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Enzymatic Preparation of Homochiral 2-(N-Carbobenzyloxypiperid-4-yl)-1,3propanediol Monoacetate. A Facile Entry to Both Enantiomers of 3-Hydroxymethylquinuclidine.

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Abstract: Both enantiomers of monoacetate 1 have been prepared in high ee by lipase catalyzed monohydrolysis of the prochiral diacetate 3 and by monoacetylation of the corresponding diol 4.3-Hydroxymethylquinuclidine 2 was then synthesized in both enantiomeric forms with an ee up to 98% starting from (+)-1.

Enantiomerically pure molecules containing a nitrogen-heterocycle are useful units that can be used as starting material in the synthesis of alkaloids and many other important biologically active targets. Unfortunately their number is rather limited and continuous efforts are currently made in order to improve the so called "heterocycle chiral pool". ²

Biocatalysis is a very powerful tool in organic synthesis for the high chemo, regio and stereoselectivity often exhibited by many enzyme assisted processes. Lipases have proved to be one of the most effective class of biological catalysis being easily available, inexpensive and requiring no cofactor. Moreover, they accept a broad range of substrates often still maintaining high stereoselectivity, 3.4

As part of our ongoing programme on the use of enzymes in the asymmetric synthesis of new chiral building blocks starting from prochiral substrates with two enantiotopic groups, \(^{1.4a,5}\) we report here our results on the lipase-catalyzed preparation of both enantiomers of 3-acetoxy-2-(N-carbobenzyloxypiperid-4-yl)-1-propanol 1,6 a new piperidine ring containing chiral building block useful for many synthetic applications. Afterwards, as a first example of the usefulness and stereochemical versatility of 1, we report herein the synthesis of both enantiomers of 3-hydroxymethylquinuclidine 2, an intermediate of considerable interest for the synthesis of mequitazine and other 3-substituted quinuclidine derived drugs.

First we studied the enantioselective hydrolysis in the presence of PPL of the diester 3,8 a derivative of 2-(4-piperidyl)-1,3-propanediol in which the two hydroxy groups were acetylated and the nitrogen of the piperidine ring was protected as carbobenzyloxy derivative. When the reaction was stopped at 47% conversion, monoacetate (-)-1 was isolated in 49% chemical yield and in 44% ee (Table 1). Some attempts to improve either chemical yield and enantiomeric excess using different lipases and degrees of conversion were made the best outcome, in terms of enantiomeric excess, was achieved using PPL and stopping the reaction at 73% conversion (entry 2); in these conditions the monoacetate (-)-1 was obtained in 74% ee, but only in 39% chemical yield.

Since the results were acceptable but not excellent we turned our attention to the enantioselective esterification of the prochiral diol 4 in organic solvent. We examined in all cases the acetylation reaction, using various lipases and different series of conditions (solvent, acylating agent, degree of conversion, weight ratio enzyme/diol, enzyme supportation). The most significant results are reported in Table 2. As expected the

Table 1: Lipase Catalyzed Monohydrolysis of Diacetate 3a

diacet	Phosp	monoacctate(-)-1 + diol 4			
entry	lipase ^b	reaction time (h)	conversion ^c	(-)-1 (yield %)	eed (%)
1	PPL	5.5	47	49	44
2	PPL_	19	73	39	74
3	CCL	18	65	45	56
4	AYL.	8	45	45	60

a) All reactions were carried out on a 0.6 mmol scale using a ponderal ratio enzyme-diacetate = 0.75 b) See note 9 c) Conversion means the percentage of initial acetyl groups which have been hydrolysed d) Determined by formation of Mosher's esters and ¹HNMR analysis of diastereoisomeric compositions. Major enantioner was always (-)-1 (abs. conf. S)

Table 2: Lipase Catalyzed Monoacetylation of Diol 4a

		diol 4 —	-	monoacet	ate(+)-1 +	diacetate 3		
entry	lipase ^b	weight ratio enzyme/4	acyl donor ^e	solvent ^c	reaction time (h)	conver- sion ^d (%)	(+)-1 (yield %)	ee c (%)
1	PPL	1	VA	VA	8.5	58	83	92
2	PPL	1	VA	VA	30	73	53	95
3	PPL	1.5	VA	VA	15	64	63	94
4	PPI.	1.5	VA	$\mathbf{V}\mathbf{A}$	23	74	52	>98f
5	PPL	l	Ac ₂ O	Toluene	8	62	71	60
6	PPL.	1	AcOEt	AcOEt	26	37	75	82
7	PPL.	1	AcOEt	ΛcΟEt	92	55	89	85
8	SAM-II	0.2	VA	VA	8	59	81	76
9	SAM-II	0.2	VA	VA	30	84	31	46
10	CAL	0.25	VA	VA	9.5	85	30g	9()g
11	CCL	3	VA	VA	7 days	48	70	30
12	PPL on celite	3.3	VA	VA	3 *	56	87	90
13	PPL on celite	3.3	VA	VA	10	74	52	94

a) All reactions were carried out on a 0.3 minol scale using 3 ml of solvent, in the presence of 3 Å molecular sieves. When the acyl donor was not used as solvent, 5 eq. were employed, b) See note 9 c) VA = vinyl acctate. d) Conversion means the percentage of initial hydroxy groups which have been acctylated e) Determined by formation of Mosher's esters and ¹HNMR analysis of disstereoisomenic compositions. Major enautioner was always: (+)-1 (absolute configuration R), except in entry 10 f) $\{\alpha\}_D = +9.1$ (CHCl3, c = 1.1) g) Major enautioner was (-)-1 (absolute configuration S)

