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Lignans from Bark of Fraxinus mandshurica var. japonica and F. japonica

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Three new lignans, (+)-fraxiresinol [(1S,2R,5R,6S)-1-hydroxy-2-(3,5-dimethoxy-4-hydroxyphenyl)-6-(4-hydroxy-3-methoxyphenyl)-3,7-dioxabicyclo[3.3.0]octane] (3), (+)-1-hydroxysyringaresinol (4) and (+)-1-hydroxypinoresinol-4'- β -D-glucoside (8), and five known lignans, (+)-pinoresinol (1), (+)-1-hydroxypinoresinol (2), (-)-olivil (5), (+)-cyclo-olivil (6) and (+)-pinoresinol- β -D-glucoside (7), were isolated from the bark of *Fraxinus mandshurica* Rupr. var. *japonica* Maxim and the bark of *F. japonica* Blume (Oleaceae).

Their structures were elucidated on the basis of spectroscopic analysis and chemical evidence.

Keywords—Fraxinus mandshurica var. japonica; Fraxinus japonica; Oleaceae; lignan; (+)-fraxiresinol; (+)-1-hydroxysyringaresinol; (+)-1-hydroxypinoresinol; (+)-pinoresinol; (-)-olivil; (+)-cyclo-olivil

Fraxinus mandshurica RUPR. var. japonica MAXIM (Oleaceae) is widely distributed in Hokkaido, the northern part of Japan. In China, the dried bark of F. mandshurica is sometimes used as a substitute for the Chinese crude drug "qin pi (Cortex Fraxini, 秦皮)". Further, the dried bark of F. japonica BLUME, which is on the market as "shinpi (秦皮)" in Japan, has been used since olden times as a diuretic, an antifebrile, an analgesic and an antirheumatic. ²⁾

Various coumarins, *i.e.* fraxinol, fraxetin, fraxin and mandshurin from F. mandshurica var. $japonica^{3a,b)}$ and fraxetin, esculetin, fraxin and esculin from F. $japonica,^{4a-c)}$ have been isolated from these barks so far. Among them, esculin and esculetin are known to be active principles of "shinpi".⁵⁾ Recently, compounds which show anti-inflammatory and antiplatelet-aggregating activities were also reported from the bark of F. $japonica.^{6)}$

Thus, our interest has been directed to the investigation of the constituents of these *Fraxinus* barks. This paper describes the isolation of three new lignans, 3, 4 and 8, along with five known lignans, (+)-pinoresinol (1), (+)-1-hydroxypinoresinol (2), (-)-olivil (5), (+)-cyclo-olivil (6) and (+)-pinoresinol- β -D-glucoside (7), from the bark of F. mandshurica var. japonica and the bark of F. japonica, and their structure elucidation on the basis of spectroscopic analysis and chemical evidence. The extraction and separation were carried out as described in Experimental.

Lignans 1 and 2 were identified as (+)-pinoresinol and (+)-1-hydroxypinoresinol, respectively, by direct comparison with authentic samples.^{7,8)}

The lignan 3 was obtained as an amorphous powder, $C_{21}H_{24}O_8$, $[\alpha]_D^{25} + 29.3^{\circ}$ (chloroform). The infrared (IR) spectrum of 3 suggested the presence of aromatic rings (1615 and 1510 cm⁻¹). The ultraviolet (UV) spectrum of 3 showed absorption maxima at 232.3 and 279.8 nm. The bathochromic shift of the absorption maxima in the presence of base was very similar to that of (+)-medioresinol (3c).⁹⁾ The proton nuclear magnetic resonance (¹H-NMR) spectrum of 3 exhibited signals at δ 3.85 (9H, s) due to aromatic methoxy protons and at δ 6.58 (2H, s) and 6.73—7.03 (3H, m) due to aromatic protons, and the signals of other protons

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1:
$$R_1 = R_2 = R_3 = H$$

1a: $R_1 = H$

2: $R_1 = OH$

2: $R_1 = OH$

3: $R_1 = H$

2: $R_1 = OH$

4: $R_2 = R_3 = CH_3$

2: $R_1 = OH$

8: $R_2 = glc$

8: $R_1 = OH$

8: $R_2 = glc$

8: $R_1 = OH$

8: $R_2 = glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

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8: $R_1 = OH$

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8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCCH_3$

8: $R_1 = OH$

8: $R_2 = Glc$

8: $R_3 = OCH_3$

8: $R_1 = OH$

8: $R_2 = CH_3$

8: $R_3 = OCH_3$

8: $R_1 = OH$

8: $R_1 = OH$

8: $R_2 = CH_3$

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8: $R_1 = OH$

8: $R_1 = OH$

8: $R_2 = CH_3$

8: $R_3 = OCH_3$

8: $R_3 = OCH_3$

8: $R_1 = OH$

8: $R_1 = OH$

8: $R_2 = CH_3$

8: $R_3 = OCH_3$

8

resembled those of 2. Methylation of 3 with diazomethane gave 3a as an amorphous powder, $C_{23}H_{28}O_8$, $[\alpha]_D^{25}+13.1^\circ$ (ethanol). The ¹H-NMR spectrum of 3a showed the presence of five aromatic methoxy groups (δ 3.84). The UV spectrum of 3a was very similar to that of (+)-medioresinol dimethyl ether (3d).⁹⁾

These data indicated that 3 contains a 2,6-diaryl-1-hydroxy-3,7-dioxabicyclo[3.3.0]-octane ring and that the aryl groups of 3 consist of a guaiacyl unit and a syringyl unit.

