## Cyclocondensation of Acylketene S,N- and N,N-acetals with Maleic Anhydride and Maleimide: A Facile One-Step Synthesis of Pyrano[3,4-c]pyrrole, Pyrrolo [3,4-c]pyridine and Condensed Pyrrole Derivatives<sup>1</sup>

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The acylketene S,N-acetals  $1\mathbf{a}-\mathbf{c}$  react with maleic anhydride (2) in refluxing acetonitrile to give the corresponding 3-pyrrolin-2-one-3-acetic acid derivatives  $3\mathbf{a}-\mathbf{c}$  in good yields. Condensation of S,N-acetals  $1\mathbf{a}-\mathbf{j}$  and 2 in presence of acetic anhydride directly afforded the corresponding 2-substituted-3-methylthio-4-aryl (or methyl)-1,6-dioxo-2,3-dihydropyrano[3,4-c]pyrroles  $4\mathbf{a}-\mathbf{j}$  in excellent yields. The reaction of cyclic S,N-  $(5\mathbf{a}-\mathbf{c})$  and N,N-  $(5\mathbf{d}-\mathbf{f})$  acetals with 2 in refluxing acetonitrile gave the corresponding pyrrolo[2,1-b]thiazole  $(6\mathbf{a}-\mathbf{c})$  and pyrrolo[1,2-a]imidazole  $(6\mathbf{d}-\mathbf{f})$  derivatives, respectively, in good yields. Similarly, the condensation of S,N-acetals  $1\mathbf{a}$  and  $1\mathbf{d}$  with maleimide directly yielded the respective pyrrolo[3,4-c]pyridine derivatives  $9\mathbf{a}$  and  $9\mathbf{b}$ , while the corresponding pyrrolinone-3-acetamide derivatives  $8\mathbf{a}$  and  $8\mathbf{b}$  were obtained under similar conditions, when the reaction time was reduced.

In our recent report, we have described the synthesis of novel 2(1H)-pyridones by cyclocondensation of acylketene S,N-acetals with malonyl chloride. In continuation of these studies, we now report the reactions of these intermediates and few cyclic N,N-acetals with maleic anhydride and maleimide, which provide facile one step synthesis of functionalized pyrano[3,4-c]pyrrole, pyrrolo[3,4-c]pyridine and other condensed pyrrole derivatives.

When the S,N-acetal 1a was reacted with maleic anhydride in refluxing acetonitrile, work-up of the reaction mixture afforded a product (87%), which was characterized as pyrrolinone-3-acetic acid (3a). The corresponding N-benzyl-, 1b, and N-phenyl-, 1c, S,N-acetals similarly gave the respective pyrrolinones 3b and 3c in good yields. The pyrrolinone 3a was cyclized in presence of acetic anhydride, when the corresponding 2-ethyl-3-methylthio-4-phenyl-1,6-dioxo-2,3-dihydropyrano[3,4-c]pyrrole (4a) was obtained in 92% yield. In an alternative experiment, the pyranopyrrole 4a was directly obtained in one step in comparable

$$\begin{array}{c} R^{2} \\ CH_{3}CN \\ \Delta, 5h \\ \hline 79-88\% & CH_{3}S_{5}N^{2} = C_{2}H_{5} \\ DR^{1} = C_{6}H_{5}, R^{2} = C_{2}H_{5} \\ DR^{1} = C_{6}H_{5}, R^{2} = C_{8}H_{5}CH_{2} \\ CR^{1} = R^{2} = C_{6}H_{5} \\ \end{array}$$

$$\begin{array}{c} CH_{3}CN \\ CR^{1} = R^{2} = C_{6}H_{5} \\ CR^{1} = R^{2} = C_{6}H_{5} \\ CH_{3}CN \\ CH_{3}CO)_{2}O \\ \Delta, 30 \text{ min} \\ \end{array}$$

$$\begin{array}{c} CH_{3}CN \\ CH_{3}CO)_{2}O \\ \Delta, 8h \\ \hline 75-88\% & CH_{3}S_{3} = N \end{array}$$

