# **One-Pot Synthesis of 3-Arylquinazoline-2,4**(1*H*,3*H*)-dithiones by the Reaction of 2-Lithiophenyl Isothiocyanates with Aryl Isothiocyanates

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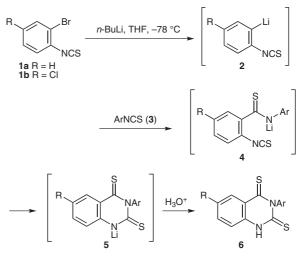
**Abstract:** 3-Arylquinazoline-2,4(1*H*,3*H*)-dithiones were synthesized in satisfactory yields from 2-bromophenyl isothiocyanates in one pot via generation of the corresponding 2-lithiophenyl isothiocyanates by bromine–lithium exchange with butyllithium followed by treatment with aryl isothiocyanates.

Key words: quinazoline-2,4(1H,3H)-dithiones, 2-lithiophenyl isothiocyanates, 2-bromophenyl isothiocyanates, aryl isothiocyanates, benzothioamides

Although the first synthesis of quinazoline-2,4(1H,3H)dithione by the reaction of 2-aminobenzonitrile with carbon disulfide in refluxing pyridine was described by Taylor and co-workers in 1966,<sup>1</sup> there has been but limited attention paid to compounds having this skeleton until the recent reports on the biological evaluation of 3arylquinazoline-2,4(1H,3H)-dithiones and related compounds as antimycobacterial agents,<sup>2a</sup> phosphodiesterase 7 inhibitors,<sup>2b</sup> and so on.<sup>3</sup> The synthesis of these compounds was generally based on the treatment of the corresponding 2-thioxo-2,3-dihydroquinazolin-4(1H)-ones with phosphorous pentasulfide<sup>2a</sup> or Lawesson's reagent.<sup>2b</sup> On the other hand, we have demonstrated, in a recent publication, that 2-lithiophenyl isothiocyanates can be generated by the treatment of 2-bromophenyl isothiocyanates with butyllithium and can serve as intermediates for the 1,4-dihydro-3,1-benzoxazine-2-thiones synthesis of through reaction with aldehydes, ketones, or butanolide.<sup>4</sup> As part of our efforts to explore the synthetic utility and potential of these lithium products, we have investigated the possibility of their use in the preparation of quinazoline-2,4(1H,3H)-dithiones by reacting them with isothiocyanates. In this paper, the results of our studies, which offer a new and efficient route to 3-arylquinazoline-2,4(1H,3H)-dithiones by the reaction of these lithium products with aryl isothiocyanates, are reported.

The synthesis of 3-arylquinazoline-2,4(1*H*,3*H*)-dithiones **6** from 2-bromophenyl isothiocyanates **1** was conducted as illustrated in Scheme 1. Treatment of **1** with an equimolar amount of butyllithium at -78 °C in THF generated 2-lithiophenyl isothiocyanates **2**, which were allowed to react with aryl isothiocyanates **3** to give the lithium thioamide intermediates **4**. Cyclization of these in-

SYNTHESIS 2011, No. 10, pp 1526–1528 Advanced online publication: 19.04.2011 DOI: 10.1055/s-0030-1260016; Art ID: F19411SS © Georg Thieme Verlag Stuttgart · New York termediates gave 3-aryl-1-lithioquinazoline-2,4(1*H*,3*H*)dithione intermediates **5**, which were protonated by aqueous workup to give the desired products **6**. Yields of these products after purification by recrystallization are summarized in Table 1, which are generally fair-to-good. The structure determination of **6a** was performed by comparison of its <sup>1</sup>H and <sup>13</sup>C NMR spectra to those reported previously.<sup>2b</sup> Each of the products **6** was isolated by recrystallization and each <sup>13</sup>C NMR spectrum uniformly shows signals at ca.  $\delta = 172$  and 188 ppm assignable to 2and 4-thioxo carbons, respectively.



