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A convenient three-component reaction for the construction of multifunctional 6-fluoroalkyl-1,2,3,4-tetrahydropyrimidines

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Abstract

 β -Fluoroalkylenaminoketones were prepared and their reaction with primary amines and formaldehyde were studied. Multifunctional 6-fluoroalkyl-1,2,3,4-tetrahydropyrimidines were synthesized conveniently from the three-component reaction in good yields under mild conditions. \bigcirc 2005 Elsevier B.V. All rights reserved.

Keywords: β-Fluoroalkylenaminoketone; Fluoroalkylated 1,2,3,4-tetrahydropyrimidine; Primary amine; Formaldehyde

1. Introduction

The ability of fluorine atom to enhance biological and therapeutic activities of certain organic compounds has led to widespread interest in selective introduction of fluorine atom and fluoroalkyl groups into organic molecules [1-5], especially those heterocyclic compounds which have potential biological activities. 1,2,3,4-Tetrahydropyrimidine ring is an important moiety in both natural and synthetic organic compounds, and many compounds containing it show interesting biological activities [6-12]. Accordingly many methods have been developed for the synthesis of various 1,2,3,4-tetrahydropyrimidines [13]. However, the synthesis of fluoroalkylated 1,2,3,4-tetrahydropyrimidines is less studied [14] and their biological activities remain unexplored. Therefore, it is our interest to develop synthetic methodologies for those fluorinecontaining heterocycles which might have unique properties by the introduction of fluoroalkyl groups.

Recently, it was found in our laboratory that the reaction of fluorine-containing enaminoketones or 3-fluoroalkylanilinoacrylic acid esters with primary amines and formaldehyde could afford multifunctional 1,2,3,4-tetrahydropyrimidines under mild conditions [14]. To explore the scope of this procedure further, β -fluoroalkylenaminoketones were prepared and their reaction with primary amines and formaldehyde were

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investigated. The results of this three-component reaction are reported in this paper.

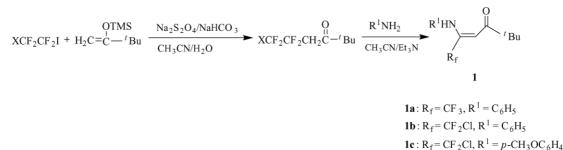
2. Results and discussion

 β -Fluoroalkylenaminoketones (1) were prepared from commercial available fluoroalkyl iodides as shown in Scheme 1 [15].

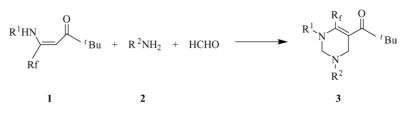
The reaction of **1b** with methylamine and formaldehyde was first carried out in refluxing acetonitrile with a mole ratio of 1:2:4. It was found that the reaction was very slow and only trace product was formed after 24 h. Then the amount of methylamine and formaldehyde was increased. With a ratio of 1:8:18 (**1**/ methylamine/formaldehyde), the reaction was completed in 10 h. After workup, a light yellow solid was obtained. Spectral and elemental analysis showed that it was the expected product, 6-chlorodifluoromethyl-5-*tert*-butylcarbonyl-3-methyl-1-phenyl-1,2,3,4-tetrahydropyrimidine (**3ba**) (Scheme 2).

To find suitable conditions, the reaction of **1b**, methylamine and formaldehyde under various conditions was investigated. As shown in Table 1, the yield of **3ba** was influenced by the ratio of three reactants and better result was obtained when more methylamine and formaldehyde were used. Solvent and temperature also played important roles in the reaction. When the reaction was carried out in DMF at 110 °C, the reaction time was fantastically shortened, and much less methylamine and formaldehyde were needed for a satisfactory conversion of **1b**. Increasing temperature further made the reaction more

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Scheme 1.





complex. Therefore, an optimized condition was chosen for this three-component reaction as follows: in DMF at 110 $^{\circ}$ C with a mole ratio of 1:3:6.5 (**1b**:CH₃NH₂:HCHO).

Using the optimized conditions, various β -fluoroalkylenaminoketones and primary amines were examined. As shown in Table 2, aliphatic primary amines reacted readily with **1** and formaldehyde to give the desired product in good yields. The length and bulkiness of primary amine had little influence on the reaction. However, the reaction was very complicated when aromatic amines such as aniline and *para*-methoxy aniline were used and no expected cyclic products were obtained. The electron charge of the aromatic group in enaminoketones had little influence on the reaction. Enaminoketones containing either electron-rich or electron-deficient aromatic ring gave satisfactory results.

