

Nickel-Catalyzed Remote C4–H Arylation of 8-Aminoquinolines

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Supporting Information



ABSTRACT: A useful and convenient method for C–H bond arylation of 8-aminoquinoline motifs on the remote C4 position was developed. This method shows good functional group tolerance toward various Grignard reagents and aminoquinoline via a nickel catalysis, giving the desired arylated products in good yields. The present method affords an efficient access to construct multisubstituted aminoquinolines.

iven the importance and versatility of quinoline J derivatives as bioactive natural products, medicines, functional materials (Figure 1), 1 and directing groups for C-



Figure 1. Representative biologically active compounds.

H functionalization in transition-metal-catalyzed reactions,² developing highly efficient and environmentally benign protocols for the synthesis of various functionalized quinolines is highly desired. In the past several decades, tremendous progress has been made for the construction of quinoline motifs via cyclization of anilines derivatives³ or cyclization of substituted alkynes⁴ intermolecularly or intramolecularly.

However, the above-mentioned methods suffer from drawbacks such as high prefunctionalized starting materials and low selectivity in some cases.^{4d} An alternative synthetic method for functionalized quinolines is the site-selective modification of readily accessible quinoline frameworks via C-H functionalization. This method provides a quick access to functionalized quinolines. However, the regioselectivity has to be controlled. Many reactions for the C-H bond functionalization of the quinoline core are mainly on the position of $C2^{5}_{1}$, $C3^{6}_{1}$, $C5^{7}_{1}$ and C8,⁸ which are easily accessible. However, as for the functionalization of the important C4 position,^{1d,e} which is relatively less electrophilic (more nucleophilic) among the 8aminoquinioline core, progress is very slow.9 In recent years, there are several reports concerning the functionalization on the C4 position of pyridine¹⁰ or quinoline motif,¹¹ which

provide some idea for the functionalization of 8-aminoquinoline. However, as for the remote C4-H functionalization of aminoquinolines, only three examples have been disclosed (Scheme 1).⁹ In 2014, Zeng et al. reported the first two substrates of Fe-catalyzed remote C4-H allylation of 8aminoquinolines (Scheme 1a).^{9a} Wu et al. developed phosphonation of 8-aminoquinoline at C4 position via the combination of photocatalysis and transition metal catalysis

Scheme 1. C4-H Functionalization of 8-Aminoquinolines



Received: July 11, 2019

(Scheme 1b).^{9b} Ranu et al. reported high efficient cobaltcatalyzed remote C4–H alkylation of 8-aminoquinoline with photocatalyst (Scheme 1c).^{9c} As is known, the C–H bond arylation attracts immense attention among the various kinds of C–H functionalizations, which can greatly improve the complexity of the obtained substrates due to the obtained valuable biaryl structural motifs. However, the research of a method to build up C4-arylated aminoquinolines has never been disclosed.

Our group is very interested in the functionalization of aminoquinoline core^{7d,12} and the cross-coupling reaction using Grignard reagents as important partner.¹³ Herein, we reported the first nickel-catalyzed C4–H arylation of 8-aminoquinolines with aryl Grignard reagents (Scheme 1d). This method affords a useful and convenient access to construct multisubstituted aminoquinolines.

The cross-coupling between N-(quinolin-8-yl)pivalamide (1a) and phenylmagnesium bromide (2a) was examined as a model reaction (see Supporting Information). After optimization of reaction conditions, the C4 arylated product N-(4-phenylquinolin-8-yl)pivalamide (3a) was generated in 78% yield when Ni(OTf)₂ was used as catalyst in the presence of 2 equiv of ^tBuOK at 60 °C (Table 1, entry 1). If no nickle

Table 1. Optimization of the Reaction Conditions^a

^t Bu N H 1a	+ Ph-MgBr 1.0 M in THF 2a, 4.0 equiv 60 °C /	I0 mol %) aquiv) mL) 4 h 3a
entry	variation from standard condit	tions yield of $3a^b$ (%)
1	none	80 (78)
2	no Ni(OTf) ₂	<1 ^c
3	Ni(COD) ₂ instead of Ni(OT	ſſ) ₂ 75
4	$NiCl_2$ instead of $Ni(OTf)_2$	66
5	no ^t BuOK	45
6	PhMgBr (3 equiv)	50
-	- />	/ ->

