A Stereocontrolled Entry to 3-Functionalized *cis*-3a-Methyloctahydroindoles: Building Blocks for Daphniphyllum Alkaloid Synthesis

Alejandro Cordero-Vargas, Xavier Urbaneja, Josep Bonjoch*

Laboratorio de Química Orgànica, Facultat de Farmàcia, Universitat de Barcelona, Av. Joan XXIII s/n, 08028 Barcelona, Spain Fax +34(93)4024539; E-mail: josep.bonjoch@ub.edu

Received 15 May 2007

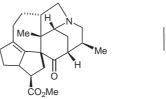
Abstract: A synthetic entry to cis-3a-methyl-3-methyleneoctahydroindol-5-ones employing ozonolysis, chemoselective methylenation, and double reductive amination of 2-(1-formylvinyl)-2methyl-1,4-cyclohexanedione monoethylene acetal is described. The same process using a 2-methoxycarbonyl derivative gave a trans-diastereoselectivity in the formation of the azabicyclic compound. Diastereoselective hydroboration of the exocyclic methylene of cis-octahydroindole derivative gives a valuable synthetic intermediate for Daphniphyllum alkaloid synthesis.

Key words: alkenation, aminations, fused-ring systems, ring closure, stereoselectivity

Functionalized cis-3a-methyloctahydroindoles are a key architectural feature in a wide array of Daphniphyllum alkaloids¹ as well as dendrobine-type alkaloids² (Figure 1). Our interest in *Daphniphyllum* alkaloids³ with the hexacyclic ring of daphniyunnine A,4 its N-oxide calyciphilline A⁵ and related congeners⁶ has directed our attention to the synthesis of 3-functionalized cis-3a-methvloctahydroindoles with a trans-relationship between the substituent at C-3 and the methyl group at C-3a. Our approach to the synthesis of these building blocks involves the preparation of a 3-methylene derivative, which could then give access to several functionalities (Scheme 1) to be used in synthetic approaches to the aforementioned natural products embodying this structural motif.

As envisioned in Scheme 1, cis-3-methylene-3a-alkyloctahydroindoles have been previously obtained by formation of the C-3-C-3a bond from 2-alkenyl N-propargylaminocyclohexanes through a cyclization catalyzed by transition metals $[Zr(II), {}^{7}Pt(II), {}^{8}or W(0)^{9}]$.¹⁰ In this letter, we report a different approach based on the interand intramolecular double reductive amination of α -(1formylvinyl)cyclohexanones. This procedure allows the incorporation of additional functionalization at the carbocyclic ring to gain access to cis-3-methylene-3a-methyloctahydroindol-5-one derivatives, which could be considered valuable intermediates in the Daphniphyllum alkaloid synthesis, particularly if the methylene group could be stereoselectively transformed into a functionalized appendage at C-3 with the same relative configuration as that of the natural products.

SYNLETT 2007, No. 15, pp 2379-2382 Advanced online publication: 28.08.2007 DOI: 10.1055/s-2007-986633; Art ID: D14807ST © Georg Thieme Verlag Stuttgart · New York





