Pd- and Ni-Catalyzed Cross-Coupling Reactions of Functionalized Organozinc Reagents with Unsaturated Thioethers

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Abstract: A variety of unsaturated thioethers have been subjected to crosscoupling reactions with functionalized zinc reagents in the presence of a transition-metal catalyst. Three different catalytic systems based on $Pd(OAc)_2$ or $[Ni(acac)_2]$ and the ligands S-Phos or DPE-Phos gave the best results. N-Heterocyclic thioethers based on a pyridine, pyrimidine, pyrazine, pyridazine, triazine, benzothiazole, benzoxazole, pyrrole, or quinazoline ring, as well as thiomethylacetylenes, serve as electrophiles in this cross-coupling reaction. Aryl-, heteroaryl-, benzylic, and alkyl-

Keywords: cross-coupling • heterocycles • homogeneous catalysis • nickel • palladium • zinc zinc halides with sensitive functionalities, such as ester, nitrile, or ketone groups react at ambient temperature with unsaturated thioethers using a Ni catalyst. The corresponding Pd-catalyzed reactions require slightly higher temperatures. Large-scale cross-coupling experiments (10–20 mmol) with N-heterocycles are also reported.

Introduction

The conversion of a carbon-sulfur bond into a carboncarbon bond under transition-metal catalysis is a useful cross-coupling procedure, which is especially important when the corresponding organic halides are unstable or difficult to prepare. In 1979, the groups of Wenkert and Takei described the cross-coupling between unsaturated thioethers or thiols and Grignard reagents under nickel catalysis.^[1] Later, Fukuyama and co-workers, and in particular Liebeskind and co-workers, extended the scope of this cross-coupling reaction, which led to a general ketone synthesis that involved reacting functionalized thioesters with organozinc reagents or organoboronic acids in the presence of a palladium catalyst.^[2] Furthermore, organostannanes,^[3] as well as organoindium reagents,^[4] have also been used as nucleophilic partners for the direct synthesis of ketones starting from the corresponding thioesters. More recently, this Pd-catalyzed reaction has been modified to allow the coupling of organoboronic acids or organostannanes with heteroaromatic or alkynyl thioethers.^[5] All of these cross-coupling reactions require stoichiometric amounts of Cu^I salts to proceed. Zinc organometallics are very useful reagents for organic synthesis due to their high tolerance to a broad range of functional groups.^[6] However, they have rarely been used for crosscoupling reactions with unsaturated thio derivatives.^[7] Recently, we reported Pd- and Ni-catalyzed cross-coupling re-

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actions of unsaturated heterocyclic^[8] and alkynyl^[9] thioethers with organozinc reagents under mild conditions. Herein, we report the full scope of these novel cross-coupling reactions.

Results and Discussion

Preparation of unsaturated thioethers: Thioether-substituted N-heterocycles, as well as acetylenes can be easily prepared by several methods, as shown in Scheme 1. The thioether moiety can be attached to N-heterocycles by the direct substitution of a halogen atom using the corresponding thiolate. Thus, the chloro-substituted triazine 1a reacted with sodium thiomethanolate in DMF or MeCN at 25°C to provide the corresponding 2,4-dimethoxy-6-(methylthio)-1,3,5-triazine (2a) in 87% yield (Scheme 1a).^[8a] Alternatively, heterocycles may be metallated and then quenched with MeSSMe. Thus, *N*-methylpyrrole (1b) was lithiated with $nBuLi^{[10]}$ and the resulting lithium reagent was treated with dimethyl disulfide to afford 1-methyl-2-(methylthio)pyrrole (2b) in 68% yield (Scheme 1b). The methylation of heterocyclic thiols constitutes another method. Thus, successive treatment of benzoxazole-2-thiol (1c) at 25°C with sodium hydride and iodomethane gave 2-(methylthio)-1,3-benzoxazole (2c) in almost quantitative yield (Scheme 1c).^[11] This methylation step could be combined with a condensation reaction.^[12] The reaction of 4,4,4-trifluoro-1-(2-thienyl)butane-1,3-dione (1d) with thiourea, followed by the above steps with sodium hydride and iodomethane, gave the functionalized pyrimidine 2d in 19% overall yield (Scheme 1d). In the case of the polyfunctional alkynyl steroid 1e,^[13] deprotonation with *i*PrMgCl·LiCl^[14] and subsequent treatment with Smethyl methanethiosulfonate afforded the thiomethylated ethynyl estradiol derivative **3a** in 83% yield (Scheme 1e).^[9]

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Scheme 1. Various pathways for the synthesis of unsaturated thioethers; MOM = methoxymethyl.

Bis(thiomethyl)acetylene (**3b**) was prepared in 55% yield by using sodium acetylide (**1f**) and methyl thiocyanate (25°C, 1 h, Scheme 1f).^[15] These various synthetic methods allowed us to prepare a variety of functionalized unsaturated thioethers, which were used in the subsequent Pd- and Ni-catalyzed cross-coupling reactions.

Pd-catalyzed cross-coupling reactions of unsaturated N-heterocycles with organozinc reagents: Pd⁰-catalyzed cross-coupling reactions of functionalized aryl-, benzylic-, and alkylzinc reagents (1.5 equiv) with various thiomethyl-substituted N-heterocycles (1.0 equiv) were best carried out using Pd-(OAc)₂/S-Phos as the catalytic system. This efficient combination was originally introduced by Buchwald^[16] (Scheme 2). Most of the reactions described herein were carried out at 25 °C within 2–20 h.

