 (d, *J* = 8.0 Hz, 1H), 7.05 (d, *J* = 8.0 Hz, 1H), 6.89 (dt, *J* = 10, 3.5 Hz, 2H), 6.32 (dt, *J* = 10, 3.0 Hz, 2H), 4.21 (s, 2H), 2.24 (s, 3H), 2.21 (s, 3H), 1.49 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 185.1, 151.1, 137.5, 136.6, 132.7, 130.3, 129.5, 129.1, 119.6, 87.1, 84.8, 73.1, 54.6, 26.4, 19.7, 19.5; IR (neat) ν_{max} 2932, 2224, 1662, 1627, 1493, 1449, 1372, 1309, 1245, 1182 cm⁻¹; HRMS (ESI) *m*/*z* [M + H]⁺ calcd for C₁₈H₁₉O₂⁺ 267.1380, found 267.1389.

4-Pentyl-4-((3-phenylprop-2-yn-1-yl)oxy)cyclohexa-2,5-dienone (1u). Compound 1u (218 mg) was obtained in 74% yield: colorless solid; mp = 64–66 °C; R_f = 0.53 (9:1 hexane/EtOAc) [silica, UV, and I_2]; ¹H NMR (500 MHz, CDCI₃) δ 7.42–7.39 (m, 2H), 7.32–7.28 (m, 3H), 6.85 (dt, *J* = 10, 3.0 Hz, 2H), 6.37 (dt, *J* = 10.5, 3.0 Hz, 2H), 4.23 (s, 2H), 1.82–1.77 (m, 2H), 1.29–1.22 (m, 6H), 0.85 (t, *J* = 6.5 Hz, 3H); ¹³C{¹H} NMR (126 MHz, CDCI₃) δ 185.4, 150.4, 131.6, 131.3, 128.5, 128.2, 122.4, 86.7, 85.9, 76.4, 54.3, 39.3, 31.8, 23.1, 22.3, 13.9; IR (neat) ν_{max} 2964, 2932, 1662, 1621, 1496, 1449, 1374, 1334, 1264, 1182 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₂₃O₂⁺ 295.1693, found 295.1704.

General Procedure for the Synthesis of α -Aryl- α , β -unsaturated Ketones **2**. A mixture of alkynyl-cyclohexadienone **1** (0.3 mmol) and Cu(OTf)₂ (11 mg, 0.03 mmol) in THF (3.0 mL) was taken in a screw-cap sealed tube. The reaction mixture was heated to 95 °C and stirred overnight. The reaction progress was monitored by thin-layer chromatography (TLC) analysis. Upon completion, the reaction mixture was cooled to room temperature and diluted with dichloromethane (3.0 mL). The crude residue was purified using column chromatography on silica gel (100–200 mesh) to provide desired α -aryl- α , β -unsaturated ketones **2**.

2-(5-Hydroxy-2-methylphenyl)-1-phenylprop-2-en-1-one (2a). Following the general procedure, compound 2a (68 mg) was obtained in 95% yield: yellow gummy liquid; $R_f = 0.54$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (600 MHz, CDCI₃) δ 7.88 (d, J = 7.8 Hz, 2H), 7.56 (t, J = 7.2 Hz, 1H), 7.45 (t, J = 7.2 Hz, 2H), 7.00 (d, J = 7.2 Hz, 1H), 6.76 (d, J = 2.4 Hz, 1H), 6.72 (dd, J = 7.8, 3.0 Hz, 1H), 6.00 (s, 1H), 5.97 (s, 1H), 5.09 (s, 1H), 2.12 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCI₃) δ 197.3, 154.1, 148.8, 138.8, 136.9, 132.8, 131.2, 129.8, 128.4, 128.3, 127.0, 116.8, 115.5, 19.4; IR (neat) ν_{max} 3368, 3058, 2922, 1649, 1596, 1576, 1495, 1446, 1281, 1253, 1228 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₅O₂⁺ 239.1067, found 239.1073.

2-(5-Hydroxy-2-methylphenyl)-1-(p-tolyl)prop-2-en-1-one (**2b**). Following the general procedure, compound **2b** (73 mg) was obtained in 96% yield: colorless solid; mp = 148–151 °C; R_f = 0.46 (4:1 hexane/ EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.81 (dd, *J* = 8.4, 1.6 Hz, 2H), 7.23 (d, *J* = 8.0 Hz, 2H), 6.95 (d, *J* = 8.0 Hz, 1H), 6.74 (bd, *J* = 2.4 Hz, 1H), 6.68 (dd, *J* = 8.4, 2.8 Hz, 1H), 6.40 (bs, 1H), 5.94 (s, 1H), 5.92 (s, 1H), 2.41 (s, 3H), 2.09 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 197.1, 154.1, 148.9, 143.8, 139.0, 134.2, 131.2, 130.1, 129.0, 127.7, 127.0, 116.9, 115.5, 21.6, 19.5; IR (neat) ν_{max} 3331, 3031, 2920, 1631, 1597, 1563, 1498, 1437, 1408, 1286, 1218 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₂⁺ 253.1223, found 253.1228.

2-(5-Hydroxy-2-methylphenyl)-1-(4-methoxyphenyl)prop-2-en-1-one (**2c**). Following the general procedure, compound **2c** (76 mg) was obtained in 94% yield: colorless solid; mp = 178–180 °C; R_f = 0.35 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, DMSO-d₆) δ 9.29 (s, 1H), 7.84 (d, *J* = 8.5 Hz, 2H), 7.04 (d, *J* = 9.0 Hz, 2H), 6.95 (d, *J* = 8.0 Hz, 1H), 6.69–6.64 (m, 2H), 5.91 (s, 1H), 5.79 (s, 1H), 3.83 (s, 3H), 1.99 (s, 3H); ¹³C{¹H} NMR (126 MHz, DMSO-d₆) δ 194.9, 163.5, 155.8, 149.1, 139.6, 132.2, 131.5, 129.6, 126.2, 125.4, 116.7, 115.5, 114.3, 56.0, 19.5; IR (neat) ν_{max} 3282, 3022, 2918, 1624, 1591, 1563, 1313, 1291, 1255, 1217 cm⁻¹; HRMS (ESI) *m*/*z* [M + Na]⁺ calcd for C₁₇H₁₆NaO₃⁺ 291.0992, found 291.0998.

1-(4-(tert-Butyl)phenyl)-2-(5-hydroxy-2-methylphenyl)prop-2en-1-one (2d). Following the general procedure, compound 2d (86 mg) was obtained in 97% yield: pale yellow solid; mp = 127–130 °C; R_f = 0.64 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.86 (dt, J = 8.4, 2.0 Hz, 2H), 7.46 (dt, J = 8.8, 2.0 Hz, 2H), 6.98 (d, J = 8.4 Hz, 1H), 6.75 (d, J = 2.8 Hz, 1H), 6.70 (dd, J = 8.0, 2.8 Hz, 1H), 5.95 (dd, J = 14, 1.2 Hz, 2H), 5.85 (bs, 1H), 2.11 (s, 3H), 1.34 (s, 9H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 196.7, 156.6, 154.0, 149.0, 139.2, 134.3, 131.2, 129.9, 127.5, 127.3, 125.3, 116.9, 115.4, 35.1, 31.1, 19.5; IR (neat) ν_{max} 3382, 3055, 1602, 1497, 1461, 1375, 1264 cm⁻¹; HRMS (ESI) m/z [M + Na]⁺ calcd for C₂₀H₂₂NaO₂⁺ 317.1512, found 317.1517.