monoacetate of opposite configuration to that derived from hydrolysis of 3 was generally obtained. However, while with PPL, SAM-II and CCI, the same prochiral hydroxymethyl was preferentially acetylated to give (+)-1, in the case of CAL (entry 10) the opposite enantiomer (-)-1 was predominant. The absolute configuration R was assigned to monoacetate (+)-1 by chemical correlation with (+)-(R)-3-hydroxymethylquinuclidine into which (+)-1 was converted as described below. These facts indicate that, in the PPL catalyzed reactions of 3 and 4, the *pro-R* CH₂OR of the prochiral substrate (either diol and diacetate) reacts preferentially: this result is in agreement with the empirical model of the active site of PPL recently proposed by us. 4a

It is noteworthy that pushing the conversion to higher degrees an improvement of enantioselectivity is generally observed. However, in the case of Lipase SAM-II (entry 8 and 9) an increase of conversion degree brings about a drastic loss of ee. According to the collected data the best combination of the various parameters seems to be the one described in entry 4: using crude PPL purchased from SIGMA, employing vinyl acetate both as acylating agent and solvent, and a 1.5 weight ratio enzyme/substrate, a chemical yield of 52% with an excellent ee > 98% was obtained when the reaction was stopped after 23 h at 74% conversion.

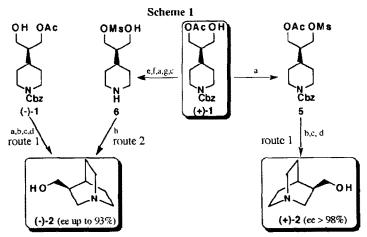
Having in our hands (+)-1 in high optical purity, we could start to examine some synthetic applications of this new chiral building block. As a first exploitation we report here the preparation of both enantiomers of 3-hydroxymethylquinuclidine 2 (Scheme 1).

The synthesis of (+)-2 was realized transforming (+)-1 (ec > 98%, entry 4 in Table 2) into the corresponding mesylate 5 followed by hydrolysis of the acetate (LiOH in 2/1 THF/H₂O), hydrogenolysis of Cbz group and heating in acetonitrile in the presence of Li₂CO₃.

From the same (R)-monoacetate (+)-1 we synthesized (-)-(S)-2¹⁰ cyclizing 6, obtained by protection of the hydroxyl group of 1 as THP, followed by hydrolysis of the acetate, mesylation of OH and removal of THP and Cbz groups. Of course this enantiomer could be obtained more directly from (-)-1 using the same process (route 1) just described for achieving (+)-2 from (+)-1. However, since (-)-1 was obtained with lower ec, ¹¹ we preferred to follow the two steps longer route 2

Formation of Mosher's esters from the products (+)- and (-)-2 and ¹HNMR analysis of their diastereoisomeric compositions allowed to establish that, while the route 1 is completely enantiospecific ((+)-2 could be obtained with ec > 98%), in the case of route 2 a slight racemization occurred at the level of THP protection so that (-)-2 could be obtained from (+)-1 with ee up to 93%.

Other synthetic applications of this new chiral building block are now in progress in our laboratories. We wish to thank C.N.R., Progetto Finalizzato Chimica Fine, and M.U.R.S.T. for financial assistance.



Reagents and conditions a) MsCl, Et3N, CH2Cl2, -30°C, 5', b) LiOH, THF/H2O 2/1, r t, 6 h, c) H2, Pd/C 10%, EtOH/AcOH, r, t; d) Lt2CO3, CH3CN, reflux, 2 h, 63% (4 steps); e) 3,4-dihydro-2*H* -pyran, *p*-TSA, CH2Cl2, 0°C, 5', f) LiOH, THF/H2O 2/1, r, t, 8 h, 95% (2 steps), g) *p*-TSA, MeOH, r t, overnight; h) Lt2CO3, CH3CN, reflux, 2 h, 60% (4 steps).

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- 8. Preparation of **3** from the commercially available 1-benzyl-4-piperidone: **a**) 2,2-dimethyl-1,3-dioxane-4,6-dione (Meldrum's acid), pyndine, r. t., 48 h, 89%; **b**) H₂, Pd/C 10%, MeOH, r. t., overnight, 92%; **c**) HCl/MeOH, reflux, 1 h; **d**) BzlOCOCl, NaOH, H₂O, 0°C, 95% (2 steps); **e**) NaBH₄, LiCl, EtOH/THF, r. t., 83% **f**) Ac₂O, Et₃N, DMAP, CH₂Cl₂, r. t., 30°, 90%
- 9 Crude PPL (50 U/mg of powder, measured using olive oil as standard) and CCL (from Candida cilyindracea, 500 U/mg of powder) were purchased from Sigma. Lipase SAM-II (from Pseudomonas fluorescens, 42 U/mg) was purchased from Fluka. Lipases Amano AY (from Candida, 30 U/mg) and Amano PS (from Pseudomonas cepacia; 30 U/mg) were kindly provided to us by Mitsubishi Italia CAL (lipase Novozym 435, from Candida antarctica; 7 U/mg) was kindly provided to us by Novo Nordisk. PPL on celite was prepared simply liophylising a mixture of crude PPL and celite in pH 7 phosphate buffer, according to the procedure reported in ref. 5b.
- 10 Renault, C; Le Fur, G. Fr. Demand FR 2,522,660; 5th March 1982
- 11. We are still working in order to improve this result