Thus, 2,4-dimethoxy-6-(methylthio)-1,3,5-triazine (2a, 1.0 equiv) reacted with 4-methoxyphenylzinc iodide (4a, 1.5 equiv) in THF (25 °C, 5 h) to provide the triazine 5a in 83 % yield (Table 1, entry 1). Furthermore, the triazine 2a was used for a cross-coupling reaction with cyclohexylzinc bromide (4b) in THF (25 °C, 20 h), which led to the functionalized triazine 5b in 43 % yield (Table 1, entry 2). Moreover, 4-cyanophenylzinc iodide (4c) reacted smoothly with



Scheme 2. Pd-catalyzed cross-coupling reaction of thiomethyl-substituted N-heterocycles 2 with functionalized organozinc reagents 4; Cy=cyclohexyl.

methylthio-substituted pyridazine 2e to give the functionalized pyridazine 5c in 76% yield (25°C, 4h, Table 1, entry 3). Functionalized benzylic zinc reagents reacted straightforwardly and the coupling of 3-cyanobenzylzinc chloride (4d) and pyridazine 2e afforded the substituted diarylmethane 5d in 71% yield (50°C, 14 h, Table 1, entry 4). Other diaza heterocycles, such as 2-thiomethylpyrazine (2 f)or 2-thiomethylpyrimidine (2g), also underwent smooth cross-coupling reactions. Thus, the reactions of 1-naphthylzinc iodide (4e) or 3-(trifluoromethyl)benzylzinc chloride (4 f) with methylthio-substituted pyrazine 2 f, as well as pyrimidine 2g led to the desired compounds 5e-5g in yields of 59-83% within 16-20 h at 25°C (Table 1, entries 5-7). Furthermore, thioether-substituted pyridines were also suitable coupling partners for the described reaction protocol. Thus, nicotinic ester **2h** underwent an efficient cross-coupling with the electron-poor arylzinc reagent 4g to afford ethyl 2-[4-(ethoxycarbonyl)phenyl]nicotinate (5h) in 69% yield (25°C, 6 h, Table 1, entry 8). Similarly, trifluoromethyl-substituted pyridine 2i gave, after cross-coupling with 4-cyanopropylzinc bromide (4h, 25°C, 16h), the alkylated heterocycle 5i in 84% yield (Table 1, entry 9). The reaction of 6,7-dimethoxy-4-(methylthio)quinazoline (2j) with the same alkylzinc reagent 4h or with the ester-substituted benzylic zinc reagent 4i yielded the functionalized quinazolines 5j and 5k in yields of 74 and 78%, respectively (Table 1, entries 10 and 11). Finally, this Pd-catalyzed cross-coupling reaction could also be performed with electron-rich N-heterocycles, such as the Boc-protected pyrrole 2k. Reaction of the latter with 4methoxyphenylzinc iodide (4a) smoothly afforded tert-butyl 2-(4-methoxyphenyl)-1*H*-pyrrole-1-carboxylate (51) in 67% yield (25 °C, 2 h, Table 1, entry 12).

Selective bis-functionalization of pyrimidines at their 2and 4-positions could be achieved. Cross-coupling occurs at the 2- or 4-position depending on the substrate. Thus, the reaction of 2-bromo-4-(methylthio)pyrimidine (**21**) with 4-methoxybenzylzinc chloride (**4j**, 1.02 equiv) proceeded rapidly in the presence of $[Pd(dba)_2]/tfp$ (dba=dibenzylideneacetone, tfp=tri(2-furyl)phosphine) within 3 h at 25°C (Scheme 3a).^[17] After direct addition of a second catalyst system (Pd(OAc)₂/S-Phos) to the reaction mixture, a second cross-coupling occurred with 4-(ethoxycarbonyl)phenylzinc iodide (**4g**, 1.50 equiv) to provide the 2,4-disubstituted pyrimidine **5m** in 67% overall yield. Alternatively, 4-iodo-2-(methylthio)pyrimidine (**2m**) was converted into the regioisomeric 2,4-disubstituted pyrimidine **5n** by first performing

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erocycles. Entry Electrophile Zinc reagent Product, Conditions, Yield^[a] OMe Znl·LiCl OMe [≿]N ⊳ N 1 MeO N MeO 'N SMe ÓМе OMe 4a 2a5a: 25°C, 5h, 83% OMe ОMе ZnBr∙LiCl ≿ N N. N 2 MeC MeC N SMe 2 a 5b: 25°C, 20 h, 43% 4b ZnH iCl MeC MeO 3 Ň 'N SMe СN ĊΝ 2 e 5c: 25°C, 4 h, 76% 4c ZnCI·LiCI MeO MeC 4 Ň 'N SMe CN CN 2 e 4d 5d: 50°C, 14 h, 71% Znl·LiCl 5 SMe 2 f 5e: 25°C, 18 h, 83% 4 e Znl·LiCl 6 SMe 2g 5 f: 25 °C, 20 h, 75 % 4e ZnCl·LiCl 7 CF₃ SMe CE/ 5g: 25°C, 16 h, 59% 2g 4 f Znl·LiCl CO₂Et CO₂Et 8 SMe CO₂Et ĊO₂Et 2h 42 5h: 25°C, 6h, 69% F_3C F_3C ZnBr∙LiCl 9 CN CN SMe 2 i 4h 5i: 25°C, 16 h, 84% CN SMe ZnBr∙LiCl MeO MeC 10 CN N MeO MeC 5j: 25°C, 5 h, 74% 2j 4h ZnCl·LiCl SMe CO₂Et MeO 11 MeO N MeO CO₂Et MeC 2j 5k: 50°C, 12 h, 78% 4 i Znl·LiC SMe 12 OMe Boc ÓМе 51: 25°C, 2 h, 67% 2k 4a