The carbon-13 nuclear magnetic resonance (13 C-NMR) spectra of 3 and 3a were compared with those of known lignans, 1, 1a, 2, 2a, 3c and 3d. The 13 C-NMR data of 3 indicated that 3 bears stereochemically the same structural skeleton as 2. The differences of chemical shifts at the C-1', C-2', C-3', C-5' and C-6' carbons between 3 and 3c ($\Delta\delta$ -4.3, +1.9, -0.5, -0.5 and +1.9) and between 3a and 3d ($\Delta\delta$ -4.4, +1.9, -0.6, -0.6 and +1.9), which are due to the effect of the alcoholic hydroxyl group at the C-1 position of 3 and 3a, clearly indicated that the aryl group at the C-2 position of 3 is the syringyl unit and the aryl group at the C-6 position of 3 is the guaiacyl unit. The mass spectral (MS) fragmentation patterns of 3 and 3a were also in good agreement with the results of 13 C-NMR analysis.

Chemical evidence for the structure 3 was obtained as follows. The sodium-ammonia reduction of $\bf 3a$ afforded $\bf 3b$ as a colorless syrup, $C_{23}H_{32}O_8$, $[\alpha]_D^{22}-15.4^\circ$ (chloroform). All the spectral data for $\bf 3b$ were in good agreement with those for the triol $\bf 3b$ which was obtained by the reduction of hydroxythujaplicatin trimethyl ether (di-O-methylhydroxythujaplicatin methyl ether)¹⁰⁾ with lithium aluminum hydride.

Thus, the structure of 3 has been established as (1S, 2R, 5R, 6S)-1-hydroxy-2-(3, 5-1)-1-hydroxy-4-hydroxyphenyl)-6-(4-1)-1-hydroxy-3-methoxyphenyl)-3,7-dioxabicyclo[3.3.0]-0ctane, and this compound has been designated as (+)-fraxiresinol.

The lignan 4 was obtained as a colorless crystalline powder, C₂₂H₂₆O₉, mp 95—97 °C,

Table I. ¹³C-NMR Chemical Shifts^{a)}

$\frac{\Delta\delta}{\Delta}$ (4b – 4d)		+1.9	9.0 –	-0.6 +1.9	
b	53.7 53.7 71.2 71.2 85.0 85.0	137.2 137.2 103.1 103.1	152.8 152.8 136.7 136.7	152.8 152.8 103.1 103.1	55.7 59.8
4 9	91.2 60.8 70.4 74.7 86.9 85.1	132.8 137.1 105.0 103.4	152.2 152.7 136.8 136.8	152.2 152.7 105.0 103.4	55.7 59.8
4 – 4c)		+1.8 +1.8	9.0-	-0.6 +1.8	
34	53.6 53.6 71.0 71.0 85.3 85.3	131.5 131.5 103.7 103.7	147.9 147.9 134.9 134.9	147.9 147.9 103.7	56.0
4	91.0 60.8 70.2 74.7 87.1 85.4	127.1 131.4 105.5 104.1	147.3 147.8 134.9 134.9	147.3 147.8 105.5 104.1	55.9
$\frac{d\delta}{(3\mathbf{a} - 3\mathbf{d})}$		-4.4 +1.9	9.0-	-0.6 +1.9	
34	53.7 53.5 71.1 71.0 85.0 84.8	137.2 133.8 103.1 109.9	152.8 148.7 136.6 148.1	152.8 111.6 103.1 118.1	55.4 55.8 59.8
3a	91.2 60.8 70.3 74.6 87.0 85.0	132.8 133.9 105.0 110.3	152.2 148.7 136.8 148.2	152.2 111.7 105.0 118.4	55.4 55.7 59.8
$\frac{\Delta\delta}{(3-3c)}$		-4.3 +1.9	-0.5	-0.5 +1.9	
36	53.6 53.4 71.0 70.8 85.2 85.1	131.4 132.2 103.6 110.4	147.9 147.5 134.8 145.9	147.9 115.1 103.6 118.5	55.6 55.9
е п	91.1 60.7 70.2 74.6 87.2 85.3	127.1 132.3 105.5 110.8	147.4 147.8 134.9 145.9	147.4 115.1 105.5 118.8	55.6 55.9
$\frac{d\delta}{(2\mathbf{a} - 1\mathbf{a})}$		-4.3 +1.9	-0.8	-0.5 +1.6	
1a	53.7 53.7 71.0 71.0 85.0 85.0	134.0 134.0 110.0	148.9 148.9 148.3 148.3	111.7 111.7 118.1 118.1	55.4
2a	91.0 60.8 70.2 74.7 86.9 85.1	129.7 133.9 111.9 110.3	148.1 148.7 148.2 148.2	111.2 111.7 119.7 118.3	55.4
$\frac{d\delta}{(2-1)}$		-4.2 +1.8	-0.7	-0.7 +1.6	
_	53.6 53.6 70.9 70.9 85.2 85.2	132.3 132.3 110.5 110.5	147.6 147.6 145.9 145.9	115.2 115.2 118.6 118.6	55.6
7	91.0 60.8 70.2 74.7 87.1 85.4	128.1 132.3 112.3 110.8	146.9 147.5 145.9 145.9	114.5 115.1 120.2 118.8	55.6
,	C-1 C-5 C-8 C-2 C-6 C-6	C-1, C-1, C-2, C-2,	C. C. 3. 4. 7. 4. 7. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4. 4.	C-5′ C-8′ C-6′	ОСН ₃

a) The spectra were taken in micro cells with a JNM-FX 60 spectrometer (15.00 MHz) in DMSO-d₆ with TMS as an internal reference.

