A NEW SYNTHETIC METHOD FOR THE FRAMEWORK OF ASPIDOSPERMA ALKALOIDS USING SINGLET OXYGEN CHEMISTRY

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Abstract—A singlet oxygen adduct of 1-benzyloxycarbonyl-1,2-dihydro-5-(2-methyl-1,3-dioxolan-2yl)pyridine (21) was treated with indole in the presence of stannous chloride to yield directly the complex compound 22, which will act as a suitable starting material for the synthesis of various kinds of indole alkaloids. Further transformation from 22, including a new device for construction of the aspidosperma framework, was investigated and pentacyclic compounds (25) (Y = H and OH) were synthesized in moderate yields. The structure of 25 (Y = H) = 38 was confirmed by correlation with 1-acetyl-20-deethylaspidospermidine (44). Determination of the structure of a by-product (39) was carried out by single crystal X-ray analysis.

In 1979 we reported a novel type of reaction using singlet oxygen chemistry, in which a C-C bond was formed on the piperidine ring.¹ The reaction consists of two successive operations which are carried out in one pot (Chart 1). (i) The sensitized photo-oxygenation reaction is applied to dihydropyridine derivatives (2)^{2,3} obtained from variously substituted pyridines (1). For convenience to the next step, dichloromethane† is used as a solvent and the oxygenation is performed by bubbling dry oxygen at $ca - 50^{\circ}$ into a solution of 2 using a 500 W halogen lamp as a light source and methylene blue as a sensitizer.⁴ The (4+2) adduct (3) of singlet oxygen is formed as a single product, as shown when the mixture is examined by TLC.[‡] The stereochemistry of the endoperoxide bridge in 3 is determined to be trans with respect to the substituent R². This high selectivity is probably due to the exclusive

approach from the opposite side of \mathbb{R}^2 of the bulky dycstuff on which the singlet oxygen is generated.

(ii) Without isolation of 3, a nucleophile is added to the cooled solution, followed by the addition of a suspension of stannous chloride in ethyl acetate. The temperature of the reaction mixture is then gradually raised to $ca - 10^{\circ}$, resulting in the formation of a reaction product. Carbon nucleophiles, such as enol ethers, indole, furan and pyrrole derivatives can be caused to react with 3 to produce mostly 2,6-*cis* substituted derivatives (4), accompanied by occasional formation of 2,6-*trans* derivatives (5) as by-products. Predominant formation of 4 is postulated to originate from the S_N2 attack of a nucleophile on the activated centre at the C-6 position of the possible intermediate (6).

The important features of this reaction are: (i)

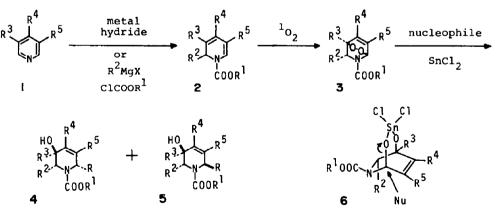


Chart 1.

 $^{1}_{1}$ ¹H-NMR spectral study of 3 (R¹ = Me, R² = R³ = R⁴ = R⁵ = H) and 3 (R² = PhCOOCH₂, R³ = R⁴ = H, R⁵ = CN, N—COPh) confirmed their structures. The latter is stable crystals.⁴ introduction of the substituent R is achieved without the use of carbanions; (ii) variously substituted products can be obtained by selecting a combination of starting pyridines and R^2 which designates hydrogen, alkyl, alkenyl or alkynyl groups; (iii) stereochemically definite compounds can be prepared in good yields; and (iv) further structural elaboration is possible by modifying an allyl alcoholic function of 4. Thus, using

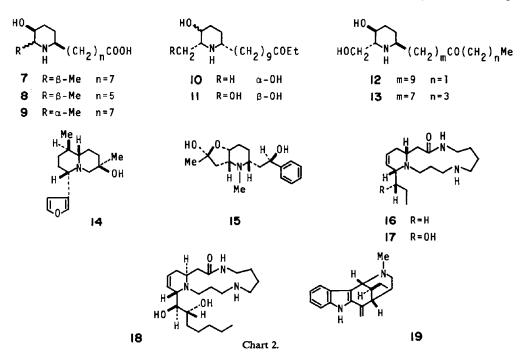
 $[\]dagger$ Commercial CH₂Cl₂ was washed with H₂O, dried over CaCl₂ and distilled in the presence of P₂O₅. Otherwise reaction with MeOH instead of a carbon nucleophile is inevitable.

reaction products 4 and 5 as starting materials, we have accomplished total synthesis of a variety of natural products (Chart 2), such as carpamic acid (7),⁵ azimic acid (8),⁵ pseudocarpamic acid (9),⁶ prosafrinine (10),⁶ prosophyllin (11),⁷ prosopinine (12),⁸ isoprosopinine B (13),⁸ nupharolutine (14),⁹ sedacryptine (15),¹⁰ 'deoxypalustrine' (16),¹¹† 'palustrine' (17)¹²† cannabisativine (18),¹³ and epiulein (19).¹⁴ A preliminary study of the synthesis of an aspidosperma type of indole alkaloids ^{15,16} has also been reported.¹⁷

Table 1. Reduction of 20 with metal hydrides in the presence of PhCH₂OCOCl at -70°

Reagent	Solvent	Ratio of 21 : 26 + 27*
NaBH	THF-MeOH	2:5
LiAl (t-BuO) ₃ H	THF	3:2
NaAl (MeOCH ₂ CH ₂ O) ₂ H ₂	THF	4:1

Ratio was estimated from analysis of ¹H-NMR signals.



In order to develop an alternative, more effective procedure for the preparation of the framework of aspidosperma alkaloids, we envisaged a synthetic route, outlined in Chart 3, applying the above reaction as the key step. For the realization of this scheme, a crucial problem had to be solved beforehand: how in general to prepare 1-alkoxycarbonyl-5-alkyl-1,2-dihydropyridine derivatives, 2 ($R^1 = R^5 = alkyl$, $R^2 =$ $R^3 = R^4 = H$), since the usual procedure^{3a} furnished only 1-alkoxycarbonyl-3-alkyl derivatives, 2 ($R^1 = R^3 = alkyl$, $R^2 = R^4 = R^5 = H$).¹⁸‡ Fortunately, the C-2 position of our starting material (20) was sterically congested owing to the neighbouring 2'methyl-1',3'-dioxolanyl group, so it was anticipated that the desired dihydropyridine derivative (21) could be obtained in competition with the formation of 1,4and 1,2-dihydro derivatives (26 and 27) if the reduction was carried out using bulky reagent in the presence of benzyl chloroformate. As shown in Table 1, sodium bis(2-methoxyethoxy)aluminium hydride (Vitride) was found to be suitable and rapid separation over silica gel afforded unstable 21 in 58% yield, while a crude mixture of 26 and 27 was formed in 16% yield.

The sensitized photo-oxygenation reaction on 21 proceeded readily, as expected, and the subsequent treatment with indole in the presence of stannous chloride produced the requisite compound (22) in 57% yield, and a by-product (28) in 2% yield.

