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Oxidative Cyclization of 1-(Pyridin-2-yl)guanidine Derivatives: A Synthesis of [1,2,4]Triazolo[1,5-*a*]pyridin-2-amines and An Unexpected Synthesis of [1,2,4]Triazolo[4,3-*a*]pyridin-3-amines

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Oxidative Cyclization of 1-(Pyridin-2-yl)guanidine Derivatives: A Synthesis of [1,2,4]Triazolo[1,5-*a*]pyridin-2-amines and An Unexpected Synthesis of [1,2,4]Triazolo[4,3-*a*]pyridin-3-amines

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1. Introduction

N-fused heterocycles are one of the important classes of molecules that are found in a variety of natural products and biologically active compounds. Among a diverse array of *N*-fused heterocycles, [1,2,4]triazolo[1,5-*a*]pyridin-2-amine derivatives have attracted much attention due to their significant bioactivities in the field of pharmaceuticals. In fact, a large number of compounds that contain a [1,2,4]triazolo[1,5-*a*]pyridin-2-amine core have been recently reported as drug candidates, including PI3Kyinhibitors,¹ JAK2 inhibitors,² CRK3 inhibitors,⁵ TYK2 inhibitors,⁶ and Src kinase inhibitors⁷.

The most common synthetic approach to [1,2,4]triazolo[1,5-a]pyridin-2-amines **3** is described in Scheme 1.⁸ The reaction of 2-aminopyridines **1** with ethoxycarbonyl isothiocyanate gives thioureas **2**, which are treated with NH₂OH, hydrochloric acid and diisopropylethylamine (DIPEA) to give **3** in good to excellent yield, accompanied by evolution of H₂S and CO₂.⁹ This synthesis is so robust that it was applicable to our multi-hundred gram scale synthesis of an intermediate for a VEGFR-2 kinase inhibitor.¹⁰ However, despite the good yield and robustness, the synthesis suffered from the odor, toxicity, corrosivity, and

ABSTRACT

Oxidative cyclization of 1-(pyridin-2-yl)guanidine derivatives using *N*-chlorosuccinimide and aqueous potassium carbonate has been investigated. Chlorination of 1-(5-nitropyridin-2-yl)guanidine by *N*-chlorosuccinimide in methanol followed by addition of aqueous potassium carbonate gave rise to cyclization and afforded 6-nitro-[1,2,4]triazolo[1,5-a]pyridin-2-amine in one-pot. In the course of studying the scope and limitation of the reaction, it was found that some of the examined 1-(pyridin-2-yl)guanidine derivatives gave not only the desired [1,2,4]triazolo[1,5-a]pyridin-3-amine products. As plausible reaction mechanisms of this oxidative cyclization, diazirine formation and nitrene formation are presented.

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flammability of H_2S generated during the reaction, and an alternative convenient synthesis of **3** was highly desired.



Scheme 1. The most common synthetic approach to [1,2,4]triazolo[1,5-*a*]pyridin-2-amines **3**

2. Results and discussion

Grenda *et al.* reported that *N*-(pyridin-2-yl)benzimidamide was subjected to chlorination when treated with aqueous NaOCl and 1 M HCl, and treatment of the obtained *N*-chloroamidine with aqueous Na₂CO₃ gave rise to cyclization to afford

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[1,2,4]triazolo[1,5-*a*]pyridine.¹¹ On the basis of their work, we M thought that **3** could be synthesized from 1-(pyridin-2-yl)guanidines **5** using a similar reaction (Scheme 2). The guanidines **5** were envisaged to be prepared by S_NAr reaction of 2-halopyridines **4** with guanidine. It was assumed that oxidation of **5** would give intermediate **6**, and treatment of **6** with a base would lead to cyclization and afford **3**. To the best of our knowledge, oxidation of **5** followed by ring closure with a base has not been hitherto reported. Thus, in order to verify our concept, we first investigated chlorination of a known compound 1-(5-nitropyridin-2-yl)guanidine **5a**¹² (Table 2, entry 1) using *N*-chlorosuccinimide (NCS) and ring closure of the obtained chlorinated intermediate **6a** by aqueous K₂CO₃ (Scheme 3, (a)).



Scheme 2. Proposed synthesis of **3** from 1-(pyridin-2yl)guanidines **4**

When a slurry of **5a** in MeCN was treated with 1.1 equiv of NCS at 40 °C, generation of yellow solids was observed in a few minutes. After cooling to room temperature, the yellow solids were isolated by filtration. HRMS and NMR analysis revealed that the structure of the obtained compound was not **6b** but **6a** (Figure 1).^{13,14} Fortunately, ring closure of **6a** proceeded smoothly in MeCN by addition of 2.1 equiv of K₂CO₃ in H₂O at 40 °C, affording **3a** in 93% yield. Compound **3a** was also synthesized from 2-amino-5-nitropyridine according to the procedure described in Scheme 1 (See Experimental Section), and we confirmed that the spectral data of **3a** prepared by these two methods was identical.¹⁵

For the synthesis of **3a**, it was possible to use MeCN as the solvent for both the chlorination of **5a** and the ring closure of **6a**. These results led us to explore the possibility of a convenient one-pot synthesis of **3a** from **5a** in MeCN (Scheme 3, (b)). After **5a** was reacted with 1.1 equiv of NCS in MeCN at 40 °C for 15 min, 2.1 equiv of K₂CO₃ in H₂O was added, and the mixture was stirred for 30 min at the same temperature. Following extraction with EtOAc and washing with 10% aqueous NaCl, the organic

Mayer was analyzed by HPLC using an authentic sample, which
showed that 3a was obtained in 79% yield.



Figure 1. Structure determination of 6a

With this promising result in hand, optimization of the reaction condition was conducted through a screening of oxidizing agents, bases and solvents (Table 1). Among the screened chlorinating agents, *t*-BuOCl gave **3a** in good yield comparable to NCS (80%, entry 5). Other chlorinating agents such as aqueous NaOCl, chloramine-T, trichloroisocyanuric acid (TCCA), 1,3-dichloro-5,5-dimethylhydantoin (NDDH) all decreased the yield (entries 1–4). The use of other oxidizing agents was also examined. While *N*-bromosuccinimide (NBS) gave the product in low yield (14%, entry 6), *N*-iodosuccinimide (NIS) did not give the product (entry 7). Taking ease of handling and price of the reagent into consideration, NCS was selected as the oxidizing agent for further optimization.

Organic bases such as pyridine, triethylamine and DBU gave the product in lower yield than aqueous K_2CO_3 (entries 8–10). Interestingly, solid K_2CO_3 gave the product in lower yield (34%, entry 11) than aqueous K_2CO_3 . The use of 1M NaOH gave the product in comparable yield to aqueous K_2CO_3 (77%, entry 12), whereas the use of aqueous KOAc largely decreased the yield (5%, entry 13). These results indicate that the use of a strong inorganic base and the presence of H_2O are essential for high conversion of **6a** to **3a**.

It was found that alcohol solvents could also be used for this one-pot process (entries 14–16). When alcohol solvents were used for the reaction, precipitation of the product was observed after addition of aqueous K_2CO_3 . Following further addition of H_2O to the reaction mixture, **3a** was easily isolated by filtration. The best yield (87% isolated yield) was achieved when MeOH was used as the solvent (entry 14).



Scheme 3. Synthesis of 3a from 5a via 6a

Table 1. Optimization of the reaction condition for the one-D MANUSCRIP pot synthesis of 3a 2 A

3a



entry	oxidizing agent	solvent	base	yield ^a
				(%)
1	aq NaOCl	MeCN	aq K ₂ CO ₃	41
2	chloramine-T	MeCN	aq K ₂ CO ₃	42
3	TCCA ^b	MeCN	aq K ₂ CO ₃	69
4	NDDH	MeCN	aq K ₂ CO ₃	42
5	t-BuOCl	MeCN	aq K ₂ CO ₃	80
6	NBS	MeCN	aq K ₂ CO ₃	13
7	NIS	MeCN	aq K ₂ CO ₃	0
8	NCS	MeCN	pyridine	32
9	NCS	MeCN	triethylamine	15
10	NCS	MeCN	DBU	51
11	NCS	MeCN	K ₂ CO ₃	34
12	NCS	MeCN	1M NaOH	77
13	NCS	MeCN	aq KOAc	5
14	NCS	MeOH	aq K ₂ CO ₃	87 ^c
15	NCS	EtOH	aq K ₂ CO ₃	74 ^c
16	NCS	<i>i</i> -PrOH	aq K ₂ CO ₃	68 ^c

^a HPLC assay yield.

^b 0.37 equiv of TCCA was used for the reaction.

^c Isolated yield.

In order to study the scope and limitation of the reaction, a series of 1-(pyridin-2-yl)guanidine derivatives **5b–5m** were prepared by S_NAr reaction of 2-halopyridine derivatives **4** with guanidine carbonate or guanidine hydrochloride as described in Scheme 2 (Table 2). Although many kinds of 2-halopyridines were examined, the S_NAr reaction proceeded only when 2-halopyridines possessed electron withdrawing groups (entries 1–9). Due to the low solubility of **5** in common organic solvents, isolation of **5** required a rather troublesome work-up and **5** was obtained in low to moderate yields.

Table 2. Preparation of 1-(pyridin-2-yl)guanidine derivatives 5





^a Reaction conditions described in the table are as follows:

A guanidine hydrochloride (10 equiv), K_2CO_3 (15 equiv), *t*-BuOH, reflux.

B guanidine carbonate (3 equiv), K_2CO_3 (5 equiv), 1-methylpyrrolidin-2-one (NMP), 120–140 °C.

C guanidine carbonate (1 equiv), K_2CO_3 (3 equiv), NMP, 100–120 °C.

D guanidine carbonate (3 equiv), K_2CO_3 (5 equiv), *N*,*N*-dimethylacetamide, 120 °C.

^b Isolated yield.

^c 0.9 NMP solvate was obtained.

