

Enantioselective Iridium-Catalyzed Allylation of Nitroalkanes: Entry to β -Stereogenic α -Quaternary Primary Amines

Woo-Ok Jung, Binh Khanh Mai, Brian J. Spinello, Zachary J. Dubey, Seung Wook Kim, Craig E. Stivala,* Jason R. Zbieg,* Peng Liu,* and Michael J. Krische*

Cite This: *J. Am. Chem. Soc.* 2021, 143, 9343–9349

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ABSTRACT: The first systematic study of simple nitronate nucleophiles in iridium-catalyzed allylic alkylation is described. Using a tol-BINAP-modified π -allyliridium C,O-benzoate catalyst, α,α -disubstituted nitronates substitute racemic branched alkyl-substituted allylic acetates, thus providing entry to β -stereogenic α -quaternary primary amines. DFT calculations reveal early transition states that render the reaction less sensitive to steric effects and distinct trans-effects of diastereomeric chiral-at-iridium π -allyl complexes that facilitate formation of congested tertiary–quaternary C–C bonds.

Chiral amines are prevalent among FDA-approved drugs,¹ as are protocols for their synthesis.² The vast majority of catalytic enantioselective methods for the construction of chiral amines deliver *N*-substituted carbon stereocenters.² Catalytic enantioselective methods applicable to the formation of acyclic chiral β -stereogenic amines are far less common and encompass asymmetric hydrogenation of enamines,³ allylic amines/amides⁴ or nitroolefins,⁵ 1,4-reductions and 1,4-additions to nitroolefins,^{6,7} and dynamic kinetic resolutions via aldehyde reductive amination.⁸ These methods do not deliver chiral β -stereogenic α -quaternary primary amines. Catalytic enantioselective Tsuji–Trost nitronate allylic alkylation–reduction potentially provides access to acyclic chiral β -stereogenic amines, but such methods are underdeveloped.^{9–12} Palladium-catalyzed allylic alkylations typically display linear regioselectivity and consequently focus on reactions that proceed via symmetric π -allylpalladium intermediates⁹ or prochiral nucleophiles (α -nitroesters).^{10a} Beyond vinyl epoxides,^{10a} only a single system for catalytic asymmetric allylic alkylation of unactivated nitronates with monosubstituted π -allyl precursors that displays branched regioselectivity has been described.^{10b,c} Only two examples of corresponding iridium-catalyzed nitronate allylic alkylations are reported.¹¹ Both the palladium- and iridium-based catalyst systems rely on linear aryl-substituted π -allyl precursors and display incomplete diastereo- and regioselectivity (Figure 1).

In connection with studies of π -allyliridium C,O-benzoate-catalyzed nucleophilic allylations of carbonyl compounds,¹³ we recently found these same complexes are competent catalysts for electrophilic allylation, as demonstrated in regio- and enantioselective aminations of racemic branched alkyl-substituted allylic acetates to form acyclic chiral α -stereogenic amines.¹⁴ Aspiring to access corresponding chiral β -stereogenic amines, a study of nitronate nucleophiles was undertaken. Here, we demonstrate that α,α -disubstituted nitroalkanes participate in highly enantioselective allylic alkylation with racemic branched alkyl-substituted allylic acetates. Despite forming congested contiguous quaternary–tertiary C–C

