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Discovery of a novel series of thienopyrimidine as highly potent and selective PI3K inhibitors

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KEYWORDS: thienopyrimidines, phosphoinositide 3-kinase, mammalian target of rapamycin, dual inhibitors, selective PI3K inhibitors

ABSTRACT: Inhibition of the phosphoinositide 3-kinase (PI3K)/AKT/mammalian target of rapamycin (mTOR) signaling pathway provides a promising new approach for cancer therapy. Through a rational design, a novel series of thienopyrimidine was discovered as highly potent and selective PI3K inhibitors. These thienopyrimidine derivatives were demonstrated to bear nanomolar PI3K α inhibitory potency with over 100-fold selectivity against mTOR kinase. The lead compounds **6g** and **6k** showed good developability profiles in cell-based proliferation and ADME assays. In this communication, their design, synthesis, structure-activity relationship, selectivity, and some developability properties are described.

The phosphoinositide 3-kinase (PI3K)/AKT/the mammalian target of rapamycin (mTOR) signal transduction pathway is involved in many essential cellular functions including cell growth, proliferation, differentiation, motility, survival, and intracellular trafficking.^{1, 2} The PI₃Ks constitute a lipid kinase family and are divided into three different classes (class I, II & III) according to their sequence, homology and substrate preferences.² Of them, class I PI₃Ks are the most widely understood and are subdivided into class IA (PI₃K α , β and δ) isoforms and class IB (PI3Ky) isoforms.² Activated by receptor tyrosine kinases and Tas and Rho family GTPases, the PI3Ks phosphorylate the 3-hydroxyl position of PIP2 (phosphatidylinositol 4,5-diphosphate) to generate PIP3 (phosphatidylinositol 3,4,5-triphosphate), a potent secondary messenger that results in the activation of several downstream effectors, including the serine-threonine kinase, AKT (also known as protein kinase B or PKB).² Subsequently, activated AKT triggers a signal transduction cascade that ultimately stimulates mTOR. mTOR containing two distinct complexes (mTORC1 and mTORC2) plays an integral key role in regulating PI3K/AKT activation and signaling through positive and negative feedback loops.2, 3

Abundant evidence from genomic analysis has revealed that many components of the PI₃K/AKT/mTOR pathway were frequently mutated or altered in numerous forms of human cancers.^{2, 4} For example, the PI₃K catalytic subunit α -isoform gene encoding the p110 α (PIK₃CA) is one of the most highly mutated oncogenes and its high mutational frequencies have been demonstrated in breast, colorectal, liver and other cancers. In addition, mTOR was the first node of this pathway to be targeted in the clinic and the association between mTOR and cancers has been validated by several approved drugs such as Afinitor[™] and Torisel[™]. Hence, targeting this pathway for cancer therapeutics has been intensively pursued by many industrial and academic research groups. Noteworthy, the kinase domain of mTOR and the puoa catalytic subunit of the class I PI3Ks share a high degree of similarity/sequence homology,^{5,6} prompting to generate PI₃K/mTOR dual inhibitors which target two nodal points in the pathway concurrently to offer an augment on better efficacy and less likelihood to induce drug resistance, etc.^{2, 7} Parallel with PI3K/mTOR dual inhibitors, the selective PI3K inhibitors to avoid extra toxicity carried by mTOR kinase inhibitory activity for achieving a balanced efficacy and safety profile has also received great attention.⁸ Both PI₃K/mTOR dual inhibitors (e.g., GSK2126458 and PKI-587, Figure 1) and selective PI3K inhibitors (e.g., GDC-0941 and BKM120, Figure 1) have been advanced into clinical evaluation.9-12



Figure 1. PI₃K/mTOR dual inhibitors (1 and 2) and the class I selective PI3K inhibitors (3 and 4).

Among PI₃K/mTOR dual inhibitors reported to date, GSK2126458, as shown in Figure 1, is one of most potent structures with a Ki of 0.019 nM, 0.18 nM and 0.30 nM for PI₃Kα, mTORC₁ and mTORC₂ respectively.⁹ Its high in vitro potencies were well translated into in vivo efficacies across a broad spectrum of antitumor models, paving the way for its clinical development.13 Its chemical structure features quinoline, sulfonamide, and methoxyl pyridine moieties, which forms key hydrogen bondings with Val882, Lys833 and an active water molecule respectively as determined from a cocrystal structure with PI₃Ky.⁹ Apparently, guinoline core is a cornerstone of the structure, for not only forming its own critical hydrogen bonding with the target enzyme but projecting other moieties appropriately to access additional interactions. Although quite a few PI3K/mTOR inhibitors were derived from this chemical series, structure-activity relationship (SAR) focused on the quinoline region has barely been reported. Possibilities of further optimizing the potency as well as other drug-like properties by replacement of quinoline still remain unclear. Herein, we describe our research progress that led to generate a structurally novel thienopyrimidine series as potent PI3K inhibitors with good selectivity against mTOR.

