## Fluorine-Sacrificial Cyclizations as an Access to 5-Fluoropyrazoles

Jean-Noël Volle<sup>[a]</sup> and Manfred Schlosser\*<sup>[a]</sup>

Keywords: Fluorine-containing heterocycles / Methyl trifluoropyruvate / Fluoride elimination / Cyclization reactions / Hydrazone / Hydrazine tautomerism

amines.

Methyl 3-methoxy-2-trifluoromethylacrylate 1, readily prepared by Wittig reaction from methyl 3,3,3-trifluoropyruvate, has been treated with a number of aryl- (or hetaryl-) hydrazines. Under mild base-catalysis, the resulting 3-hydrazinoacrylates 6 undergo consecutive hydrogen fluoride elimination and intramolecular nucleophilic addition to afford

#### Introduction

Previously, this laboratory has devoted considerable effort towards the preparation of "fluoroisoprene" (2-fluoro-3-methyl-1,3-butadiene)<sup>[1-3]</sup> and its homologues,<sup>[2-5]</sup> including congeners bearing electron-withdrawing<sup>[2,6]</sup> and electron-donating<sup>[6–9]</sup> heterofunctional groups, as well as 2fluoro-2-alkenals<sup>[10]</sup> and the corresponding N-tert-butyl imines<sup>[11]</sup> and N,N-dimethylhydrazones.<sup>[12]</sup> These fluorinated building blocks have been employed as versatile modules in chain-lengthening ("fluoroisoprenylation")<sup>[11]</sup> or Diels-Alder cycloaddition<sup>[2,4,7,8,12]</sup> reactions. More recently, the acyl chloride and the methyl ester of 2-fluoro-3-methoxyacrylic acid have been used as electrophilic components in Knorr-Effenberger type and other cyclization reactions, allowing access to a host of fluorine-bearing pyrazolones, uracils, 2-quinolones, 2-chromenones (2H-1-benzopyran-2ones), 1-thio-2-chromenones, and benzothiazepinones.[9,13,14]

We decided to prepare methyl 3-methoxy-2-trifluoromethyl-2-propenoate (1) in order to extend the latter studies towards the assembly of trifluoromethyl-substituted heterocycles. We had hitherto gained access to this class of compounds through the use of two other building blocks, namely 3-trifluoromethyl-1-trimethylsilyloxy-1,3-butadiand 4-tert-butylamino-1,1,1-trifluoro-3-buten-2ene<sup>[15]</sup> one,<sup>[16,17]</sup> a synthetic equivalent of 4,4,4-trifluoro-3-oxobutanal.

The required key intermediate 1 was readily obtained by a Wittig reaction between methyl 3,3,3-trifluoropyruvate (methyl 3,3,3-trifluoro-2-oxopropanoate)<sup>[18,19]</sup> and (triphenylphosphonio)methoxymethanide.<sup>[20,21]</sup> The product was isolated as a 2:3 (Z/E) mixture and was used as such in all subsequent reactions.

Eur. J. Org. Chem. 2000, 823-828

© WILEY-VCH Verlag GmbH, D-69451 Weinheim, 2000



methyl 1-(het)aryl-5-fluoropyrazole-4-carboxylates 7. 5-Ami-

nopyrazoles 8 have been obtained by direct reaction of the

ester 7a with a lithium amide, whereas 5-fluoro-1-phenylpyr-

azole-4-carboxamides 10 have been formed by condensation

of the 5-fluoro-1-phenylpyrazole-4-carboxylic acid 9 with

Upon gentle warming, ester 1 underwent a condensation reaction with aniline to afford (E)-methyl 3-anilino-2-trifluoromethyl-2-propenoate (2a; 92%). Heating this intermediate to 100 °C in the presence of polyphosphoric acid gave 4-quinolone as the sole identifiable product (3a; 24%). Apparently, repeated 1,4-dehydrofluorination followed by addition of water led to a transformation of the trifluoromethyl substituent into a carboxy function and then the latter was lost in a typical  $\beta$ -oxo-assisted decarboxylation process. With *N*-methylaniline, the reaction sequence took a slightly different course. (N-Methylanilido)acrylate (2b; 52%), derived from this amine as a 1:4 (Z/E) mixture, was converted into methyl 1-methyl-4-quinolone-3-carboxylate (3b; 44%) as the major cyclization product. Lastly, methyl 3-(2-aminophenyl)amino-2-trifluoromethyl-2-propenoate (4; 89%) and methvl 1H-4-isopropyloxy-1,5-benzodiazepine-3-carboxylate (5b; 7%) were obtained when 1,2-phenylenediamine was employed as the starting material.

Fluoride elimination/nucleophile addition sequences are responsible for the lability of 2- and 4-(trifluoromethyl)phenol under basic conditions.<sup>[22,23]</sup> Similar transformations of trifluoromethyl groups by two-step nucleophilic substitutions have been reported for 4-(trifluoromethyl)ani-

<sup>[</sup>a] Section de Chimie (BCh), Université, CH-1015 Lausanne, Switzerland Fax: (internat.) + 41-21/692 39 65

### **FULL PAPER**



line,<sup>[24,25]</sup> 4-(nonofluorobutyl)aniline,<sup>[26]</sup> 2- and 3-(trifluoromethyl)indole,<sup>[27,28]</sup> and 5-amino-2-chloro-4-(trifluoromethyl)thiazole.<sup>[29]</sup> The irreversible inhibition of decarboxylases, transaminases, and racemases by a variety of βfluoroamines also involves the elimination of hydrogen fluoride and subsequent binding to the enzyme through a cysteine residue.<sup>[30]</sup> 1-(2-Hydroxyphenyl)-2,3,3,3-tetrafluoropropanone<sup>[31,32]</sup> and N-(2-hydroxyphenyl)-2,2,2-trifluoro-1-(trifluoromethyl)ethylamine<sup>[33]</sup> undergo base-promoted dehydrofluorination followed by intramolecular nucleophilic attack of the phenolate oxygen to afford a 2-substituted 3-fluoro-4H-1-benzopyran-4-one (a chromenone) and 2fluoro-3-trifluoromethyl-1,4-benzoxazine, respectively. N-Alkylidene-2-(trifluoromethyl)anilines react with nucleophiles such as butyllithium, lithium diisopropylamide, or potassium tert-butoxide to afford quinolines bearing the nucleophile moiety at the 4-position.<sup>[34–37]</sup> 4-Fluoroquinolines can be prepared by trapping of (6-difluoromethylene-2,4cvclohexadienvlidene)amine, generated from 2-(trifluoromethyl)aniline, with nucleophiles such as acetylides, enolates, or eneimides (α-deprotonated aliphatic nitriles).<sup>[26,38]</sup> Other methods involving dehydrofluorination and cyclization through intramolecular nucleophilic addition lead to 1,2-dihydro-3-trifluoromethyl-4-pyridazinones,<sup>[39]</sup> to 2fluoroindoles, -benzofurans, and -benzothiophenes,<sup>[40]</sup> and to 3-trifluoromethyl- (or 3-perfluoroalkyl-), 4-fluoro-, or both 3-trifluoromethyl- and 4-fluoro-substituted pyrazoles.[41-43]