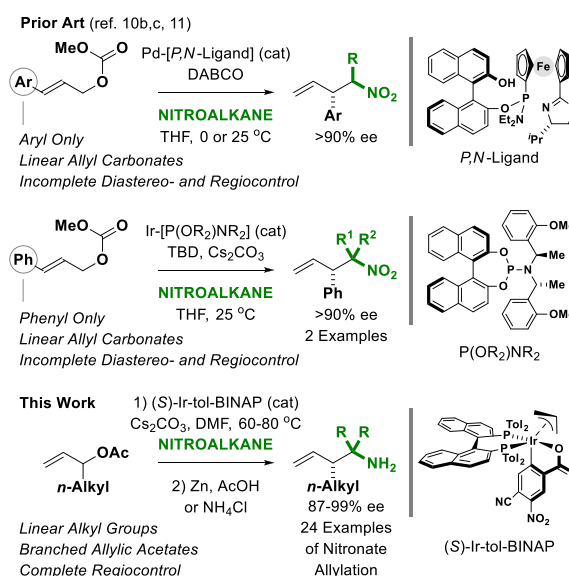


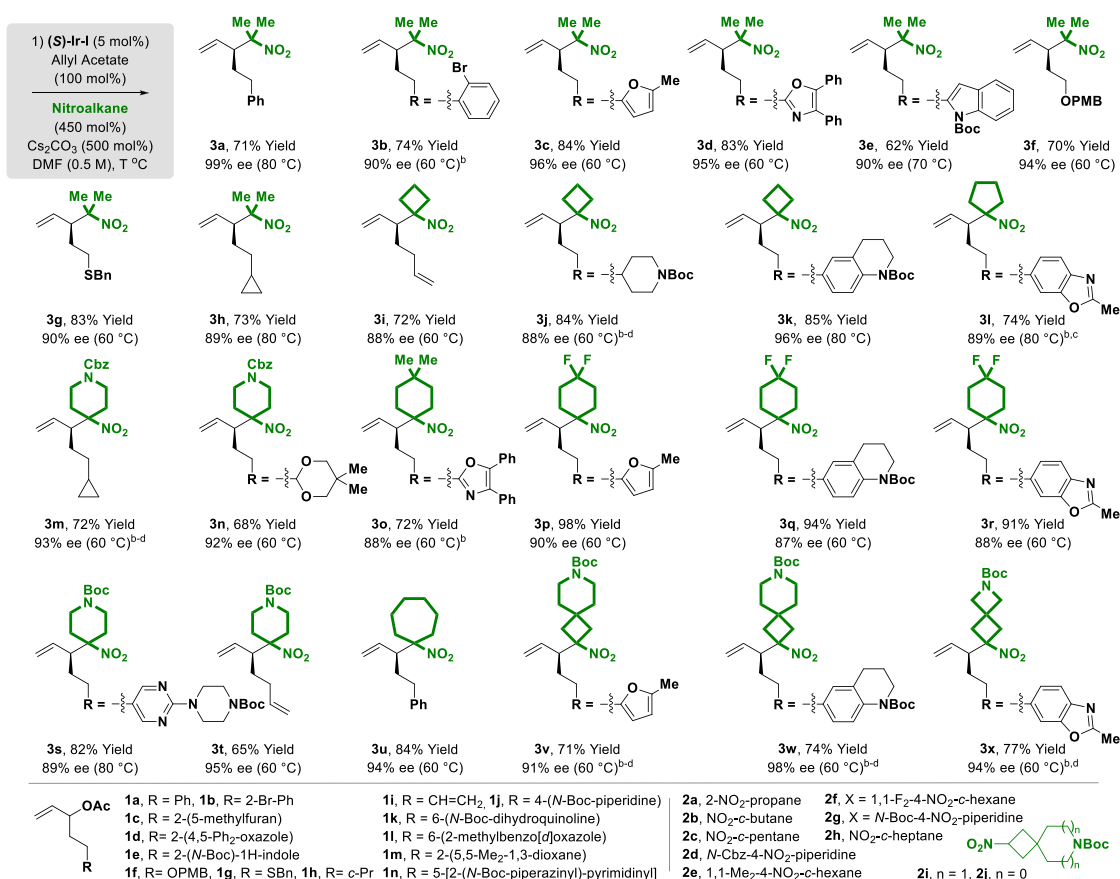
Figure 1. Unactivated nitronates in catalytic enantioselective allylic alkylations of monosubstituted π -allyl precursors.

bonds, complete branched regioselectivities are observed. DFT calculations reveal early transition states that render the reaction less sensitive to steric effects and distinct trans-effects of diastereomeric chiral-at-iridium π -allyl complexes that facilitate formation of congested tertiary–quaternary C–C bonds.