With these substrates in hand, the one-pot oxidative cyclization was applied to compounds 5b-5m (Table 3). When 1-(3-nitropyridin-2-yl)guanidine 5b was subjected to the abovementioned one-pot chlorination-cyclization procedure, the product precipitated in the reaction mixture. After addition of

4	Tetrahedron
H ₂ O to the reaction mi	xture, 3b was isolated by filtration in MANUSCRIP5b
86% yield (entry 1).	

The reaction of 1-(3-chloropyridin-2-yl)guanidine 5c, in contrast, gave a completely different result from the results of the reactions of 5a and 5b (entry 2). When 5c was treated with 1.1 equiv of NCS, chlorination proceeded smoothly to give a single chlorinated product. However, treatment of the chlorinated intermediate with aqueous K₂CO₃ unexpectedly gave two main products with several kinds of byproducts. After isolation by column chromatography, the two products were analyzed by HRMS and NMR. Interestingly, HRMS analysis showed that the two compounds have the same accurate molecular mass corresponding to molecular formula C₆H₅ClN₄. In addition, ¹H NMR spectra of the two compounds showed a similar coupling pattern: one triplet-like peak and two doublet-like peaks in the aromatic region, and one peak corresponding to NH₂ protons, whereas the chemical shifts of the corresponding peaks in the two compounds were different. These results implied that treatment of the chlorinated intermediate with aqueous K₂CO₃ gave not only the desired 3c but a compound having a similar structure to 3c via an unexpected reaction pathway. Subsequent twodimensional NMR experiments such as HMBC and NOESY spectra indicated that the structure of the unexpected product was **7c** depicted in Figure 2.¹⁶ The structures of **3c** and **7c** obtained in this one-pot oxidative cyclization were confirmed by comparing the spectral data with that of 3c and 7c prepared according to known procedures (Scheme 4).^{8, 17}

In the case of 1-(5-chloropyridin-2-yl)guanidine 5d, compound 7d was obtained as the main product in 50% yield with a trace amount of 3d (entry 3). Then 1-(pyridin-2yl)guanidines bearing a CF3 group at three different positions were subjected to the reaction. Compound 5e bearing a 3-CF₃ group gave 3e as the main product in 44% yield and 7e as the minor product in 20% yield, respectively (entry 4). On the other hand, in the case of the 4-CF₃ and 5-CF₃ derivatives, compound 7 became the main product. Compound 5f gave 7f in 52% yield (entry 5), and 5g gave 7g in 50% yield (entry 6), respectively. Interestingly, the reaction of **5h** bearing a 2-F group gave neither 3h nor 7h (entry 7). Although the chlorination of 5h proceeded cleanly, addition of aqueous K₂CO₃ made the reaction mixture a dark purple solution, and generation of several kinds of unidentified products was observed. Compound 5i bearing a 3-Cl group and a 5-CF₃ group gave 3i as the main product in 57% yield (entry 8).

The reaction was also applied to guanidines containing a bicyclic ring such as 1-(quinolin-2-yl)guanidine **5j** and 1-(isoquinolin-1-yl)guanidine **5k**. Fortunately, both **5j** and **5k** were successfully chlorinated by NCS, and treatment with aqueous K_2CO_3 gave tricyclic compounds. When **5j** was subjected to the reaction, **7j** was obtained as the main product in 39% yield (entry 9). On the other hand, the reaction of **5k** gave **3k** as the main product in 35% yield (entry 10). Unfortunately, in the case of 1-(benzo[d]oxazol-2-yl)guanidine **5m**, the cyclized products were not obtained (entry 11).

Table 3. One-pot oxidative cyclization of 1-(pyridin-2yl)guanidine derivatives $\mathbf{5}^{a}$

entry	substrate	product		
1	$\bigcup_{NO_2}^{N} \bigcup_{NH_2}^{NH}$			



^a The substrates in MeOH were treated with 1.1 equiv of NCS at 40 °C for 15 min. Then 2.1 equiv of K_2CO_3 in H_2O was added, and the reaction mixture was stirred at 40 °C for 30 min.

^bNot Detected.

^c 0.9 NMP solvate of **5f** was used as the substrate.



NOE

Figure 2. Structure determination of **7c**

In Table 3, the substrates having an electron withdrawing group at 3-position of the pyridine ring (**5b**, **5c**, **5e**) gave **3** as the major product. Meanwhile, while **5d** and **5f**, which have an electron withdrawing group at 5-position, mainly gave **7**, compound **5a** with a strong electron withdrawing group (NO₂) at 5-position gave exclusively compound **3**. Interestingly, although compound **5f** gave mainly compound **7**, compound **5i**, which have one more additional electron withdrawing group (Cl) at 3-position, showed a different reactivity from **5f** and gave mainly compound **3**. These results may indicate that the type, position and number of substituents affect the reaction pathway.¹⁸

mentioned previously, oxidation of 1-(pyridin-2-As yl)guanidine derivatives 5 and their ring closure with a base has not been previously reported. Meanwhile, a literature survey shows that treatment of amidines with aqueous NaOCl/NaOBr gives 3-halo-3-substituted diazirines via an interesting reaction mechanism (Scheme 5).¹⁹ In this diazirine formation reported first by Graham, intermediates other than N-chloro and N,Ndichloroamidines have not been directly observed.²⁰ Thus, it is not clear whether the diazirine formation proceeds via direct displacement of a halogen atom with a nitrogen anion or via formation of iminonitrene. Moss et al. intensively studied this reaction and reported that dichlorination of amidines is prerequisite for the base promoted diazirine formation, while Nchloro and N,N',N'-trichloroamidines did not afford diazirine.21 Probably, the second chlorination of a NH2 group would be essential because the chlorine atom acts as an electron withdrawing group and promotes deprotonation of the NH proton by a base.



Scheme 5. Formation of 3-halo-3-substituted diazirines by Graham's procedure (ref 19)

In the case of **5**, an electron deficient pyridine ring is thought to increase acidity of the neighboring NH proton, which would cause base promoted formation of diazirine **12** via **11** (Scheme 6, (a)). Ring opening of **12** is assumed to proceed in two different pathways. The attack of a nitrogen atom of a pyridine ring to the nitrogen atom of the diazirine would give **3** via N-N bond cleavage (path A), while the attack of the nitrogen atom of the pyridine ring to the carbon atom of the C=N double bond of the diazirine would give **7** via C-N bond cleavage (path B).

An alternative possible reaction mechanism is the one involving nitrene formation (Scheme 6, (b)).²² Anion 11 could generate nitrene 13 by loss of a chlorine atom, and ring closure of 13 would give 3 (path C).¹¹ At the same time, there is a possibility that nitrene 14, which is an isomer of 13, could give carbodiimide 15 via Curtius-type rearrangement.^{23, 24} The attack of the nitrogen atom of the pyridine ring to the carbon atom of the carbodiimide would give 7 (path D).

At present, it is difficult to predict which reaction pathway mainly works in each substrate. Further detailed investigation is required to elucidate the factors (substituent, oxidizing agent, temperature, base and solvent) that control the dominant reaction pathway.

3. Conclusions

We have shown that chlorination of 1-(5-nitropyridin-2yl)guanidine **5a** by NCS followed by treatment with aqueous K_2CO_3 gives rise to cyclization and affords 6-nitro-[1,2,4]triazolo[1,5-*a*]pyridin-2-amine **3a**. It was possible to conduct this reaction sequence in one-pot by using MeOH as the solvent. In the course of studying the scope and limitation of this one-pot oxidative cyclization, it was found that some of the 1-(pyridin-2-yl)guanidine derivatives **5** gave not only [1,2,4]triazolo[1,5-*a*]pyridin-2-amines **3** but [1,2,4]triazolo[4,3*a*]pyridin-3-amines **7**. Although we proposed plausible reaction mechanisms based on the diazirine formation and the nitrene formation, the precise reaction mechanism remains to be elucidated. A more detailed study on this reaction is to be performed in the future.



51%



7c

Scheme 4. Preparation of 3c and 7c according to known procedures

32%

10



Scheme 6. Plausible reaction mechanisms of the oxidative cyclization of 5

4.1. General

All materials were purchased from commercial suppliers and used without any additional purification. NH silica gel (CHROMATOREX) was purchased from Fuji Silysia Chemical Ltd. Melting points were determined on a Büchi Melting Point B-540 and on a Stanford Research Systems OptiMelt MPA 100, and are uncorrected unless otherwise noted. ¹H and ¹³C NMR spectra were recorded on a BRUKER AVANCE 500 spectrometer and on a BRUKER AVANCE 600 spectrometer with tetramethylsilane as an internal standard. Chemical shifts are shown in ppm. HPLC analysis of the compounds and reaction monitoring was carried out on a Shimadzu LC-2010C_{HT}. Highresolution mass spectrometry (HRMS) data was obtained on a Shimadzu Prominence UFLC system with a Thermofisher LTQ Orbitrap Discovery. IR spectra were recorded on a Thermo Electron FT-IR Nicolet 4700 (ATR). Elemental analyses were recorded on an Elementar vario MICRO cube. HPLC conditions are as follows: Inertsil ODS-3 column, 5 $\mu m,\,250\;mm\times4.6\;mm$ i.d.; UV detector at 254 nm; isocratic elution with CH₃CN/50 mM aqueous KH₂PO₄ (pH 7) (30:70) at 1.0 mL/min flow rate; column temperature: 25 °C. Retention times: 5a (3.9 min), 3a (4.9 min).

4.2. General procedure for the one-pot oxidative cyclization

To a solution of 1-(3-chloropyridin-2-yl)guanidine 5c (1.00 g, 5.86 mmol) in MeOH (100 mL) was added N-chlorosuccinimide (861 mg, 6.45 mmol) at 40 °C, and the mixture was stirred at the same temperature for 15 min. To the resulting slurry was added K_2CO_3 (1.70 g, 12.3 mmol) in H_2O (20 mL) at 40 °C, and the mixture was stirred at the same temperature for 30 min. After cooling to room temperature, the solvent was concentrated in vacuo. To the residue were added EtOAc (100 mL) and 10% aqueous NaCl (50 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (2×100 mL), and the combined organic layer was washed with 10% aqueous K₂CO₃ (2 \times 50 mL) and 10% aqueous NaCl (50 mL). The organic layer was concentrated in vacuo and the residue was purified by column chromatography (NH silica gel, 1:4 EtOAc/hexane to 3:97 MeOH/EtOAc) to give crude 3c (275 mg) and 7c (190 mg, 19%, pale brown solid). Crude 3c (150 mg) was triturated with EtOAc/hexnae (1:1, 2 mL) and filtered to give 3c (148 mg, 27%) as a pale brown solid.

4.3. Syntheses of 1-(pyridin-2-yl)guanidine derivatives 5

4.3.1. 1-(5-Nitropyridin-2-yl)guanidine (5a)¹²

To a suspension of 2-chloro-5-nitropyridine 4a (5.00 g, 31.5 mmol) and guanidine hydrochloride (30.1 g, 315 mmol) in t-BuOH (250 mL) was added K₂CO₃ (65.3 g, 473 mmol), and the mixture was refluxed for 50 h. The solvent was removed by evaporation, and H₂O (200 mL) was added to the residue. The slurry was stirred at room temperature for 1 h and then filtered. The wet cake was washed with H_2O (2 × 10 mL). The resulting solids were suspended in EtOH/H2O (1:6, 11 mL), and the mixture was stirred at 60 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered. The wet cake was washed with $H_2O(3 \times 2 \text{ mL})$ and dried in vacuo to give the crude product (3.69 g). The crude product (3.00 g) was suspended in EtOAc/hexane (1:2, 27 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered. The obtained solids were suspended in EtOAc/hexane (1:1, 12 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 5a as an

DMSO- d_6) δ 6.58 (d, J = 9.5 Hz, 1H), 7.30 (br s, 4H), 8.10 (dd, J = 9.3, 3.0 Hz, 1H), 8.93 (d, J = 2.8 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 118.0, 131.3, 134.8, 144.6, 159.7, 167.2; IR (ATR) 3406, 3033, 1664, 1599, 1563, 1509, 1461, 1412, 1313, 1262, 1111, 992, 947, 934, 864, 836, 771, 726, 713, 531, 496, 452, 420 cm⁻¹; HRMS-ESI (m/z): [M + H]⁺ calcd for C₆H₈N₅O₂, 182.0678; found, 182.0673.

