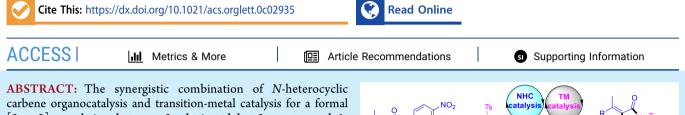




Synergistic *N*-Heterocyclic Carbene/Palladium-Catalyzed [3 + 2] Annulation of Vinyl Enolates with 1-Tosyl-2-vinylaziridine

Jian Gao, Jianming Zhang, Shuaishuai Fang, Jie Feng, Tao Lu, and Ding Du*



carbene organocatalysis and transition-metal catalysis for a formal [3 + 2] annulation between 3-substituted but-2-enoates and 1-tosyl-2-vinylaziridine was developed. This cooperative strategy provides a facile and efficient access to various functionalized (*E*)-3-ethylidene-4-vinylpyrrolidin-2-ones in a regioselective and stereo-

R = aryl, alkyl R = aryl, aryl

selective manner. The preliminary asymmetric studies were also performed, which indicated a potential for enantioselective annulation of vinyl enolate intermediates with transition-metal– π -allyl species.

O ver the past decades, conventional transition-metal (TM) catalysis¹ and organocatalysis² have been widely explored and applied in the activation and functionalization of various chemical bonds to accomplish useful chemical transformations. Nevertheless, with the development of the modern organic synthesis, the traditionally single catalytic systems have also encountered some challenges, such as the activation efficiency for new types of substrates, and the control for the reactivity and selectivity. Aiming to get out of this dilemma, synergistic or cooperative catalysis, especially synergistic TM catalysis and organocatalysis, has drawn much attention and has been proven to be a general and effective strategy to realize inaccessible transformations by single catalytic approaches previously via independent activation of separate substrates simultaneously.³

In recent years, N-heterocyclic carbenes (NHCs) have emerged as efficient and powerful organocatalysts for the discovery of novel and unconventional synthetic transformations with readily access to a wide array of heterocycles and bioactive molecules.⁴ Especially via umpolung activation mode, a variety of NHC-bound active intermediates have been approached and widely applied for chemical bond construction. However, because of the strong affinity of NHCs to TM centers through irreversible binding and coordination, a disruption of the individually desired catalytic activities of both NHCs and TMs usually happens (Scheme 1a).⁵ Thus, the design and development of synergistic catalysis by merging NHCs with TMs remain a formidable challenge and only limited examples were explored (Scheme 1b).⁶ In 2014, the Scheidt group pioneered an intramolecular allylation and subsequent acylation reaction of NHC-bound Breslow intermediates for the construction of dihydrocoumarins derivatives co-catalyzed by NHC/palladium catalyst. Following this pioneering work, intermolecular allylation and benzylation of NHC-bound Breslow intermediates were also achieved by Ohmiya and co-workers.^{8–11} Furthermore, other than Breslow intermediates, NHC-bound active species such as homoenolates, enolates, and azolium enamides have also been developed as C2 or C3 synthons for diverse [n+m] annulation^{12–18} and arylation¹⁹ reactions by synergistic NHC/TMs (Pd, Cu, Ir, Au) co-catalysis. In view of the importance of synergistic catalysis and the diversity of NHC-bound intermediates, it is still highly desirable to explore novel and efficient catalytic systems based on cooperative NHC/TM catalysis.

