

# Highly Diastereoselective Synthesis of Trifluoromethyl Indolines by Interceptive Benzylic Decarboxylative Cycloaddition of Nonvinyl, Trifluoromethyl Benzoxazinanones with Sulfur Ylides under Palladium Catalysis

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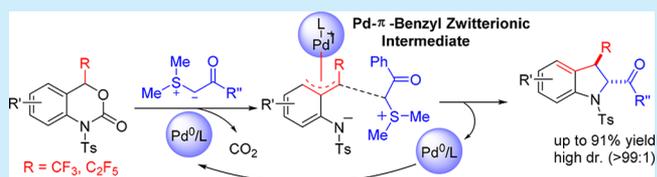
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**S** Supporting Information

**ABSTRACT:** A highly diastereoselective synthesis of trifluoromethyl-substituted indolines under palladium catalysis is disclosed. The reaction proceeds by interceptive decarboxylative benzylic cycloaddition (IDBC) of nonvinyl, trifluoromethyl benzoxazinanones with sulfur ylides. The palladium– $\pi$ -benzyl zwitterionic intermediates are suggested for this transformation, and this would be the first example of an IDBC reaction.

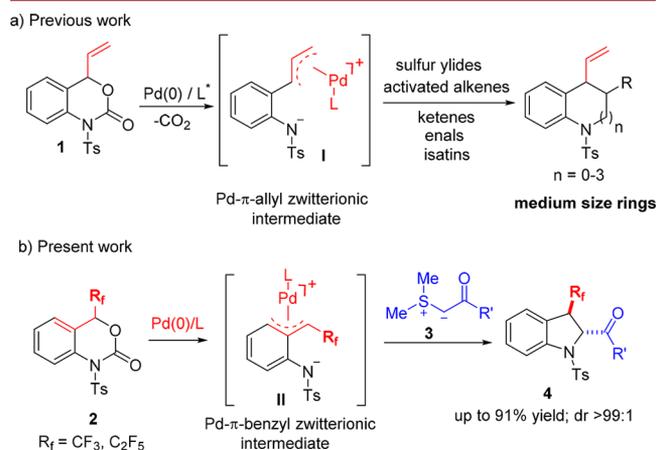


Heterocyclic compounds containing a fluorine (F) or trifluoromethyl (CF<sub>3</sub>) group on their core moieties are well studied in the field of pharmaceuticals and agrochemicals, and some of them are successfully distributed in the drug market.<sup>1</sup> Among them, fluorinated indoles and oxindoles have received a great deal of attention in the literature because of the exceptional influence of the fluorine atom in terms of chemical and physiological stability.<sup>2</sup> On the other hand, the CF<sub>3</sub>-substituted indolines are not developed, despite the attractive biological properties of indolines, which are well distributed in natural products and pharmaceuticals.<sup>3</sup> We report herein the highly diastereoselective synthesis of medicinally attractive CF<sub>3</sub>-substituted indolines by interceptive decarboxylative benzylic cycloaddition (IDBC) via a palladium– $\pi$ -benzyl zwitterionic intermediate.

The interceptive decarboxylative allylic cycloaddition (IDAC) reaction of cyclic allylic carbamates under palladium catalysis<sup>4</sup> is a sophisticated extension of the Tsuji–Trost decarboxylative allylation (DA) reaction.<sup>5</sup> A zwitterionic  $\pi$ -allyl palladium complex is a reactive intermediate that is then intercepted by a sufficiently reactive substrate, providing a medium-sized ring skeleton.<sup>4</sup> Among the cyclic allylic carbamates in this transformation, vinyl benzoxazinones **1** are efficient multifarious synthons for the IDAC reaction to afford biologically pertinent heterocyclic molecules.<sup>6</sup> The palladium– $\pi$ -allyl zwitterionic intermediate **I** generated from **1** was disclosed by Tunge et al. and successfully trapped with benzylidene malononitriles, producing tetrahydroquinolines via the [4 + 2] IDAC reaction.<sup>6a,b</sup> This seminal work stimulated chemists into further extensions of the IDAC reaction of **1** with other interceptors such

as sulfur ylides **3**,<sup>6c–e</sup> activated alkenes,<sup>6a,b,f</sup> enals,<sup>6g,h</sup> ketenes,<sup>6i</sup> and isatins<sup>6j</sup> under palladium or other metal catalysis (Figure 1a).