Received: May 20, 2021

Published: June 21, 2021



Scheme 1. Iridium-Catalyzed Allylation of Nitroalkanes 2a–2j Using Allylic Acetates 1a–1n to Form Homoallylic Nitroalkanes 3a–3x^a

^aYields are of material isolated by silica gel chromatography. Enantioselectivities were determined by chiral stationary phase HPLC analysis. ^bNMP was used as the reaction solvent. ^c(S)-Ir-tol-BINAP, X = OMe. ^dIr-catalyst (10 mol%). See the Supporting Information for experimental details.

In initial experiments, allylic acetate **1a** (100 mol%) was exposed to 2-nitropropane **2a** (1125 mol%) as solvent (1.0 M) in the presence of Cs₂CO₃ (500 mol%) and the π -allyliridium C,O-benzoate complex modified by (S)-tol-BINAP, Ir-I, at 100 °C. The homoallylic nitroalkane **3a** was formed in 51% yield and 95% ee (Table 1, entry 1). Lower loading of 2-nitropropane **2a** (450 mol%) with DMF (0.5 M) as solvent improved yield and enantioselectivity (Table 1, entry 2), but further reduction in the loading of **2a** led to a small decrease in yield (Table 1, entry 3). A slight decrease in temperature (80 °C) improved the yield of **3a** without diminishing enantioselectivity (Table 1, entry 4). Lower loading of Cs₂CO₃ decreased the yield of **3a** (Table 1, entry 5), and water (100 mol%) dramatically decreased conversion of **1a** (Table 1, entry 6). Other π -allyliridium C,O-benzoates were evaluated but did not improve the yield of **3a** (Table 1, entries 7–10).

To evaluate reaction scope, optimal conditions for the formation of **3a** (Table 1, entry 4) were applied to the allylic alkylation of α,α -disubstituted nitroalkanes **2a–2j** with racemic branched alkyl-substituted allylic acetates **1a–1n** (Scheme 1). As illustrated by the conversion of 2-nitropropane **2a** to adducts **3a–3h**, diverse (hetero)aromatic groups (**3a–3e**), (thio)ethers (**3f** and **3g**), and cyclopropyl groups (**3h**) are tolerated. As demonstrated by the formation of adducts **3i–3x**, cyclic nitroalkanes **2b–2h** are also competent partners for allylic alkylation. This includes nitrocyclobutane (**2b**), nitrocyclopentane (**2c**), N-Cbz- and N-Boc-4-nitropiperidines (**2d**

Table 1. Selected Optimization Experiments in the Enantioselective Ir-Catalyzed Allylic Alkylation of Racemic Branched Alkyl-Substituted Allylic Acetate **1a** with Nitronate **2a**^a

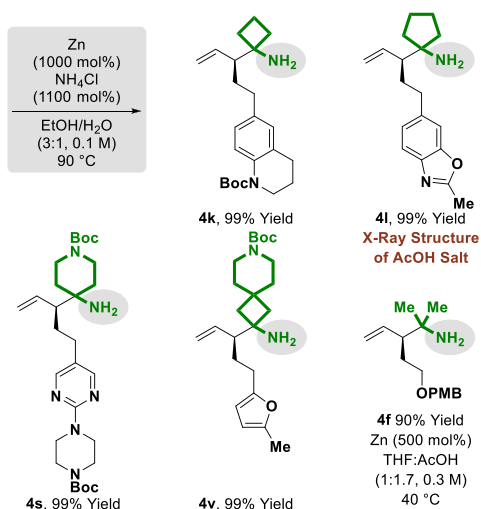
Entry	2a (mol%)	catalyst	Cs ₂ CO ₃ (mol%)	T °C	Yield (%)	ee (%)
1 ^b	1125	Ir-I	500 mol%	100	51	95
2	450	Ir-I	500 mol%	100	67	99
3	240	Ir-I	500 mol%	100	35	99
4	450	Ir-I	500 mol%	80	71	99
5	450	Ir-I	300 mol%	80	56	99
6 ^c	450	Ir-I	500 mol%	80	<10	-
7	450	Ir-II	500 mol%	80	51	99
8	450	Ir-III	500 mol%	80	53	99
9	450	Ir-IV	500 mol%	80	43	99
10	450	Ir-V	500 mol%	80	50	99