Our main objective in this synthesis is to convert 22 to 23 having an activated C-2 side chain at the N_b group, followed by a novel type of ring closure reaction $(23 \rightarrow 24 \rightarrow 25)$ which consists of two successive C—C bond formations at the C-3 and C-2 positions in the indole ring during a single operation. This provides a new methodology for the construction of the aspidosperma alkaloids. In practice, our intention was realized as described below.

At first, removal of the OH group in 22 was attempted to simplify further steps (Chart 4). The ethylene ketal group in 22 was hydrolysed to 29 in 93% yield and its methanesulphonyl (mesyl) derivative (30) was hydrogenated over 10% Pd–C. Addition of a small amount of triethylamine was essential for the selective cleavage of the O-mesyl group located at the special position of a neighbour of the α,β -unsaturated ketone system. Both the double bond and the N-protecting group remained intact and 31 was obtained in 93.5% yield. Further hydrogenation was carried out by catalysis of PtO₂ to afford 32 and 33 in 56% and 17% yields, respectively. The former compound exhibiting H-2 and H-3 signals at $\delta 6.45$ (d, J = 5.5 Hz) and $\delta 2.93$ (ddd, J = 10.5, 5.5, 5.5 Hz) in its ¹H-NMR spectrum is

t We synthesized the compounds bearing proposed structures. Lack of identity required reinvestigation of the structure of natural products.

[‡]Multi-step preparation of 1-acyl-5-ethyl-1,2-dihydropyridines was reported.¹⁹

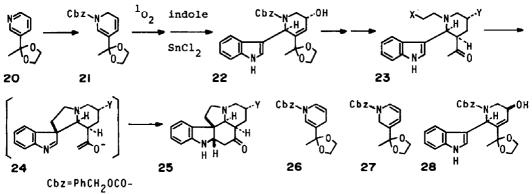
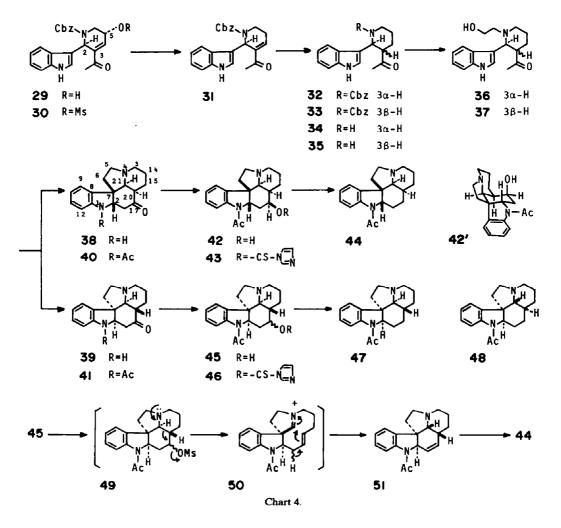


Chart 3.

assigned as a *cis* product, since the indolyl group at the α -position of the N-acylpiperidine system is determined to be axially oriented²⁰ and, hence, H-3 should be axially situated judging from the above coupling pattern. The benzyloxycarbonyl function of **32** and **33** was split off by 10% Pd–C catalysed hydrogenation to furnish **34** (70% yield) and **35** (57% yield) which were treated with ethylene oxide to give **36** and **37** in 44% and 75% yields, respectively.

As the key reaction of the present synthesis, the Omesylate of the cis-36 was prepared by treatment with methanesulphonyl chloride and potassium carbonate at room temperature. The desired cyclization reaction was effected only by reaction of potassium bis(trimethylsilyl)amide²¹ with the crude mesylate in tetrahydrofuran initially at -70° and then at room temperature (*ca* 20°) to afford the two products, **38** and **39**, in 35% and 3% yields. The same operation applied to *trans*-**37** also produced the same derivatives, **38** and **39**, in 14% and 24% yields, respectively. Both these derivatives exhibit almost identical mass spectra with fragment ions m/z 268, 226 and 96, suggesting that the pentacyclic structure can be assigned to **38** and **39** on the basis of the report of Wenkert *et al.*²²



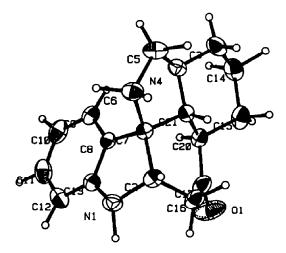
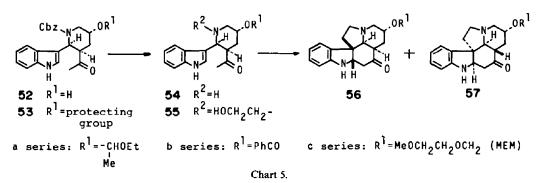


Fig. 1. The ORTEP drawing of 39.

The structural determination of **38** was attained by correlation with 1-acetyl-20-deethylaspidospermidine (**44**),²³ obtained in our previous study.¹⁷ N-Acetate (**40**), derived from **38** in 73% yield, was reduced with sodium borohydride to give a single compound, **42**, in 81% yield. In its ¹H-NMR spectrum, a H-17 signal appears at δ 3.96 as a broad singlet and this indicates the stereochemistry of the OH group as shown, assuming that **42** possesses a conformational structure **42**' analogous to that of (-)-17-O-methylaspidospermine benzene solution of **45**-mesylate with 1.8diazabicyclo[5.4.0]undec-7-ene (DBU) for 12 hr. Successive hydrogenation of 51 took place readily with PtO₂ in 99% yield but, surprisingly, the product was found to be 44. Therefore, a crystal of 39 was subjected to X-ray crystallographic analysis and the stereostructure (Fig. 1) confirmed the stereochemical arrangement of 39 as a hitherto unencountered pattern in the natural products. As the structure of 45 is definitely established, the above coincidence is now interpreted to occur due to bond cleavage between C-20 and C-21 (49 \rightarrow 50), followed by closing the rings again to form the thermodynamically favourable 51.

Next, the synthesis of pentacyclic cyclization products (56) bearing an OH function at the D ring was attempted starting from the catalytic hydrogenation of 29 with PtO_2 in dimethoxyethane to afford a single product, 52, in 70% yield (Chart 5). This time, the Pt catalyst approaches the double bond exclusively from the opposite side of the indolyl group, probably due to coordination of the catalyst with the OH group. The OH function of 52 was protected by 1-ethoxy-1-ethyl, benzoyl, or methoxyethoxymethyl (MEM) groups and 53a, 53b and 53c were each converted to 56 and 57, as shown in Table 2. A benzoate (55b) directly afforded 56 $(R^1 = H)$ in 41% yield, which was also obtained from 56a in 78% yield by acid hydrolysis [1% HClin MeOH-H₂O (10:1), room temp, 10 min] and from 56c in 39% yield by treatment with titanium tetrachloride²⁶ in dichloromethane at 0° for 1 hr. Compound 57 ($R^1 = H$) was prepared from 57a by hydrolysis with 1% HCl in a mixture of methanol and water (9:1).



hydrobromide, which was determined by X-ray analysis.²⁴ The OH function of **42** was removed by Barton's procedure.²⁵ Compound **42** was treated with 1,1'-thiocarbonyldiimidazole in hot dichloroethane to afford **43** in 62% yield and this was reduced with tri-nbutyltin hydride to produce, in 93% yield, the final compound, **44**, which was identical with the previous sample in all respects.