Chart 2. The Mass Fragmentation Patterns of 3 and 3a

Chart 3

 $[\alpha]_D^{24.5} + 24.6^{\circ}$ (chloroform). The IR spectrum of 4 suggested the presence of aromatic rings (1615 and 1515 cm⁻¹). The absorption maxima in the UV spectrum of 4 and its bathochromic shift in the presence of base were very similar to those of (+)-syringaresinol (4c). (4c). The HNMR spectrum of 3 exhibited signals at δ 3.85 (12H, s) due to aromatic methoxy protons and at δ 6.60 (2H, s) and 6.62 (2H, s) due to aromatic protons. Acetylation of 4 with acetic anhydride-pyridine gave 4a as a colorless crystalline powder, $C_{28}H_{32}O_{12}$, mp 81—83 °C, $[\alpha]_D^{25} + 12.0^{\circ}$ (chloroform). The H-NMR spectrum of 4a showed the presence of an alcoholic acetoxy group (δ 1.70), two phenolic acetoxy groups (δ 2.30 and 2.32) and four aromatic methoxy groups (δ 3.79). Methylation of 4 with diazomethane gave 4b as an amorphous powder, $C_{24}H_{30}O_9$, $[\alpha]_D^{22} + 11.9^{\circ}$ (chloroform). The H-NMR spectrum of 4b showed the presence of six aromatic methoxy groups (δ 3.84). The UV spectrum of 4b was very similar to that of the known compound (+)-syringaresinol dimethyl ether (4d). The oxidation of 4b with potassium permanganate gave only 3,4,5-trimethoxybenzoic acid. These data suggest that 4 bears a marked structural resemblance to 2 and 3, and that the aryl groups of 4 are syringyl units.

In the 13 C-NMR spectrum, the differences of chemical shifts at the C-1', C-2', C-3', C-5' and C-6' carbons between **4** and **4c** ($\Delta\delta$ -4.4, +1.8, -0.6, -0.6 and +1.8) and between **4b**

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	TABLE II. Molecular Optical Rotation Differences			
	[α] _D (°)	[M] (°)	<i>∆</i> [<i>M</i>] (°)	
4	+24.6	+106.8	-153.2	
4c	+62.2	+260.0		
2	+39.0	+145.9	-131.6	
1	+77.5	+277.5		
3	+29.3	+118.4	104.2	
3c	+57.4	+222.7	-104.3	

TABLE II. Molecular Optical Rotation Differences

and 4d ($\Delta\delta$ -4.4, +1.9, -0.6, -0.6 and +1.9) supported the view that the aryl groups at both the C-2 and C-6 positions of 4 are syringyl units.

With regard to the problem of the absolute configuration of 4, a comparison of the molecular optical rotation differences between 4 and 4c with those between the known compounds, 2 and 1, and 3 and 3c suggested that 4 has the same absolute configuration as 2 and 3. Consequently, the structure of 4 has been established as (1S, 2R, 5R, 6S)-1-hydroxy-2,6-bis(3,5-dimethoxy-4-hydroxyphenyl)-3,7-dioxabicyclo[3.3.0]octane. This compound has been designated as (+)-1-hydroxysyringaresinol.

Lignans 5 and 6 were identified as (–)-olivil and (+)-cyclo-olivil, respectively, by direct comparison with authentic samples. The lignan 7 was obtained as an amorphous powder, $C_{20}H_{22}O_6$, mp 107—109 °C, $[\alpha]_D^{20} + 8.6^{\circ}$ (ethanol). The enzymatic hydrolysis of 7 gave 1 and D-glucose. The lignan 7 was confirmed to be (+)-pinoresinol- β -D-glucoside by direct comparison with authentic sample.⁷⁾

The lignan **8** was obtained as colorless plates, $C_{26}H_{32}O_{12}$, mp $127-129\,^{\circ}C$, $[\alpha]_D^{23}-9.3^{\circ}$ (methanol). The absorption maxima in the UV spectrum of **8** and its bathochromic shift in the presence of base were very similar to those of **7**. Acetylation of **8** with acetic anhydride—pyridine gave **8a** as a colorless syrup, $C_{38}H_{44}O_{18}$, $[\alpha]_D^{20}-6.3^{\circ}$ (ethanol). The ¹H-NMR spectrum of **8a** showed the presence of five alcoholic acetoxy groups (δ 1.67, 2.03 and 2.10), a phenolic acetoxy group (δ 2.33) and two aromatic methoxy groups (δ 3.83 and 3.87). The enzymatic hydrolysis of **8** gave **2** and D-glucose. The ¹³C-NMR spectrum of **8** also revealed that **8** is a monoglucoside of **2**. Methylation of **8** with diazomethane gave **8b** as an amorphous powder, $C_{27}H_{34}O_{12}\cdot 1.5H_2O$, mp $117-120\,^{\circ}C$, $[\alpha]_D^{23}-1.3^{\circ}$ (ethanol). The enzymatic hydrolysis of **8b** gave **8c** as an amorphous powder, $C_{21}H_{24}O_7$, $[\alpha]_D^{23}+37.9^{\circ}$ (chloroform) and D-glucose. Compound **8c** was identified as (+)-1-hydroxypinoresinol 4''-O-methyl ether by direct comparison with an authentic sample. Therefore, as regards the position of the glucose linkage in **8**, these data indicate that D-glucose is attached to the 4'-O-position of **2**. Thus, the structure of **8** has been established as (+)-1-hydroxypinoresinol-4'- β -D-glucoside.

