

### **Combined Directed Ortho Metalation/Cross-Coupling Strategies:** Synthesis of the Tetracyclic A/B/C/D Ring Core of the Antitumor **Agent Camptothecin**

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A convergent synthesis of the A/B/C/D ring fragment 5 of camptothecin using a combination of directed ortho metalation and Negishi cross-coupling is described. The key features of the synthetic sequence are an anionic ortho-Fries rearrangement  $(10 \rightarrow 12)$ , a Negishi cross-coupling  $(7 \rightarrow 6)$ , and a terminal modified von Braun reaction  $(16 \rightarrow 5)$  that leads to tetracyclic derivative 5 in 7 steps and 11% overall yield.

#### Introduction

(20S)-Camptothecin (1a), one of the most potent antitumor natural products, was first isolated in 1966 from *Camptotheca acuminata* Nyssaceae by Wall et al.<sup>1,2</sup> Its unusual structure and intriguing speculative biosynthetic pathway<sup>3</sup> stimulated immediate and intense synthetic activity worldwide,<sup>4,5</sup> which subsequently precipitously declined due to the finding of discouraging biological activity data of 1a<sup>5</sup> showing serious toxic side effects. In the past 15 years, renewed synthetic interest in **1a** and analogues<sup>6</sup> evolved due to findings that substituted derivatives of **1a** such as 9-aminocamptothecin (**1b**).<sup>7</sup> 9-((dimethylamino)methyl)-10-hydroxycamptothecin (Topotecan, 1c),<sup>8</sup> and Irinotecan  $(1d)^9$  show low overall toxicity, higher solubility, and still impressive in vivo activity against certain solid tumors. In fact, Topotecan 1c and Irinotecan 1d were recently approved by the FDA

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for the treatment of ovarian cancer and small-cell lung cancer<sup>10</sup> and refractory colorectal cancer, respectively.<sup>11</sup>



Alkaloids structurally related to camptothecin (1a) such as homocamptothecin (2),  $^{12g,13}$  mappicine (3),  $^{14,15}$  and mappicine ketone  $(4)^{15}$  are also of clinical relevance. Homocamptothecin (2) and its derivatives show similar therapeutic activities to the camptothecins (1). The E-ring lactone in **1a** easily hydrolyzes at physiological pH leading to the biologically inactive carboxylate, an inherent deficiency of the parent compound 1a.<sup>16</sup> In the E-ring lactone of **2** the tertiary alcohol and the lactone carbonyl are separated by a methylene spacer. It was found in bioassays that the longevity of activity of 2 was increased compared to that of **1a**.<sup>17</sup> Mappicine ketone (**4**) has been identified as an antiviral lead compound with selective

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FIGURE 1. Selected strategies for the synthesis of camptothecin.

activities against herpes viruses HSV-1 and HSV-2 and human cytomegalovirus.<sup>18</sup>



2: Homocamptothecin

R=OH, R'=H (Mappicine) 4: R, R'=O (Mappicine ketone)

The antitumor activity of the camptothecins is now accepted to be associated with the inhibition of DNA relaxation by specific interference of the function of DNA topoisomerase I.<sup>19,20</sup> Interestingly, it has been shown that the tetracyclic A/B/C/D ring core of 1a functions as the key binding site to DNA.<sup>21</sup>

Camptothecin and its analogues have provided a rich playing field for development of convergent total synthesis strategies. To date, the shortest asymmetric synthesis of **1a** by Comins involves the formation of the C-ring by connecting the A/B- and D/E-fragments via an N-alkylation and a key intramolecular Heck ring closure reaction (Figure 1, A).<sup>22</sup> Curran devised an imaginative strategy in which the appropriately functionalized A- and D/E-fragments (Figure 1, B) participate in a free-radical cascade leading to the formation of the B- and C-rings of 1a.12 A different concomitant formation of the B- and C-rings was reported by Fortunak using an efficient intramolecular Diels-Alder reaction (Figure 1, C) that is now used on an industrial scale.<sup>23</sup> Boger devised an approach in which a D-ring forming intermolecular Diels-Alder process precedes a C-ring cyclization (Figure 1, D).<sup>17</sup> The key features of an inventive strategy<sup>24</sup> by Bosch consisted of an intramolecular radical cyclization to form the C-ring followed by asymmetric construction of the E-ring using enolate chemistry (Figure 1, E).