Ir-I L = (S)-tol-BINAP, X = CN
 Ir-II L = (S)-tol-BINAP, X = NO₂
 Ir-III L = (S)-BINAP, X = CN
 Ir-IV L = (S)-SEGPHOS, X = CN
 Ir-V L = (S)-MeO-BIPHEP, X = CN

^aYields are of material isolated by silica gel chromatography. Enantioselectivities were determined by HPLC analysis. ^b**2a** = solvent (1.0 M). ^cH₂O (100 mol%). See the Supporting Information for experimental details.

and **2g**, respectively), as well as 1,1-disubstituted 4-nitrocyclohexanes (**2e** and **2f**), nitrocycloheptane (**2h**), and the spirocyclic nitroalkanes (**2i** and **2j**). In all cases, good to excellent yields of adducts **3a–3x** were obtained with high levels of enantiomeric enrichment (87–99% ee) and complete branched regioselectivity. The nonsymmetric nitroalkane (1-nitroethyl)benzene provided high yields of allylic alkylation product, but low diastereomeric ratios were observed. Low conversions were observed using allylic acetates bearing α - or β -branched alkyl moieties. As revealed by the conversion of **3f**, **3k**, **3l**, **3s**, and **3v** to amines **4f**, **4k**, **4l**, **4s** and **4v**, respectively, zinc-mediated reduction of the allylation products provides entry to chiral β -stereogenic α -quaternary primary amines (Scheme 2).

Scheme 2. Zinc-Mediated Reduction of **3f**, **3k**, **3l**, **3s**, and **3v** to Form β -Stereogenic α -Quaternary Primary Amines **4f**, **4k**, **4l**, **4s**, and **4v**^a



^aYields are of material isolated after filtration through Celite. See the Supporting Information for experimental details.

Density functional theory (DFT) calculations were carried out to investigate the origin of regio- and enantioselectivity of the Ir-catalyzed allylic alkylation.^{15,16} As numerous diastereomeric π -allyliridium C,O-benzoate complexes may exist in equilibrium prior to the allylation, the relative stabilities, reactivities, and regioselectivities of all 16 possible stereoisomers of the π -crotyliridium(III) complex were computed (Figure 2). These include diastereomers A, B, C, and D, where the C,O-benzoate resides in different coordination sites of the octahedral Ir complex,¹⁷ with four different coordination modes of the π -crotyl ligand for each diastereomer. Here, the suffix *exo* or *endo* describes whether the allyl C2–H points away or toward the C,O-benzoate ligand, and *distal* or *proximal* describes if the C3–methyl points away or toward the C,O-benzoate (see Figure S2 for the relative stabilities of all 16 isomers). Computed activation barriers for the addition of $^-CMe_2NO_2$ anion to each π -crotyliridium(III) isomer leading to branched and linear products are shown in Figure 2B. The 16 π -crotyliridium(III) isomers exhibited very different reactivity and regioselectivity. The most reactive isomer, **D_{exo proximal}**, strongly favors the branched product ($\Delta\Delta G^\ddagger(B-L) = -8.5$ kcal/mol), while some less reactive isomers, such as **A_{endo proximal}** and **B_{endo proximal}**,