It became apparent that methyl 3-methoxy-2-(trifluoromethyl)acrylate (1) described above offers a convenient access to 5-fluoropyrazole derivatives. When the ester 1 was heated in the presence of an aryl- (or hetaryl)- hydrazine, smooth replacement of the methoxy group occurred. The methyl 3-(het)arylhydrazino-2-(trifluoromethyl)acrylates **6** thus formed were found to exist in equilibrium with the tautomeric species **6**'. Fluorine-sacrificial cyclization occurred under weakly basic conditions, a first elimination of hydrogen fluoride initiating intramolecular addition of the terminal nitrogen atom to the halogen-bearing carbon, and a second one, as the last stage of the ring-closure process, affording methyl 1-(het)aryl-5-fluoropyrazole-4-carboxylates **7** as the final products.



When solutions of ester 7a and lithium anilide or lithium *p*-anisidide in tetrahydrofuran were mixed, the methyl 5-amino-1-phenylpyrazole-4-carboxylates **8a** (84%) and **8b** (74%) were rapidly formed in a nucleophilic 1,4-addition/ fluoride elimination sequence. Alternatively, ester **7a** could

be saponified to give the carboxylic acid **9**, and the latter could then be converted into the carboxamides **10** by condensation with various aromatic amines in the presence of (1-benzotriazolyloxy)tripyrrolidinophosphonium hexafluorophosphate (BTP-HFP).<sup>[44]</sup>

The new access to 5-fluoropyrazoles 7 reported herein has its limitations. We were unable to isolate any heterocyclic compounds following attempted reactions between acrylate 1 and *alkyl*hydrazines. We are currently focussing on acylhydrazines and hydroxylamines as potential cyclization components.

### **Experimental Section**

**General Remarks:** For standard operations and abbreviations, refer to recent publications from this laboratory.<sup>[45,46]</sup> <sup>1</sup>H- and <sup>19</sup>F NMR spectra were recorded in deuteriochloroform solution at 400 and 376 MHz, respectively. Mass spectra were obtained after chemical ionization (CI) and only <sup>35</sup>Cl isotopomers are listed in the case of chlorinated compounds.

**Preparation of the Key Starting Material: Methyl 3-Methoxy-2-trifluoromethyl-2-propenoate** (1): (Methoxymethyl)triphenylphosphonium bromide (58 g, 0.15 mol) was added to a solution of *tert*butyllithium (0.15 mol) in a 2:1 mixture (250 mL) of tetrahydrofuran and pentane cooled to -75 °C. After stirring vigorously for 15 min at -25 °C, the mixture was cooled to -75 °C once more, whereupon methyl trifluoropyruvate<sup>[18,19]</sup> (23 g, 0.15 mol) was added. The reaction mixture was allowed to warm to 25 °C and then poured into cold (-25 °C) pentane (250 mL). After the addition of diatomaceous earth (kieselguhr), the solid material was removed by suction filtration and the filtrate was concentrated. On distillation under reduced pressure, a colorless liquid was collected; 14.1 g (51%); b.p. 79–81 °C/5 Torr. – C<sub>6</sub>H<sub>7</sub>F<sub>3</sub>O<sub>3</sub> (184.11): calcd. C 39.14, H 3.83; found C 39.31, H 3.93.

On the basis of gas chromatographic (30 m, DB-1701, 100 °C; 30 m, DB-FFAP, 100 °C) and NMR analyses, the product was found to consist of the (*Z*) and (*E*) isomers in a 2:3 ratio.  $^{-1}$ H NMR of *Z*-1:  $\delta$  = 7.24 (q, *J* = 1.5, 1 H), 4.02 (s, 3 H), 3.80 (s, 3 H).  $^{-1}$ H NMR of *E*-1:  $\delta$  = 7.62 (s, 1 H), 4.01 (s, 3 H), 3.78 (s, 3 H).  $^{-19}$ F NMR of *Z*-1:  $\delta$  = -61.8 (d, *J* = 1.5 Hz).  $^{-19}$ F NMR of *E*-1:  $\delta$  = -58.8 (s).  $^{-1}$ MS (CI, GC-coupled) of *Z*-1: *m*/*z* (%) = 202 (46) [M<sup>+</sup> + NH<sub>4</sub>], 185 (83) [M<sup>+</sup> + 1], 184 (16) [M<sup>+</sup>], 153 (100).  $^{-1}$ MS (CI, GC-coupled) of *E*-1: *m*/*z* (%) = 202 (63) [M<sup>+</sup> + NH<sub>4</sub>], 185 (35) [M<sup>+</sup> + 1], 184 (20) [M<sup>+</sup>], 153 (100).

3-Amino-2-(trifluoromethyl)acrylates and Cyclization Products Derived Therefrom: Methyl 3-Anilino-2-trifluoromethyl-2-propenoate (2a): A mixture of methyl 3-methoxy-2-trifluoromethyl-2-propenoate 1 (5.5 g, 30 mmol) and aniline (2.7 mL, 2.8 g, 30 mmol) was heated at 100 °C for 1 h. The product was then absorbed on silica and eluted with a 1:20 mixture ( $\nu/\nu$ ) of ethyl acetate and hexanes. Upon recrystallization of the white solid from hexanes, colorless needles were obtained; m.p. 68–70 °C; 6.8 g (92%). – <sup>1</sup>H NMR:  $\delta = 10.47$  (br. d, J = 13.3 Hz, 1 H), 7.85 (d, J = 13.3 Hz, 1 H), 7.36 (t, J = 7.9 Hz, 2 H), 7.14 (t, J = 7.9 Hz, 1 H), 7.06 (d, J = 7.9 Hz, 2 H), 3.84 (s, 3 H). – <sup>19</sup>F NMR:  $\delta = -60.0$  (s). – MS (CI): m/z (%) = 263 (33) [M<sup>+</sup> + NH<sub>4</sub>], 246 (100) [M<sup>+</sup> + 1], 245 (20) [M<sup>+</sup>], 226 (90). – C<sub>11</sub>H<sub>10</sub>F<sub>3</sub>NO<sub>2</sub> (245.20): calcd. C 53.88, H 4.11; found C 53.73, H 4.15.