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Intermolecular<sup>25</sup> (Figure 1, F) and intramolecular<sup>26</sup> (Figure 1, G) Michael addition reactions by Ciufolini and Chavan, respectively, were utilized for the D-ring construction of **1a**. Finally, in a route that has been widely traveled from the beginning of camptothecin synthetic work, a classical Friedlander reaction of 2-aminobenzal-dehyde (A-ring) with the assembled C/D/E-ring framework (Figure 1, H) was used by Henegar for B-ring annelation of **1a**.<sup>11</sup>

Herein, we report a new synthesis of the A/B/C/D-ring core  $5^{27}$  of 1a using a combined directed ortho metalation (DoM)-transition metal catalyzed cross-coupling tactic.<sup>28</sup> Thus, the construction of **5** is based on the initial coupling of **7**, prepared by anionic Fries rearrangement,<sup>29</sup> with the organozinc species **8** to afford **6**, which, by simple functional group manipulation-cyclization leads to C-ring formation (Scheme 1).

#### **Results and Discussion**

As an appropriate precursor of coupling partner 7, 2-quinolone (9) (Scheme 2), prepared in sizable quantities following Henze's procedure,<sup>30</sup> was converted into the O-carbamate 10 by treatment with sodium hydride and diethyl carbamoyl chloride in refluxing THF<sup>31</sup> followed by purification by flash chromatography on  $SiO_2$ . In an alternative purification of carbamate **10** via short path distillation, a clean thermal 1,3-carbamoyl rearrangement was observed to form urea **11**.<sup>32</sup> As reported by Queguiner,<sup>31</sup> treatment of carbamate 10 with LDA resulted in, even at -75 °C, C-3 metalation-anionic Fries rearrangement to give the 3-amidoquinolone 12. However, in contrast to the observations of Queguiner, apart from **12** (61%), the self-condensation product **13** was also isolated in 13% yield. When the reaction was carried out at -42 °C, compound 13 was not formed and the yield of the desired rearrangement product **12** increased to 71%. The increased selectivity favoring the intramolecular rearrangement at higher temperature may be rationalized by the difference in temperature dependence of a

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#### SCHEME 3



uni- vs. bimolecular reaction. To complete the synthesis of the A/B-fragment, the quinolone **12** was transformed into the triflate **7** with use of standard conditions.

As a model reaction, the cross-coupling reaction between triflate **7** and 2-bromopyridine,<sup>33</sup> representing the D-ring fragment, was undertaken (Scheme 3) with use of the Negishi protocol<sup>34</sup> with Pd(PPh<sub>3</sub>)<sub>4</sub> as catalyst and provided the biaryl **14** in satisfactory yield (57%). With this procedure, 2-bromo-6-methoxypyridine, prepared from 2,6-dibromopyridine (86% yield),<sup>35</sup> was sequentially treated with 2 equiv of *t*-BuLi at -78 °C and anhydrous ZnBr<sub>2</sub>. The resulting organozinc species **8** was subjected to the Pd<sup>0</sup>-catalyzed cross-coupling procedure with triflate **7** to afford the biaryl **6** in 59% yield.

To avoid complications of reduction of  $\pi$ -deficient heterocyclic rings, the biaryl **6** was subjected to the mild reduction protocol of Raucher.<sup>36</sup> Thus, **6** was converted with use of Lawesson's reagent into the corresponding thioamide **15**, which, upon sequential ethylation with ethyl-Meerwein salt and reduction with NaBH<sub>4</sub>, was transformed into the tertiary amine **16** in 83% yield. The final cyclization of **16** to the tetracycle **5**<sup>27</sup> was achieved in 62% yield via a modified von Braun reaction with ethyl chloroformate and potassium carbonate.<sup>37</sup> Compound **5** was found to be identical by comparison of physical and spectroscopic properties with those reported (see the Experimental Section).