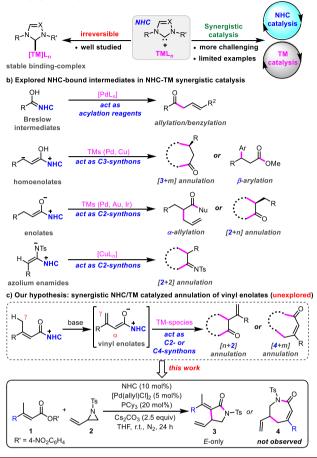
Recently, NHC-bound vinyl enolates derived from $\alpha_{\beta}\beta_{\beta}$ unsaturated acids or their esters, or (α -bromo)enals via γ activation, have been used as reactive nucleophilic C4 synthons or C2 synthons to enable diverse annulation reactions at the γ or α -position.²⁰ However, to the best of our knowledge, NHC/ TM co-catalyzed reactions of vinyl enolates have not been reported yet. We envisioned that NHC-bound vinyl enolates would also be applicable in combination with TM-activated electrophiles to realize various formal [n+2] or [4+m]annulations (Scheme 1c). Herein, as part of our continuous efforts to enrich vinyl enolate chemistry,²¹ we report an NHC/ Pd co-catalyzed [3 + 2] annulation of α_{β} -unsaturated esters 1 with 1-tosyl-2-vinylaziridine 2, a versatile substrate for the generation of Pd- π -allyl species,²² for the facile construction of (E)-3-ethylidene-4-vinylpyrrolidin-2-one skeleton 3 that is frequently found in numerous synthetic compounds or natural products.²³ Notably, (E)-[3 + 2] adducts 3 were obtained exclusively in a regioselective and stereoselective manner, and the corresponding [4 + 3] adducts 4 were not observed.

Received: September 1, 2020



Scheme 1. Merged *N*-Heterocyclic Carbene and Transition-Metal Co-catalysis

a) Merging N-heterocyclic carbene (NHC) catalysis and transition metal (TM) catalysis



We commenced the evaluation of our hypothesis by the reaction of 4-nitrophenyl-3-phenylbut-2-enoate 1a with 1tosyl-2-vinylaziridine 2 using Cs₂CO₃ as the base in THF under NHC/palladium co-catalysis (for details, see the Supporting Information (SI)). We were delighted to find that the reaction proceeded smoothly to give the formal [3 +2] annulation product **3a** in 11% yield, rather than the [4 + 3]annulation product 4a, in the presence of carbene precursor A and $Pd_2(dba)_3$ ·CHCl₃ with PPh₃ as the ligand (Table 1, entry 1). Subsequent screening of other carbene precursors indicated that NHC precatalyst D displayed the best catalytic efficiency for this [3 + 2] annulation reaction (Table 1, entries 2–4). Notably, the control experiments showed that 3a was not formed at all in the absence of either NHC or a palladium catalyst, indicating the synergistically catalytic roles of both carbene and the palladium catalyst (Table 1, entries 5 and 6). Different ligands, under the co-catalysis of carbene D and Pd₂(dba)₃•CHCl₃, were then further examined (Table 1, entries 7–13). As a result, tricyclohexylphosphine (PCy_3) proved to be the best ligand for this annulation, and 38% yield of 3a could be obtained (Table 1, entry 9). Encouraged by this result, we then tested various other palladium catalysts (Table 1, entries 14-20). Gratifyingly, a 46% yield of 3a could be achieved when [Pd(allyl)Cl]₂ was used as the metallic catalyst in combination with PCy₃ (Table 1, entry 15). Lastly, we found that an excess amount of 2 was critical to this transformation, because of the dimerization of 2, and the reaction yield could be enhanced to 75% (Table 1, entry 20).