In order to get information about the structure of another cyclization product, 39, the same transformation sequence as above was applied to it by way of 41 (87% yield), 45 with uncharacterized configuration of the OH group (70% yield) and 46 (65% yield), then converting into the corresponding N-acetyldeoxo derivative, 47 (71% yield from 46). However, 47 was found to be different from the known 48.¹⁷ Removal of the OH group of 45 was tried by chance through a dehydration route. A compound (51) having a disubstituted double bond, evidenced by its ¹H-NMR signals, was obtained in 46% yield by gently refluxing a In conclusion, the present work demonstrates further the utility of our reaction involving the singlet oxygen adducts of 1,2-dihydropyridines in the preparation of complex molecules and opens a way to total synthesis of the aspidosperma type of indole alkaloids of biological interest.

EXPERIMENTAL

All m.p.s were taken on Yanagimoto micro-melting point apparatus and are uncorrected. IR spectra were recorded on Hitachi 215 IR spectrophotometer. ¹H-NMR spectra were determined on Varian EM-390 (90 MHz) spectrometer with TMS as an internal standard. Chemical shifts (δ) are expressed in ppm. Coupling constants are reported in Hz and splitting patterns are designated as s (singlet), d (doublet), t (triplet), q (quartet), m (multiplet) and br (broad). Mass spectra were obtained on a Hitachi RMS-4 spectrometer with a direct inlet system operating at 70 eV. Merck silica gel PF₂₅₄ was used for preparative TLC. Elemental analyses were performed by Shionogi Research Laboratories. For experiments requiring

33	3	\$3	56 57
∕∕OEt, TsOH · Pyridine CH ₂ Cl ₂ , room temp, 16 hr, 89%	H ₃ , 10% Pd-C 95% EtOH, room temp, 14 hr, 67%	Toom temp, 14 hr, 81%	 (i) MsCl, K₂CO₃, CH₂Cl₂ room temp, 15 hr (ii) (Me₃Sl)₂NK, THF, -60°, 15 min, 0°, 1.5 hr 28%
PhCOCl, Pyridine, room temp, 14 hr, 89%	H ₂ , 10% Pd-C 95% EtOH, room temp, 14 hr, 73%	, McOH room temp, 14 hr, 88.5%	 (i) MsCl, K₁CO₃, CH₂Cl₂, room temp. 18 hr (ii) (Me₃Sl₃)₂ NK, THF, -60°, 20 min, room temp. 1 hr 41% (R¹ = H) None
MEMCI, iso-Pr ₂ NEt CH ₂ Cl ₂ , room temp, 15 hr, 87%	H ₂ , 10% Pd–C 95% EtOH room temp, 2 hr, 75%	room temp, 14 hr, 85%	 (i) MsCl, K₂CO₃, CH₂Cl₂ room temp, 18 hr (ii) (Me₃Sl)₂NK, THF, - 70°, 30 min, room temp, 45 min 32.5% None

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dry solvents, CH_2Cl_2 and dichloroethane were distilled over P_2O_3 . THF and Et_3N were distilled over LAH. Pyridine was distilled over CaH₂. Xylene was distilled over Na.

Reduction of 20. To a stirred soln of 20 (204 mg, 1.27 mmol) in dry THF (10 ml) was added a 30-35% toluene soln of PhCH₂OCOCl (0.8 ml, ca 1.6 mmol) at - 70°. After 30 min, a soln of 70% NaAl (MeOCH₂CH₂O)₂H₂ in toluene (451 mg, 1.56 mmol) in THF (4 ml) was added dropwise at -70° with stirring. After 30 min, the mixture was poured into H₂O and extracted with Et₂O. The extract was dried over Na₂SO₄ and evaporated at reduced pressure at ca 30°. The residue was separated by preparative TLC (CH₂Cl₂). Elution of the upper layer gave a mixture of 26 and 27 (59 mg, 16%) as an unstable oil. ¹H-NMR (CDCl₃) δ 1.47 (s, Me), 2.70–2.87 (m, H-4 of 26), 3.63-4.10 (m, OCH2CH2O), 4.32 (br s, H-2 of 27), 6.73 (br d, J = 7.5 Hz, H-6 of 27) and 6.85-7.10 (m, H-2 and H-6 of 26). Elution of the lower layer gave 21 (214 mg, 58%) as an unstable oil. IR v^{film}_{max} 1710, 1665 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.49 (3H, s, Me), 3.63-4.10 (4H, m, OCH₂CH₂O), 4.34 (2H, dd, J = 3.5, 2 Hz, H-6), $5.22(2H, s, OCH_2Ph)$, 5.55(1H, brd, J = 10.5 Hz, H-4), 5.94 (1H, ddd, J = 10.5, 3.5, 2 Hz, H-5), 6.92 (1H, br s, H-2), 7.36 (5H, s, Ph).

Sensitized photo-oxygenation of 21. O2 gas was bubbled into a soln of 21 (310 mg, 1.03 mmol) and methylene blue (41 mg) in purified CH₂Cl₂ (see footnote on page 2115) (100 ml) at -60° for 30 min, while the mixture was irradiated externally by an Iwasaki 500 W halogen lamp (JD 110 V 500 W-M5). After terminating the irradiation, a soln of indole (133 mg, 1.13 mmol) in purified CH2Cl2 (5 ml) and a soln of SnCl2 (251 mg, 1.32 mmol) in dry EtOAc (100 ml) were successively added at -60° with stirring. The mixture was stirred at -60° for 10 min and the cooling bath was removed. After stirring at -60 to -5° for 2 hr, sat NaHCO₃ aq was added and the ppt was removed by filtration. The CH₂Cl₂ extract was washed with H2O, dried over Na2SO4 and evaporated in vacuo to leave a crystalline solid, which was recrystallized from Et₂O-CH₂Cl₂ to afford 22 (224 mg) as colourless prisms, m.p. 192.5-193°. Separation of the mother liquor by preparative TLC (hexane-EtOAc, 1:1) gave a further yield of 22 (32 mg) and a byproduct, 28 (9 mg, 2%) as colourless glass. The total yield of 22 was 256 mg (57%). Compound 22: (Found: C, 69.04; H, 6.05; N, 6.51%. Calc for $C_{25}H_{26}N_2O_5$: C, 69.11; H, 6.03; N, 6.45%.) IR v_{max}^{KBr} 3355, 1660 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 1.28 (3H, s, IR VM Me), 3.31 (1H, dd, J = 15, 4 Hz, H-6), 3.53-3.97 (4H, m, OCH₂CH₂O), 5.23 (2H, s, OCH₂Ph), 6.30 (1H, d, J = 3 Hz, H-4), 6.33 (1H, s, H-2), 7.35 (s, Ph). MS m/z 434 [M]⁺, 416, 345, 343, 299, 255, 237, 91 (base peak). Compound 28: IR vmax 3420, 1676 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 1.26(3H, s, Me), 2.99(1H, dd, J = 13.5, 9 Hz, H-6), 3.47-3.90 (4H, m, OCH₂CH₂O), 5.17 (2H, s, OCH, Ph), 6.19 (2H, br s, H-2 and H-4), 7.32 (s, Ph). MS m/z 434 [M]⁺, 416, 345, 343, 299, 255, 237, 91 (base peak).