Table 1. Pd-catalyzed cross-coupling reactions of thiomethylated N-het-



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Scheme 3. Selective one-pot cross-coupling reactions of a) 2-bromo-4-(methylthio)pyrimidine (21) and b) 4-iodo-2-(methylthio)pyrimidine (2m)

a cross-coupling with 4-methoxybenzylzinc chloride (4j, 1.02 equiv) using [Pd(dba)₂]/tfp (25 °C, 10 min) and then carrying out a second cross-coupling with the arylzinc reagent 4g (1.50 equiv) in the presence of Pd(OAc)₂/S-Phos (25°C, 20 h). The pyrimidine **5n** was obtained in 80% overall yield by this one-pot double cross-coupling sequence (Scheme 3b).

Ni-catalyzed cross-coupling reactions of thiomethyl-substituted N-heterocycles with organozinc reagents: Thiomethylated N-heterocycles (1.0 equiv) could also be cross-coupled with functionalized aryl-, benzylic-, and alkylzinc reagents (1.5 equiv) under Ni⁰ catalysis. By using the inexpensive and robust system of [Ni(acac)₂] (2.5 mol%) and bis(2-diphenylphosphinophenyl)ether (DPE-Phos, 5.0 mol%), the reactions took place within 5-20 h at 25 °C in THF (Scheme 4).



Scheme 4. Ni-catalyzed cross-coupling reaction of thiomethyl-substituted N-heterocycles 2 with functionalized organozinc reagents 4.

Utilizing this protocol, the triazine 2a (1.0 equiv) underwent cross-coupling with the trifluoromethyl-substituted benzylic zinc reagent 4f (1.5 equiv) in 4 h at 25 °C, to afford 2.4-dimethoxy-6-[3-(trifluoromethyl)benzyl]-1,3,5-triazine (50) in 72% yield (Table 2, entry 1). Similarly, triazine 2a reacted with 4-fluorobenzylzinc chloride (4k) or 3-penta-

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Table 2. Ni-catalyzed cross-coupling reactions of thiomethylated N-het-erocycles.

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noylbenzylzinc chloride (41) to give the diarylmethanes 5p and 5q in yields of 89 and 82%, respectively (Table 2, entries 2 and 3). Thiomethyl-substituted pyridazine 2e underwent a straightforward cross-coupling with 4-chlorophenylzinc iodide (4m) to afford the 3,6-difunctionalized pyridazine 5r in 72% yield (Table 2, entry 4). Cross-coupling of 4methyl-2-(methylthio)pyrimidine (2n) with the electron-deficient ester-substituted arylzinc reagent 4n provided the 2arylpyrimidine 5s in 95% yield (Table 2, entry 5). The reaction of trifluoromethylated pyrimidine 2d with 2-thienylzinc iodide (40) gave the trisubstituted pyrimidine 5t in 94% yield (Table 2, entry 6). Alkylzinc reagents were also found to be well suited to these Ni-catalyzed reactions. Thus, as expected, reactions of 2-methylthiobenzoxazole (2c) with either the chloro-substituted benzylic zinc reagent 4p or 5cyano-5-methylhexylzinc bromide (4q) at 25°C provided the benzoxazole derivatives 5u and 5v in yields of 76 and 82%, respectively (Table 2, entries 7 and 8). Similarly, the crosscoupling reaction of 4-(N,N-dimethylamino)phenylzinc iodide $(4\mathbf{r})$ with benzo[d]thiazole **20** gave the 2-arylbenzothiazole 5w in 89% yield (25°C, 5h, Table 2, entry 9). Furthermore, reactions of 2-(methylthio)nicotinonitrile (2p) with the organozinc reagents 4g and 4p led to the 2,3-disubstituted pyridines 5x and 5y in yields of 67 and 69%, respectively (Table 2, entries 10 and 11). Finally, the electronrich 1-methyl-2-(methylthio)-1H-pyrrole (2b) also reacted well with 4-fluorobenzylzinc chloride $(4\mathbf{k})$ using $[Ni(acac)_2]/$ DPE-Phos to afford the functionalized pyrrole 5z in 42% yield (50°C, 18 h, Table 2, entry 12).