Table 1. Selected Results for the Optimization of Reaction Conditions a

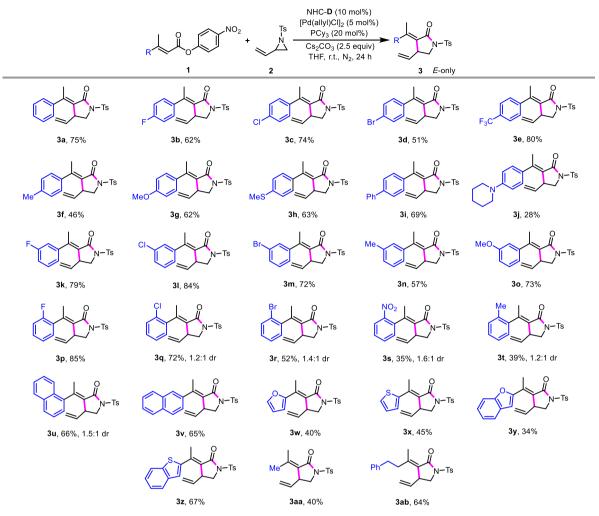
Ph	NO ₂	NHC-precat. (10 mol% Cs ₂ CO ₃ (2.5 equiv) Pd-cat. (5 mol%) ligand (20 mol%) 2 THF, r.t., N ₂ , 24 h	Ph Bh Ja E-only	Ph 4a
			F F Cy-	Cy P <i>i</i> -Pr <i>j</i> -Pr <i>j</i> -Pr xphos
entry	NHC-precat.	Pd-cat.	ligand	yield of $3a^b$
1	Α	Pd ₂ (dba) ₃ •CHCl ₃	PPh ₃	11
2	В	Pd ₂ (dba) ₃ •CHCl ₃	PPh ₃	trace
3	С	$Pd_2(dba)_3 \cdot CHCl_3$	PPh ₃	0
4	D	Pd ₂ (dba) ₃ •CHCl ₃	PPh ₃	21
5	-	$Pd_2(dba)_3 \cdot CHCl_3$	PPh ₃	0
6	D	-	-	0
7	D	$Pd_2(dba)_3 \cdot CHCl_3$	$P(4-FPh)_3$	13
8	D	Pd ₂ (dba) ₃ •CHCl ₃	$P(4-MePh)_3$	26
9	D	$Pd_2(dba)_3 \cdot CHCl_3$	PCy ₃	38
10	D	Pd ₂ (dba) ₃ •CHCl ₃	P ^t BuCy ₂	14
11	D	Pd ₂ (dba) ₃ •CHCl ₃	xphos	0
12	D	$Pd_2(dba)_3 \cdot CHCl_3$	P ^t Bu ₃	0
13	D	$Pd_2(dba)_3 \cdot CHCl_3$	P ⁿ Hex ₃	30
14 ^c	D	$Pd(PPh_3)_2Cl_2$	PCy ₃	0
15	D	$[Pd(allyl)Cl]_2$	PCy ₃	46
16 ^c	D	$Pd(PPh_3)_4$	PCy ₃	38
17 ^c	D	$Pd(TFA)_2$	PCy ₃	0
18 ^c	D	$Pd(tBu_3P)_2$	PCy ₃	38
19 ^c	D	$Pd(OAc)_2$	PCy ₃	14
20 ^d	D	$[Pd(allyl)Cl]_2$	PCy ₃	75

^{*a*}Reaction conditions: **1a** (0.12 mmol, 1.2 equiv); **2** (0.1 mmol, 1.0 equiv); NHC-precat. (10 mol%); Cs_2CO_3 (2.5 equiv); Pd-cat. (5 mol%); ligand (20 mol%); THF, 2 mL; N_2 ; rt; 24 h. ^{*b*}Isolated yields. ^cligand (10 mol%). ^{*d*}**1a** (0.2 mmol, 1.0 equiv); **2** (0.6 mmol, 3.0 equiv); THF, 4 mL. dba, dibenzylideneacetone; Cy, cyclohexyl; "Hex, *n*-hexyl; TFA, trifluoroacetate.

Note that the configuration of the C==C double bond of the product can be established by NMR analysis of 3a, which is a known compound.²⁴

With the optimal reaction conditions in hand, the substrate scope of the annulation between 3-substituted-(4nitrophenyl)but-2-enoates 1 and 1-tosyl-2-vinylaziridine 2 was investigated as shown in Scheme 2. Inspiringly, a wide array of 3-(substituted phenyl)-(4-nitrophenyl)but-2-enoates were compatible to the present annulation system. The nature and position of the substituents on the phenyl ring seemed to have certain impact on the reaction results. Substrates bearing either electron-withdrawing groups or electron-donating groups at the para position of the phenyl ring could be welltolerated, affording the corresponding desired products 3b-3i in moderate to good yields. However, when a cyclic amino group was anchored on the para position of the phenyl ring, only 28% yield was achieved for the corresponding product 3j. In terms of 3-(meta-substituted phenyl)-(4-nitrophenyl)but-2enoates, substrates with electron-withdrawing groups, such as halogens (3k-3m) and methoxy group (3o) were found to be better than those bearing an electron-donating group, such as methyl (3n). When it comes to 3-(ortho-substituted phenyl)-(4-nitrophenyl)but-2-enoates, the steric effect from the substituents was obvious. The yields decreased significantly with the increase of the volume of substituents at the ortho

