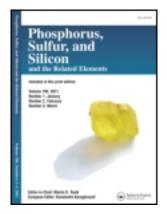
This article was downloaded by: [McGill University Library]

On: 19 November 2012, At: 12:41

Publisher: Taylor & Francis

Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered

office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information:

http://www.tandfonline.com/loi/gpss20

Synthesis and Antimicrobial Activity of Some New 5-Arylazothiazole, Pyrazolo[1,5-a] Pyrimidine, [1,2,4]Triazolo[4,3-a]Pyrimidine, and Pyrimido[1,2-a]Benzimidazole Derivatives Containing the Thiazole Moiety

Abdou O. Abdelhamid ^a , Eman K. A. Abdelall ^b , Nadia A. Abdel-Riheem ^a & Sayed A. Ahmed ^c

Version of record first published: 23 Mar 2010.

To cite this article: Abdou O. Abdelhamid, Eman K. A. Abdelall, Nadia A. Abdel-Riheem & Sayed A. Ahmed (2010): Synthesis and Antimicrobial Activity of Some New 5-Arylazothiazole, Pyrazolo[1,5-a] Pyrimidine, [1,2,4]Triazolo[4,3-a]Pyrimidine, and Pyrimido[1,2-a]Benzimidazole Derivatives Containing the Thiazole Moiety, Phosphorus, Sulfur, and Silicon and the Related Elements, 185:4, 709-718

To link to this article: http://dx.doi.org/10.1080/10426500902922933

PLEASE SCROLL DOWN FOR ARTICLE

Full terms and conditions of use: http://www.tandfonline.com/page/terms-and-conditions

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any

^a Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt

^b Department of Pharmaceutical Organic Chemistry, Faculty of Pharmacy, Beni-Suef University, Beni, Suef, Egypt

^c Department of Chemistry, Faculty of Science, Beni-Suef University, Beni, Suef, Egypt

instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.

Phosphorus, Sulfur, and Silicon, 185:709–718, 2010 Copyright © Taylor & Francis Group, LLC

Copyright © Taylor & Francis Group, LLC ISSN: 1042-6507 print / 1563-5325 online DOI: 10.1080/10426500902922933



SYNTHESIS AND ANTIMICROBIAL ACTIVITY OF SOME NEW 5-ARYLAZOTHIAZOLE, PYRAZOLO[1,5-a] PYRIMIDINE, [1,2,4]TRIAZOLO[4,3-a]PYRIMIDINE, AND PYRIMIDO[1,2-a]BENZIMIDAZOLE DERIVATIVES CONTAINING THE THIAZOLE MOIETY

Abdou O. Abdelhamid, Eman K. A. Abdelall, Nadia A. Abdel-Riheem, and Sayed A. Ahmed

¹Department of Chemistry, Faculty of Science, Cairo University, Giza, Egypt
²Department of Pharmaceutical Organic Chemistry, Faculty of Pharmacy,
Beni-Suef University, Beni Suef, Egypt
³Department of Chemistry, Faculty of Science, Beni-Suef University,
Beni Suef, Egypt

1-(2-(4,5-Dihydro-3-(4-methyl-2-phenylthiazol-5-yl)-5-phenylpyrazol-1-yl)-4-substituted thiazol-5-yl)-2-phenyldiazene were synthesized from hydrazonoyl halides and 3-(4-methyl-2-phenyl-1,3-thiazol-5-yl)-5-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide in ethanolic triethylamine. Also, pyrazolo[5,1-a]pyrimidines, 2,3,6-trisubstituted pyridines, and pyrazolo[3,4-d]pyridazines were obtained from sodium salt of 3-hydroxy-1-(4-methyl-2-phenylthiazol-5-yl)prop-2-en-1-one and different heterocyclic amines. All structures of the newly synthesized compounds were elucidated by elemental analysis, spectral data, and alternative synthetic route whenever possible. The newly synthesized compounds were tested towards different microorganisms.

Supplemental materials are available for this article. Go to the publisher's online edition of Phosphorus, Sulfur, and Silicon and the Related Elements to view the free supplemental file.

Keywords Hydrazonoyl halides; 5-phenylazothiazoles; pyrazolines; pyrazolo[1,5-a]pyrimidines; 2,3,6-trisubstituted pyridines

INTRODUCTION

Differently substituted pyrazolines and their derivatives are important biological agents, and a significant amount of research activity has been directed towards this class. In particular, they are used as antitumor, antibacterial, antifungal, antiviral, antiparasitic, antitubercular, and insecticidal agents. Some of these compounds also have anti-inflammatory, antidiabetic, anaesthetic, and analgesic properties. In addition, pyrazolopyrimidine systems are reported as inhibitors for the synthesis of DNA and RNA in

Received 28 January 2009; accepted 24 March 2009.

Address correspondence to Abdou O. Abdelhamid, Department of Chemistry, Faculty of Science, Cairo University, Giza 12613, Egypt. E-mail: aabdou_abdelhamid@yahoo.com

the cells of some types of cancer¹¹ and viruses.¹² We report in this article the synthesis and antimicrobial studies of some new 5-arylazothiazole, pyrazolo[1,5-a]pyrimidine, [1,2,4]triazolo[4,3-a]pyrimidine, and pyrimido[1,2-a]benzimidazole derivatives.