1-(4-Acetylphenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (**2e**). Following the general procedure, compound **2e** (52 mg) was obtained in 62% yield: yellow solid; mp = 193–196 °C; R_f = 0.21 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, DMSO–D₆) δ 9.05 (s, 1H), 8.00 (s, 1H), 7.99 (d, *J* = 3.6 Hz, 1H), 7.87 (d, *J* = 8.4 Hz, 2H), 6.91 (d, *J* = 8.8 Hz, 1H), 6.70–6.63 (m, 2H), 6.01 (s, 1H), 5.91 (s, 1H), 2.60 (s, 3H), 2.01 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃ + DMSO-d₆) δ 195.4, 193.7, 153.9, 147.0, 138.9, 137.9, 136.8, 129.3, 127.9, 126.9, 126.5, 123.6, 114.9, 113.8, 25.3, 17.6; IR (neat) ν_{max} 3310, 2919, 1683, 1637, 1604, 1560, 1494, 1451, 1354, 1286, 1253 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₇O₃⁺ 281.1172, found 281.1179. 1-(4-Fluorophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (**2f**). Following the general procedure, compound **2f** (62 mg) was obtained in 80% yield: colorless liquid; $R_f = 0.64$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃): δ 7.95–7.89 (m, 2H), 7.13–7.07 (m, 2H), 6.96 (d, J = 8.0 Hz, 1H), 6.73 (d, J = 2.5 Hz, 1H), 6.69 (dd, J = 8.0, 2.5 Hz, 1H), 6.33 (s, 1H), 5.96 (d, J = 0.5 Hz, 1H), 5.93 (d, J = 0.5 Hz, 1H), 2.07 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 195.7, 165.6 (d, J = 256 Hz, 1C), 154.1, 148.7, 138.7, 133.1 (d, J = 2.5 Hz, 1C), 132.5 (d, J = 10 Hz, 1C), 131.3, 128.0, 127.1, 116.8, 115.6, 115.4, 19.4; ¹⁹F NMR (471 MHz, CDCl₃) δ – 104.9; IR (neat) $ν_{max}$ 3368, 2923, 1643, 1596, 1502, 1282, 1228, 1154 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄FO₂⁺ 257.0972, found 257.0977.

1-(4-Chlorophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (**2g**). Following the general procedure, compound **2g** (61 mg) was obtained in 75% yield: pale yellow solid; mp = 131–134 °C; $R_f = 0.50$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCI₃) δ 7.81 (dd, J = 8.4, 1.6 Hz, 2H), 7.39 (dd, J = 8.4, 1.6 Hz, 2H), 6.96 (d, J = 8.0 Hz, 1H), 6.73 (d, J = 2.8 Hz, 1H), 6.68 (dd, J = 8.4, 2.4 Hz, 1H), 6.33 (bs, 1H), 5.97 (s, 1H), 5.94 (s, 1H), 2.07 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCI₃) δ 195.9, 154.0, 148.6, 139.3, 138.6, 135.1, 131.3, 131.2, 128.7, 128.4, 127.1, 116.7, 115.6, 19.4; IR (neat) ν_{max} 3361, 2922, 1630, 1602, 1583, 1562, 1498, 1432, 1400, 1346, 1284, 1207 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄ClO₂⁺ 273.0677, found 273.0683.

1-(4-Bromophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (2h). Following the general procedure, compound 2h (55 mg) was obtained in 58% yield: yellow solid; mp = 157–160 °C; R_f = 0.50 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.73 (d, J = 8.8 Hz, 2H), 7.57 (d, J = 8.8 Hz, 2H), 6.97 (d, J = 8.0 Hz, 1H), 6.73 (bd, J = 2.8 Hz, 1H), 6.69 (dd, J = 8.0, 2.8 Hz, 1H), 5.98 (d, J = 0.8 Hz, 1H), 5.95 (d, J = 0.8 Hz, 1H), 2.07 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 195.9, 154.0, 148.6, 138.7, 135.6, 131.7, 131.4, 131.3, 128.4, 128.0, 127.2, 116.7, 115.6, 19.4; IR (neat) ν_{max} 3359, 2922, 1628, 1603, 1580, 1497, 1451, 1398, 1283, 1254 cm⁻¹; HRMS (ESI) m/z [M + Na]⁺ calcd for C₁₆H₁₃BrNaO₂⁺ 338.9991, found 339.0008.

2-(5-Hydroxy-2-methylphenyl)-1-(4-iodophenyl)prop-2-en-1-one (2i). Following the general procedure, compound 2i (67 mg) was obtained in 61% yield: colorless solid; mp = 158–160 °C; R_f = 0.58 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, DMSO-d₆) δ 9.33 (s, 1H), 7.91 (d, *J* = 8.4 Hz, 2H), 7.58 (d, *J* = 8.4 Hz, 2H), 6.66 (bd, *J* = 2.4 Hz, 1H), 6.65 (s, 1H), 6.03 (s, 1H), 5.88 (s, 1H), 1.97 (s, 3H); ¹³C{¹H} NMR (101 MHz, DMSO-d₆) δ 195.9, 155.9, 148.5, 139.2, 138.1, 136.6, 131.6, 131.5, 128.7, 125.5, 116.9, 115.7, 102.0, 19.6; IR (neat) ν_{max} 3357, 1627, 1575, 1496, 1432, 1390, 1344, 1283 cm⁻¹; HRMS (ESI) *m*/*z* [M + H]⁺ calcd for C₁₆H₁₄IO₂⁺ 365.0033, found 365.0042.

2-(5-Hydroxy-2-methylphenyl)-1-(m-tolyl)prop-2-en-1-one (2j). Following the general procedure, compound 2j (59 mg) was obtained in 78% yield: pale yellow solid; mp = 152–154 °C; R_f = 0.57 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.72 (s, 1H), 7.67 (d, *J* = 7.6 Hz, 1H), 7.37 (d, *J* = 7.6 Hz, 1H), 7.32 (t, *J* = 7.6 Hz, 1H), 6.96 (d, *J* = 8.4 Hz, 1H), 6.74 (d, *J* = 2.8 Hz, 1H), 6.69 (dd, *J* = 8.4, 2.8 Hz, 1H), 6.61 (bs, 1H), 5.98 (bd, *J* = 1.2 Hz, 1H), 5.93 (bd, *J* = 1.2 Hz, 1H), 2.39 (s, 3H), 2.10 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 197.5, 154.1, 148.9, 138.8, 138.2, 137.0, 133.6, 131.1, 130.2, 128.3, 128.1, 127.2, 127.0, 116.9, 115.5, 21.3, 19.4; IR (neat) ν_{max} 3383, 1654, 1602, 1581, 1496, 1358, 1264 cm⁻¹; HRMS (ESI) *m*/*z* [M + H]⁺ calcd for C₁₇H₁₇O₂⁺ 253.1223, found 253.1232.

1-(3-Fluorophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (2k). Following the general procedure, compound 2k (42 mg) was obtained in 54% yield: pale yellow solid; mp = 120–122 °C; R_f = 0.53 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCI₃) δ 7.65 (d, J = 8.0 Hz, 1H), 7.59–7.54 (m. 1H), 7.44–7.39 (m, 1H), 7.28–7.23 (m, 1H), 6.99 (d, J = 8.0 Hz, 1H), 6.74–6.69 (m, 2H), 6.02 (s, 1H), 5.98 (s, 1H), 5.74 (bs, 1H), 2.09 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCI₃) δ 195.3, 162.5 (d, J = 250 Hz, 1C), 153.9, 148.5, 139.1 (d, J = 6.3 Hz, 1C), 138.7, 131.4, 130.0 (d, J = 7.6 Hz, 1C), 128.6, 127.3, 125.5 (d, J = 2.5 Hz, 1C), 119.8 (d, J = 21 Hz, 1C), 116.7, 116.5 (d, J = 22 Hz, 1C), 115.6, 19.4; ¹⁹F NMR (376 MHz, CDCI₃) δ – 111.7; IR (neat) ν_{max} 3371, 3056, 1656, 1585, 1497, 1440, 1264 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄FO₂⁺ 257.0972, found 257.0977.