1,4	R <sup>1</sup>	R <sup>2</sup>	1,4	R¹	R²
a b c d	C <sub>6</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> 4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> C <sub>6</sub> H <sub>5</sub> CH <sub>3</sub> CH <sub>3</sub>	f g h i	4-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub> 4-ClC <sub>6</sub> H <sub>4</sub> 4-ClC <sub>6</sub> H <sub>4</sub> CH <sub>3</sub> 4-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	C <sub>2</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub> C <sub>2</sub> H <sub>5</sub> C <sub>6</sub> H <sub>5</sub> CH <sub>2</sub>

Table 1. Products 3a-c and 8a-b Prepared

Prod- uct	Yield (%)	mp <sup>a</sup> (°C)	Molecular Formula <sup>b</sup>	IR (KBr) <sup>c</sup> v (cm <sup>-1</sup> )	$^{1}$ H-NMR (CDCl <sub>3</sub> ) $\delta$ , $J$ (Hz)	MS (70 eV) <sup>e</sup> m/z (%)
3a	87	138	C <sub>16</sub> H <sub>17</sub> NO <sub>4</sub> S (319.4)	1720, 1672, 1642	1.15 (t, 3H, $J = 7$ , $CH_2CH_3$ ); 1.67 (s, 3H, $SCH_3$ ); 3.25 (s, 2H, $CH_2CO_2H$ ); 3.32–4.00 (m, 2H, $CH_2CH_3$ ); 5.37 (s, 1H, H-5); 7.33–7.82 (m, $5H_{arom}$ ); 9.29 (s, 1H, OH, exchangeable with $D_2O$ )	319 (M <sup>+</sup> , 1); 275 (9); 228 (100)
3b	79	154	C <sub>21</sub> H <sub>19</sub> NO <sub>4</sub> S (381.5)	1705, 1684, 1660	1.74 (s, 3 H, SCH <sub>3</sub> ); 3.30 (s, 2 H, CH <sub>2</sub> CO <sub>2</sub> H); 4.20 (d, 1 H, $J = 17$ , H <sub>A</sub> of CH <sub>2</sub> ); 5.15 (d, 1 H, $J = 17$ , H <sub>B</sub> of CH <sub>2</sub> ); 5.16 (s, 1 H, H-5); 7.20 (s, 5 H <sub>arom</sub> ); 7.33–7.80 (m, 5 H <sub>arom</sub> ); 9.13 (br s, 1 H, OH, exchangeable with D <sub>2</sub> O)	337 (M <sup>+</sup> – 44,4); 290 (28)
3e	82	168	C <sub>20</sub> H <sub>17</sub> NO <sub>4</sub> S (367.4)	1716, 1660, 1640	1.68 (s, 3H, SCH <sub>3</sub> ); 3.39 (s, 2H, CH <sub>2</sub> ); 5.95 (s, 1H, H-5); 7.10–7.93 (m, $10\mathrm{H_{arom}}$ ); 9.08 (br s, 1H, OH, exchangeable with D <sub>2</sub> O)	367 (M <sup>+</sup> , 3); 323 (9); 276 (83)
8a	88	156	$C_{15}H_{16}N_2O_3S$ (304.4)	3360, 3170, 1680, 1650, 1624	1.73 (s, 3H, SCH <sub>3</sub> ); 3.03 (s, 3H, NCH <sub>3</sub> ); 3.33 (s, 2H, CH <sub>2</sub> ); 5.37 (s, 1H, H-5); 6.00, 6.80 (br s, 1H each, NH <sub>2</sub> , exchangeable with D <sub>2</sub> O); 7.37-7.68 (m, 3H <sub>arom</sub> ); 7.84-8.03 (m, 2H <sub>arom</sub> )	304 (M <sup>+</sup> , 9); 257 (19); 214 (100)
8b	85	158	C <sub>16</sub> H <sub>18</sub> N <sub>2</sub> O <sub>3</sub> S (318.4)	3360, 3168, 1678, 1650, 1625	1.20 (t, 3H, $J = 7$ , $CH_2CH_3$ ); 1.71 (s, 3H, $SCH_3$ ); 3.23 (s, 2H, $CH_2CO$ ); 3.15–3.63 (dq, 1H, $J = 17$ , $H_A$ of $CH_2CH_3$ ); 3.63–4.00 (dq, 1H, $J = 17$ , $H_B$ of $CH_2CH_3$ ); 5.56 (s, 1H, H-5); 6.60, 7.22 (br s, 1H each, $NH_2$ , exchangeable with $D_2O$ ); 7.38–8.13 (m, $5H_{arom}$ ) <sup>r</sup>	318 (M <sup>+</sup> , 11); 228 (M <sup>+</sup> – 90, 100)