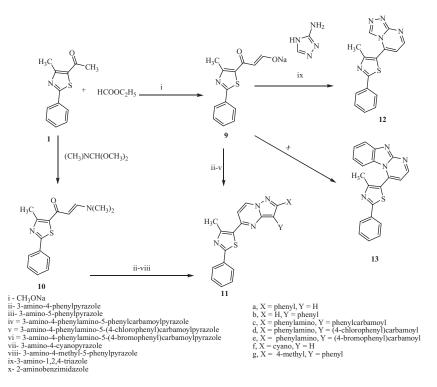
RESULTS AND DISCUSSION

Treatment of 1-(4-methyl-2-phenyl-1,3-thiazol-5-yl)-3-phenylprop-2-en-1-one¹³ (2) with thiosemicarbazide in ethanol under reflux gave 3-(4-methyl-2-phenyl-1,3-thiazol-5yl)-5-phenyl-4,5-dihydro-1*H*-pyrazole-1-carbothioamide (3). Structure 3 was elucidated by elemental analysis, spectral data, and chemical transformation. The ¹H NMR spectrum of 3 showed signals at $\delta = 2.66$ (s, 3H, CH₃), 3.14–3.16 (d, 1H, J = 6 Hz, pyrazoline) 3.32–3.35 (d, 1H, J = 6 Hz, pyrazoline), 6.00–6.02, 6.05–6.07 (q, 1H, J =6 Hz, pyrazoline), 6.15 (s, br., 2H, NH₂), 6.95–7.95 (m, 10H, ArH's). The mass spectrum showed peaks m/e at 380 (M⁺², 2.24%), 379 (M⁺¹, 21.3%), 378 (M, 100%). Thus, compound 3 reacted with C-benzoyl-N-phenylhydrazonoyl bromide in ethanolic triethylamine to afford 1-(2-(4,5-dihydro-3-(4-methyl-2-phenylthiazol-5-yl)-5-phenylpyrazol-1yl)-4-phenylthiazol-5-yl)-2-phenyldiazene (4a) (Scheme 1). Structure 4a was confirmed by elemental analysis, spectral data, and an alternate synthetic route. The ¹H NMR spectrum of 4a showed signals at $\delta = 2.76$ (2, 3H, CH₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline), 3.90 (dd, 1H, J = 12 Hz, pyrazoline), 5.76 (dd, 1H, J = 6 Hz, pyrazoline), and 7.26–8.15 (m, 20H, ArH's). The ¹³C NMR spectrum showed peaks at $\delta = 18.13$ (CH₃), 45.91 (CH₂), 63.74 (CH), 119.78, 119.78, 122.35, 123.54, 126.33, 127.61, 128.11, 128.95, 129.02, 129.24, 130.30, 130.65, 132.89, 134.11, 140.74, 142.87, 148.34, 152.61, 154.89,

Scheme 1

163.69, 167.34. The mass spectrum of $\bf 4a$ showed m/e = 582 (M⁺). Compound $\bf 3$ reacted with ω -bromoacetophenone in ethanol to give 4,5-dihydro-3-(4-methyl-2-phenylthiazol-5-yl)-5-phenyl-1-(4-phenylthiazol-2-yl)-1*H*-pyrazole ($\bf 5a$). The latter was coupled with benzenediazonium chloride in alcoholic sodium acetate to afford a product identical in all respects (mp, mixed mp, and spectra) to $\bf 4a$. An unequivocal support for structure $\bf 4a$ came from the reaction of each of the compounds $\bf 7a$, which was prepared via $\bf 6^{14}$ with ω -bromoacetophenone, with benzaldehyde in ethanolic sodium hydroxide followed by benzenediazonium chloride in alcoholic sodium acetate or by reaction with compound $\bf 8a$, which was prepared via reaction of $\bf 6$ with *C*-benzoyl-*N*-phenyl hydrazonoyl bromide, with benzaldehyde in ethanolic sodium hydroxide, and gave a product identical in all respects (mp, mixed mp, and spectra) with $\bf 4a$.

On the other hand, 1-(4-methyl-2-phenyl-1,3-thiazol-5-yl)ethanone (1) reacted with ethyl formate in dried ether containing sodium methoxide to afford a sodium salt of 3-hydroxy-1-(4-methyl-2-phenylthiazol-5-yl)prop-2-en-1-one (9). Structure 9 was confirmed by its reactions. Thus, treatment of 3-amino-4-phenyl-1*H*-pyrazole in acetic acid containing piperidine acetate afforded a product, namely 5-(4-methyl-2-phenylthiazol-5-yl)-3-phenylpyrazolo[1,5-a]pyrimidine (11a) (Scheme 2). Structure 11a was elucidated by elemental analysis, spectral data and an alternate synthetic route. The ¹H NMR spectrum of 11a showed signals at $\delta = 2.53$ (s, 3H, CH₃) and 7.44–7.82 (m, 13H, ArH's, pyrazole H-5, pyrimidine H4 & H5). Thus, treatment of 3-(dimethylamino)-1-(4-methyl-2-phenylthiazol-5-yl)prop-2-en-1-one (10)¹⁵ with 3-amino-4-phenylpyrazole in acetic acid and ammonium



Scheme 2

acetate by heating under reflux gave a product identical in all respects (mp, mixed mp, spectra) with 11a.

