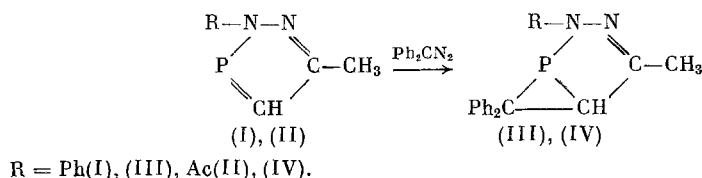


REACTION OF SUBSTITUTED 5-METHYL 1,2,3-DIAZAPHOSPHOLES  
AND 1,2,3-DIAZAARSOLES WITH DIPHENYLDIAZOMETHANE AND NITRONES

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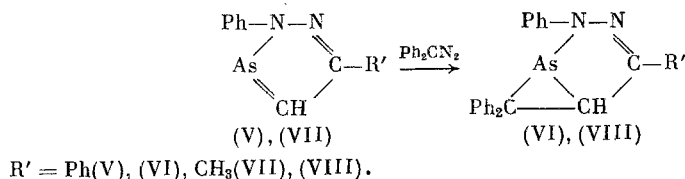
UDC 542.91:547.794

Earlier, by reaction of 5-methyl-2-phenyl(acetyl)-1,2,3-diazaphospholes (I) and (II) with diphenyldiazomethane in various solvents, bicyclic phosphiranes (III) and (IV) were obtained [1].



However, it is known that reaction of  $\alpha,\beta$ -unsaturated carbonyl compounds with diazo compounds in pyridine leads to [2 + 3]-cycloaddition to diazaphospholes, the reaction of diazaphosphole (II) with diphenyldiazomethane in pyridine was carried out. It was found that also in this solvent 2-acetyl-5-methyl-6,6-diphenyl-1-phospha-2,3-diazabicyclo[3.1.0]hex-3-ene (IV) is formed. Thus, the nature of the solvent does not influence the reaction of diazaphospholes with  $\text{Ph}_2\text{CN}_2$ .

Earlier [3] we found that heating 2,5-diphenyl-1,2,3-diazaarsole (V) with  $\text{Ph}_2\text{CN}_2$  without solvent results in bicyclic arsirane (VI)



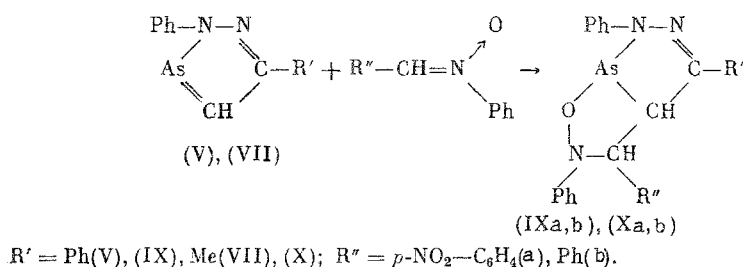
It was of interest to study the behavior of 5-methyl-2-phenyl-1,2,3-diazaarsole (VII) in the reaction with  $\text{Ph}_2\text{CN}_2$  and compare the reactivity of diazaphospholes and diazaarsoles. Diazaarsole (VII) reacts with  $\text{Ph}_2\text{CN}_2$  without heating in the absence of solvent forming 4-methyl-2,6,6-triphenyl-1-arsa-2,3-diazabicyclo[3.1.0]hex-3-ene (VIII) the structure of which was proved by its mass spectrum, elemental analysis, and IR and PMR spectra.

Thus, diazaphospholes (I) and (II) and diazaarsoles (V) and (VII) react with  $\text{Ph}_2\text{CN}_2$  analogously. Comparison of the reactivity of 5-methyl-2-phenyl-1,2,3-diazaphosphole (I) and 5-methyl-2-phenyl-1,2,3-diazaarsole (VII), which have the same substituents, showed that diazaarsole (VII) is more active toward  $\text{Ph}_2\text{CN}_2$ . It is interesting that with methylphenyldiazomethane diazaphospholes (I) and (II) and diazaarsoles (V) and (VII) behave differently and form different products, which will be reported elsewhere.