Deketalization of 22. A soln of 22 (50 mg, 0.12 mmol) in MeOH (10 ml) containing 10% HCl (2 ml) was stirred at room temp for 10 min. The mixture was neutralized with sat NaHCO₃ aq and extracted with CH₂Cl₂. The extract was washed with H₂O, dried over Na₂SO₄ and evaporated under reduced pressure. The residue was purified by preparative TLC (hexane-EtOAc, 1:1) to afford 29 (42 mg, 93%) as a colourless glass. IR v_{max}^{KB} 3420, 1678 cm⁻¹. ¹H-NMR (CDCl₃) δ 2.21 (3H, s, Me), 3.10 (1H, dd, J = 15, 3 Hz, H-6), 4.15 (1H, br d, J = 15 Hz, H-6), 4.23 (1H, br s, H-5), 5.16 (2H, s, OCH₂Ph), 6.58 (1H, br s, H-2), 6.93 (d, J = 5 Hz, H-4), 7.33 (br s, Ph). MS m/z 390 [M]⁺ (base peak), 372, 299, 255, 237, 195, 91.

Formation of 31. To an ice-cooled soln of 29 (281 mg, 0.72 mmol) in CH₂Cl₂ (5 ml) containing Et₃N (0.5 ml), MeSO₂Cl (0.1 ml, 1.3 mmol) was added with stirring. After 15 min, sat NaHCO₃ aq was added. The CH₂Cl₂ extract was washed with 10% HCl, sat NaHCO₃ aq and H₂O, and dried over Na₂SO₄. The solvent was evaporated to dryness *in vacuo* at *ca* 30°. A soln of the dried residue of 30 (354 mg) in MeOH (15 ml) containing Et₃N (1 ml) was hydrogenated over 10% Pd-C (71 mg) for 1.5 hr at atmospheric pressure. The catalyst was removed by filtration and the filtrate evaporated in vacuo. After the work-up described above, the residue was purified by

preparative TLC (hexane-EtOAc, 1:1) to afford 252 mg (93.5%) of 31. IR v_{max}^{KBr} 3375, 1695–1664 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.17 (s, Me), 2.97 (1H, ddd, J = 13.5, 10.5, 5 Hz, H-6), 4.01 (1H, br dd, J = 13.5, 5 Hz, H-6), 5.13 (1H, d, J = 13.5 Hz) and 5.24 (1H, d, J = 13.5 Hz) (OCH₂Ph), 6.55 (1H, br s, H-2), 7.31 (br s, Ph). MS m/z 374 [M]⁺, 283, 239 (base peak), 195, 168, 91.

Hydrogenation of 31. A soln of 31 (1.583 g, 4.23 mmol) in MeOH (30 ml) was hydrogenated over PtO₂ (124 mg) for 2 hr at atmospheric pressure. The catalyst was removed by filtration and the filtrate was evaporated in vacuo. The residue was chromatographed over silica gel (50 g) with CH_2Cl_2 . The first elution afforded the cis isomer, 32 (883 mg, 56%), as colourless prisms, m.p. 153.5-154.5°, from MeOH. (Found : C, 73.42; H, 6.26; N, 7.41%. Calc for $C_{23}H_{24}N_2O_3$: C, 73.38; H, 6.43; N, 7.44%.) IR v_{max}^{KBr} 3410, 1678 cm⁻¹. ¹H-NMR (CDCl₃, 70°) $\delta 2.04$ (s, Me), 2.68 (1H, ddd, J = 13.5, 13.5, 3 Hz, H-6), 2.93 $(1H, ddd, J = 10.5, 5.5, 5.5 Hz, H-3), 3.93(1H, brd, J = 13.5 Hz, H_3), 3.93(1H, brd, J = 13.5 Hz$ H-6), 5.27 (2H, s, OCH₂Ph), 6.45 (1H, d, J = 5.5 Hz, H-2), 7.36 (s, Ph). MS m/z 376 [M]⁺, 285, 241 (base peak), 225, 199, 91. The second elution afforded 33 (268 mg, 17%) as colourless prisms (from MeOH), m.p. 132-134°. (Found: C, 73.52; H, 6.31; N, 7.37%. Calc for $C_{23}H_{24}N_2O_3$: C, 73.38; H, 6.43; N, 7.44%.) IR v_{max}^{EB} 3380, 1714, 1692 cm ⁻¹. ¹H-NMR (CDCl₃) δ 2.30 (3H, s, Me), 4.02 (1H, br d, J = 13.5 Hz, H-6), 5.20 (2H, s, OCH₂Ph), 6.32 (1H, br s, H-2), 7.32 (s, Ph). MS m/z 376 [M]⁺, 285, 241 (base peak), 225, 199, 91.

Removal of the Cbz group from 32. A soln of 32 (776 mg, 2.06 mmol) in MeOH (35 ml) containing 10% Pd–C (122 mg) was hydrogenated for 4 hr at atmospheric pressure. The catalyst was removed by filtration and the filtrate evaporated *in vacuo* to leave a crystalline solid, which was recrystallized from CHCl₃ to afford 34 (316 mg) as slightly yellow needles, m.p. 146–147°. The mother liquor was purified by preparative TLC (CH₂Cl₂-MeOH, 9:1) to give a further yield (36 mg) of 34. Total yield was 352 mg (70%). (Found: C, 74.16; H, 7.51; N, 11.39%. Calc for C₁₃H₁₈N₂O: C, 74.35; H, 7.49; N, 11.56%,) IR v^{KBP} 3455, 1695 cm⁻¹. ¹H-NMR (CDCl₃-CD₃OD, 1:4) δ 1.77(s, Me), 4.42(1H, d, J = 3.5 Hz, H-2), 7.01 (s, H-2 of indole). MS *m/z* 242 [M]⁺ (base peak), 223, 199, 171.

Removal of the Cbz group from 33. A mixture of 33 (74 mg, 0.20 mmol) in 95% EtOH (20 ml) containing 10% HCl (0.8 ml) was hydrogenated over 10% Pd–C (51 mg) at atmospheric pressure. After 4 hr, the catalyst was filtered off and the filtrate neutralized with an ion-exchanger resin IRA-400 (OH⁻). The solvent was removed *in vacuo* to leave an oil, which was purified by preparative TLC (CH₂Cl₂-MeOH, 9:1) to afford 27 mg (57%) of 35. IR v_{max}^{CMCl₃} 3500, 3340, 1705 cm^{-1.} ¹H-NMR (CDCl₃) δ 1.71 (s, Me), 4.05 (1H, d, J = 10 Hz, H-2), 6.83–7.35 (4H, m, aromatic H), 7.63–7.84 (1H, m, aromatic H). MS *m/z* 242 [M]⁺, 223, 199, 171 (base peak).