4.3.2. 1-(3-Nitropyridin-2-yl)guanidine (5b)

To a suspension of 2-chloro-3-nitropyridine 4b (5.00 g, 31.5 mmol) and guanidine hydrochloride (30.1 g, 315 mmol) in t-BuOH (150 mL) was added K₂CO₃ (65.4 g, 473 mmol), and the mixture was refluxed for 40 h. The solvent was removed by evaporation, and EtOAc (500 mL) was added to the residue. The suspension was filtered through a pad of NH silica gel, and the pad was washed with EtOAc (2×100 mL). The combined filtrate was washed with saturated brine $(3 \times 15 \text{ mL})$ and 10% aqueous NaCl $(2 \times 15 \text{ mL})$, and concentrated in vacuo. The resulting solids were suspended in EtOAc/hexane (1:2, 45 mL) and stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtrated to give the crude product (3.99 g). The crude product (1.00 g) was suspended in EtOAc/hexane (1:2, 9 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtrated to give 5b as a yellow solid (462 mg, 32%). mp 140–141 °C (lit.²⁵ mp 143–144 °C); ¹H NMR (500 MHz, DMSO- d_6) δ 6.73 (dd, J = 7.6, 5.0 Hz, 1H), 6.96 (br s, 4H), 7.86 (dd, J = 7.9, 1.9 Hz, 1H), 8.21 (dd, J = 4.7, 1.9 Hz, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 111.9, 130.8, 140.3, 148.8, 154.7, 158.4; IR (ATR) 3458, 3412, 3128, 2362, 1651, 1581, 1536, 1491, 1417, 1348, 1325, 1257, 1044, 877, 837, 760, 730, 714, 586, 555, 527, 447 cm⁻¹; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₆H₈N₅O₂, 182.0678; found, 182.0670.

4.3.3. 1-(3-Chloropyridin-2-yl)guanidine (5c)

To a suspension of 2,3-dichloropyridine 4c (4.50 g, 30.4 mmol) and guanidine carbonate (16.4 g, 91.2 mmol) in 1methylpyrrolidin-2-one (NMP) (40 mL) was added K₂CO₃ (21.0 g, 152 mmol), and the mixture was heated to 140 °C and stirred for 30 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (5 \times 90 mL). The combined filtrate was washed with H₂O (45 mL) and 10% aqueous NaCl (3×45 mL), and concentrated in vacuo. The residue was suspended in EtOAc/hexane (1:2, 45 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered. The obtained solids were suspended in H₂O (45 mL), and the slurry was stirred at 50 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered to give 5c (2.01 g, 39%) as a white solid. mp 138 °C; ¹H NMR (500 MHz, DMSO- d_6) δ 6.63 (dd, J =7.7, 4.9 Hz, 1H), 6.83 (br s, 4H), 7.61 (dd, J = 7.7, 1.7 Hz, 1H), 7.98 (dd, J = 4.9, 1.7 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 113.8, 123.0, 136.9, 144.1, 157.8, 159.2; IR (ATR) 3474, 3365, 3148, 2357, 1609, 1555, 1512, 1421, 1311, 1230, 1154, 1121, 1036, 997, 951, 855, 764, 731, 668, 646, 558, 523, 476, 455, 431, 405 cm⁻¹; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₆H₈ClN₄, 171.0437; found, 171.0432.

4.3.4. 1-(5-Chloropyridin-2-yl)guanidine (5d)

To a suspension of 5-chloro-2-fluoropyridine **4d** (5.00 g, 38.0 mmol) and guanidine carbonate (20.5 g, 114 mmol) in NMP (50 mL) was added K_2CO_3 (26.3 g, 190 mmol), and the mixture was heated to 130 °C and stirred for 8 h. To the mixture at room temperature were added EtOAc (100 mL) and H₂O (200 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (3 × 100 mL), and the combined organic layer was

washed with 10% aqueous NaCl (3×50 mL) and concentrated \bigvee in vacuo. To the resulting residue was added EtOAc (500 mL), and the solution was filtered through a pad of NH silica gel. The filtrate was evaporated, and the residue was suspended in EtOAc/hexane (1:3, 40 mL). The mixture was stirred at 40 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered. The obtained solids were dissolved into H₂O (15 mL) at 50 °C, and the solution was gradually cooled to room temperature. The slurry was stirred at room temperature for 1 h, and then filtered to give 5d (3.03 g, 47%) as a white solid. mp 162–163 °C; ¹H NMR (500 MHz, DMSO- d_6) δ 6.60 (dd, J =8.8, 0.6 Hz, 1H), 6.75 (br s, 4H), 7.48 (dd, J = 8.8, 2.8 Hz, 1H), 8.04 (d, J = 2.5 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 119.3, 120.0, 136.5, 143.7, 157.6, 162.1; IR (ATR) 3468, 3414, 2362, 1627, 1573, 1543, 1518, 1463, 1370, 1314, 1278, 1229, 1133, 1110, 1004, 914, 857, 834, 756, 730, 614, 572, 520 459, 436, 405 cm⁻¹; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₆H₈ClN₄ 171.0437; found, 171.0432.

4.3.5. 1-(3-(Trifluoromethyl)pyridin-2-yl)guanidine (5e)

To a suspension of 2-chloro-3-(trifluoromethyl)pyridine 4e (10.0 g, 55.3 mmol) and guanidine carbonate (29.9 g, 166 mmol) in NMP (100 mL) was added K₂CO₃ (38.2 g, 277 mmol), and the mixture was heated to 120 °C and stirred for 40 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (5 \times 200 mL). The combined filtrate was washed with H₂O (100 mL) and 10% aqueous NaCl $(3 \times 100 \text{ mL})$, and concentrated in vacuo. The residue was purified by column chromatography (NH silica gel, EtOAc/hexane 1:10 to 2:3) to give the crude product. To the crude product in EtOAc (200 mL) was added activated carbon (500 mg), and the mixture was stirred at room temperature for 30 min. The activated carbon was filtered off, and the filtrate was concentrated in vacuo. The residue was dissolved into EtOAc (100 mL), and 4M HCl in EtOAc (14 mL, 55.3 mmol) was added at 5 °C. The mixture was stirred at 5 °C for 30 min, and then filtered. To the obtained solids were added EtOAc (200 mL) and aqueous 5% NaHCO₃ (50 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (100 mL), and the combined organic layer was concentrated in vacuo. To the residue was added EtOAc/hexane (1:3, 10 mL), and the slurry was stirred at 5 °C for 1 h, and then filtered to give 5e (1.31 g, 12%) as a white solid. mp 128–130 °C; ¹H NMR (500 MHz, DMSO- d_6) δ 6.72 (dd, J = 7.4, 4.9 Hz, 1H), 6.87 (br s, 4H), 7.78 (dd, J = 7.6, 1.9 Hz, 1H), 8.23 (dd, J = 5.0, 1.6 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 111.9, 115.8 (q, ${}^2J_{CF} = 28.8$ Hz), 124.3 (q, ${}^{1}J_{CF} = 270.0$ Hz), 135.1 (q, ${}^{3}J_{CF} = 6.3$ Hz), 149.7, 157.7, 160.5; IR (ATR) 3483, 3449, 3384, 3164, 1614, 1592, 1561, 1513, 1430, 1334, 1301, 1256, 1226, 1158, 1127, 1102, 1070, 1027, 965, 859, 806, 779, 742, 695, 615, 583, 543, 483, 471, 448, 435, 422 cm⁻¹; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₇H₈F₃N₄, 205.0701; found, 205.0696.

4.3.6. 1-(5-(Trifluoromethyl)pyridin-2-yl)guanidine 0.9 1-methylpyrrolidin-2-one solvate (5f)

To a suspension of 2-chloro-5-(trifluoromethyl)pyridine **4f** (8.00 g, 44.1 mmol) and guanidine carbonate (7.94 g, 44.1 mmol) in NMP (80 mL) was added K₂CO₃ (18.3 g, 132 mmol), and the mixture was heated to 120 °C and stirred for 7 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (5×160 mL). The combined filtrate was washed with H₂O (80 mL) and 10% aqueous NaCl (3×80 mL), and concentrated *in vacuo*. To the residue was added EtOAc/hexane (1:2, 40 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered to give the crude product (6.13 g). The crude

product (1.00 g) was suspended in EtOAc/hexane (1:2, 9 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 5f (829 mg, 39%) as a white solid. mp 95–97 °C; 1 H NMR (500 MHz, DMSO-*d*₆) δ 1.85–1.96 (m, 1.8H, NMP), 2.18 (t, J = 8.0 Hz, 1.8H, NMP), 2.70 (s, 2.7H, NMP), 3.30 (t, J = 6.9 Hz, 1.8H, NMP), 6.66 (d, J = 8.8 Hz, 1H), 7.01 (br s, 4H), 7.67 (dd, J = 8.8, 2.5 Hz, 1H), 8.35 (br s, 1H); ¹³C NMR (125 MHz, DMSO-d₆) δ 17.2 (NMP), 28.9 (NMP), 30.1 (NMP), 48.4 (NMP), 113.9 (q, ${}^{2}J_{CF} = 32.1$ Hz), 118.3, 124.9 (q, ${}^{1}J_{CF} = 268.3$ Hz), 133.1 (q, ${}^{3}J_{CF} = 2.5$ Hz), 143.5 (q, ${}^{3}J_{CF} = 5.0$ Hz), 158.8, 166.1, 173.7 (NMP); IR (ATR) 3483, 3449, 3384, 3163, 1614, 1591, 1561, 1512, 1430, 1334, 1301, 1256, 1226, 1158, 1127, 1102, 1070, 1027, 965, 859, 806, 779, 742, 695, 651, 614, 584, 542, 480, 470, 448, 435, 422, 410 cm⁻¹; HRMS-ESI (m/z): [M + H_{1}^{+} calcd for $C_{7}H_{8}F_{3}N_{4}$, 205.0701; found, 205.0694; Anal. Calcd for C_{11.5}H_{15.1}N_{4.9}F₃O_{0.9}: C, 47.08; H, 5.19; N, 23.39. Found: C, 46.96 H, 5.13; N, 23.17.

