

SYNTHETIC APPROACH TO STEMODIN---A NOVEL STEREOCONTROLLED CONSTRUCTION OF THE STEMODANE SYSTEM BY THE SUCCESSIVE INTRAMOLECULAR DIELS-ALDER REACTIONS

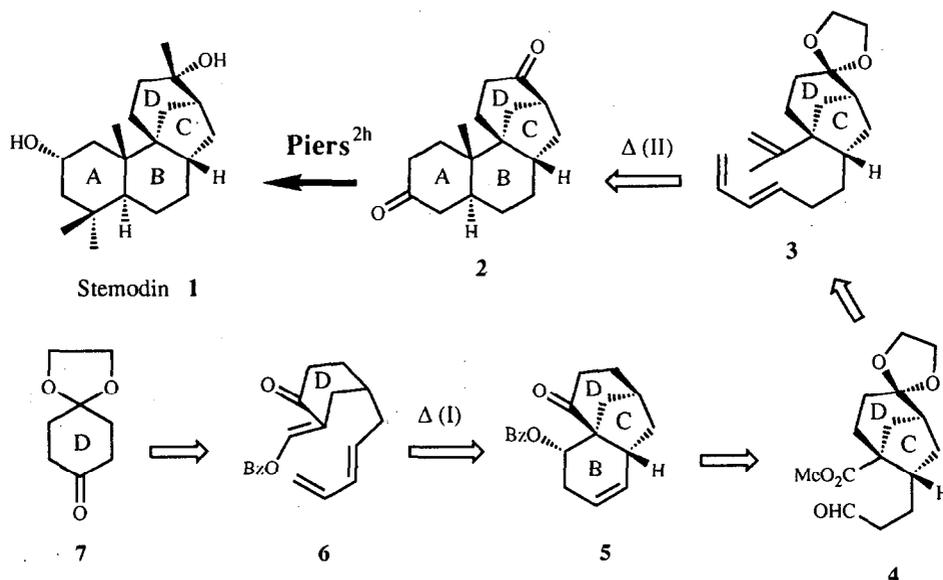
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Abstract: The tetracyclic compound **18**, possessing the stemodane skeleton, was synthesized from 1,4-cyclohexanedione monoethylene ketal **7**. The key steps, **6**→**5** and **16**→**18**, involve intramolecular Diels-Alder reaction.

Because of unique carbon skeleton and alleged medicinal properties of *Stemodia maritima* L.,¹ the stemodane group of diterpenoids has received considerable attention as targets for synthesis.² Herein we report an efficient synthesis of the stemodane nucleus **18**, as part of a study directed toward the total synthesis of stemodin **1**. Our strategy based upon the synthetic analysis is shown below. Namely, the intramolecular Diels-Alder reaction of the triene **6**, obtained from **7**, would give the tricyclic ketone **5**, which could be transformed into **3** via the aldehyde **4**. The second intramolecular Diels-Alder reaction of **3** would afford the tetracyclic compound **2**, convertible into stemodin **1**.

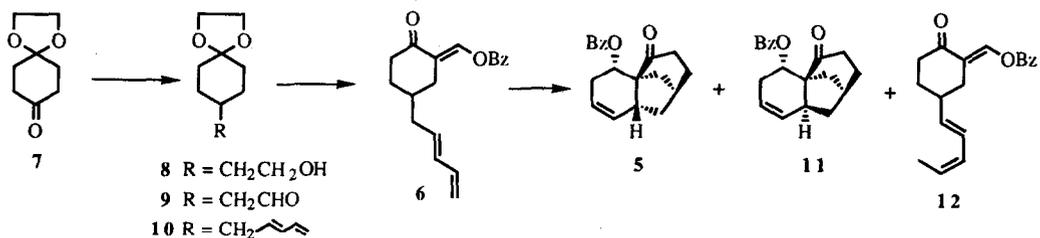
Scheme 1



In order to explore the feasibility of the designed synthetic strategy, stereocontrolled synthesis of **18** was first examined as a model experiment for the construction of **2**. The requisite triene **6** for the initial Diels-Alder reaction was readily prepared as described below. A commercially available **7** was converted into the alcohol **8** in the usual manner (1. (EtO)₂P(O)CH₂CO₂Et / NaH in DME, 2. H₂ / 10% Pd-C in EtOAc, 3. LAH in THF, 95% overall yield), which was followed by oxidation³ with SO₃-Py / DMSO in the presence of Et₃N to afford the aldehyde **9** (77%). Selective preparation of the (E)-dienes, developed by Yamamoto⁴