Analogously, either compound **9** or **10** reacted with the appropriate 3-aminopyrazoles, 3-amino-1,2,4-triazole, or 2-aminobenzimidazole gave pyrazolo[1,5-*a*]pyrimidines **11b–g**, 5-(4-methyl-2-phenylthiazol-5-yl)-[1,2,4]triazolo[4,3-*a*]pyrimidine (**12**), and 4-(4-methyl-2-phenyl-4,5-dihydro-1,3-thiazol-5-yl)pyrimido[1,2-*a*]benzimidazole (**13**), respectively.

Treatment of 3-(dimethylamino)-1-(4-methyl-2-phenylthiazol-5-yl)prop-2-en-1-one (10) with 2,4-pentanedione or ethyl 3-oxobutanoate in boiling acetic acid and ammonium acetate under reflux gave 3-acetyl-2-methyl-6-(4-methyl-2-pheny)lthiazol-2-ylpyridine (14) and ethyl 2-methyl-6-(4-methyl-2-phenyl)thiazol-2-ylpyridine-3-carboxylate (15), respectively (Scheme 3).

Scheme 3

Finally, treatment of the appropriate of **9** or **10** with the arenediazonium chloride in ethanol containing sodium acetate as a buffer solution yielded 2-(2-phenylhydrazono)-3-(4-methyl-2-phenylthiazol-5-yl)-3-oxopropanal (**16a**) and 2-(2-(4-tolyl)hydrazono)-3-(4-methyl-2-phenylthiazol-5-yl)-3-oxopropanal (**16b**), respectively. Structure **16** was confirmed by elemental analysis, spectral data, and chemical transformation. The ¹H NMR spectrum of **16a** showed signals at $\delta = 2.46$ (s, 3H, CH₃), 6.67–8.05 (m, 10H, ArH's), 9.98 (s, 1H, -CHO), and 14.39 (s, br., 1H, NH). Thus, **16a** was reacted with hydrazine hydrate in boiling ethanol under reflux and gave 1-(3-(4-methyl-2-phenylthiazol-5-yl)-4H-pyrazol-4-ylidene)-2-phenylhydrazine (**17a**) (Scheme 3).

PHARMACOLOGY: ANTIMICROBIAL SCREENING

Sixteen compounds were screened by the Agar dilution technique¹⁶ for their antimicrobial activity (see Supplemental Materials, Table S1, available online).

EXPERIMENTAL

All melting points were determined on an electrothermal apparatus and are uncorrected. IR spectra were recorded (KBr discs) on a Shimadzu FT-IR 8201 PC spectrophotometer. 1H NMR and ^{13}C NMR spectra were recorded in CDCl $_3$ and (CD $_3$) $_2SO$ solutions on a Varian Gemini 300 MHz spectrometer, and chemical shifts are expressed as δ using TMS as an internal reference. Mass spectra were recorded on a GC-MS QP1000. Elemental analyses were carried out at the Microanalytical Center of Cairo University. The hydrazonoyl halides were prepared as previously described. $^{17-19}$

3-(4-Methyl-2-phenyl-1,3-thiazol-5-yl)-5-phenyl-4,5-dihydro-1H-pyrazole-1-carbothioamide 3

To a mixture of 1-(4-methyl-2-phenyl-1,3-thiazol-5-yl)-3-phenylprop-2-en-1-one (2) (1.55 g, 5 mmol) and thiosemicarbazide (0.46 g, 5 mmol) in ethanol (20 mL), a catalytic amount of triethylamine was added, then heated under reflux for 6 h. The resulting solid was collected and recrystallized from ethanol to give 3 (Tables I and II).

4,5-Dihydro-3-(4-methyl-2-phenylthiazol-5-yl)-1-(4-substituted thiazol-2-yl)-5-phenyl-1H-pyrazole 5a-c

A mixture of compound 3 (1.8 g, 5 mmol) and the appropriate of ω -bromoacetophenone, chloroacetone, or 5-bromoacetyl-4-methyl-2-phenylthiazole (5 mmol) in ethanol (20 mL) was heated under reflux for 2 h. The resulting solid was neutralized with sodium bicarbonate (10%), collected, and recrystallized from ethanol to give compounds **5a–c** (Tables I and II).

1-(2-(4,5-Dihydro-3-(4-methyl-2-phenylthiazol-5-yl)-5-phenylpyrazol-1-yl)-4-substituted thiazol-5-yl)-2-phenyldiazene 4a-c

Method A. A mixture of the appropriate hydrazonoyl halides (5 mmol), **3** (1.8 g, 5 mmol), and triethylamine (0.5 g, 0.75 mL, 5 mmol) in ethanol (20 mL) was boiled under reflux for 2 h. The resulting solid was collected and recrystallized from DMF-EtOH mixture to give **4a–c** (Tables I and II).