Large-scale cross-coupling reactions of thiomethylated Nheterocycles: To demonstrate that these cross-coupling protocols can be scaled-up, additional Pd- and Ni-catalyzed reactions were carried out on a larger scale. Thus, pyrimidine 2n (1.0 equiv) was reacted with 4-(ethoxycarbonyl)phenylzinc iodide (4g, 1.5 equiv) on a 10 mmol scale under Pd catalysis (25°C, 3 h), which gave ethyl 4-(4-methylpyrimidin-2yl)benzoate (5aa) in 91% yield (Scheme 5). This arylzinc iodide 4g also underwent a straightforward cross-coupling reaction with 2,4-dimethoxy-6-(methylthio)-1,3,5-triazine (2a) on a 20 mmol scale, which gave the triazine 5ab in 86% yield (25°C, 18 h). 2-Methylthiopyrazine (2 f) was used in a Ni-catalyzed cross-coupling reaction with 4-methoxybenzylzinc chloride (4i), which led to the functionalized diarylmethane 5ac (25°C, 15 h, 10 mmol scale, 84% yield). Finally, the arylzinc reagent 4r reacted smoothly with the thiomethyl-substituted benzothiazole 20 to give 4-(benzo[d]thiazol-2-yl)-N,N-dimethylaniline (5w) in 85% yield (25°C, 5h, 20 mmol scale).

Pd-catalyzed cross-coupling reactions of ethynyl thioethers: Pd⁰-catalyzed cross-coupling reactions of functionalized aryl-, heteroaryl-, and alkylzinc reagents (1.5 equiv) with various thiomethyl-substituted acetylenes (1.0 equiv) were achieved by using Pd(OAc)₂/DPE-Phos as the catalytic system (Scheme 6). These reactions could be carried out at 25–50 °C and were complete within 6–24 h.

[[]a] Yield of isolated, analytically pure product.



Scheme 5. Large-scale cross-couplings of thiomethyl-substituted N-heterocycles **2** with organozinc reagents **4** under Pd or Ni catalysis.



Scheme 6. Pd-catalyzed cross-coupling reaction of thiomethyl-substituted acetylenes **3** with functionalized organozinc reagents **4**.

Thus, [(methylthio)ethynyl]benzene (3c, 1.0 equiv) reacted with 4-chlorophenylzinc iodide (4m, 1.5 equiv) to afford the bis-arylacetylene 6a in 68% yield (25°C, 24 h, Table 3, entry 1). Acetylene 3c was also used for cross-coupling reactions with functionalized alkylzinc reagents. For example, 4cyanopropylzinc bromide (4h) underwent a smooth crosscoupling reaction with 3c to afford alkyne 6b in 52% yield (50°C, 16 h, Table 3, entry 2). Moreover, 1-chloro-4-[(methylthio)ethynyl]benzene (3d) reacted well with 4-chlorophenylzinc iodide (4m) to afford the symmetrically substituted bis-arylacetylene 6c in 73% yield (50°C, 12h, Table 3, entry 3). Sensitive functional groups, such as an ester group, were also tolerated in this cross-coupling reaction. Thus, reaction of the chloro-substituted phenylacetylene 3d with 4-(ethoxycarbonyl)phenylzinc iodide (4g) provided the functionalized bis-arylacetylene 6d in 53% yield (50°C, 6h, Table 3, entry 4). Similarly, 1-trifluoromethyl-4-[(methyl-

thio)ethynyl]benzene (3e) reacted smoothly with the electron-deficient zinc reagents 4m and 4g at 25 °C to give the expected products 6e and 6f in yields of 76 and 77%, respectively (Table 3, entries 5 and 6). Secondary alkylzinc reagents could also be used. Cyclohexylzinc bromide (4b) underwent a smooth cross-coupling reaction with the unsaturated thioether 3e (25°C, 18 h), from which the substituted cyclohexylethylene derivative 6g was isolated in 66% yield (Table 3, entry 7). The use of heterocyclic reagents as both the electrophile and the organozinc species proved to be compatible with these reaction conditions. Thus, the pyridine derivative 3f could be coupled with 2-thienylzinc iodide (40) to give the bis-heteroarylacetylene 6h in 72% vield (50°C, 8 h, Table 3, entry 8). Alkyl-substituted acetylenes have also been successfully used, and the cross-coupling reactions of 1-[(methylthio)ethynyl]cyclohexene (3g) with 2,4-dichlorophenylzinc iodide (4s) or 2-thienylzinc iodide (40, 25°C, 10-16 h) led to the expected cyclohexenylacetylenes 6i and 6j in yields of 69 and 65%, respectively (Table 3, entries 9 and 10). Remarkably, bis(methylthio)acetylene (3b, 1.0 equiv) could be easily cross-coupled using the catalytic system and 3.0 equivalents of the organozinc compound. This is especially useful since the corresponding dihaloacetylenes are known to be toxic, carcinogenic, and explosive, and consequently not practical substrates. Thus, the thioether-substituted alkyne 3b was functionalized with heteroarylzinc reagent 40 or 2-benzofuranylzinc iodide (4t)to afford the symmetrically substituted acetylenes 6k and 6l in yields of 67 and 74%, respectively (50°C, 3-20 h, Table 3, entries 11 and 12). Electron-rich arylzinc reagents 4u and 4v, as well as their electron-deficient counterparts, such as 4-(ethoxycarbonyl)phenylzinc iodide (4g) and 4-cyanophenylzinc iodide (4c), each reacted smoothly with bis(methylthio)acetylene (3b) to deliver the bis-arylacetylenes 6m-6p in yields of 64-79% (25-50°C, 10-16 h, Table 3, entries 13-16).