Eur. J. Org. Chem. 2000, 823-828

**Methyl 3-Methylanilino-2-trifluoromethyl-2-propenoate (2b):** The above procedure was repeated by using *N*-methylaniline (3.3 mL, 3.2 g, 30 mmol) instead of aniline and heating at 100 °C was prolonged to 5 h. The corresponding substitution product, isolated by column chromatography, was found to exist as a 1:4 (*Z/E*) mixture; b.p. 98–99 °C/0.8 Torr; 4.0 g (52%). – <sup>1</sup>H NMR of *Z*-**2b**:  $\delta$  = 7.97 (q, *J* = 1.5 Hz, 1 H), 7.4 (m, 5 H), 3.76 (s, 3 H), 3.46 (q, *J* = 3.0 Hz, 3 H). – <sup>1</sup>H NMR of *Z*-**2b**:  $\delta$  = 7.3 (m, 6 H), 3.69 (s, 3 H), 3.38 (s, 3 H). – <sup>19</sup>F NMR of *Z*-**2b**:  $\delta$  = -51.3 (s). – <sup>19</sup>F NMR of *E*-**2b**:  $\delta$  = -59.0 (s). – C<sub>12</sub>H<sub>12</sub>F<sub>3</sub>NO<sub>2</sub> (259.23): calcd. C 55.60, H 4.66; found C 55.61, H 4.75.

Methyl 3-(2-Aminoanilino)-2-trifluoromethyl-2-propenoate (4): A similar reaction as above was performed with 1,2-phenylenediamine (3.2 g, 30 mmol), but this substrate was dissolved together with the acrylate 1 in methanol (100 mL) and the resulting solution was heated under reflux for 2 h. After evaporation of the solvent, the crude product was purified by chromatography on silica, again by using a 1:20 (*v*/*v*) mixture of ethyl acetate and hexanes as eluent; m.p. 113–115 °C (from hexanes); 6.9 g (89%). – <sup>1</sup>H NMR: δ = 10.24 (br. d, *J* = 13.0 Hz, 1 H), 7.73 (dd, *J* = 13.0, 1.0 Hz, 1 H), 7.0 (m, 2 H), 6.85 (td, *J* = 7.5, 1.0 Hz, 1 H), 6.81 (dd, *J* = 8.0, 1.0 Hz, 1 H), 3.84 (s, 3 H), 3.63 (br. s, 2 H). – <sup>19</sup>F NMR: δ = -60.2 (d, *J* = 1.0 Hz). – MS (CI): *m*/*z* (%) = 278 (12) [M<sup>+</sup> + NH<sub>4</sub>], 261 (100) [M<sup>+</sup> + 1], 260 (22) [M<sup>+</sup>], 241 (35), 119 (49). – C<sub>11</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub> (260.21): calcd. C 50.77, H 4.26; found C 50.98, H 3.89.

**4H-4-Quinolinone (3a):** The 3-aminoacrylate **2a** (2.5 g, 10 mmol) was added to polyphosphoric acid (15 mL) and the mixture was heated at 100 °C for 1 h. Neutralization, extraction, and subsequent chromatography (see above) afforded 4-quinolinone (4-hydroxyquinoline) as a white solid; m.p. and mixed m.p. 194–197 °C (ref.<sup>[47]</sup> 196–197 °C); 0.35 g (24%; 22% based on **1**).

**Methyl 1,4-Dihydro-1-methyl-4-oxoquinoline-3-carboxylate (3b):** Analogous treatment of aminoacrylate **2b** (10 mmol) afforded the ester **3b**; m.p. 189–191 °C (after sublimation; ref.<sup>[48]</sup> 185–186 °C); 0.50 g (44%; 23% based on 1). – <sup>1</sup>H NMR:  $\delta = 8.51$  (dd, J = 8.0, 1.5 Hz, 1 H), 8.46 (s, 1 H), 7.70 (ddd, J = 8.0, 7.0, 1.5 Hz, 1 H), 7.4 (m, 2 H), 3.92 (s, 3 H), 3.88 (s, 3 H). – MS (CI): *m/z* (%) = 218 (100) [M<sup>+</sup> + 1], 217 (18) [M<sup>+</sup>], 186 (35), 159 (55).

**Methyl 4-Isopropyloxy-1***H***-1,5-benzodiazepine-3-carboxylate (5b):** The aminoacrylate 4 (10 mmol) was added to a suspension of potassium carbonate (2.8 g, 20 mmol) in propan-2-ol (100 mL). The mixture was heated under reflux for 45 min, the solvent was then evaporated, and the product was isolated by chromatography; m.p. 170–171 °C (after sublimation); 0.18 g (7%; 6% based on 1). – <sup>1</sup>H NMR:  $\delta$  = 7.47 (d, *J* = 8.0 Hz, 1 H), 7.0 (m, 1 H), 6.88 (dd, *J* = 8.0, 1.5 Hz, 1 H), 6.82 (td, *J* = 8.0, 1.5 Hz, 1 H), 6.36 (dd, *J* = 8.0, 1.5 Hz, 1 H), 5.38 (br. d, *J* = 8.0 Hz, 1 H), 5.24 (sept, *J* = 6.3 Hz, 1 H), 3.69 (s, 3 H), 1.31 (d, *J* = 6.3 Hz, 6 H). – MS (CI): *m/z* (%) = 261 (79) [M<sup>+</sup> + 1], 260 (47) [M<sup>+</sup>], 219 (50), 186 (49), 119 (100). – C<sub>14</sub>H<sub>16</sub>N<sub>2</sub>O<sub>3</sub> (260.29): calcd. C 64.60, H 6.20; found C 65.19, H 5.66.

3-*N*-(Het)Arylhydrazino-2-(trifluoromethyl)acrylates and Cyclization Products Derived Therefrom: Methyl 3-*N*-Phenylhydrazino-2trifluoromethyl-2-propenoate (6a): A mixture of the acrylate 1 (7.4 g, 40 mmol) and phenylhydrazine (3.9 mL, 4.3 g, 40 mmol) was heated under reflux for 2 h. A 2:1 mixture of the 3-hydrazinoacrylate 6a and the 3-hydrazonopropionate 6a' was obtained; 8.6 g (83%) after purification by chromatography on silica. – <sup>1</sup>H NMR:  $\delta = 9.45$  (br. d, J = 10.9 Hz, 0.7 H), 7.79 (s, 0.3 H), 7.54 (d, J =10.9 Hz, 0.7 H), 7.26 (t, J = 7.9 Hz, 2 H), 7.0 (m, 1.6 H), 6.89 (t, J = 7.4 Hz, 0.3 H), 6.78 (d, J = 7.9 Hz, 1.4 H), 6.09 (br. s, 0.7 H), 4.13 (qd, J = 8.5, 1.0 Hz, 0.3 H), 3.80 (s, 0.9 H), 3.78 (s, 2.1 H).

