SYNTHESIS OF HEXA- AND OCTAHYDRO-PYRIDO[3'2':4,5]THIENO[3,2-*d*]PYRIMIDINES*

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The reaction of N-methylmorpholinium 4-(2-chlorophenyl)-5-cyano-2-oxo-1,2,3,4-tetrahydropyridine-6thiolate with chloroacetamide in DMF in the presence of an excess of KOH gave 3-amino-2-carbamoyl-4-(2-chlorophenyl)-6-oxo-4,5,6,7-tetrahydrothieno[2,3-b]pyridine. Refluxing the latter with acyl chlorides in AcOH or heating in formic acid gave hexahydropyrido[3',2':4,5]thieno[3,2-d]pyrimidine derivatives and reaction with cyclohexanone gave a spiro-linked octahydropyrido[3',2':4,5]thieno-[3,2-d]pyrimidine.

Keywords: tetrahydropyridinethiolate, thienopyridine, partially hydrogenated pyridothienopyrimidines, cyclization.

Substituted 4,7-dihydrothieno[2,3-*b*]pyridines and their dehydrogenated analogs have been successfully used previously as starting reagents in the synthesis of polycondensed heterocycles, including pyrido[3',2':4,5]thieno[3,2-*d*]pyrimidines [1-3]. Amongst compounds with a similar structure there have been discovered substances with antianaphylactic [4, 5] and antimicrobial [6] activity. With this in mind, the development of convenient methods for the synthesis of such compounds seems timely and promising.

In continuation of our study of the chemical properties of tetrahydropyridones [7, 8] and based on the tetrahydropyridinethiolate 1 we have carried out the synthesis of the substituted tetrahydrothieno[2,3-*b*]pyridine 2. By further reaction it is transformed to the previously unknown partially hydrogenated pyrido[3',2':4,5]thieno[3,2-*d*]pyrimidines.

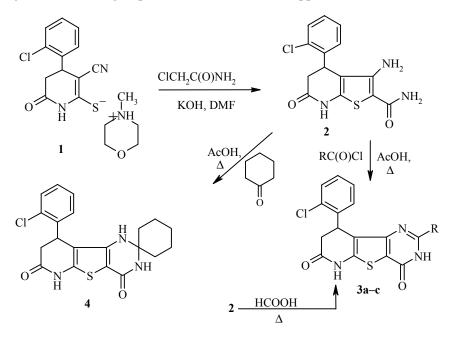
Hence the reaction of salt 1 with chloroacetamide in the presence of an excess of 10% KOH solution in DMF gives a 91% yield of the target product 2. The latter reacts with excess acetyl chloride or chloroacetyl chloride when refluxed in glacial AcOH to form the condensed structures 3a,b. The reaction of compound 2 with cyclohexanone gives a 65.5% yield of octahydropyrido[3',2':4,5]thieno[3,2-d]pyrimidine 4. Reaction of the thienopyridone 2 with 99% formic acid leads to the formation of compound 3c via a sequence of N-formylation and cyclization processes.

The IR spectra of the synthesized compounds show absorption bands for the stretching vibrations of the amino (for **2**) and imino groups (for **2-4**) in the range 3550-3150 cm⁻¹. The signals for the carbonyl groups are seen in the range 1655-1680 cm⁻¹. The ¹H NMR spectra of all of the synthesized samples show characteristic bands for the CH(Ar)–CH₂ fragment of the tetrahydropyridine ring [7, 9] as three double doublets for CH₂ (δ 2.59-2.64 and 3.05-3.17 ppm) and CH (δ 4.54-4.87 ppm). In a number of examples the signals for the

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indicated protons appear as broadened pseudo doublets. From a number of signals confirming the pyridothienopyrimidine structure in compounds **3a-c** the most diagnostic are the singlet for the N₍₃₎H group proton (δ 12.24-12.67 ppm) and the corresponding signals for the other substituents. The presence of the pyrimidine fragment in the structure of compound **4** is confirmed by the multiplet in the range δ 1.11-1.87 ppm ((CH₂)₅) and the singlets for the NH groups (δ 10.81, 7.15, and 6.16 ppm).