<sup>&</sup>lt;sup>a</sup> Uncorrected, recorded on Thomas Hoover apparatus.

<sup>&</sup>lt;sup>b</sup> Satisfactory microanalyses obtained:  $C \pm 0.31$ ,  $H \pm 0.29$ ,  $N \pm 0.28$ .

Recorded on Perkin-Elmer 297 spectrometer.

<sup>&</sup>lt;sup>d</sup> Recorded on 90 MHz using a Varian EM-390 spectrometer.

Recorded on Jeol JMS-D 300 spectrometer.

f In DMSO-d<sub>6</sub> CDCl<sub>3</sub>.

yield (85%), when 1a and 2 were refluxed in acetonitrile in the presence of acetic anhydride. The other substituted pyranopyrrole 4b-j were similarly obtained from the respective 1b-j and 2 in 75-88% overall yields. The reaction was extended to cyclic S,N-5a-c and N,N-5d-f acetals also. Thus 5a-c and 2 in refluxing acetonitrile afforded the corresponding pyrrolo[2,1-b]thiazole derivatives 6a-c in 85-91% overall yields. Similarly, 5d-f under identical conditions yielded the respective pyrrolo[1,2-a]imidazolines 6d-f in good yields. The structures of 6a-f were established with the help of spectral and analytical data (Table 3). Attempted cyclization of 6a or 6d in refluxing acetic anhydride, however gave only intractable mixture of products.

6 5, 6 R X 5, 6 R X 4-ClC<sub>6</sub>H<sub>4</sub> d C<sub>6</sub>H<sub>5</sub> NH 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> S 4-ClC<sub>6</sub>H<sub>4</sub> NH 4-CH<sub>3</sub>OC<sub>6</sub>H<sub>4</sub> CH<sub>3</sub> NH Reaction of 1 with maleimide was next investigated. A few enaminoesters/nitriles and 2-(benzylamino) propenylmethyl ketone are reported<sup>4</sup> to have been condensed with maleimide recently to give 5-oxo-2-pyrrolin-4-acetamide derivatives, which were further cyclized in the presence of base to afford pyrrolo[3,4-c]pyridine derivatives in good yields. Incidentally, when 1 a and 1 d were refluxed with maleimide in acetonitrile for 8 hours, the corresponding pyrrolopyridines 9 a and 9 b were directly obtained in good yields. The open-chain analogs 2-oxo-3-pyrrolin-3-acetamide derivatives 8 a and 8 b were also formed under similar conditions, when reaction time was reduced (2 hours).

1a, 1d + NH 
$$R^{1}$$
  $CH_{3}CN, \Delta, 2h$   $R^{1}$   $CONH_{2}$   $R^{1}$   $CONH_{2}$   $R^{1}$   $CONH_{2}$   $R^{1}$   $R^{1}$ 