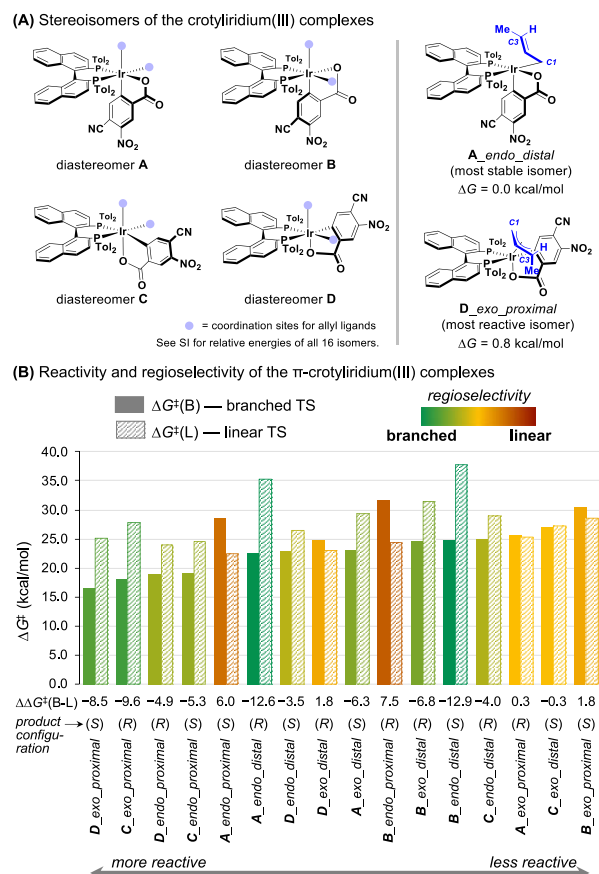


Figure 2. Computed activation free energies of the addition of $^-CMe_2NO_2$ to different π -crotyliridium(III) isomers. All activation barriers ($\Delta G^\ddagger$) are with respect to **A_{endo distal}**.

prefer the linear product. These results suggest stereogenicity at iridium affects not only reactivity but also regioselectivity. To assess the origin of these effects, computational analyses on the electronic and structural properties of the diastereomeric π -crotyliridium(III) complexes were performed.

The computed transition-state structures of the outer sphere addition of the nitronate nucleophile all involve long forming C–C distances (usually >2.4 Å in branch-selective TSs and >2.2 Å in linear-selective TSs (see Figure 3 and Figure S4 in the Supporting Information). These early transition states suggest that the addition is not sensitive to steric effects, and thus, the regioselectivity is mainly controlled by the electronic properties of the π -allyl complexes. The computed ground-state properties of the π -crotyliridium(III) isomers and the distortion/interaction model analysis of the allylation transition states revealed enormously different electronic properties and their influences on the regioselectivity. In branch-selective isomers, such as **D_{exo proximal}** (Figure 3A), the Ir–C3 distance is much longer than Ir–C1. This geometry leads to weaker $d \rightarrow \pi^*$ backbonding at C3 and more positive charge on C3 compared to C1 (Figure 3A),¹⁸ making the more substituted C3 terminus more electrophilic.¹⁹ In addition, the transition state of nitronate addition to C3 (TS-1) is promoted by the smaller distortion energy of the π -crotyliridium complex ($\Delta E_{\text{dist-Ir}}$) as the ground-state Ir–C3 bond of **D_{exo proximal}** is pre-distorted to a longer distance (2.50 Å) that is closer to that in TS-1 (2.87 Å). In contrast, electronic effects strongly disfavor branch-selective addition with