Treatment of 34 with ethylene oxide. A soln of 34 (84 mg, 0.35 mmol) and ethylene oxide (2 ml) in MeOH (15 ml) was stirred at room temp for 14 hr. After removal of the solvent, the residue was purified by preparative TLC (CH_2CI_2 -MeOH, 9:1) to afford 36 (44 mg, 44%) as slightly yellow prisms, m.p. 152.5-153.5° from MeOH. (Found: C, 71.44; H, 7.96; N, 9.7%). Calc for $C_{17}H_{22}N_2O_2$: C, 71.30; H, 7.74; N, 9.78%.) IR ν_{max}^{RM} 3410, 3250, 1692 cm^{-1.} ¹H-NMR (CDCI₃) δ 1.93 (s, Me), 2.93-3.37 (1H, m, H-3), 4.91 (1H, d, J = 5 Hz, H-2), 7.00-7.43 (4H, m, aromatic H), 7.54-7.78 (1H, m, aromatic H). MS *m*/z 286 [M]⁺, 255 (base peak).

Treatment of 35 with ethylene oxide. A soln of 35 (26 mg, 0.11 mmol) and ethylene oxide (1 ml) in MeOH (10 ml) was stirred at room temp for 6 hr. After removal of the solvent, the residue was purified by preparative TLC (CH₂Cl₂-MeOH, 9:1) to afford 23 mg (75%) of 37 as a slightly yellow oil. IR v_{max}^{BB} 3400, 1700 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.61 (s, Me), 2.63 (br s, OH), 3.65 (1H, d, J = 10.5 Hz, H-2), δ .87–7.39 (4H, m, aromatic H), 7.67–7.93 (1H, m, aromatic H). MS m/z 286 [M]⁺, 255 (base peak).

Cyclization of 36. To a mixture of 36 (100 mg, 0.35 mmol) and anhyd. K_2CO_3 (551 mg) in dry CH₂Cl₂ (25 ml), MeSO₂Cl (63 mg, 0.55 mmol) was added at room temp with stirring. After 16

hr, H₂O was added and the mixture extracted with CH₂Cl₂. The extract was washed with H₂O, dried over Na₂SO₄ and evaporated to dryness in vacuo at ca 30°. To a soln of this dried residue (106 mg) in dry THF (20 ml), (Me₃Si)₂NK²¹ (241 mg, 1.21 mmol) was added with stirring at -70° under N₂. The mixture was stirred for 45 min at -70° and then for 1 hr at room temp. The mixture was diluted with brine, extracted with CH₂Cl₂-MeOH (9:1) and the extract dried over Na₂SO₄. After evaporation of solvent, the residue was separated by preparative TLC (hexane-EtOAc, 1:1) to afford 38 (33 mg, 35%) and 39 (3 mg, 3%) both as colourless prisms. Compound **38**: m.p. 136–137° (**MeOH**) (lit.²² m.p. 113–115°). (Found : C, 75.95; H, 7.40: N, 10.36%. Calc for $C_{1,7}H_{20}N_2O$: C, 76.08; H, 7.51; N, 10.44%.) IR ν_{max}^{KBr} 3350, 1700 cm⁻¹. IR $\nu_{max}^{CHCl_3}$ 3410, 1712 ¹. ¹H-NMR (CDCl₃) δ 2.12(d, J = 3 Hz, H-21), 2.52(dd, J cm⁻ = 17, 3 Hz, H-16), 3.08 (dd, J = 17, 3 Hz, H-16), 3.93 (1H, dd, J = 3, 3 Hz, H-2), 6.59 (1H, d, J = 7 Hz, aromatic H), 6.79 (1H, dd, J = 7, 7 Hz, aromatic H), 6.93-7.22 (2H, m, aromatic H). MS m/z 268 [M]⁺, 226, 130, 96 (base peak). Compound 39: m.p. 172-173.5° (MeOH-H₂O). (Found : C, 76.39; H, 7.70; N, 10.42%. Calc for $C_{17}H_{20}N_2O$: \dot{C} , 76.08; H, 7.51; \dot{N} , 10.44%.) IR v_{max}^{MB} 3365, 1695 cm⁻¹. IR v_{max}^{CHC1} 3400, 1714 cm⁻¹. ¹H-NMR $(CDCl_3) \delta 2.17 (dd, J = 16, 6 Hz, H-16), 2.72 (1H, dd, J = 16, 6 Hz, H-16)$ Hz, H-16), 3.88 (1H, dd, J = 6, 6 Hz, H-2), 6.54-6.87 (2H, m, aromatic H), 7.06 (1H, dd, J = 7.5, 7.5 Hz, aromatic H), 7.62 (1H, d, J = 7.5 Hz, aromatic H). MS $m/z 268 [M]^+, 226, 130, 96$ (base peak).

Cyclization of 37. To a stirred mixture of 37 (124 mg, 0.434 mmol) and anhyd. K_2CO_3 (511 mg) in CH₂Cl₂ (20 ml), MeSO₂Cl (0.1 ml, 0.8 mmol) was added at room temp. After stirring for 19 hr, the mixture was treated as above. To a soln of the dried residue (161 mg) in dry THF (25 ml), (Me₃Si)₂NK²¹ (312 mg, 1.57 mmol) was added at -70° with stirring under N₂. After 20 min, the cooling bath was removed and then the mixture was stirred at room temp for 40 min. After work-up as above, the crude product was separated by preparative TLC (hexane-EtOAc, 3:2) to afford 16 mg (14%) of 38 and 28 mg (24%) of 39 both as colourless prisms.

Acetylation of 38. A soln of 38 (32 mg, 0.12 mmol) and Ac₂O (0.7 ml) in pyridine (1 ml) was stirred at room temp for 3 hr. The mixture was neutralized with sat NaHCO₃ aq and extracted with CH₂Cl₂. The extract was washed with H₂O and dried over Na₂SO₄. Removal of the solvent gave an oil which was purified by preparative TLC (hexane-EtOAc, 1: 1) to afford 27 mg (73%) of 40. IR $v_{max}^{CHCl_3}$ 1720, 1650 cm⁻¹. ¹H-NMR (CDCl₃) δ 2.30 (s, Me), 2.36 (dd, J = 15, 9 Hz, H-16), 2.83 (dd, J = 15, 6 Hz, H-16), 4.54 (1H, dd, J = 9, 6 Hz, H-2), 6.97-7.42 (3H, m, aromatic H), 7.94 (1H, br d, J = 7.5 Hz, H-12). MS *m/z* 310 [M]⁺, 268, 144, 130, 96 (base peak).

