CHEMISTRY LETTERS, pp. 385-388, 1988.

© 1988 The Chemical Society of Japan

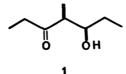
Chiral α -Methyl-homoallylic Alcohols from Yeast-Generated Precursors. Synthesis of $(4\underline{R},5\underline{S})$ Sitophilure

Giovanni FRONZA, Claudio FUGANTI, Hans-Erik HÖGBERG,¹⁾ Giuseppe PEDROCCHI-FANTONI, and Stefano SERVI

CNR Centro per lo Studio delle Sostanze Organiche Naturali. Dipartimento di Chimica Politecnico di Milano, Piazza L. da Vinci 32, 20133 Milano, Italy

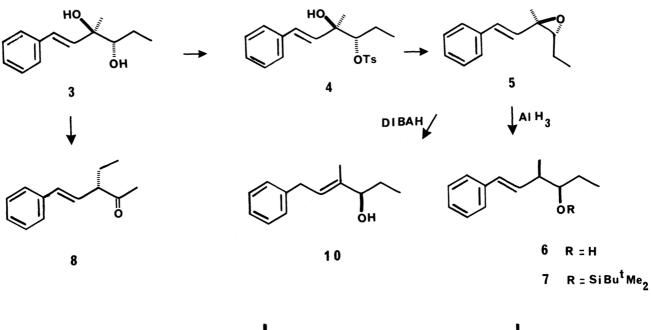
Optically active $(3\underline{S},4\underline{R})-3,4-dihydroxy-4-methyl-6-phenyl-hex-5$ ene and $(2\underline{S},3\underline{R}\underline{S})-2,3-dihydroxy-3-methyl-5-phenyl-pent-4-ene, previous$ ly obtained from fermenting baker's yeast reduction of the correspon $ding ketones, are transformed either into optically active <math>\alpha$ -ethyl and α -methyl ketones, or into α -methyl homoallylic alcohols from which enantiomerically pure $(4\underline{S},5\underline{R})$ sitophilure, $(2\underline{S},3\underline{R})-2$ -methyl-1,3butandiol and its $(2\underline{S},3\underline{S})$ -diastereoisomer have been prepared.

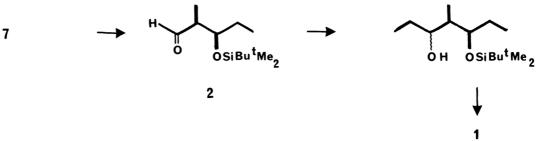
A recent report²⁾ on the synthesis of the $(4\underline{R},5\underline{S})$ form 1 of the pheromone sitophilure³⁾ and of its enantiomer from yeast generated methyl $(3\underline{R})$ 3-hydroxy pentanoate through the key intermediacy of $(2\underline{S},3\underline{R})$ -2-methyl-3-^tbutyl-dimethyl-silyloxy-pentanal 2 induces us to present a short preparation of 1 realized in a study designed to obtain synthetically useful chiral α -methyl homoallylic alcohols from $(3\underline{S},4\underline{R})$ -3,4-dihydroxy-4-methyl-6-phenyl-hex-5-ene 3 and $(2\underline{S},3\underline{RS})$ -2,3-dihydroxy-3methyl-5-phenyl-pent-4-ene 11, obtained from baker's yeast treatment of the corresponding ketones.⁴)



We aimed to convert 3 into 2 through the tosylate 4, the epoxide 5 and the product of 1,2-opening 6, according to earlier experience.⁵⁾ Ozonolysis of the $\underline{0}$ -protected product 7 would subsequently produce the required 2. Rather unexpectedly,

the diol 3 when treated with 1 mol equiv. of TsCl in pyridine at 0 °C, did not give the anticipated tosylate 4 but instead the ethyl ketone 8 (vide infra). However, this difficulty was circumvented by trapping the unstable tosylate as the required epo-xide 5 which was isolated in 70% yield when the diol 3 was reacted with excess solid TsCl in 1,2-dimethoxyethane in the presence of powdered KOH.⁶⁾ Treatment of 5 with DIBAH in THF at -50 °C afforded the product of 1,4-opening 10, in 90% yield, whereas with AlH₃ in Et₂O at -30 °C gave rise to the required carbinol 6, which was converted into the protected compound 7, $[\alpha]_D^{20}$ -14.6° (c 1, CHCl₃), in <u>ca.</u> 40% overall yield. Product 7 on ozonolysis in CH₂Cl₂ at -78 °C and Ph₃P treatment afforded benzaldehyde and compound 2. This mixture was immediately reacted with ethereal BrMgCH₂CH₃. The adduct was separated by chromatography from 1-phenylpropan-1-ol, and after Swern oxidation followed by deprotection, it afforded as reported, ²⁾ compound (1), $[\alpha]_D^{20} + 26.2^\circ$ (c 1, Et₂O) (96% by GLC) (lit.²⁾ +26.7°) in 45% from 7.

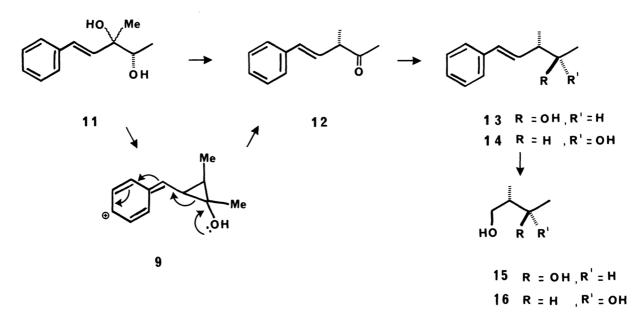




Scheme 1.