Although the IDAC reactions of **1** have significant potential with many reactive interceptive substrates, a diversity of benzoxazinone **1** are highly limited to “vinyl”-substituted compounds because of the vital role of the vinyl group in generating a palladium-stabilized  $\pi$ -allyl complex. Xiao expanded the [4 + 1] interceptive decarboxylative cycloaddition reaction to



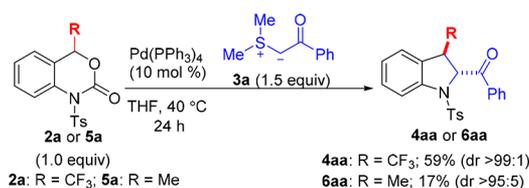
**Figure 1.** (a) IDAC reaction via zwitterionic  $\pi$ -allyl palladium complex (previous work); (b) IDBC reaction via zwitterionic  $\pi$ -benzyl palladium complex (this work).

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the propargylic analogues of **1** with sulfur ylides **3** under copper catalysts, furnishing a series of alkynyl indolines.<sup>6c</sup> However, the fundamental issue of the use of a nonvinyl substrate remains a great challenge.<sup>6</sup> In this context, we envisaged the generation of palladium- $\pi$ -benzyl zwitterionic intermediate **II**, an isoelectronic variant of Tsuji–Trost allylic substitution, could be equally possible, but in fact, there is no example of this type of benzylic cycloaddition. The reaction involving  $\pi$ -benzyl–palladium complex is rather limited to the acyclic reactions.<sup>7</sup> The low reactivity of benzylic substitution of this type is explained by the aromaticity of the benzene ring, while allylic compounds are inherently reactive. Recently, a Suzuki–Miyaura cross-coupling reaction with secondary  $\alpha$ -(trifluoromethyl)benzyl tosylates was reported by Tredwell et al.<sup>7k</sup> Inspired by this result, we assumed that the CF<sub>3</sub> group at the benzylic position (C4) induces a strong negative inductive effect on the neighboring carbon, allowing the benzylic carbon to become electron deficient. Thus, a highly electrophilic palladium- $\pi$ -benzyl zwitterionic intermediate **II** should be generated by oxidative addition of Pd(0) to amenable IDBC reaction to produce biologically important CF<sub>3</sub>-indolines (Figure 1b). These results are the first example of a palladium-catalyzed IDBC reaction using nonvinyl benzoxazinones **2**. The CF<sub>3</sub> group plays an important role in this transformation.

First, we designed the previously unknown CF<sub>3</sub>-benzoxazinone **2a** and investigated its suitability for IDBC reactions.<sup>4,5</sup> We rapidly discovered that CF<sub>3</sub>-benzoxazinone **2a** reacted with sulfur ylide **3a** under palladium catalysis to provide the desired [4 + 1] IDBC product **4aa** in moderate yield. This reaction performed with Pd(PPh<sub>3</sub>)<sub>4</sub> (10 mol %) at 40 °C in THF gave CF<sub>3</sub>-indoline **4aa** in 59% yield with high diastereoselectivity (dr >99:1). CF<sub>3</sub> effect is very impressive as the methyl-substituted benzoxazinone **5a** gave Me-indoline **6aa** only in 17% yield (dr >95:5) (Scheme 1). Encouraged by this result, we optimized the