3 \mathbf{a} R = Me; \mathbf{b} R = CH₂Cl; \mathbf{c} R = H

EXPERIMENTAL

¹H NMR spectra were taken on a Gemini 200 instrument (200 MHz) using DMSO-d₆ and TMS as internal standard. IR spectra were obtained on an IKS-29 spectrophotometer and elemental analysis was performed on a Perkin-Elmer C,H,N analyser. The course of the reaction and purity of the materials were monitored using TLC on Silufol UV-254 with acetone–heptane (1:1) as eluent. Melting points were measured on Koffler stage.

3-Amino-2-carbamoyl-4-(2-chlorophenyl)-6-oxo-4,5,6,7-tetrahydrothieno[2,3-*b***]pyridine (2). Aqueous KOH solution (10%, 23 ml, 41 mmol) was added to a solution of the thiolate 1** (15 g, 41 mmol) in DMF (60 ml). The mixture was taken to reflux and passed through filter paper into a solution of chloroacetamide (4 g, 43 mmol) in DMF (20 ml). The reaction mass was stirred for 12 h, a further 23 ml of the 10% KOH was added, the stirring was continued at reflux for 10 min, and after 48 h it was diluted with water to a volume of 230 ml and held for 7 days. The precipitate was filtered off and washed with water and ethanol. Yield 12 g (91%); mp 259-261°C (EtOH) as a white, finely crystalline powder. IR spectrum, v, cm⁻¹: 1670 (C=O); 3450, 3265, 3150 (NH₂ and CONH₂). ¹H NMR spectrum, δ , ppm (*J*, Hz): 10.71 (1H, s, NH); 6.75-7.44 (4H, m, Ar); 6.47 (2H, br. s, CONH₂); 6.16 (2H, br. s, NH₂); 4.54 (1H, br. pseudo-d, C₍₄₎H); 3.07 (1H, m, C₍₅₎H); 2.64 (1H, br. pseudo-d, ²*J* = 15.7, C₍₅₎H). Found, %: C 53.00; H 3.72; N 12.98. C₁₄H₁₂ClN₃O₂S. Calculated, %: C 52.26; H 3.76; N 13.06.

9-(2-Chlorophenyl)-2-methyl-4,7-dioxo-3,4,6,7,8,9-hexahydropyrido[3',2':4,5]thieno[3,2-d]pyrimidine (3a). Acetyl chloride (0.44 ml, 6.2 mmol) was added to a suspension of the thienopyridine **2** (1 g, 3.1 mmol) in AcOH (10 ml) and the mixture was refluxed for 6 h. The precipitate formed was filtered off and washed with

ethanol. Yield 0.81 g (75%); mp >300°C as a white, finely crystalline powder. IR spectrum, v, cm⁻¹: 1660, 1680 (2 C=O); 3420, 3465 (2 NH). ¹H NMR spectrum, δ , ppm (*J*, Hz): 12.24 (1H, br. s, N₍₃₎H); 11.03 (1H, s, N₍₆₎H); 6.64-7.46 (4H, m, Ar); 4.80 (1H, br. pseudo-d, C₍₉₎H); 3.15 (1H, m, C₍₈₎H); 2.59 (1H, br. pseudo-d, ²*J* = 16.2, C₍₈₎H); 2.24 (3H, s, Me). Found, %: C 56.50; H 3.60; N 12.12. C₁₆H₁₂ClN₃O₂S. Calculated, %: C 55.57; H 3.58; N 12.15.

4,7-Dioxo-2-chloromethyl-9-(2-chlorophenyl)-3,4,6,7,8,9-hexahydropyrido[**3',2':4,5**]**thieno**[**3,2-***d*]**-pyrimidine (3b)** was obtained similarly to product **3a** using chloroacetyl chloride (0.49 ml, 6.2 mmol). Yield 0.98 g (83%); mp >300°C as a white, finely crystalline powder. IR spectrum, v, cm⁻¹: 1660, 1675 (2 C=O); 3390, 3470 (2 NH). ¹H NMR spectrum, δ , ppm (*J*, Hz): 12.67 (1H, br. s, N₍₃₎H); 11.12 (1H, s, N₍₆₎H); 6.68-7.48 (4H, m, Ar); 4.86 (1H, br. pseudo-d, C₍₉₎H); 4.38 (2H, s, CH₂Cl); 3.17 (1H, dd, ²*J* = 16.3, ³*J* = 8.1, C₍₈₎H); 2.59 (1H, br. pseudo-d, ²*J* = 16.3, C₍₈₎H). Found, %: C 50.98; H 2.93; N 11.01. C₁₆H₁₁Cl₂N₃O₂S. Calculated, %: C 50.54; H 2.92; N 11.05.