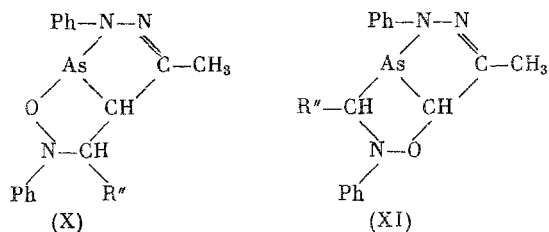
The stabilities of phosphirane- and arsirane-containing bicycles are varied. Thus, 4-methyl-2-phenyl(acetyl)-6,6-diphenyl-1-phospha-2,3-diazabicyclo[3.1.0]hex-3-enes (III) and (IV) are stable in crystalline form and in solutions. They can be recrystallized from MeOH. 2,4,6,6-Tetraphenyl-1-arsa-2,3-diazabicyclo[3.1.0]hex-3-ene (VI) is unstable in solutions and upon prolonged storage it decomposes to the initial diazaarsole. 4-Methyl-2-phenyl-6,6-diphenyl-1-arsa-2,3-diazabicyclo[3.1.0]hex-3-ene (VIII) is stable in crystalline form. In benzene,  $\text{CHCl}_3$ , and  $\text{CCl}_4$  it resinifies, but it can be recrystallized from EtOH/MeCN.

A. M. Butlerov Chemical Institute, V. I. Ul'yanov-Lenin Kazan State University. Translated from *Izvestiya Akademii Nauk SSSR, Seriya Khimicheskaya*, No. 12, pp. 2819-2821, December, 1987. Original article submitted May 14, 1986.

Cycloaddition of diazaarsoles to the As=C bond was first realized in 1979 [4] with the reaction of 2,5-diphenyl-1,2,3-diazaarsole (V) with nitrones. The structure of the product (IXb) was proved by X-ray analysis



In continuation of these investigations we studied the reaction of nitrones with 5-methyl-2-phenyl-1,2,3-diazaarsole (VII). Crystalline products were obtained which by mass spectrum and elemental analysis are the 1:1 adducts (X) and (XI)



The PMR spectra of the obtained compounds are analogous to the spectra of the compounds (IXa, b) synthesized earlier, in which the methine proton signals appeared as singlets because the C-H bonds lie in mutually perpendicular planes. Therefore, for the reaction products of 5-methyl-2-phenyl-1,2,3-diazaarsole with nitrones, 2,7-diphenyl-4-methyl-6-R''-1-arsa-8-oxa-2,3,7-triazabicyclo[3.3.0]oct-3-ene (X) is the preferred structure. Moreover, if products of type (XI) were formed, then the methine signals in the spectrum would not differ so strongly. PMR spectrum (acetone-d<sub>6</sub>,  $\delta$ , ppm) of compound (Xa): 2.16 s (CH<sub>3</sub>), 4.60 s (H-C-N), 6.23 s (H-C-As), 7.16 m (C<sub>6</sub>H<sub>5</sub>). PMR spectrum (acetone-d<sub>6</sub>,  $\delta$ , ppm) of compound (Xb): 2.12 s (CH<sub>3</sub>), 4.53 s (H-C-N), 6.03 s (H-C-As), 7.14 m (C<sub>6</sub>H<sub>5</sub>).

#### EXPERIMENTAL

IR spectra were recorded on a Specord 75 IR instrument, PMR spectra on a Tesla BS-467 NMR spectrometer (working frequency 60 MHz) relative to HMDS. Mass spectra were recorded on a Finnegan MAT-212 instrument at 50-120°C and Vu = 50 V.

2-Acetyl-5-methyl-6,6-diphenyl-1-phospha-2,3-diazabicyclo-[3.1.0]hex-3-ene (IV). To 1.5 g Ph<sub>2</sub>CN<sub>2</sub> in 10 ml Py at 10°C, 1.1 g diazaphosphole (II) was added dropwise in an atmosphere of dry nitrogen. In 4 h the color of the mixture changed from crimson to yellow. After 24 h the crystalline product was filtered off and washed with ether. There was obtained 2.1 g (88%) of compound (IV) with m.p. 143-145°C.

Mixing with an authentic sample does not decrease the melting point. IR spectrum ( $\nu$ , cm, white mineral oil): 515 w, 710 s, 725 s, 780 m, 1000 w, 1040 w, 1208 m, 1279 w, 1320 s, 1350 s, 1400 v. s, 1498 m, 1605 m (Ph), 1660 v. s (C=O), and others. Found: M 308 (mass spectrum). C<sub>18</sub>H<sub>17</sub>N<sub>2</sub>OP. Calculated: M 308.