Table 1. PI₃Kα and mTOR potency data for compounds 5a-h¹⁴

Ň	_N	OMe	
		୍ଦୁ୦	~
Bicyclic	~	`N´``	
ring	5a-h	F	√∕_F

IC₅₀

(nM)

0.10

34.7

1.02

0.095

(nM)

0.75

66.7

3.04

0.20

5e	N N N N N N N N N N N N N N N N N N N	51.7	62.3	1.2
5f	N _{NN} St	0.27	21.6	80
5g	H _I N N N	35.6	18.4	0.52
5h	N H2N N	0.07	0.29	4.1

During the course of our SAR investigation, compound 5a with a quinoline core was firstly synthesized and evaluated as a close analogue to GSK2126458 for a reference standard (Table 1). Next, the aromatic CH at 8-position of quinoline was replaced by a nitrogen and the resulting naphthyridine 5b displayed a significant drop of potency in both PI3Ka and mTOR assays relative to **5a** (~350-fold decrease for PI3Kα and ~90-fold decrease for mTOR). Apparently, this position is not well tolerated with nitrogen and the corresponding loss of potency could be explained by unfavorable interaction between nitrogen and the carbonyl of Glu880 in the target protein as observed from a Amgen chemical series.¹⁵ When nitrogen was moved from 8- to 7-position, 1,7-naphthyridine 5c gained back some degree of potency, but was still around 10and 4-fold less potent against PI3Ka and mTOR enzymes than 5a, respectively. With SAR expansion, quinazoline 5d, pyridopyrimidine 5e and cinnoline 5f and were also investigated. Noteworthy, N-3 nitrogen (5d and 5e) retains the PI₃Kα and mTOR potency (5d vs 5a and 5e vs 5b) while N-2 nitrogen (cinnoline **5f**) exhibited a slight decrease of potency for PI₃K α (**5f** vs **5a**: < 3-fold) and a pronounced decrease for mTOR (5f vs 5a: > 25-fold). Through scanning the above 2-, 3-, 7-, and 8-position of quinoline, it was clearly found that only 3-position was tolerated for nitrogen replacement without loss of any PI₃Kα and mTOR inhibitory activities. When 2-amino group was incorporated, pyridopyrimidine 5g showed slightly enhanced PI₃K α inhibitory activities (5g vs 5e) and a similar trend was also observed for quinazoline 5h (**5h** vs **5d**). Of particular note, selectivity of PI₃K α against mTOR was all less than 10-fold for all compounds listed in Table 1 except for compound **5f** (~80 fold). Synthesis of compounds 5a-h was performed according to the revised literature procedures.¹⁶⁻²¹

Table 2. PI3K and mTOR potency data for compounds **6a-m**¹⁴



	Compd	R	Ar	PI3Kα IC ₅₀ (nM)	mTOR IC ₅₀ (nM)	Selectivi- ty (mTOR/ PI ₃ Kα)
	6a	Н	32 N OMe	1228	ND^{a}	/
-	6b	Н	N OMe	3494	ND^{a}	/

ring

Compd

5a

5b

5C

5d

Selectivity

(mTOR/PI3Kα)

7.5

1.9

3.0

2.1

6с	Н	N OMe	>2500	ND^{a}	/
6d	Н	N OMe N OSCH3	43.1	ND^{a}	/
бе	Me	F F	72.3	ND^{a}	/
6f	Н	N OMe O O H CF3	17.0	ND^{a}	/
6g	Н	N OMe O O F F	2.07	218	105
6h	Н	N OMe N S H CN	3.60	363	100
6i	Н	N OMe N S S S-CI	1.30	67.9	52
6j	Н	N Me O O H F	1.70	345	203
6k	Н	N O O N F F	0.23	53.6	233
61	Н	N CI O SO H	0.30	43.7	146
6m	Н	N CI S S S CI	0.15	25.8	172

^{*a*} ND = not determined.