1-(3-Chlorophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (2l). Following the general procedure, compound 2l (53 mg) was obtained in 65% yield: colorless solid; mp = 134–136 °C; R_f = 0.48 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.86 (bt, *J* = 1.5 Hz, 1H), 7.73 (d, *J* = 7.5 Hz, 1H), 7.54–7.50 (m, 1H), 7.37 (t, *J* = 8.0 Hz, 1H), 6.97 (d, *J* = 8.5 Hz, 1H), 6.72 (d, *J* = 3.0 Hz, 1H), 6.69 (dd, *J* = 8.0, 2.5 Hz, 1H), 6.40 (bs, 1H), 6.02 (s, 1H), 5.96 (s, 1H), 2.08 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 195.6, 154.0, 148.5, 138.6, 138.4, 134.6, 132.7, 131.3, 129.64, 129.59, 128.9, 127.9, 127.1, 116.8, 115.7, 19.4; IR (neat) ν_{max} 3383, 3064, 1661, 1569, 1497, 1419, 1235 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄ClO₂⁺ 273.0677, found 273.0682.

2-(5-Hydroxy-2-methylphenyl)-1-(3-iodophenyl)prop-2-en-1-one (2m). Following the general procedure, compound 2m (68 mg) was obtained in 62% yield: colorless semisolid; mp = $125-127 \,^{\circ}C$; $R_f = 0.49$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCI₃) δ 8.01–7.97 (m, 1H), 7.68 (dd, J = 8.0, 0.8 Hz, 1H), 7.60 (dd, J = 8.0, 1.2 Hz, 1H), 7.00 (t, J = 7.6 Hz, 1H), 6.81–6.75 (m, 1H), 6.60–6.54 (m, 2H), 5.82 (s, 1H), 5.76 (s, 1H), 1.88 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCI₃ + DMSO-d₆) δ 194.3, 154.9, 148.2, 140.8, 138.6, 138.0, 137.7, 130.7, 129.6, 128.4, 127.9, 125.4, 116.2, 115.2, 93.6, 19.0; IR (neat) ν_{max} 3360, 1656, 1579, 1495, 1457, 1410, 1229 cm⁻¹; HRMS (ESI) $m/z [M + H]^+$ calcd for C₁₆H₁₄IO₂+ 365.0033, found 365.0040. The –OH proton is not detected by ¹H NMR.

2-(5-Hydroxy-2-methylphenyl)-1-(o-tolyl)prop-2-en-1-one (2n). Following the general procedure, compound 2n (57 mg) was obtained in 75% yield: pale yellow solid; mp = 156–158 °C; $R_f = 0.60$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.45 (dd, *J* = 7.6, 1.2 Hz, 1H), 7.39–7.34 (m, 1H), 7.29–7.20 (m, 2H), 7.05 (d, *J* = 8.4 Hz, 1H), 6.72 (dd, *J* = 8.4, 2.8 Hz, 1H), 6.68 (d, *J* = 2.4 Hz, 1H), 6.09 (bd, *J* = 0.8 Hz, 1H), 5.97 (bd, *J* = 1.2 Hz, 1H), 5.28 (bs, 1H), 2.43 (s, 3H), 2.17 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 198.6, 153.6, 150.1, 138.3, 138.1, 137.2, 131.9, 131.2, 131.1, 130.4, 128.6, 127.9, 125.0, 116.8, 115.2, 20.0, 19.3; IR (neat) ν_{max} 3385, 3055, 1659, 1497, 1320, 1264 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₂⁺ 253.1223, found 253.1229.

2-(5-Hydroxy-2-methylphenyl)-1-(2-methoxyphenyl)prop-2-en-1-one (**20**). Following the general procedure, compound **20** (80 mg) was obtained in 99% yield: yellow liquid; $R_f = 0.36$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.44–7.39 (m, 2H), 7.00 (d, J = 7.5 Hz, 2H), 6.92 (d, J = 8.5 Hz, 1H), 6.68 (dd, J = 8.0, 3.0 Hz, 1H), 6.66 (d, J = 3.0 Hz, 1H), 5.98 (d, J = 1.0 Hz, 1H), 5.96 (d, J = 1.0 Hz, 1H), 5.83 (bs, 1H), 3.78 (s, 3H), 2.16 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 197.2, 157.3, 153.6, 150.1, 138.2, 132.0, 130.9, 130.4, 129.7, 128.7, 127.8, 120.3, 116.9, 115.1, 111.4, 55.4, 19.2; IR (neat) ν_{max} 3379, 2924, 1651, 1597, 1487, 1460, 1285, 1248 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₃⁺ 269.1172, found 269.1179.

1-(2-Chlorophenyl)-2-(5-hydroxy-2-methylphenyl)prop-2-en-1one (**2p**). Following the general procedure, compound **2p** (36 mg) was obtained in 44% yield: yellow semisolid; $R_f = 0.50$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.45–7.42 (m, 1H), 7.41–7.38 (m, 2H), 7.35–7.31 (m, 1H), 7.06 (d, J = 8.5 Hz, 1H), 6.72 (dd, J = 8.0, 2.5 Hz, 1H), 6.67 (d, J = 2.5 Hz, 1H), 6.16 (d, J = 0.5 Hz, 1H), 6.02 (d, J = 0.5 Hz, 1H), 5.50 (bs, 1H), 2.20 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 195.5, 153.5, 149.0, 138.4, 137.3, 133.7, 131.2, 131.1, 131.0, 130.1, 129.0, 128.1, 126.5, 116.8, 115.4, 19.2; IR (neat) ν_{max} 3349, 3056, 2922, 1641, 1604, 1587, 1496, 1346, 1286, 1205 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₆H₁₄ClO₂⁺ 273.0677, found 273.0686.

1-(3,4-Dimethylphenyl)-2-(5-hydroxy-2-methylphenyl)prop-2en-1-one (2q). Following the general procedure, compound 2q (73 mg) was obtained in 91% yield: pale yellow solid; mp = 150–152 °C; R_f = 0.44 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.71 (s, 1H), 7.64 (dd, *J* = 7.6, 1.6 Hz, 1H), 7.19 (d, *J* = 8.0 Hz, 1H), 6.95 (d, *J* = 8.0 Hz, 1H), 6.72 (d, *J* = 2.4 Hz, 1H), 6.67 (dd, *J* = 8.4, 2.8 Hz, 1H), 6.33 (bs, 1H), 5.95 (bd, *J* = 1.2 Hz, 1H), 5.90 (bd, *J* = 1.2 Hz, 1H), 2.32 (s, 3H), 2.30 (s, 3H), 2.09 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 197.3, 154.1, 149.0, 142.6, 139.0, 136.8, 134.6, 131.2, 131.0, 129.5, 127.9, 127.6, 127.0, 116.9, 115.5, 20.0, 19.7, 19.5; IR (neat) ν_{max} 3337, 2921, 2854, 1626, 1600, 1563, 1498, 1437, 1356, 1291, 1224 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₉O₂⁺ 267.1380, found 267.1385.