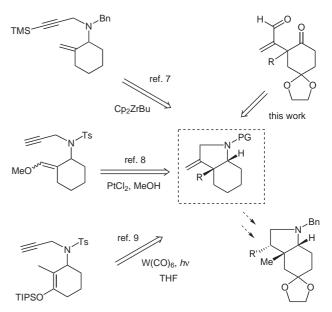


dendrobine

Figure 1

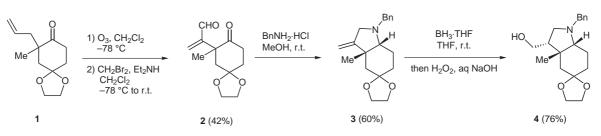






Scheme 1 Synthetic approaches to cis-3-methylene-3a-methyloctahydroindoles

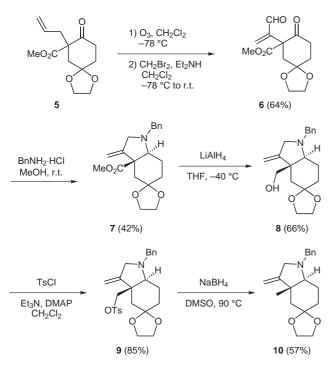
The starting material for the synthetic entry to our azabicyclic target was ketone 1 (Scheme 2), which was prepared from the monoethylene acetal of 1,4-cyclohexanedione by α -methylation (LiHMDS, MeI, THF, 70%)¹¹ followed by regioselective allylation under thermodynamic conditions using sodium hydride as the base (NaH, BrCH₂CH=CH₂, NaI, THF, r.t., 66%). This alternative to our previously reported procedure³ for the α, α -disubstituted cylohexanedione 1,¹² although giving a lower overall yield, requires only two steps instead of five. Ketone 1 was submitted to a one-pot procedure reductive ozonolysis-methylenation process, using the protocol developed by Hon¹³ (CH₂Br₂-Et₃N) in the second step, giving keto aldehyde 2 in 42% yield.^{14,15} The double reductive



Scheme 2 Stereoselective synthesis of 3-hydroxymethyl-3a-methyloctahydroindolone 4

amination of **2** using benzylamine hydrochloride and sodium cyanoborohydride diastereoselectively furnished *cis*-octahydroindole **3** in 60% overall yield.^{16,17} Diastereoselective hydroboration of **3** allowed alcohol **4** to be obtained with the desired relative configuration at the three stereogenic centers.¹⁸ The synthetic sequence herein described is the first for octahydroindole derivatives with this stereochemical arrangement and tetrasubstitution pattern of great interest for the development of a *Daphniphyllum* synthetic approach.

This promising result prompted us to extend the methodology to the synthesis of compounds embodying a different substituent at C-3a. Under the same reaction conditions used for the aminocyclization of **2**, keto aldehyde **6** did not give a *cis*-octahydroindole, the *trans*-octahydroindole **7** being formed instead (Scheme 3). For conclusive evidence of the structural assignment, ester **7** was reduced to alcohol **8**,¹⁹ which in turn was reduced again to give the 3a-methyl derivative **10**,²⁰ epimer of **3**. This result shows that the methyl- and methoxycarbonylsubstituted derivatives behave differently, probably due to a different preferred conformation of the bicyclic iminium salt intermediates (see below).



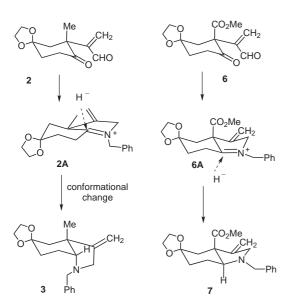
Scheme 3 Synthesis of *trans*-3-methylene-3a-methyloctahydroindoles

Synlett 2007, No. 15, 2379-2382 © Thieme Stuttgart · New York

The relative stereochemistry of **3** and **10** was elucidated by 2D NMR spectra (COSY and HSQC). The key evidence for the stereochemical and, in the *cis*-series, conformational elucidation of these octahydroindoles was found in the ¹H NMR coupling pattern for the methine proton at C-7a, which appears as a triplet (J = 2.7 Hz) for the *cis*isomer **3** and a doublet of doublets (J = 11.9 and 3.4 Hz) for the *trans*-isomer. The chemical shifts of C-6 and C-7a are of diagnostic value for the assignment of *cis/trans* configuration in this series of compounds, their signals appearing deshielded ($\delta = 34.6$ and 71.8 ppm for **10**) in the *trans*-series compared with the *cis*-derivatives ($\delta = 28.6$ and 67.8 ppm for **3**).