In addition to the approach of inserting nitrogen into quinoline ring as illustrated above, the bicyclic thienopyrimidine core was further explored for a new structural scaffold (Table 2). The simple methoxyl pyridine **6a** only displayed singledigit micromolar enzymatic PI₃K α potency with an IC₅₀ of 1.22 μ M. The potency of pyrimidine **6b** fell into the same order of magnitude with that of the pyridine (1.22 μ M vs 3.49 μM). The small electron-withdrawing fluoride group (e.g., 6c) on the pyridine did not offer the potency benefit. Similar to several other published chemical series, incorporation of sulfonamide functionality to the 3-position of the methoxyl pyridine on this thienopyrimidine scaffold significantly boosted the potency (e.g. 6d-6i). This great improvement could be explained by the strong interaction picked up by the sulfonamide with Lys833 within the affinity pocket, more specifically, the charge interaction derived from the deprotonated sulfonamide nitrogen at physiological conditions.9, 22 In particular, compound 6g exhibited an IC₅₀ of 2.07 nM in PI3Ka assay (Table 2), which is around 20-fold less potent compared to 5d. It is likely that the sufur in the thienopyrimidine core causes a less favorable interaction with the carbonyl of Glu880 in a similar situation as described earlier for the nitrogen at the 8-position of the pyridopyrimidine core in **5e**.¹⁵ The binding mode between **6g** and PI₃Ky as a surrogate protein was proposed in Figure 2, illustrating the possible lone pair repulsion between the sulfur in the thienopyrimidine core and carbonyl oxygen of Glu880. However, relative to the pyridopyrimidine **5e**, the bicyclic thienopyrimidine confers the PI₃Kα potency enhancement (6g vs 5e: 2.07

vs 51.7 nM). Modification on the thienopyrimidine ring by attaching a methyl group (**6e**) resulted in a significant potency drop. With no apparent advantages gained from this variation, further SAR investigation retained the parent thienopyrimidine core to focus on the methoxyl pyridine ring. When methoxyl group was replaced by other simple substituents (Me or Cl), it was found that in comparison to **6g**, methyl pyridine **6j** exhibited similar PI₃K α potency (1.70 nM vs 2.07 nM), while chloride pyridine **6k** generated around 10-fold PI₃K α potency enhancement (0.23 nM vs 2.07 nM). This significant potency enhancement was also demonstrated by two more chlorine substituted pyridines **6l** and **6m**. Notably, with sub-nanomolar IC₅₀s achieved, PI₃K α potency level from thienopyrimidine series has been optimized up to the similarly high level from quinoline series.



Figure 2. Predicted binding mode for **6g** (shown in stick representation with carbon atoms colored magenta) with PI₃K γ (PDB ID: 3Lo8). Hydrogen bonding interactions are shown in yellow dashed lines to the hinge region (Val882), Lys8₃₃ and the conserved water molecule in the selectivity pocket. The interaction between the sulfur in the thienopyrimidine core of **6g** and carbonyl of Glu88o is demonstrated in magenta dashed line. Images generated using PyMol.

The general synthetic approach for above thienopyrimidines was described in Scheme 1: cyclization of compound 7 with formamidine acetate afforded the bicyclic thienopyrimidine core **8**, which was then brominated with liquid bromine to give compound **9**. Compound **9** was then treated with phosphorus oxychloride and sodium iodide subsequently to afford the key intermediate 6-bromo-4-iodiothienopyrimidine **11**, which was further coupled with pyridine-4-boronic acid and then various aryl borate or boronic acid to give compounds **6a-m**.

Scheme 1. Synthesis of compounds 6a-m.^a



^{*a*} Reagents and conditions. (a) formamidine acetate, NMP, 135°C, Ar; (b) Br₂, HOAc, rt; (c) POCl₃, reflux; (d) NaI, anhydrous dioxane, 105°C, Ar; (e) pyridine-4-boronic acid, 2N K_2CO_3 , Pd(PPh₃)₂Cl₂, dioxane, 100°C; (f) aryl borate or aryl boronic acid, PdCl₂(dppf), 2N K_2CO_3 , dioxane, 100°C.

Of particular interest, these thienopyrimidines showed good PI₃K α /mTOR selectivity and the selectivity was, in general, determined to be around 100-fold. This is a much improved selectivity profile relative to six-member fused bicyclic rings such as quinoline, naphthyridine and quinazoline illustrated in Table 1. For the selectivity, we hypothesize that the sulfur in thienopyrimidine was closer to Tyr2225 in mTOR than the corresponding Tyr residue in PI₃K α , rendering a stronger repulsion for mTOR kinase.