**Scheme 1. Initial Attempts for IDBC Reaction of 2a or 5a with 3a Using Pd(PPh<sub>3</sub>)<sub>4</sub> Catalyst**



reaction of **2a** with **3a** by screening different phosphine ligands (dppe, dppf, PET<sub>3</sub>, tBuXPhos, PCy<sub>3</sub>, and Binap) and solvents in the presence of Pd(0) catalyst (Table 1 and Tables S1 and S2 in Supporting Information). Gratifyingly, **4aa** was obtained in excellent yield (98%) and very high diastereocontrol (dr = 99:1) using PCy<sub>3</sub> (10 mol %) and Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (5 mol %) at 40 °C in THF for 1 h (run 1, Table 1). The amount of Pd catalyst could be reduced to 1 mol % and 2.5 mol %, resulting in the required product with moderate yields (Table S2). No reaction occurred in the absence of the palladium catalyst (runs 3 and 4), and the absence of PCy<sub>3</sub> gave **4aa** in low yield after longer reaction time (run 2). Increasing the reaction temperature from rt to 40 °C accelerated the reaction significantly from 24 to 1 h (runs 1 and 5). The Me-substituted benzoxazinone **5a** underwent IDBC and gave Me-indoline **5aa** in 42% yield under the same reaction conditions optimized for **2a**, albeit only after extended reaction time (run 6: 42%, 24 h vs run 1: 98%, 1 h).

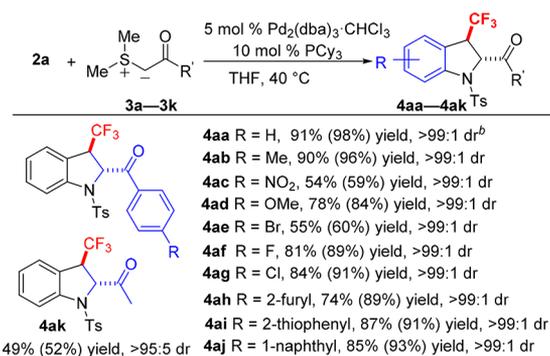
**Table 1. Effects of Pd, Ligand, and Temperature<sup>a</sup>**

run	Pd	PCy <sub>3</sub>	t (°C)	time (h)	yield <sup>b</sup> (%)	dr <sup>c</sup>
1	yes	yes	40	1	98	>99:1
2	yes	no	40	24	24	>99:1
3	no	yes	40	24	NR	
4	no	no	60	24	NR	
5	yes	yes	rt	24	91	>99:1
6 <sup>d</sup>	yes	yes	40	24	42 <sup>e</sup>	>95:5

<sup>a</sup>Experiments were performed with **2a** (0.15 mmol), **3a** (0.225 mmol), Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (0.0075 mmol or absence), PCy<sub>3</sub> (0.015 mmol or absence) in 1.0 mL of dry THF. <sup>b</sup>Yields are <sup>19</sup>F NMR yields with internal standard PhCF<sub>3</sub>. <sup>c</sup>Diastereomeric ratio was determined by <sup>19</sup>F NMR analysis of the reaction mixture. <sup>d</sup>Compound **5a** was used instead of **2a**. <sup>e</sup>Yield of **6aa**.

With the optimized reaction conditions in hand, the generality of the IDBC reaction was examined by reacting CF<sub>3</sub>-benzoxazinone **2a** with sulfur ylides **3a–k**. The results are summarized in Scheme 2. Sulfur ylides **3b,d**, which have electron-

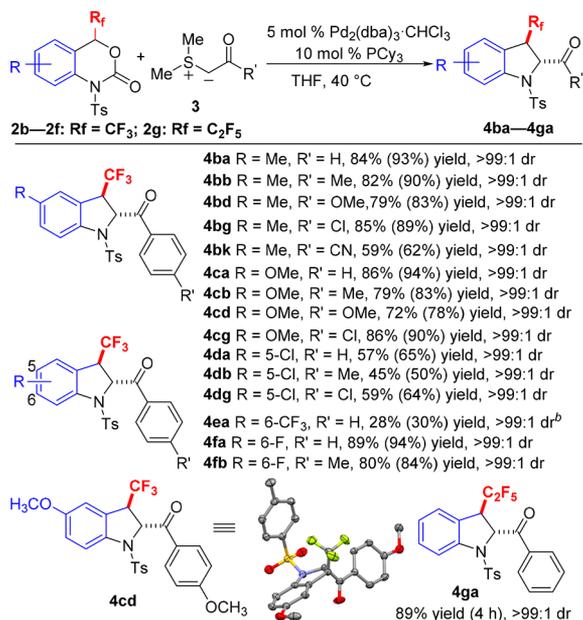
**Scheme 2. Scope of Sulfur Ylides 3<sup>a</sup>**