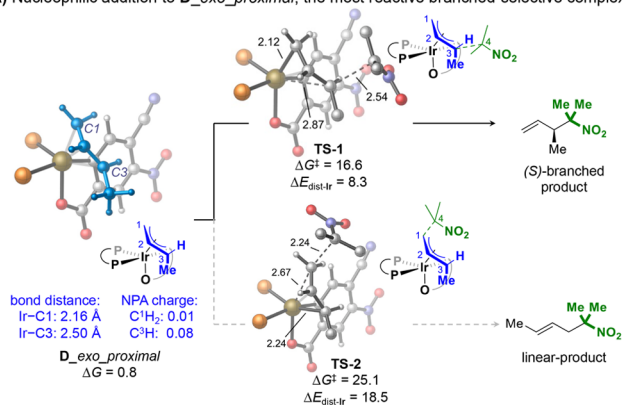
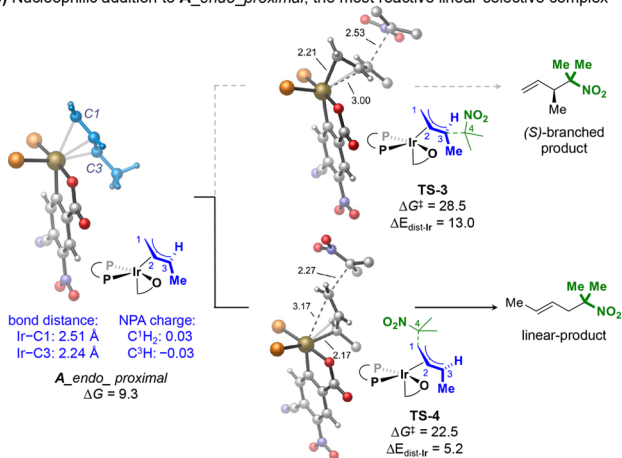
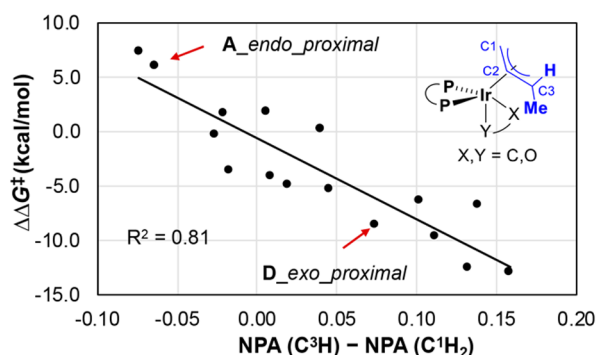
(A) Nucleophilic addition to **D_{exo}proximal**, the most reactive branched-selective complex(B) Nucleophilic addition to **A_{endo}proximal**, the most reactive linear-selective complex

Figure 3. Origin of regioselectivity. Bisphosphine ligand and some H atoms are omitted for clarity. All Gibbs free energies are in kcal/mol with respect to **A_{endo}distal**. ΔE_{dist-Ir}: distortion energy of the π -crotyliridium complex to reach its geometry in the TS for nucleophilic addition.

A_{endo}proximal (Figure 2B), because the more substituted terminus C3 is less electrophilic than C1, as evidenced by the negative charge on C3. In addition, the Ir–C3 bond becomes shorter than Ir–C1, leading to greater distortion energy (ΔE_{dist-Ir}) to elongate the Ir–C3 bond in the branch-selective transition state (TS-3).

The same analyses on all 16 isomers of the π -crotyliridium complex revealed good correlations between the computed regioselectivity for each isomer and the C3/C1 difference in NPA charge and Ir–C3 bond distance; that is, diastereomers with more positive charge on C3 and a longer Ir–C3 bond give higher branch-selectivity (Figure 4).²⁰ These results are consistent with our findings that regioselectivity is controlled by electronic effects, including the allyl electrophilicity and the distortion of the π -allyl complex. The electronic properties of the diastereomeric π -crotyl complexes also are affected by the trans-effect²¹ of the bisphosphine and the chelating C,O-benzoate ligands on the stereogenic Ir center. In diastereomers **A** and **B**, because of the stronger trans-effect of the aryl versus phosphine groups, addition occurs at sites *trans* to the aryl group, leading to branched products in “*distal*” isomers and linear products in “*proximal*” isomers. In diastereomers **C** and **D**, “*proximal*” isomers, in which C3 is *trans* to a phosphorus atom, give higher branched regioselectivity. This is consistent with the stronger trans-effects of phosphine versus carboxylate