Thus, the tetracyclic A/B/C/D-ring system **5** of camptothecin (**1a**) has been synthesized from quinolone **9** in 7 steps and 11% overall yield, which represents a valuable alternative to previously achieved syntheses (7 steps, 6%;<sup>27a</sup> 2 steps, 27%;<sup>27b</sup> and 5 steps, 17%<sup>27c</sup>). The simplicity of the steps, the ready availability of starting materials of both A/B- and D-ring fragments by DoM chemistry, and the potential further modification of ring D in 5, taken in sum, offer consideration of additional avenues for synthetic excursions in the camptothecin field.

#### **Experimental Section**

N.N-Diethyl O-(Quinolyl-2) Carbamate (10). To a suspension of NaH (959 mg, 60% in mineral oil, 24.0 mmol) in anhydrous THF (45 mL) was added 9<sup>30</sup> (2.88 g, 19.8 mmol) portionwise at room temperature under N<sub>2</sub> atmosphere. The greenish suspension was stirred at room temperature for 30 min and then ClCONEt<sub>2</sub> (3.75 g, 27.6 mmol) was added via syringe. The reaction mixture was refluxed for 2 h. Since TLC showed only about 50% conversion, more  $ClCONEt_2$  (3.75 g, 27.6 mmol) was added and the mixture was stirred for an additional hour. The reaction mixture was cooled in ice and carefully treated with saturated aq  $NH_4Cl$  (20 mL). The aqueous layer was separated and extracted with  $Et_2O~(2 \times 20$ mL). The combined organic phase was washed with  $H_2O$  (50 mL) and brine (50 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent gave the crude product as a brown oil, which was purified by flash chromatography on  $SiO_2$  (Et<sub>2</sub>O/hexane (1:1)) to yield 10 as a yellowish oil (4.26 g, 88%): IR (film) 3062, 2977, 2935, 1721, 1598, 1505, 1417, 1216, 1150, 1045, 978, 779, 752 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  1.23 (t, J = 7.1 Hz, 3H), 1.30 (t, J = 7.1 Hz, 3H), 3.43 (q, J = 7.1 Hz, 2H), 3.53 (q, J = 7.1 Hz, 2H), 7.24 (d, J = 8.7 Hz, 1H), 7.51 (td, J = 7.5, 1.1Hz, 1H), 7.69 (td, J = 7.7, 1.5 Hz, 1H), 7.81 (d, J = 8.1 Hz, 1H), 8.01 (d, J = 8.4 Hz, 1H), 8.19 (d, J = 8.7 Hz, 1H); MS (EI, 70 eV) m/z 244 (18, M<sup>+</sup>), 227 (17), 145 (27), 128 (30), 116 (36), 100 (100), 72 (80).

*N*-(*N*,*N*-Diethylaminocarbonyl)-2-quinolone (11). Short path distillation (173−175 °C/0.5 Torr) of 4.26 g of 10 yielded a mixture of 10 and a new product. Flash chromatography on SiO<sub>2</sub> (Et<sub>2</sub>O) gave recovered 10 (2.39 g) and a colorless crystalline material (1.81 g), which upon recrystallization from CHCl<sub>3</sub>/ Et<sub>2</sub>O (1:3) yielded pure 11 (1.07 g, 25%): mp 108.5−110 °C; IR (KBr) 2988, 2943, 2880, 1695, 1659, 1591, 1430, 829 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  1.06 (t, J = 7.2 Hz, 3H), 1.41 (t, J = 7.2 Hz, 3H), 3.04−3.31 (m, 2H), 3.52−3.86 (m, 2H), 6.4 (d, J = 9.6 Hz, 1H), 7.13 (d, J = 8.5 Hz, 1H), 7.25 (td, J = 7.6, 1.0 Hz, 1H), 7.47−7.60 (m, 2H), 7.74 (d, J = 9.6 Hz, 1H); <sup>13</sup>C NMR (50.3 MHz, CDCl<sub>3</sub>)  $\delta$  12.6, 13.7, 41.8, 43.4, 114.4, 119.9, 121.6, 123.1, 128.6, 131.0, 137.3, 140.7, 152.0, 160.2; MS (EI,

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70 eV) m/z 244 (18, M<sup>+</sup>), 227 (9), 145 (12), 100 (100), 72 (67). Anal. Calcd for  $C_{14}H_{16}N_2O_2$ : C, 68.83; H, 6.60; N, 11.47. Found: C, 68.75; H, 6.70; N, 11.48.