Table 2. Products 4a-j and 9a-b Prepared

Prod- uct	Yield (%)	mp <sup>a</sup> (°C)	Molecular Formula <sup>b</sup>	$IR (KBr)^{c}$ $v (cm^{-1})$	$^{1}$ H-NMR (Solvent/TMS) $^{d}$ $\delta$ , $J$ (Hz)	MS (70 eV) <sup>e</sup> m/z (%)
4a	85	148	C <sub>16</sub> H <sub>15</sub> NO <sub>3</sub> S (301.4)	1727, 1693	DMSO- $d_6$ : 1.26 (t, 3H, $J = 7$ , CH <sub>2</sub> CH <sub>3</sub> ); 1.50 (s, 3H, SCH <sub>3</sub> ); 3.48 (dq, 1H, $J = 17$ , H <sub>A</sub> of CH <sub>2</sub> CH <sub>3</sub> ); 3.78 (dq, 1H, $J = 17$ , H <sub>B</sub> of CH <sub>2</sub> CH <sub>3</sub> ); 6.36 (s, 1H, H-3); 6.49 (s, 1H, H-7); 7.40–7.45 (m, 3H <sub>2</sub> ); 7.75 8.12 (m, 3H <sub>3</sub> ); 7.75	254 (M <sup>+</sup> – 47, 100); 226 (16)
4b	80	134	C <sub>21</sub> H <sub>17</sub> NO <sub>3</sub> S (363.4)	1719, 1710	$^{3}$ H <sub>arom</sub> ); 7.75–8.12 (m, $^{2}$ H <sub>arom</sub> ) CF <sub>3</sub> CO <sub>2</sub> H: 1.60 (s, $^{3}$ H, SCH <sub>3</sub> ); 4.56 (d, $^{1}$ H, $^{J}$ = 15, H <sub>A</sub> of CH <sub>2</sub> ); 5.38 (d, $^{1}$ H, $^{J}$ = 15, H <sub>B</sub> of CH <sub>2</sub> ); 5.68 (s, $^{1}$ H, H-3); 6.98 (s, $^{1}$ H, H-7); 7.18–7.84 (m, $^{1}$ 0 H <sub>arom</sub> )	316 (M <sup>+</sup> - 47, 100)
4c	86	192	C <sub>20</sub> H <sub>15</sub> NO <sub>3</sub> S (349.4)	1720, 1705	DMSO-d <sub>6</sub> : 1.53 (s, 3 H, SCH <sub>3</sub> ); 6.65 (s, 1 H, H-3); 7.08 (s, 1 H, H-7); 7.32-7.82 (m, 8 H <sub>arom</sub> ); 7.83-8.00 (m, 2 H <sub>arom</sub> )	302 (M <sup>+</sup> - 47, 100); 274 (10)
4d	84	176	C <sub>15</sub> H <sub>13</sub> NO <sub>3</sub> S (287.3)	1721, 1698	CDCl <sub>3</sub> : 1.66 (s, 3 H, SCH <sub>3</sub> ); 3.14 (s, 3 H, NCH <sub>3</sub> ); 5.70 (s, 1 H, H-3); 6.60 (s, 1 H, H-7); 7.32–7.65 (m, 3 H <sub>arom</sub> ); 7.65–7.94 (m, 2 H <sub>arom</sub> )	240 (M <sup>+</sup> – 47, 100); 212 (18)
<b>4e</b>	81	164	C <sub>16</sub> H <sub>15</sub> NO <sub>4</sub> S (317.4)	1720, 1685	CF <sub>3</sub> CO <sub>2</sub> H: 1.20 (s, 3H, SCH <sub>3</sub> ); 2.89 (s, 3H, NCH <sub>3</sub> ); 3.53 (s, 3H, OCH <sub>3</sub> ); 5.45 (s, 1H, H-3); 6.43 (s, 1H, H-7); 6.70 (d, A <sub>2</sub> B <sub>3</sub> ).	370 (M <sup>+</sup> – 47, 100); 242 (12)
4f	84	158	C <sub>17</sub> H <sub>17</sub> NO <sub>4</sub> S (331.4)	1711 (br)	$2H_{arom}$ ); 7.43 (d, $A_2B_2$ , $2H_{arom}$ ) $CF_3CO_2H$ : 1.40 (t, 3H, $J=7$ , $CH_2CH_3$ ); 1.60 (s, 3H, $SCH_3$ ); 3.50–4.28 (m, 2H, $NCH_2$ ); 3.97 (s, 3H, $OCH_3$ ); 5.97 (s, 1H, H-3);	284 (M <sup>+</sup> – 47, 100); 256 (10)
