View Online / Journal Homepage / Table of Contents for this issue

Studies of [3,3]Sigmatropic Rearrangements: Rearrangement of 3-(4-*p*-Tolyloxybut-2-ynyloxy)[1]benzopyran-2-one

K. C. Majumdar,** R. N. De,* A. T. Khan,* S. K. Chattopadhyay,* K. Dey,* and A. Patrabt

^a Department of Chemistry, University of Kalyani, Kalyani 741235, W. B., India ^b University College of Science, 92 A.P.C. Road, Calcutta 700009, India

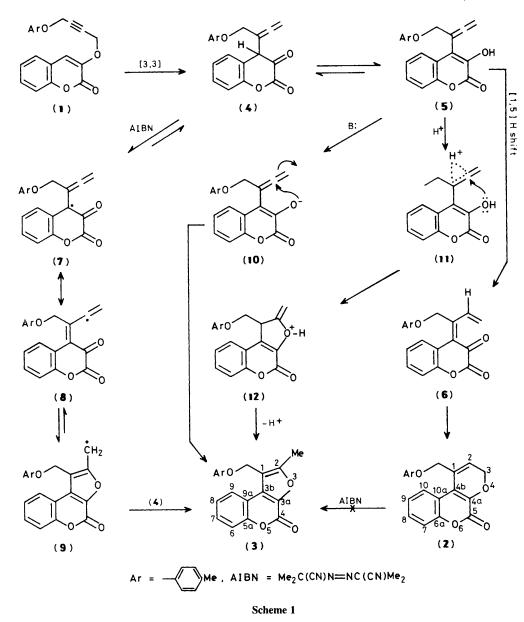
3-(4-*p*-Tolyloxybut-2-ynyloxy)[1]benzopyran-2-one (1), when refluxed in chlorobenzene, ethyl benzene, or xylene, gave exclusively 1-(*p*-tolyloxymethyl)pyrano[2,3-*c*][1]benzopyran-5(3*H*)-one (2), whereas through an ionic or radical pathway 2-methyl-1-(*p*-tolyloxymethyl)furo[2,3-*c*][1]benzopyran-4-one (3) was the exclusive product.

3-Alkyl and 4-alkyl coumarins ([1]benzopyran-2-ones) are reported¹⁻⁴ to have anthelmintic, hypnotic, insecticidal, antifungal activities and anticoagulant effect on blood. A considerable amount of work has been done on the synthesis of these compounds, but not much on the synthesis of 3,4-fused coumarins.⁵⁻⁷ Recently we have achieved the regioselective synthesis⁸ of 4-aryloxymethyl pyrano [3,2-c]-[1]benzopyran-2-ones from 4-(4-aryloxybut-2-ynyloxy)-[1]benzopyran-2-ones. Here we report the results of the [3,3]-sigmatropic rearrangement of 3-(4-*p*-tolyloxybut-2ynyloxy)[1]benzopyran-2-one (1) (Scheme 1).

Compound (1), m.p. 155 °C, was synthesised from 4-chloro-1-(*p*-tolyloxy)-but-2-yne and 3-hydroxy[1]benzopyran-2-one in anhydrous Me₂CO–K₂CO₃ (yield 80%). The product (1) was refluxed in carefully purified‡ chlorobenzene (b.p. 132 °C) for 4 h, then chlorobenzene was removed *in vacuo* and the crude mass was subjected to column chromatography over

 \ddagger The solvents, chlorobenzene, ethylbenzene, and xylene were purified by successively washing with dil. H_2SO_4 , dil. NaOH, water, and freshly prepared iron(II) sulphate solution, dried (Na₂SO₄), refluxed with hydroquinone, and distilled twice before use.

[†] The homodecoupling, ¹³C n.m.r., and heteroatom correlation (HETCOR) experiments on compound (3) were performed by Dr. A. Patra.



silica gel. The exclusive product was 1-(p-tolyloxymethyl)pyrano[2,3-c][1]benzopyran-5(3H)-one (2) (yield 92%; m.p. 135 °C). The same reaction in refluxing ethylbenzene (b.p. 136 °C) or xylene‡ (140 °C) for 4 h also gave the product (2) exclusively; no furo[2,3-c]benzopyran-4-one was obtained.

We then studied the rearrangement of (1) in the presence of a radical initiator. When compound (1) was heated in chlorobenzene‡ in the presence of azobisisobutyronitrile (AIBN) (5 mg) for 2.5 h, 2-methyl-1-(*p*-tolyloxymethyl)furo[2,3-*c*][1]benzopyran-4-one (3), m.p. 194 °C, was obtained in 80% yield, along with a 20% yield of (2). Compound (3) was characterised by elemental analysis and spectral data and the structure was confirmed by homodecoupling, ¹³C n.m.r., and HETCOR experiments. Protonated carbon resonance assignments were established by direct correlation with proton resonances by HETCOR with normal one-bond C-H coupling, and non-protonated carbon resonances were established by a long-range (J 7 Hz) HETCOR experiment (primarily three-bond coupling). The benzene ring methyl proton resonance was correlated with the carbon resonances of C-3' and C-5'. The 2-methyl proton resonance was correlated with the non-protonated carbon resonance of C-1. The 1-CH₂ proton resonance was correlated with the non-protonated carbon resonances of C-3b and C-2. The aromatic C-9 proton was correlated with the non-protonated carbon resonances of C-5a and C-3b.

When compound (1) was heated at 140 °C for 4 h in non-polar solvents, viz. polyethylene glycol (PEG-600) or N,N-dimethylaniline, the furo[2,3-c]benzopyran-4-one (3) was obtained in 90% yield. When (1) was refluxed in pyridine (116 °C) for 20 h an 80% yield of (3) was obtained along with 20% of unchanged starting material. When (1) was refluxed in ethylbenzene (140 °C) in the presence of toluene-p-sulphonic acid, (3) was again obtained, in 90% yield. In these cases no pyrano[2,3-c]benzopyran-5-one was obtained.

A possible mechanism^{9,10} for the formation of (2) and (3) is outlined in Scheme 1.

We thank the C.S.I.R. (New Delhi) and the University of

Kalyani for financial assistance. We are also grateful to the referees for comments and suggestions.

Received, 31st July 1987;§ Com. 1113

References

- 1 P. Von Lauger, H. Martin, and P. Muller, Helv. Chim. Acta, 1944, 27, 892.
- 2 H. Kitagawal, R. Iwaki, B. Yanagi, and T. Sato, J. Pharm. Soc. Jpn., 1956, 76, 186.

§ Received in revised form, 15th March 1988.

- 3 T. O. Soine, J. Pharm. Sci., 1964, 53, 231.
- 4 F. M. Dean, 'Naturally Occurring Oxygen Ring Compounds,' Butterworths, London, 1963.
- 5 V. N. Dholakia and K. N. Trivedi, J. Indian Chem. Soc., 1971, 48, 344.
- 6 V. K. Ahluwalia, M. C. Gupta, and S. Mehta, Indian J. Chem., Sect. B, 1979, 17, 395. 7 A. Patra, A. K. Mukhopadhyay, and A. K. Mitra, Indian J.
- Chem., Sect. B, 1979, 17, 638.
- 8 K. C. Majumdar, D. P. Das, and A. T. Khan, Synth. Commun., in the press; K. C. Majumdar, A. T. Khan, and R. N. De, *ibid.*, in the press.
- 9 J. Zsindely and H. Schmid, Helv. Chim. Acta, 1968, 51, 1510.
- 10 N. Sarcevic, J. Zsindley, and H. Schmid, Helv. Chim. Acta, 1973, 56, 1457.