Scheme 1 Preparation of 3-arylquinazoline-2,4(1*H*,3*H*)-dithiones 6

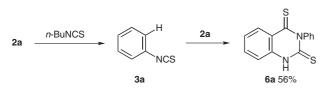
Our attempt to obtain 3-butylquinazoline-2,4(1*H*,3*H*)dithione by the reaction of 2-lithiophenyl isothiocyanate (**2a**) with *n*-butyl isothiocyanate was unsuccessful. Instead of the expected product, 3-phenylquinazoline-2,4(1*H*,3*H*)-dithione (**6a**) was obtained in 56% yield based on **1a**. The formation of **6a** may be interpreted as follows. Part of lithiated compound **2a** is protonated probably by proton abstraction from the  $\alpha$ -position of *n*-butyl isothiocyanate to give phenyl isothiocyanate (**3a**), which reacts with remaining **2a** to provide **6a**, as shown in Scheme 2.

3-Aryl-2-thioxo-2,3-dihydroquinazolin-4(1H)-ones could be expected to be formed by the application of the present reaction using aryl isocyanates in place of aryl isothiocyanates. But, the reaction of 2-lithiophenyl isothiocyanate (**2a**) with phenyl isocyanate only resulted in the formation

 Table 1
 Preparation of 3-Arylquinazoline-2,4(1H,3H)-dithiones 6

Entry	1	Ar in <b>3</b>	6	Yield (%) <sup>a</sup>
1	1a	Ph ( <b>3a</b> )	6a	74
2	1a	$3-MeC_{6}H_{4}(\mathbf{3b})$	6b	71
3	1a	$3,5-Me_2C_6H_3(3c)$	6c	72
4	1a	$2\text{-ClC}_{6}\text{H}_{4}(\mathbf{3d})$	6d	62
5	1a	$3-ClC_{6}H_{4}(3e)$	6e	80
6	1a	$2-BrC_{6}H_{4}(\mathbf{3f})$	6f	58
7	1a	$3\text{-MeOC}_{6}\text{H}_{4}(\mathbf{3g})$	6g	70
8	1a	naphthalen-2-yl (3h)	6h	64
9	1b	Ph ( <b>3a</b> )	6i	63
10	1b	2-Br,4-ClC <sub>6</sub> H <sub>3</sub> ( <b>3j = 1b</b> )	6j	61

<sup>a</sup> Isolated yields.



Scheme 2 Attempted synthesis of 3-butylquinazoline-2,4(1*H*,3*H*)dithione

of intractable mixture of products, from which no more than a trace amount of the desired product could be obtained in a pure form. This result is most likely due to susceptibility of isocyanates to oligomerization under the reaction conditions.

In conclusion, we have demonstrated that the reaction of 2-lithiophenyl isothiocyanates with aryl isothiocyanates provides rapid access to 3-arylquinazoline-2,4(1H,3H)-dithiones. The present new method may find some value in synthesis, because it is operationally simple and the starting materials are readily available.

All melting points were obtained on a Laboratory Devices MEL-TEMP II melting apparatus and are uncorrected. IR spectra were determined as KBr disks with a Shimadzu FTIR-8300 spectrophotometer. The <sup>1</sup>H NMR spectra were determined using TMS as an internal reference with a JEOL ECP500 FT NMR spectrometer operating at 500 MHz or a JEOL LA400 FT NMR spectrometer operating at 400 MHz. The <sup>13</sup>C NMR spectra were determined using TMS as an internal reference with a JEOL ECP500 FT NMR spectrometer operating at 125 MHz or a JEOL LA400 FT NMR spectrometer operating at 100 MHz. Low-resolution mass spectra (EI, 70 eV) were measured by a JEOL JMS AX505 HA spectrometer. TLC was carried out on a Merck Kieselgel 60 PF254. Column chromatography was performed using Wako Gel C-200E. All of the organic solvents used in this study were dried over appropriate drying agents and distilled prior to use. All chemicals, excluding 2-bromo-4-chlorophenyl isothiocyanate (1b) and 3,5-dimethylphenyl isothiocyanate (3c), were commercially available. Compound 1b was prepared according to the procedure reported previously.<sup>4</sup>