4g	88	218	C <sub>20</sub> H <sub>14</sub> ClNO <sub>3</sub> S (383.9)	1727, 1708	6.89 (s, 1 H, H-7); 7.15 (d, A <sub>2</sub> B <sub>2</sub> , 2 H <sub>arom</sub> ); 7.87 (d, A <sub>2</sub> B <sub>2</sub> , 2 H <sub>arom</sub> ) CF <sub>3</sub> CO <sub>2</sub> H: 1.73 (s, 3 H, SCH <sub>3</sub> ); 6.57 (s, 1 H, H-3); 7.18 (s, 1 H, H-7); 7.45–7.80 (m, 7 H <sub>arom</sub> ); 7.81–8.11 (d, A <sub>2</sub> B <sub>2</sub> , 2 H <sub>arom</sub> )	338 (36); 336 (M <sup>+</sup> – 47, 100)
4h	80	124	C <sub>21</sub> H <sub>16</sub> ClNO <sub>3</sub> S (397.9)	1718, 1704	CDCl <sub>3</sub> : 1.58 (s, 3 H, SCH <sub>3</sub> ); 4.36 (d, 1 H, $J = 15$ , H <sub>A</sub> of CH <sub>2</sub> ); 5.33 (d, 1 H, $J = 15$ , H <sub>B</sub> of CH <sub>2</sub> ); 5.48 (s, 1 H, H-3); 6.70 (s, 1 H, H-7);	308 (8) 352 (9); 350 (M <sup>+</sup> - 47, 18)
4i	79	108	C <sub>11</sub> H <sub>13</sub> NO <sub>3</sub> S (239.3)	1728, 1692	7.33 (s, $5H_{arom}$ ); 7.30–7.71 (dd, $A_2B_2$ , $4H_{arom}$ ) CF <sub>3</sub> CO <sub>2</sub> H: 1.34 (t, $3H$ , $J=7$ , CH <sub>2</sub> CH <sub>3</sub> ); 1.74 (s, $3H$ , SCH <sub>3</sub> ); 2.60 (s, $3H$ , ArCH <sub>3</sub> ); 3.45–4.32 (m, $2H$ , CH <sub>2</sub> CH <sub>3</sub> ); 5.73 (s, $1H$ , H-3);	239 (M <sup>+</sup> , 19); 192 (M <sup>+</sup> – 47,
4j	75	91	C <sub>22</sub> H <sub>19</sub> NO <sub>3</sub> S (377.5)	1715, 1700	6.92 (s, 1H, H-7) $CDCl_3$ : 1.55 (s, 3H, $SCH_3$ ); 2.40 (s, 3H, $CH_3$ ); 4.40 (d, 1H, $J = 15$ , $H_A$ of $CH_2$ ); 5.35 (d, 1H, $J = 15$ , $H_B$ of $CH_2$ ); 5.58 (s, 1H, H-3); 6.72 (s, 1H, H-7); 7.41 (s, 5H <sub>aram</sub> ); 7.28 (d, A <sub>2</sub> B <sub>2</sub> , 2H <sub>aram</sub> ); 7.64 (d.	100); 164 (30) 330 (M <sup>+</sup> – 47, 37)
9a	70	249	$C_{15}H_{14}N_2O_2S$ (286.4)	1685, 1660, 1623	A <sub>2</sub> B <sub>2</sub> , 2H <sub>arom</sub> ) DMSO-d <sub>6</sub> : 1.33 (s, 3H, SCH <sub>3</sub> ); 3.04 (s, 3H, CH <sub>3</sub> ); 6.07 (s, 1H, H-3); 6.63 (s, 1H, H-7); 7.38-7.78 (m, 5H <sub>arom</sub> ); 11.88 (br s, 1H, NH,	239 (M <sup>+</sup> – 47, 100)
)b	82	218	$C_{16}H_{16}N_2O_2S$ (300.4)	1684, 1660, 1625	exchangeable with $D_2O$ ) $CDCl_3$ : 1.24 (t, 3H, $J=7$ , $CH_2CH_3$ ); 1.38 (s, 3H, $SCH_3$ ); 3.41 (dq, 1H, $J=17,7$ , $H_A$ of $CH_2$ ); 3.90 (dq, 1H, $J=17,7$ , $H_B$ of $CH_2$ ); 5.70 (s, 1H, H-3); 6.85 (s, 1H, H-7); 7.58 (br s, $5H_{arom}$ )	253 (M <sup>+</sup> – 47, 100); 225 (26)