N.N-Diethvl 1.2-Dihydro-2-oxo-3-quinolinecarboxamide (12). To a stirred LDA solution (diisopropylamine (592 mg, 5.9 mmol) and n-BuLi (3.9 mL (1.6 M in hexane), 6.2 mmol)) in THF (30 mL), a solution of 10 (1.10 g, 4.5 mmol) in THF (8 mL) (precooled to -78 °C) was added via cannula at -78 °C under N<sub>2</sub> atmosphere. After the solution was stirred for 1 h, anhydrous MeOH (562 mg, 17.5 mmol) was added and the reaction mixture was stirred for an additional hour at -78°C. After quenching with saturated aq NH<sub>4</sub>Cl (15 mL), the aqueous phase was extracted with  $CHCl_3$  (3  $\times$  30 mL). The combined organic phase was washed with H<sub>2</sub>O and dried (Na<sub>2</sub>- $SO_4$ ). Removal of the solvent gave the crude product as yellow crystals (1.11 g). Flash chromatography on SiO<sub>2</sub> (EtOAc/MeOH (95:5)) yielded 12 as pale yellow crystals (666 mg, 61%): mp 166-167 °C (toluene) (lit.<sup>31</sup> mp 167 °C); IR (KBr) 3063, 2968, 2897, 1662, 1617, 1569, 1433, 1221, 946, 749 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  1.17 (t, J=6.6 Hz, 3H), 1.32 (t, J=6.6Hz, 3H), 3.34 (q, J = 6.8 Hz, 2H), 3.63 (q, J = 6.7 Hz, 2H), 7.24 (td, J = 7.5, 1.3 Hz, 1H), 7.41 (d, J = 7.9 Hz, 1H), 7.49-7.60 (m, 2H), 7.87 (s, 1H), 12.36 (s br, 1H, exchangeable with  $D_2O).$ 

A second fraction was obtained and recrystallized from EtOAc/MeOH (2:1) to yield **13** as colorless crystals (122 mg, 13%): mp >180 °C dec; IR (KBr) 3063, 2974, 2938, 2892, 2853, 1718, 1658, 1619, 1591, 1564, 1399, 1275, 1196, 1080, 754 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>/DMSO-d<sub>6</sub> (2:1))  $\delta$  0.87 (t, J = 7.0 Hz, 3H), 0.94 (t, J = 7.0 Hz, 3H), 2.98–3.06 (m, 4H), 7.22 (t, J = 7.7 Hz, 1H), 7.39 (d, J = 8.2 Hz, 1H), 7.54–7.67 (m, 2H), 7.77–7.87 (m, 2H), 7.96 (d, J = 8.2 Hz, 1H), 8.12 (d, J = 7.9 Hz, 1H), 8.35 (s, 1H), 8.70 (s, 1H), 12.10 (s, 1H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>/DMSO-d<sub>6</sub> (2:1))  $\delta$  11.1, 11.8, 39.3, 39.7, 113.7, 116.7, 120.7, 124.8, 125.1, 125.9, 126.7, 127.2, 127.8, 129.0, 130.0, 130.9, 138.7, 138.8, 141.2, 145.0, 150.7, 151.8, 158.5, 188.7.