Scheme 2. Substrate Scope^a



^aReaction conditions: 1 (0.2 mmol, 1.0 equiv), 2 (0.6 mmol, 3.0 equiv), NHC-D (10 mol %), Cs_2CO_3 (2.5 equiv), $[Pd(allyl)]Cl_2$ (5 mol %), PCy₃ (20 mol %), THF, 4 mL, rt, 24 h; isolated yields are based on 1.

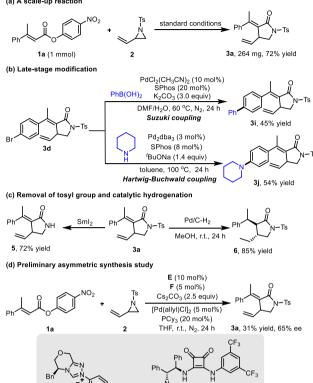
position (3p-3t). Furthermore, atropisomerism was observed for products 3q-3t bearing bigger groups at ortho position, because of the generation of a chiral axial. Similarly, the reaction of 3-(1-naphthyl) enoate gave the corresponding product 3u in 66% yield with a diastereomeric ratio (dr) of 1.5:1. 3-(2-Naphthyl and 2-heteroaryl phenyl) enoates were also applicable to the present protocol. Although lower yields were obtained for 2-(2-furyl)-, 2-(2-thienyl)-, and 2-(2benzofuryl)-substituted enoates (3w-3y), upper-medium yields could be obtained for 2-naphthyl and 2-benzothienylderived enoates (3v and 3z). Lastly, 3-alkyl-(4-nitrophenyl) but-2-enoates were also tested to evaluate the generality of our present reaction system. Fortunately, medium yields could be achieved under the identical reaction conditions for two typical aliphatic substrates (3aa and 3ab).

To further explore the synthetic utility of this methodology, a scale-up synthesis of product **3a** and several synthetic applications were then performed (see Scheme 3). Product **3a** was obtained in a maintained yield with a 1 mmol scale under the standard conditions (see Scheme 3a). Moreover, **3d** could be used for late-stage modification for different synthetic purposes. For example, the Suzuki coupling with phenylboronic acid and Hartwig-Buchwald coupling with piperidine could proceed smoothly to afford the target compounds **3i** and **3**j, respectively (Scheme 3b), indicating the generality for latestage functionalization of the annulation products. In addition, the *N*-protecting group could also be easily removed under mild conditions to produce pyrrolidin-2-one derivative **5** in a good yield (see Scheme 3c). Under catalytic hydrogenation conditions, the two C=C double bonds of product **3a** were completely reduced together, giving a *trans*-reduced product **6** in a high yield (Scheme 3c). Finally, a preliminary asymmetric annulation study was performed (Scheme 2d). Unsatisfyingly, after the screening of various reaction parameters (for details, see the SI), **3a** could be obtained in 31% yield with 65% enantiomeric excess (ee) in the presence of chiral carbene precursor **E**, using chiral diamine **F** as the additive.

A tentative mechanism of the NHC/palladium co-catalyzed annulation of vinyl enolates with 1-tosyl-2-vinylaziridine was proposed as outlined in Scheme 4.^{7a,13-15} Herein, this synergistic annulation was successfully realized through the concomitant catalytic generation of two reactive species: a nucleophilic NHC-bound vinyl enolate II and a highly electrophilic Pd- π -allyl species IV. The nucleophilic attack of carbene **D**' generated upon the deprotonation of **D** with Cs₂CO₃ to 1a afforded the acyl azolium intermediate I. Subsequent deprotonation of γ -H of I with a base produced NHC-bound vinyl enolate II. In a parallel palladium catalytic