4,7-Dioxo-9-(2-chlorophenyl)-3,4,6,7,8,9-hexahydropyrido[**3',2':4,5]thieno**[**3,2-***d*]**pyrimidine** (**3c**). A solution of the thienopyridine **2** (1 g, 3.1 mmol) in 99% formic acid (10 ml) was refluxed for 20 h and the mixture was then diluted with water (15 ml). The precipitate formed was filtered off and recrystallized from AcOH (50 ml). Yield 0.80 g (78%); mp >300°C as a white, finely crystalline powder. IR spectrum, v, cm⁻¹: 1650, 1665 (2 C=O); 3425, 3480 (2 NH). ¹H NMR spectrum, δ , ppm (*J*, Hz): 12.67 (1H, br. s, N₍₃₎H); 11.07 (1H, br. s, N₍₆₎H); 7.86 (1H, br. s, C₍₂₎H); 6.67-7.45 (4H, m, Ar); 4.87 (1H, br. pseudo-d, C₍₉₎H); 3.10 (1H, m, C₍₈₎H); 2.61 (1H, br. pseudo-d, ²*J* = 16.0, C₍₈₎H). Found, %: C 54.91; H 3.01; N 12.60. C₁₅H₁₀ClN₃O₂S. Calculated, %: C 54.30; H 3.04; N 12.66.

4,7-Dioxo-9-(2-chlorophenyl)-2-(1'-cyclohexanespiro)-1,2,3,4,6,7,8,9-octahydropyrido[3',2':4,5]thieno[3,2-d]pyrimidine (4) was obtained similarly to pyrimidine **3a** using cyclohexanone (1.93 ml, 18.6 mmol) and reflux time of 12 h. Yield 0.78 g (65.5%); mp >300°C as colorless crystals. IR spectrum, v, cm⁻¹: 1665, 1680 (2 C=O); 3465, 3435, 3400, 3250 (3 NH). ¹H NMR spectrum, δ , ppm (*J*, Hz): 10.81 (1H, s, N₍₆₎H); 6.71-7.48 (4H, m, Ar); 7.15 (1H, s, N₍₃₎H); 6.16 (1H, s, N₍₁₎H); 4.61 (1H, br. pseudo-d, C₍₉₎H); 3.05 (1H, dd, ²*J* = 16.5, ³*J* = 7.7, C₍₈₎H); 2.64 (1H, br. pseudo-d, ²*J* = 16.5, C₍₈₎H); 1.11-1.87 (10H, m, (CH₂)₅). Found, %: C 60.09; H 4.97; N 10.41. C₂₀H₂₀ClN₃O₂. Calculated, %: C 59.77; H 5.02; N 10.45.

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REFERENCES

- 1. V. P. Litvinov, S. G. Krivokolysko, and V. D. Dyachenko, *Khim. Geterotsikl. Soedin.*, 579 (1999).
- 2. V. P. Litvinov, Izv. Akad. Nauk, Ser. Khim., 2123 (1998).
- 3. V. P. Litvinov, V. K. Promonenkov, Yu. A. Sharanin, and A. M. Shestopalov in: *Summaries of Science and Technology. Organic Chemistry Series* [in Russian], Vol. 17, Moscow (1989), p.72.
- 4. G. Wagner, S. Leistner, H. Vieweg, U. Krasselt, and J. Prantz, *Pharmazie*, 48, 342 (1993).
- 5. N. Boehm, U. Krasselt, S. Leistner, and G. Wagner, *Pharmazie*, 47, 897 (1992).
- 6. M. A. J. Awad, A. E. Abdel-Rahman, and E. A. Bakhtie, *Phosphorus, Sulfur and Silicon and Related Elements*, **57**, 293 (1991).
- 7. V. N. Nesterov, S. G. Krivokolysko, V. D. Dyachenko, V. V. Dotsenko, and V. P. Litvinov, *Izv. Akad. Nauk, Ser. Khim.*, 1029 (1997).
- 8. V. V. Dotsenko, S. G. Krivokolysko, V. P. Litvinov, and A. N. Chernega, *Izv. Akad. Nauk, Ser. Khim.*, 339 (2002).
- 9. S. G. Krivokolysko, V. D. Dyachenko, and V. P. Litvinov, Izv. Akad. Nauk, Ser. Khim., 2333 (1999).