# **FULL PAPER**

When the tautomeric mixture was triturated with pentanes, the pure hydrazino derivative **6a** crystallized; 39%; m.p. 97–99 °C (after sublimation). – <sup>19</sup>F NMR:  $\delta$  = –60.3 (s). – MS (CI): *m/z* (%) = 278 (22) [M<sup>+</sup> + NH<sub>4</sub>], 261 (94) [M<sup>+</sup> + 1], 260 (10) [M<sup>+</sup>], 241 (23), 94 (100). – C<sub>11</sub>H<sub>11</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub> (260.21): calcd. C 50.77, H 4.26; found C 50.91, H 4.20.

3-N'-(2-Fluorophenyl)hydrazino-2-trifluoromethyl-2-pro-Methyl penoate (6b): Analogously formed as a 3:2 hydrazino/hydrazono mixture (6b/6b') by using 2-fluorophenylhydrazine<sup>[49,50]</sup> (5.0 g, 40 mmol) in the above procedure; 9.1 g (82%) after purification by chromatography on silica.  $-{}^{1}$ H NMR:  $\delta = 9.42$  (br. d, J = 10.9 Hz, 0.6 H), 7.97 (br. s, 0.4 H), 7.55 (d, J = 10.9 Hz, 0.6 H), 7.41 (td, J = 8.5, 1.5 Hz, 0.4 H), 7.11 (d, J = 7.0 Hz, 0.4 H), 7.0 (m, 2 H), 6.9 (m, 1.2 H), 6.8 (m, 0.4 H), 6.40 (br. s, 0.6 H), 4.14 (qd, J = 8.5, 1.0 Hz, 0.4 H), 3.82 (s, 1.2 H), 3.79 (s, 1.8 H). - The pure hydrazino tautomer 6b (30%) crystallized from the mixture of 6b and 6b' upon trituration with pentanes; m.p. 102-104 °C (after sublimation). -<sup>19</sup>F NMR:  $\delta$  = -60.5 (s, 3 F), -134.8 (m, 1 F). - MS (CI): *m*/*z* (%) = 296 (4)  $[M^+ + NH_4]$ , 279 (42)  $[M^+ + 1]$ , 278 (19)  $[M^+]$ , 112 (100).  $- C_{11}H_{10}F_4N_4O_2$  (278.20): calcd. C 47.49, H 3.62; found C 47.29, H 3.66.

Methyl 3-N'-(2-Chlorophenyl)hydrazino-2-trifluoromethyl-2-propenoate (6c): Use of 2-chlorophenylhydrazine<sup>[50,51]</sup> (5.7 g, 40 mmol) in the above procedure again afforded a 3:2 hydrazino/hydrazono mixture (6c/6c'); 9.4 g (80%) after purification by chromatography on silica. – <sup>1</sup>H NMR:  $\delta$  = 9.46 (br. d, J = 10.9 Hz, 0.6 H), 8.21 (br. s, 0.4 H), 7.55 (d, J = 11.0 Hz, 0.6 H), 7.44 (dd, J = 8.5, 1.5 Hz, 0.4 H), 7.31 (dd, J = 7.8, 1.5 Hz, 0.6 H), 7.25 (td, J = 7.6, 1.5 Hz, 0.6 H), 7.22 (d, J = 7.6 Hz, 0.8 H), 7.16 (d, J = 7.3 Hz, 0.4 H), 6.9 (m, 1.2 H), 6.82 (td, J = 7.6, 1.5 Hz, 0.4 H), 6.61 (br. s, 0.6 H), 4.15 (qd, J = 8.5, 1.0 Hz, 0.4 H), 3.83 (s, 1.2 H), 3.81 (s, 1.8 H). - The pure, crystalline hydrazino tautomer 6c (27%) was isolated upon trituration of the tautomeric mixture of 6c and 6c' with pentanes; m.p. 97–99 °C (after sublimation). – <sup>19</sup>F NMR:  $\delta = -60.5$  (s). - MS (CI): m/z (%) = 314 (6), 312 (15) [M<sup>+</sup> +  $NH_4$ ], 297 (33), 296 (16), 295 (100)  $[M^+ + 1]$ , 294 (9)  $[M^+]$ . -C<sub>11</sub>H<sub>10</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub> (294.66): calcd. C 44.84, H 3.42; found C 44.51, H 3.33.

Methyl 3-*N*-(2,4,6-Trichlorophenyl)hydrazino-2-trifluoromethyl-2propenoate (6d): Use of 2,4,6-trichlorophenylhydrazine (8.5 g, 40 mmol) in the above procedure gave a 9:1 hydrazino/hydrazono mixture (6d/6d'); 13.1 g (90%) after extraction and drying. – <sup>1</sup>H NMR: δ = 9.43 (br. d, *J* = 10.9 Hz, 0.9 H), 7.70 (d, *J* = 10.9 Hz, 0.9 H), 7.59 (br. s, 0.1 H), 7.35 (s, 2.0 H), 6.46 (br. s, 0.9 H), 4.09 (quint., *J* = 8.2 Hz, 0.1 H), 3.82 (s, 0.3 H), 3.76 (br. s, 2.7 H). – The hydrazino tautomer 6d (77%) was obtained in a pure state by precipitation from a concentrated methanol solution by adding pentanes and subsequent recrystallization from methanol; m.p. 146–150 °C. – <sup>19</sup>F NMR: δ = –60.4 (s). – MS (CI): *m/z* (%) = 368 (7), 367 (30), 366 (25), 365 (93), 364 (55), 363 (100) [M<sup>+</sup> + 1], 362 (37) [M<sup>+</sup>]. – C<sub>11</sub>H<sub>8</sub>Cl<sub>3</sub>F<sub>3</sub>N<sub>2</sub>O<sub>2</sub> (363.55): calcd. C 36.34, H 2.22; found C 36.37, H 2.10.

Exposure of solutions of the pure hydrazino tautomers **6** to trace amounts of triethylamine led to rapid re-establishment of the hydrazino/hydrazono equilibria. The **6/6'** proportions thus obtained were identical with those initially found (2:1, 3:2, 3:2, and 9:1 in the case of compounds **6a**, **6b**, **6c**, and **6d**, respectively).