In conclusion, the three-component reaction of β -fluoroalkylenaminoketone, primary amine and formaldehyde was achieved under mild conditions, providing a convenient method for the synthesis of fluoroalkylated multifunctional 1,2,3,4-

Table 1

The reaction of ${\bf 1b}$ with methylamine and formal dehyde under different conditions $^{\rm a}$

Entry	Solvent	1b :CH ₃ NH ₂ : НСНО	Temperature (°C)	Time (h)	Yield (%) ^b
1	CH ₃ CN	1:2:4	Reflux	24	Trace
2	CH ₃ CN	1:4:9	Reflux	17	78
3	CH ₃ CN	1:8:18	Reflux	10	87
4	DMF	1:8:18	80	9	90
5	DMF	1:4:9	80	15	81
6	DMF	1:4:9	110	2.5	90
7	DMF	1:3:6.5	110	3	88
8	DMF	1:2:4	110	10	50
9	DMF	1:3:6.5	140	2	Complex

^a All reactions were carried out with **1** (1 mmol) in solvent (5 mL). ^b Determined by ¹⁹F NMR. tetrahydropyrimidines. Using this method, a series of novel multifunctional 6-fluoroalkyl-1,2,3,4-tetrahydropyrimidines were synthesized from easily available starting materials for the first time.

1d: $R_f = CF_2Cl, R^1 = p-BrC_6H_4$

3. Experimental

3.1. General

Melting points were uncorrected. IR spectra were taken on a Perkin-Elmer Jeol 983 spectrophotometer. ¹H NMR spectra were measured on a Bruker AM300 (300 MHz) spectrometer using TMS as internal standard. ¹⁹F NMR spectra were taken on a Bruker AM300 (282 MHz) spectrometer using CFCl₃ as external standard. Mass spectra were obtained on a Hewlett– Packard HP-5989A spectrometer. Column chromatography was performed on silica gel H, particle size 10–40 µm.

Table	2

The reaction of 1 with primary amines and formaldehyde^a

Entry	1	Time (h)	$R_{\rm f}$	R^1	R ²	Product	Yield (%) ^b
1	1a	2	CF ₃	Ph	Me	3aa	85
2	1a	3	CF ₃	Ph	Et	3ab	83
3	1a	3	CF ₃	Ph	<i>n</i> -Pr	3ac	80
4	1a	3	CF ₃	Ph	t-Bu	3ad	84
5	1b	3	CF_2Cl	Ph	Me	3ba	85
6	1b	4	CF ₂ Cl	Ph	Et	3bb	80
7	1b	4	CF ₂ Cl	Ph	<i>n</i> -Pr	3bc	80
8	1b	5	CF ₂ Cl	Ph	t-Bu	3bd	80
9	1c	3	CF ₂ Cl	p-OMe-Ph	Me	3ca	82
10	1c	3	CF_2Cl	p-OMe-Ph	Et	3cb	80
11	1c	4	CF ₂ Cl	p-OMe-Ph	<i>n</i> -Pr	3cc	78
12	1c	5	CF ₂ Cl	<i>p</i> -OMe-Ph	t-Bu	3cd	80
13	1d	4	CF ₂ Cl	<i>p</i> -Br-Ph	Me	3da	79

 $^{\rm a}$ All reactions were carried out with 1 (1 mmol) in DMF (5 mL). $^{\rm b}$ Isolated yields based on 1.

3.2. Typical procedure for the synthesis of 6-fluoroalkyl-1,2,3,4-tetrahydropyrimidines (3)

To a solution of **1** (1 mmol) in DMF (5 mL), were added primary amines **2** (3 mmol) and 37% formaldehyde (6.5 mmol). The mixture was stirred at 110 °C for 2–5 h (monitored by TLC or ¹⁹F NMR). After completion of the reaction, the mixture was cooled to room temperature, diluted with water and extracted with Et₂O. The organic layer was washed with saturated NaCl solution and dried over anhydrous Na₂SO₄. The residue obtained after evaporation of the solvent was purified by flash chromatography on silica gel eluting with petroleum-ethyl acetate to give **3**.