2-(5-Hydroxy-2-methylphenyl)-1-(thiophen-2-yl)prop-2-en-1one (**2***r*). Following the general procedure, compound **2r** (72 mg) was obtained in 98% yield: yellow liquid; $R_f = 0.55$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.63 (dd, J = 5.0, 1.0Hz, 1H), 7.53 (dd, J = 4.0, 1.0 Hz, 1H), 7.05–7.02 (m, 1H), 6.96 (d, J = 8.0 Hz, 1H), 6.79 (d, J = 3.0 Hz, 1H), 6.74 (dd, J = 8.5, 2.5 Hz, 1H), 6.13 (bd, J = 1.0 Hz, 1H), 5.81 (bd, J = 1.0 Hz, 1H), 2.07 (s, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 188.2, 154.2, 148.7, 143.1, 138.7, 134.8, 134.6, 131.3, 128.1, 127.4, 126.5, 116.9, 115.8, 19.4; IR (neat) ν_{max} 3365, 2920, 1605, 1577, 1495, 1407, 1353, 1233, 1052 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₄H₁₃O₂S⁺ calcd 245.0631, found 245.0637. The –OH proton is not detected by ¹H NMR.

2-(2-Ethyl-5-hydroxyphenyl)-1-phenylprop-2-en-1-one (**2s**). Following the general procedure, compound **2s** (58 mg) was obtained in 76% yield: yellow liquid; $R_f = 0.64$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.89 (dd, J = 8.0, 1.0 Hz, 2H), 7.59–7.54 (m, 1H), 7.45 (t, J = 8.0 Hz, 2H), 7.02 (d, J = 8.0 Hz, 1H), 6.74 (dd, J = 8.5, 3.0 Hz, 1H), 6.72 (d, J = 2.5 Hz, 1H), 6.31 (bs, 1H), 6.00 (bd, J = 0.5 Hz, 1H), 5.95 (bd, J = 1.0 Hz, 1H), 2.43 (q, J = 7.5 Hz, 2H), 1.09 (t, J = 7.5 Hz, 3H); ¹³C{¹H} NMR (126 MHz, CDCl₃) δ 197.3, 153.9, 148.7, 138.4, 137.0, 133.3, 132.8, 129.8, 129.4, 128.8, 128.3, 117.1, 115.8, 25.8, 15.3; IR (neat) ν_{max} 3367, 2964, 2871, 1648, 1597, 1576, 1495, 1446, 1260, 1226 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₂⁺ 253.1223, found 253.1233.

2-(5-Hydroxy-2-propylphenyl)-1-phenylprop-2-en-1-one (2t). Following general the procedure, compound 2t (64 mg) was obtained in 80% yield: pale yellow solid; mp = 116–118 °C; R_f = 0.56 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 7.90 (dd, *J* = 7.6, 0.4 Hz, 2H), 7.60–7.55 (m, 1H), 7.45 (t, *J* = 8.0 Hz, 2H), 7.02–6.97 (m, 1H), 6.75–6.70 (m, 2H), 6.48 (bs, 1H), 5.99 (bd, *J* = 1.2 Hz, 1H), 5.94 (bd, *J* = 0.8 Hz, 1H), 2.36 (t, *J* = 7.6 Hz, 2H), 1.55– 1.45 (m, 2H), 0.83 (t, *J* = 7.2 Hz, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 197.4, 153.9, 148.8, 138.6, 136.9, 132.8, 131.8, 130.2, 129.8, 128.8, 128.3, 117.3, 115.6, 35.1, 24.4, 13.9; IR (neat) ν_{max} 3367, 2956, 2867, 1629, 1595, 1493, 1441, 1345, 1318, 1253 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₉O₂⁺ 267.1380, found 267.1386.

2-(5-Hydroxy-2-pentylphenyl)-1-phenylprop-2-en-1-one (2u). Following the general procedure, compound 2u (45 mg) was obtained in 51% yield: colorless gummy liquid; $R_f = 0.70$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.90 (dd, J = 8.0, 1.0Hz, 2H), 7.59–7.55 (m, 1H), 7.45 (t, J = 8.0 Hz, 2H), 7.00 (dd, J = 7.0, 1.5 Hz, 1H), 6.75–6.71 (m, 2H), 6.12 (bs, 1H), 5.99 (bd, J = 1.5 Hz, 1H), 5.95 (bd, J = 1.5 Hz, 1H), 2.37 (t, J = 8.0 Hz, 2H), 1.50–1.43 (m, 2H), 1.27–1.15 (m, 4H), 0.82 (t, J = 7.0 Hz, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 197.2, 153.8, 148.8, 138.6, 137.0, 132.8, 132.2, 130.2, 129.9, 128.7, 128.3, 117.2, 115.6, 33.0, 31.6, 30.9, 22.4, 14.0; IR (neat) $ν_{max}$ 3381, 2954, 2926, 2857, 1648, 1597, 1576, 1494, 1445, 1260, 1228 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₀H₂₃O₂⁺ 295.1693, found 295.1695.

2-(5-Hydroxy-2,4-dimethylphenyl)-1-phenylprop-2-en-1-one (**2v**). Following the general procedure, compound **2v** (64 mg) was obtained in 85% yield: pale yellow solid; mp = 102–105 °C; R_f = 0.60 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (600 MHz, CDCI₃) δ 7.89 (dd, *J* = 8.4, 1.2 Hz, 2H), 7.57–7.53 (m, 1H), 7.44 (t, *J* = 7.8 Hz, 2H), 6.89 (s, 1H), 6.66 (s, 1H), 5.96 (bd, *J* = 1.2 Hz, 1H), 5.91 (bd, *J* = 1.2 Hz, 1H), 5.58 (bs, 1H), 2.18 (s, 3H), 2.09 (s, 3H); ¹³C{¹H} NMR (151 MHz, CDCI₃) δ 197.2, 152.1, 148.8, 137.1, 136.3, 132.8, 132.7, 129.9, 128.3, 127.8, 127.1, 124.3, 116.4, 19.4, 15.5; IR (neat) ν_{max} 3372, 1630, 1592, 1507, 1445, 1412, 1348, 1324, 1270 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₇H₁₇O₂⁺ 253.1223, found 253.1240.

2-(4-Hydroxy-[1,1'-biphenyl]-2-yl)-1-phenylprop-2-en-1-one (2w). Following the general procedure, compound 2w (60 mg) was obtained in 66% yield: colorless solid; mp = $205-207 \,^{\circ}$ C; $R_f = 0.41$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (600 MHz, CDCl₃) δ 7.46–7.40 (m, 3H), 7.28–7.23 (m, 2H), 7.19 (t, J = 7.2 Hz, 2H), 7.16– 7.09 (m, 4H), 6.92 (d, J = 3.0 Hz, 1H), 6.84 (dd, J = 9.0, 3.0 Hz, 1H), 6.25 (bs, 1H), 6.08 (s, 1H), 5.91 (s, 1H); $^{13}C\{^{1}H\}$ NMR (151 MHz, CDCl₃) δ 196.6, 155.4, 149.1, 140.7, 138.5, 136.7, 133.6, 132.3, 131.2, 129.9, 129.6, 128.8, 128.0, 127.7, 126.7, 117.5, 115.8; IR (neat) $\nu_{\rm max}$ 3324, 1978, 1640, 1597, 1447, 1271, 1200, 980 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₂₁H₁₇O₂+ 301.1223, found 301.1230.