<sup>\*</sup> As in Table 1.

<sup>&</sup>lt;sup>b</sup> Satisfactory microanalyses obtained:  $C \pm 0.28$ ,  $H \pm 0.32$ , N + 0.30.

SYNTHESIS

Table 3. Products 6a-f Prepared

Product	Yield (%)	mp <sup>a</sup> (°C)	Molecular Formula <sup>b</sup>	IR (KBr) <sup>c</sup> ν (cm <sup>-1</sup> )	$^{1}\text{H-NMR (DMSO-}d_{6}/\text{TMS})^{4}$ $\delta,~J(\text{Hz})$	MS (70 eV)° m/z (%)
6a	85	220	C <sub>15</sub> H <sub>12</sub> CINO <sub>4</sub> S (337.8)	1733, 1647, 1612	2.70-2.91 (m, 2H, CH <sub>2</sub> CO); 3.00-3.28 (m, 2H, SCH <sub>2</sub> ); 3.55-3.86 (m, 2H, NCH <sub>2</sub> ); 4.29-4.60 (m, 1H, H-6); 7.52 (s, 4H <sub>arom</sub> )	339 (8); 337 (M <sup>+</sup> , 24); 295 (34); 293 (M <sup>+</sup> - 44, 100)
6b	87	170	C <sub>16</sub> H <sub>15</sub> NO <sub>5</sub> S (333.4)	1722, 1646, 1603	2.80 (d, 2H, $J = 5.5$ , CH <sub>2</sub> CO); 2.97–3.18 (m, 2H, SCH <sub>2</sub> ); 3.57–3.97 (m, 2H, NCH <sub>2</sub> ); 3.83 (s, 3H, OCH <sub>3</sub> ); 4.23–4.68 (m, 1H, H-6); 6.90 (d, A <sub>2</sub> B <sub>2</sub> , 2H <sub>arom</sub> ); 7.52 (d, A <sub>2</sub> B <sub>2</sub> , 2H <sub>arom</sub> )	333 (M <sup>+</sup> , 4); 289 (M <sup>+</sup> -44, 34)
6c	91	222	C <sub>10</sub> H <sub>11</sub> NO <sub>4</sub> S (241.2)	1720, 1650, 1640	2.31 (s, 3H, CH <sub>3</sub> ); 2.66–2.89 (m, 2H, CH <sub>2</sub> CO); 2.90–3.28 (m, 2H, SCH <sub>2</sub> ); 3.44–3.96 (m, 2H, NCH <sub>2</sub> ); 4.18–4.50 (m, 1H, H-6)	241 (M <sup>+</sup> , 4); 197 (M <sup>+</sup> - 44, 70)
6d	82	198	$C_{15}H_{14}N_2O_4$ (286.3)	3258, 1718, 1680, 1620	2.59-2.80 (m, 2H, CH <sub>2</sub> CO); 3.43-4.04 [m, 5H, N(CH <sub>2</sub> ) <sub>2</sub> and H-6]; 7.35 (s, 5H <sub>arom</sub> ); 9.55 (br s, 1H, OH, exchangeable with D <sub>2</sub> O)	268 (M <sup>+</sup> – 18, 10); 242 (M <sup>+</sup> – 44,5)
6e	87	150	C <sub>15</sub> H <sub>13</sub> ClN <sub>2</sub> O <sub>4</sub> (320.7)	3265, 1735, 1712, 1639	2.28–2.81 (m, 2H, $CH_2CO$ ); 3.38–4.12 [m, 5H, $N(CH_2)_2$ and H-6]; 7.23–7.61 (dd, $A_2B_2$ , $4H_{arom}$ ); 7.97 (br s, 1H, OH, exchangeable with $D_2O$ )	304 (1); 302 (M <sup>+</sup> – 18, 5); 278 (6); 276 (M <sup>+</sup> – 44,2)
6f	88	168	$C_{16}H_{16}N_2O_5$ (316.3)	3222, 1708, 1691, 1629	2.56–2.78 (m, $^{2}$ H, $^{2}$ CH); $^{2}$ 3.48–4.15 [m, $^{2}$ 5H, $^{2}$ N( $^{2}$ CH) <sub>2</sub> and $^{2}$ H-6]; $^{2}$ 3.77 (s, $^{2}$ 3H, $^{2}$ OCH <sub>3</sub> ); $^{2}$ 6.88 (d, $^{2}$ B <sub>2</sub> , $^{2}$ 2H <sub>arom</sub> ); $^{2}$ 7.35 (d, $^{2}$ B <sub>2</sub> , $^{2}$ 2H <sub>arom</sub> ); $^{2}$ 9.50 (br s, $^{2}$ 1H, OH, exchangeable with $^{2}$ D <sub>2</sub> O)	298 (M <sup>+</sup> – 18,2)