4.3.7. 1-(4-(Trifluoromethyl)pyridin-2-yl)guanidine (5g)

To a suspension of 2-chloro-4-(trifluoromethyl)pyridine 4g (4.50 g, 24.8 mmol) and guanidine carbonate (13.4 g, 74.4 mmol) in NMP (45 mL) was added K₂CO₃ (17.1 g, 124 mmol), and the mixture was heated to 120 °C and stirred for 18 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (5 \times 90 mL). The combined filtrate was washed with H₂O (45 mL) and 10% aqueous NaCl (3 \times 45 mL), and concentrated *in vacuo*. The residue was suspended in EtOAc/hexane (1:2, 23 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered. The obtained solids were suspended in H₂O (13 mL) and the slurry was stirred at 50 °C for 1 h. The mixture was gradually cooled by ice bath and stirred for 1 h, and then filtered to give 5g (2.39 g, 47%) as a white solid. mp 141-142 °C; ¹H NMR (500 MHz, DMSO- d_6) δ 6.75 (s, 1H), 6.85 (dd, J =5.4, 1.6 Hz, 1H), 6.92 (br s, 4H), 8.25 (d, J = 5.4 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 107.5 (q, ${}^{3}J_{CF} = 2.5$ Hz), 113.9 (q, ${}^{3}J_{\rm CF} = 3.8$ Hz), 123.2 (q, ${}^{1}J_{\rm CF} = 271.3$ Hz), 137.3 (q, ${}^{2}J_{\rm CF} = 32.5$ Hz), 147.7, 158.3, 164.2; IR (ATR) 3388, 3045, 1663, 1595, 1527, 1411, 1335, 1293, 1271, 1171, 1145, 1122, 1078, 980, 871, $817, 793, 758, 731, 689, 667, 561, 536, 467, 444, 421, 401 \text{ cm}^{-1}$; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₇H₈F₃N₄, 205.0701; found, 205.0693.

4.3.8. 1-(6-Fluoropyridin-2-yl)guanidine (5h)

To a suspension of 2,6-difluoropyridine 4h (4.00 g, 34.8 mmol) and guanidine carbonate (12.5 g, 69.6 mmol) in N,Ndimethylacetamide (80 mL) was added K₂CO₃ (14.4 g, 104 mmol), and the mixture was heated to 120 °C and stirred for 6 h. At this point, guanidine carbonate (6.27 g, 34.8 mmol) and K₂CO₃ (9.62 g, 69.6 mmol) were added, and the mixture was stirred at 120 °C for an additional 3 h. After cooling to room temperature, EtOAc (160 mL) and H₂O (240 mL) were added, and the layers were separated. The aqueous layer was extracted with EtOAc (5 \times 40 mL), and the combined organic layer was washed with saturated brine $(2 \times 40 \text{ mL})$ and 10% aqueous NaCl $(3 \times 40 \text{ mL})$, and concentrated *in vacuo*. The resulting residue was purified by column chromatography (NH silica gel, EtOAc/hexane 1:1). The obtained solids were suspended in EtOAc/hexane (1:2, 36 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered. The obtained solids were suspended in H₂O (12 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give the crude product (2.31 g). The crude product (1.00 g) was suspended in EtOAc/hexane (1:3, 5 mL),

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and the slurry was stirred at 50 °C for 1 h. The mixture was gradually cooled to room temperature and stirred for 1 h, and then filtered to give **5h** (866 mg, 37%) as a white solid. mp 121–122 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 6.27 (dd, *J* = 7.6, 2.2 Hz, 1H), 6.48 (dd, *J* = 8.0, 2.1 Hz, 1H), 6.67 (br s, 4H), 7.56 (dt, *J* = 9.8, 7.9 Hz, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 96.4 (d, ²*J* = 36.3 Hz), 115.6 (d, ⁴*J*_{HF} = 5.0 Hz), 141.5 (d, ³*J*_{HF} = 6.3 Hz), 158.0, 161.2 (d, ¹*J*_{HF} = 233.8 Hz), 162.6 (d, ³*J*_{HF} = 12.6 Hz); IR (ATR) 3485, 3310, 3137, 2255, 2179, 1709, 1635, 1608, 1564, 1537, 1487, 1464, 1416, 1334, 1260, 1243, 1222, 1151, 1055, 1035, 988, 957, 866, 781, 726, 696, 673, 596, 572, 555, 472, 422, 404 cm⁻¹; HRMS-ESI (m/z): [M + H]⁺ calcd for C₆H₈FN₄, 155.0733; found, 155.0725.

4.3.9. 1-(3-Chloro-5-(trifluoromethyl)pyridin-2yl)guanidine (5i)

To a suspension of 2,3-dichloro-5-(trifluoromethyl)pyridine 4i (10.0 g, 46.3 mmol) and guanidine carbonate (8.34 g, 46.3 mmol) in NMP (100 mL) was added K₂CO₃ (19.2 g, 139 mmol), and the mixture was heated to 100 °C and stirred for 5 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (200 mL). To the combined filtrate was added 10% aqueous NaCl (100 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (3 \times 100 mL). The combined organic layer was washed with 10% aqueous NaCl $(3 \times 100 \text{ mL})$ and concentrated in vacuo. To the residue was added EtOAc (100 mL), and the solution was cooled by ice bath. To the solution was added 4M HCl in EtOAc (23 mL, 92.6 mmol), and the mixture was stirred for 2 h, and then filtered. To the obtained solids were added EtOAc (200 mL), H₂O (50 mL) and 8M NaOH (12 mL, 92.6 mL), and the layers were separated. The aqueous layer was extracted with EtOAc (3 \times 100 mL). The combined organic layer was concentrated in vacuo. The residue was purified by column chromatography (NH silica gel, EtOAc/hexane 1:3 to 1:1) to give the crude product. To the crude product was added H₂O (10 mL), and the mixture was stirred at 60 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 5i (4.02 g, 36%) as a white solid. mp 146–149 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 7.14 (br s, 4H), 7.91 (d, *J* = 2.2 Hz, 1H), 8.30 (dd, *J* = 2.4, 1.1 Hz, 1H); ¹³C NMR (125 MHz, DMSO- d_6) δ 113.9 (q, ${}^{2}J_{\rm CF}$ = 32.5 Hz), 122.8, 124.0 (q, ${}^{1}J_{\rm CF}$ = 268.8 Hz), 132.9 (q, ${}^{3}J_{\rm CF}$ = 2.5 Hz), 141.7 (q, ${}^{3}J_{CF}$ = 5.0 Hz), 159.2, 161.5; IR (ATR) 3511, 3457, 3405, 3324, 1624, 1600, 1542, 1519, 1461, 1405, 1382, 1331, 1307, 1257, 1153, 1112, 1088, 1054, 999, 937, 910, 878, $809, 773, 721, 670, 641, 598, 543, 481, 465, 421, 405 \text{ cm}^{-1};$ HRMS-ESI (m/z): $[M + H]^+$ calcd for $C_7H_7ClF_3N_4$, 239.0311; found, 239.0309.

4.3.10. 1-(Quinolin-2-yl)guanidine (5j)

To a suspension of 2-chloroquinoline 4i (4.50 g, 27.5 mmol) and guanidine carbonate (14.9 g, 82.5 mmol) in NMP (45 mL) was added K₂CO₃ (19.0 g, 138 mmol), and the mixture was heated to 120 °C and stirred for 17 h. After cooling to room temperature, EtOAc (90 mL) was added, and the mixture was filtered through a glass filter. The cake was washed with EtOAc $(4 \times 90 \text{ mL})$. To the combined filtrate was added H₂O (45 mL), and the layers were separated. The organic layer was washed with 10% aqueous NaCl (2×45 mL) and concentrated in vacuo. The residue was suspended in EtOAc/hexane (1:2, 45 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered. The obtained solids were suspended in H₂O (14 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered to give 5j (2.54 g, 50%) as a white solid. mp 218–219 °C; ¹H NMR (500 MHz, DMSO- d_6) δ 6.80 (d, J = 8.8 Hz, 1H), 7.18 (br s, 4H), 7.22 (td, J = 7.4, 1.3 Hz,

4H), 7.43–7.53 (m, 1H) 7.59 (d, J = 8.5 Hz, 1H), 7.65 (dd, J = 8.0, 1.1 Hz, 1H), 7.90 (d, J = 8.8 Hz, 1H),; ¹³C NMR (125 MHz, DMSO- d_6) δ 121.7, 122.2, 123.4, 125.7, 127.1, 128.6, 135.9, 146.4, 158.9, 162.9; IR (ATR) 3411, 2997, 1657, 1620, 1585, 1535, 1491, 1444, 1417, 1378, 1349, 1309, 1284, 1246, 1212, 1142, 1119, 1016, 1003, 974, 944, 929, 828, 788, 763, 752, 684, 640, 593, 550, 478, 467, 447, 416 cm⁻¹; HRMS-ESI (m/z): [M + H]⁺ calcd for C₁₀H₁₁N₄, 187.0984; found, 187.0976.

4.3.11. 1-(Isoquinolin-1-yl)guanidine (5k)

To a suspension of 1-chloroisoquinoline 4k (4.50 g, 27.5 mmol) and guanidine carbonate (14.9 g, 82.5 mmol) in NMP (45 mL) was added K₂CO₃ (19.0 g, 138 mmol), and the mixture was heated to 120 °C and stirred for 15 h. The mixture was filtered through a glass filter at approximately 100 °C, and the cake was washed with EtOAc (5 \times 90 mL). To the combined filtrate was added H₂O (45 mL) and the layers were separated. The organic layer was washed with 10% aqueous NaCl (3 \times 45 mL) and concentrated in vacuo. The residue was purified by column chromatography (NH silica gel, EtOAc/hexane 1:1). The obtained solids were suspended in H₂O (45 mL) and the slurry was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give the crude product (3.30 g). The crude product (700 mg) was suspended in EtOAc/hexane (1:2, 11 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 5k (0.627 g, 58%) as a white solid. mp 193–194 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 6.98 (d, J = 6.0 Hz, 1H), 7.11 (br s, 4H), 7.43 (ddd, J = 8.2, 6.9, 1.3 Hz, 1H), 7.58 (ddd, J = 7.6, 7.6, 1.3 Hz, 1H), 7.61–7.67 (m, 1H), 7.90 (d, J = 5.7 Hz, 1H), 8.62 (d, J = 8.5 Hz, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 111.3, 125.5, 125.6, 126.0, 126.8, 130.0, 137.3, 140.4, 158.9, 162.0; IR (ATR) 3382, 3046, 2179, 1586, 1519, 1489, 1443, 1390, 1363, 1315, 1277, 1211, 1139, 1089, 1020, 964, 950, 877, 810, 797, 757, 728, 682, 658, 584, 520, 463, 425, 404 cm⁻¹; HRMS-ESI (m/z): $[M + H]^+$ calcd for $C_{10}H_{11}N_4$, 187.0984; found, 187.0976.