N,N-Diethyl 2-Trifluoromethanesulfonyloxy-3-quino**linecarboxamide (7).** To a solution of **12** (441 mg, 2.0 mmol) in CH<sub>2</sub>Cl<sub>2</sub> (18 mL) at 0 °C was added NEt<sub>3</sub> (365 mg, 3.6 mmol) and Tf<sub>2</sub>O (560 mg, 2.0 mmol) dropwise under N<sub>2</sub> atmosphere. The solution was stirred at 0 °C for 1 h and quenched with saturated aq NaHCO<sub>3</sub> (20 mL). The aqueous phase was extracted with  $CH_2Cl_2$  (2  $\times$  10 mL) and the combined organic phase was washed H<sub>2</sub>O and dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent gave the crude product as yellow crystals (887 mg). Flash chromatography on  $SiO_2$  (Et<sub>2</sub>O) afforded 7 as pale yellow crystals (478 mg, 70%): mp 89.5-91 °C (CHCl<sub>3</sub>/hexane (4:1)); IR (film) 3059, 2981, 2939, 1635, 1573, 1424, 1217, 1140, 1052, 918, 884, 847, 801, 759 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$ 1.12 (t, J = 7.1 Hz, 3H), 1.31 (t, J = 7.1 Hz, 3H), 3.25 (q, J = 7.1 Hz), 3.257.1 Hz, 2H), 3.61 (s br, 2H), 7.67 (td, J = 7.5, 1.2 Hz, 1H), 7.84 (td, J = 8.4, 1.5 Hz, 1H), 7.90 (d, J = 8.4 Hz, 1H), 8.06 (d, JJ=8.5 Hz, 1H), 8.28 (s, 1H);  $^{13}\mathrm{C}$  NMR (62.9 MHz, CDCl\_3)  $\delta$ 12.4, 13.9, 39.5, 43.1, 118.5 (q, 321), 122.6, 127.1, 127.8, 128.3, 128.6, 131.7, 139.0, 145.4, 149.6, 163.8; MS (EI, 70 eV) m/z 376 (36, M<sup>+</sup>), 304 (100), 227 (49), 172 (82), 143 (66); HRMS calcd for C<sub>15</sub>H<sub>15</sub>F<sub>3</sub>N<sub>2</sub>O<sub>4</sub>S 376.0705, found 376.0693.

*N,N*-Diethyl 2-(2-Pyridyl)-3-quinolinecarboxamide (14). A solution of 2-bromopyridine (117 mg, 741  $\mu$ mol) in THF (3 mL) was treated at -78 °C with *t*-BuLi (1.7 M in pentane, 867  $\mu$ L, 1.5 mmol) under N<sub>2</sub> atmosphere. After the dark red solution was stirred for 15 min at -78 °C, a solution of anhydrous ZnBr<sub>2</sub> (183 mg, 813  $\mu$ mol) in THF (2 mL) was added via cannula. The resulting orange solution was stirred at -78 °C for 1 h and then allowed to warm to room temperature. A solution of 7 (185 mg, 492  $\mu$ mol) and Pd(PPh<sub>3</sub>)<sub>4</sub> (28 mg, 24  $\mu$ mol) in THF (3 mL), which had been stirred for 15 min at room temperature, was added via cannula. The reaction mixture was refluxed (60 h) and quenched with saturated aq NH<sub>4</sub>Cl (2 mL). The aqueous phase was extracted with CH<sub>2</sub>Cl<sub>2</sub> (1 × 20 mL, 2 × 10 mL) and the combined organic phase was

washed with H<sub>2</sub>O and dried (Na<sub>2</sub>SO<sub>4</sub>). Evaporation to dryness in vacuo gave the crude product as a brownish oil (208 mg), which was purified by flash chromatography on SiO<sub>2</sub> (CHCl<sub>3</sub>) to yield **14** as a yellowish oil (85 mg, 57%): IR (CDCl<sub>3</sub>) 3064, 2981, 2837, 2875, 1723, 1618, 1559, 1480, 1279, 1097 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  1.00 (t, J = 7.1 Hz, 3H), 1.30 (t, J = 7.1 Hz, 3H), 3.21 (s br, 2H), 3.93 (s br, 2H), 7.28–7.34 (m, 1H), 7.58 (t, J = 7.2 Hz, 1H), 7.72–7.87 (m, 3H), 8.14 (s, 1H), 8.19 (d, J = 8.5 Hz, 1H), 8.42 (d, J = 7.9 Hz, 1H), 8.59 (s br, 1H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>)  $\delta$  11.7, 13.3, 38.7, 42.7, 123.1, 123.8, 127.0, 127.4, 127.5, 129.6, 130.1, 130.7, 134.8, 136.6, 147.3, 148.2, 153.4, 156.2, 170.3; MS (EI, 70 eV) *m/z* (305 (<1, M<sup>+</sup>), 233 (100), 205 (19); HRMS calcd for C<sub>19</sub>H<sub>19</sub>N<sub>3</sub>O 305.1528, found 305.1500.