Finally, the (methylthio)ethynyl-substituted steroid **3a** was prepared starting from commercially available ethynyl estradiol (Scheme 1e) and was easily functionalized by using 4-(dimethylamino)phenylzinc iodide (**4r**, 1.5 equiv, 25 °C, 8 h) to afford the protected ethynyl estradiol derivative **6q** in 75 % yield (Scheme 7).



Scheme 7. Pd-catalyzed cross-coupling reaction of the MOM-protected ethynyl estradiol derivative 3a with 4-(*N*,*N*-dimethylamino)phenylzinc iodide (4r).

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Entry	Electrophile	Zinc reagent	Product, Conditions, Yield ^[a]	Entry	Electrophile	Zinc reagent	Product, Conditions, Yield ^[a]
1	SMe	Znl·LiCl	CI	9	SMe	Znl·LiCl Cl	
2	3c SMe	4m ZnBr·LiCl CN	6a : 25°C, 24 h, 68%	10	3g SMe	4s	6i: 25°C, 10 h, 69%
	3c	4 h	6b : 50°C, 16 h, 52 %			40	6j : 25 °C, 16 h, 65 %
3	SMe	Znl·LiCl	ciCi	11	SMe SMe	KS ZnŀLiCl	$ [s]_{s} = [s]_{s}] $
	3d	4 m	6 c : 50 °C, 12 h, 73 %		3 b	40	6k : 50°C, 20 h, 67% ^[b]
4	Sime	Znl·LiCl CO ₂ Et	CI-CO2Et	12	SMe SMe	ZnŀLiCl	
	3d	4g	6d : 50 °C, 6 h, 53 %		3b	4t	61 : 50 °C, 3 h, 74 % ^[b]
5	CF ₂	ZnI·LiCl	F ₃ C-	13	SMe SMe	ZnI-LiCl OMe	MeO MeO OMe
	3e	4 m	6e : 25°C, 12 h, 76%		3b	4u	6 m : 25°C, 10 h, 79% ^[b]
6		Znl·LiCl CO ₂ Et	F ₃ C-	14	SMe SMe	Znŀ-LiCl SMe	MeS-
	3e SMo	4g	6 f : 25 °C, 20 h, 77 %		3b	4 v	6 n : 25 °C, 14 h, 73 % ^[b]
7		ZnBr·LiCl	F ₃ C-	15	SMe SMe	Zni-LiCl CO ₂ Et	EtO ₂ C-
	ог ₃ 3е	4b	6g : 25°C, 18 h, 66 %		3b	4g	60 : 50 °C, 12 h, 67 % ^[b]
8	SMe	∠Znl·LiCl	$ \underset{N}{\overset{S}{\longrightarrow}} = \overset{S}{\overset{S}{\longrightarrow}} $	16	SMe SMe	Znl·LiCl	NC
	3 f	40	6 h : 50 °C, 8 h, 72 %		3b	4c	6 p : 50 °C, 16 h, 64 % ^[b]

Table 3. Pd-catalyzed cross-coupling reactions of thiomethylated acetylenes.

[a] Yield of isolated, analytically pure product. [b] 3.0 equivalents of the zinc reagent were used.

Comparison between Pd and Ni catalysts used in cross-coupling reactions of organozinc reagents with N-heterocyclic methyl thioethers: We have compared the reactivities of the two different catalytic systems used for the cross-coupling reactions between various functionalized zinc reagents (1.5 equiv) and thiomethyl-substituted N-heterocycles (1.0 equiv). Six different compounds were prepared using both of the described protocols. The reaction times, temperatures, and yields of the isolated products were compared, and the results of these experiments are shown in Table 4. Arylzinc reagents were investigated first. Thus, the electronrich organozinc reagent **4a** (1.5 equiv) was treated with 2thiomethylbenzothiazole **2o** (1.0 equiv) with $Pd(OAc)_2/S$ -Phos or [Ni(acac)_2]/DPE-Phos (2.5 mol%). Full

Entry	Thioether	Zinc re-	Product, Conditions, Yield ^[a]
		agent	
		Znl·LiCl	
1	SMe	OMe	S N OMe
	20	4a	7a : Pd: 25°C. 2 h. 73 %: Ni: 25°C.
			3 h, 85 %
	Me	Znl·LiCl	Me
		\langle	N
2		\square	N N
	'N' 'SMe	ĊN	L CN
	2 n	4 c	7b: Pd: 25°C, 16 h, 74%; Ni:
			25°C, 18 h, 73 %
	OMe	Znl·LiCl	Оме _
2	NKN	\bigcirc	N ^{/ ⊗} N
3		\forall	MeO
	MeO N SMe	ĊO ₂ Et	CO ₂ Et
	2 a	4 g	5ab: Pd: 25°C, 5 h, 84%; Ni:
		ZhOLLICI	25°C, 7 h, 81 %
	MeO 🔿		MeQ \sim \sim F
4		\square	
	N N SMe	\triangleleft	NN
		ŕ	
	2 e	4 k	7c : Pd: 25 °C, 5 h, 73 %; Ni: 25 °C,
		.ZnCl·LiCl	5 h, 74 %
	Me		Me
5	Ň	\square	N OMe
	Me N SMe	\searrow	Me
	_	OMe	
	2 q	4j	7d : Pd: 50 °C, 8 h, 94 %; Ni: 25 °C,
	OMe		0 II, 63 % OMe
6	NKN	ZnBr·LiCl	N
U		CN	Man CN
	2a	4 h	7e : Pd: 25°C 5 h 66% Ni · 25°C
			48 h, 68 %

Table 4. Comparison of Pd- and Ni-catalyzed cross-coupling reactions of thiomethylated N-heterocycles with organozinc reagents.