<sup>&</sup>lt;sup>a</sup> As in Table 1.

The condensation of S,N- and N,N-acetals 1 and 5 with maleic anhydride and maleimide provides a convenient one step method for the synthesis of a variety of pyrano[3,4-c]pyrrole, pyrrolo[3,4-c]pyridine and condensed pyrrole derivatives under simple reaction conditions. Few pyrano[3,4-c]pyrrole derivatives are described in the literature,  $^{5-7}$  which are obtained through multistep routes and in one case, by acid catalyzed cyclization of naturally occurring Kainic acid. The pyrrolo[2,1-b]thiazolines  $\bf 6a-c$  possess structural framework similar to  $\gamma$ -lactam analogs of penicillanic acid. with potential biological activity.

The starting S,N-acetals 1a-j, 5a-c and N,N-acetals 5d-f were prepared according to earlier reported procedures.

1-Alkyl/aryl-4-aroyl-5-methylthio-2-oxo-3-pyrrolin-3-acetic Acid (3a-c); 7-Acetyl/aroyl-5-oxo-2,3,5,6-tetrahydropyrrolo[2,1-b]thiazole-6-acetic Acid (6a-c); 7-Aroyl-5-oxo-2,3,5,6-tetrahydro-1 H-pyrrolo[1,2-a]imid-azole-6-acetic Acid (6d-f); 1-Alkyl-4-benzoyl-5-methylthio-2-oxo-3-pyrrolin-3-acetamide(8a-b), and 2-Alkyl-3-methylthio-4-phenyl-1,6-di-oxo-2,3,5,6-tetrahydro-5 H-pyrrolo[3,4-c]pyridine (9a-b): General Procedure:

A solution of 1 or 5 (20 mmol) and 2 or 7 (20 mmol) in MeCN (30 mL) is refluxed for 2 h ( $6\mathbf{a}$ - $\mathbf{f}$ ,  $8\mathbf{a}$ - $\mathbf{b}$ ), 5 h ( $3\mathbf{a}$ - $\mathbf{c}$ ) and 8 h ( $9\mathbf{a}$ - $\mathbf{b}$ ). The mixture is cooled, poured into ice-water (50 mL), extracted with CHCl<sub>3</sub> ( $2 \times 50$  mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give crude products, which are purified either by crystallization from MeOH ( $6\mathbf{a}$ - $\mathbf{f}$ ,  $9\mathbf{a}$ - $\mathbf{b}$ ) or by filtering through a silica gel column ( $3\mathbf{a}$ - $\mathbf{c}$ ,  $8\mathbf{a}$ - $\mathbf{b}$ ) using hexane/EtOAc (9:1) as eluent (see Tables 1-3).

## 2-Alkyl/aryl-4-aryl/methyl-3-methylthio-1,6-dioxo-2,3-dihydropyrano-[3,4-c]pyrrole (4a-j); General Procedure:

To a solution of 1 (20 mmol) and 2 (1.96 g, 20 mmol) in MeCN (30 mL), freshly distilled acetic anhydride (2.04 g, 20 mmol) is added and the mixture is refluxed for 8 h, cooled and poured into ice-water (50 mL), The product is extracted with CHCl<sub>3</sub> (2×50 mL), dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated to give crude products, which are purified by filtering through silica gel column using hexane/EtOAc (9:1) as eluent and subsequently crystallized from CHCl<sub>3</sub>.