The results reported here constitute an example of the influence of the substituent at C-3a on the iminium salt reduction process in 3a-substituted hexahydro-2Hindoles.²¹ The diastereoselectivity observed could be explained by the different conformational behavior of bicyclic iminium intermediates 2A and 6A (Scheme 4), formed when keto aldehydes 2 and 6 were submitted to the double reductive amination.²² Thus, in order to avoid the steric crowding between the C-3a-Me and C-5-O bond of the acetal group, as well as to diminish the A^{1,3} strain between the exocyclic methylene at C-3 and the methylene at C-4, compound 2A could adopt a conformation in which the B ring is located below the quasiplanar ring A, implying that the acetal group blocks the bottom face of 2a. Consequently, the hydride approach takes place from the top face, giving the reduced compound 3 with a *cis*-diastereoselectivity (Scheme 4). In contrast, the methoxycarbonyl group at C-3a in 6A is less sterically demanding, allowing a conformation with the acetal group located above. The hydride attack thus occurs from the more accessible bottom face and the resulting reduced octahydroindole 7 has a trans-relationship in the azabicyclic ring. Although electronic effects, as observed by Woerpel in oxacarbenium ions,²³ cannot be discarded when trying to understand the stereoselectivity in the reduction involving a cyclic iminium ion bearing an electron-withdrawing substituent such as the methoxycarbonyl group, it should be mentioned that in our previous study³ a *cis*-diastereoselectivity was found in the octahydroindole ring formation from 5, which lacks the methylene group.

In summary, the synthesis of *cis*-3a-methyl-3-methyleneoctahydroindol-5-one derivative **3** and alcohol **4**, bearing the same relative configuration in its three stereogenic centers as daphniyunnine A, has been reported. The



Scheme 4 Reaction pathways for the reduction of *N*-benzylhexahydroindole iminium salts

usefulness of these new building blocks in the total synthesis of *Daphniphyllum* alkaloids is currently being studied.

Acknowledgment

This research was supported by the MEC (Spain)-FEDER through project CTQ2004-04701/BQU. Thanks are also due to the DURSI (Catalonia) for Grant 2005SGR-00442 and the Ministry of Education and Science (Spain) for a fellowship to A.C.

References and Notes

- (1) Kobayashi, J.; Morita, H. The Alkaloids 2003, 60, 165.
- (2) Morita, H.; Fujiwara, M.; Yoshida, N.; Kobayashi, J. *Tetrahedron* **2000**, *56*, 5801.
- (3) Solé, D.; Urbaneja, X.; Bonjoch, J. Org. Lett. 2005, 7, 5461.
- (4) Zhang, H.; Yang, S.-P.; Fan, C.-Q.; Ding, J.; Yue, J.-M. J. Nat. Prod. 2006, 69, 533.
- (5) Morita, H.; Kobayashi, J. Org. Lett. 2003, 5, 2895.
- (6) (a) Saito, S.; Kubota, T.; Fukushi, E.; Kawabata, J.; Zhang, H.; Kobayashi, J. *Tetrahedron Lett.* 2007, *48*, 1587. (b) Li, C.; He, H.; Wang, Y.; Mu, S.; Li, S.; Gao, Z.; Gao, Z.; Hao, X. *Tetrahedron Lett.* 2007, *48*, 2737; and references therein.
- (7) By Zr-promoted reductive cyclization: Mori, M.; Uesaka, N.; Saitoh, F.; Shibasaki, M. J. Org. Chem. 1994, 59, 5643.
- (8) By intramolecular reaction of enol ethers with alkynes catalyzed by Pt(II): Nevado, C.; Cárdenas, D. J.; Echavarren, A. M. *Chem. Eur. J.* 2003, *9*, 2627.
- (9) By W(CO)₅(L)-catalyzed cyclization of ω-acetylenic silyl enol ethers: Grandmarre, A.; Kusama, H.; Iwasawa, N. *Chem. Lett.* 2007, *36*, 66.
- (10) For the synthesis of 3a-unsubstituted *cis*-3-methylene-hydoindoles, see: (a) Solé, D.; Cancho, Y.; Llebaría, A.; Moretó, J. M.; Delgado, A. *J. Org. Chem.* **1996**, *61*, 5895. (b) Lemaire-Audoire, S.; Savignac, M.; Dupuis, C.; Genêt, J.-P. *Tetrahedron Lett.* **1996**, *37*, 2003. (c) Berteina, S.; De Mesmaeker, A.; Wendeborn, S. *Synlett* **1999**, 1121. (d) Gordon, G. J.; Luker, T.; Tuckett, M. W.; Whitby, R. J. *Tetrahedron* **2000**, *56*, 2113.