Our results show that there is no difference between the lignans contained in the barks of *F. mandshurica* var. *japonica* and *F. japonica*, despite the difference in the coumarins contained.

In regard to the biological activity of the isolated lignans, lignans 1 and 7 showed high inhibitory activity against cyclic adenosine monophosphate (cAMP)-phosphodiesterase in vitro (IC₅₀ ($\times 10^{-5}$ M): 7.5 and 14.2). Weinryb et al. reported that a considerable number of therapeutic agents used as antipsychotics, antianxiety agents, antihypertensives and so on showed inhibitory effects against phosphodiesterase. Thus, these lignans might possess some pharmacological activity.

Experimental

All melting points were determined on a Yanagimoto micro-melting point apparatus and are uncorrected. The

following instruments were used: optical rotations, Yanaco OR-50D; UV spectra, Shimadzu UV-210; IR spectra, Shimadzu IR-400 and Hitachi 270-30; circular dichroism (CD) curves, Jasco J-40; 1 H-NMR spectra, JEOL JNM-PMX 60 and Hitachi R-40 with tetramethylsilane (TMS, δ =0) as an internal reference; 13 C-NMR spectra, JEOL JNM-FX 60, equipped with a JEC-980 computer; MS, Hitachi RMU-7L and Shimadzu LKB-9000. The abbreviations used are as follows: s, singlet; d, doublet; m, multiplet; t, triplet; q, quartet; sh, shoulder. Precoated thin-layer chromatography (TLC) plates, Silica gel 60_{F254} (Merck), were used for TLC and preparative TLC. The spots were detected by spraying the plates with 10% H₂SO₄ soln. and heating. Silica gel (100 mesh, Mallinckrodt) was used for column chromatography.

Isolation—Dry powdered bark of *Fraxinus mandshurica* var. *japonica* (4.3 kg), collected in December 1979 at our University, Hokkaido, Japan, were extracted four times with hot MeOH. The MeOH solution was concentrated to a small volume under reduced pressure, diluted with water and filtered. The filtrate was extracted successively with ether, CHCl₃ and BuOH.

The ether layer was evaporated to dryness. The ether extract $(15.7 \,\mathrm{g})$ was subjected to column chromatography; elution was carried out with a CHCl₃-AcOEt solvent system with gradually increasing proportions of AcOEt. The fractions were monitored by TLC developed with CHCl₃-AcOEt (1:2). The fractions $(100 \,\mathrm{ml})$ each) showing a TLC spot at Rf 0.59 were concentrated, and the residue was purified by preparative TLC using CHCl₃-AcOEt (1:1) to give 133.7 mg of 1. When treated in the same way as described for 1, the fractions showing TLC spots at Rf 0.37, 0.30, 0.24, 0.19 and 0.09 gave 121.2 mg of 2, 141.0 mg of 3, 94.9 mg of 4, 115.6 mg of 5 and 65.3 mg of 6, respectively.

The CHCl₃ layer was evaporated to dryness. The CHCl₃ extract $(20.2\,\mathrm{g})$ was subjected to column chromatography; elution was carried out with a CHCl₃-EtOH solvent system with gradually increasing proportions of EtOH. The fractions were monitored by TLC developed with CHCl₃-EtOH (4:1). The fractions $(100\,\mathrm{ml}$ each) showing a TLC spot at Rf 0.35 were concentrated, and the residue was purified by preparative TLC using CHCl₃-EtOH (4:1) to give 53.4 mg of 7.

The BuOH layer was evaporated to dryness. The BuOH extract (170.7 g) was subjected to column chromatography; elution was carried out with a CHCl₃-EtOH solvent system with gradually increasing proportions of EtOH. The fractions were monitored by TLC developed with CHCl₃-EtOH (4:1), no TLC spot of lignans was detected.

Dry powdered bark (3.5 kg) of Fraxinus japonica was treated in the same manner as described for Fraxinus mandshurica var. japonica. The ether extract (16.7 g) gave 108.0 mg of 1, 38 mg of 2, 29.2 mg of 3, 29 mg of 5 and 8 mg of 6. The CHCl₃ extract (14.4 g) gave 89.3 mg of 7. The BuOH layer was evaporated to dryness. The BuOH extract (40.0 g) was subjected to column chromatography; elution was carried out with a CHCl₃-EtOH solvent system with gradually increasing proportions of EtOH. The fractions were monitored by TLC developed with CHCl₃-EtOH (4:1). The fractions (100 ml each) showing a TLC spot at Rf 0.15 were concentrated, and the residue was purified by preparative TLC using CHCl₃-MeOH (4:1) to give 43.0 mg of 8.