Compound  $3c^5$  was prepared from 3,5-dimethylphenyl isocyanide<sup>6</sup> according to the procedure employed for the preparation of 1b.<sup>4</sup>

# 3-Phenylquinazoline-2,4(1*H*,3*H*)-dithione (6a);<sup>2</sup> Typical Procedure

To a stirred solution of **1a** (0.21 g, 1.0 mmol) in THF (4 mL) at -78 °C was added *n*-BuLi (1.6 M in hexane, 0.03 mL, 1.0 mmol) dropwise. After 1 min, **3a** (0.14 g, 1.0 mmol) was added and the stirring was continued for an additional 10 min at the same temperature before sat. aq NH<sub>4</sub>Cl (10 mL) was added. The mixture was warmed to r.t. and extracted with EtOAc (3 × 10 mL). The combined extracts were washed with brine (10 mL) and dried (Na<sub>2</sub>SO<sub>4</sub>). Evaporation of the solvent gave a residual solid, which was recrystallized from hexane–CHCl<sub>3</sub> to give **6a** as an orange solid; yield: 0.20 g (74%); mp 253–255 °C (Lit.<sup>2b</sup> mp 253–256 °C).

<sup>1</sup>H and <sup>13</sup>C NMR spectral data for this product were identical with those reported previously.<sup>2b</sup>

### **3-(3-Methylphenyl)quinazoline-2,4(1***H***,3***H***)-dithione (6b) Yellow solid; mp 176–178 °C (hexane–THF).**

IR (KBr): 3158, 1616, 1604, 1371 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta = 2.31$  (s, 3 H), 6.98–7.01 (m, 2 H), 7.17 (d, J = 7.3 Hz, 1 H), 7.32–7.37 (m, 2 H), 7.42 (d, J = 7.8 Hz, 1 H), 7.78 (ddd, J = 8.3, 7.8, 1.5 Hz, 1 H), 8.32 (d, J = 8.3 Hz, 1 H), 13.44 (br s, 1 H).

<sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>): δ = 20.9, 116.0, 123.7, 125.3, 125.6, 128.67, 128.69, 129.1, 131.8, 135.3, 135.8, 138.6, 144.1, 172.8, 189.7.

MS: m/z (%) = 284 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{15}H_{12}N_2S_2$ : C, 63.35; H, 4.25; N, 9.85. Found: C, 63.18; H, 4.30; N, 9.69.

### **3-(3,5-Dimethylphenyl)quinazoline-2,4(1***H***,3***H***)-dithione (6c) Pale-yellow solid; mp 188–189 °C (hexane–THF).**

IR (KBr): 3160, 1605, 1368 cm<sup>-1</sup>.

<sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ ):  $\delta = 2.71$  (s, 6 H), 6.51 (s, 2 H), 6.83 (s, 1 H), 7.35 (ddd, J = 8.4, 6.9, 1.5 Hz, 1 H), 7.46 (d, J = 8.4 Hz, 1 H), 7.66 (ddd, J = 8.4, 6.9, 1.5 Hz, 1 H), 8.14 (dd, J = 8.4, 1.5 Hz, 1 H), 13.32 (br s, 1 H).

<sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>): δ = 21.0, 116.7, 118.1, 119.1, 125.5, 125.8, 126.6, 134.0, 138.8, 139.0, 148.7, 172.5, 185.8.

MS: m/z (%) = 298 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{16}H_{14}N_2S_2{:}$  C, 64.39; H, 4.73; N, 9.39. Found: C, 64.29; H, 4.76; N, 9.22.

**3-(2-Chlorophenyl)quinazoline-2,4(1***H***,3***H***)-dithione (6d) Orange solid; mp 239–241 °C (hexane–THF).** 

IR (KBr): 3161, 1614, 1398 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ): δ = 7.33–7.49 (m, 5 H), 7.57 (dd, J = 8.8, 1.5 Hz, 1 H), 7.82 (ddd, J = 8.3, 7.3, 1.5 Hz, 1 H), 8.32 (dd, J = 8.3, 0.9 Hz, 1 H), 13.58 (br s, 1 H).

<sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ):  $\delta = 116.2$ , 123.2, 125.6, 128.4, 129.9, 130.0, 130.8, 130.9, 131.6, 135.3, 136.3, 141.0, 171.9, 188.7.

MS: m/z (%) = 304 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{14}H_9ClN_2S_2$ : C, 55.16; H, 2.98; N, 9.19. Found: C, 55.35; H, 2.76; N, 9.05.

**3-(3-Chlorophenyl)quinazoline-2,4(1***H***,3***H***)-dithione (6e) Orange solid; mp 256–258 °C (hexane–THF).** 

IR (KBr): 3165, 1618, 1603, 1400 cm<sup>-1</sup>.

<sup>1</sup>H NMR (500 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 7.22 (dd, *J* = 8.4, 1.5 Hz, 1 H), 7.35 (ddd, *J* = 8.4, 6.9, 1.5 Hz, 1 H), 7.40–7.45 (m, 3 H), 7.50 (t, *J* = 7.6 Hz, 1 H), 7.79 (ddd, *J* = 8.4, 6.9, 1.5 Hz, 1 H), 8.32 (dd, *J* = 8.4, 1.5 Hz, 1 H), 13.50 (br s, 1 H).

<sup>13</sup>C NMR (125 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 116.0, 123.7, 125.3, 127.7, 128.1, 128.8, 130.8, 131.6, 133.2, 135.3, 135.9, 145.3, 172.6, 189.8. MS: *m/z* (%) = 304 ([M<sup>+</sup>], 100).

Anal. Calcd for C<sub>14</sub>H<sub>9</sub>ClN<sub>2</sub>S<sub>2</sub>: C, 55.16; H, 2.98; N, 9.19. Found: C, 55.18; H, 3.02; N, 9.25.

# 3-(2-Bromophenyl)quinazoline-2,4(1H,3H)-dithione (6f)

Orange solid; mp 250-254 °C (hexane-THF).

IR (KBr): 3161, 1614, 1395 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta$  = 7.31–7.38 (m, 2 H), 7.43–7.45 (m, 2 H), 7.49 (t, *J* = 7.8 Hz, 1 H), 7.71 (dd, *J* = 7.8, 7.3 Hz, 1 H), 7.81 (dd, *J* = 7.8, 7.3 Hz, 1 H), 8.32 (d, *J* = 7.8 Hz, 1 H), 13.57 (br s, 1 H).

<sup>13</sup>C NMR (100 MHz, DMSO-*d*<sub>6</sub>):  $\delta$  = 116.1, 121.8, 123.3 125.5, 128.9, 130.0, 130.9, 131.6, 133.1, 135.3, 136.2, 142.5, 171.9, 188.5.

MS: m/z (%) = 348 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{14}H_9BrN_2S_2{:}$  C, 48.14; H, 2.60; N, 8.02. Found: C, 48.02; H, 2.71; N, 8.04.

## 3-(3-Methoxyphenyl)quinazoline-2,4(1H,3H)-dithione (6g)

Orange solid; mp 269-271 °C (hexane-THF).

IR (KBr): 3153, 1620, 1605, 1396 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ):  $\delta = 3.73$  (s, 3 H), 6.78 (dd, J = 7.8, 2.0 Hz, 1 H), 6.82 (t, J = 2.0 Hz, 1 H), 6.94 (dd, J = 8.3, 2.0 Hz, 1 H), 7.33 (t, J = 7.8 Hz, 1 H), 7.37 (dd, J = 8.3, 7.8 Hz, 1 H), 7.41 (d, J = 8.3 Hz, 1 H), 7.77 (ddd, J = 8.3, 7.8, 1.5 Hz, 1 H), 8.31 (d, J = 7.8 Hz, 1 H), 13.40 (br s, 1 H)

<sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ): δ = 55.3, 113.4, 114.6, 116.1, 120.8, 123.8, 125.3, 130.0, 131.8, 135.4, 135.9, 145.2, 160.2, 172.8, 189.6.

