

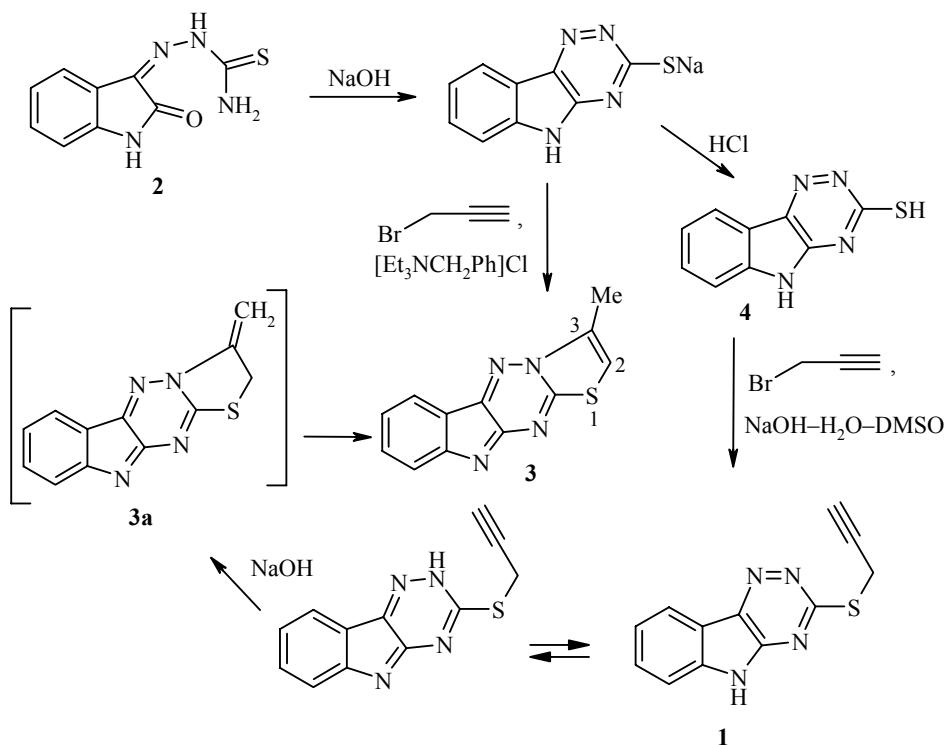
## NEW SYNTHESIS OF THE [1,3]THIAZOLO-[3',2':2,3][1,2,4]TRIAZINO[5,6-*b*]INDOLE SYSTEM

D. G. Kim<sup>1</sup>\* and A. V. Zhuravlyova<sup>1</sup>

**Keywords:** 3-propargylthio-5H-[1,2,4]triazino[5,6-*b*]indole, 3-methyl[1,3]thiazolo[3',2':2,3][1,2,4]triazino[5,6-*b*]indole, one-pot synthesis.

Tomchin [1] was the first to report the preparation of the new 2,3-diphenyl[1,3]thiazolo[3',2':2,3]-[1,2,4]triazino[5,6-*b*]indole system by the reaction of 3-benzoylphenylmethylthio[1,2,4]triazino[5,6-*b*]indole with PPA but no spectral evidence was given for the formation of this product.

In an attempt to obtain 3-propargylthio-5H-[1,2,4]triazino[5,6-*b*]indole (**1**) by a one-pot synthesis from isatin-β-thiosemicarbazone (**2**) by the action of aqueous alkali and propargyl bromide under phase-transfer



\* To whom correspondence should be addressed, e-mail: kim\_dg48@mail.ru, dgkim@susu.ac.ru.

<sup>1</sup>South Urals State University, Chelyabinsk 454080, Russia.

Translated from Khimiya Geterotsiklicheskikh Soedinenii, No. 10, pp. 1590-1592, October, 2009.  
Original article submitted September 1, 2009.

catalysis conditions, we unexpectedly obtained another representative of this heterocyclic system, namely, 3-methyl[1,3]thiazolo[3',2':2,3][1,2,4]triazino[5,6-*b*]indole (**3**). Triazinoindole **1** is formed by the reaction of 5H-[1,2,4]triazino[5,6-*b*]indole-3-thiol (**4**) with propargyl bromide in the NaOH-H<sub>2</sub>O-DMSO superbase medium [2]. Triazinoindole **3** is probably formed by intramolecular cyclization of triazinoindole **1** through intermediate **3a**. Thus, heating triazinoindole **1** in aqueous NaOH at reflux leads to triazinoindole **3**.

The <sup>1</sup>H NMR spectrum of triazinoindole **3** shows singlets for the methyl group protons at 2.64 ppm and for H-2 at 7.47 ppm. The mass spectrum has a molecular ion peak (M<sup>+</sup> 240).

The <sup>1</sup>H NMR spectrum was taken on a Bruker spectrometer at 400 MHz in DMSO-d<sub>6</sub> with TMS as the internal standard. The electron impact mass spectra were taken on a Hewlett-Packard GC/MS computer with an HP-5890 gas chromatograph at 70 eV.

**3-Propargylthio-5H-[1,2,4]triazino[5,6-*b*]indole (1).** Propargyl bromide (0.152 g, 1 mmol) was added to a mixture of indolethiol **4** (0.202 g, 1 mmol) and NaOH (0.04 g, 1 mmol) in DMSO (10 ml). The reaction mixture was stirred for 1.5 h at room temperature. Then, water (50 ml) was added. The colorless light brown precipitate was filtered off, washed with water, and dried to give 0.202 g (84%) triazinoindole **1**; mp ~173°C. <sup>1</sup>H NMR spectrum, δ, ppm: 3.17 (1H, s, CH≡); 4.13 (2H, s, S-CH<sub>2</sub>); 7.45, 7.60, 7.71, 8.31 (4H, benzene ring).

**3-Methyl[1,3]triazolo[3',2':2,3][1,2,4]triazino[5,6-*b*]indole (3).** A. A solution of thiosemicarbazone **2** (0.5 g, 2 mmol) [3] in 1 N NaOH (40 ml) was heated at reflux for 3 h. Then, propargyl bromide (0.24 g, 2 mmol) and triethylbenzylammonium chloride (50 ml) as the phase-transfer catalyst were added and the mixture was stirred for 1.5 h at room temperature. The red precipitate was filtered off, washed with water, and dried to give triazinoindole **3** in 43% yield; mp 140°C (ethanol, dec.). <sup>1</sup>H NMR spectrum, δ, ppm: 2.64 (3H, s, CH<sub>3</sub>); 7.47 (1H, s, H-2); 7.33, 7.69, 8.23 (4H, benzene ring). Found, %: C 59.74; H 3.31; N 23.42. C<sub>12</sub>H<sub>8</sub>N<sub>4</sub>S. Calculated, %: C 59.98; H 3.36; N 23.32.

B. A solution of triazinoindole **1** (0.1 g, 0.4 mmol) in 1 N NaOH (5 ml) was heated at reflux for 1.5 h. The red precipitate was filtered off, washed with water, and dried to give triazinoindole **3** in 60% yield. This product was identical to the product obtained by method A and the melting point of a mixed sample was not depressed.

## REFERENCES

1. A. B. Tomchin, *Zh. Org. Khim.*, **18**, 1272 (1982).
2. A. V. Zhuravleva, A. A. Koryukova, and D. G. Kim, in: *XI Youth Conference on Organic Chemistry* [in Russian], Yekaterinburg (2008), p. 332.
3. G. I. Zhungietu and M. A. Rekhter, *Isatin and Its Derivatives* [in Russian], Shtunitsa, Kishinev (1977), p. 109.