(A) Correlation between NPA charge difference between allyl termini and computed regioselectivity



(B) Correlation between Ir–C3 distance and computed regioselectivity

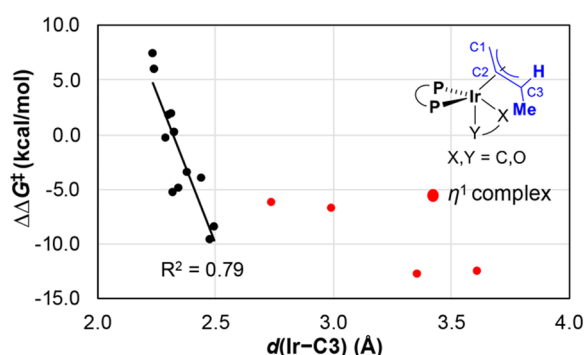


Figure 4. Electronic effects on allylation regioselectivity of different stereoisomers of the π -crotyliridium complex.

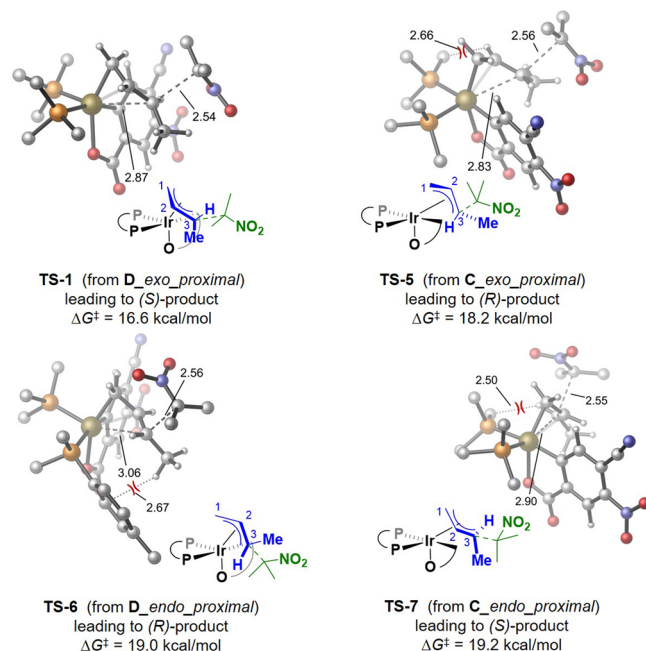


Figure 5. Optimized structures for additions of [−]CMe₂NO₂ anion giving branched product. Gibbs free energies are with respect to **A_{endo}distal**.

ligands. These branch-selective “*proximal*” isomers of **C** and **D** are also among the most reactive in all 16 stereoisomers (Figure 2B), as the π -acceptor ability of the phosphine ligand promotes addition at sites *trans* to phosphorus.²²

Stereogenicity at iridium is also a critical determinant of enantioselectivity. Because of trans-effects (*vide supra*), the “proximal” isomers of **C** and **D** are the most reactive among all 16 stereoisomers. The four most favorable pathways involve **D**_{exo}_{proximal} (TS-1) and **C**_{endo}_{proximal} (TS-7), which give (S)-product, and **D**_{endo}_{proximal} (TS-6) and **C**_{exo}_{proximal} (TS-5), which give (R)-product. Optimized structures of these branch-selective transition states are shown in Figure 5. The two transition states leading to the (R)-product, TS-5 and TS-6, are destabilized because the allyl C2–H moiety in TS-5 and the C3–methyl in TS-6 clash with a *P*-tolyl group of (S)-tol-BINAP. Thus, TS-5 and TS-6 are 1.6 and 1.8 kcal/mol less stable than the lowest-energy transition state, TS-1, that gives the (S)-product. In TS-1, the steric repulsions between tol-BINAP and the allyl group are absent. Our calculations are consistent with the experimentally observed enantioselectivity for the (S)-product.