3.2.1. 5-tert-Butylcarbonyl-3-methyl-1-phenyl-6trifluoromethyl-1,2,3,4-tetrahydropyrimidine (**3aa**)

Light yellow liquid. IR (film): ν 3062, 2975, 1695, 1646, 1596, 1492, 1370, 1183, 1158, 1135, 865, 767, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.28–7.33 (2H, m), 7.14–7.17 (3H, m), 3.92 (2H, s), 3.35 (2H, s), 2.29 (3H, s), 1.26 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ –60.71 (s); EIMS (70 eV) *m*/*z* (%): 326 [*M*⁺] (23), 269 (53), 57 (100). Anal. Calcd. for C₁₇H₂₁F₃N₂O: C, 62.56; H, 6.49; N, 8.58. Found: C, 62.53; H, 6.59; N, 8.34.

3.2.2. 5-tert-Butylcarbonyl-3-ethyl-1-phenyl-6trifluoromethyl-1,2,3,4-tetrahydropyrimidine (**3ab**)

Light yellow liquid. IR (film): ν 3062, 2974, 1694, 1645, 1597, 1493, 1369, 1159, 1135, 1071, 990, 763, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.28–7.33 (2H, m), 7.14–7.17 (3H, m), 4.04 (2H, s), 3.43 (2H, s), 2.50 (2H, q, *J* = 7.2 Hz), 1.29 (9H, s), 0.95 (3H, t, *J* = 7.2 Hz); ¹⁹F NMR (282 MHz, CDCl₃): δ –60.44 (s). EIMS (70 eV) *m*/*z* (%): 340 [*M*⁺] (25), 283 (100), 227 (79), 57 (44). Anal. Calcd. for C₁₈H₂₃F₃N₂O: C, 63.51; H, 6.81; N, 8.23. Found: C, 63.43; H, 6.83; N, 8.03.

3.2.3. 5-tert-Butylcarbonyl-1-phenyl-3-propyl-6trifluoromethyl-1,2,3,4-tetrahydropyrimidine (**3ac**)

Light yellow liquid. IR (film): ν 3061, 2968, 1693, 1647, 1597, 1493, 1479, 1465, 1367, 1297, 1180, 1158, 1136, 927, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.26–7.32 (2H, m), 7.11–7.17 (3H, m), 4.04 (2H, s), 3.42 (2H, s), 2.38 (2H, t, J = 7.5 Hz), 1.27–1.37 (2H, m), 1.26 (9H, s), 0.74 (3H, t, J = 7.2 Hz,); ¹⁹F NMR (282 MHz, CDCl₃): δ –60.42 (s); EIMS (70 eV) m/z (%): 354 [M^+] (16), 297 (44), 226 (55), 57 (100). Anal. Calcd. for C₁₉H₂₅F₃N₂O: C, 64.39; H, 7.11; N, 7.90. Found: C, 4.28; H, 7.08; N, 7.86.

3.2.4. 5-tert-Butylcarbonyl-1-phenyl-3-tert-butyl-6trifluoromethyl-1,2,3,4-tetrahydropyrimidine (**3ad**)

Light yellow solid: mp 54–55 °C; IR (KBr): ν 3061, 2955, 1693, 1644, 1597, 1492, 1360, 1199, 1181, 1164, 1128, 917, 753, 698 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.26–7.32 (2H, m), 7.11–7.17 (3H, m), 3.94 (2H, s), 3.46 (2H, s), 1.27 (9H, s), 0.92 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ –61.61 (s); EIMS (70 eV) *m/z* (%): 330 [*M*⁺] (3), 368 (13), 226 (100), 283 (60).

Anal. Calcd. for $C_{20}H_{27}F_3N_2O$: C, 65.20; H 7.39, N, 7.60. Found: C, 65.08; H, 7.45; N, 7.49.

3.2.5. 5-tert-Butylcarbonyl-6-chlorodifluoromethyl-3methyl-1-phenyl-1,2,3,4-tetrahydropyrimidine (**3ba**)

Light yellow solid: mp 55–57 °C; IR (KBr): ν 3062, 3040, 2976, 2849, 1687, 1639, 1595, 1495, 1181, 1154, 1136, 1114, 896, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.27–7.31 (2H, m), 7.15–7.19 (3H, m), 3.90 (2H, s), 3.35 (2H, s), 2.29 (3H, s), 1.29 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ –48.11 (s). EIMS (70 eV) m/z (%): 342 [M^+] (5), 214 (100), 57 (21). Anal. Calcd. for C₁₇H₂₁ClF₂N₂O: C, 59.66; H, 6.17; N, 8.17. Found: C, 59.61; H, 6.25; N, 7.98.