"Reaction conditions: N-(quinolin-8-yl)pivalamide (1a, 0.2 mmol), phenylmagnesium bromide (2a,1.0 M in THF, 4.0 equiv), solvent (1.0 mL), N₂ atmosphere. ^bGC Yield using *n*-dodecane as internal standard; the isolated yield was given in parentheses. ^cC2 arylated product (3a') was obtained in 10%. The crude mixture contains a small amount of 1,2-dihydroquinoline derivative. After quenched by NH₄Cl (aq), it was oxidized to quinoline by adding 1 equiv of DDQ and stirring for 10 min at ambient temperature (procedures are similar hereinafter).

catalyst was used, **3a** was formed in less than 1% yield with C2 arylated product (**3a**') generated in 10% yield (entry 2). When Ni(COD)₂ was used as catalyst under the standard condition instead of Ni(OTf)₂, the desired product **3a** was generated in 75% yield, indicating that the reactive species may be Ni(0). NiCl₂ can also catalyze this reaction, giving **3a** in 66% yield (entry 4). Without 'BuOK, the reaction can be proceeded but in lower yield (entry 5). Lower the amount of Grignard reagent used in the reaction decreased the yield of **3a** (entry 6). The structure of **3a** was confirmed by single crystal X-ray analysis (figure in Scheme 2).

After having the optimized reaction conditions, we investigated the substrate scopes of Grignard reagents (Scheme 2). For various kinds of aryl Grignard reagents, the reaction proceeded smoothly to be arylated on C4 position selectively, yielding the desired products 3a-i in good yields. Methyl

Scheme 2. Substrate Scope for Aryl Grignard Reagents^a



^{*a*}Reaction conditions: 1 (0.2 mmol), 2 (4.0 equiv), N₂, isolated yield. ^{*b*}Reaction conditions: 24 h.

groups attached to the phenyl Grignard reagents at different positions show no obvious steric hindrance, giving 3b-d in middle to good yields (61-70%). When electron-donating group 4-Me, 4-^tBu, or 4-OMe was attached to the para position, the corresponding 3d-f were yielded in 57%-73% yield. Aryl Grignard reagents possessing electron-withdrawing substituents (4-F, 4-Cl) gave 3g,h in modest yields. Biphenyl Grignard reagent gave 3i in a slightly lower yield. The substituents on the quinoline core motifs were also investigated. When an aminoquinoline having a methoxy group on C6 position was employed, 3j was generated in 79% yield. However, if a Br or Cl atom is attached to the C5 position, the desired products (31,m) were not yielded, but C5 arylated product 1f was obtained in 55%, 70%, respectively, indicating that the Kumada coupling is predominated. C5 arylated aminoquinoline can also tolerated to give 3n in 58% yield, which shows no obvious steric hindrance to the C4 functionalization. Alkyl Grignard reagent such as butylmagnesium bromide and heteroaromatic Grignard reagent such as thiophen-2-ylmagnesium bromide failed to give the corresponding products (see Supporting Information).

As shown in Scheme 3, the reactions of phenylmagnesium bromide (2a) with various kinds of aminoquinolines were investigated. When replacing 1a with 2,2-dimethyl-*N*-(quino-lin-8-yl)butanamide, the corresponding product 4a was yielded in 61% yield. For long-chain alkyl groups and cyclohexyl group, the arylation proceeded smoothly to give desired products 4b-d in 65%–68% yield. It is noted that benzoyl aminoquinolines can also be tolerated in this system without the generation of *ortho* C–H bond arylated byproducts directed by 8-amino-quinoline group (4e-j).

In order to demonstrate the synthetic utility, the model reaction was enlarged to a gram scale, and 1.3 g **3a** was generated in 61% yield (Scheme 4). The amide group could be hydrolyzed under basic conditions to give C4-arylated

Scheme 3. Substrate Scope for Aminoquinolines^a



^aReaction condition: 1 (0.2 mmol), 2a (4.0 equiv), N₂, isolated yield.