Method B. An equimolar amount of the appropriate **8a-c** (5mmol) and benzaldehyde (0.53g, 5 mmol) in ethanol (20 mL) was cooled at 5°C, then sodium hydroxide solution (10 mL, 4N) was added portionwise while stirring. The resulting solid was collected and recrystallized from DMF/ethanol to give identical product in all aspects (mp, mixed mp, and spectra) with corresponding **4a-c** (Tables I and II).

Method C. A solution of benzenediazonium chloride (10 mmol) was added dropwise to a stirred solution of the appropriate **5a-c** in ethanol (50 mL) containing sodium acetate trihydrate (1.3g, 10 mmol) at 0–5°C. The reaction mixture was stirred for 3 h at 0–5°C. The resulting solid was collected and crystallized from DMF/ethanol to give **4a-c** identical in all aspects (mp, mixed mp, and spectra) with those above.

Synthesis of N-[1-(4-Methyl-2-substituted thiazol-5-yl)-ethylidene]-N'-(4-phenyl-thiazol-2-yl)-hydrazine 7a-c

A mixture of **6** (1.45 g, 5 mmol) and the appropriate of ω -bromoacetophenone, chlroacetone, or 5-bromoacetyl-4-methyl-2-phenylthiazole (5 mmol) in ethanol (20 mL)

Table I Characterization data of the newly synthesized compounds

% Analyses, Calcd./ Found			und	Mol. Formula/	Color/	Mp, °C/	Compd.
	N	Н	С	Mol. Wt.	Yield%	Solvent	No.
94	14.80	4.79	63.47	C ₂₀ H ₁₈ N ₄ S ₂	Yellow	210–213	3
8	14.61	4.82	63.33	378.5	80	EtOH	
00	14.42	4.40	70.08	$C_{34}H_{26}N_6S_2$	Violet	254-256	4a
32	14.72	4.35	69.85	582.74	86	DMF/EtOH	
32	16.14	4.65	66.90	$C_{29}H_{24}N_6S_2$	Violet	248-250	4b
22	16.01	4.60	66.81	520.67	82	DMF/EtOH	
.5	14.42	4.30	67.13	$C_{38}H_{29}N_7S_3$	Violet	262-263	4c
)1	14.30	4.14	67.30	679.87	80	DMF/EtOH	
9	13.45	4.84	66.32	$C_{23}H_{20}N_5$	Brown	192-195	5a
7	13.21	4.62	66.45	416.56	66	EtOH	
10	11.70	4.63	70.26	$C_{28}H_{22}N_4S_2$	Orange	202-204	5b
21	11.50	4.55	70.10	478.63	69	EtOH	
0	12.16	4.38	66.75	$C_{32}H_{25}N_5S_3$	Orange	195–197	5c
8	12.21	4.50	66.58	575.76	62	EtOH	
2	14.35	4.65	64.59	$C_{21}H_{18}N_4S_2$	Brown	145–147	7a
23	14.20	4.60	64.42	390.52	80	EtOH	
52	17.06	4.91	58.51	$C_{16}H_{16}N_4S_2$	Green	151–153	7b
80	17.09	5.72	58.20	494.63	90	EtOH	7.0
2	14.36	4.34	61.57	C ₂₅ H ₂₁ N ₅ S ₃	Brown	161–163	7c
53	14.21	4.49	61.33	487.66	69	EtOH	70
)6	16.99	4.48	65.56	$C_{27}H_{22}N_6S_2$	Red	164–166	8a
)1	16.67	4.46	65.55	494.63	80	EtOH	oa
32		4.66				119–121	8b
	19.43	4.50	61.09	C ₂₂ H ₂₀ N ₆ S ₂ 432.56	Red 82	EtOH	อม
59	19.43		61.00				0-
25	16.57	4.26	62.92	$C_{31}H_{25}N_7S_3$	Red	143–145	8c
.9	16.52	4.22	62.78	591.77	69 V II	EtOH	11
0	15.21	4.38	71.72	$C_{22}H_{16}N_4S$	Yellow	149–150	11a
57	15.10	4.22	71.60	368.4	75	EtOH	111
0	15.21	4.38	71.72	$C_{22}H_{16}N_4S$	Orange	238–240	11b
0	15.00	4.28	71.55	368.4	89	DMF	
88	16.72	4.41	69.31	$C_{29}H_{22}N_6OS$	Yellow	260–261	11c
54	16.85	4.32	69.51	502.58	66	EtOH	
7	15.65	3.94	64.86	C ₂₉ H ₂₁ ClN ₆ OS	Yellow	245–247	11d
2	15.56	4.12	64.74	537.05	66	DMF/EtOH	
1	14.45	3.64	59.90	$C_{29}H_{21}$ BrN ₆ OS	Orange	263–265	11e
25	14.65	3.75	60.12	581.50	70	DMF/EtOH	
.0	22.07	3.49	64.34	$C_{17}H_{11}N_5S$	Pale yellow	252–255	11f
00	21.85	3.56	64.34	317.35	85	AcOH	
88	14.65	4.74	72.23	$C_{23}H_{18}N_4S$	Yellow	154–157	11g
12	14.56	4.85	72.13	382.49	78	Benzene	
93	23.87	3.78	61.42	$C_{15}H_{11}N_5S$	Colorless	190-192	12
2	23.95	3.87	61.66	293.35	82	AcOH	
35	16.36	4.12	70.15	$C_{20}H_{14}N_4S$	Yellow	160–163	13
3	16.46	3.90	70.00	342.43	78	AcOH	
10	9.08	5.23	70.10	$C_{18}H_{16}N_2OS$	Colorless	138-140	14
25	9.12	5.32	70.20	308.41	77	EtOH	
17	8.28	5.36	67.43	$C_{19}H_{18}N_2O_2S$	Colorless	80-81	15
57	8.42	5.52	67.34	338.43	67	EtOH	
8	12.03	4.33	65.31	$C_{19}H_{15} N_3O_2S$	Orange	222-224	16a
00	11.85	4.35	65.23	349.41	78	EtOH	
1	11.56	4.71	66.10	$C_{20}H_{17}\ N_3O_2S$	Orange	232-235	16b
60	11.45	4.65	66.20	363.44	75	EtOH	
28	20.27	4.38	66.07	$C_{19}H_{15}N_5S$	Pale yellow	254-257	17a
34	20.05	4.52	66.12	345.43	68	EtOH	