**Methyl 5-Fluoro-1-phenyl-1***H***-pyrazole-4-carboxylate (7a):** Potassium carbonate (2.8 g, 20 mmol) was added to a solution of the hydrazino/hydrazono derivative **6a/6a'** in propan-2-ol (200 mL). The mixture was heated under reflux for 45 min and then concentrated to dryness. The residue was purified by chromatography on

silica by using a 1:9 ( $\nu/\nu$ ) mixture of ethyl acetate and hexanes as eluent; 3.5 g (79%). An additional sublimation was required to obtain the product in a colorless state; m.p. 82–83 °C. – <sup>1</sup>H NMR:  $\delta$  = 7.96 (d, J = 2.5 Hz, 1 H), 7.64 (dd, J = 7.9, 1.3 Hz, 2 H), 7.51 (t, J = 7.9 Hz, 2 H), 7.41 (t, J = 7.9 Hz, 1 H), 3.88 (s, 3 H). – <sup>19</sup>F NMR:  $\delta$  = –123.3 (m). – MS (CI): m/z (%) = 238 (5) [M<sup>+</sup> + NH<sub>4</sub>], 221 (100) [M<sup>+</sup> + 1], 220 (37) [M<sup>+</sup>], 189 (26). – C<sub>11</sub>H<sub>9</sub>FN<sub>2</sub>O<sub>2</sub> (220.20): calcd. C 60.00, H 4.12; found C 59.94, H 4.19.

**Methyl 5-Fluoro-1-(2-fluorophenyl)-1***H*-pyrazole-4-carboxylate (7b): Prepared analogously from the hydrazino/hydrazono compound **6b/6b'** (5.6 g, 20 mmol); m.p. 75–77 °C (after sublimation); 2.9 g (61%). – <sup>1</sup>H NMR: δ = 8.01 (d, *J* = 2.5 Hz, 1 H), 7.54 (td, *J* = 7.6, 1.5 Hz, 1 H), 7.5 (m, 1 H), 7.3 (m, 2 H), 3.88 (s, 3 H). – <sup>19</sup>F NMR: δ = -121.9 (dd, *J* = 22.0, 2.5 Hz), -122.3 (m). – MS (CI): *m*/*z* (%) = 256 (7) [M<sup>+</sup> + NH<sub>4</sub>], 239 (100) [M<sup>+</sup> + 1], 238 (12) [M<sup>+</sup>], 207 (10). – C<sub>11</sub>H<sub>8</sub>F<sub>2</sub>N<sub>2</sub>O<sub>2</sub> (238.19): calcd. C 55.47, H 3.39; found C 55.39, H 3.16.

Methyl 5-Fluoro-1-(2-chlorophenyl)-1*H*-pyrazole-4-carboxylate (7c): Prepared analogously from the hydrazino/hydrazono compound 6c/ 6c' (5.9 g, 20 mmol); m.p. 64–65 °C (after sublimation); 2.0 g (39%). – <sup>1</sup>H NMR:  $\delta$  = 8.01 (d, *J* = 2.5 Hz, 1 H), 7.6 (m, 1 H), 7.5 (m, 3 H), 3.88 (s, 3 H). – <sup>19</sup>F NMR:  $\delta$  = –121.3 (t, *J* = 2.5 Hz). – MS (CI): *m/z* (%) = 257 (33), 256 (22), 255 (100) [M<sup>+</sup> + 1], 254 (31) [M<sup>+</sup>], 223 (47), 225 (17). – C<sub>11</sub>H<sub>8</sub>CIFN<sub>2</sub>O<sub>2</sub> (254.65): calcd. C 51.88, H 3.17; found C 51.97, H 3.17.

Methyl 5-Fluoro-1-(2,4,6-trichlorophenyl)-1*H*-pyrazole-4-carboxylate (7d): Prepared analogously from the hydrazino/hydrazono compound **6d/6d'** (7.3 g, 20 mmol); m.p. 86–88 °C (after sublimation); 1.3 g (20%).  $^{-1}$ H NMR:  $\delta = 8.06$  (d, J = 2.4 Hz, 1 H), 7.53 (s, 2 H), 3.89 (s, 3 H).  $^{-19}$ F NMR:  $\delta = -122.1$  (d, J = 2.4 Hz).  $^{-1}$ MS (CI): m/z (%) = 328 (5), 327 (32), 326 (20), 325 (100), 324 (43), 323 (93), 322 (34) [M<sup>+</sup>].  $^{-1}$ C<sub>11</sub>H<sub>6</sub>Cl<sub>3</sub>FN<sub>2</sub>O<sub>2</sub> (323.54): calcd. C 40.84, H 1.87; found C 40.90, H 2.17.

Methyl 5-Fluoro-1-(2-pyridyl)-1H-pyrazole-4-carboxylate (7e): In contrast to the four examples given above, the pyrazole 7e was prepared without isolating and purifying the corresponding hydrazino/ hydrazono intermediate 6e. Thus, the acrylate 1 (3.7 g, 20 mmol), 2-pyridylhydrazine<sup>[52]</sup> (2.2 g, 20 mmol), and pyridine (4.9 mL, 4.8 g, 60 mmol) were dissolved in propan-2-ol (200 mL), the mixture was heated under reflux for 2 h, and then concentrated to dryness. The residue was purified by chromatography on silica by using an ethyl acetate/hexanes mixture (1:3, v/v) as eluent; m.p. 96-98 °C (after sublimation); 0.95 g (21%). – <sup>1</sup>H NMR:  $\delta$  = 8.57 (dd, J = 4.8, 1.8 Hz, 1 H), 7.97 (d, J = 2.5 Hz, 1 H), 7.89 (ddd, J = 8.2, 7.6, 1.8 Hz, 1 H), 7.76 (dd, J = 8.2, 0.9 Hz, 1 H), 7.34 (ddd, J = 7.6, 4.8, 0.9 Hz, 1 H), 3.89 (s, 3 H).  $-{}^{19}$ F NMR:  $\delta = -118.8$  (d, J =2.5 Hz). – MS (CI): m/z (%) = 222 (100) [M<sup>+</sup> + 1], 221 (8) [M<sup>+</sup>], 190 (18). - C<sub>10</sub>H<sub>8</sub>FN<sub>3</sub>O<sub>2</sub> (221.19): calcd. C 54.30, H 3.65; found C 54.48. H 3.71.