MS: m/z (%) = 300 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{15}H_{12}N_2OS_2:$  C, 59.97; H, 4.03; N, 9.33. Found: C, 59.85; H, 4.05; N, 9.27.

# 3-(Naphthalen-2-yl)quinazoline-2,4(1H,3H)-dithione (6h)

Pale-yellow solid; mp 171-173 °C (hexane-THF).

IR (KBr): 3160, 1614, 1373 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, DMSO- $d_6$ ): δ = 7.29–7.60 (m, 5 H), 7.70–7.89 (m, 2 H), 7.91 (d, J = 7.8 Hz, 1 H), 7.97 (d, J = 8.3 Hz, 1 H), 8.00 (d, J = 8.8 Hz, 1 H), 8.33 (dd, J = 8.3, 1.0 Hz, 1 H), 13.55 (br s, 1 H).

<sup>13</sup>C NMR (100 MHz, DMSO- $d_6$ ):  $\delta = 116.1$ , 123.8, 125.4, 126.1, 126.6, 126.8, 126.9, 127.7, 128.1, 128.9, 131.7, 132.4, 133.5, 135.4, 135.9, 141.7, 173.0, 190.0.

MS: m/z (%) = 320 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{18}H_{12}N_2S_2$ : C, 67.47; H, 3.77; N, 8.74. Found: C, 67.21; H, 3.95; N, 8.65.

## 6-Chloro-3-phenylquinazoline-2,4(1H,3H)-dithione (6i)

Pale-pink solid; mp 225-230 °C (hexane-CHCl<sub>3</sub>).

IR (KBr): 3150, 1611, 1387 cm<sup>-1</sup>.

<sup>1</sup>H NMR (400 MHz, CDCl<sub>3</sub>):  $\delta$  = 7.10 (d, *J* = 8.8 Hz, 1 H), 7.21 (d, *J* = 7.8 Hz, 2 H), 7.50–7.60 (m, 4 H), 8.47 (d, *J* = 2.4 Hz, 1 H), 10.87 (br s, 1 H).

<sup>13</sup>C NMR (100 MHz, CDCl<sub>3</sub>): δ = 116.5, 125.4, 128.1, 129.0, 129.9, 132.2, 132.5, 135.5, 135.5, 143.7, 173.2, 188.5.

MS: m/z (%) = 304 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{14}H_9ClN_2S_2$ : C, 55.16; H, 2.98; N, 9.19. Found: C, 54.93; H, 3.23; N, 9.03.

# 3-(2-Bromo-4-chlorophenyl)-6-chloroquinazoline-2,4(1*H*,3*H*)-dithione (6j)

Orange solid; mp 188-190 °C (hexane-THF).

IR (KBr): 3154, 1611, 1386 cm<sup>-1</sup>.

<sup>1</sup>H NMR (500 MHz, DMSO- $d_6$ ):  $\delta$  = 7.46 (d, J = 8.6 Hz, 1 H), 7.49 (d, J = 8.6 Hz, 1 H), 7.60 (dd, J = 8.6, 2.3 Hz, 1 H), 7.88 (dd, J = 8.6, 2.3 Hz, 1 H), 7.89 (d, J = 2.3 Hz, 1 H), 8.24 (d, J = 2.3 Hz, 1 H), 13.51 (br s, 1 H).

<sup>13</sup>C NMR (125 MHz, CDCl<sub>3</sub>): δ = 118.6, 122.8, 124.3, 129.3, 129.7, 130.1, 132.2, 132.6, 133.7, 134.2, 136.1, 141.6, 171.7, 187.2.

MS: *m*/*z* (%) = 415 ([M<sup>+</sup>], 100).

Anal. Calcd for  $C_{14}H_7BrCl_2N_2S_2;\,C,\,40.21;\,H,\,1.69;\,N,\,6.70.$  Found: C, 40.24; H, 1.75; N, 6.55.

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