[a] Yield of isolated, analytically pure product.

conversion was achieved with the Pd catalyst after 2 h at 25°C, giving the product in 73% yield, whereas by using the Ni catalyst, the reaction took place within 3 h, delivering the cross-coupled product 7a in 85% yield (Table 4, entry 1). Electron-deficient 4-cyanophenylzinc iodide (4c) reacted with pyrimidine 2n at 25°C in the presence of either catalytic system in 16-18 h to give 2-arylpyrimidine 7b in comparable yields (73 and 74%, Table 4, entry 2). Also, the reaction of triazine 2a and 4-(ethoxycarbonyl)phenylzinc iodide (4g) led to the trisubstituted triazine 5ab after similar reaction times at 25 °C and in similar yields (Pd: 5 h, 84 % yield; Ni: 7 h, 81 % yield, Table 4, entry 3). The reaction behavior of benzylic zinc reagents was also investigated. Whereas the cross-coupling of electron-deficient 4-fluorobenzylzinc chloride (4k) with pyridazine 2e to afford heterocycle 7c gave similar results using both catalytic systems (25°C, 5 h, 73 and 74% yields, Table 4, entry 4), electron-rich reagents showed different activity. 4,6-Dimethyl-2-methylthiopyrimidine (2q) could only be functionalized with 4-methoxybenzylzinc chloride (4j) utilizing the $Pd(OAc)_2/S$ -Phos system in 8 h at 50 °C, whereby the diarylmethane 7d was obtained in 94% yield. When the same reaction was performed with $[Ni(acac)_2]/DPE$ -Phos as the catalytic system, full conversion of the pyrimidine 2q was observed after just 3 h at 25 °C, although the yield of 7d was lower at 85% (Table 4, entry 5). Furthermore, alkylzinc reagents were also studied, which showed different reactivities with the respective catalysts. Thus, triazine 2a underwent a smooth cross-coupling reaction with 4-cyanopropylzinc bromide (4h) (25 °C, 5 h, 66% yield) in the presence of $Pd(OAc)_2$, but by using [Ni-(acac)_2] it took 48 h at 25 °C to deliver a slightly higher yield of alkyltriazine 7e (68%, Table 4, entry 6).

Thus, the two catalytic systems used here, $Pd(OAc)_2/S$ -Phos and $[Ni(acac)_2]/DPE$ -Phos, perform equally well for most substrates. Although the Ni-catalyzed reactions usually require longer reaction times, especially for alkylzinc reagents, the yields of the isolated products are very close to those achieved under Pd catalysis. An advantage of the [Ni- $(acac)_2$] system is that full conversions of all electrophiles are obtained at just 25 °C, whereas in some cases the Pd- $(OAc)_2$ system requires elevated temperatures (50 °C).

Cross-coupling reactions of thiophenylated N-heterocycles: Despite the thiomethyl moiety being the most economic thioester leaving group in the cross-coupling reactions, the advantages of these methods were further stressed by employing thiophenylated N-heterocycles. These could easily be prepared by nucleophilic attack of thiophenolate on heterocyclic chlorides in analogy to Scheme 1a. The reaction of 2-(phenylthio)nicotinonitrile (2r, 1.0 equiv) with 3-(ethoxy-carbonyl)benzylzinc chloride (4i, 1.5 equiv) using Pd(OAc)₂ and S-Phos (2.5 mol%/5.0 mol%) provided the substituted pyridine 7f in 91% yield after 8 h at 25 °C (Table 5, entry 1). The same pyridine 2r underwent a smooth cross-coupling reaction with *ortho*-functionalized benzylzinc reagent 4p and the functionalized nicotinonitrile 5y was isolated in 87% yield (25 °C, 14 h, entry 2). 3-Chlorobenzylzinc

Table 5. Pd-catalyzed cross-coupling reactions of thiophenylated N-heterocycles.



[a] Yield of isolated, analytically pure product.

chloride (4p) could also be cross-coupled with phenylthioether 2s to afford the trichloro-substituted bis-arylmethane 7g in 68% yield (25°C, 16 h, Table 5, entry 3). These examples demonstrate the versatility of these reaction protocols and the scope of thioethers that are compatible with them.

Conclusion

We have developed two different, very efficient catalytic systems, Pd(OAc)₂/S-Phos and [Ni(acac)₂]/DPE-Phos, for the cross-coupling of a range of thiomethylated N-heterocycles with functionalized organozinc reagents. Aryl-, benzylic, as well as alkylzinc halides, with sensitive functional groups, such as ester, nitrile, or ketone, can be used, and all of the Ni-catalyzed, as well as most of the Pd-catalyzed reactions can be conducted at 25 °C. No copper(I) salts are necessary to achieve good yields of the cross-coupling products and these protocols are suitable for multigram reactions. A third catalytic system (Pd(OAc)₂/DPE-Phos) has been used for thiomethylated alkynes, as well as bis(thiomethyl)acetylene. In a series of comparison experiments, the different activities of the catalysts towards different combinations of heterocycles and organometallic reagents have been evaluated. The possibility of using different classes of thioethers as electrophiles has been demonstrated, along with the different synthetic approaches for their preparation. Further extensions of this methodology, as well as applications in natural product synthesis and materials chemistry, are currently underway in our laboratories.