- (+)-Pinoresinol (1)—Colorless prisms, mp 119—120 °C, [α] $_{0}^{21}$ + 77.5° (c = 0.26 in CHCl $_{3}$), UV $\lambda_{\max}^{\text{EiOH}}$ nm (log ε): 232.0 (4.14), 281.0 (3.75). IR $\nu_{\max}^{\text{CHCl}_{3}}$ cm $^{-1}$: 3545 (OH), 1610, 1510 (arom. C=C). MS m/z: 358 (M $^{+}$, C $_{20}$ H $_{22}$ O $_{6}$). 1 H-NMR (in CDCl $_{3}$) δ: 2.87—3.33 (2H, m, C $_{1.5}$ -H), 3.87 (6H, s, 2 × OCH $_{3}$), 3.56—4.46 (4H, m, C $_{4.8}$ -H), 4.73 (2H, d, J=5 Hz, C $_{2.6}$ -H), 6.65—7.05 (6H, m, arom. H). The identity of this compound was confirmed by direct comparison with authentic (+)-pinoresinol.
- (+)-1-Hydroxypinoresinol (2)—Colorless crystalline powder, mp 183—185 °C, [α]₁₅ + 39.0° (c=0.65 in EtOH). UV $\lambda_{\max}^{\text{EtOH}}$ nm (log ε): 232 (4.27), 281.0 (3.88). UV $\lambda_{\max}^{\text{EtOH}+\text{NaOH}}$ nm: 253, 293.3. IR $\nu_{\max}^{\text{CHCl}_3}$ cm $^{-1}$: 3540 (OH), 1605, 1510 (arom. C=C). MS: Calcd for C₂₀H₂₂O₇, 374.1364. Obsd., 374.1367. CD (c=1.458 × 10⁻⁴, ethanol) [θ]²⁰ × 10⁻³ (nm): +1.03 (277), -2.91 (238), +8.22 (212). ¹H-NMR (in CDCl₃) δ: 3.00—3.23 (1H, m, C₅-H), 3.87 (6H, s, 2 × OCH₃), 3.65—4.22 (3H, m, C_{4.8}-H), 4.53 (1H, dd, J=8 and 9 Hz, C_{4.6}-H), 4.80 (1H, s, C₂-H), 4.85 (1H, d, J=5 Hz, C₆-H), 6.67—7.10 (6H, m, arom. H). The identity of this compound was confirmed by direct comparison with authentic (+)-1-hydroxypinoresinol.
- (+)-Fraxiresinol (3)——Amorphous powder, [α] $_{0}^{25}$ + 29.3° (c = 0.61 in CHCl $_{3}$). UV $\lambda_{\max}^{\text{EiOH}}$ nm (log ϵ): 232.3 (4.10), 279.8 (3.54). UV $\lambda_{\max}^{\text{EiOH}}$ nm: 235.7, 254.9, 284. IR $\nu_{\max}^{\text{CHCl}_{3}}$ cm $^{-1}$: 3545 (OH), 1615, 1510 (arom. C = C). MS: Calcd for C $_{21}$ H $_{24}$ O $_{8}$, 404.1470. Obsd., 404.1476. CD (c = 4.862 × 10 $^{-4}$, ethanol) [θ] 20 × 10 $^{-3}$ (nm): +1.202 (233.0), +0.926 (270.0). ¹H-NMR (in CDCl $_{3}$) δ : 2.95—3.22 (1H, m, C $_{5}$ -H), 3.85 (9H, s, 3 × OCH $_{3}$), 3.70—4.17 (3H, m, C $_{4,8}$ -H), 4.52 (1H, dd, J = 9 and 9 Hz, C $_{4c}$ -H), 4.80 (1H, s, C $_{2}$ -H), 4.85 (1H, d, J = 5 Hz, C $_{6}$ -H), 6.58 (2H, s, arom. C $_{2',6'}$ -H), 6.73—7.03 (3H, m, arom. C $_{2',5',6'}$ -H).
- (+)-Fraxiresinol Dimethyl Ether (3a)——3 (50 mg) was methylated with diazomethane in the usual way. The crude product was purified by preparative TLC using CHCl₃-AcOEt (2:1) to give 3a (34.8 mg) as an amorphous powder. $[\alpha]_{D}^{25} + 13.1^{\circ}$ (c = 0.58 in EtOH). UV λ_{\max}^{EtOH} nm (log ε): 229.6 (4.22), 277.9 (3.58). UV $\lambda_{\max}^{EtOH+NaOH}$ nm: 229.6 sh, 277.9. IR ν_{\max}^{KBr} cm⁻¹: 3450 (OH), 1590, 1510 (arom. C=C). MS: Calcd for C₂₃H₂₈O₈, 432.1781. Obsd., 432.1765. ¹H-NMR (in CDCl₃) δ: 2.94—3.48 (1H, m, C₅-H), 3.84 (15H, s, 5 × OCH₃), 3.68—4.16 (3H, m, C_{4.8}-H), 4.52 (1H, dd, J = 9 and 9 Hz, C_{4.e}-H), 4.80 (1H, s, C₂-H), 4.83 (1H, d, J = 5 Hz, C₆-H), 6.58 (2H, s, arom. C_{2',6'}-H), 6.71—7.03 (3H, m, arom. C_{2'',5'',6''}-H).

Sodium-Ammonia Reduction of (+)-Fraxiresinol Dimethyl Ether (3a)—3a (30 mg) in pure tetrahydrofuran (THF) (2 ml) was added to liquid ammonia, and the stirred solution was treated with sodium (30 mg). After the