## Cyclization of 3a: 2-Ethyl-3-methylthio-4-phenyl-1,6-dioxo-2,3-dihydro-pyrano[3,4-c]pyrrole (4a); Typical Procedure:

To a solution of **3a** (3.20 g, 10 mmol) in acetonitrile (20 mL), freshly distilled acetic anhydride (2.04 g, 20 mmol) is added and the mixture is refluxed for 30 min, worked-up as described for **4a**—j to give crude **4a**, which on column chromatography over silica gel (hexane/EtOAc eluent 9:1) gives pure **4a**; yield: 2.7 g (92 %) (mixed mp, superimposable IR and <sup>1</sup>H-NMR spectra).

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- (1) Part 66 of the series on Polarized Ketene S,S- and S,N-acetals. Part 65: Singh, L. W., Ila, H., Junjappa, H. J. Chem. Soc. Perkin Trans. 1, in press.
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<sup>&</sup>lt;sup>b</sup> Satisfactory microanalyses obtained:  $C \pm 0.34$ ,  $H \pm 0.26$ ,  $N \pm 0.31$ .

c-e As in Table 1.

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- Baldo, M.A., Chessa, G., Marangoni, G., Pitteri, B. Synthesis 1987, 720. On p. 722, line 6, "degree of functionalization" should read "yield of binding". On the same page in the preparation of bis-hydrazone 5, line 8, "solid product" should read "oil".
- Singh, L.W., Ila, H., Junjappa, H. Synthesis 1988, 89. On p. 90 in the <sup>1</sup>H-NMR data for dioxime 4, line 2, "N<sub>2</sub>" should read "NH<sub>2</sub>".
- Burger, K., Hübl, D., Geith, K. Synthesis 1988, 194. On p. 196 in the table, for entries 41, 4m, and 4n, Y = O, and Nu = Cl, Br, and  $C_6H_5$ , respectively.
- -- Tolstikov, A.G., Khakhalina, N.V., Spirikhin, L.V. Synthesis 1988, 221. In the title and abstract, benzyl esters should read benzyl ethers.
- Gupta, A.K., Ila, H., Junjappa, H. *Synthesis* **1988**, 284. Compounds **4** are 1,6-dioxo-1,2,3,6-tetrahydropyrano[3,4-c]pyrroles; compounds **9** are 1,6-dioxo-2,3,5,6-tetrahydro-1*H*-pyrrolo[3,4-c]pyridines.
- Keshavarz-K., M., Cox, S.D., Angus, R.O., Jr., Wudl, F.
   Synthesis 1988, 641. On p. 642 the IR spectra shown in Figures 2 and 3 should be interchanged.

- Rodriguez, J., Waegell, B. Synthesis **1988**, 534. On p. 535, the first line of the general procedure should read: "DMAP (0.92 g, 7.5 mmol) and then α,β-unsaturated aldehyde **1** (0.1 mol)…"
- Zbiral, E., Drescher, M. Synthesis 1988, 735. On p. 738 in the last procedure, the name for compounds 14 should read: (5-Oxo-5,6-dihydroimidazo[1,2-c]pyrimidin-3-yl)methylphosphonsäuren.
- Valerio, R. M., Alewood, P. F., Johns, R. B. Synthesis 1988.
  786. On p. 787 formula 2 should be:

Also on p. 787 in the reaction of 5 in the scheme on the right side, the reagent should be:

- Garrigues, B., Mulliez, M. *Synthesis* **1988**, 810. The title should read: Salts of *N*-(Sulfoalkyl)ureas and -thioureas.
- Yokoyama, M., Watanabe, S., Seki, T. Synthesis 1988, 879.
   On p. 880 the name of compound 3a in the first procedure should be azido(2-benzyloxyethoxy)methane.