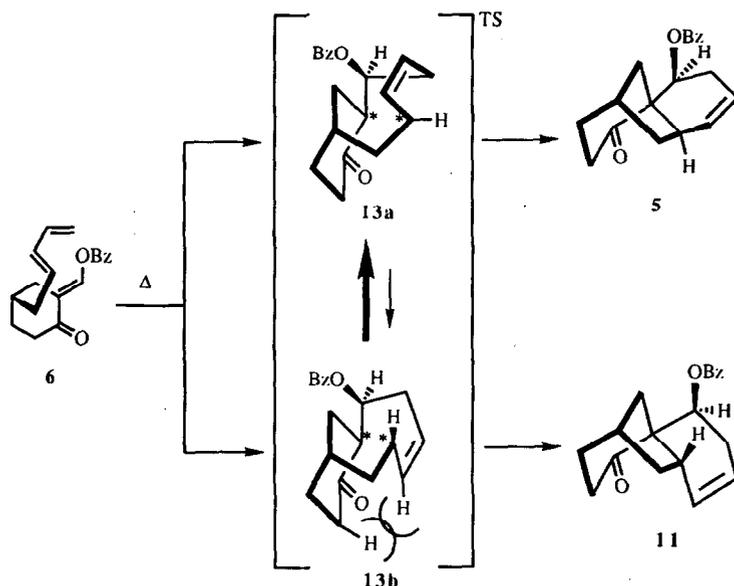
($\text{Ph}_2\text{P}(\text{O})\text{CH}_2\text{CH}=\text{CH}_2 / ^n\text{BuLi} / \text{HMPA}$ in THF), was applied to **9** to give rise to **10** (61%),⁵ which was converted into the triene **6** in three steps (1. 10% HClO_4 in THF, 2. $\text{HCO}_2\text{Et} / \text{NaH}$ in C_6H_6 , 3. $\text{Bz}_2\text{O} / \text{Py} / \text{DMAP}$ in CH_2Cl_2 , 87%). Heating **6** in the presence of a catalytic amount of methylene blue⁶ in *o*-dichlorobenzene at 180 °C for 6 h produced a mixture of two tetracyclic compounds (**5** and **11**) and the isomerization product **12**⁷ (70%) in a ratio of 7 : 1 : 0.5. The structure of the major component **5** was determined by X-ray analysis.⁸

Scheme 2



The preferred formation of **5** could be due to a "concerted but nonsynchronous" transition state for the cyclization.^{6b,9} The steric congestion between the olefinic hydrogen and the axial hydrogen in the nine membered ring transition state **13b**, first partially formed, makes it less favorable than the alternative transition state **13a** which affords the desired product **5**.

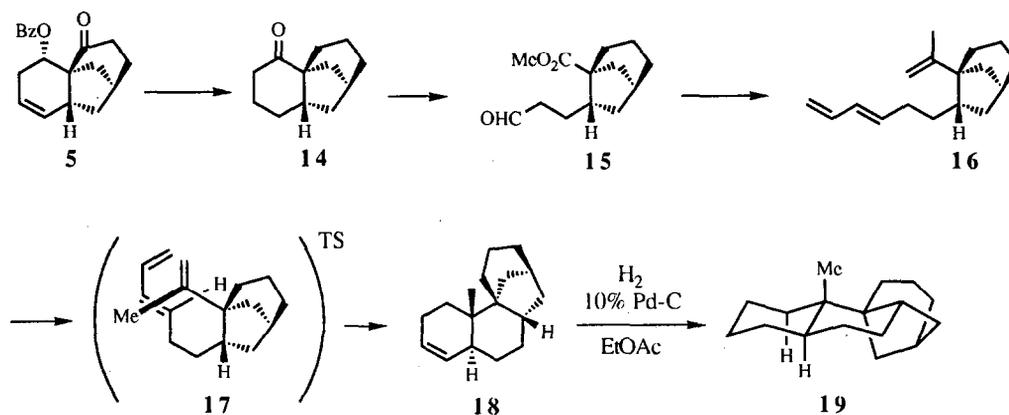
Figure 1



Our synthetic efforts were next focused on the ring opening and subsequent introduction of diene and dienophile portions for the second intramolecular Diels-Alder reaction. Toward this end, the carbonyl moiety of **5** was removed by Wolff-Kishner reduction¹⁰ and the resulting alcohol was transformed into the ketone **14** in two steps (1. $\text{H}_2 / 10\% \text{Pd-C}$ in EtOAc , 2. $\text{PCC} / \text{SiO}_2$ in CH_2Cl_2 , 68%). The treatment of **14** with Vedejs' reagent¹¹ in THF (-78°C, 1.5 h), followed by oxidation with $\text{Pb}(\text{OAc})_4$ in $\text{MeOH-C}_6\text{H}_6$ (1:3 v/v) at 0°C for 1.5 h gave **15** (35%). After Wittig reaction in the same manner as previously, transformation of the resulting ester to the triene **16** was carried out by two steps, methylation with MeLi (28%)¹² and dehydration with SOCl_2 in Py (86%). An

intramolecular Diels-Alder reaction was conducted in the presence of methylene blue⁶ in toluene at 220 °C for 96 h in a sealed tube to produce the tetracyclic compound **18** in 89% yield as a single product. The stereochemistry of **18** was deduced on the preference of the *exo*-conformer **17** in the transition state during the thermolysis and the spectral evidence of **19**; particularly due to the similarity of the half-band width ($Wh/2=1.00$ Hz) of the angular methyl group with the proposed¹³ that ($Wh/2=0.91$ Hz) of the methyl group possessing two anticoplanar protons at the C₁₀ position in *trans*-decalin derivatives.