- (11) Carcache, D. A.; Cho, Y. S.; Hua, Z.; Tian, Y.; Li, Y.-M.; Danishefsky, S. J. J. Am. Chem. Soc. 2006, 128, 1016.
- (12) For the enantioselective synthesis of 1, see: (a) Behenna, D. C.; Stoltz, B. M. *J. Am. Chem. Soc.* 2004, *126*, 15044.
 (b) Mohr, J. T.; Behenna, D. C.; Harned, A. M.; Stoltz, B. M. *Angew. Chem. Int. Ed.* 2005, *44*, 6924.
- (13) Hon, Y.-S.; Chang, F.-J.; Liu, L.; Lin, W.-C. *Tetrahedron* 1998, *54*, 5233.
- (14) This was higher than the 30% yield obtained when the intermediate aldehyde was submitted to methylenation with the Eschenmoser reagent.
- 2-[(1-Methylene)formylmethyl]-2-methylcyclohexane-(15)1,4-dione Monoethylene Acetal (2) A stirred solution of ketone 1 (1.1 g, 5.23 mmol) in CH₂Cl₂ (105 mL) at -78 °C was treated with a constant stream of ozone. When the solution turned a characteristic pale blue, it was purged with oxygen. To this solution (at -78 °C) was added a mixture of CH₂Br₂ (1.84 mL, 4.54 g, 26.15 mmol) and Et₂NH (8.15 mL, 5.74 g, 78.47 mmol; preheated to 55 °C for 1.5 h and then cooled to r.t.) and the resulting vellow solution was stirred at r.t. for 2.5 h. The reaction mixture was concentrated, Et₂O was added (most of the ammonium salts precipitated), and the mixture was filtered and washed with Et₂O. The residue was purified by flash column chromatography (silica gel, hexane-EtOAc, 2:1 to 1:1) to give aldehyde 2 (42% yield) as a colorless oil: ¹H NMR (300 MHz, CDCl₃): δ = 1.44 (s, 3 H, CH₃), 1.62 (dd, J = 13.9, 2.9 Hz, 1 H, H-3), 2.00 (dm, J = 12.8 Hz, 1 H, H-5eq), 2.32 (td, 1 H, J = 12.8, 5.1 Hz, H-5ax), 2.42 (d, J = 13.9 Hz, 1 H, H-3), 2.59 (dt, J = 16.5, 4.8 Hz, 1 H, H-6eq), 2.75 (ddd, J = 16.6, 12.4, 5.7 Hz, 1 H, H-6ax), 4.05–3.91 (m, 4 H, OCH₂), 6.16 (s, 1 H, =CH), 6.35 (s, 1 H, =CH), 9.46 (s, 1 H, CHO). ¹³C NMR (75 MHz, CDCl₃): $\delta = 23.5$ (Me), 32.8 (C-5), 35.9 (C-6), 43.7 (C-3), 49.5 (C-2), 64.2 and 64.6 (OCH₂), 107.4 (C-4), 134.8 (=CH₂), 140.3 (=C), 193.3 (CHO), 211.1 (C-1). IR (NaCl, neat): 2969, 2888, 1711, 1688, 1282, 1114, 1072. 1036 cm⁻¹.
- (16) *cis*-1-Benzyl-3a-methyl-3-methyleneoctahydroindol-5one Ethylene Acetal (3)