2-Bromo-6-methoxypyridine. A solution of NaOMe in MeOH (Na (1.33 g, 58 mmol) in anhydrous MeOH (14 mL)) was added to a suspension of 2,6-dibromopyridine (8.0 g, 34 mmol) in anhydrous MeOH (22 mL) and the resulting mixture was refluxed for 25 h. The reaction mixture was allowed to cool to room temperature, cold 5% aq NaHCO<sub>3</sub> (25 mL) was added, and the mixture was extracted with  $Et_2O$  (1  $\times$  30 mL,  $2\,\times\,20$  mL). The combined organic phase was concentrated, the resulting residue was taken up in  $Et_2O$  (30 mL), and the Et<sub>2</sub>O solution was washed with brine (20 mL) and dried (K<sub>2</sub>- $CO_3$ ). Removal of the solvent gave a yellowish liquid (5.72 g). Kugelrohr distillation yielded 2-bromo-6-methoxypyridine as a colorless liquid (5.48 g, 86%): bp 87-91 °C/15 Torr (lit.<sup>35</sup> bp 85-95°/15 Torr); IR (film) 2985, 2952, 2850, 1594, 1581, 1556, 1467, 1408, 1296, 1020, 854 cm  $^{-1}$ ; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  3.93 (s, 3H), 6.68 (d, J = 7.7 Hz, 1H), 7.05 (d, J = 7.4 Hz, 1H), 7.40 (t, J = 7.8 Hz, 1H).

N,N-Diethyl 2-(2-Methoxypyridin-6-yl)-3-quinolinecarboxamide (6). To a solution of 2-bromo-6-methoxypyridine (301 mg, 1.6 mmol) in THF (6 mL) at -78 °C was added dropwise t-BuLi (1.7 M in pentane, 1.9 mL, 3.2 mmol) under N2 atmosphere. The resulting pale yellow solution was stirred for 15 min at -78 °C. A solution of anhydrous ZnBr<sub>2</sub> (409 mg, 1.8 mmol) in THF (4 mL) was added via cannula and the mixture was stirred for 70 min at -78 °C. The reaction mixture was allowed to warm to room temperature and was added via cannula to a solution of 7 (399 mg, 1.1 mmol) and  $Pd(PPh_{3})_{4}$ (62 mg, 54  $\mu mol)$  in THF (4 mL), which had been stirred for 20 min at room temperature. The resulting solution was refluxed for 45 h and the solvent was removed in vacuo. The residue was dissolved in CHCl<sub>3</sub>/MeOH (9:1) (15 mL) and washed with  $Na_2EDTA \cdot 2H_2O$  (1.35 g in 15 mL of  $H_2O$ ). The aqueous phase was extracted with CHCl<sub>3</sub>/MeOH (9:1) (5  $\times$  12 mL) and the combined organic phase was dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent gave a yellow gum (569 mg) that was purified by flash chromatography on  $SiO_2$  (EtOAc/hexane (2: 1)) to give 6 as a pale yellow oil (211 mg, 59%): IR (film) 3060, 2979, 2934, 2874, 1634, 1574, 1556, 1267, 1047, 921, 888, 792 cm<sup>-1</sup>; <sup>1</sup>H NMR (200 MHz, CDCl<sub>3</sub>)  $\delta$  0.84 (t, J = 7.1 Hz, 3H), 1.22 (t, J = 7.1 Hz, 3H), 2.90-2.99 (m, 1H), 3.16-3.36 (m, 2H),3.79-3.87 (m, 1H), 3.97 (s, 3H), 6.79 (d, J = 8.2 Hz, 1H), 7.56(t, J = 7.9 Hz, 1H), 7.69–7.77 (m, 2H), 7.84 (d, J = 8.1 Hz, 1H), 7.97 (d, J = 7.4 Hz, 1H), 8.17 (s, 1H), 8.18 (d, J = 5.8 Hz, 1H);  $^{13}{\rm C}$  NMR (50 MHz, CDCl\_3)  $\delta$  12.8, 13.4, 38.9, 43.2, 53.7,  $111.3,\ 116.3,\ 126.9,\ 127.4,\ 127.5,\ 129.5,\ 130.1,\ 130.3,\ 135.9,$ 139.2, 147.3, 153.4, 154.0, 163.4, 169.8; MS (EI, 70 eV) m/z 335 (7, M<sup>+</sup>), 263 (100), 235 (23); HRMS calcd for C<sub>20</sub>H<sub>21</sub>N<sub>3</sub>O<sub>2</sub> 335.1634, found 335.1602.