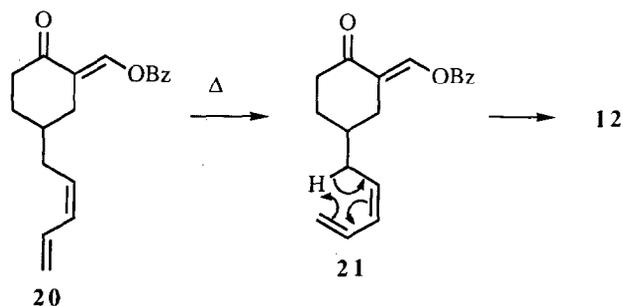
Scheme 3



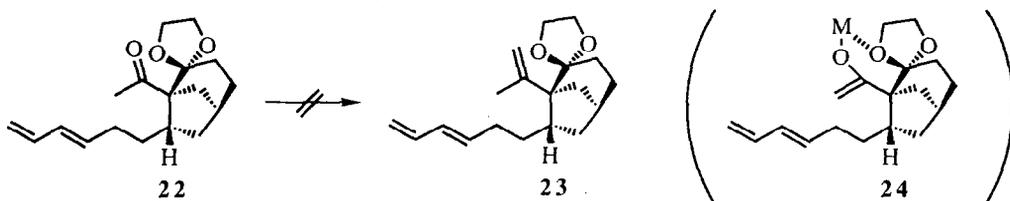
Application of this novel methodology to the synthesis of stemodin **1** is in progress.

References and Notes

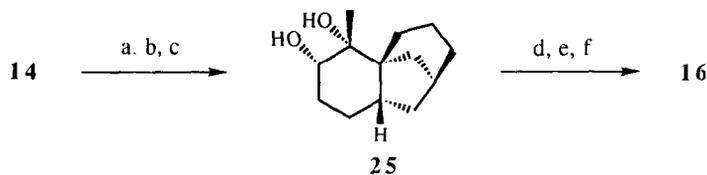
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3. J. R. Parikh and W. E. Doering, *J. Am. Chem. Soc.*, **89**, 5505 (1967).
4. J. Ukai, Y. Ikeda, N. Ikeda, and H. Yamamoto, *Tetrahedron Lett.*, **24**, 4029 (1983).
5. E/Z ratio (17:1) was determined after the initial intramolecular Diels-Alder reaction (**6** → **5** + **11** + **12**).
6. (a) D. F. Taber and S. A. Saleh, *J. Am. Chem. Soc.*, **102**, 5085 (1980); (b) D. F. Taber, C. Campbell, B. P. Gunn, and I-C. Chiu, *Tetrahedron Lett.*, **22**, 5141 (1981).
7. The compound **12** was obtained through 1,5-sigmatropic rearrangement of **20** under the thermal conditions.



8. The compound **5** crystallizes in the monoclinic $P2_1/n$ space group with $a=10.743$ (1), $b=12.322$ (3), $c=12.108$ (1) Å, $\alpha=90.0$, $\beta=90.719$ (12), $\gamma=90.0^\circ$, $V=1602.8$ (4), and $Z=4$. The final coordinates were solved by direct methods and refined by block diagonal least squares methods with $R=0.071$, $R_w=0.062$. Final crystallographic coordinates are deposited in Cambridge Crystallographic Data Center.
9. E. Ciganek, *Organic Reactions*, **32**, 1 (1984), and refs cited therein.
10. All attempts (a: $\text{Ph}_3\text{P}^+\text{MeBr}^-$, $^n\text{BuLi}$; b: CH_2Br_2 , Zn, TiCl_4 ; c: MeLi , hexane then SOCl_2) in the conversion (**22**→**23**) led to failure, and the starting material **22** was recovered unchanged probably due to the metal chelated intermediate **24**.



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12. By using different synthetic route shown below, it is possible to obtain fair yield of the triene **16**.



Reagents and reaction conditions

a) MeLi , hexane; b) SOCl_2 , Py; c) OsO_4 , NMO, Et_2O , H_2O , 33% from **14**; d) NaIO_4 , Et_2O , H_2O ; e) $\text{Ph}_2\text{P}(\text{O})\text{CH}_2\text{CH}=\text{CH}_2$, $^n\text{BuLi}$, THF, HMPA; f) $\text{Ph}_3\text{P}^+\text{MeBr}^-$, $^n\text{BuLi}$, DME, 39% from **25**.

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