2-(5-Hydroxy-2-methylphenyl)-1-(o-tolyl)but-2-en-1-one (2aa). Following the general procedure, compound 2aa (60 mg) was obtained in 75% yield: pale yellow liquid; $R_f = 0.39$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (600 MHz, CDCl₃) δ 7.36–7.28 (m, 2H), 7.25–7.19 (m, 2H), 7.12 (d, J = 7.8 Hz, 1H), 6.72 (dd, J = 7.8, 2.4 Hz, 1H), 6.65 (q, J = 6.6 Hz, 1H), 6.55 (d, J = 2.4 Hz, 1H), 5.04 (bs, 1H), 2.36 (s, 3H), 2.12 (s, 3H), 1.68 (d, J = 7.2 Hz, 3H); ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 199.0, 153.5, 146.7, 144.5, 139.6, 135.8, 135.7, 131.1, 130.7, 129.5, 128.4, 127.5, 125.1, 116.6, 115.0, 19.63, 18.74, 16.02; IR (neat) ν_{max} 3317, 2943, 2831, 1449, 1416, 1264, 1021 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₉O₂⁺ 267.1380, found 267.1394.

2-(5-Hydroxy-2-methylphenyl)-1-(o-tolyl)but-2-en-1-one (2ab). Following the general procedure, compound 2ab (62 mg) was obtained in 73% overall yield (inseparable mixture of diasteriomers): colorless solid; mp = 158–160 °C; R_f = 0.33 (4:1 hexane/EtOAc) [silica, UV, and I_2]; ¹H NMR (600 MHz, CDCl₃) δ 7.99 (dd, J = 7.2, 1.8 Hz, 0.77H), 7.80 (dd, J = 6.6, 1.8 Hz, 2H), 7.09-7.03 (m, 1.37H), 6.98-6.88 (m, 3H), 6.71 (dd, J = 8.4, 3.0 Hz, 1H), 6.67–6.59 (m, 3H), 3.88 (s, 0.67H), 3.86 (s, 3H), 2.19 (s, 0.94H), 2.10 (s, 3H), 1.83 (d, J = 7.2 Hz, 0.94H), 1.75 (d, J = 7.2 Hz, 3H); ${}^{13}C{}^{1}H{}$ NMR (151 MHz, CDCl₃) *δ* 196.5, 195.6, 163.6, 162.8, 153.9, 153.6, 142.4, 140.6, 136.8, 134.8, 132.0, 131.3, 131.1, 130.9, 130.8, 130.7, 129.8, 128.9, 128.5, 128.2, 117.2, 117.0, 114.9, 114.0, 113.9, 113.83, 113.78, 113.4, 55.5, 55.4, 43.1, 18.9, 18.85, 15.6; IR (neat) ν_{max} 3282, 3011, 2921, 1727, 1651, 1617, 1592, 1502, 1418, 1280 cm⁻¹; HRMS (ESI) m/z [M + H] calcd for C₁₈H₁₉O₃⁺ 283.1329, found 283.1338. The -OH proton is not detected by ¹H NMR.

2-(5-Hydroxy-2-methylphenyl)-1-(4-methoxyphenyl)but-2-en-1one (**2ac**). Following the general procedure, compound **2ac** (73 mg) was obtained in 86% overall yield (inseparable mixture of diasteriomers): colorless solid; mp = 164–166 °C; R_f = 0.34 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (600 MHz, CDCl₃) δ 8.03 (d, *J* = 8.4 Hz, 0.11H), 7.99 (d, *J* = 9.6 Hz, 0.81H), 7.80 (d, *J* = 9.6 Hz, 2H), 7.05 (t, *J* = 8.4 Hz, 1.18H), 6.98–6.88 (m, 3.20H), 6.70 (dd, *J* = 7.8, 2.4 Hz, 1H), 6.66–6.58 (m, 2.28H), 5.24 (bs, 1H), 5.11 (bs, 0.12H), 4.18 (s, 0.32H), 3.87 (s, 3H), 2.18 (s, 0.28H), 2.10 (s, 3H), 1.75 (d, *J* = 6.6 Hz, 3H), 1.31 (d, *J* = 6.6 Hz, 0.20H); ¹³C{¹H} NMR (151 MHz, CDCl₃) δ 196.4, 195.5, 163.6, 162.8, 153.8, 153.5, 142.4, 140.6, 131.9, 131.1, 130.7, 128.3, 117.1, 117.0, 114.8, 114.0, 113.9, 113.8, 113.4, 55.5, 55.4, 43.1, 18.91, 18.86, 15.6; IR (neat) ν_{max} 2985, 2907, 2257, 1737, 1447, 1373, 1235, 1098, 1044 cm⁻¹; HRMS (ESI) m/z [M + H]⁺ calcd for C₁₈H₁₉O₃⁺ 283.1329, found 283.1339.

Synthesis of 2-(5-Hydroxy-2-methylphenyl)-2,3-dihydro-1Hinden-1-one (3). A mixture of 2a (0.3 mmol, 72 mg) in triflic acid (30 mmol, 2.3 mL) was taken in a screw-cap sealed tube. The reaction mixture was stirred at room temperature for 72 h. Upon completion, the whole mixture was poured into ice-water (30 mL) and extracted with CH_2Cl_2 (15 mL \times 3). The combined organic layers were washed with brine (30 mL), dried over Na₂SO₄, filtered, and evaporated to give a residue, which was purified by silica gel (100-200 mesh) column chromatography (hexane/ethyl acetate) to give the desired product 3 in 63% yield (45 mg) as a colorless solid: mp = 166–168 °C; R_f = 0.40 (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, CDCl₃) δ 8.34 (s, 1H), 7.64–7.58 (m, 1H), 7.52–7.43 (m, 1H), 7.36 (d, J = 7.6 Hz, 1H), 6.84 (dd, J = 8.0, 2.8 Hz, 1H), 6.51–6.44 (m, 1H), 6.31–6.24 (m, 1H), 3.86 (dd, J = 8.4, 4.4 Hz, 1H), 3.56–3.46 (m, 1H), 2.96 (dd, J = 17.2, 3.6 Hz, 1H), 2.73 (bs, 1H), 2.08 (s, 3H); ¹³C{¹H} NMR (101 MHz, CDCl₃) δ 206.4, 155.1, 153.2, 139.0, 136.1, 134.6, 131.0, 127.2, 126.6, 126.2, 123.6, 114.0, 113.7, 50.3, 35.1, 18.6; IR (neat) ν_{max} 3252, 1679, 1603, 1584, 1500, 1462, 1435, 1381, 1279 cm⁻¹; HRMS (ESI) $m/z [M + H]^+$ calcd for $C_{16}H_{15}O_2^+$ 239.1067, found 239.1080.

Synthesis of 2-(5-Hydroxy-2-methylphenyl)-1-(4-methoxyphenyl)propan-1-one (4). A mixture of compound <math>2c (0.2 mmol, 54 mg) and NaBH₄ (0.4 mmol, 15 mg) was taken in a Schlenk

tube. To the above mixture was added methanol (2.5 mL) at 0 °C, and the mixture was stirred for 20 h at the same temperature. Upon completion, the reaction was quenched with water (20 mL) and extracted in ethyl acetate (15 mL \times 3). The combined organic layers were dried with anhydrous Na2SO4, filtered, and concentrated in reduced pressure. The crude residue was purified by silica gel column chromatography (100-200 mesh) to provide 4 in 68% yield (37 mg) as a colorless solid: mp = 163-165 °C; $R_f = 0.50$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, CDCl₃) δ 7.76 (d, J = 8.0 Hz, 2H), 7.08 (d, J = 8.5 Hz, 1H), 6.79 (d, J = 8.0 Hz, 2H), 6.66 (dd, J = 8.0, 2.0 Hz, 1H), 6.63 (s, 1H), 6.59 (d, J = 2.5 Hz, 1H), 4.67 (q, J = 6.5 Hz, 1H), 3.79 (s, 3H), 2.43 (s, 3H), 1.40 (d, J = 7.0 Hz, 3H); ${}^{13}C{}^{1}H$ NMR (126 MHz, CDCl₃) δ 200.5, 163.3, 154.9, 141.5, 132.0, 130.9, 129.2, 125.7, 114.0, 113.7, 113.6, 55.4, 44.2, 18.6, 18.0; IR (neat) ν_{max} 3317, 1653, 1591, 1511, 1449, 1340, 1303, 1271 cm⁻¹; HRMS (ESI) *m*/*z* [M + H]⁺ calcd for $C_{17}H_{19}O_3^+$ 271.1329, found 271.1342.