3.2.6. 5-tert-Butylcarbonyl-6-chlorodifluoromethyl-3ethyl-1-phenyl-1,2,3,4-tetrahydropyrimidine (**3bb**)

Light yellow liquid. IR (film): ν 3064, 2974, 1695, 1638, 1597, 1494, 1180, 1153, 1129, 896, 761, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.26–7.36 (2H, m), 7.14–7.20 (3H, m) 3.99 (2H, s), 3.41 (2H, s), 2.49 (2H, q, *J* = 7.2 Hz), 1.29 (9H, s), 0.95 (3H, t, *J* = 7.2 Hz); ¹⁹F NMR (282 MHz, CDCl₃): δ –47.68 (s); EIMS (70 eV) *m*/*z* (%): 356 [*M*⁺] (4), 299 (18), 214 (100). Anal. Calcd. for C₁₈H₂₃ClF₂N₂O: C, 60.59; H, 6.50; N, 7.85. Found: C, 60.56; H, 6.28; N, 7.82.

3.2.7. 5-tert-Butylcarbonyl-6-chlorodifluoromethy-1phenyl-3-propyl-1,2,3,4-tetrahydropyrimidine (**3bc**)

Light yellow liquid. IR (film): ν 3064, 2968, 1694, 1638, 1597, 1494, 1479, 1193, 1133, 912, 699 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.26–7.32 (2H, m), 7.14–7.20 (3H, m), 3.97 (2H, s), 3.41 (2H, s), 2.38 (2H, t, *J* = 7.2 Hz), 1.30–1.36 (2H, m), 1.29 (9H, s), 0.75 (3H, t, *J* = 7.2 Hz); ¹⁹F NMR (282 MHz, CDCl₃): δ –47.72 (s); EIMS (70 eV) *m*/*z* (%): 370 [*M*⁺] (10), 214 (100), 57 (72). Anal. Calcd. for C₁₉H₂₅ClF₂N₂O: C, 61.53; H, 6.79; N, 7.55. Found: C, 61.56; H, 6.99; N, 7.35.

3.2.8. 3-tert-Butyl-5-tert-butylcarbonyl-6-chlorodifluoromethyl-1-phenyl-1,2,3,4-tetrahydropyrimidine (**3bd**)

Light yellow solid: mp 64–66 °C; IR (KBr): ν 3062, 2974, 1684, 1630, 1595, 1493, 1479, 1366, 1289, 1181, 1133, 1124, 887, 696 cm⁻¹; ¹H NMR (300 MHz,CDCl₃): δ 7.26–7.29 (2H, m), 7.16–7.18 (3H, m), 3.95 (2H, s), 3.47 (2H, s), 1.27 (9H, s), 0.93 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ –48.53 (s); EIMS (70 eV) *m*/*z* (%): 384 [*M*⁺] (4), 214 (100), 57 (38). Anal. Calcd. for C₂₀H₂₇ClF₂N₂O: C, 62.41; H, 7.07; N, 7.28. Found: C, 62.62; H, 7.30; N, 7.00.

3.2.9. 5-tert-Butylcarbonyl-6-chlorodifluoromethyl-3methyl-1-(4-methoxyphenyl)-1,2,3,4-tetrahydropyrimidine (**3ca**)

Light yellow liquid. IR (film): ν 3041, 2972, 1693, 1637, 1584, 1509, 1464, 1183, 1153, 1125, 1068, 902, 836, 794 cm⁻¹; ¹H NMR (300 MHz,CDCl₃): δ 7.12–7.16 (2H, m), 6.82–6.86 (2H, m), 3.78–3.81 (5H, m), 3.31 (2H, s), 2.30 (3H, s), 1.28 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ –47.30 (s); EIMS (70eV) *m*/*z* (%): 372 [*M*⁺] (6), 244 (100), 57 (84). Anal. Calcd. for

C₁₈H₂₃ClF₂N₂O₂: C, 57.99; H, 6.22; N, 7.51. Found: C, 57.85; H, 6.31; N, 7.48.