4.3.12. 1-(Benzo[d]oxazol-2-yl)guanidine (5m)

To a suspension of 2-chlorobenzo[d]oxazole 4m (4.50 g, 29.3 mmol) and guanidine carbonate (5.28 g, 29.3 mmol) in NMP (90 mL) was added K₂CO₃ (12.1 g, 87.9 mmol), and the mixture was heated to 100 °C and stirred for 1 h. After cooling to room temperature, EtOAc (135 mL) and H₂O (225 mL) were added, and the layers were separated. The aqueous layer was extracted with EtOAc (3 \times 90 mL). The combined organic layer was washed with 10% aqueous NaCl (3×23 mL) and concentrated in vacuo. The residue was purified by column chromatography (NH silica gel, EtOAc/hexane 1:3 to 1:1). The obtained solids were suspended in H₂O (23 mL) and the mixture was stirred at 40 °C for 1 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered to give the crude product (3.83 g). The crude product (700 mg) was suspended in EtOAc/hexane (1:2, 11 mL), and the mixture was stirred at 50 °C for 1 h. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 5m as a white solid (535 mg, 57%). mp 185-186 °C (lit.²⁶ 186 °C); ¹H NMR (500 MHz, DMSO- d_6) δ 7.03 (dd, J =7.7, 1.1 Hz, 1H), 7.07–7.20 (m, 1H), 7.18 (br s, 4H), 7.20–7.30 (m, 1H), 7.30–7.40 (m, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 108.5, 115.3, 120.8, 123.0, 142.5, 146.6, 159.8, 166.4; IR (ATR) 3448, 3340, 3196, 3056, 1609, 1547, 1455, 1344, 1316, 1251, 1177, 1101, 1029, 1007, 962, 919, 755, 728, 504, 434, 416, 409 cm^{-1} ; HRMS-ESI (m/z): $[M + H]^+$ calcd for C₈H₉N₄O, 177.0776; found, 177.0768.

4.4. 2-Chloro-1-(5-nitropyridin-2-yl)guanidine (6a)

To a suspension of **5a** (544 mg, 3.00 mmol) in MeCN (27 mL) was added *N*-chlorosuccinimide (NCS, 441 mg, 3.30 mmol) at 40 °C, and the mixture was stirred at this temperature for 15 min. The mixture was cooled by ice bath and stirred for 30 min, and then filtered to give **6a** (584 mg, 90%) as a pale yellow solid. mp 154–155 °C; ¹H NMR (500 MHz, DMSO-*d*₆) δ 7.35 (d, *J* = 9.5 Hz, 1H), 7.58 (br s, 2H), 8.44 (dd, *J* = 9.5, 2.8 Hz, 1H), 9.05 (d, *J* = 2.5 Hz, 1H), 10.3 (s, 1H); ¹³C NMR (125 MHz, DMSO-*d*₆) δ 111.2, 133.7, 137.8, 144.5, 157.5, 157.9; IR (ATR) 3461, 3228, 3056, 1659, 1599, 1562, 1523, 1478, 1413, 1322, 1287, 1243, 1159, 1111, 1014, 978, 954, 889, 850, 833, 756, 723, 697, 609, 528, 499, 416 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₆H₇N₅O₂Cl, 216.0288; found, 216.0287.

4.5. Syntheses of [1,2,4]triazolo[1,5-a]pyridin-2-amines **3** and [1,2,4]triazolo[4,3-a]pyridin-3-amines **7**

4.5.1. 6-Nitro-[1,2,4]triazolo[1,5-a]pyridin-2amine (**3a**)

4.5.1.1. Synthesis from 6a

To a suspension of **6a** (431 mg, 2.00 mmol) in MeCN (22 mL) was added K₂CO₃ (581 mg, 4.20 mmol) in H₂O (9 mL) at 40 °C, and the mixture was stirred at this temperature for 30 min. After cooling to room temperature, H₂O (35 mL) was added, and the mixture was cooled by ice bath and stirred for 30 min, and then filtered to give **3a** (332 mg, 93%) as a yellow solid. mp 311–312 °C (lit.¹⁵ mp 310 °C); ¹H NMR (600 MHz, DMSO-*d*₆) δ 6.68 (s, 2H), 7.47 (d, *J* = 9.5 Hz, 1H), 8.16 (dd, *J* = 9.4, 2.3 Hz, 1H), 9.64 (d, *J* = 1.9 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 110.9, 123.8, 126.9, 135.1, 152.6, 168.7; IR (ATR) 3449, 3320, 3237, 3110, 3070, 3033, 1633, 1561, 1522, 1488, 1448, 1422, 1328, 1290, 1256, 1205, 1139, 1081, 1038, 946, 888, 831, 760, 744, 725, 697, 610, 582, 550, 489, 424 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₆H₆N₅O₂, 180.0521; found, 180.0514.

4.5.1.2. One-pot synthesis from 5a

To a suspension of **5a** (362 mg, 2.00 mmol) in MeOH (36 mL) was added NCS (294 mg, 2.20 mmol) at 40 °C, and the mixture was stirred at this temperature for 15 min. To the resulting slurry was added K_2CO_3 (581 mg, 4.20 mmol) in H_2O (7 mL) at 40 °C, and the mixture was stirred at this temperature for 30 min. After cooling to room temperature, H_2O (29 mL) was added, and the mixture was stirred for 30 min, and then filtered to give **3a** (312 mg, 87%) as a yellow solid.

4.5.2. 8-Nitro[1,2,4]triazolo[1,5-a]pyridin-2-amine (**3b**)

Compound **3b** was synthesized from **5b** (544 mg, 3.00 mmol) according to the procedure described for the one-pot synthesis of **3a**. Yellow solid, 461 mg, 86% yield. mp 350 °C (decompose); ¹H NMR (600 MHz, DMSO- d_6) δ 6.71 (s, 2H), 7.04 (dd, J = 7.5, 7.5 Hz, 1H), 8.41 (dd, J = 8.3, 1.1 Hz, 1H), 8.99 (dd, J = 6.4, 1.1 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 109.2, 127.2, 132.7, 133.4, 145.2, 167.7; IR (ATR) 3361, 3309, 3220, 3173, 3101, 1631, 1561, 1519, 1442, 1413, 1340, 1272, 1194, 1149, 1077, 1038, 984, 898, 841, 835, 795, 752, 738, 686, 596, 563, 543, 474, 452 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₆H₆N₅O₂, 180.0521; found, 180.0513.

4.5.3. 8-Chloro-[1,2,4]triazolo[1,5-a]pyridin-2amine (**3c**)

mp 225 °C; ¹H NMR (600 MHz, DMSO-*d*₆) δ 6.25 (s, 2H), 6.86 (dd, J = 7.7, 6.6 Hz, 1H), 7.60 (dd, J = 7.7, 0.9 Hz, 1H), 8.55 (dd, J = 6.8, 1.1 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 110.9, 116.9, 126.7, 128.1, 148.4, 166.1; IR (ATR) 3323, 3164, 1641, 1627, 1544, 1506, 1407, 1308, 1221, 1203, 1136, 1121, A062, 1029, 962, 907, 794, 779, 772, 749, 675, 635, 579, 561, 537, 408 cm⁻¹; HRMS (ESI): $[M+H^+]$ calcd for C₆H₆ClN₄, 169.0281; found, 169.0273.

4.5.4. 8-Chloro-[1,2,4]triazolo[4,3-a]pyridin-3amine (7c)

mp 270-271 °C (decompose) (lit.¹⁶ 275–276 °C); ¹H NMR (600 MHz, DMSO- d_6) δ 6.56 (s, 2H), 6.75 (dd, J = 7.0, 7.0 Hz, 1H), 7.28 (d, J = 7.1 Hz, 1H), 8.06 (dd, J = 6.8, 0.8 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 111.2, 120.0, 121.6, 124.0, 143.2, 149.8; IR (ATR) 3265, 3106, 1658, 1629, 1575, 1517, 1500, 1439, 1414, 1378, 1331, 1154, 1128, 1043, 1036, 943, 935, 882, 868, 781, 758, 727, 687, 651, 568, 557, 539, 510, 409, 404 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₆H₆ClN₄, 169.0281; found, 169.0273.

4.5.5. 6-Chloro-[1,2,4]triazolo[4,3-a]pyridin-3amine (7d)

Compound 5d (512 mg, 3.00 mmol) in MeOH (51 mL) was treated with NCS (441 mg, 3.30 mmol) and K₂CO₃ (871 mg, 6.3 mmol) in H₂O (10 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:1 to EtOAc) to give crude 7d (262 mg). Crude 7d (150 mg) was triturated with EtOAc/hexane (2:1, 2 mL) and filtered to give 7d (145 mg, 50%) as a pale yellow solid. mp 256 °C (decompose); ¹H NMR (600 MHz, DMSO- d_6) δ 6.48 (s, 2H), 7.08 (dd, J = 9.8, 1.9 Hz, 1H), 7.50 (dd, J = 9.8, 0.8 Hz, 1H), 8.34 (dd, J = 1.7, 0.9 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 116.5, 118.3, 120.1, 125.8, 144.1, 148.5; IR (ATR) 3248, 3094, 1664, 1630, 1578, 1509, 1447, 1391, 1359, 1337, 1165, 1137, 1034, 928, 851, 824, 817, 790, 734, 676, 572, 540, 416 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₆H₆ClN₄, 169.0281; found, 169.0279.

4.5.6. 8-(Trifluoromethyl)[1,2,4]triazolo[1,5a]pyridin-2-amine (3e)

Compound 5e (817 mg, 4.00 mmol) in MeOH (82 mL) was treated with NCS (588 mg, 4.40 mmol) and K₂CO₃ (1.16 g, 8.40 mmol) in H_2O (16 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:4 to EtOAc) to give crude 3e (473 mg) and crude 7e (189 mg). Crude 3e (200 mg) was purified by column chromatography (NH silica gel, EtOAc/hexane 1:4 to 1:1), and the obtained solids were triturated with EtOAc/hexane (1:1, 1 mL) and filtered to give 3e (151 mg, 44%) as a white solid. mp 221 °C; ¹H NMR (600 MHz, DMSO-*d*₆) δ 6.44 (s, 2H), 7.00 (t, J = 7.2 Hz, 1H), 7.84 (d, J = 7.6 Hz, 1H), 8.84 (d, J = 6.8 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 109.9, 112.6 (q, ² J_{CF} = 33.0 Hz), 122.9 (q, ${}^{1}J_{CF}$ = 270.0 Hz), 127.2 (q, ${}^{3}J_{CF}$ = 4.5 Hz), 131.2, 146.8, 166.8; IR (ATR) 3342, 3209, 1640, 1582, 1524, 1509, 1460, 1421, 1352, 1330, 1303, 1226, 1199, 1166, 1112, 1068, 1028, 959, 929, 906, 807, 794, 762, 717, 667, 593, 526, 507 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₇H₆F₃N₄, 203.0545; found, 203.0540.