4-Methyl-2-phenyl-6,6-diphenyl-1-arsa-2,3-diazabicyclo[3.1.0]hex-3-ene (VIII). To 1.6 g Ph<sub>2</sub>CN<sub>2</sub> 1.8 g of diazaarsole (VII) was added dropwise. Evolution of nitrogen occurred and a crystalline product was formed. The mixture was kept for 1.5 days, then the crystals were filtered off and washed with petroleum ether. There was obtained 2.2 g (69%) of compound (VIII) with m.p. 129-131°C (from EtOH/MeCN). IR spectrum ( $\nu$ , cm<sup>-1</sup>, white mineral oil): 475 w, 495 w, 505 w, 555 w, 630 s, 787 v. s, 830 v. s, 977 w, 1000 w, 1019 w, 1190 m, 1285 s, 1480 v. s, 1578 s, and others. PMR spectrum (acetone-d<sub>6</sub>,  $\delta$ , ppm): 2.00 s (CH<sub>3</sub>), 4.65 s (CH), 7.06 m (C<sub>6</sub>H<sub>5</sub>). Found, %: C 68.32, H 4.90, N 7.16; M 386. C<sub>22</sub>H<sub>19</sub>AsN<sub>2</sub>. Calculated, %: C 68.40, H 4.96, N 7.25; M 386.

Reaction of 5-Methyl-2-phenyl-1,2,3-diazaarsole (VII) with C-p-nitrophenyl-N-phenyl-nitrone. To a suspension of 0.42 g nitrone in  $\text{CH}_2\text{Cl}_2$  0.38 g of diazaarsole (VII) was added dropwise. The mixture was kept for 3 days, then the crystals were filtered off and washed with  $\text{CH}_2\text{Cl}_2$ . There was obtained 0.65 g (81%) of compound (Xa) with m.p. 156.5–157.5°C. IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ , white mineral oil): 505 w, 575 m, 630 m, 668 m, 695 s, 740 s, 815 m, 985 w, 1025 w, 1170 w, 1248 m, 1310 s, 1485 s, 1505 s, 740 s, 815 m, 985 w, 1025 w, 1170 w, 1248 m, 1310 s, 1485 s, 1505 s, 1587 s, and others. Found, %: C 56.59, H 4.22, N 11.70.  $\text{C}_{22}\text{H}_{19}\text{AsN}_4\text{O}_3$ . Calculated, %: C 57.11, H 4.13, N 12.11. The mass spectrum has an intense signal at M 220 (diazaarsole (VII)) and M 242 (nitrone).

Reaction of 5-Methyl-2-phenyl-1,2,3-diazaarsole (VII) with C,N-diphenylnitrone. To a suspension of 1.28 g nitrone in  $\text{CH}_2\text{Cl}_2$  1.43 g of diazaarsole (VII) was added dropwise. The mixture was kept for 40 days and then alcohol was added until dissolution. Upon cooling crystals precipitated which were filtered off and washed with EtOH. There was obtained 1.3 g (48%) of (Xb) with m.p. 109–111.5°C (from MeOH–petroleum ether). IR spectrum ( $\nu$ ,  $\text{cm}^{-1}$ , white mineral oil): 500 w, 570 m, 687 s, 730 s, 740 s, 1020 w, 1070 w, 1170 w, 1238 m, 1305 m, 1480 s, 1577 m, and others. Found, %: C 63.23, H 4.89, N 10.00.  $\text{C}_{22}\text{H}_{20}\text{AsN}_3\text{O}$ . Calculated, %: C 63.32, H 4.83, N 10.07.

#### CONCLUSIONS

1. Reaction of 2-acetyl-5-methyl-1,2,3-diazaphosphole and 5-methyl-2-phenyl-1,2,3-diazaarsole with diphenyldiazomethane leads to formation of phosphirane- and arsirane-containing bicyclic compounds.

2. In reaction with nitrones 5-methyl-2-phenyl-1,2,3-diazaarsole, as well as 2,5-diphenyl-1,2,3-diazaarsole, forms analogous [2 + 3]-cycloadducts.

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