<sup>a</sup>Experiments were performed with **2a** (0.15 mmol), **3a–k** (0.225 mmol), Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> (0.0075 mmol), PCy<sub>3</sub> (0.015 mmol) in 1.0 mL of dry THF with stirring at 40 °C for 1–8 h. Yields and <sup>19</sup>F NMR yields with internal standard PhCF<sub>3</sub> are shown in parentheses. The dr was determined by <sup>19</sup>F NMR analysis of the reaction mixture. <sup>b</sup>Reaction performed with 1 mmol scale of **2a** and 1.5 mmol of **3a** (see the Supporting Information for details).

donating groups (Me and OMe), afforded the desired CF<sub>3</sub>-indolines in excellent yields (90% of **4ab** and 78% of **4ad**), whereas the sulfur ylide **3c** with an electron-withdrawing group (NO<sub>2</sub>) resulted in moderate yield (54% of **4ac**). Halogen-substituted sulfur ylides **3e–g** also underwent IDBC with moderate to very good yields (55% of **4ae**, 81% of **4af**, and 84% of **4ag**). The heteroaryl systems **3h,i** (2-furyl and 2-thiophenyl), extended  $\pi$ -conjugate naphthalene-derived sulfur ylide **3j**, and aliphatic acyl-stabilized ylide **3k** are also well tolerated to afford the CF<sub>3</sub>-indolines **4ah** (74%), **4ai** (87%), **4aj** (85%), and **4ak** (49%). All of these reactions proceeded with complete control over diastereoselectivity (Scheme 2).

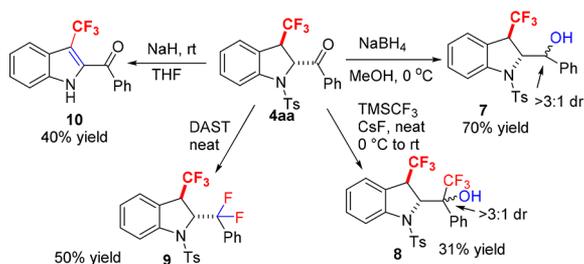
Next, a range of differently substituted CF<sub>3</sub>-benzoxazinones **2b–f** was examined to delineate the generality of the IDBC reaction (Scheme 3). Various substituents on the benzene ring with electronically dissimilar groups at different positions were

Scheme 3. Scope of Benzoxazinanones 2<sup>a</sup>

<sup>a</sup>Unless otherwise noted, the reaction was performed with 0.15 mmol scale of **2** as mentioned in Scheme 2. The dr values and yields were calculated as mentioned in Scheme 2. <sup>b</sup>10 mol % of Pd<sub>2</sub>(dba)<sub>3</sub>·CHCl<sub>3</sub> and 20 mol % of PCy<sub>3</sub> were used.

well tolerated in moderate to good yields with high diastereoselectivities. The halogen-substituted CF<sub>3</sub>-benzoxazinanones **2d** and **2f** (Cl and F) produced CF<sub>3</sub>-indolines **4** in moderate to good yields (57% of **4da** and 89% of **4fa**). Substrates **2b,c** with electron-donating groups on the benzene ring such as Me and OMe were successfully transformed into corresponding CF<sub>3</sub>-indolines in excellent yields (84% of **4ba** and 86% of **4ca**) with high dr. An electron-withdrawing CF<sub>3</sub> group on the benzene ring of **2e** reduced the yield of CF<sub>3</sub>-indoline **4ea** by 28%. The *trans*-configuration of **4** was confirmed by the X-ray crystallographic analysis of **4cd** (CCDC 1556229). The pentafluoroethyl (C<sub>2</sub>F<sub>5</sub>)-benzoxazinanone **2g** also underwent the IDBC reaction smoothly to furnish C<sub>2</sub>F<sub>5</sub>-indoline **4ga** with excellent yield of 89% and high selectivity (>99:1) in 4 h (Scheme 3).

