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LETTERS TO THE EDITOR

Reaction of Catechol with α-Aminoacetals. Synthesis of New Polyphenols

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The basic method for the synthesis of calix[4] resorcinarenes is a condensation of resorcinol and their derivatives with aliphatic and aromatic aldehydes [1–3]. Recently it was shown that the condensation of resorcinol and 2-methylresorcinol with dimethylacetal of α -methylaminoacetic aldehyde in aqueous solutions of hydrogen halides results in hydrohalides of substituted diarylethylmethylamines as the products of the interaction of two molecules of resorcinol with one molecule of aminoacetal [4–7]. At the same time, in

dioxane in the presence of trifluoromethanesulfonic acid these reactions lead to the formation of the corresponding calixarenes [8].

To determine the effect of the nature of the used phenol on the reaction result we carried out the reactions of α -aminoacetic aldehyde derivatives **Ia–Ic** with pyrocatechol. The reactions afford the corresponding polyphenols **IIa–IIc** comprising two catechol moieties with yields of 60–90%.



The reaction of acetal derivative **Ib** with alkyl halides in ethanol results in acetals **IIIa–IIIc** containing an onium moiety in the molecule. The reaction of pyrocatechol acetals in an aqueous alcohol medium in the presence of hydrobromic acid leads to the formation of polyphenols **IVa–IVc**. Due to the presence of the quaternary ammonium moiety the latter are of interest as antibiotics, antivirals and antifungicidal agents.

The structure and composition of the products obtained were confirmed by spectroscopic methods and elemental analysis.

2,2-Bis(3,4-dihydroxyphenyl)ethylamine hydrobromide (IIa). To a solution of 0.24 g of acetal Ia and 0.50 g of catechol in 4 ml of ethanol was added 1 ml of concentrated hydrobromic acid. The reaction mixture was refluxed for 5 h. After the removal of the solvent the residue was washed with ether (3 × 50 ml) and dried in a vacuum (10 h, 15 mm Hg). Yield 0.77 g (98%), mp 172°C. IR spectrum, v, cm⁻¹: 1607 (Ar), 2723 (NH⁺), 3320 (OH). ¹H NMR spectrum (CD₃OD), δ , ppm (*J*, Hz): 3.47 d (2H, CH₂, *J* 8.24), 4.05 t (1H, CH, *J* 8.20), 6.69 d. d (2H, CH_{Ar}, *J* 8.13, 2.10), 6.74 d (2H, CH_{Ar}, *J* 2.10), 6.78 d (2H, CH_{Ar}, *J* 8.13). Found, %: C 49.01; H 4.81; Br 23.63; N 4.11. C₁₄H₁₆BrNO₄. Calculated, %: C 49.14; H 4.71; Br 23.35; N 4.09.

2,2-Bis(3,4-dihydroxyphenyl)-*N*-methylethylamine hydrobromide (IIb) was prepared similarly



 $R^{1} = Me(\mathbf{a}), n-Bu(\mathbf{b}), n-C_{10}H_{21}(\mathbf{c}).$

from 1.35 g of acetal **Ib**, 2.50 g of pyrocatechol, and 3 ml of concentrated hydrobromic acid. Yield 2.80 g (89%), mp 160°C. IR spectrum, v, cm⁻¹: 1610 (Ar), 2730 (NH⁺), 3180 (OH). ¹H NMR spectrum (D₂O), δ , ppm (*J*, Hz): 2.62 s (3H, CH₃N), 3.44 d (2H, CH₂, *J* 8.25), 4.01 t (1H, CH, *J* 8.20), 6.63 d. d (2H, CH_{Ar}, *J* 8.16, 1.92), 6.77 d (2H, CH_{Ar}, *J* 1.92), 6.77 d (2H, CH_{Ar}, *J* 8.16). Found, %: C 50.03; H 5.22; Br 22.51; N 3.83. C₁₅H₁₈BrNO₄. Calculated, %: C 50.58; H 5.09; Br 22.43; N 3.93.