Experimental Section

For experimental procedures, analytical data, and NMR spectra, see the Supporting Information.

Representative preparation of 5q: In a dry, argon-flushed Schlenk flask equipped with a septum and a magnetic stirring bar, compound 2a (187 mg, 1.00 mmol), [Ni(acac)₂] (6.4 mg, 2.5 mol%), and DPE-Phos (27 mg, 5.0 mol%) were dissolved in THF (1 mL). After 10 min of stirring, a solution of (3-pentanoyl)benzylzinc chloride (41) in THF (0.44 M, 3.41 mL, 1.50 mmol) was added dropwise, and the reaction mixture was stirred for 5 h at 25 °C until GC analysis of a hydrolyzed aliquot showed full conversion of the electrophile. The reaction mixture was then quenched with a saturated aqueous solution of K2CO3 (15 mL) and extracted with EtOAc (3×25 mL). The combined organic layers were dried (Na₂SO₄), and the solvent was removed in vacuo. Purification by flash chromatography (silica gel; pentane/Et₂O, 1:1) afforded the triazine 5q (258 mg, 82%) as a clear oil. ¹H NMR (CDCl₃, 600 MHz): $\delta = 7.96-7.95$ (m, 1H), 7.83-7.79 (m, 1H), 7.58-7.54 (m, 1H), 7.50-7.35 (m, 1H), 4.06 (s, 2H), 3.99 (s, 6H), 2.93 (t, J=7.43 Hz, 2H), 1.73-1.63 (m, 2H), 1.44-1.32 (m, 2H), 0.92 ppm (t, J=7.31 Hz, 3H); ¹³C NMR (CDCl₃, 150 MHz): δ=200.3, 180.9, 172.5, 137.3, 136.9, 133.9, 129.0, 128.7, 126.7, 55.1, 44.9, 38.4, 26.4, 22.4, 13.9 ppm; IR (ATR): $\tilde{v} = 2955$ (w), 1682 (m), 1546 (vs), 1500 (s), 1458 (m), 1378 (m), 1350 (vs), 1264 (m), 1231 (m), 1202 (m), 1106 (m), 1091 (m), 1069 (m), 820 (m), 731 (m), 690 cm⁻¹ (m); MS (EI, 70 eV): m/z (%): 316 (4), 315 (M⁺, 15), 274 (7), 273 (40), 259 (15), 258 (100), 245 (6), 231 (11), 230 (4), 158 (6); HRMS (EI): m/z: calcd for C₁₇H₂₁N₃O₃: 315.1583; found: 315.1577.