Acetylation of 39. A soln of 39 (17 mg, 0.63 mmol) and Ac₂O (0.5 ml) in pyridine (0.8 ml) was stirred at room temp for 3 hr. After the usual work-up, the residue was purified by preparative TLC (hexane-EtOAc, 1:1) to give 17 mg (87%) of 41. IR v_{max}^{CHC1} 1716, 1650 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.30(s, Me), 2.92 (1H, dd, J = 17, 7.5 Hz, H-16), 4.18-5.03 (1H, br s, H-2), 6.87 7.36 (2H, m, aromatic H), 7.80 (1H, d, J = 7.5 Hz, aromatic H), 7.50-8.37 (1H, br s, aromatic H). MS m/z 310 [M]⁺, 309, 291, 268, 267, 130, 96 (base peak).

Reduction of 40 with NaBH₄. To a soln of 40 (32 mg, 0.103 mmol) in MeOH (5 ml), NaBH₄ (14 mg, 0.37 mmol) was added at room temp with stirring. After 10 min, the mixture was diluted with H₂O, extracted with CH₂Cl₂ and the extract dried over Na₂SO₄. Evaporation of the solvent followed by preparative TLC (PhH-EtOAc, 1:1) gave 26 mg(81%) of 42 as colourless oil. IR v_{max}^{CHC1} 3270, 1652 cm⁻¹. ¹H-NMR (CDCl₃) δ 2.32(s, Me), 2.78 (1H, br s, H-21), 3.96 (1H, br s, H-17), 4.35 (1H, br d, J = 11, 6Hz, H-2), 6.90-7.33 (3H, m, aromatic H), 8.12 (1H, br d, J = 7.5 Hz, H-12). MS m/z 312 [M]⁺, 311, 269, 241, 144, 130, 96 (base peak).

Formation of 43. A mixture of 42 (25 mg, 0.081 mmol) and 1,1'-thiocarbonyldiimidazole (43 mg, 0.24 mmol) in 1,2-dichloroethane (3 ml) was gently refluxed for 3 hr and evaporated in vacuo. The residue was purified by preparative TLC (PhH-EtOAc, 2:3) to afford 43 (21 mg, 62%). IR v_{max}^{BP}

1664 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.20 (3H, s, Me), 3.87– 4.55(1H, br s, H-2), 5.79–5.96(1H, br s, H-17), 6.90–7.38(3H, m, aromatic H), 7.06 (1H, br s) and 7.84(1H, d, J = 1 Hz) (H-4 and H-5 of imidazole), 8.53 (1H, br s, H-2 of imidazole).

Formation of 44. To a refluxed soln of Bu_3SnH (71 mg, 0.24 mmol) in dry xylene (3 ml) was added dropwise a soln of 43 (20 mg, 0.047 mmol) in dry xylene (3 ml) under N₂. After 2 hr, the solvent was removed under reduced pressure. The residue was purified by preparative TLC (CH₂Cl₂-MeOH, 19:1) to afford 13 mg (93%) of 44. Picrate : m.p. 206-209°. Identification with the authentic sample was confirmed by comparison of IR and NMR spectra and also by admixture of picrate.

Reduction of 41 with NaBH₄. A mixture of 41 (17 mg, 0.055 mmol) and NaBH₄ (10 mg, 0.26 mmol) in MeOH (3 ml) was stirred at room temp for 10 min. After the usual work-up, 12 mg (70%) of 45 was obtained by preparative TLC (PhH-EtOAc, 1:1). IR v_{max}^{CHC1} , 3400, 1646 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 1.51 (br s, OH), 2.28 (s, Me), 3.65–4.82 (1H, br s, H-2), 6.75–7.30 (2H, m, aromatic H), 7.78 (1H, br d, J = 7.5 Hz, H-9 or H-12), 7.83–8.29 (1H, br s, H-9 or H-12). MS m/z 312 [M]⁺, 311, 293, 268, 225, 144, 130, 96 (base peak).

Formation of 46. A mixture of 45 (25 mg, 0.081 mmol) and 1,1'-thiocarbonyldiimidazole (47 mg, 0.26 mmol) in dichloroethane (3 ml) was gently refluxed for 20 hr under N₂. Removal of the solvent and preparative TLC(CHCl₃) afforded 22 mg (65%) of 46. IR $v_{max}^{CHCl_3}$: 1650 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.30 (3H, s, Me), 3.90–4.87 (1H, br s, H-2), 5.19–5.55 (1H, br s, H-17), 6.94 (1H, br s) and 7.44 (1H, br s) (H-4 and H-5 of imidazole), 7.84 (d, J = 7.5 Hz, aromatic H), 8.14 (1H, s, H-2 of imidazole).

Formation of 47. A soln of 46 (20 mg, 0.047 mmol) in dry xylene (3 ml) was added dropwise to a refluxed soln of Bu₃SnH (73 mg, 0.25 mmol) in dry xylene (3 ml). After 2 hr, the solvent was removed in vacuo. The residue was purified twice by preparative TLC (PhH-EtOAc, 1:1 and then $CH_2Cl_2-MeOH, 99:1$) to give 10 mg (71%) of 47. IR $\nu_{max}^{CHCl_3}$ 1645 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.26 (s, Me), 3.73–4.31 (1H, br s, H-2), 6.77–7.31 (2H, m, aromatic H), 7.77 (1H, br d, J = 7 Hz, H-9 or H-12), 7.90–8.30 (1H, br s, H-9 or H-12). MS m/z 296 [M]⁺, 295, 281, 268, 253, 223, 144, 130, 96 (base peak).

Dehydration of 45. A methanesulphonate was formed from 45 (16 mg, 0.051 mmol), Et₃N (0.3 ml) and MeSO₂Cl (19 mg, 0.17 mmol) in CH₂Cl₂(4 ml) as above. A soln of the sulphonate and DBU (1.5 ml) in benzene (1.5 ml) was gently refluxed for 12 hr with stirring under N₂. The cooled mixture was diluted with CH₂Cl₂ and washed successively with sat CuSO₄ aq, sat NaHCO₃ aq and H₂O. The CH₂Cl₂ layer was dried over Na₂SO₄ and evaporated to leave the residue, which was purified by preparative TLC (hexane-EtOAc, 1:1) to give 7 mg(46%) of 51. IR $v_{max}^{CHCI_3}$ 1642 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.32 (3H,s, Me), 4.40–4.84 (1H, br s, H-2), 5.57 (2H, br s, olefinic protons), 6.79–7.28 (2H, m, aromatic H), 7.70 (1H, br d, J = 7 Hz, H-9 or H-12), 7.87–8.27 (1H, br s, H-9 or H-12). MS m/z 294 [M]⁺ (base peak), 293, 279, 266, 251, 223, 180, 167.

Hydrogenation of 51. A soln of 51 (6 mg) in MeOH (20 ml) was hydrogenated over PtO_2 (9 mg) for 11 hr at atmospheric pressure. After removal of the catalyst, the filtrate was evaporated to dryness. The residue was purified by preparative TLC (CH₂Cl₂-MeOH, 94:6) to afford 44 (6 mg, 99%).