**Methyl 5-Anilino-1-phenyl-1***H***-pyrazole-4-carboxylate (8a):** At 0 °C, butyllithium (10 mmol) in hexanes (7 mL) followed by methyl 5-fluoro-1-phenyl-1*H*-pyrazole-3-carboxylate (**7a**; 1.10 g, 5.0 mmol) were added to a solution of aniline (0.91 mL, 0.93 g, 10 mmol) in tetrahydrofuran (15 mL). After 1 h at 25 °C, the mixture was neutralized with ethereal hydrogen chloride, then concentrated and absorbed on silica. Elution with an ethyl acetate/hexanes mixture (1:7,  $\nu/\nu$ ) afforded product **8a** as colorless needles; m.p. 74–75 °C; 1.2 g (84%). – <sup>1</sup>H NMR: δ = 7.97 (s, 1 H), 7.82 (br. s, 1 H), 7.45 (d, *J* = 7.9 Hz, 2 H), 7.19 (t, *J* = 7.9 Hz, 2 H), 7.11 (t, *J* = 7.9 Hz, 1 H), 6.99 (t, *J* = 7.9 Hz, 2 H), 6.81 (t, *J* = 7.9 Hz, 1 H), 6.68 (d, *J* = 7.9 Hz, 2 H), 3.85 (s, 3 H). – MS (CI): m/z (%) = 294 (100) [M<sup>+</sup> + 1], 293 (20) [M<sup>+</sup>], 261 (48). – C<sub>17</sub>H<sub>15</sub>N<sub>3</sub>O<sub>2</sub> (293.32): calcd. C 69.61, H 5.15; found C 69.74, H 4.97.

**Methyl 5**-*p*-Anisidino-1-phenyl-1*H*-pyrazole-4-carboxylate (8b): Analogously, by using *p*-anisidine, product **8b** was prepared and isolated; m.p. 94–96 °C; 1.2 g (74%). – <sup>1</sup>H NMR:  $\delta$  = 7.93 (s, 1 H), 7.84 (br. s, 1 H), 7.36 (d, *J* = 7.4 Hz, 2 H), 7.16 (t, *J* = 7.4 Hz, 2 H), 7.10 (t, *J* = 7.4 Hz, 1 H), 6.65 (d, *J* = 9.0 Hz, 2 H), 6.52 (d, *J* = 9.0 Hz, 2 H), 3.86 (s, 3 H), 3.65 (s, 3 H). – MS (CI): *m/z* (%) = 324 (100) [M<sup>+</sup> + 1], 323 (25) [M<sup>+</sup>], 291 (33). – C<sub>18</sub>H<sub>17</sub>N<sub>3</sub>O<sub>2</sub> (323.35): calcd. C 66.86, H 5.30; found C 67.05, H 5.00.

**5-Fluoro-1-phenyl-1***H***-pyrazole-4-carboxylic Acid (9):** The ester **7a** (4.4 g, 20 mmol) was dissolved in a 3:1 ( $\nu/\nu$ ) mixture (200 mL) of tetrahydrofuran and water and lithium hydroxide monohydrate (0.84 g, 20 mmol) was added portionwise. The solution was kept at 25 °C for 2 h and then acidified with 2 M hydrochloric acid to pH 1. The thick white precipitate thus formed was collected by filtration, dried, and crystallized in the form of tiny needles from a mixture of ethyl acetate and hexanes; m.p. 166–167 °C; 2.8 g (68%). – <sup>1</sup>H NMR:  $\delta$  = 8.02 (d, *J* = 2.5 Hz, 1 H), 7.65 (d, *J* = 7.9 Hz, 2 H), 7.53 (t, *J* = 7.9 Hz, 2 H), 7.43 (t, *J* = 7.9 Hz, 1 H). – <sup>19</sup>F NMR:  $\delta$  = –121.6 (m). – MS (CI): *m/z* (%) = 224 (3) [M<sup>+</sup> + NH<sub>4</sub>], 207 (100) [M<sup>+</sup> + 1], 206 (46) [M<sup>+</sup>]. – C<sub>10</sub>H<sub>7</sub>FN<sub>2</sub>O<sub>2</sub> (206.18): calcd. C 58.26, H 3.42; found C 57.99, H 3.42.

5-Fluoro-N-1-diphenyl-1H-pyrazole-4-carboxamide (10a): (1-Benzo $triazolyloxy) tripyrrolidinophosphonium \qquad hexa fluorophosphate \cite{44}$ (2.6 g, 5.0 mmol), aniline hydrochloride (0.71 g, 5.5 mmol), and diisopropylethylamine (2.1 mL, 1.6 g, 13 mmol) were successively added to a solution of the 5-fluoropyrazole acid 9 (1.0 g, 5.0 mmol) in dichloromethane (25 mL). The resulting heterogeneous mixture was stirred for 2 h at 25 °C, then the solvent was evaporated and the crude product was purified by chromatography on silica by using a 1:4 (v/v) mixture of ethyl acetate and hexanes as eluent; m.p. 153– 155 °C (from hexanes); 0.58 g (41%).  $-{}^{1}$ H NMR:  $\delta = 8.06$  (d, J =2.7 Hz, 1 H), 7.7 (m, 2 H), 7.60 (d, J = 7.4 Hz, 2 H), 7.54 (br. s, 1 H), 7.53 (t, J = 7.9 Hz, 2 H), 7.43 (t, J = 7.4 Hz, 1 H), 7.37 (t, J = 7.9 Hz, 2 H), 7.15 (t, J = 7.4 Hz, 1 H). – <sup>19</sup>F NMR:  $\delta = -$ 127.3 (br. s). – MS (CI): m/z (%) = 299 (5) [M<sup>+</sup> + NH<sub>4</sub>], 282 (100)  $[M^+ + 1]$ , 281 (15)  $[M^+]$ , 189 (16). –  $C_{16}H_{12}FN_3O$  (281.29): calcd. C 68.32, H 4.30; found C 68.35, H 4.20.

**5-Fluoro-***N***-2-methoxyphenyl-1-phenyl-1***H***-pyrazole-4-carboxamide** (10b): Prepared analogously as described above by using *o*-anisidine hydrochloride (0.88 g, 5.5 mmol) instead of aniline hydrochloride; m.p. 109–111 °C (from ethyl acetate/hexanes); 0.57 g (37%). – <sup>1</sup>H NMR:  $\delta = 8.47$  (dd, J = 7.9, 1.5 Hz, 1 H), 8.30 (br. s, 1 H), 8.06 (d, J = 3.0 Hz, 1 H), 7.7 (m, 2 H), 7.53 (t, J = 7.9 Hz, 2 H), 7.43 (t, J = 7.9 Hz, 1 H), 7.08 (td, J = 7.9, 1.5 Hz, 1 H), 7.00 (td, J = 7.9, 1.5 Hz, 1 H), 6.91 (dd, J = 7.9, 1.5 Hz, 1 H), 3.93 (s, 3 H). – <sup>19</sup>F NMR:  $\delta = -127.5$  (br. s). – MS (CI): *mlz* (%) = 312 (100) [M<sup>+</sup> + 1], 311 (15) [M<sup>+</sup>], 189 (16). – C<sub>17</sub>H<sub>14</sub>FNO<sub>2</sub> (311.31): calcd. C 65.59, H 4.53; found C 65.58, H 4.69.