Synthesis of 2-(5-Hydroxy-2-methylphenyl)-1-(4-methoxyphenyl)-3-(p-tolylthio)propan-1-one (5). To a solution of 2c (0.2 mmol, 54 mg) in CHCl₃ (2.5 mL) were added 4-methylbenzenethiol (0.4 mmol, 50 mg) and trimethylamine (0.6 mmol, 83 μ L) subsequently. The resulting mixture was stirred at room temperature for 48 h. Upon completion, the crude reaction mixture was purified using column chromatography (hexane/ethyl acetate) on silica gel (100-200 mesh) to afford 5 in 91% yield (72 mg) as a colorless solid: mp = 129-131 °C; $R_f = 0.42$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (500 MHz, $(DCl_3) \delta 8.42$ (s, 1H), 7.42 (d, J = 8.5 Hz, 2H), 6.92 (d, J = 8.0 Hz, 2H), 6.77 (d, J = 7.5 Hz, 2H), 6.62 (d, J = 9.0 Hz, 1H), 6.51 (d, J = 9.0 Hz, 2H), 6.26–6.15 (m, 2H), 4.44 (dd, J = 9.5, 4.0 Hz, 1H), 3.46 (s, 3H), 3.28 (dd, J = 13.5, 10 Hz, 1H), 2.66 (dd, J = 13.5, 4.0 Hz, 1H), 1.99 (s, 3H), 1.81 (s, 3H); ${}^{13}C{}^{1}H$ NMR (126 MHz, CDCl₃) δ 196.1, 162.5, 155.0, 136.8, 135.9, 131.4, 131.1, 130.5, 129.7, 128.9, 128.8, 124.3, 113.9, 113.0, 112.9, 54.6, 48.3, 37.0, 20.2, 17.6; IR (neat) $\nu_{\rm max}$ 3369, 1670, 1597, 1506, 1459, 1412, 1261, 1235 cm⁻¹; HRMS (ESI) $m/z [M + H]^+$ calcd for $C_{24}H_{25}O_3S^+$ 393.1519, found 393.1530.

Synthesis of (2-(5-Hydroxy-2-methylphenyl)oxiran-2-yl)(4methoxyphenyl)methanone (6). A mixture of compound 2c (0.2 mmol, 54 mg) and NaOH (0.6 mmol, 24 mg) in MeOH (2.5 mL) was taken in a Schlenk tube. To the above solution was added 30% aqueous hydrogen peroxide (150 μ L) at 0 °C, and the mixture was stirred for 20 h at same temperature. Upon completion, the reaction was quenched with aqueous $Na_2S_3O_2$ solution and then extracted with ethyl acetate (3 × 15 mL). The combined organic layer was dried over anhydrous Na₂SO₄, filtered, and evaporated under reduced pressure. The residue was purified by silica gel (100-200 mesh) column chromatography (hexane/ethyl acetate) to afford product 6 in 98% yield (56 mg) as a colorless gummy liquid: $R_f = 0.29$ (4:1 hexane/EtOAc) [silica, UV, and I₂]; ¹H NMR (400 MHz, $CDCl_3$) δ 8.03 (d, J = 8.8 Hz, 2H), 7.10 (d, J = 2.8 Hz, 1H), 6.95 (d, J = 8.4 Hz, 1H), 6.82 (d, J = 9.2 Hz, 2H), 6.72 (dd, J = 8.0, 2.8 Hz, 1H), 6.26 (bs, 1H), 3.79 (s, 3H), 3.40 (d, J = 6.0 Hz, 1H), 3.12 (d, J = 5.6 Hz, 1H), 2.24 (s, 3H); ${}^{13}C{}^{1}H$ NMR (101 MHz, $\mathrm{CDCl}_3)\,\delta$ 194.2, 163.7, 154.0, 135.9, 132.0, 131.8, 129.2, 127.6, 115.7, 114.4, 113.7, 63.7, 55.4, 53.0, 18.8; IR (neat) $\nu_{\rm max}$ 3385, 1668, 1598, 1507, 1459, 1307, 1263, 1170 cm⁻¹; HRMS (ESI) m/z [M + Na]⁺ calcd for C₁₇H₁₆NaO₄⁺ 307.0941, found 307.0959.

Gram-Scale Synthesis of α -Aryl- α , β -unsaturated Ketones **2a**/**2c**. A mixture of alkynyl-cyclohexadienone **1a**/**1c** (1.0 g; 1.0 equiv) and Cu(OTf)₂ (10 mol %) in THF (25 mL) was taken in a screw-cap sealed tube. The reaction mixture was heated to 95 °C and stirred overnight. The progress of the reaction was monitored by thin-layer chromatography (TLC) analysis. Upon completion, the reaction mixture was cooled to room temperature, and the crude residue was purified using column chromatography on silica gel (100–200 mesh) to provide the desired α -aryl- α , β -unsaturated ketones **2a** and **2c** in 90% (0.9 g) and 93% (0.93 g) yields, respectively.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.joc.1c00290.

The Journal of Organic Chemistry

Detailed spectra (¹H, ¹³C, and ¹⁹F NMR) and X-ray crystallographic data (PDF)

Accession Codes

CCDC 1947873, 2054259, and 2060957 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: + 44 1223 336033.

AUTHOR INFORMATION

Corresponding Author

Akhila K. Sahoo – School of Chemistry, University of Hyderabad, Hyderabad, Telangana 500046, India; Advanced Centre of Research in High Energy Materials (ACRHEM), University of Hyderabad, Hyderabad 500046, India;
orcid.org/0000-0001-5570-4759; Email: akhilchemistry12@gmail.com, akssc@uohyd.ac.in

Authors

- Rajendra K. Mallick School of Chemistry, University of Hyderabad, Hyderabad, Telangana 500046, India; orcid.org/0000-0002-9998-6767
- Srinivas Vangara Advanced Centre of Research in High Energy Materials (ACRHEM), University of Hyderabad, Hyderabad 500046, India
- Nagarjuna Kommu Advanced Centre of Research in High Energy Materials (ACRHEM), University of Hyderabad, Hyderabad 500046, India
- **Tirumaleswararao Guntreddi** School of Chemistry, University of Hyderabad, Hyderabad, Telangana 500046, India

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.joc.1c00290

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

This research was supported by the CEFIPRA (5505-2). We thank the University of Hyderabad (UoH-IoE; UPE-CAS and PURSE-FIST) for use of the facility. R.K.M., V.S., N.K., and T.G. thank UGC, ACRHEM, and DST for financial support.

