

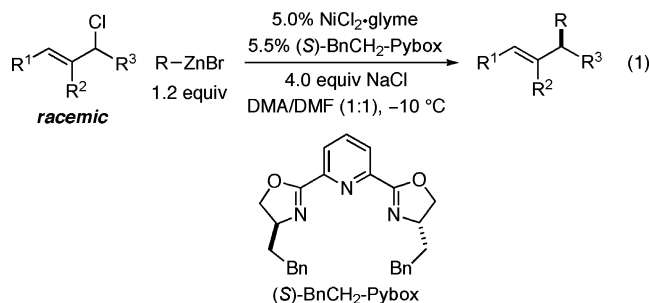
Nickel-Catalyzed Asymmetric Negishi Cross-Couplings of Secondary Allylic Chlorides with Alkylzincs

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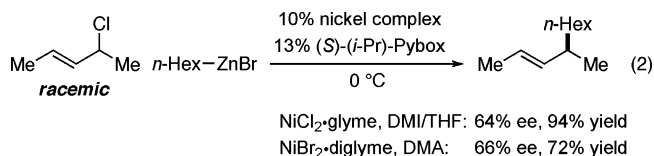
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Metal-catalyzed enantioselective couplings of allylic electrophiles with carbon nucleophiles have been intensively studied,¹ with most of the investigations focused on palladium-catalyzed reactions of allylic esters/carbonates with enolates, copper-catalyzed couplings of primary allylic electrophiles with Grignard and diorganozinc reagents (S_N2' substitution),² and nickel-catalyzed reactions of certain allylic electrophiles with Grignard reagents.^{3,4} Although powerful methods have been developed, there remains room for improvement, for example, processes that accommodate a broader range of nucleophiles and that display greater functional-group compatibility. In this report, we describe a versatile nickel-based catalyst for asymmetric couplings of racemic secondary allylic chlorides with readily available alkylzinc halides⁵ (eq 1; DMA = *N,N*-dimethylacetamide), and we apply this method to a formal total synthesis of fluvirucine A₁.



Previously, we have reported nickel-catalyzed enantioselective Negishi reactions of α -bromo amides and benzylic bromides with organozinc reagents.⁶ Although the regioselectivity of the carbon-carbon bond-forming process was not a concern for these families of substrates, we anticipated that regioselectivity *would* be an issue for couplings of allylic electrophiles. To avoid this complication during our initial studies, we chose to examine the reaction of a "symmetrical" allylic halide. Under the conditions that we had developed for enantioselective Negishi couplings of α -bromo amides and benzylic bromides, we obtained promising results for an allylic electrophile (eq 2; DMI = 1,3-dimethyl-2-imidazolidinone).



Through optimization studies, we were able to significantly improve the enantioselectivity of this Negishi cross-coupling reaction (87% ee, 95% yield; Table 1, entry 1).⁷⁻⁹ The combination of a high ee and a high yield establishes that the process is stereoconvergent: the two enantiomers of the racemic substrate are transformed into the same enantiomer of the product with good stereoselectivity.

Table 1. Enantioselective Negishi Cross-Couplings of "Symmetrical" Allylic Chlorides with Alkylzinc Reagents (for the Reaction Conditions, See eq 1)

entry	allylic chloride	R-ZnBr	ee (%)	yield (%) ^a
1		<i>n</i> -Hex-ZnBr	87	95 ^b
2			90	93
3 ^c	<i>n</i> -Pr		85	81
4	<i>n</i> -Pr		79	81
5	<i>i</i> -Pr	TBSO-CH ₂ -CH ₂ -CH ₂ -ZnBr	69	57
6			98	54

All data are the average of two experiments. ^a Isolated yield. ^b The product is volatile. The yield was determined by GC versus an internal standard. ^c Solvent: DMA/DMF (9:1).

Table 2. Enantioselective Negishi Cross-Couplings of Unsymmetrical Allylic Chlorides with Alkylzinc Reagents (for the Reaction Conditions, See eq 1)

entry	allylic chloride	R-ZnBr	ee (%)	yield (%) ^a
1 ^{b,c}		Ph-CH ₂ -CH ₂ -CH ₂ -ZnBr	83	97
2 ^c	<i>i</i> -Pr		84	95
3 ^c	<i>t</i> -Bu	MeO ₂ C-CH ₂ -CH ₂ -CH ₂ -ZnBr	81	85
4	CO ₂ Et		96	86
5	CONEt ₂	Et-ZnBr	91	57
6	CON(OMe)Me	TBSO-CH ₂ -CH ₂ -CH ₂ -ZnBr	93	91
7	PO(OEt) ₂	<i>n</i> -Hex-ZnBr	90	63

All data are the average of two experiments. Regioselectivity, >20:1, except for entry 1. ^a Isolated yield. ^b Regioselectivity, 1.9:1; ee of the minor regioisomer, 88%. ^c The allylic chloride is a mixture of regioisomers.

As the steric demand of the R¹ substituent increases, the enantioselectivity of the cross-coupling decreases (Table 1, entries 1-5). Thus, good ee's are generally obtained if the group is unbranched (entries 1-4), but an erosion in stereoselection is observed for a hindered diisopropyl-substituted allylic chloride (entry 5). The Ni/Pybox catalyst can achieve an asymmetric Negishi