Hydrogenation of **29**. A mixture of **29** (887 mg, 2.27 mmol) and PtO₂ (70 mg) in 1,2-dimethoxyethane (30 ml) was hydrogenated for 3 hr at atmospheric pressure. Filtration of the catalyst followed by concentration gave a crystalline solid, which was recrystallized from CH₂Cl₂ to afford 625 mg (70%) of **52** as colourless prisms, m.p. 205.5–207°. (Found: C, 69.79; H, 6.08; N, 7.10%. Calc for C₂₃H₂₄N₂O₄: C, 70.39; H, 6.16; N, 7.14%.) IR v^{KBr}_{max}: 3410, 1700, 1662 cm⁻¹. ¹H-NMR (DMSOd₆) δ 2.02(s, Me), 2.72(1H, br d, J = 15 Hz, H-6), 3.65–4.04(2H, m, H-5 and H-6), 4.75 (1H, br d, J = 3 Hz, OH), 5.18 (2H, br s, OCH₂Ph), 6.35 (1H, br d, J = 6 Hz, H-2). MS m/z 392 [M]⁺, 374, 301, 257 (base peak), 215, 187, 91.

Compound 53a. Colourless oil, purified by preparative TLC (hexane-EtOAc, 1:1). IR v_{max}^{KBr} 3375, 1690 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.10(3H, br t, J = 7 Hz, Me), 1.22(3H, br d, J = 6 Hz,

Me), 2.10 (br s, COMe), 2.71 (1H, dd, J = 13.5, 7.5 Hz, H-6), 4.57-5.00 (1H, m, O-C<u>H</u>Me-O), 6.29-6.65 (1H, m, H-2), 7.37 (s, Ph), 7.73 (1H, br d, J = 7.5 Hz, aromatic H). MS *m*/z 464 [M]⁺ (base peak), 419, 418, 391, 375, 347, 329, 257, 239, 91.

Compound **53b**. Colourless oil, purified by preparative TLC (hexane-EtOAc, 1:1). IR $v_{\text{MB}}^{\text{KB}}$: 3395, 1720–1680 cm⁻¹. ¹H-NMR (CDCl₃) δ 2.12 (3H, s, Me), 2.21–2.70 (2H, m, H-4), 2.88 (1H, br d, J = 15 Hz, H-6), 3.24–3.73 (1H, m, H-3), 4.80 (1H, d, J = 12 Hz), 5.16 (1H, d, J = 12 Hz, OCH₂Ph), 6.51 (d, J = 5 Hz), 6.65 (d, J = 6 Hz, H-2 of rotamers). MS m/z 496 [M]⁺, 361, 105, 91 (base peak).

Compound 53c. Colourless oil, purified by preparative TLC (hexane-EtOAc, 1:1). IR v_{max}^{fina} : 3340, 1715–1670 cm⁻¹. ¹H-NMR (CDCl₃, 70°) δ 2.10(s, Me), 2.68 (1H, d, J = 15 Hz, H-6), 3.35 (s, OMe), 5.19 (1H, d, J = 13.5 Hz), 5.34 (1H, d, J = 13.5 Hz, OCH₂Ph), 6.50 (1H, br d, J = 4.5 Hz, H-2), 7.37 (br s, Ph). MS *m/z* 480 [M]⁺, 405, 404, 391, 375, 345, 269, 241, 239, 91 (base peak).

Compound **54a**. Colourless oil, purified by preparative TLC (CH₂Cl₂-MeOH, 9:1). IR ν_{max}^{KBr} 3380, 1702 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.20(3H, t, J = 7 Hz, Me), 1.33(3H, d, J = 6 Hz, Me), 1.70 (3H, s, COMe), 3.88–4.27 (1H, m, H-5), 4.49 (1H, d, J = 4 Hz, H-2), 4.77 (1H, q, J = 6 Hz, O—C<u>H</u>Me—O), 6.93 (1H, br s, H-2 of indole). MS m/z 330 [M]⁺, 301, 285, 257, 242, 241, 187 (base peak), 185.

Compound 54b. Colourless needles, purified by preparative TLC (PhH-EtOAc, 1:1), followed by recrystallization from Et₂O-CH₂Cl₂, m.p. 168-169°. (Found: C, 72.52; H, 5.96; N, 7.73%. Calc for C₂₂H₂₂N₂O₃: C, 72.91; H, 6.12; N, 7.73%.) IR v_{max}^{KBr} : 3430, 1715, 1703 (sh) cm⁻¹. ¹H-NMR (CDCl₃-D₂O) δ 1.82 (3H, s, Me), 2.13 (1H, ddd, J = 14, 9, 4.5 Hz, H-4), 2.47 (1H, ddd, J = 14, 4.5, 4.5 Hz, H-4), 2.90 (1H, dd, J = 13.5, 7.5 Hz, H-6), 3.35 (1H, dd, J = 13.5, 4.5 Hz, H-6), 3.54 (1H, ddd, J = 4.5, 4.5, 4.4 Hz, H-3), 4.68 (1H, d, J = 4 Hz, H-2), 5.43 (1H, ddd, J = 9, 7.5, 4.5, 4.5 Hz, H-5). MS *m/z* 362 [M]⁺, 240, 222, 197, 187, 105 (base peak).

Compound 54c. Colourless oil, purified by preparative TLC (CH₂Cl₂-MeOH, 9:1). IR γ_{max}^{flim} 3340, 1705 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.72 (3H, s, Me), 1.89 (1H, ddd, J = 13.5, 8, 5 Hz, H-4), 2.25 (1H, ddd, J = 13.5, 4.5, 4.5 Hz, H-4), 2.67 (1H, dd, J = 13.5, 8 Hz, H-6), 3.22 (1H, dd, J = 13.5, 4 Hz, H-6), 3.39 (s, OMe), 4.04 (1H, dddd, J = 8, 8, 4.5, 4 Hz, H-5), 4.54 (1H, d, J = 4.5 Hz, H-2), 4.80 (2H, s, OCH₂O), 6.88 (1H, d, J = 2Hz, H-2) of indole), 6.97-7.45 (3H, m, aromatic H), 7.45-7.80 (1H, m, aromatic H). MS *m*/2 346 [M]⁺, 328, 294, 271, 269, 241, 223, 187 (base peak), 185.

Compound **55a**. Colourless oil, purified by preparative TLC (CH₂Cl₂-MeOH, 19:1). IR v_{max}^{KBr} : 3380, 1707 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.22 (3H, t, J = 7 Hz, Me), 1.34 (d, J = 5 Hz), 1.35 (d, J = 5 Hz, 3H, O—CH<u>Me</u>—O), 1.91 (s, COMe), 3.02 (1H, br s, OH), 4.07 (1H, br s, H-5), 6.90–7.50 (4H, m, aromatic H), 7.50–7.77 (1H, m, aromatic H). MS m/z 374 [M]⁺, 343 (base peak), 341, 329, 301, 286, 285, 271, 253.