To a solution of aldehyde 2 (0.4 g, 1.78 mmol) in MeOH (9 mL) were added first benzylamine hydrochloride (1.13 g, 7.85 mmol) and then NaBH₃CN (0.09 g, 1.43 mmol). After stirring for 30 min, an additional portion of NaBH₃CN (0.09 g, 1.43 mmol) was added and stirring was continued for 1 h. A third portion of NaBH₃CN (0.25 g, 3.92 mmol) was then added, and stirring was continued overnight. After removing the MeOH, CH₂Cl₂ was added and the resulting organic solution was washed with sat. aq NaHCO₃ solution, dried and concentrated. The resulting mixture was purified by column chromatography (silica gel, hexane-EtOAc, 4:1 to 2:1) to give amine **3** (60% yield) as a yellow oil: ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3, \text{gCOSY}): \delta = 1.21 \text{ (s, 3 H, H-9)}, 1.35 \text{ (dd,})$ *J* = 13.7, 2.5 Hz, 1 H, H-4), 1.46 (ddd, *J* = 12.5, 6.0, 2.9 Hz, 1 H, H-7), 1.77–2.10 (m, 4 H, H-4, H-6, and H-7), 2.23 (t, J = 2.7 Hz, 1 H, H-7a), 2.79 (dt, J = 13.7, 2.5 Hz, 1 H, H-2), $3.02 (d, J = 13.2 Hz, 1 H, CH_2Ph), 3.64 (dt, J = 14.5, 1.8 Hz)$ 1 H, H-2), 3.88–4.01 (m, 4 H, OCH₂), 4.10 (d, J = 13.2 Hz, 1 H, CH₂Ph), 4.66 (t, J = 2.1 Hz, 1 H, =CH), 4.69 (t, J = 2.6 Hz, 1 H, =CH), 7.21–7.35 (m, 5 H, ArH). ¹³C NMR (75 MHz, CDCl₃): δ = 21.4 (C-7), 21.8 (CH₃), 28.6 (C-6), 42.3 (C-4), 45.5 (C-3a), 57.5 (NCH₂Ar), 57.9 (C-2), 63.6 and 64.3 (OCH₂), 67.8 (C-7a), 101.5 (=CH₂), 102.1 (C-5), 123.7 (ArH), 128.3 (ArH), 128.4 (ArH), 139.7 (Ar), 157.4 (C-3).

(17) For the isolation of the 5-desoxo derivative, see: Mori, M.; Saitoh, F.; Uesaka, N.; Okamura, K.; Date, T. J. Org. Chem. 1994, 59, 4993.