4.5.7. 8-(Trifluoromethyl)[1,2,4]triazolo[4,3a]pyridin-3-amine (7e)

Crude **7e** (100 mg) was triturated with EtOAc/hexane (1:1, 1 mL) and filtered to give **7e** (87 mg, 20%) as a yellow solid. mp 258 °C (decompose); ¹H NMR (600 MHz, DMSO- d_6) δ 6.64 (s, 2H), 6.88 (t, J = 7.0 Hz, 1H), 7.58 (d, J = 6.8 Hz, 1H), 8.30 (d, J = 7.2 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 109.9, 115.9 (q, ² J_{CF} = 33.0 Hz), 122.5 (q, ¹ J_{CF} = 270.0 Hz), 125.0 (q, ³ J_{CF} = 6.0 Hz), 126.6, 140.6, 149.2; IR (ATR) 3304, 3153, 1661, 1637, 1585, 1558, 1508, 1450, 1424, 1390, 1338, 1313, 1240, 1163,

1139, 1114, 1046, 1031, 946, 872, 773, 742, 733, 690, 642, 611, M A24.8 (q, ²J_{CF} = 33.0 Hz), 143.9, 149.4; IR (ATR) 3272, 3081

564, 520, 504 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₇H₆F₃N₄, 203.0545; found, 203.0540.

4.5.8. 6-(Trifluoromethyl)[1,2,4]triazolo[1,5a]pyridin-2-amine (**3f**)

Compound 5f (1.17 g, 4.00 mmol) in MeOH (117 mL) was treated with NCS (588 mg, 4.40 mmol) and K₂CO₃ (1.16 g, 8.40 mmol) in H₂O (23 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:1 to 4:1) to give 3f (184 mg, 23%) as a white solid and crude 7f (454 mg). mp 182 °C; ¹H NMR (600 MHz, DMSO- d_6) δ 6.37 (s, 2H), 7.52 (d, J = 9.4 Hz, 1H), 7.68 (dd, J = 9.1, 1.9 Hz, 1H), 9.18 (s, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 112.8, 113.4 (q, ${}^2J_{CF} = 33.5$ Hz), 123.7 (q, ${}^1J_{CF} =$ 269.0 Hz), 124.9 (q, ${}^{3}J_{CF} = 3.0$ Hz), 126.6 (q, ${}^{3}J_{CF} = 4.5$ Hz), 151.4, 167.5; IR (ATR) 3343, 3172, 2360, 2341, 1648, 1564, 1543, 1522, 1428, 1358, 1331, 1307, 1181, 1120, 1058, 1034, 938, 864, 809, 761, 740, 674, 639, 562, 537, 438, 415 cm^{-1} ; HRMS (ESI): $[M+H^+]$ calcd for $C_7H_6F_3N_4$, 203.0545; found, 203.0538.

4.5.9. 6-(Trifluoromethyl)[1,2,4]triazolo[4,3a]pyridin-3-amine (**7f**)

Crude **7f** (200 mg) was dissolved in EtOAc and washed with 5% aqueous K₂CO₃. After evaporation of the solvent, the obtained solids were triturated with EtOAc/hexane (1:1, 2 mL) and filtered to give **7f** (184 mg, 52%) as a yellow solid. mp 249 °C (decompose); ¹H NMR (600 MHz, DMSO-*d*₆) δ 6.72 (s, 2H), 7.23 (dd, *J* = 9.8, 1.5 Hz, 1H), 7.63 (d, *J* = 9.8 Hz, 1H), 8.77 (d, *J* = 1.5 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 114.0 (q, ²*J*_{CF} = 33.5 Hz), 117.0, 119.8 (q, ³*J*_{CF} = 3.0 Hz), 123.2 (q, ³*J*_{CF} = 6.0 Hz), 123.6 (q, ¹*J*_{CF} = 269.0 Hz), 144.7, 149.5; IR (ATR) 3257, 3105, 1666, 1650, 1588, 1531, 1450, 1405, 1376, 1322, 1170, 1119, 1031, 930, 889, 833, 801, 777, 743, 651, 633, 584, 566, 542, 522, 425, 410 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₇H₆F₃N₄, 203.0545; found, 203.0539.

4.5.10. 7-(Trifluoromethyl)[1,2,4]triazolo[1,5a]pyridin-2-amine (**3g**)

Compound **5g** (817 mg, 4.00 mmol) in MeOH (82 mL) was treated with NCS (588 mg, 4.40 mmol) and K_2CO_3 (1.16 g, 8.40 mmol) in H_2O (16 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:4 to 10:1) to give **3g** (131 mg, 16%) as a pale yellow solid and crude **7g** (496 mg). mp 167–168 °C; ¹H NMR (600 MHz, DMSO- d_6) δ 6.35 (s, 2H), 7.15 (dd, J = 7.0, 2.1 Hz, 1H), 7.82 (s, 1H), 8.78 (d, J = 7.2, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 106.7 (q, ³ $J_{CF} = 3.0$ Hz), 109.7 (q, ³ $J_{CF} = 3.8$ Hz), 123.3 (q, ¹ $J_{CF} = 270.8$ Hz), 128.7 (q, ² $J_{CF} = 33.0$ Hz), 128.7, 149.5, 167.4; IR (ATR) 3327, 3166, 1650, 1556, 1523, 1509, 1483, 1326, 1272, 1256, 1211, 1180, 1119, 1057, 1035, 951, 860, 789, 758, 740, 673, 582, 568, 544, 443, 413 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₇H₆F₃N₄, 203.0545; found, 203.0539.

4.5.11. 7-(Trifluoromethyl)[1,2,4]triazolo[4,3a]pyridin-3-amine (**7g**)

Crude **7g** (200 mg) was dissolved in EtOAc and washed with 10% aqueous NH₄Cl and H₂O. After evaporation of the solvent, the obtained solids were triturated with EtOAc/hexane (1:1, 4 mL) and filtered to give **7g** (162 mg, 50%) as a white solid. mp 275 °C (decompose); ¹H NMR (600 MHz, DMSO-*d*₆) δ 6.71 (s, 2H), 6.99 (dd, *J* = 7.6, 1.5 Hz, 1H), 7.99 (s, 1H), 8.25 (d, *J* = 7.6 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 106.6 (q, ³*J*_{CF} = 3.0 Hz), 114.6 (q, ³*J*_{CF} = 5.5 Hz), 123.3 (q, ¹*J*_{CF} = 270.5 Hz), 124.1,

1651, 1572, 1509, 1481, 1428, 1371, 1336, 1293, 1263, 1145, 1050, 1035, 937, 887, 826, 793, 779, 741, 731, 688, 674, 639, 606, 576, 526, 459, 433, 407 cm⁻¹; HRMS (ESI): $[M+H^+]$ calcd for $C_7H_6F_3N_4$, 203.0545; found, 203.0540.

4.5.12. 8-Chloro-6-

(trifluoromethyl)[1,2,4]triazolo[1,5-a]pyridin-2amine (**3i**)

Compound 5i (716 mg, 3.00 mmol) in MeOH (72 mL) was treated with NCS (441 mg, 3.30 mmol) and K₂CO₃ (871 mg, 6.30 mmol) in H₂O (14 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:5 to 5:1) to give crude 3i (420 mg) and crude 7i (129 mg). Crude 3i (200 mg) was triturated with EtOAc/hexane (1:1, 2 mL) and filtered to give 3i (193 mg, 57%) as a white solid. mp 238-239 °C; ¹H NMR (600 MHz, DMSO d_6) δ 6.62 (s, 2H), 7.99 (d, J = 1.5 Hz, 1H), 9.22 (s, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 113.3 (q, ${}^2J_{CF}$ = 34.5 Hz), 117.5, 123.1 (q, ${}^{1}J_{CF} = 268.5$ Hz), 124.2 (q, ${}^{3}J_{CF} = 1.5$ Hz), 125.9 (q, ${}^{3}J_{CF}$ = 4.5 Hz), 149.5, 167.3; IR (ATR) 3379, 3299, 3165, 3111, 1627, 1542, 1401, 1351, 1321, 1306, 1219, 1170, 1116, 1074, 1031, 975, 920, 883, 869, 841, 758, 691, 652, 628, 551, 532, 502, 449, 410 cm⁻¹; HRMS (ESI): $[M+H^+]$ calcd for $C_7H_5ClF_3N_4$, 237.0155; found, 237.0148.

4.5.13. 8-Chloro-6-

(trifluoromethyl)[1,2,4]triazolo[4,3-a]pyridin-3amine (7i)

Crude **7i** (129 mg) was triturated with EtOAc/hexane (1:1, 1 mL) and filtered to give **7i** (72 mg, 9%) as a white solid. mp 266 °C (decompose); ¹H NMR (600 MHz, DMSO- d_6) δ 6.89 (s, 2H), 7.58 (d, J = 1.1 Hz, 1H), 8.79 (s, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 113.9 (q, ² $J_{CF} = 34.0$ Hz), 119.0 (q, ³ $J_{CF} = 2.7$ Hz), 121.7, 122.6 (q, ³ $J_{CF} = 6.0$ Hz), 123.0 (q, ¹ $J_{CF} = 269.5$ Hz), 142.5, 150.9; IR (ATR) 3472, 3297, 3228, 3104, 2998, 1647, 1572, 1542, 1518, 1441, 1377, 1345, 1312, 1234, 1163, 1115, 1069, 1037, 948, 905, 884, 873, 865, 831, 754, 732, 685, 651, 630, 550, 446, 424 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₇H₅ClF₃N₄, 237.0155; found, 237.0151.

4.5.14. [1,2,4]Triazolo[1,5-a]quinolin-2-amine (3j)

Compound 5j (745 mg, 4.00 mmol) in MeOH (75 mL) was treated with NCS (588 mg, 4.40 mmol) and K₂CO₃ (1.16 g, 8.40 mmol) in H₂O (15 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:5 to EtOAc/MeOH 99:5) to give crude 3j (168 mg) and crude 7j (370 mg). Crude 3j (130 mg) was purified by column chromatography (NH silica gel, EtOAc/hexane 1:2), and the obtained solids were triturated with EtOAc/hexane (1:3, 3 mL) and filtered to give 3j (115 mg, 20%) as a pale pink solid. mp 196 °C; ¹H NMR (600 MHz, DMSO-d₆) δ 6.09 (s, 2H), 7.45–7.54 (m, 2H), 7.78 (t, J = 7.9 Hz, 1H), 7.94 (d, J = 9.4 Hz, 1H), 8.02 (d, J = 7.6 Hz, 1H), 8.15 (d, J = 8.3 Hz,1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 113.2, 114.5, 121.9, 124.0, 129.0, 129.7, 130.2, 132.5, 148.5, 165.6; IR (ATR) 3328, 3178, 1651, 1611, 1561, 1530, 1449, 1418, 1392, 1355, 1330, 1215, 1092, 1054, 810, 767, 756, 745, 691, 643, 615, 554, 522, 508, 477, 424, 409 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₁₀H₉N₄, 185.0827; found, 185.0820.