Table II Spectral data of some newly synthesized compounds

Spectral data	Compound No.
IR: 3380, 2160 (NH ₂), 3056, 2980 (CH, aromatic and aliphatic), 1623 (C=N), 1370 (CH ₃),	3
1430 (CH ₂).	
¹ H NMR: $\delta = 2.66$ (s, 3H, CH ₃), 3.14–3.16 (d, 1H, $J = 6$ Hz, pyrazoline) 3.32–3.35 (d, 1H,	
J = 6 Hz, pyrazoline), 6.00–6.02, 6.05–6.07 (q, 1H, $J = 6$ Hz, pyrazoline), 6.15 (s, br., 2H, NH ₂), 6.95–7.95 (m, 10H, ArH's). Ms: m/e = 380 (M ⁺² , 2.24%), 379 (M ⁺¹ , 21.3%), 378 (M, 100%).	
IR: 3051, 2994 (CH, aromatic and aliphatic), 1623 (C=N), 1370 (CH ₃), 1433 (CH ₂).	4 a
¹ H NMR: δ = 2.76 (2, 3H, CH ₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline), 3.90 (dd, 1H, J = 12	
Hz, pyrazoline), 5.76 (dd, 1H, $J = 6$ Hz, pyrazoline) and 7.26–8.15 (m, 20 H, ArH's).	
¹³ C NMR: $\delta = 18.13$ (CH ₃), 45.91 (CH ₂), 63.74 (CH), 119.78, 119.78, 122.35, 123.54,	
126.33, 127.61, 128.11, 128.95, 129.02, 129.24, 130.30, 130.65, 132.89, 134.11, 140.74,	
142.87, 148.34, 152.61, 154.89, 163.69, 167.34.	
Ms: $m/e = 582$ (M).	
IR: 3050, 2990 (CH, aromatic and aliphatic), 1623 (C=N), 1365 (CH ₃), 1435 (CH ₂).	4b
¹ H NMR: δ = 2.42 (s, 3H, CH ₃ CO), 2.76 (2, 3H, CH ₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline),	
3.90 (dd, 1H, J = 12 Hz, pyrazoline), 5.76 (dd, 1H, J = 6 Hz, pyrazoline) and $7.26-8.15$	
(m, 15 H, ArH's).	
¹ H NMR: δ = 2.76 (s, 6H, 2-CH ₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline), 3.90 (dd, 1H, J = 12	4c
Hz, pyrazoline), 5.76 (dd, 1H, $J = 6$ Hz, pyrazoline) and 7.26—8.15 (m, 20 H, ArH's).	_
IR: 3056, 2992 (CH, aromatic and aliphatic), 1623 (C=N), 1345 (CH ₃), 1437 (CH ₂).	5a
¹ H NMR: δ = 2.42 (s, 3H, CH ₃ CO), 2.76 (2, 3H, CH ₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline),	
3.90 (dd, 1H, $J = 12$ Hz, pyrazoline), 5.76 (dd, 1H, $J = 6$ Hz, pyrazoline) and 7.26–8.15	
(m, 17 H, ArH's). IR: 3058, 2985 (CH, aromatic and aliphatic), 1623 (C=N), 1370 (CH ₃), 1440 (CH ₂).	5b
¹ H NMR: $\delta = 2.76$ (2, 3H, CH ₃), 3.32 (dd, 1H, $J = 6$ Hz, pyrazoline), 3.90 (dd, 1H, $J = 12$	30
Hz, pyrazoline), 5.76 (dd, 1H, $J = 6$ Hz, pyrazoline) and 7.26–8.15 (m, 16 H, ArH's).	
IR: 3056, 2980 (CH, aromatic and aliphatic), 1623 (C=N), 1370 (CH ₃), 1440 (CH ₂).	5e
¹ H NMR: δ = 2.76 (s, 6H, 2-CH ₃), 3.32 (dd, 1H, J = 6 Hz, pyrazoline), 3.90 (dd, 1H, J = 12	
Hz, pyrazoline), 5.76 (dd, 1H, $J = 6$ Hz, pyrazoline) and 7.26–8.15 (m, 16 H, ArH's).	
IR: 3350 (NH), 3056, 2988 (CH, aromatic and aliphatic), 1623 (C=N), 1350 (CH ₃).	7a
¹ H NMR: $\delta = 2.29$ (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 7.16 (s, 1H, thiazole H-5), 7.27–7.89 (m,	
10 H, ArH's) and 8.99 (s, br., 1H, NH).	
IR: 3350 (NH), 3056, 2993 (CH, aromatic and aliphatic), 1623 (C=N), 1356 (CH ₃).	7b
¹ H NMR: $\delta = 2.19$ (s, 3H, CH ₃), 2.29 (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 7.16 (s, 1H, thiazole	
H-5), 7.67–7.89 (m, 5 H, ArH's) and 8.62 (s, br., 1H, NH).	
IR: 3350 (NH), 3050, 2987 (CH, aromatic and aliphatic), 1623 (C=N), 1360 (CH ₃).	7c
¹ H NMR: δ = 2.29 (s, 3H, CH ₃), 2.46 (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 6.96 (s, 1H, thiazole	
H-5), 7.66–7.88 (m, 10 H, ArH's) and 8.99 (s, br., 1H, NH).	0-
IR: 3350 (NH), 3056, 2988 (CH, aromatic and aliphatic), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: $\delta = 2.29$ (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 7.37–8.00 (m, 15 H, ArH's) and 9.74 (s,	8a
br., 1H, NH).	
IR: 3350 (NH), 3062, 2980 (CH, aromatic and aliphatic), 1623 (C=N), 1350 (CH ₃).	8b
¹ H NMR: δ = 2.29 (s, 3H, CH ₃), 2.54 (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 7.50–8.07 (m, 10 H,	O.D
ArH's) and 8.89 (s, br., 1H, NH).	
IR: 3350 (NH) 3050, 2985 (CH, aromatic and aliphatic), 1623 (C=N), 1347 (CH ₃).	8c
¹ H NMR: δ = 2.29 (s, 3H, CH ₃), 2.46 (s, 3H, CH ₃), 2.59 (s, 3H, CH ₃), 7.50–7.08 (m, 15 H,	
ArH's) and 9.47 (s, br., 1H, NH).	
IR: 3310 (NH) 3066, 2992 (CH, aromatic and aliphatic), 1623 (C=N), 1340 (CH ₃).	11a
¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 6.53 (s, 1H, pyrazole H-4), 7.13 (d, 1H, J = 4Hz, pyrimidine	
H-5), 7.46–7.82 (m, 10H, ArH's) and 8.74 (d, 1H, 8Hz, pyrimidine H-6).	