REFERENCES

(1) (a) Jin, H.; Fürstner, A. Regioselective trans-Carboboration of Propargyl Alcohols. Org. Lett. **2019**, *21*, 3446. (b) Trost, B. M.; Rudd, M. T. A Mechanistic Dichotomy in Ruthenium-Catalyzed Propargyl Alcohol Reactivity: A Novel Hydrative Diyne Cyclization. J. Am. Chem. Soc. **2003**, *125*, 11516. (c) Luzung, M. R.; Toste, F. D. Rhenium-Catalyzed Coupling of Propargyl Alcohols and Allyl Silanes. J. Am. Chem. Soc. **2003**, *125*, 15760. (d) Pang, Y.; Liu, G.; Huang, C.; Yuan, X.-A.; Li, W.; Xie, J. A Highly Efficient Dimeric Manganese-Catalyzed Selective Hydroarylation of Internal Alkynes. Angew. Chem., Int. Ed. **2020**, *59*, 12789. (e) Liu, N.; Yao, J.; Yin, L.; Lu, T.; Tian, Z.; Dou, X. Rhodium-Catalyzed Expeditious Synthesis of Indenes from Propargyl Alcohols and Organoboronic Acids by Selective 1,4-Rhodium Migration over β -Oxygen Elimination. ACS Catal. **2019**, *9*, 6857.

(2) (a) Trost, B. M.; Tracy, J. S. Vanadium-Catalyzed Synthesis of Geometrically Defined Acyclic Tri- and Tetrasubstituted Olefins From Propargyl Alcohols. *ACS Catal.* **2019**, *9*, 1584. (b) Wu, X.; Wang, B.; Zhou, S.; Zhou, Y.; Liu, H. Ruthenium-Catalyzed Redox-Neutral [4 + 1] Annulation of Benzamides and Propargyl Alcohols via C-H Bond Activation. *ACS Catal.* **2017**, *7*, 2494. (c) Sultana, S.; Shim, J.-J.; Kim, S.

H.; Lee, Y. R. Silver(I)/base-promoted propargyl alcohol-controlled regio- or stereoselective synthesis offuran-3-carboxamides and (Z)-enaminones. Org. Biomol. Chem. **2018**, *16*, 6749. (d) Zhao, S.; Wang, X.; Wang, P.; Wang, G.; Zhao, W.; Tang, X.; Guo, M. BF₃·OEt₂-Promoted Propargyl Alcohol Rearrangement/[1,5]-Hydride Transfer/Cyclization Cascade Affording Tetrahydroquinolines. Org. Lett. **2019**, *21*, 3990. (e) Leduc, A. B.; Lebold, T. P.; Kerr, M. A. Synthesis of Tetrahydropyrans from Propargyl Alcohols and 1, 1-Cyclopropane-diesters: A One-pot Ring-Opening/Conia-Ene Proto-col. J. Org. Chem. **2009**, *74*, 8414.

(3) (a) Meyer, K. H.; Schuster, K. Ber. Dtsch. Chem. Ges. B **1922**, 55, 819. (b) Engel, D. A.; Dudley, G. B. The Meyer–Schuster rearrangement for the synthesis of $\alpha_i\beta$ -unsaturated carbonyl compounds. Org. Biomol. Chem. **2009**, 7, 4149. (c) Sugawara, Y.; Yamada, W.; Yoshida, S.; Ikeno, T.; Yamada, T. Carbon Dioxide-Mediated Catalytic Rearrangement of Propargyl Alcohols into $\alpha_i\beta$ -Unsaturated Ketones. J. Am. Chem. Soc. **2007**, 129, 12902. (d) Trost, B. M.; Chung, C. K. Vanadium-Catalyzed Addition of Propargyl Alcohols and Imines. J. Am. Chem. Soc. **2006**, 128, 10358.

(4) Roy, D.; Tharra, P.; Baire, B. Intercepted Meyer-Schuster Rearrangements in Organic Synthesis. *Asian J. Org. Chem.* 2018, 7, 1015.

(5) (a) Ye, L.; Zhang, L. Practical Synthesis of Linear α -Iodo/Bromo- $\alpha_{,\beta}$ -unsaturated Aldehydes/Ketones from Propargylic Alcohols via Au/ Mo Bimetallic Catalysis. Org. Lett. 2009, 11, 3646. (b) Zhao, M.; Mohr, J. T. Vanadium(V)-mediated rearrangement/halogenation cascade: Synthesis of α -haloenones from propargyl alcohols. *Tetrahedron* **2017**, 73, 4115. (c) Kramer, S.; Friis, S. D.; Xin, Z.; Odabachian, Y.; Skrydstrup, T. Metal-Free Halonium Mediated Acetate Shifts of Ynamides To Access α -Halo Acrylamides/Acrylimides. Org. Lett. 2011, 13, 1750. (d) Collins, B. S. L.; Suero, M. G.; Gaunt, M. J. Copper-Catalyzed Arylative Meyer-Schuster Rearrangement of Propargylic Alcohols to Complex Enones Using Diaryliodonium Salts. Angew. Chem., Int. Ed. 2013, 52, 5799. (e) Tlahuext-Aca, A.; Hopkinson, M. N.; Garza-Sanchez, R. A.; Glorius, F. Alkyne Difunctionalization by Dual Gold/Photoredox Catalysis. Chem. - Eur. J. 2016, 22, 5909. (f) Xiong, Y.-P.; Wu, M.-Y.; Zhang, X.-Y.; Ma, C.-L.; Huang, L.; Zhao, L.-J.; Tan, B.; Liu, X.-Y. Direct Access to α -Trifluoromethyl Enones via Efficient Copper-Catalyzed Trifluoromethylation of Meyer-Schuster Rearrangement. Org. Lett. 2014, 16, 1000. (g) Lin, Y.; Kong, W.; Song, Q. Palladium-Catalyzed Nitration of Meyer-Schuster Intermediates with ^tBuONO as Nitrogen Source at Ambient Temperature. Org. Lett. 2016, 18, 3702.

(6) (a) Cai, S.; Liu, Z.; Zhang, W.; Zhao, X.; Wang, Z. D. Gold-Catalyzed [3 + 2] Cycloaddition/Hydrolytic Michael Addition/Retro-Aldol Reactions of Propargylic Esters Tethered to Cyclohexadi-enones. Angew. Chem., Int. Ed. 2011, 50, 11133. (b) Anugu, R. R.; Chegondi, R. Tunable Diastereoselective Desymmetrization of Cyclohexadienones Triggered by Copper-Catalyzed Three-Component Coupling Reaction. J. Org. Chem. 2017, 82, 6786. (c) Shu, T.; Zhao, L.; Li, S.; Chen, X.-Y.; von Essen, C.; Rissanen, K.; Enders, D. Asymmetric Synthesis of Spirocyclic β -Lactams through Copper-Catalyzed Kinugasa/Michael Domino Reactions. Angew. Chem., Int. Ed. 2018, 57, 10985. (d) Lu, H.; Fan, Z.; Xiong, C.; Zhang, A. Highly Stereoselective Assembly of Polycyclic Molecules from 1,6-Enynes Triggered by Rhodium(III)-Catalyzed C-H Activation. Org. Lett. 2018, 20, 3065. (e) Tan, Y.-X.; Liu, X.-Y.; Zhao, Y.-S.; Tian, P.; Lin, G.-Q. Arylation/Intramolecular Conjugate Addition of 1,6-Enynes Enabled by Manganese(I)-Catalyzed C-H Bond Activation. Org. Lett. 2019, 21, 5.