Scheme 3. Synthetic Applications and Preliminary Enantioselective Study





cycle, through the coordination of 2 to the palladium catalyst, an active electrophilic Pd- π -allyl species IV could be generated via the ring opening of aziridine. At this point, the NHCbound vinyl enolate II could undergo a nucleophilic attack to the *in-situ*-formed Pd- π -allyl species IV with the formation of intermediate V. Following this C–C bond formation, the release of the Pd(0) gave rise to acyl azolium VI. This species then underwent a *N*-acylation annulation to produce the cyclic intermediate VII and regenerate carbene D' for the next

Scheme 4. Proposed Reaction Mechanism

catalytic cycle. Lastly, tautomerization of VII furnished the more stable final product 3a.

In conclusion, we have demonstrated the first [3 + 2] annulation reaction of NHC-bound vinyl enolates with a Pd- π allyl species via a dual organo-metal catalytic process. Under the properly optimized conditions, the free carbene catalyst and the Pd catalyst could work well independently to accomplish their own catalytic roles without quenching each other. The rapid and stereoselective construction of (*E*)-3ethylidene-4-vinylpyrrolidin-2-one skeletons could be achieved successfully in medium to good yields with excellent compatibility of functional groups. In addition, encouraged by the preliminary results of the asymmetric study, further enantioselective annulation of NHC-bound vinyl enolates with diverse transition metal- π -allyl species and other synthetic applications are underway in our laboratory.

ASSOCIATED CONTENT

3 Supporting Information

The Supporting Information is available free of charge at https://pubs.acs.org/doi/10.1021/acs.orglett.0c02935.

General information, experimental procedures, compound characterizations, and NMR spectra (PDF)

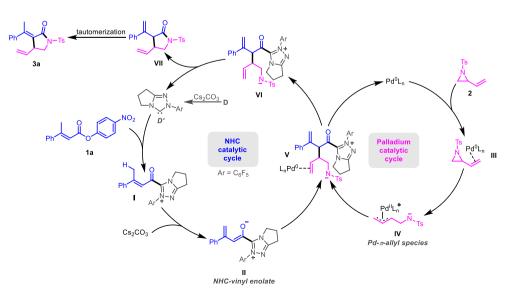
AUTHOR INFORMATION

Corresponding Author

Ding Du – Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China; ⊚ orcid.org/0000-0002-4615-5433; Email: ddmn9999@cpu.edu.cn

Authors

- Jian Gao Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China; Orcid.org/0000-0003-0071-1972
- Jianming Zhang Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China



Shuaishuai Fang – Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China

Jie Feng – Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China; Orcid.org/0000-0002-8662-2198

Tao Lu – Department of Chemistry, State Key Laboratory of Natural Medicines, China Pharmaceutical University, Nanjing 210009, People's Republic of China

Complete contact information is available at: https://pubs.acs.org/10.1021/acs.orglett.0c02935

Author Contributions

All authors have given approval to the final version of the manuscript.

Notes

The authors declare no competing financial interest.

ACKNOWLEDGMENTS

We are grateful to the National Natural Science Foundation of China (Nos. 21572270 and 21702232) and the "Double First-Class" University Project (Nos. CPU2018-GY02 and CPU2018GY35) for financial support.

REFERENCES

(1) (a) Magano, J.; Dunetz, J. R. Chem. Rev. 2011, 111, 2177.
(b) Crawley, M. L.; Trost, B. M., Ed.; Applications of Transition Metal Catalysis in Drug Discovery and Development: An Industrial Perspective; Wiley, 2012.

(2) (a) Movassaghi, M.; Jacobsen, E. N. Science 2002, 298, 1904.
(b) List, B.; Yang, J. W. Science 2006, 313, 1584. (c) MacMillan, D. W. C. Nature 2008, 455, 304.

(3) (a) Du, Z.; Shao, Z. Chem. Soc. Rev. 2013, 42, 1337. (b) Chen, D. F.; Han, Z. Y.; Zhou, X. L.; Gong, L. Z. Acc. Chem. Res. 2014, 47, 2365. (c) Afewerki, S.; Cordova, A. Chem. Rev. 2016, 116, 13512. (d) Kim, D. S.; Park, W. J.; Jun, C. H. Chem. Rev. 2017, 117, 8977. (e) Allen, A. E.; Macmillan, D. W. C. Chem. Sci. 2012, 3, 633.