To demonstrate the synthetic utility of CF<sub>3</sub>-indolines **4**, several reactions were carried out, as shown in Scheme 4. CF<sub>3</sub>-

Scheme 4. Derivatization of CF<sub>3</sub>-Indoline 4aa

indoline **4aa** was successfully transformed to alcohols **7** and **8** having three consecutive stereogenic centers by reduction with NaBH<sub>4</sub> (70% yield) and by reaction with Me<sub>3</sub>SiCF<sub>3</sub> (31% yield), respectively (diastereoselectivities are >3:1). Indoline **4aa** was also converted to difluorinated derivative **9** with 50% yield by the reaction with DAST. Furthermore, **4aa** was converted into CF<sub>3</sub>-

substituted indole **10** with 40% yield in the presence of NaH in THF (Scheme 4).

A plausible catalytic cycle of the Pd-catalyzed IDBC reaction of **2a** with **3a** to **4aa** is depicted in Figure 2, based on the reported

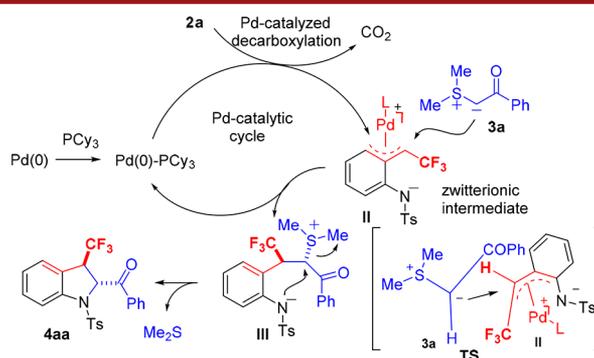


Figure 2. Proposed mechanism including stepwise formation of zwitterionic intermediate II.

mechanism of the IDBC reaction of **2** with **3a** via the Pd- $\pi$ -benzyl zwitterionic intermediate.<sup>6a–c</sup> The catalytic cycle is first initiated by the oxidative addition of Pd(0) with **2a** followed by decarboxylation, which leads to the benzyl-Pd(II) zwitterionic intermediate II. Intermediate II is very electrophilic due to the CF<sub>3</sub> group, and subsequently, a new C–C bond is formed by the addition of sulfur ylide **3a** to complex II, diastereoselectively resulting in the zwitterionic complex III with the reductive elimination of the Pd(0) catalyst. Finally, the zwitterionic complex III undergoes an intramolecular S<sub>N</sub>2-type *anti*-periplanar reaction from nitrogen to the stereogenic carbon center attached to the sulfonium moiety in III, furnishing the corresponding *trans*-CF<sub>3</sub>-substituted indoline **4aa** with the release of Me<sub>2</sub>S. The highly diastereoselective formation of the *trans*-isomer of **4aa** could be explained as follows. The zwitterionic complex III can be formed via transition state TS, due to the favorable steric repulsion between CF<sub>3</sub> and SME<sub>2</sub> moieties (Figure 2).

In conclusion, we disclose a novel method to synthesize diverse substituted trifluoromethyl indolines **4** from **2** under palladium catalysis in a highly diastereoselective fashion. In this transformation, a CF<sub>3</sub> substitution at the benzylic position of benzoxazinanones **2** is highly effective. Pentafluoroethyl (C<sub>2</sub>F<sub>5</sub>) substitution was also suitable under the same reaction conditions, whereas a nonfluorinated Me group gave a lower yield. The electron-withdrawing nature of the perfluoroalkyl group played a crucial role in this transformation. A series of CF<sub>3</sub>-benzoxazinanones **2** was smoothly reacted with sulfur ylides **3** under palladium catalysis in the presence of tricyclohexylphosphine (PCy<sub>3</sub>) to provide pharmaceutically attractive CF<sub>3</sub>-substituted indolines **4** in high to excellent yields up to 91% with high diastereoselectivities greater than 99:1 dr. This type of IDAC reaction is believed to use vinyl-substituted substrates, thus, these results are the first example of a palladium-catalyzed nonvinyl, IDBC reaction, and CF<sub>3</sub> is highly advantageous for the efficient transformation. Moreover, the CF<sub>3</sub>-indolines **4** are prospective synthetic building blocks for biologically attractive drug candidates.<sup>1</sup> Extension of this work is currently in progress, including isolation of Pd- $\pi$ -benzyl zwitterionic intermediates and enantioselective variants of the reaction (for preliminary studies of the asymmetric reaction, see Table S3).