Synlett 2007, No. 15, 2379-2382 © Thieme Stuttgart · New York

- (18) (3RS,3aRS,7aRS)-1-Benzyl-3-hydroxymethyl-3amethyloctahydroindol-5-one Ethylene Acetal (4) To a solution of 3 (0.9 g, 3.0 mmol) in anhyd THF (30 mL) at r.t. was added dropwise BH3. THF complex (15 mL, 15.0 mmol; 1 M solution in THF). When the starting material was completely consumed, the reaction was cooled to 0 °C and 3 N NaOH (6 mL) and H₂O₂ (3 mL of an 30% aq solution) were carefully added to the solution. After stirring for an hour, brine was added to the reaction and extracted with EtOAc. The organic layers were dried, filtered, and concentrated. The residue was purified by column chromatography (SiO₂, CH₂Cl₂-EtOAc, 98:2 to 95:5) to afford alcohol 4 (705 mg, 76%) as a yellow oil: ¹H NMR $(300 \text{ MHz}, \text{CDCl}_3, \text{gCOSY}): \delta = 1.21 (s, 3 \text{ H}, \text{CH}_3), 1.28 (dd, 1.28)$ *J* = 13.1, 2.2 Hz, 1 H, H-4), 1.45 (ddd, *J* = 12.4, 5.9, 3.2 Hz, 1 H, H-7), 1.87–1.76 (m, 2 H, H-4 and H-6), 1.91–2.11 (m, 2 H, H-6 and H-7), 2.38 (t, J = 2.6 Hz, 1 H, H-7a), 2.52 (t, *J* = 10.1 Hz, 1 H, H-2), 2.74 (dd, *J* = 10.1, 8.2 Hz, 1 H, H-2), 3.19 (d, J = 13.4 Hz, 1 H, CH₂Ph), 3.47 (dd, J = 10.4, 8.4 Hz, 1 H, CH₂OH), 3.69 (dd, *J* = 10.4, 5.9 Hz, 1 H, CH₂OH), 3.87-3.99 (m, 4 H, OCH₂), 4.02 (d, J = 13.5 Hz, 1 H, CH₂Ph), 7.20–7.32 (m, 5 H, ArH). ¹³C NMR (300 MHz, CDCl₃, gHSQC): δ = 21.5 (C-6), 21.9 (Me), 28.6 (C-7), 36.5 (C-4), 43.0 (C-3a), 51.0 (C-3), 55.1 (C-2), 57.9 (NCH₂Ar), 62.6 (CH₂OH), 63.5 and 64.4 (OCH₂), 68.5 (C-7a), 109.6 (C-5), 126.6 (ArH), 128.1 (ArH), 128.3 (ArH), 140.2 (Ar). IR (NaCl, neat): 3440, 2926, 2877, 2788, 1359, 1091 cm⁻¹.
- (19) *trans*-1-Benzyl-3a-hydroxymethyl-3-methyleneoctahydroindol-5-one Ethylene Acetal (8) ¹H NMR (300 MHz, CDCl₃, NOESY): $\delta = 1.57-1.60$ (m, 2 H, H-4 and H-7), 1.74–1.89 (m, 2 H, H-4 and H-6), 1.89– 2.01 (m, 2 H, H-6 and H-7), 2.40 (dd, J = 11.9, 3.4 Hz, 1 H, H-7a), 2.82 (td, J = 14.4, 2.4 Hz, 1 H, H-2), 3.12 (d, J = 12.9Hz, 1 H, NCH₂Ar), 3.44 (d, J = 11.0 Hz, 1 H, CH₂OH), 3.69 (d, J = 14.4 Hz, 1 H, H-2), 3.86–4.00 (m, 4 H, OCH₂), 4.04 (d, J = 13 Hz, 1 H, NCH₂Ar), 4.40 (d, J = 11.0 Hz, 1 H, CH₂OH), 4.80 (t, J = 2.5 Hz, 1 H, =CH₂), 4.88 (t, J = 2.0 Hz, 1 H, =CH₂), 7.26–7.30 (m, 5 H, ArH). ¹³C NMR (75 MHz, CDCl₃, gHSQC): $\delta = 21.1$ (C-7), 34.5 (C-6), 40.3 (C-4), 49.5 (C-3a), 58.0 (CH₂Ph), 59.9 (C-2), 63.9 and 64.7 (OCH₂), 66.3 (CH₂OH), 72.3 (C-7a), 103.7 (=CH₂), 109.0 (C-5),

127.1 (ArH), 128.3 (ArH), 128.5 (ArH), 138.0 (Ar), 152,1 (C-3). IR (NaCl, neat): 3410, 3300, 2947, 2881, 2804, 1452, 1359, 1252, 1206, 1142, 1110, 1059, 1011 cm⁻¹.