disappearance of the blue color, a little water was added to the yellow solution and the ammonia was evaporated off. The residue was treated with H_2O , the mixture was extracted with $CHCl_3$, and the extract was concentrated. The residue was purified by preparative TLC using $CHCl_3$ –AcOEt (1:1) to give **3b** as a colorless syrup. $[\alpha]_{D}^{22}-15.4^{\circ}$ (c=0.15 in $CHCl_3$). UV λ_{\max}^{EIOH} nm $(\log\epsilon)$: 224.5 (4.16), 278.5 (3.52). IR ν_{\max}^{KBr} cm⁻¹: 3424 (OH), 1592, 1514 (arom. C=C). MS: Calcd for $C_{23}H_{32}O_8$, 436.2094. Obsd., 436.2085. MS m/z: 436 (M⁺, $C_{23}H_{32}O_8$, 5.4%), 254 ($C_{13}H_{18}O_5$, 4.8%), 237 ($C_{13}H_{17}O_4$, 4.3%), 219 ($C_{13}H_{15}O_3$, 5.4%), 189 ($C_{12}H_{13}O_2$, 7.0%), 182 ($C_{10}H_{14}O_3$, 100%), 181 ($C_{10}H_{13}O_3$, 31.7%), 167 ($C_9H_{11}O_3$, 25.8%), 151 ($C_9H_{11}O_2$, 75.3%), 137 ($C_8H_9O_2$, 25.3%). ¹H-NMR (in $CDCl_3$) δ : 1.91—2.27 (1H, m, C_3 –H), 2.44—2.71 (2H, m, C_4 –H), 2.90 (2H, s, C_1 –H), 3.35—3.80 (4H, m, $C_{2a,3a}$ –H), 3.53 (2H, s, 2×OH, quenched by addition of D_2O), 3.83 (15H, s, 5×OCH₃), 6.50 (2H, s, arom. $C_{2',6}$ –H), 6.63—6.90 (3H, m, arom. $C_{2'',5'',6''}$ –H).

Reduction of Hydroxythujaplicatin Trimethyl Ether with Lithium Aluminum Hydride——A solution of hydroxythujaplicatin trimethyl ether (136 mg) in THF (5 ml) was added dropwise to a suspension of LiAlH₄ (136 mg) in THF (5 ml). The mixture was stirred for 8 h at room temperature and then poured into ice-cold water. The whole was carefully acidified with 10% H₂SO₄ soln. and extracted with ether. The ether soln. was washed with water and concentrated *in vacuo*. The residue showed a spot of Rf 0.29 on TLC using CHCl₃–AcOEt (1:2). Purification by TLC afforded the triol 3b as a colorless syrup. All the spectral data of the product were in good agreement with those of the triol 3b.

- (+)-1-Hydroxysyringaresinol (4)—Colorless crystalline powder, mp 95—97 °C. [α]_D^{24.5} + 24.6° (c=0.12 in CHCl₃). UV $\lambda_{\text{max}}^{\text{EIOH}}$ nm (log ε): 237.0 (4.10), 272.2 (3.40), 281.0 (3.32). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 3545 (OH), 1615, 1515 (arom. C=C). MS: Calcd for C₂₂H₂₆O₉, 434.1574. Obsd., 434.1562. CD (c=5.423 × 10⁻⁴, ethanol) [θ]²⁰ × 10⁻³ (nm): +4.05 (262). ¹H-NMR (in CDCl₃) δ : 3.00—3.20 (1H, m, C₅-H), 3.85 (12H, s, 4 × OCH₃), 3.66—4.20 (3H, m, C_{4,8}-H), 4.51 (1H, dd, J=9 and 9 Hz, C_{4e}-H), 4.78 (1H, s, C₂-H), 4.81 (1H, d, J=5 Hz, C₆-H), 6.60 (2H, s, arom. H), 6.62 (2H, s, arom. H).
- (+)-1-Hydroxysyringaresinol Triacetate (4a)——4 (50 mg) was acetylated with acetic anhydride–pyridine in the usual way. The crude acetate was purified by preparative TLC using CHCl₃–AcOEt (1:1) to give 4a (38.0 mg) as a colorless crystalline powder. mp 81—83 °C. [α]_D²⁵ +12.0° (c=0.63 in CHCl₃). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ϵ): 223.5 (4.23) sh, 274.4 (3.40), 277.8 (3.39). IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm⁻¹: 1758, 1737 (C=O), 1600, 1504 (arom. C=C). MS: Calcd for C₂₈H₃₂O₁₂, 560.1890. Obsd., 560.1889. ¹H-NMR (in CDCl₃) δ: 1.70 (3H, s, alcoholic OCOCH₃), 2.30, 2.32 (6H, each s, 2 × phenolic OCOCH₃), 3.17—3.48 (1H, m, C₅–H), 3.79 (12H, s, 4 × OCH₃), 3.53—4.46 (3H, m, C_{4,8}–H), 4.52 (1H, dd, J=9 and 9 Hz, C_{4e}–H), 4.78 (1H, d, J=5 Hz, C₆–H), 5.03 (1H, s, C₂–H), 6.55 (2H, s, arom. H), 6.60 (2H, s, arom. H).
- (+)-1-Hydroxysyringaresinol Dimethyl Ether (4b)——4 (25 mg) was methylated with diazomethane in the usual way. The crude product was purified by preparative TLC using CHCl₃-AcOEt (2:1) to give **4b** (14.6 mg) as an amorphous powder. [α]_D²² +11.9° (c=0.076 in CHCl₃). UV $\lambda_{\rm max}^{\rm EIOH}$ nm (log ε): 226.1 (4.19) sh, 275.7 (3.53).IR $\nu_{\rm max}^{\rm KBr}$ cm⁻¹: 3430 (OH), 1590, 1512 (arom. C=C). MS m/z: 462 (M⁺, C₂₄H₃₀O₉). ¹H-NMR (in CDCl₃) δ: 2.90—3.49 (1H, m, C₅-H), 3.84 (18H, s, 6 × OCH₃), 3.50—4.40 (3H, m, C_{4,8}-H), 4.52 (1H, dd, J=9 and 9 Hz, C_{4e}-H), 4.80 (1H, s, C₂-H), 4.83 (1H, d, J=5 Hz, C₆-H), 6.57 (2H, s, arom. H), 6.60 (2H, s, arom. H).