(7) (a) Keilitz, J.; Newman, S. G.; Lautens, M. Enantioselective Rh-Catalyzed Domino Transformations of Alkynyl cyclohexadienones with Organoboron Reagents. *Org. Lett.* **2013**, *15*, 1148. (b) He, Z.-T.; Tian, B.; Fukui, Y.; Tong, X.; Tian, P.; Lin, G.-Q. Rhodium-Catalyzed Asymmetric Arylative Cyclization of meso-1,6-Dienynes Leading to Enantioenriched cis-Hydrobenzofurans. *Angew. Chem., Int. Ed.* **2013**, *52*, 5314. (c) Liu, P.; Fukui, Y.; Tian, P.; He, Z.-T.; Sun, C.-Y.; Wu, N.-Y.; Lin, G.-Q. Cu-Catalyzed Asymmetric Borylative Cyclization of Cyclohexadienone-Containing 1,6-Enynes. *J. Am. Chem. Soc.* **2013**, *135*, 11700. (d) Murthy, A. S.; Donikela, S.; Reddy, C. S.; Chegondi, R.

The Journal of Organic Chemistry

Palladium-Catalyzed Regioselective Domino Cyclization of Cyclohexadienones. J. Org. Chem. 2015, 80, 5566.

(8) (a) Tello-Aburto, R.; Harned, A. M. Palladium-Catalyzed Reactions of Cyclohexadienones: Regioselective Cyclizations Triggered by Alkyne Acetoxylation. Org. Lett. 2009, 11, 3998. (b) Takenaka, K.; Mohanta, S. C.; Sasai, H. Palladium Enolate Umpolung: Cyclative Diacetoxylation of Alkynyl Cyclohexadienones Promoted by a Pd/ SPRIX Catalyst. Angew. Chem., Int. Ed. 2014, 53, 4675. (c) Gollapelli, K. K.; Donikela, S.; Manjula, N.; Chegondi, R. Rhodium-Catalyzed Highly Regio- and Enantioselective Reductive Cyclization of Alkyne-Tethered Cyclohexadienones. ACS Catal. 2018, 8, 1440. (d) He, C.-Y.; Xie, L. B.; Ding, R.; Tian, P.; Lin, G.-Q. Copper-catalyzed asymmetric silylative cyclization of cyclohexadienone-containing 1,6-enynes. Tetrahedron 2019, 75, 1682.

(9) Mallick, R. K.; Dutta, S.; Vanjari, R.; Voituriez, A.; Sahoo, A. K. Thioarylative Radical Cyclization of Yne-Dienone. *J. Org. Chem.* **2019**, *84*, 10509.

(10) Basavaiah, D.; Naganaboina, R. T. The Baylis—Hillman reaction: a new continent in organic chemistry — our philosophy, vision and over three decades of research. *New J. Chem.* **2018**, *42*, 14036.

(11) (a) Munakala, A.; Chegondi, R. Silver(I)-Catalyzed Enyne Cyclization/Aromatization of Alkyne-Tethered Cyclohexadienones to Access Meta-Substituted Phenols. *Org. Lett.* **2021**, *23*, 317. (b) Nair, A. M.; Halder, I.; Sharma, R.; Volla, C. M. R. Water Mediated Rearrangement of Alkynyl Cyclohexadienones: Access to meta-Alkenylated Phenols. *Org. Lett.* **2021**, *23*, 1840.

(12) Suzuki, T.; Ohwada, T.; Shudo, K. Superacid-Catalyzed Electrocyclization of 1-Phenyl-2-propen-1-ones to 1-Indanones.Kinetic and Theoretical Studies of Electrocyclization of Oxonium-Carbenium Dications. J. Am. Chem. Soc. **1997**, *119*, 6774.

(13) Ruano, J. L. G.; Fernández-Ibáñez, M. Á.; Fernández-Salas, J. A.; Maestro, M. C.; Márquez-López, P.; Rodríguez-Fernández, M. M. Remote Stereo-control Mediated by a Sulfinyl Group: Synthesis of Allylic Alcoholsvia Chemoselective and Diastereoselective Reduction of γ -Methylene δ -Ketosulfoxides. J. Org. Chem. **2009**, 74, 1200.

(14) Li, K.; Jin, Z.; Chan, W.-L.; Lu, Y. Enantioselective Construction of Bicyclic Pyran and Hydrindane Scaffolds via Intramolecular Rauhut-Currier Reactions Catalyzed by Thiourea-Phosphines. *ACS Catal.* **2018**, *8*, 8810.

(15) Zhao, Z.; Bagdi, P. R.; Yang, S.; Liu, J.; Xu, W.; Fang, X. Stereodivergent Access to Enantioenriched Epoxy Alcohols with Three Stereogenic Centers via Ruthenium-Catalyzed Transfer Hydrogenation. *Org. Lett.* **2019**, *21*, 5491.

(16) (a) Wolff, S. K.; Grimwood, D. J.; McKinnon, J. J.; Turner, M. J.; Jayatilaka, D.; Spackman, M. A. CrystalExplorer, version 3.1; University of Western Australia: Crawley, Australia, 2012. (b) Spackman, M. A.; Jayatilaka, D. Hirshfeld surface analysis. *CrystEngComm* **2009**, *11*, 19– 32. (c) Spackman, M. A.; McKinnon, J. J. Fingerprinting intermolecular interactions in molecular crystals. *CrystEngComm* **2002**, *4*, 378–392.

(17) (a) Krasik, P.; Bohemier-Bernard, M.; Yu, Q. Cu(OTf)₂-Catalyzed Selective Opening of Aryl and Vinyl Epoxides with Carbonyl Compounds to Give 1,3-Dioxolanes. *Synlett* **2005**, *5*, 0854. (b) Tao, L.; Shi, M. (CH₃)₂CuLi/Cu(OTf)₂ Mediated N-or O-Cyclization of Urea-Tethered Cyclobuta[b]indolines. *Org. Lett.* **2019**, *21*, 129. (c) Lee, J.; Wang, S.; Callahan, M.; Nagorny, P. Copper(II)-Catalyzed Tandem Decarboxylative Michael/Aldol Reactions Leading to the Formation of Functionalized Cyclohexenones. *Org. Lett.* **2018**, *20*, 2067. (d) Alavala, G. K. R.; Sajjad, F.; Shi, T.; Kang, Z.; Ma, M.; Xing, D.; Hu, W. Diastereoselective synthesis of isochromans via the Cu(II)-catalysed intramolecular Michael-type trapping of oxonium ylides. *Chem. Commun.* **2018**, *54*, 12650.

(18) (a) Hexum, J.; Tello-Aburto, K. R.; Struntz, N. B.; Harned, A. M.; Harki, D. A. Bicyclic Cyclohexenones as Inhibitors of NF-κB Signaling. *ACS Med. Chem. Lett.* **2012**, *3*, 459. (b) Fukui, Y.; Liu, P.; Liu, Q.; He, Z.-T.; Wu, N.-Y.; Tian, P.; Lin, G.-Q. Tunable Arylative Cyclization of 1,6-Enynes Triggered by Rhodium(III)-Catalyzed C–H Activation. *J. Am. Chem. Soc.* **2014**, *136*, 15607. (c) Zhou, X.; Pan, Y.; Li, X. Catalyst-Controlled Regiodivergent Alkyne Insertion in the Context of C–H Activation and Diels–Alder Reactions: Synthesis of Fused and Bridged Cycles. *Angew. Chem., Int. Ed.* **2017**, *56*, 8163.

(19) Sonogashira, K.; Tohda, Y.; Hagihara, N. A convenient synthesis of acetylenes: catalytic substitutions of acetylenic hydrogen with bromoalkenes, iodoarenes and bromopyridines. *Tetrahedron Lett.* **1975**, *16*, 4467.