In summary, we report iridium-catalyzed allylic alkylations of nitronate nucleophiles. This method, which employs an air- and water-stable π -allyliridium C,O-benzoate catalyst modified by tol-BINAP, enables highly regio- and enantioselective substitution of racemic branched alkyl-substituted allylic acetates by α,α -disubstituted nitronates and, hence, entry to β -stereogenic α -quaternary primary amines. As revealed by DFT calculations, early transition states that render the reaction less sensitive to steric effects and distinct trans-effects of diastereomeric chiral-at-iridium π -allyl complexes facilitate formation of congested tertiary–quaternary C–C bonds. Related asymmetric allylic alkylations of nonstabilized carbanions are currently underway.²³

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge at <https://pubs.acs.org/doi/10.1021/jacs.1c05212>.

Experimental procedures; spectroscopic and chromatographic data for all new compounds (¹H NMR, ¹³C NMR, IR, and HRMS), including HPLC traces for racemic and enantiomerically enriched compounds; computational details; single-crystal X-ray diffraction data for compound **4l** (PDF)

Accession Codes

CCDC 2081056 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

■ AUTHOR INFORMATION

Corresponding Authors

Craig E. Stivala – *Discovery Chemistry, Genentech, Inc., South San Francisco, California 94080, United States*; orcid.org/0000-0002-5024-6346; Email: stivala.craig@gene.com

Jason R. Zbieg – *Discovery Chemistry, Genentech, Inc., South San Francisco, California 94080, United States*; orcid.org/0000-0002-2897-8911; Email: zbieg.jason@gene.com

Peng Liu – *Department of Chemistry, University of Pittsburgh, Pittsburgh, Pennsylvania 15260, United States*; orcid.org/0000-0002-8188-632X; Email: pengliu@pitt.edu

Michael J. Krische – *Department of Chemistry, University of Texas at Austin, Austin, Texas 78712, United States*; orcid.org/0000-0001-8418-9709; Email: mkrische@mail.utexas.edu

Authors

Woo-Ok Jung – *Department of Chemistry, University of Texas at Austin, Austin, Texas 78712, United States*

Binh Khanh Mai – *Department of Chemistry, University of Pittsburgh, Pittsburgh, Pennsylvania 15260, United States*; orcid.org/0000-0001-8487-1417

Brian J. Spinello – *Department of Chemistry, University of Texas at Austin, Austin, Texas 78712, United States*

Zachary J. Dubey – *Department of Chemistry, University of Texas at Austin, Austin, Texas 78712, United States*

Seung Wook Kim – *Department of Chemistry, University of Texas at Austin, Austin, Texas 78712, United States*

Complete contact information is available at:

<https://pubs.acs.org/doi/10.1021/jacs.1c05212>

Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The Robert A. Welch Foundation (F-0038) and the NIH-NIGMS (RO1-GM069445, 1 S10 OD021508-01, R35 GM128779) are acknowledged. We thank Mr. Weijia Shen and Ms. Sakiho Sumikura for skillful technical assistance. Genentech is acknowledged for summer predoctoral internship support (W.-O.J., B.J.S., and S.W.K.). DFT calculations were carried out at the Center for Research Computing at the University of Pittsburgh, the Extreme Science and Engineering Discovery Environment (XSEDE), and the TACC Frontera Supercomputer supported by the National Science Foundation Grant Number ACI-1548562. We thank Dr. Michael Ruf of Bruker AXS for acquisition of crystallographic data.

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