## ■ ASSOCIATED CONTENT

### Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.orglett.8b00237.

Syntheses and NMR spectra (PDF)

### Accession Codes

CCDC 1556229 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via [www.ccdc.cam.ac.uk/data\\_request/cif](http://www.ccdc.cam.ac.uk/data_request/cif), or by emailing [data\\_request@ccdc.cam.ac.uk](mailto:data_request@ccdc.cam.ac.uk), or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

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### Notes

The authors declare no competing financial interest.

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## ■ REFERENCES

- (1) (a) Petrov, V. A. *Fluorinated Heterocyclic Compounds: Synthesis, Chemistry and Applications*; John Wiley & Sons, Inc.: Hoboken, NJ, 2009. (b) Nenajdenko, V. *Fluorine in Heterocyclic Chemistry*; Springer: Cham, Switzerland, 2014; Vol. 1. (c) Nenajdenko, V. *Fluorine in Heterocyclic Chemistry*; Springer: Cham, Switzerland, 2014; Vol. 2. (d) Wang, J.; Sanchez-Rosello, M.; Acena, J. L.; del Pozo, C.; Sorochinsky, A. E.; Fustero, S.; Soloshonok, V. A.; Liu, H. *Chem. Rev.* **2014**, *114*, 2432–2506. (e) Xu, X.-H.; Matsuzaki, K.; Shibata, N. *Chem. Rev.* **2015**, *115*, 731–764. (f) Li, S.; Ma, J.-A. *Chem. Soc. Rev.* **2015**, *44*, 7439–7448.
- (2) (a) Dolensky, B.; Nam, G.; Deng, W.-P.; Narayanan, J.; Fan, J.; Kirk, K. L. *J. Fluorine Chem.* **2004**, *125*, 501–508. (b) Faivre, S.; Demetri, G.; Sargent, W.; Raymond, E. *Nat. Rev. Drug Discovery* **2007**, *6*, 734–745. (c) Laderoute, K. R.; Calaoagan, J. M.; Madrid, P. B.; Klon, A. E.; Ehrlich, P. J. *Cancer Biol. Ther.* **2010**, *10*, 68–76. (d) Zhang, C.; Zhuang, D.-M.; Li, J.; Chen, S.-Y.; Du, X.-L.; Wang, J.-Y.; Li, J.-Y.; Jiang, B.; Yao, J.-H. *Org. Biomol. Chem.* **2013**, *11*, 5621–5633.
- (3) (a) Patil, S. A.; Patil, R.; Miller, D. D. *Future Med. Chem.* **2012**, *4*, 2085–2115. (b) Furman, S.; Nissim-Bardugo, E.; Zeeli, S.; Weitman, M.; Nudelman, A.; Finkin-Groner, E.; Moradov, D.; Shifrin, H.; Schorer-Apelbaum, D.; Weinstock, M. *Bioorg. Med. Chem. Lett.* **2014**, *24*, 2283–2287. (c) Bisai, V.; Suneja, A.; Singh, V. K. *Angew. Chem., Int. Ed.* **2014**, *53*, 10737–10741. (d) Patil, R.; Patil, S. A.; Beaman, K. D.; Patil, S. A. *Future Med. Chem.* **2016**, *8*, 1291–1316.
- (4) (a) Weaver, J. D.; Recio, A., III; Grenning, A. J.; Tunge, J. A. *Chem. Rev.* **2011**, *111*, 1846–1913. (b) Shintani, R.; Park, S.; Hayashi, T. *J. Am. Chem. Soc.* **2007**, *129*, 14866–14867. (c) Shintani, R.; Park, S.; Shirozu, F.; Murakami, M.; Hayashi, T. *J. Am. Chem. Soc.* **2008**, *130*, 16174–16175. (d) Yeagley, A. A.; Lowder, M. A.; Chruma, J. J. *Org. Lett.* **2009**, *11*, 4022–4025. (e) Streuff, J.; White, D. E.; Virgil, S. C.; Stoltz, B. M. *Nat. Chem.* **2010**, *2*, 192–196. (f) Shintani, R. *Bull. Chem. Soc. Jpn.* **2012**, *85*, 931–939. (g) Schmitt, M.; Grenning, A. J.; Tunge, J. A. *Tetrahedron*