Oxidation of (+)-1-Hydroxysyringaresinol Dimethyl Ether (4b) with Potassium Permanganate—4b (10 mg) was dissolved in 10 ml of 1 N NaOH soln. with warming on a water bath. The solution was treated with 2% KMnO₄ soln. in small portions at 35 °C (pink end point). The color was discharged with sodium bisulfite solution and the precipitate was filtered off. The filtrate, after being acidified with dil. H_2SO_4 soln., was extracted with ether. The ether solution was evaporated to yield the oxidation product as colorless needles, mp 169—171 °C. The identity of this product was confirmed by direct comparison with authentic 3,4,5-trimethoxybenzoic acid.

- (-)-Olivil (5)—Colorless needles from EtOH, mp 121—123 °C. [α] $_D^{23}$ –23.9° (c = 1.29 in EtOH). UV λ_{max}^{EtOH} nm (log ε): 230.4 (4.10), 281.5 (3.72). IR ν_{max}^{KBr} cm $^{-1}$: 3400 (OH), 1600, 1510 (arom. C = C). MS m/z: 376 (M $^+$, C $_{20}$ H $_{24}$ O $_{7}$). ¹H-NMR (in CD $_3$ OD) δ : 2.10—2.83 (1H, m, C $_3$ —H), 2.92 (2H, s, –C $_{12}$ —), 3.76 (6H, s, 2 × OCH $_3$), 3.38—3.98 (2H, m, –C $_{12}$ OH), 3.97—4.50 (2H, m, C $_5$ —H), 6.46—7.15 (6H, m, arom. H). The identity of this product was confirmed by direct comparison with authentic (—)-olivil.
- (+)-Cyclo-olivil (6)—Colorless needles from EtOH, mp 168—170 °C. [α]_D²³ +51.8° (c=0.77 in EtOH). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 230.5 (4.03), 284 (3.69). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3430 (OH), 1620, 1520 (arom. C=C). MS m/z: 376 (M⁺, C₂₀H₂₄O₇). ¹H-NMR (in CD₃OD) δ: 1.90—2.24 (1H, m, C₂-H), 2.60 (1H, d, J=17 Hz, C₄-H), 3.20 (1H, d, J=17 Hz, C₄-H), 3.73, 3.75 (6H, each s, 2×OCH₃), 3.40—4.35 (5H, m, C_{1,2a,3a}-H), 6.15—6.85 (5H, m, arom. H). The identity of this product was confirmed by direct comparison with authentic (+)-cyclo-olivil.
- (+)-Pinoresinol-β-D-glucoside (7)—Amorphous powder, mp 107—109 °C. [α]_D²⁰ + 8.6° (c = 0.79 in EtOH). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 228.3 (4.14), 280.5 (3.66). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1600, 1510 (arom. C=C). MS m/z: 358, C₂₀H₂₂O₆). ¹H-NMR (in CD₃OD) δ: 2.93—3.18 (2H, m, C_{1.5}–H), 3.80 (6H, s, 2×OCH₃), 3.58—4.44 (4H, m, C_{4.8}–H), 6.60—7.21 (6H, m, arom. H). ¹³C-NMR (in DMSO- d_6) δ: 53.5 (C-1, 5), 84.8 (C-2), 70.9 (C-4, 8), 85.1 (C-6), 135.2 (C-1'), 110.4 (C-2', 2''), 148.9 (C-3', 3''), 145.9 (C-4', 4''), 115.1 (C-5', 5''), 118.1 (C-6'), 132.1 (C-1''), 118.6 (C-6''), 55.6 (OCH₃), 100.1 (Glc-1), 73.1 (Glc-2), 76.9 (Glc-3, 5), 69.6 (Glc-4), 60.6 (Glc-6).

Hydrolysis of (+)-Pinoresinol- β -D-glucoside (7) with Emulsin—7 (10 mg) was hydrolyzed with emulsin in the usual way to give 1 and D-glucose. The lignan 7 was identified by direct comparison with authentic (+)-pinoresinol- β -

D-glucoside.

(+)-1-Hydroxypinoresinol-4′-β-D-glucoside (8)—Colorless plates, mp 127—129 °C. [α] $_{\rm D}^{23}$ -9.3° (c=0.42 in MeOH). UV $\lambda_{\rm max}^{\rm EIOH}$ nm (log ε): 228.0 (4.11), 279.8 (3.66). UV $\lambda_{\rm max}^{\rm EIOH+NaOH}$ nm: 252, 280, 292 sh. IR $\nu_{\rm max}^{\rm KBr}$ cm $^{-1}$: 3400 (OH), 1600, 1515 (arom. C=C). CD (c=3.444 × 10 $^{-4}$, ethanol) [θ] 20 × 10 $^{-3}$ (nm): -0.59 (271), -1.90 (239), +1.77 (214). FD-MS m/z: 536 (M $^+$, C₂₆H₃₂O₁₂). 1 H-NMR (in CD₃OD) δ : 2.96—3.14 (1H, m, C₅–H), 3.80, 3.83 (6H, each s, 2 × OCH₃), 3.47—4.58 (4H, m, C_{4.8}–H), 4.80 (1H, s, C₂–H), 6.60—6.80 (6H, m, arom. H). 13 C-NMR (in DMSO- d_6) δ : 91.1 (C-1), 86.8 (C-2), 70.2 (C-4), 60.8 (C-5), 85.4 (C-6), 74.7 (C-8), 131.1 (C-1′), 112.6 (C-2′), 148.3 (C-3′), 146.0 (C-4′, 4′′), 114.6 (C-5′), 119.7 (C-6′), 132.3 (C-1′′), 110.8 (C-2′′), 147.5 (C-3′′), 115.1 (C-5′′), 118.8 (C-6′′), 55.6 (OCH₃), 100.4 (Glc-1), 73.2 (Glc-2), 76.8 (Glc-3, 5), 69.7 (Glc-4), 60.8 (Glc-6).