Scheme 4. Synthetic Applications

^{*a*}Reaction conditions: Ni(OTf)₂ (10 mol %), **2a** (4 equiv), ^{*t*}BuOK (2 equiv), THF, 60 °C/12 h. ^{*b*}Reaction conditions: NaOH (10 equiv), EtOH, 100 °C/24 h. ^{*c*}Reaction conditions: NCS (1.1 equiv), DMF (2 mL), 50 °C/2 h, N₂. ^{*d*}Reaction conditions: NBS (1.1 equiv), DMF (2 mL), 50 °C/2 h, N₂. ^{*c*}Reaction conditions: 3-thiopheneboronic acid (2 equiv), Na₂CO₃ (2 equiv), Pd(PPh₃)₄ (10 mol %), DMSO, N₂, 140 °C/10 h. ^{*f*}Reaction conditions: m-CPBA (2 equiv), CH₂Cl₂, 25 °C/12 h, air. ^{*g*}Reaction conditions: R-Li (5 equiv), THF (2 mL), 0 °C/12 h, N₂. See Supporting Information for details.

quinoline-8-amine **5** in 90% yield. Furthermore, C5-position could selectively brominated and chlorinated to give **31**,**m** in high yields according our previous report.^{12b} It should be noted that these bromo- and cholor- products **31**,**m** cannot be obtained by arylation of C5 bromo- or chloroquinolines (Scheme 2). With application of a Suzuki coupling reaction to this brominated compound,^{12b} the thienyl group can be introduced at the C5 position to give **6a** in 81% yield. The C2 position was arylated and alkylated to give **3k**, **7a**,**b** in good yields by the reaction with *m*-CPBA.

To shed some light on the possible mechanism of the C4 arylation on 8-aminoquinlines, we performed several preliminary experiments as depicted in Scheme 5. When *tert*-butyl quinolin-8-ylcarbamate (9a), quinolin-8-yl pivalate (9b), or free amine quinolin-8-amine (9c) instead of 1a was treated with phenylmagnesium bromide (2a) under the standard condition, no corresponding product was detected (Scheme 4a). The above-mentioned results indicate the important role

Scheme 5. Control Experiments



of the amide group at C8 position. *N*-(Naphthalen-1-yl)pivalamide (9d) was also not arylated. When unsubstituted quinoline (9e) was reacted with 2f, only C2-arylted product 10 was obtained in 65% yield, as previous reported by Da et al.¹⁴ In addition, by careful analysis of the atmosphere after reaction, a small amount of hydrogen was detected.

On the basis of the above results and literature,^{10a} we propose here a possible pathway for the C4 arylation of 8-aminoquinoline derivatives (Scheme 6). Initially, deprotona-

Scheme 6. Possible Mechanism



tion of amide 1 by ^tBuOK followed by isomerization forms the intermediate **B**. Coordination of the Ni(0) species which was reduced by Grignard reagent from Ni(II) with aminoquinoline forms intermediate **C**. Then, nucleophilic addition of Grignard reagent generates intermediate **D**. The intermediate **D** then undergoes elimination and protonation to deliver the desired product 3 and generate Ni(0) species for the next catalytic cycle. A nickel insertion mechanism may also be possible (see Supporting Information).^{10b,c,11b}

In summary, we demonstrated a useful and convenient nickel catalyzed protocol for C4-arylation of 8-aminoquinoline motifs. The reaction shows good tolerance toward various aryl Grignard reagents and aminoquinoline, giving the desired products in good yields. This method affords an efficient access to construct multisubstituted aminoquinolines.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b02403.

Detailed experimental procedures, characterization data, spectra copies of the ¹H, ¹³C NMR for obtained products, and X-ray data of **3a** (PDF)

Accession Codes

CCDC 1842336 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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Notes

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ACKNOWLEDGMENTS

The authors thank the Natural Science Foundation of China (Nos. 21676076, 21878071), the Hu-Xiang High Talent Project in Hunan Province (2018RS3042), and Recruitment Program for Foreign Experts of China (WQ20164300353) for financial support.

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