Table II Spectral data of some newly synthesized compounds (Continued)

Spectral data	Compound No.
IR: 3320 (NH), 3050, 2990 (CH, aromatic and aliphatic), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: $\delta = 2.46$ (s, 3H, CH ₃), 6.90 (d, 2H, J = 8H, ArH's), 7.14 (d, 1H, J = 4Hz, pyrimidine	11b
H-5), 7.56–7.78 (m, 8H, ArH's), 8.74 (d, 1H, 8Hz, pyrimidine H-6 and 9.05 (s, 1H, pyrazole H-3). IR: 3299 (NH), 3065, 2994 (CH, aromatic and aliphatic), 1662 (CO), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 6.98 (d, 2H, J = 8H, pyrimidine H-5), 7.07–8.00 (m, 15H, ArH's), 8.49 (d, 1H, 8Hz, pyrimidine H-6) and 11.34 (s, br., 1H, 2NH).	11c
IR: 3299 (NH), 3056, 2980 (CH, aromatic and aliphatic), 1662 (CO), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 6.98 (d, 2H, J = 8H, pyrimidine H-5), 7.07–8.00 (m, 14H, ArH's), 8.49 (d, 1H, 8Hz, pyrimidine H-6) and 11.34 (s, br., 1H, 2NH).	11d
IR: 3299 (NH), 3056, 2990 (CH, aromatic and aliphatic), 1665 (CO), 1620 (C=N), 1340 (CH ₃). ¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 6.98 (d, 2H, J = 8H, pyrimidine H-5), 7.07–8.00 (m, 14H, ArH's), 8.49 (d, 1H, 8Hz, pyrimidine H-6) and 11.34 (s, br., 1H, 2NH).	11e
IR: 3056, 2996 (CH, aromatic and aliphatic), 2160 (CN), 1623 (C=N), 1340 (CH ₃). 1 H NMR: $\delta = 2.46$ (s, 3H, CH ₃), 7.35 (d, 2H, J = 8H, pyrimidine H-5), 7.62–7.79 (m, 5H. ArH's), 9.23 (s, 1H, pyrazole H-5), and 9.50 (d, 1H, 8Hz, pyrimidine H-6).	11f
¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 2.47 (s, 3H, CH ₃), 7.11 (d, 2H, J = 8H, pyrimidine H-5), 7.23–7.85 (m, 10H, ArH's) and 8.60 (d, 1H, 8Hz, pyrimidine H-6).	11g
¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 6.99 (d, 2H, J = 4H, pyrimidine H-5), 7.62–7.79 (m, 5H, ArH's), 9.01 (d, 4H, 8Hz, pyrimidine H-6) and 9.57 (s, 1H, triazole H-5).	12
¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 7.26 (d, 2H, J = 8H, pyrimidine H-5), 7.32–7.94 (m, 9H, ArH's) and 8.88 (d, 4H, 8Hz, pyrimidine H-6)	13
IR: 3056, 2980 (CH, aromatic and aliphatic), 1675 (CO), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: δ = 2.49 (s, 3H, CH ₃), 2.62 (s, 3H, CH ₃), 2.66 (s, 3H, CH ₃), 7.62–7.76 (m, 5H, ArH's), 7.86 (d, 1H, 4Hz, pyridine H-5) and 7.06 (d, 1H, 4Hz, pyridine H-6).	14
IR: 3056, 2980 (CH, aromatic and aliphatic), 1713 (CO), 1623 (C=N), 1340 (CH ₃). ¹ H NMR: δ = 1.34 (t, J = 7Hz, CH ₂ CH ₃), 2.73 (s, 3H, CH ₃), 2.75 (s, 3H, CH ₃), 4.31 (q, J = 7H, CH ₂ CH ₃), 7.62–7.92 (m, 5H, ArH's), 8.98(d, 1H, 4Hz, pyridine H-5) and 8.29 (d, 1H, 4Hz, pyridine H-6).	15