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- a) E. Wenkert, T. W. Ferreira, E. L. Michelotti, J. Chem. Soc. Chem. Commun. 1979, 637; b) H. Okamura, M. Miura, H. Takei, Tetrahedron Lett. 1979, 20, 43; c) H. Takei, M. Miura, H. Sugimura, H. Okamura, Chem. Lett. 1979, 1447; d) E. Wenkert, T. W. Ferreira, J. Chem. Soc. Chem. Commun. 1982, 840; e) E. Wenkert, M. E. Shepard, A. T. McPhail, J. Chem. Soc. Chem. Commun. 1986, 1390; f) E. Wenkert, D. Chianelli, J. Chem. Soc. Chem. Commun. 1991, 627; g) for selected reviews, see: h) S. R. Dubbaka, P. Vogel, Angew. Chem. 2005, 117, 7848; Angew. Chem. Int. Ed. 2005, 44, 7674; i) H. Prokopcová, C. O. Kappe, Angew. Chem. 2009, 121, 2312; Angew. Chem. Int. Ed. 2009, 48, 2276.
- [2] a) H. Tokuyama, S. Yokoshima, T. Yamashita, T. Fukuyama, *Tetrahedron Lett.* 1998, 39, 3189; b) L. S. Liebeskind, J. Srogl, J. Am. Chem. Soc. 2000, 122, 11260; c) J. Srogl, G. D. Allred, L. S. Liebeskind, J. Am. Chem. Soc. 1997, 119, 12376; d) C. Savarin, J. Srogl, L. S. Liebeskind, Org. Lett. 2000, 2, 3229; e) J. M. Villalobos, J. Srogl, L. S. Liebeskind, J. Am. Chem. Soc. 2007, 129, 15734; f) L. S. Liebeskind, H. Yang, H. Li, Angew. Chem. 2009, 121, 1445; Angew. Chem. Int. Ed. 2009, 48, 1417; g) Y. Yu, L. S. Liebeskind, J. Org. Chem. 2004, 69, 3554; h) C. L. Kusturin, L. S. Liebeskind, W. L. Neumann, Org. Lett. 2002, 4, 983.
- [3] R. Wittenberg, J. Srogl, M. Egi, L. S. Liebeskind, Org. Lett. 2003, 5, 3033.
- [4] B. W. Fausett, L. S. Liebeskind, J. Org. Chem. 2005, 70, 4851.
- [5] a) L. S. Liebeskind, J. Srogl, Org. Lett. 2002, 4, 979; b) S. Oumouch, M. Bourotte, M. Schmitt, J.-J. Bourguignon, Synthesis 2005, 25; c) A. Aguilar-Aguilar, E. Pena-Cabrera, Org. Lett. 2007, 9, 4163; d) A. Lengar, C. O. Kappe, Org. Lett. 2004, 6, 771; e) H. Prokopcová, C. O. Kappe, J. Org. Chem. 2007, 72, 4440; f) W. van Rossom, W. Maes, L. Kishore, M. Ovaere, L. van Meervelt, W. Dehaen, Org. Lett. 2008, 10, 585; g) K. Itami, D. Yamazaki, J. Yoshida, J. Am. Chem. Soc. 2004, 126, 15396; h) C. Kusturin, L. S. Liebeskind, H. Rahman, K. Sample, B. Schweitzer, J. Srogl, W. L. Neumann, Org. Lett. 2003, 5, 4349; i) M. Egi, L. S. Liebeskind, Org. Lett. 2001, 3, 91.
- [6] a) P. Knochel, N. Millot, A. L. Rodriguez, C. E. Tucker, Org. React. 2001, 58, 417; b) P. Knochel, H. Leuser, L.-Z. Gong, S. Perrone, F. F. Kneisel, Handbook of Functionalized Organometallics (Ed.: P. Knochel), Wiley-VCH, Weinheim, 2005; c) P. Knochel, J. J. Almena Perea, P. Jones, Tetrahedron 1998, 54, 8275; d) P. Knochel, P. Jones (Eds.), Organozinc Reagents, Oxford University Press, New York, 1999; e) P. Knochel, F. Langer, A. longeau, M. Rottländer, T. Stüdemann, Chem. Ber. 1997, 130, 1021; f) P. Knochel, S. Vettel, C. Eisenberg, Appl. Organomet. Chem. 1995, 9, 175.
- [7] a) M. E. Angiolelli, A. L. Casalnuovo, T. P. Selby, Synlett 2000, 905;
 b) K. Lee, C. M. Counceller, J. P. Stambuli, Org. Lett. 2009, 11, 1457;
 c) J. Srogl, W. Liu, D. Marshall, L. S. Liebeskind, J. Am. Chem. Soc. 1999, 121, 9449; for Pd-catalyzed reactions of organozinc reagents with thioimidates, see: d) I. Ghosh, P. A. Jacobi, J. Org. Chem. 2002, 67, 9304; e) W. P. Roberts, I. Ghosh, P. A. Jacobi, Can. J. Chem. 2004, 82, 279; f) D. M. Mans, W. H. Pearson, J. Org. Chem. 2004, 69, 6419; for cross-coupling reactions of vinyl sulfides with benzylic and allylic zinc reagents under nickel catalysis, see: g) Y. Baba, A. Toshimitsu, S. Matsubara, Synlett 2008, 2061; for a cobalt-catalyzed activation of methylthio-substituted N-heterocycles, see: h) J.-M. Begouin, M. Rivard, C. Gosmini, Chem. Commun. 2010, 46, 5972.
- [8] a) A. Metzger, L. Melzig, C. Despotopoulou, P. Knochel, Org. Lett.
 2009, 11, 4228; b) L. Melzig, A. Metzger, P. Knochel, J. Org. Chem.
 2010, 75, 2131; c) A. Metzger, L. Melzig, P. Knochel, Synthesis 2010, 2853.
- [9] L. Melzig, J. Stemper, P. Knochel, Synthesis 2010, 2085.

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- [10] J. M. Brittain, R. A. Jones, J. S. Arques, T. A. Saliente, Synth. Commun. 1982, 12, 231.
- [11] L. Bethge, D. V. Jarikote, O. Seitz, *Bioorg. Med. Chem.* 2008, 16, 114.
- [12] A. Rodriguez, J. Arturo Garcia-Vazquez, A. Sousa-Pedrares, J. Romero, A. Sousa, *Inorg. Chem. Commun.* 2003, 6, 619.
- [13] The MOM-protection of ethynyl estradiol was performed according to: F. Wüst, K. E. Carlson, J. A. Katzenellenbogen, H. Spies, B. Johannsen, *Steroids* 1998, 63, 665.
- [14] A. Krasovskiy, P. Knochel, Angew. Chem. 2004, 116, 3396; Angew. Chem. Int. Ed. 2004, 43, 3333.
- [15] H. D. Verkruijsse, L. Brandsma, Synthesis 1991, 818.
- [16] a) R. A. Altman, S. L. Buchwald, *Nat. Protoc.* 2007, 2, 3115; b) T. E. Barder, S. L. Buchwald, *J. Am. Chem. Soc.* 2007, *129*, 5096; c) J. E. Milne, S. L. Buchwald, *J. Am. Chem. Soc.* 2004, *126*, 13028.
- [17] This catalytic system proved to be the best for achieving a selective coupling with the carbon-halogen bonds of pyrimidines 21 and 2m. When using Pd(OAc)₂/S-Phos for the first step of Scheme 3, a mixture of the two possible cross-coupling products was obtained.

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