*N,N*-Diethyl 2-(2-Methoxypyridin-6-yl)-3-quinolinethiocarboxamide (15). A solution of 6 (120 mg, 358  $\mu$ mol) and Lawesson's reagent (241 mg, 596  $\mu$ mol) in anhydrous benzene (3 mL) was refluxed for 8 h under N<sub>2</sub> atmosphere. The reaction mixture was passed through a cotton plug and the filtrate was concentrated. The resulting orange oil was purified by flash chromatography on SiO<sub>2</sub> (CH<sub>2</sub>Cl<sub>2</sub>/Et<sub>2</sub>O (9:1)) to yield **15** as a yellow viscous liquid (105 mg, 83%): IR (film) 3064, 2982, 2941, 2876, 2221, 1592, 1575, 1498, 1266, 816 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>)  $\delta$  1.03 (t, J = 7.2 Hz, 3H), 1.27 (t, J = 7.1 Hz, 3H), 3.25 (dq, J = 14.3, 7.1 Hz, 1H), 3.66 (dq, J = 14.3, 7.2 Hz, 1H), 3.95 (s, 3H), 4.03–4.19 (m, 2H), 6.78 (dd, J = 8.0, 0.6 Hz, 1H), 7.56 (td, J = 8.0, 1.0 Hz, 1H), 7.68–7.76 (m, 2H), 7.82 (dd, J = 8.1, 0.7 Hz, 2H), 8.09 (s, 1H), 8.17 (d, J = 8.5 Hz, 1H); <sup>13</sup>C NMR (50 MHz, CDCl<sub>3</sub>)  $\delta$  11.1, 13.3, 45.9, 48.7, 53.9, 110.9, 117.3, 126.8, 127.4, 127.5, 129.2, 130.1, 134.3, 135.8, 139.1, 146.7, 152.2, 154.2, 163.2, 197.6; MS (EI, 70 eV) m/z 351 (25, M<sup>+</sup>), 280 (47), 265 (100), 243 (12).