Compound **55b**. Colourless needles, recrystallized from MeOH, m.p. 189–190°. (Found : C, 70.58; H, 6.35; N, 6.84%. Calc for $C_{24}H_{26}N_2O_4$: C, 70.91; H, 6.45; N, 6.89%.) IR v_{max}^{Max} 3405, 1712 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.80(3H, s, Me), 2.63 (br s, OH), 5.05 (1H, d, J = 6 Hz, H-2), 5.37–5.57 (1H, m, H-5). MS *m/z* 375, 268, 253, 248, 222, 105 (base peak).

Compound 55c. Colourless oil, purified by preparative TLC (CH₂Cl₂-MeOH, 19:1). IR v_{mex}^{film} 3350, 1710 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.90 (3H, s, Me), 3.39 (s, OMe), 4.06 (1H, br s, H-5), 4.80 (2H, s, OCH₂O), 4.89 (1H, d, J = 5 Hz, H-2), 6.87-7.41 (4H, m, aromatic H), 7.49-7.76 (1H, m, aromatic H). MS *m/z* 390 [M]⁺, 374, 359, 315, 285, 231, 188, 169, 130 (base peak).

Compound **56a**. Colourless oil, purified by preparative TLC (hexane-EtOAc, 1:1). IR v_{max}^{OHC1} 1712 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.20 (t, J = 7 Hz, Me), 1.27 (d, J = 5 Hz, Me), 3.93 (1H, dd, J = 3, 3 Hz, H-2), 4.71 (1H, q, J = 5 Hz, O-C<u>H</u>Me-O), 6.61 (1H, d, J = 7 Hz, aromatic H), 6.80 (1H, dd, J = 7.5, 7 Hz, aromatic H), 6.96-7.22 (2H, m, aromatic H). MS m/z 356 [M]⁺, 283, 267, 242, 184 (base peak), 130, 112.

m/z 356 [M]⁺, 283, 267, 242, 184 (base peak), 130, 112. *Compound* 57a. Colourless oil, separated from 56a by preparative TLC as above. IR v^{KBr}_{Max} 3365, 1710 cm⁻¹ ¹H-NMR $(CDCl_3) \delta 1.20 (t, J = 6 Hz), 1.23 (t, J = 6 Hz, OCH_2Me), 1.32 (d, J = 5 Hz, O-CHMe-O), 2.74 (1H, dd, J = 16, 7 Hz, H-16), 4.80 (1H, q, J = 5 Hz, O-CHMe-O), 6.47-6.91 (2H, m, aromatic H), 7.07 (1H, ddd, J = 7.5, 7.5, 1 Hz, aromatic H), 7.57 (1H, brd, J = 7.5 Hz, aromatic H). MS m/z 356 [M]⁺, 327, 311, 283, 267, 242, 241, 184 (base peak), 130.$

Compound 56 (R¹ = H). Colourless oil, purified by preparative TLC (PhH-EtOAc, 1:1). IR v_{max}^{KBt} 3375, 1703 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.07 (1H, ddd, J = 13.5, 11.5, 6 Hz, H-15), 2.98 (dd, J = 13.5, 3 Hz, H-16), 3.67 (1H, dddd, J = 11.5, 10.5, 5.5, 5.5 Hz, H-14), 3.93 (1H, dd, J = 3, 3 Hz, H-2), 6.61 (1H, d, J = 7.5 Hz, H-9 or H-12), 6.81 (1H, dd, J = 7.5, 7.5 Hz, H-10 or H-11), 7.09 (1H, dd, J = 7.5, 7.5 Hz, H-10 or H-11), 7.13 (1H, d, J = 7.5 Hz, H-9 or H-12). MS m/z 284 [M]⁺, 242, 154, 144, 130, 112 (base peak).

Compound **56**c. Colourless oil, purified by preparative TLC (hexane-EtOAc, 1:1). IR $v_{max}^{CMCJ_3}$ 1715 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.11 (1H, ddd, J = 11, 11, 4 Hz, H-15), 2.52 (dd, J = 15, 3 Hz, H-16), 3.02 (1H, dd, J = 15, 3 Hz, H-16), 3.37 (s, OMe), 3.74 (1H, dd, J = 3, 3 Hz, H-2), 4.73 (2H, s, OCH₂O), 6.62 (1H, d, J = 7.5 Hz, H-9 or H-12), 6.80 (1H, br dd, J = 7.5, 7.5 Hz, H-10 or H-11), 7.10 (1H, br dd, J = 7.5, 7.5 Hz, H-10 or H-11), 7.14 (1H, br d, J = 7.5 Hz, H-9 or H-12). MS m/z 372 [M]⁺, 341, 330, 313, 283, 268, 267, 242, 200 (base peak).

 \bar{H} ydrolysis of 57a. A soln of 57a (62 mg, 1.74 mmol) in MeOH (4.5 ml) containing 10% HCl (0.5 ml) was stirred at room temp for 5 min. The mixture was neutralized with sat NaHCO₃ aq and extracted with CH₂Cl₂. The extract was washed with H₂O, dried over Na₂SO₄ and evaporated to give the residue. Preparative TLC (hexane–EtOAc, 1 : 1) afforded 36 mg (73%) of 57 (R¹ = H). IR $\nu_{max}^{CHCl_3}$ 3400, 1712 cm⁻¹. ¹H-NMR (CDCl₃) δ 1.17(1H, ddd, J = 12, 12, 12 Hz, H-15), 3.67–4.08 (2H, m, H-2 and H-14), 6.48–6.87 (2H, m, aromatic H), 7.03 (1H, ddd, J = 7.5, 7.5, 1 Hz, H-10 or H-11), 7.52 (1H, br d, J = 7.5 Hz, H-9 or H-12). MS m/z 284 [M]⁺, 242, 223, 154, 144, 130, 112 (base peak).

Crystallographic data and structure analysis. Compound 39 was shown to have the following crystal data: triclinic, PI, a = 11.129(6), b = 14.624(6), c = 10.465(4)Å, $\alpha = 101.67(4), c = 10.465(4)$ Å $\beta = 84.67$ (4), $\gamma = 59.73$ (4)° and Z = 4. Lattice constants and intensity data were measured by using graphitemonochromated MoK_a radiation on a Rigaku AFC-5 diffractometer. A total of 4442 unique reflections with Fo > 4σ (Fo) were obtained using the $\omega - 2\theta$ scanning method at 2° /min in 2θ in the range up to $2\theta = 55^{\circ}$. The structure was solved by the structure determination package RASA-11 system of Rigaku Corp. based on the direct method.27 All the C, N, and O atoms were allocated and, at this stage of the work, the anisotropic temp factors were assumed for all the C, N, O atoms except C-8, C-13, C-8' and C-13'; for these four carbons the isotropic temp factors were assumed. The subsequent structural work²⁸ was transferred to a Hitac M280H system computer at the Computer Center of Tokyo University, which enables us to refine all the C, N, and O atoms anisotropically and to find all the hydrogen atoms by the difference Fourier synthesis. The resulting structure was finally refined by two cycles of full-matrix least-squares calculation to the R value of 0.058.† The ORTEP drawing²⁹ of the fully deduced structure is shown in Fig. 1.

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