3.2.10. 5-tert-Butylcarbonyl-6-chlorodifluoromethyl-3ethyl-1-(4-methoxyphenyl)-1,2,3,4-tetrahydropyrimidine (*3cb*)

Light yellow liquid. IR (film): ν 3041, 2972, 1693, 1637, 1584, 1508, 1494, 1479, 1199, 1180, 1127, 920, 899, 836, 791 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.16 (2H, d, J = 9.0 Hz), 6.83 (2H, d, J = 9.0 Hz), 3. 88 (2H, s), 3.78 (3H, s), 3.37 (2H, s), 2.48 (2H, q, J = 7.2 Hz), 1.27 (9H, s), 0.97 (3H, t, J = 7.2 Hz); ¹⁹F NMR (282 MHz, CDCl₃): δ -46.98 (s); EIMS (70 eV) m/z (%): 386 [M^+] (7), 329 (36), 244 (100). Anal. Calcd. for C₁₉H₂₅ClF₂N₂O₂: C, 58.99; H, 6.51; N, 7.24; Found: C, 59.17; H, 6.62; N, 7.20.

3.2.11. 5-tert-Butylcarbonyl-6-chlorodifluoromethyl -1-(4methoxyphenyl)-3-propyl-1,2,3,4-tetrahydropyrimidine (**3cc**)

Light yellow liquid. IR (film): ν 3041, 2966, 1693, 1637, 1584, 1509, 1475, 1465, 1246, 1193, 1180, 1153, 1127, 1034, 1008, 904, 836, 796 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.16 (2H, d, *J* = 8.7 Hz), 6.83 (2H, d, *J* = 8.7Hz), 3.84 (2H, s), 3.78 (3H, s), 3.35 (2H, s), 2.37 (2H, t, *J* = 7.2 Hz), 1.33–1.40 (2H, m), 1.26 (9H, s), 0.78 (3H, t, *J* = 7.2 Hz); ¹⁹F NMR (282 MHz, CDCl₃): δ -47.90 (s); EIMS (70 eV) *m*/*z* (%): 400 [*M*⁺] (7), 329 (40), 244 (100), 57 (53). Anal. Calcd. for C₂₀H₂₇ClF₂N₂O₂: C, 59.92; H, 6.79; N, 6.99. Found: C, 59.83; H, 6.80; N, 6.80.

3.2.12. 6-Chlorodifluoromethyl-1-(4-methoxyphenyl)-3tert-butyl-5-tert-butylcarbonyl-1,2,3,4-tetrahydropyrimidine (**3cd**)

Light yellow liquid. IR (film): ν 2972, 1693, 1637, 1610, 1584, 1508, 1478, 1465, 1290, 1247, 1199, 1183, 1129, 1034, 921, 895, 836 cm⁻¹; ¹H NMR (300 MHz,CDCl₃): δ 7.14–7.16 (2H, m), 6.79–6.83 (2H, m), 3.85 (2H, s), 3.78 (3H, s), 3.44 (2H, s), 1.26 (9H, s), 0.94 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ 48.16 (s); EIMS (70 eV) m/z (%): 414 [M^+] (7), 329 (37), 244 (100), 57 (14). Anal. Calcd. for C₂₁H₂₉ClF₂N₂O₂: C, 60.79; H, 7.04; N, 6.75. Found: C, 60.62; H, 7.02; N, 6.63.

3.2.13. 1-(4-Bromophenyl)-5-tert-butylcarbonyl-6chlorodifluoromethyl-3-methyl-1,2,3,4tetrahydropyrimidine (**3da**)

Light yellow liquid. IR (film): ν 2974, 1693, 1638, 1588, 1487, 1181, 1152, 1139, 1066, 1008, 900, 852, 768 cm⁻¹; ¹H NMR (300 MHz, CDCl₃): δ 7.44 (2H, d, J = 9.0 Hz), 7.05 (2H, d, J = 9.0Hz), 3.85 (2H, s), 3.33 (2H, s), 2.28 (3H, s), 1.28 (9H, s); ¹⁹F NMR (282 MHz, CDCl₃): δ 48.46 (s). EIMS (70 eV) m/z (%): 422 [M^+] (8), 379 (27), 292 (73), 57 (100). Anal. Calcd. for C₁₈H₂₃ClF₂N₂O₂: C, 48.42; H, 4.78; N, 6.64. Found: C, 48.64; H, 4.97; N, 6.50.

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