**5-Fluoro-***N***-3-methoxyphenyl-1-phenyl-1***H***-pyrazole-4-carboxamide** (**10c**): Prepared analogously as described above by using *m*-anisidine hydrochloride (0.88 g, 5.5 mmol) instead of aniline hydrochloride; m.p. 120–122 °C (from ethyl acetate/hexanes); 0.65 g (42%). – <sup>1</sup>H NMR:  $\delta = 8.05$  (d, J = 2.7 Hz, 1 H), 7.7 (m, 2 H), 7.5 (m, 3 H), 7.43 (tt, J = 7.4, 1.2 Hz, 1 H), 7.39 (t, J = 2.4 Hz, 1 H), 7.25 (t, J = 4.1 Hz, 1 H), 7.04 (ddd, J = 8.2, 1.8, 0.9 Hz, 1 H), 6.71 (ddd, J = 8.2, 2.4, 0.9 Hz, 1 H), 3.82 (s, 3 H). – <sup>19</sup>F NMR:  $\delta = -127.3$  (br. s). – MS (CI): *m*/*z* (%) = 329 (3) [M<sup>+</sup> + NH<sub>4</sub>], 312 (100) [M<sup>+</sup> + 1], 311 (23) [M<sup>+</sup>], 189 (15). – C<sub>17</sub>H<sub>14</sub>FN<sub>3</sub>O<sub>2</sub> (311.31): calcd. C 65.59, H 4.53; found C 66.01, H 4.74.

5-Fluoro-*N*-4-methoxyphenyl-1-phenyl-1*H*-pyrazole-4-carboxamide (10d): Prepared analogously as described above by using *p*-anisidine hydrochloride (0.88 g, 5.5 mmol) instead of aniline hydrochloride; m.p. 163–165 °C (from ethyl acetate/hexanes); 0.90 g (58%).  $^{-1}$ H NMR: δ = 8.04 (br. s, 1 H), 7.66 (dd, *J* = 7.9, 1.5 Hz, 2 H), 7.5 (m, 4 H), 7.4 (m, 2 H), 6.9 (m, 2 H), 3.81 (s, 3 H).  $^{-19}$ F NMR: δ =  $^{-127.5}$  (br. s).  $^{-}$ MS (CI): *m/z* (%) = 329 (1) [M<sup>+</sup> + NH<sub>4</sub>], 312 (100) [M<sup>+</sup> + 1], 311 (23) [M<sup>+</sup>], 189 (15).  $^{-}$ C<sub>17</sub>H<sub>14</sub>FN<sub>3</sub>O<sub>2</sub> (311.32): calcd. C 65.59, H 4.53; found C 65.22, H 4.53.

#### Acknowledgments

This work was financially supported by the Schweizerische Nationalfond zur Förderung der wissenschaftlichen Forschung, Bern (grant 20-49'307-96).

- M. Schlosser, B. Spahić, C. Tarchini, Angew. Chem. 1975, 87, 346–347; Angew. Chem. Int. Ed. Engl. 1975, 14, 365–366.
- [2] M. Schlosser, B. Spahić, *Helv. Chim. Acta* 1980, 63, 1223–1235.
- [3] M. Schlosser, R. Dahan, S. Cottens, *Helv. Chim. Acta* 1984, 67, 284–288.
- [4] B. Spahić, T. M. T. Truong-Nguyen, M. Schlosser, *Helv. Chim. Acta* 1980, 63, 1236–1241.
- <sup>[5]</sup> B. Spahić, M. Schlosser, *Helv. Chim. Acta* 1980, 63, 1242–1256.
   <sup>[6]</sup> K. Kondo, S. Cottens, M. Schlosser, *Chem. Lett.* 1984, 2149–
- 2156.
  <sup>[7]</sup> G.-q. Shi, S. Cottens, S. A. Shiba, M. Schlosser, *Tetrahedron* 1992, 48, 10569–10574.
- <sup>[8]</sup> G.-q. Shi, M. Schlosser, *Tetrahedron* **1993**, *49*, 1445–1456.
- [9] G.-q. Shi, Q. Wang, M. Schlosser, Tetrahedron 1996, 52, 4403–4410.
- [10] Y. Bessière, D. Ngoc-Huê Savary, M. Schlosser, *Helv. Chim. Acta* 1977, 60, 1739–1746.
- <sup>[11]</sup> T. M. T. Truong-Nguyen, H. Togo, M. Schlosser, *Tetrahedron* 1994, 50, 7827–7836.
- <sup>[12]</sup> S. N. Gosh, M. Schlosser, J. Fluorine Chem. 1994, 67, 53–56.
- <sup>[13]</sup> G.-q. Shi, S. Takagishi, M. Schlosser, *Tetrahedron* **1994**, *50*, 1129–1134.
- <sup>[14]</sup> U. Mävers, F. Berruex, M. Schlosser, *Tetrahedron* **1996**, 52, 3223–3228.
- <sup>[15]</sup> M. Schlosser, H. Keller, *Liebigs Ann.* 1995, 1587–1589.
- <sup>[16]</sup> H. Keller, M. Schlosser, Tetrahedron 1996, 52, 4637-4644.
- <sup>[17]</sup> M. Schlosser, H. Keller, S.-i. Sumida, J. Yang, *Tetrahedron Lett.* 1997, 38, 8523–8526.
- <sup>[18]</sup> A. Pasetti, F. Tarli, D. Sianesi, *Gazz. Chim. Ital.* **1968**, *98*, 277–289; *Chem. Abstr.* **1968**, *69*, 77016s.
- [19] I. L. Knunyants, V. V. Shokina, V. V. Tyuleneva, Dokl. Akad. Nauk SSSR 1966, 169, 594–597; Dokl. Chem. (Engl. Transl.) 1966, 169, 722–725; Chem. Abstr. 1966, 65, 15218e.
- <sup>[20]</sup> G. Wittig, M. Schlosser, Chem. Ber. 1961, 94, 1373–1383.
- <sup>[21]</sup> M. El-Khoury, Q. Wang, M. Schlosser, *Tetrahedron Lett.* **1996**, *37*, 9047–9048.
- [22] R. G. Jones, J. Am. Chem. Soc. 1947, 69, 2346-2350.
- [23] R. Belcher, M. Stacey, A. Sykes, J. C. Tatlow, J. Chem. Soc. 1954, 3846–3851.
- <sup>[24]</sup> Y. Kobayashi, I. Kumadaki, Y. Hanzawa, M. Mimura, *Chem. Pharm. Bull.* **1975**, *23*, 636–639.
- <sup>[25]</sup> R. L. Wydra, S. E. Patterson, L. Strekowski, J. Heterocycl. Chem. 1990, 27, 803–805.
- <sup>[26]</sup> L. Strekowski, S.-Y. Lin, H. Lee, J. C. Mason, *Tetrahedron Lett.* 1996, 37, 4655–4658.
- [27] Y. Kobayashi, I. Kumadaki, Y. Hirose, Y. Hanzawa, J. Org. Chem. 1974, 39, 1836–1838.
- <sup>[28]</sup> Y. Kobayashi, I. Kumadaki, Acc. Chem. Res. 1978, 11, 197– 204.
- <sup>[29]</sup> M. S. South, J. Heterocycl. Chem. 1991, 28, 1013–1016.
- [<sup>30]</sup> J. Kollonitsch, L. Barash, J. Am. Chem. Soc., **1976**, 98, 5591–5593; R. B. Silverman, M. A. Levy, J. Org. Chem. **1980**, 45, 815–818; C. W. Fearon, J. A. Rodkey, R. H. Abeles, Biochemistry **1982**, 21, 3790–3794; P. J. Reider, R. S. Eichen, P. Davis, V. J. Grenda, A. J. Zambito, E. J. J. Grabowski, J. Org. Chem. **1987**, 52, 3326–3334.
- <sup>[31]</sup> W. Dmowski, J. Fluorine Chem. 1982, 20, 589-598.
- <sup>[32]</sup> W. Dmowski, Synthesis 1983, 396–397.
- <sup>[33]</sup> T. Kubota, K. Yamamoto, T. Tanaka, *Chem. Lett.* **1983**, 167–168.