*Lett.* **2012**, *53*, 4494–4497. (h) Baiju, T. V.; Joseph, N.; Ajit, J.; Prakash, P.; Radhakrishnan, K. V.; Varughese, S.; Yamamoto, Y. *Synlett* **2014**, *25*, 1246–1257. (i) Maji, T.; Tunge, J. A. *Org. Lett.* **2015**, *17*, 4766–4769. (j) Khan, A.; Zhang, Y. J. *Synlett* **2015**, *26*, 853–860. (k) Yang, L.; Khan, A.; Zheng, R.; Jin, L. Y.; Zhang, Y. J. *Org. Lett.* **2015**, *17*, 6230–6233. (l) Khan, A.; Xing, J.; Zhao, J.; Kan, Y.; Zhang, W.; Zhang, Y. J. *Chem. - Eur. J.* **2015**, *21*, 120–124. (m) Allen, B. D. W.; Lakeland, C. P.; Harrity, J. P. A. *Chem. - Eur. J.* **2017**, *23*, 13830–13857. (n) De, N.; Yoo, E. J. *ACS Catal.* **2018**, *8*, 48–58.

(5) Selective examples of decarboxylative allylation: (a) Tsuji, J. *Transition Metal Reagents and Catalysts: Innovations in Organic Synthesis*; John Wiley & Sons: Chichester, U.K., 2000. (b) Trost, B. M.; Lee, C. *Asymmetric Allylic Alkylation Reactions. In Catalytic Asymmetric Synthesis*, 2nd ed.; Ojima, I., Ed.; Wiley-VCH: New York, 2000; pp 593–649. (c) Trost, B. M.; Van Vranken, D. L. *Chem. Rev.* **1996**, *96*, 395–422. (d) Trost, B. M.; Crawley, M. L. *Chem. Rev.* **2003**, *103*, 2921–2944. (e) Mohr, J. T.; Stoltz, B. M. *Chem. - Asian J.* **2007**, *2*, 1476–1491. (f) Guerrero Rios, I.; Rosas-Hernandez, A.; Martin, E. *Molecules* **2011**, *16*, 970. (g) Shibata, N.; Suzuki, S.; Furukawa, T.; Kawai, H.; Tokunaga, E.; Yuan, Z.; Cahard, D. *Adv. Synth. Catal.* **2011**, *353*, 2037–2041. (h) Shibata, N.; Fukushi, K.; Furukawa, T.; Suzuki, S.; Tokunaga, E.; Cahard, D. *Org. Lett.* **2012**, *14*, 5366–5369. (i) Liu, Y.; Han, S.-J.; Liu, W.-B.; Stoltz, B. M. *Acc. Chem. Res.* **2015**, *48*, 740–751. (j) Maeno, M.; Kondo, H.; Tokunaga, E.; Shibata, N. *RSC Adv.* **2016**, *6*, 85058–85062.