The $(2\underline{S}, 3\underline{RS})$ diol 11 behaved similarly to compound 3 when reacted with TsCl/pyridine, furnishing 3-methyl-5-phenylpent-4-en-2-one, 12 $[\alpha]_D^{20}$ +76.3° (c 1,

CHCl₃), containing an excess of the $(3\underline{S})$ enantiomer, and proceeding through 9 as a possible intermediate, in a not unprecedented fashion.⁷⁾ Indeed, product 12 on DIBAH reduction afforded $(2\underline{R},3\underline{S})$ 13, $\begin{bmatrix} \alpha \end{bmatrix}_{D}^{20}$ -29.3° (c 1, CHCl₃) and 14, $\begin{bmatrix} \alpha \end{bmatrix}_{D}^{20}$ -16.5° (c 1, CHCl₃), in 2:1 ratio, separated by fractional crystallization of the 3,5-dinitrobenzoates, in 70% overall yield.



Scheme 2.

These materials, in separate runs, were benzoylated and treated sequentially with O_3 in CH₂Cl₂/MeOH, NaBH₄ and aqueous NaOH to give $(3\underline{R}, 2\underline{S})$ 15, $[\alpha]_D^{20}$ -3.7° (c 3.5, EtOH)⁸) and $(2\underline{S}, 3\underline{S})$ 16, $[\alpha]_D^{20}$ +2.5° (c 3.5, EtOH)⁸) of <u>ca</u>. 0.6 ee, as confirmed by Mosher's analysis ⁹) of the intermediate 3-benzoate yielding 15. These results, seen together, indicate that the conversion of the tosylate in the α -methyl ketone 12 is accompanied by inversion of configuration at C-2, being the loss of optical purity possibly due to the poor configurational stability of the intermediate ketone 12 to the basic conditions. Subsequently, the diol 11 was converted ⁷¹ into a mixture of diastereoisomeric epoxides, yielding on hydride opening, as above, product 13, $[\alpha]_D^{20}$ -49° and the $(2\underline{R},3\underline{R})$ diastereoisomer, $[\alpha]_D^{20}$ +30.2° (c 1, CHCl₃). These materials were similarly converted into optically pure 15, $[\alpha]_D^{20}$ -6.4° (lit.⁸⁾ -6.3°) and into the $(2\underline{R},3\underline{R})$ diastereoisomer, $[\alpha]_D^{20}$ -4.2° (c 3.5, EtOH) (enantiomer of 16), respectively, in <u>ca</u>. 35% overall yield from 11. Recently, ¹⁰ a ten steps preparation of $(2\underline{R},3\underline{R})$ 2-methyl-3-benzyloxy butanal, conceivably accessible from 13 <u>via</u> benzylation and ozonolysis, from $(2\underline{S})$ -3-hydroxy-2-methyl propionic acid has been reported.

The preparation from the diols 3 and 11 of the chiral α -methyl homoallylic alcohols 6, 13 and the enantiomer of 14 in optically pure form, from which synthetically

useful α -methyl- β -alkoxy C₅ and C₄ aldehydes can be obtained, is a further illustration of the significance to the synthesis of natural products of the microbial transformations of non conventional substrates leading to small highly functionalized educts.

References

- Professore a Contratto 1986/87. Politecnico di Milano. Permanent adress: University College of Sundsvall/Härnösand, Box 860, S-851 24 Sundsvall, Sweden.
- 2) K. Mori and T, Ebata, Tetrahedron, <u>42</u>, 4421(1986).
- J.K. Phillips, C.A. Walgenbach, J.A. Klein. W.E. Burkholder, N.R. Schmuff and H.M. Fales, J. Chem. Ecol., <u>11</u>, 1263(1985) and references therein.
- 4) C. Fuganti, P. Grasselli, S. Servi, F. Spreafico, C. Zirotti, and P. Casati, J. Org. Chem., <u>49</u>, 4087 (1984); G. Fronza, C. Fuganti, G. Pedrocchi-Fantoni, and S.Servi J. Org. Chem., <u>52</u>, 1141(1987).
- 5) C. Fuganti, P. Grasselli, F. Spreafico, and C. Zirotti, J. Org. Chem., <u>49</u>, 543 (1984).
- 6) S. Holand and R. Epsztein, Synthesis, 1977, 706.
- 7) S. Tsuchihashi, K. Kitajima, and S. Mitamura, Tetrahedron Lett., <u>22</u>, 4305(1981). and ref. therein
- 8) C. Najera, M. Yus, and D. Seebach, Helv. Chim. Acta, <u>67</u>, 289(1984); G. Fleet and T. Shing, Tetrahedron Lett., <u>24</u>, 3657(1983).
- 9) J.A. Dale, D.L. Dull, and H.S.Mosher, J. Org. Chem., <u>34</u>, 2543(1969).
- 10)R. Baker, M.J. O'Mahony, and C.J. Swain, Tetrahedron Lett., 27, 3059(1986).

(Received September 22, 1987)