IR: 32280 (NH), 3056, 2980 (CH, aromatic and aliphatic), 2870, 2780 (CH), 1690 (CO), 1340 (CH ₃). ¹ H NMR: δ = 2.46 (s, 3H, CH ₃), 7.28–8.05 (m, 10H, ArH's), 9.75 (s, 1H, CHO) and 15.09 (s, 1H,	16a
br., NH). IR: 32280 (NH), 3056, 2980 (CH, aromatic and aliphatic), 2870, 2780 (CH), 1690 (CO), 1340 (CH ₃).	16b
¹ H NMR: δ = 2.33 (s, 3H, CH ₃), 2.46 (s, 3H, CH ₃), 7.44–8.05 (m, 9H, ArH's), 9.75 (s, 1H, CHO) and 15.09 (s, 1H, br., NH). IR: 32280 (NH), 3056, 2980 (CH, aromatic and aliphatic), 1340 (CH ₃). ¹ H NMR: δ = 2.60 (s, 3H, CH ₃), 6.85–8.12 (m, 11H, ArH's and pyrazole H-5), and 11.40 (s, 1H, br., NH).	17a

was heated under reflux for 2 h. The resulting solid was collected and recrystallized from the proper solvent to give compounds **7a–c** (Tables I and II).

Sodium Salt of 3-Hydroxy-1-(4-methyl-2-phenylthiazol-5-yl)prop-2-en-1-one 9

A mixture of 1-(4-methyl-2-phenyl-thiazol-5-yl)ethanone ($2.17\,g$, $10\,mmol$) and ethyl formate ($0.74\,g$, $10\,mmol$) in dry ether ($20\,mL$) was added portionwise while stirring to

solution sodium methoxide (0.54 g, 10 mmol) in dry ether (10 mL) at 0–5°C. The resulting solid was collected, dried, and was used without purification.

N-[1-(4-Methyl-2-substituted thiazol-5-yl)-ethylidene]-N'-(4-phenyl-5-phenylazo-thiazol-2-yl)-hydrazine 8a-c

A mixture of **6** (1.45 g, 5 mmol), the appropriate hydrazonoyl halides **iiia-c**, and triethylamine (0.75 mL, 5 mmol) in ethanol (20 mL) was heated under reflux for 2 h. The resulting solid was collected and recrystallized from the proper solvent to give **8a-c** (Tables I and II).

Pyrazolo[5,1-a]pyrimidines 11a-g, [1,2,4]Triazolo[4,3-a]pyrimidine 12, Pyrimido[1,2-a]benzimidazole 13, and Pyridines 14, 15

Method A. A mixture of the sodium salt **9** (1.2g, 5 mmol) and the appropriate amount of 3-amino-4-phenylpyrazole, 3-amino-5-phenylpyrazole, 3-amino-4-phenylcar bamoyl-5-phenylaminopyrazole, 3-amino-4-(4-chlorophenyl)carbamoyl-5-phenylaminopyrazole, 3-amino-4-(4-bromophenyl)carbamoyl-5-phenylaminopyrazole, 3-amino-4-cyano pyrazole, 3-amino-4-methyl-5-phenylpyrazole, 3-amino-1,2,4-triazole, 2-aminobenzimida zole, acetylacetone, or ethyl acetoacetate (5 mmol) in a solution of piperidine acetate [piperidine (2.5 mL)], water (5 mL), and acetic acid (2.5 mL) was heated under reflux for about 10 min. Then acetic acid (1.5 mL) was added while boiling, and the resulting solid was collected and recrystallized from the appropriate solvents (Tables I and II).