4.5.15. [1,2,4]Triazolo[4,3-a]quinolin-1-amine (7j)

Crude **7j** (300 mg) was purified by column chromatography (NH silica gel, EtOAc), and the obtained solids were triturated with EtOAc/hexane (1:1, 3 mL) and filtered to give **7j** (231 mg,

39%) as a pink solid. mp 255–256 (lit.²⁷ mp 475–177 °C, lit.²⁸ mp 250–253 °C); ¹H NMR (600 MHz, DMSO- d_6) δ 6.31 (br s, 2H), 7.40 (d, J = 9.4 Hz, 1H), 7.45–7.55 (m, 2H), 7.65 (t, J = 7.6 Hz, 1H), 7.87 (d, J = 7.6 Hz, 1H), 8.52 (d, J = 8.3 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 115.1, 115.9, 124.0, 125.3, 127.4, 128.5, 128.6, 131.9, 146.1, 151.5; IR (ATR) 3292, 3126, 1654, 1612, 1567, 1546, 1539, 1471, 1449, 1440, 1390, 1341, 1329, 1278, 1214, 1171, 1150, 1126, 1065, 1039, 1008, 968, 930, 879, 854, 803, 754, 739, 691, 680, 609, 565, 536, 509, 440, 418 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₁₀H₉N₄, 185.0827; found, 185.0820.

4.5.16. [1,2,4]Triazolo[5,1-a]isoquinolin-2-amine (3k)

Compound 5k (745 mg, 4.00 mmol) in MeOH (75 mL) was treated with NCS (588 mg, 4.40 mmol) and K₂CO₃ (1.16 g, 8.40 mmol) in H₂O (15 mL), and work-up was conducted according to the general procedure. The residue obtained after evaporation of the organic layer was purified by column chromatography (NH silica gel, EtOAc/hexane 1:2 to EtOAc/MeOH 98:2) to give crude **3k** (361 mg) and crude **7k** (141 mg). Crude **3k** (300 mg) was purified by column chromatography (NH silica gel, EtOAc/hexane 1:2), and the obtained solids were triturated with EtOAc/hexane (2:1, 9 mL) and filtered to give 3k (212 mg, 35%) as an orange solid. mp 193–195 °C; ¹H NMR (600 MHz, DMSO d_6) δ 6.03 (s, 2H), 7.27 (d, J = 7.2 Hz, 1H), 7.70 (td, J = 7.5, 1.3 Hz, 2H), 7.95 (d, J = 7.5 Hz, 1H), 8.29 (d, J = 7.7 Hz, 1H), 8.37 (d, J = 7.2 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 110.7, 120.1, 123.1, 124.6, 127.3, 127.7, 129.2, 131.0, 148.1, 165.3; IR (ATR) 3320, 3175, 1638, 1556, 1537, 1523, 1486, 1434, 1400, 1370, 1330, 1298, 1258, 1137, 1095, 1058, 904, 823, 778, 747, 704, 682, 659, 613, 570, 548, 509, 416 cm⁻¹; HRMS (ESI): $[M+H^+]$ calcd for C₁₀H₉N₄, 185.0827; found, 185.0820.

4.5.17. [1,2,4]Triazolo[3,4-a]isoquinolin-3-amine (7k)

Crude **7k** (141 mg) was purified by column chromatography (NH silica gel, EtOAc/hexane 1:1 to EtOAc), and the obtained solids were triturated with EtOAc/hexane (1:1, 2 mL) and filtered to give **7k** (98 mg, 13%) as an orange solid. mp 263 °C (decompose); ¹H NMR (600 MHz, DMSO- d_6) δ 6.47 (s, 2H), 7.07 (d, J = 7.2 Hz, 1H), 7.63 (dt, J = 7.7 Hz, 2H), 7.79 (d, J = 7.0 Hz, 1H), 7.90 (d, J = 7.6 Hz, 1H), 8.31 (d, J = 7.1 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 112.1, 119.8, 121.6, 122.0, 127.3, 128.5, 129.0 (2C), 143.7, 149.9; IR (ATR) 3311, 3139, 1644, 1567, 1557, 1525, 1480, 1459, 1437, 1370, 1325, 1303, 1132, 1044, 982, 926, 896, 783, 769, 750, 741, 708, 700, 679, 636, 560, 507, 484, 471, 462, 443, 432, 421, 405 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₁₀H₉N₄, 185.0827; found, 185.0821.

4.6. Synthesis of **3a** from 5-nitropyridin-2-amine

To a stirred solution of 5-nitropyridin-2-amine (1.00 g, 7.19 mmol) in acetone (10 mL) was added ethoxycarbonyl isothiocyanate (1.41 g, 10.8 mmol), and the mixture was stirred at 50 °C for 22 h. After cooling to room temperature, H₂O (10 mL) was slowly added to the mixture. The slurry was stirred at room temperature for 1 h and then filtered to give ethyl [(5-nitropyridin-2-yl)carbamothioyl]carbamate (1.46 g, 75%) as a brown solid. mp 178–179 °C; ¹H NMR (600 MHz, DMSO-*d*₆) δ 1.28 (t, *J* = 7.2 Hz, 3H), 4.25 (q, *J* = 7.2 Hz, 2H), 8.69 (dd, *J* = 9.2, 3.0 Hz, 1H), 8.90 (d, *J* = 9.4 Hz, 1H), 9.22 (d, *J* = 2.6 Hz, 1H), 12.0 (br s, 1H), 12.5 (br s, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 14.0, 62.5, 113.9, 134.2, 140.8, 144.6, 153.3, 155.2, 177.9; IR (ATR) 3162, 2994, 1720, 1579, 1529, 1508, 1455, 1355, 1311, 1243, 1192, 1143, 1110, 1038, 1004, 949, 870, 840, 799, 757, 734, 714, 700, 628, 596, 525, 501, 436, 412 cm⁻¹;

77 °C, lit.²⁸ M HRMS (ESI): $[M+H^+]$ calcd for C₉H₁₁N₄O₄S, 271.0501; found, 6.31 (br s, 271.0497.

0497.

To a stirred solution of ethyl [(5-nitropyridin-2yl)carbamothioyl]carbamate (1.00 g, 3.70 mmol) in EtOH (10 mL) were added *N*,*N*-diisopropylethylamine (1.43 g, 11.1 mmol) and hydroxylamine hydrochloride (1.29 g, 18.5 mmol), and the mixture was stirred at 50 °C for 1.5 h. After cooling to room temperature, H₂O (10 mL) was slowly added to the mixture. The slurry was stirred at room temperature for 1 h and then filtered. The wet solids were suspended in EtOAc/*n*-heptane (1:1, 10 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give **3a** (587 mg, 89%) as a yellow solid.

4.7. Synthesis of 3c from 3-chloropyridin-2-amine 1c

4.7.1. Ethyl [(3-Chloropyridin-2yl)carbamothioyl]carbamate (2c)

To a stirred solution of 3-chloropyridin-2-amine 1c (800 mg, 6.22 mmol) in acetone (8 mL) was added ethoxycarbonyl isothiocyanate (1.22 g, 9.33 mmol), and the mixture was stirred at 50 °C for 1 h. After cooling to room temperature, H₂O (16 mL) was slowly added, and the mixture was stirred at 40 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 2c (1.39 g, 86%) as a white solid. mp 126–128 °C; ¹H NMR (600 MHz, DMSO- d_6) δ 1.27 (t, J = 7.2 Hz, 3H), 4.23 (q, J = 7.2 Hz, 2H), 7.42 (dd, J = 8.1, 4.7 Hz, 1H), 8.04 (dd, J = 7.9, 1.5 Hz, 1H), 8.45 (dd, J = 4.5, 1.5 Hz, 1H), 11.4–11.5 (m, 2H); ¹³C NMR (150 MHz, DMSO- d_6) δ 14.1, 62.1, 124.4, 128.5, 138.7, 147.1, 148.5, 153.4, 180.0; IR (ATR) 3152, 2989, 1723, 1578, 1510, 1474, 1447, 1423, 1370, 1321, 1266, 1235, 1171, 1137, 1068, 1041, 1031, 870, 795, 769, 753, 736, 708, 680, 647, 598, 551, 534, 418 cm⁻¹; HRMS (ESI): $[M+H^+]$ calcd for C₉H₁₁ClN₃O₂S, 260.0261; found, 260.0258.

4.7.2. Synthesis of 3c from 2c

To a stirred solution of 2c (1.00 g, 3.85 mmol) in EtOH (10 mL) were added *N*,*N*-diisopropylethylamine (1.49 g, 11.6 mmol) and hydroxylamine hydrochloride (1.34 g, 19.3 mmol), and the mixture was stirred at 50 °C for 2 h. After cooling to room temperature, H₂O (10 mL) was slowly added to the mixture. The slurry was stirred at room temperature for 1 h and then filtered. The wet solids were suspended in EtOAc/*n*-heptane (1:1, 10 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 3c (443 mg, 68%) as a white solid.

4.8. Synthesis of **7c** from 2,3-dichloropyridine **4c**

4.8.1. 3-Chloro-2-hydrazinylpyridine $(8)^{29}$

To a stirred solution of **4c** (5.00 g, 33.8 mmol) in EtOH (25 mL) was added hydrazine monohydrate (6.77 g, 135 mmol), and the mixture was heated to reflux for 20 h. At this point, additional hydrazine monohydrate (1.69 g, 33.8 mmol) was added, and the mixture was heated to reflux for 20 h. The mixture was cooled by ice bath and stirred for 1 h, and then filtered to give **8** (4.40 g, 91%) as a white solid. mp 164–166 °C (lit.²⁵ mp 163–164 °C); ¹H NMR (600 MHz, DMSO-*d*₆) δ 4.22 (br s, 2H), 6.61 (dd, *J* = 7.6, 4.9 Hz, 1H), 7.52–7.64 (m, 2H), 8.05 (dd, *J* = 4.7, 1.3 Hz, 1H); ¹³C NMR (150 MHz, DMSO-*d*₆) δ 113.2, 113.6, 136.2, 145.7, 155.7; IR (ATR) 3284, 3195, 1592, 1494, 1455, 1412, 1343, 1264, 1122, 1033, 991, 968, 957, 930, 787, 761, 750, 726, 634, 610, 543, 497, 438, 423, 416 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₅H₇ClN₃, 144.0328; found, 144.0323.

4.8.2. 2-(3-Chloropyridin-2-yl)-N-(2,4,4trimethylpentan-2-yl)hydrazinecarbothioamide (**9**)

To a stirred solution of 8 (3.00 g, 20.9 mmol) in THF (150 mL) was added 2-isothiocyanato-2,4,4-trimethylpentane (4.30 g, 25.1 mmol), and the mixture was stirred at room temperature for 14 h. The mixture was concentrated in vacuo, and the residue was purified by column chromatography (EtOAc/hexane 1:5) to give the crude product. The crude product was suspended in hexane (35 mL), and the mixture was stirred at 50 $^{\circ}\mathrm{C}$ for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 9 (5.84 g, 89%) as a white solid. mp $105-106 \,^{\circ}C$; ¹H NMR (600 MHz,CDCl₃) δ 0.93 (s, 9H), 1.60 (s, 6H), 1.91 (s, 2H), 6.86 (m, 2H), 7.11 (br s, 1H), 7.52-7.63 (m, 2H), 8.18 (d, J = 4.5 Hz, 1H); ¹³C NMR (150 MHz,CDCl₃) δ 28.9 (2C), 31.4 (3C), 31.5, 52.3, 57.6, 116.3, 118.1, 137.3, 146.5, 152.5, 181.9; IR (ATR) 3360, 3332, 3208, 2954, 1586, 1567, 1533, 1503, 1472, 1454, 1402, 1387, 1365, 1323, 1266, 1225, 1156, 1127, 1101, 1065, 1034, 1005, 979, 791, 754, 735, 668, 596, 553, 441, 412, 401 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₁₄H₂₄ClN₄S, 315.1410; found, 315.1405.