(+)-1-Hydroxypinoresinol-4'-β-D-glucoside Hexaacetate (8a)——8 (16 mg) was acetylated with acetic anhydride-pyridine in the usual way. The crude acetate was purified by preparative TLC using CHCl₃-AcOEt (1:1) to give 8a (15.5 mg) as a colorless syrup. [α]_D²⁰ -6.3° (c=1.8 in EtOH). UV $\lambda_{\text{max}}^{\text{EIOH}}$ nm (log ε): 220 (4.26), 275 (3.75), 279 (3.74). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 1720 (C=O), 1595, 1500 (arom. C=C). MS m/z: 788 (M⁺, C₃₈H₄₄O₁₈). ¹H-NMR (in CDCl₃) δ : 1.67 (3H, s, alcoholic OCOCH₃), 2.03, 2.10 (12H, each s, 4×alcoholic OCOCH₃), 2.33 (3H, s, phenolic OCOCH₃), 3.10—3.55 (1H, m, C₅-H), 3.83, 3.87 (6H, each s, 2×OCH₃), 4.10—4.55 (4H, m, C_{4.8}-H), 4.82 (1H, d, J=5 Hz, C₆-H), 5.08 (1H, s, C₂-H), 6.73—7.27 (6H, m, arom. H).

Hydrolysis of (+)-1-Hydroxypinoresinol-4'- β -D-glucoside (8) with Emulsin—8 (16 mg) was hydrolyzed with emulsin in the usual way to give 2 and D-glucose.

(+)-1-Hydroxypinoresinol 4''-*O*-Methyl Ether-4'-β-D-glucoside (8b)——8 (28.2 mg) was methylated with diazomethane in the usual way. The crude product was purified by preparative TLC using CHCl₃–EtOH (4:1) to give 8b (10 mg) as an amorphous powder. mp 117—120 °C. [α]_D²³ – 1.3° (c = 1.0 in EtOH). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 230 (4.26), 277 (3.76). IR $\nu_{\text{max}}^{\text{KBr}}$ cm⁻¹: 3400 (OH), 1590, 1510 (arom. C=C). *Anal.* Calcd for C₂₇H₃₄O₁₂·1.5H₂O: C, 56.14; H, 6.46. Found: C, 56.23; H, 6.18. ¹H-NMR (in CD₃OD) δ: 2.92—3.10 (1H, m, C₅–H), 3.77, 3.79, 3.83 (9H, each s, 3 × OCH₃), 3.52—4.38 (4H, m, C_{4.8}–H), 4.82 (1H, d, J=5 Hz, C₆–H), 4.73 (1H, s, C₂–H), 6.59—7.16 (6H, m, arom. H). ¹³C-NMR (in DMSO-d₆) δ: 91.2 (C-1), 86.8 (C-2), 70.3 (C-4), 60.8 (C-5), 85.1 (C-6), 74.7 (C-8), 131.0 (C-1'), 112.5 (C-2'), 148.3 (C-3'), 145.9 (C-4'), 114.6 (C-5'), 119.7 (C-6'), 133.9 (C-1''), 110.2 (C-2''), 148.7 (C-3''), 148.3 (C-4''), 111.6 (C-5''), 118.4 (C-6''), 55.5, 55.7 (OCH₃), 100.3 (Glc-1), 73.2 (Glc-2), 76.9 (Glc-3, 5), 69.7 (Glc-4), 60.8 (Glc-6).

Hydrolysis of (+)-1-Hydroxypinoresinol 4"-O-Methyl Ether-4'- β -D-glucoside (8b) with Emulsin—8b (25 mg) was hydrolyzed with emulsin in the usual way to give 8c (12 mg) and D-glucose.

(+)-1-Hydroxypinoresinol 4"-*O*-Methyl Ether (8c)——Amorphous powder. [α]₂²³ + 37.9 ° (c=0.81 in CHCl₃). UV $\lambda_{\text{max}}^{\text{EtOH}}$ nm (log ε): 232.5 (4.21), 280 (3.75). UV $\lambda_{\text{max}}^{\text{EtOH}+\text{NaOH}}$ nm: 235, 253, 285, 295. IR $\nu_{\text{max}}^{\text{CHCl}_3}$ cm $^{-1}$: 3540 (OH), 1605, 1590, 1510 (arom. C=C). MS: Calcd for C₂₁H₂₄O₇, 388.1520. Obsd., 388.1538 CD (c=4.727 × 10⁻⁴, ethanol) [θ]²⁰ × 10⁻³ (nm): +1.06 (278), -0.36 (242.5), +0.32 (240), +0.33 (228). ¹H-NMR (in CDCl₃) δ : 1.80 (1H, br s, OH, quenched by addition of D₂O), 2.94—3.33 (1H, m, C₅-H), 3.91 (9H, s, 3 × OCH₃), 3.42—4.72 (4H, m, C_{4.8}-H), 4.83 (1H, s, C₂-H), 4.88 (1H, d, J=5 Hz, C₆-H), 6.59—7.24 (6H, m, arom. H). The identity of this product was confirmed by direct comparison with authentic (+)-1-hydroxypinoresinol 4"-O-methyl ether.

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