Method B. An equimolar amount of **10** (1.02 g, 5 mmol), the appropriate amount of 3-amino-4-phenylpyrazole, 3-amino-5-phenylpyrazole, 3-amino-4-phenylcarbamoyl-5-phenylaminopyrazole, 3-amino-4-(4-chlorophenyl)carbamoyl-5-phenylaminopyrazole, 3-amino-4-cyanopyrazole, 3-amino-4-methyl-5-phenylpyrazole, 3-amino-1,2,4-triazole, 2-aminobenzimidazole, acetylacetone, or ethyl acetoacetate (5 mmol) and ammonium acetate (5 mmole) in acetic acid (10 mL) was heated under reflux for 4 h. The resulting solid was collected and recrystallized from the proper solvent to give products (Tables I and II).

2-(2-Arylhydrazono)-3-(4-methyl-2-phenylthiazol-5-yl)-3-oxopropanal 16a,b

Arenediazonium chloride (5 mmol) was added to the appropriate cold solution of **9** or **10** (5 mmol) and sodium acetate (0.65 g, 5 mmol) in ethanol (50 mL) while stirring. The crude solid was collected and recrystallized from ethanol to give **16** (Tables I and II).

(1-(3-(4-Methyl-2-phenylthiazol-5-yl)-4H-pyrazol-4-ylidene)-2-phenylhydrazine 17a

An equimolar amount of **16a** and hydrazine hydrate (5 mmole) in ethanol (20 mL) was boiled under refluxed for 15 min. The resulting solid was collected and recrystallized from the proper solvent to give the pyrazole **17a** (Tables I and II).

REFERENCES

- 1. E. C. Taylor, H. Patel, and H. Kumr, Tetrahedron, 48, 8089–8100 (1992).
- 2. G. H. Keats, UK Patent. 1, 209, 631 (1970); Chem. Abstr., 74, 141787b (1971).
- 3. R. M. Kedar, N. N. Vidhale, and M. M. Chincholkar, Orient J. Chem., 13, 143–148 (1997).
- 4. Z. H. Khalil and A. S. Yanni, J. Ind. Chem. Soc., 58, 168–170 (1981).
- 5. N. B. Das and A. S. Mittra, Ind. J. Chem., 16B, 638-640 (1978).
- 6. D. Azarifar and M. Shaabanzadeh, Molecules, 7, 885-895 (2002).
- B. Shivarama Holla, P. M. Akberali, and M. K. Shivanada, Farmaco, 55, 256–263 (2000); Chem. Abstr., 134, 86195n (2001).
- E. Palaska, G. Sahin, P. Kelicen, N. T. Durlu, and G. Altinok, Farmaco, 57, 101–107 (2002); Chem. Abstr., 138, 55916h (2003).
- 9. H. A. Regaila, A. K. El-Bayouk, and M. Hammad, Egypt. J. Chem., 20, 157–166 (1979).
- 10. F. Seela, C. Wei, G. Becher, M. Zulauf, and P. Leonard, *Bioorg. Med. Chem. Lett.*, 289–292 (2000).
- 11. Y. Tominaga, Y. Honkawa, M. Hara, and A. Hosomi, J. Heterocycl. Chem., 27, 775–783 (1990).
- O. Moukha-chafiq, M. L. Taha, H. B. Lazrek, J.-J. Vasseur, C. Pannecouque, M. Witvrouw, and E. De Clercq, Farmaco, 57, 27–32 (2002); Chem. Abstr., 137, 93724z (2002).
- 13. S. G. Nemlekar, L. P. Ghalasasi, and S. D. Sawant, *Indian Drugs*, 22, 73-76 (1984).
- 14. S. P. Singh, S. Sehgal, and P. K. Sharma, *Indian J. Chem.*, **29B**, 533–538 (1990).
- Y. H. Zaki, S. A. Ahmed, A. M. Hussein, and A. O. Abdelhamid, *Phosphorus, Sulfur, and Silicon*, 181, 825–837 (2006).
- 16. A. A. El-helby, *Al-Azhar J. Pharm. Sci.*, **27**, 375–386 (2001).
- 17. A. S. Shawaliand and A. O. Abdelhamid, *Bull. Soc. Jpn.*, **49**, 321–324 (1976).
- 18. N. F. Eweiss and A. O. Abdelhamid. J. Heterocycl. Chem., 17, 1713–1717 (1980).
- A. O. Abdelhamid, A. H. El-Ghandour, A. M. Hussein, and Y. H. Zaki, *Phosphorus, Sulfur, and Silicon*, 180, 2079–2109 (2005).