- (20) *trans*-1-Benzyl-3a-methyl-3-methyleneoctahydroindol-5-one Ethylene Acetal (10) ¹H NMR (300 MHz, CDCl₃, gCOSY): $\delta = 1.18$ (s, 3 H, CH₃), 1.56–1.62 (m, 3 H, H-4 and H-7), 1.72–1.75 (m, 1 H, H-7), 1.91–1.96 (m, 2 H, H-6), 1.97 (dd, J = 13.2, 2.4 Hz, 1 H, H-7), 2.04–2.12 (m, 1 H, H-7a), 2.73 (dt, J = 14.8, 2.1 Hz, 1 H, H-2), 3.18 (d, J = 13.5 Hz, 1 H, CH₂Ph), 3.70 (dt, J = 15.2, 1.9 Hz 1 H, H-2), 3.86–4.07 (m, 5 H, OCH₂ and CH₂Ph), 4.58 (t, J = 2.3 Hz 1 H, =CH₂), 7.22–7.37 (m, 5 H, ArH). ¹³C NMR (75 MHz, CDCl₃): $\delta = 20.3$ (CH₃), 21.5 (C-7), 34.6 (C-6), 42.6 (C-4), 45.4 (C-3a), 58.1 (NCH₂Ar), 58.6 (C-2), 63.6 and 63.6 (OCH₂), 71.8 (C-7a), 100.0 (=CH₂), 109.8 (C-5), 126.7 (ArH), 128.1 (ArH), 128.3 (ArH), 139.8 (Ar), 156.7 (C-3).
- (21) For the influence of the substituent at C-3a in the reduction of *N*-Boc-hexahydro-1*H*-indoles, see: Brodney, M. A.; Cole, M. L.; Freemont, J. A.; Kyi, S.; Junk, P. C.; Padwa, A.; Riches, A. G.; Ryan, J. H. *Tetrahedron Lett.* **2007**, *48*, 1939.
- (22) Both precursors **2** and **6** seem to have the same preferred conformation, according to their NMR data. For methyl derivative **2**, see ref. 15. For **6**: ¹H NMR (300 MHz, CDCl₃): $\delta = 2.00$ (ddd, J = 13.1, 6.5, 3.3 Hz, 1 H, H-5), 2.26 (dt, J = 13.5, 5.2 Hz, 1 H, H-5), 2.51 (d, J = 13.0 Hz, 1 H, H-3), 2.56 (d, J = 13.7 Hz, 1 H, H-3), 2.63 (ddd, J = 15.9, 5.2, 3.0 Hz, 1 H, H-6), 3.04 (ddd, J = 15.9, 13.7, 6.5 Hz, 1 H, H-6), 3.79 (s, 3 H, OMe), 3.91–4.04 (m, 4 H, OCH₂), 6.23 and 6.28 (2 s, 1 H each, =CH₂), 9.47 (s, 1 H, CHO). ¹³C NMR (75 MHz, CDCl₃): $\delta = 33.9$ (C-5), 37.6 (C-6), 39.7 (C-3), 53.0 (OMe), 61.4 (C-2), 64.3 and 64.8 (OCH₂), 106.4 (C-4), 135.9 (=CH₂), 147.5 (=C), 170.4 (CO₂Me), 192.0 (CHO), 203.1 (C-1). IR (NaCl, neat): 2956, 2894, 1740, 1710, 1695, 1434, 1279, 12361121, 1089, 1040 cm⁻¹.
- (23) (a) Ayala, L.; Lucero, C. G.; Romero, J. A. C.; Tabacco, S. A.; Woerpel, K. A. *J. Am. Chem. Soc.* 2003, *125*, 15521.
 (b) Smith, D. M.; Woerpel, K. A. *Org. Lett.* 2004, *6*, 2063.
 (c) Smith, D. M.; Woerpel, K. A. *Org. Biomol. Chem.* 2006, *4*, 1195.

Copyright of Synlett is the property of Georg Thieme Verlag Stuttgart and its content may not be copied or emailed to multiple sites or posted to a listserv without the copyright holder's express written permission. However, users may print, download, or email articles for individual use.