(6) (a) Wang, C.; Tunge, J. A. *J. Am. Chem. Soc.* **2008**, *130*, 8118–8119. (b) Wang, C.; Pahadi, N.; Tunge, J. A. *Tetrahedron* **2009**, *65*, 5102–5109. (c) Li, T.-R.; Tan, F.; Lu, L.-Q.; Wei, Y.; Wang, Y.-N.; Liu, Y.-Y.; Yang, Q.-Q.; Chen, J.-R.; Shi, D.-Q.; Xiao, W.-J. *Nat. Commun.* **2014**, *5*, 5500. (d) Wang, Q.; Qi, X.; Lu, L.-Q.; Li, T.-R.; Yuan, Z.-G.; Zhang, K.; Li, B.-J.; Lan, Y.; Xiao, W. J. *Angew. Chem., Int. Ed.* **2016**, *55*, 2840–2844. (e) Wang, Q.; Li, T. R.; Lu, L.-Q.; Li, M.-M.; Zhang, K.; Xiao, W.-J. *J. Am. Chem. Soc.* **2016**, *138*, 8360–8363. (f) Wei, Y.; Lu, L.-Q.; Li, T.-R.; Feng, B.; Wang, Q.; Xiao, W.-J.; Alper, H. *Angew. Chem., Int. Ed.* **2016**, *55*, 2200–2204. (g) Guo, C.; Fleige, M.; Janssen-Muller, D.; Daniliuc, C. G.; Glorius, F. *J. Am. Chem. Soc.* **2016**, *138*, 7840–7843. (h) Leth, L. A.; Glaus, F.; Meazza, M.; Fu, L.; Thogersen, M. K.; Bitsch, E. A.; Jorgensen, K. A. *Angew. Chem., Int. Ed.* **2016**, *55*, 15272–15276. (i) Li, M.-M.; Wei, Y.; Liu, J.; Chen, H.-W.; Lu, L.-Q.; Xiao, W.-J. *J. Am. Chem. Soc.* **2017**, *139*, 14707–14713. (j) Mei, G.-J.; Bian, C.-Y.; Li, G.-H.; Xu, S.-L.; Zheng, W.-Q.; Shi, F. *Org. Lett.* **2017**, *19*, 3219–3222. (k) Mei, G.-J.; Li, D.; Zhou, G.-X.; Shi, Q.; Cao, Z.; Shi, F. *Chem. Commun.* **2017**, *53*, 10030–10033.

(7) Selective examples for palladium- $\pi$ -benzyl complexes: (a) Lin, Y.-S.; Yamamoto, A. *Organometallics* **1998**, *17*, 3466–3478. (b) Lin, Y.-S.; Yamamoto, A. *Bull. Chem. Soc. Jpn.* **1998**, *71*, 723–734. (c) Kawatsura, M.; Hartwig, J. F. *J. Am. Chem. Soc.* **2000**, *122*, 9546–9547. (d) Kuwano, R.; Kondo, Y.; Matsuyama, Y. *J. Am. Chem. Soc.* **2003**, *125*, 12104–12105. (e) Kuwano, R.; Kondo, Y. *Org. Lett.* **2004**, *6*, 3545–3547. (f) Johns, A. M.; Utsunomiya, M.; Incarvito, C. D.; Hartwig, J. F. *J. Am. Chem. Soc.* **2006**, *128*, 1828–1839. (g) Johns, A. M.; Tye, J. W.; Hartwig, J. F. *J. Am. Chem. Soc.* **2006**, *128*, 16010–16011. (h) Urkalan, K. B.; Sigman, M. S. *Angew. Chem., Int. Ed.* **2009**, *48*, 3146–3149. (i) Blessley, G.; Holden, P.; Walker, M.; Brown, J. M.; Gouverneur, V. *Org. Lett.* **2012**, *14*, 2754–2757. (j) Miró, J.; del Pozo, C.; Toste, F. D.; Fustero, S. *Angew. Chem., Int. Ed.* **2016**, *55*, 9045–9049. (k) Brambilla, M.; Tredwell, M. *Angew. Chem., Int. Ed.* **2017**, *56*, 11981–11985.

(8) For reviews on the cyclization reactions of sulfur ylides, see: (a) McGarrigle, E. M.; Myers, E. L.; Illa, O.; Shaw, M. A.; Riches, S. L.; Aggarwal, V. K. *Chem. Rev.* **2007**, *107*, 5841–5883. (b) Sun, X.-L.; Tang, Y. *Acc. Chem. Res.* **2008**, *41*, 937–948. (c) Lu, L.-Q.; Li, T.-R.; Wang, Q.; Xiao, W.-J. *Chem. Soc. Rev.* **2017**, *46*, 4135–4149.