4.8.3. 8-Chloro-N-(2,4,4-trimethylpentan-2-yl)-[1,2,4]triazolo[4,3-a]pyridin-3-amine (10)

To a stirred solution of 9 (3.80 g, 12.1 mmol) in THF (114 mL) were added triethylamine (4.01 g, 29.0 mmol) and 2-chloro-1-methylpyridinium iodide (4.01 g, 15.7 mmol), and the mixture was stirred at room temperature for 5 h. To the mixture were added EtOAc (228 mL) and H₂O (38 mL), and the layers were separated. The organic layer was washed with 5% aqueous NaHCO₃ (2 \times 38 mL) and H₂O (2 \times 38 mL). To the organic layer was added activated carbon (380 mg), and the mixture was stirred at room temperature for 30 min. The mixture was filtered and the filtrate was concentrated in vacuo. The residue was purified by column chromatography twice (EtOAc/hexane 1:10 to 1:1, then EtOAc/hexane 1:3) to give the crude product. The crude product was suspended in EtOAc/n-hexane (1:5, 114 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give **10** (1.72 g, 51%) as a pale brown solid. mp 182–183 °C; ¹H NMR (600 MHz, DMSO-*d*₆) δ 0.92 (s, 9H), 1.51 (s, 6H), 1.93 (s, 2H), 6.22 (s, 1H), 6.73 (t, *J* = 7.0 Hz, 1H), 7.29 (d, *J* = 6.8 Hz, 1H), 8.26 (d, J = 7.2 Hz, 1H); ¹³C NMR (150 MHz, DMSO- d_6) δ 29.4 (3C), 31.1 (2C), 31.4, 49.9, 55.3, 110.9, 120.0, 121.8, 124.3, 142.9, 148.3; IR (ATR) 3360, 3332, 3209, 2953, 1586, 1567, 1533, 1503, 1471, 1454, 1402, 1387, 1365, 1323, 1266, 1225, 1156, 1127, 1101, 1065, 1033, 1005, 979, 791, 754, 735, 667, 593, 553, 442, 413 cm⁻¹; HRMS (ESI): [M+H⁺] calcd for C₁₄H₂₂ClN₄, 281.1533; found, 281.1530.

4.8.4. Synthesis of 7c from 10

To a stirred solution of 10 (500 mg, 1.78 mmol) in MeOH (2.5 mL) was added 6M HCl (2.5 mL), and the mixture was stirred at 50 °C for 5 h. At this point, MeOH (2.5 mL) and 6M HCl (2.5 mL) were added, and the mixture was stirred at 50 °C for 2 h. Then additional MeOH (2.5 mL) and 6M HCl (2.5 mL) were added, and the mixture was stirred at 50 °C for 1.5 h. The mixture was concentrated in vacuo. To the residue were added EtOAc (100 mL) and 5% aqueous NaHCO₃ (50 mL), and the layers were separated. The aqueous phase was extracted with EtOAc (2×100 mL). The combined organic layer was washed with 5% aqueous NaHCO₃ (2×20 mL) and 10% aqueous NaCl (20 mL) and concentrated in vacuo. The resulting residue was suspended in EtOAc/n-hexane (1:1, 10 mL), and the mixture was stirred at 50 °C for 30 min. The mixture was cooled to room temperature and stirred for 1 h, and then filtered to give 7c (97.0 mg, 32%) as a white solid.

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Supplementary data

Copies of ¹H NMR and ¹³C NMR data for all products in this article, HMBC spectrum of **6a**, HMBC and NOESY spectra of **3a**, **3b**, **3c**, **7c**, **7d**, **3e**, **7e**, **3f**, **7f**, **3g**, **7g**, **3i**, **7i**, **3j**, **7j**, **3k** and **7k**. Supplementary data associated with this article can be found in the online version, at

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Tetrahedron

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Supporting Information

Oxidative Cyclization of 1-(Pyridin-2-yl)guanidine Derivatives: A Synthesis of [1,2,4]Triazolo[1,5-*a*]pyridin-2-amines and An Unexpected Synthesis of [1,2,4]Triazolo[4,3-*a*]pyridin-3-amines

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1-(5-nitropyridin-2-yl)guanidine (5a)



1-(3-nitropyridin-2-yl)guanidine (5b)



1-(3-chloropyridin-2-yl)guanidine (5c)



1-(5-chloropyridin-2-yl)guanidine (5d)



1-(3-(trifluoromethyl)pyridin-2-yl)guanidine (5e)



1.4 = 5f proton.esp 1.3 M04(br. s.) M05(s) 1.2 -2.70 1.1 1.0 F₃(NH 0.9 0.9 NH₂ С N´ H 0.8-0.7-0.6-M02(m) M06(t) 0.5 M08(br. s.) M03(t) 3.30 M10(br. s.) 0.4 M07(d) œ -8.35 M09(dd) 3.29 2 -2.51 6.67 6.65 .92 -1.90 0.3--2.16 M01(s) 7.68 7.67 7.66 7.66 80 0.2 0.00 5 5 0.1 0 **1**.00 =2.44 C 1.75 **1**.12 94 4 -------Chemical Shift (ppm) 11 8 6 ידי 3 10 9 Т 5 13C NMR spectrum -39.86 -39.69 -39.52 -39.36 -39.36 5f carbon.esp 17.19 0.35 -30.07 118.34 -48.45 0.30 -28.94 158.79 0.25 0.20--40.03 39.02 -133.12 -143.52 0.15 166.06 173.72 ~128.16 ~126.01 ~123.86 0.10--114.01 113.76 143.48 113.50 0.05 0 Chemical Shift (ppm) 80 200 180 160 140 100 11 60 40 120

1-(5-(trifluoromethyl)pyridin-2-yl)guanidine 0.9 1-methylpyrrolidin-2-one solvate (**5f**) <u>¹H NMR spectrum</u>

1-(4-(trifluoromethyl)pyridin-2-yl)guanidine (5g)



1-(6-fluoropyridin-2-yl)guanidine (5h)



1-(3-chloro-5-(trifluoromethyl)pyridin-2-yl)guanidine (5i)





<u>¹H NMR spectrum</u>



1-(isoquinolin-1-yl)guanidine (5k)



1-(benzo[d]oxazol-2-yl)guanidine (5m)



2-chloro-1-(5-nitropyridin-2-yl)guanidine (6a)



HMBC spectrum



6-nitro-[1,2,4]triazolo[1,5-a]pyridin-2-amine (3a)





ACCEPTED MANUSCRIPT

HMBC spectrum



ACCEPTED MANUSCRIPT

NOESY spectrum



8-nitro[1,2,4]triazolo[1,5-a]pyridin-2-amine (3b)



<u>¹H NMR spectrum</u>

HMBC spectrum




8-chloro-[1,2,4]triazolo[1,5-a]pyridin-2-amine (3c)

<u>¹H NMR spectrum</u>







8-chloro-[1,2,4]triazolo[4,3-a]pyridin-3-amine (7c)

1H NMR spectrum

















8-(trifluoromethyl)[1,2,4]triazolo[1,5-a]pyridin-2-amine (**3e**)



<u>¹H NMR spectrum</u>





8-(trifluoromethyl)[1,2,4]triazolo[4,3-a]pyridin-3-amine (7e)

1H NMR spectrum







$6\-(trifluoromethyl)[1,2,4]triazolo[1,5\-a]pyridin\-2\-amine\(\mathbf{3f})$

¹H NMR spectrum

1.4-33f proton.esp 1.3 F_3C $-NH_2$ 1.2 M04(s) -6.37 1.1 1.0 0.9-0.8 M06(dd) M07(s) 0.7-M05(d) -9.18 M01(s) M03(s) -7.53 M02(m) 0.6 -0.00 -3.33 -7.68 -7.52 0.5 -2.51 7.69 5 0.4 0 0.3-0.2 0.1-0. 20 7 7 7 1 1 6. 6. 6. 6. 8 T Chemical Shift (ppm) 9 8 6 4 10 3 2 5 13C NMR spectrum -40.02 -39.88 -39.74 -39.60 -39.46 3f carbon.esp 112.76 0.13-0.12 0.11 0.10 167.53 0.09 0.08 0.07 151.43 0.06 /−126.58 ___124.85 0.05 0.04 <u>_124.63</u> 0.03 M01(m) 0.02 0.06 0.01 0 200 180 160 140 80 Chemical Shift (ppm) 120 100 40 60





6-(trifluoromethyl)[1,2,4]triazolo[4,3-a]pyridin-3-amine (7f)



¹H NMR spectrum





7-(trifluoromethyl)[1,2,4]triazolo[1,5-a]pyridin-2-amine (**3g**)



S43





7-(trifluoromethyl)[1,2,4]triazolo[4,3-a]pyridin-3-amine (7g)

1H NMR spectrum







8-chloro-6-(trifluorom	ethyl)[1,2,4]triazol	lo[1,5 - a]pyridin	-2-amine (3i
------------------------	----------------------	---------------------------	----------------------



¹H NMR spectrum





8-chloro-6-(trifluorome	thyl)[1,2,4]triazolo	o[4,3-a]pyridin-3	<i>}-amine</i> (7i)
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[1,2,4]triazolo[1,5-a]quinolin-2-amine (**3**j)

1H NMR spectrum






[1,2,4]triazolo[4,3-a]quinolin-1-amine (7j)

1H NMR spectrum



HMBC spectrum





[1,2,4]triazolo[5,1-a]isoquinolin-2-amine (3k)



HMBC spectrum





[1,2,4]triazolo[3,4-a]isoquinolin-3-amine (7k)



ACCEPTED MANUSCRIPT

HMBC spectrum



ACCEPTED MANUSCRIPT



Ethyl [(5-Nitropyridin-2-yl)carbamothioyl]carbamate







3-Chloro-2-hydrazinylpyridine (8)

1H NMR spectrum





 $\label{eq:2-(3-Chloropyridin-2-yl)-N-(2,4,4-trimethylpentan-2-yl)-hydrazine carbothio a mide (\textbf{9})$



8-Chloro-N-(2,4,4-trimethylpentan-2-yl)-[1,2,4]triazolo[4,3-a]pyridin-3-amine (10)