Table 3. Class I PI3K assay data for 6g and 6k

Compd	IC ₅₀ (nM)			
	ΡΙ3Κα	ΡΙ3Κβ	ΡΙ3Κδ	PI3Ky
6g	2.07	22.5	13.3	21.1
6k	0.23	3.6	3.1	10.2

Through above SAR investigation, novel thienopyrimidine derivatives were identified to bear decent PI3Ka potency with good mTOR selectivity. Two representative compounds 6g and 6k were selected for further profiling. Among the class I PI3K isoforms, 6g and 6k were most potent against PI3Ka and had single to low double digit nanomolar potencies against PI₃Kβ, PI₃Kδ and PI₃Kγ (Table 3). Both PAMPA and Caco-2 assays were utilized to test their property of permeability. These two in vitro ADME assay results were well correlated and showed moderate permeability for 6g and 6k (Table 4). For cellular activities, AKT (Ser 473) phosphorylation assay was conducted with T47D cells in which 6g exhibited significant inhibitory activity on the downstream AKT phosphorylation with an IC₅₀ of 84.6 nM (Table 5). Compounds 6g and 6k were also tested in different cancer cell lines for its sensitivity in proliferation cell assays. All cancer cell lines chosen for the experiment either harbor PI3KCA mutations or bear the PTEN deletion except for A549. Compounds 6g and 6k were moderately potent with at least single digit micromolar IC50s across these selected cell lines (Table 5). It was worth to note that compound 6g was more sensitive to non-small cell lung cancer cell line H460 and breast cancer line T47D with IC50s of 0.42 µM and 0.66 µM respectively, while it performed the least inhibitory effect on human brain glioma cell line U87MG (IC₅₀ = 4.4μ M). The

overall profiles of **6g** and **6k** obtained from in vitro assays support their next phase of in vivo evaluation.

Table 4. Permeability assay data for 6g and 6k

Compd	PAMPA (10 ⁻⁶ cm/s)	Caco-2 (10^{-6} cm/s)			
		Papp (A to B)	Papp (B to A)	Efflux ratio	
6g	8.0	5.6	4.5	0.8	
6k	7.0	4.8	6.9	1.4	

Table 5. Cellular assay data for 6g and 6k

Assay	Cell line	6g	6k
pAKT-S ₄₇₃ IC ₅₀ (nM)	T ₄₇ D	84.6	ND^{a}
	$U8_7MG^b$	4.40	4.43
	$T_{47}D^{c}$	0.66	1.52
Cell Growth	SKOV ₃ ^c	3.74	5.20
IC ₅₀ (μM)	H1975 [°]	1.07	2,20
	H460 ^c	0.42	2.94
	A_{549}^d	3.18	8.44

^{*a*} ND = not determined. ^{*b*} Tumor cell line with PTEN deletion. ^{*c*} Tumor cell line harboring PI₃KCA mutation. ^{*d*} Tumor cell line harboring neither PI₃KCA mutation nor PTEN deletion.

In summary, a novel series of thienopyrimidine was discovered by rational design as highly potent PI₃K inhibitors. The most potent compounds identified from this series demonstrated sub-nanomolar PI₃K α potency, comparable to those from the known quinoline scaffold. Differing from the quinoline scaffold, thienopyrimidine derivatives confer good PI₃K α /mTOR selectivity (>100-fold). The lead compounds **6g** and **6k** showed acceptable developability profiles in cell-based proliferation and ADME assays. Further optimization of this series as well as in vivo phase of testing including PK properties and in vivo efficacies will be reported in due course.

SUPPORTING INFORMATION AVAILABLE

Biological assays and experimental procedures. This material is available free of charge via the Internet at http://pubs.acs.org.

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ABBREVIATIONS

PI₃K, phosphoinositide 3-kinase; mTOR, mammalian target of rapamycin; PTEN, phosphotase and tensin homologue; PAMPA, parallel artificial membrane permeability assay.

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 $\begin{array}{l} \text{IC}_{50} \;(\text{PI3K}\alpha) = 2.07 \; \text{nM} \\ \text{IC}_{50} \;(\text{mTOR}) = \; 218 \; \text{nM} \\ \text{Selectivity} \;(\text{mTOR/PI3K}\alpha) = \; 105\text{-fold} \\ \text{T47D} \;\; \text{IC}_{50} = \; 0.66 \; \mu\text{M} \\ \text{H460} \;\; \text{IC}_{50} = \; 0.42 \; \mu\text{M} \end{array}$