2,2-Bis(3,4-dihydroxyphenyl)-*N*,*N*-dimethylethylamine hydrobromide (IIc) was prepared similarly from 0.73 g of acetal Ic, 1.0 g of catechol, and 1 ml of concentrated hydrobromic acid. Yield 0.99 g (59%), mp 154°C. IR spectrum, v, cm⁻¹: 1610 (Ar), 2730 (NH⁺), 3180 (OH). ¹H NMR spectrum (D₂O), δ , ppm (*J*, Hz): 2.59 s (6H, CH₃N), 3.31 d (2H, CH₂, *J* 8.49), 3.89 t (1H, CH, *J* 8.44), 6.49 d. d (2H, CH_{Ar}, *J* 8.25, 2.14), 6.66 d (2H, CH_{Ar}, *J* 8.25), 6.67 d (2H, CH_{Ar}, *J* 2.14). Found, %: C 51.65; H 5.53; Br 21.65; N 3.52. C₁₆H₂₀BrNO₄. Calculated, %: C 51.90; H 5.44; Br 21.58; N 3.78.

N-(2,2-Diethoxy)ethyl-*N*,*N*,*N*-trimethylammonium bromide (IIIa). To a cooled (-20° C) solution of 1.77 g of acetal Ic in 3 ml of ethanol was added 1.15 g of cooled (-20° C) methyl bromide. The reaction mixture was kept at room temperature for 12 h. After removal of the solvent, the waxy residue was dried in a vacuum (2 h, 0.01 mm Hg). Yield 2.60 g (89%). ¹H NMR spectrum (CD₃OD), δ , ppm (*J*, Hz): 1.25 t (6H, CH₃ *J* 7.06), 3.27 s (9H, CH₃N), 3.55 d (2H, CH₂, *J* 4.67), 3.69 q (2H, CH₂O, *J* 7.06), 3.79 q (2H, CH₂O, *J* 7.06), 5.06 t (1H, CH, *J* 5.03). Found, %: C 36.83; H 7.97; Br 35.01; N 6.16. C₇H₁₈BrNO₂. Calculated, %: C 36.85; H 7.95; Br 35.03; N 6.14. *N*-(2,2-Diethoxy)ethyl-*N*,*N*-dimethyl-*N*-butylammonium bromide (IIIb). To a solution of 2.65 g of acetal Ic in 5 ml of ethanol was added 2.25 g of *n*-butyl bromide. The reaction mixture was refluxed for 2 h. After removal of the solvent, the waxy residue was washed with diethyl ether and dried in a vacuum (2 h, 0.01 mm Hg). Yield 4.01 g (82%). ¹H NMR spectrum (CD₃OD), δ, ppm (*J*, Hz): 1.03 t (3H, CH₃, *J* 7.40), 1.25 t (6H, CH₃, *J* 7.05), 1.38–1.48 m (2H, CH₂), 1.76–1.86 m (2H, CH₂), 3.23 s (6H, CH₃N), 3.46– 3.50 m (2H, CH₂N), 3.53 d (2H, CH₂N, *J* 4.95), 3.71 q (2H, CH₂O, *J* 7.05), 3.78 q (2H, CH₂O, *J* 7.05), 5.07 t (1H, CH, *J* 4.92). Found, %: C 44.40; H 8.98; Br 29.55; N 5.17. C₁₀H₂₄BrNO₂. Calculated, %: C 44.45; H 8.95; Br 29.57; N 5.18.

N-(2,2-Diethoxy)ethyl-*N*,*N*-dimethyl-*N*-decylammonium bromide (IIIc) was prepared analogously from 1.0 g of acetal derivative Ic and 1.37 g of 1bromodecane as a waxy gum. Yield 2.30 g (97%). ¹H (CD₃OD), δ, ppm (*J*, Hz): 0.89 t (3H, CH₃CH₂, *J* 6.94), 1.23 t (6H, CH₃CH₂O, *J* 7.02), 1.32 m (12H, (CH₂)₆), 1.40 m (2H, CH₂CH₂CH₂CH₂N), 1.81–1.83 m (2H, CH₂CH₂N), 3.33 s (6H, CH₃N), 3.45–3.49 m (2H, CH₂CH₂N), 3.51 d (2H, CHCH₂N, *J* 4.84), 3.68–3.81 m (4H, CH₂O), 5.06 t (1H, CH, *J* 4.92). Found, %: C 56.41; H 10.40; Br 20.97; N 3.59. C₁₈H₄₀BrNO₂. Calculated, %: C 56.54; H 10.47; Br 20.94; N 3.66.