3-N,N-Diethylaminomethyl-2-(2-methoxypyridin-6-yl)quinoline (16). To a solution of 15 (105 mg, 299  $\mu$ mol) in anhydrous  $CH_2Cl_2$  (1.5 mL) at 0 °C was added  $Et_3O^+BF_4^-$  (0.5 M in  $CH_2Cl_2$ , 0.62 mL, 310  $\mu$ mol) under  $N_2$  atmosphere. The resulting mixture was stirred for 5 min at 0 °C and then for 90 min at room temperature. The solvent was removed in vacuo and the solid residue was dissolved in anhydrous MeOH (2.5 mL). NaBH<sub>4</sub> (29 mg, 767  $\mu$ mol) was added at 0 °C and the mixture was stirred for 5 min at 0 °C and then for 3 h at room temperature. HCl (5%, 2 mL) was added and the mixture was stirred for 5 min. After addition of aq NaOH solution to pH >10, the reaction mixture was extracted with  $Et_2O$  (1 × 20 mL,  $2 \times 15$  mL). The combined organic phase was washed with H<sub>2</sub>O (20 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent gave a yellow oil (101 mg) that was purified by flash chromatography on  $SiO_2$  (EtOAc) to yield **16** as a yellow gum (80 mg, 83%): IR (CDCl<sub>3</sub>) 3064, 2971, 2935, 2874, 2809, 1722, 1576, 1464, 1411, 1261 cm<sup>-1</sup>; <sup>1</sup>H NMR (250 MHz, CDCl<sub>3</sub>) δ 0.98 (t,  $J=7.1~{\rm Hz},\,6{\rm H}),\,2.54~({\rm q},\,J=7.1~{\rm Hz},\,4{\rm H}),\,3.96~({\rm s},\,3{\rm H}),\,4.09~({\rm s},\,3{\rm H}),\,3.96~({\rm s},\,3{\rm H}),\,4.09~({\rm s},\,3{\rm H}),\,3.96~({\rm s},\,3{\rm H}),\,4.09~({\rm s},\,3{\rm H}),\,3.96~({\rm s},\,3{\rm H}),\,3.96~$ 2H), 6.80 (dd, J = 8.4, 0.6 Hz, 1H), 7.49-7.55 (m, 2H), 7.64-7.77 (m, 2H), 7.86 (d, J = 8.1 Hz, 1H), 8.14 (d, J = 8.5 Hz, 1H), 8.53 (s, 1H); <sup>13</sup>C NMR (63 MHz, CDCl<sub>3</sub>) & 11.8, 47.2, 53.4, 54.9, 110.2, 117.3, 126.6, 127.4, 127.9, 129.0, 129.2, 132.3, 136.7, 139.3, 156.8, 157.2, 162.7; MS (EI, 70 eV) m/z 321 (35, M<sup>+</sup>), 292 (100), 249 (86), 235 (50), 205 (40), 185 (53); HRMS calcd for C<sub>20</sub>H<sub>23</sub>N<sub>3</sub>O 321.1841, found 321.1820.

**11***H***-Indolizino[1,2-***b***]quinolin-9-one (5). To a suspension of anhydrous K\_2CO\_3 (23 mg, 166 \mumol) and ClCOOEt (25 \muL,** 

261 µmol) in anhydrous THF (1 mL) was added a solution of 16 (67 mg, 208  $\mu$ mol) in anhydrous THF (1 mL) via cannula under N<sub>2</sub> atmosphere. The mixture was stirred at room temperature for 2.5 h. TLC showed a new spot with intense blue fluorescence under UV light. The reaction mixture was heated at reflux for 3 h. To the resulting orange suspension was added ClCOOEt (10  $\mu$ L, 105  $\mu$ mol) and the mixture was heated at reflux for 1 h. After the solution was cooled to room temperature,  $H_2O$  (10 mL) was added and the whole was extracted with  $CH_2Cl_2$  (3  $\times$  10 mL). The combined organic phase was washed with brine (10 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). Removal of the solvent gave a yellow solid (48 mg) that was purified by flash chromatography on  $SiO_2$  (EtOAc/NEt<sub>3</sub> (99: 1)) to yield  ${\bf 5}$  as pale yellow crystals (30 mg, 62%) (lit.  $^{27a}$  mp 265 °C; lit.<sup>27c</sup> mp 253-254 °C): IR (CDCl<sub>3</sub>) 3056, 1663, 1594, 1159 cm $^{-1};$   $^1\!H$  NMR (250 MHz, CDCl\_3)  $\delta$  5.25 (s, 2H), 6.73 (dd, J = 9.0, 0.7 Hz, 1H), 7.30 (dd, J = 6.1, 0.8 Hz, 1H), 7.60–7.70 (m, 2H), 7.80 (ddd, J = 8.4, 7.0, 1.4 Hz, 1H), 7.89 (d, J = 8.1Hz, 1H), 8.20 (d, J = 8.5 Hz, 1H), 8.34 (s, 1H). The spectroscopic properties are identical with those reported for 5 prepared previously.27

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**Supporting Information Available:** Spectroscopic information and reagent purification and availability. This material is available free of charge via the Internet at http://pubs.acs.org.

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