## **FULL PAPER**

- <sup>[34]</sup> L. Strekowski, R. L. Wydra, M. T. Celga, A. Czarny, D. B. Harden, S. E. Patterson, M. E. Battiste, J. M. Coxon, *J. Org. Chem.* **1990**, *55*, 4777–4779.
- <sup>[35]</sup> L. Strekowski, J. L. Mokrosz, V. A. Honkan, A. Czarny, M. T. Cegla, R. L. Wydra, S. E. Patterson, R. E. Schinazi, *J. Med. Chem.* **1991**, *34*, 1739–1746.
- <sup>[36]</sup> L. Strekowski, S. E. Patterson, L. Janda, R. L. Wydra, D. B. Harden, M. Lipowska, M. T. Cegla, *J. Org. Chem.* **1992**, *57*, 196–201.
- <sup>[37]</sup> L. Janda, J. Nguyen, S. E. Patterson, J. Heterocycl. Chem. 1992, 29, 1753–1756.
- <sup>[38]</sup> A. S. Keselyov, L. Strekowski, *Tetrahedron Lett.* 1994, 35, 7597–7600.
- <sup>[39]</sup> Y. Komitori, M. Hojo, R. Masuda, T. Ikemura, Y. Mori, *Tetrahedron Lett.* **1993**, *34*, 5135–5138.
- [40] J. Ichikawa, Y. Wada, T. Okauchi, T. Minami, J. Chem. Soc., Chem. Commun. 1997, 1537–1538.
- <sup>[41]</sup> T. Ishihara, Y. Okada, M. Kuroboshi, T. Shinozaki, T. Ando, *Chem. Lett.* **1988**, 819–822; X.-f. Shi, T. Ishihara, H. Yamanaka, J. T. Gupton, *Tetrahedron Lett.* **1995**, *36*, 1527–1530.
- <sup>[42]</sup> X.-q. Tang, C.-m. Hu, J. Fluorine Chem. 1995, 73, 229–131.
- <sup>[43]</sup> Other methods for the preparation of fluoro-, difluoromethyl-, or trifluoromethyl-substituted pyrazoles: C. L. Bumgardner, J. C. Sloop, *J. Fluorine Chem.* 1992, 56, 141–146; R. J. Linderman, K. S. Kirolles, *Tetrahedron Lett.* 1989, 30, 2049–2052; J.

- W. Lyga, R. M. Patera, J. Heterocycl. Chem. 1990, 27, 919–921; A. Said, Inform. Chim. 1985, 261, 251–257; Chem. Abstr. 1986, 104, 68282q; S. Iwata, J. Namekata, K. Tanaka, K. Mitsuhashi, J. Heterocycl. Chem. 1991, 28, 1971–1976; P. P. K. Claire, P. L. Coe, C. J. Jones, J. A. McLeverty, J. Fluorine Chem. 1991, 51, 283–289; B. J. Gaede, L. L. McDermott, J. Heterocycl. Chem. 1993, 30, 49–54; J. P. Boullon, C. Ates, Z. Janousek, H. G. Viehe, Tetrahedron Lett. 1993, 34, 5075–5078.
- <sup>[44]</sup> J. Coste, D. Le-Nguyen, B. Castro, *Tetrahedron Lett.* **1990**, *31*, 205–208.
- <sup>[45]</sup> S. Jeganathan, M. Tsukamoto, M. Schlosser, Synthesis 1990, 109–111.
- [46] Q. Wang, H.-x. Wei, M. Schlosser, Eur. J. Org. Chem., 1999, 3263–3268.
- [47] J. Franck, A. R. Katritzky, J. Chem. Soc., Perkin Trans. 2 1976, 1428–1431.
- <sup>[48]</sup> N. Takenari, A. Hideo, M. Toru, N. Mitsuo, Jap. Pat. 2166375 (to Sumitomo Chem. Co., filed on 28 Jan. 1970, issued on 18 Oct. 1973); *Chem. Abstr.* 1974, *80*, 3486a.
- <sup>[49]</sup> C. Willgerodt, Ber. Dtsch. Chem. Ges. 1891, 24, 1660–1662.
- [50] See also: I. T. Barnish, P. D. Callaghan, M. S. Gibson, J. Chem. Soc., Perkin Trans. 1 1974, 215–219.
- <sup>[51]</sup> H. Suschitzky, J. Chem. Soc. 1953, 3326–3327.
- <sup>[52]</sup> H. Gregory, L. F. Wiggins, J. Chem. Soc. **1949**, 2546–2549. Received July 13, 1999 [O99424]