N-2,2-Bis(3,4-dihydroxyphenyl)ethyl-*N*,*N*,*N*-trimethylammonium bromide (IVa). To a mixture of 2.0 g of acetal IIIa, 1.72 g of pyrocatechol, and 10 ml of ethanol was added 2 ml of concentrated hydrobromic acid. The reaction mixture was refluxed for 5 h, and the solvent was removed. The residue was washed with diethyl ether and dried in a vacuum (10 h, 15 mm Hg). Yield 2.58 g (86%), mp 145°C. IR spectrum, v, cm⁻¹: 1605 (Ar), 3300 (OH). ¹H NMR spectrum (D₂O), δ , ppm (*J*, Hz): 2.79 s (9H, CH₃N), 3.74 d (2H, CH₂, *J* 6.82), 4.13 t (1H, CH, *J* 6.62), 6.66 d. d (2H, CH_{Ar}, *J* 8.24, 2.14), 6.77 d (2H, CH_{Ar}, *J* 8.24), 6.82 d (2H, CH_{Ar}, *J* 2.14). Found, %: C 53.04; H 5.83; Br 20.70; N 3.58. C₁₇H₂₂BrNO₄. Calculated, %: C 53.14; H 5.77; Br 20.79; N 3.65.

N-2,2-Bis(3,4-dihydroxyphenyl)ethyl-*N*,*N*-dimethyl-*N*-butylammonium bromide (IVb) was prepared analogously from 0.98 g of acetal IIIb, 0.72 g of catechol, and 1 ml of concentrated hydrobromic acid. Yield 0.80 g (57%), mp 132°C. IR spectrum, v, cm⁻¹: 1609 (Ar), 3325 (OH). ¹H NMR spectrum (D₂O), δ , ppm (*J*, Hz): 0.69 t (3H, CH₃, *J* 7.33), 0.88–0.98 m (2H, CH₂), 1.28–1.38 m (2H, CH₂), 2.77 s (6H, CH₃N), 2.86–2.94 m (2H, CH₂), 3.71 d (2H, CH₂, *J* 6.42), 4.11 t (1H, CH, *J* 6.30), 6.69 d. d (2H, CH_{Ar}, *J* 8.20, 1.94), 6.79 d (2H, CH_{Ar}, *J* 8.20), 6.86 d (2H, CH_{Ar}, *J* 1.94). Found, %: C 56.30; H 6.72; Br 18.55; N 3.18. C₂₀H₂₈BrNO₄. Calculated, %: C 56.34; H 6.62; Br 18.74; N 3.29.

N-2,2-Bis(3,4-dihydroxyphenyl)ethyl-*N*,*N*-dimethyl-*N*-decylammonium bromide (IVc) was prepared analogously from 1.73 g of acetal IIIc, 1.0 g of catechol, and 1 ml of concentrated hydrobromic acid. Yield 0.85 g (37%), mp 155°C. IR spectrum, v, cm⁻¹: 1601 (Ar), 3305 (OH). ¹H NMR spectrum (CD₃OD), δ, ppm (*J*, Hz): 0.91 t (3H, C<u>H</u>₃CH₂, *J* 6.93), 1.08–1.18 m (2H, CH₃C<u>H</u>₂), 1.23–1.38 m [12H, (CH₂)₆], 1.59–1.69 m (2H, C<u>H</u>₂CH₂N), 3.05 s (6H, CH₃N), 3.20–3.27 m (2H, CH₂C<u>H</u>₂N), 4.00 d (2H, CHC<u>H</u>₂N, *J* 6.50), 4.32 t (1H, CH, *J* 6.30), 6.77 d (2H, CH_{Ar}, *J* 8.09), 6.81 d. d (2H, CH_{Ar}, *J* 8.09, 2.02), 6.89 d (2H, CH_{Ar}, *J* 2.02). Found, %: C 61.10; H 7.98; Br 15.53; N 2.68. C₂₆H₄₀BrNO₄. Calculated, %: C 61.17; H 7.90; Br 15.65; N 2.74.

The ¹H NMR spectra were recorded on an Avance 600 instrument operating at 600.13 MHz relative to the

residual protons of the deuterated solvent (CD₃OD). The IR spectra were registered on an UR-20 spectrometer in the range of 400–3600 cm⁻¹ in a slurry in mineral oil or in thin layer. All solvents were purified before use [9].

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