(4) For selected reviews, see: (a) Enders, D.; Niemeier, O.; Henseler, A. Chem. Rev. 2007, 107, 5606. (b) Bugaut, X.; Glorius, F. Chem. Soc. Rev. 2012, 41, 3511. (c) Grossmann, A.; Enders, D. Angew. Chem., Int. Ed. 2012, 51, 314. (d) Hopkinson, M. N.; Richter, C.; Schedler, M.; Glorius, F. Nature 2014, 510, 485. (e) Flanigan, D. M.; Romanov-Michailidis, F.; White, N. A.; Rovis, T. Chem. Rev. 2015, 115, 9307. (f) Zhang, C.; Hooper, J. F.; Lupton, D. W. ACS Catal. 2017, 7, 2583. (g) Chen, S.; Shi, Y. H.; Wang, M. Chem. - Asian J. 2018, 13, 2184. (h) Murauski, K. J. R.; Jaworski, A. A.; Scheidt, K. A. Chem. Soc. Rev. 2018, 47, 1773. (i) Mondal, S.; Yetra, S. R.; Mukherjee, S.; Biju, A. T. Acc. Chem. Res. 2019, 52, 425. (j) Ohmiya, H. ACS Catal. 2020, 10, 6862.

(5) (a) Fortman, G. C.; Nolan, S. P. Chem. Soc. Rev. 2011, 40, 5151.
(b) Janssen-Muller, D.; Schlepphorst, C.; Glorius, F. Chem. Soc. Rev. 2017, 46, 4845.

(6) (a) Lebeuf, R.; Hirano, K.; Glorius, F. Org. Lett. 2008, 10, 4243.
(b) Nagao, K.; Ohmiya, H. Top Curr. Chem. (Z) 2019, 377, 35.

(7) (a) Liu, K.; Hovey, M. T.; Scheidt, K. A. Chem. Sci. 2014, 5, 4026. (b) Bai, Y.; Xiang, S.; Leow, M. L.; Liu, X. W. Chem. Commun. 2014, 50, 6168.

(8) Yasuda, S.; Ishii, T.; Takemoto, S.; Haruki, H.; Ohmiya, H. Angew. Chem., Int. Ed. 2018, 57, 2938.

(9) Haruki, H.; Yasuda, S.; Nagao, K.; Ohmiya, H. *Chem. - Eur. J.* 2019, 25, 724.

(10) Ohnishi, N.; Yasuda, S.; Nagao, K.; Ohmiya, H. Asian J. Org. Chem. 2019, 8, 1133.

(11) Takemoto, S.; Ishii, T.; Yasuda, S.; Ohmiya, H. Bull. Chem. Soc. Jpn. 2019, 92, 937.

- (12) Namitharan, K.; Zhu, T.; Cheng, J.; Zheng, P.; Li, X.; Yang, S.; Song, B. A.; Chi, Y. R. *Nat. Commun.* **2014**, *5*, 3982.
- (13) Guo, C.; Fleige, M.; Janssen-Muller, D.; Daniliuc, C. G.; Glorius, F. J. Am. Chem. Soc. 2016, 138, 7840.
- (14) Guo, C.; Janssen-Muller, D.; Fleige, M.; Lerchen, A.; Daniliuc, C. G.; Glorius, F. J. Am. Chem. Soc. 2017, 139, 4443.
- (15) Singha, S.; Patra, T.; Daniliuc, C. G.; Glorius, F. J. Am. Chem. Soc. 2018, 140, 3551.
- (16) Singha, S.; Serrano, E.; Mondal, S.; Daniliuc, C. G.; Glorius, F. Nat. Catal. 2020, 3, 48.

(17) Zhang, Z. J.; Zhang, L.; Geng, R. L.; Song, J.; Chen, X. H.; Gong, L. Z. Angew. Chem., Int. Ed. **2019**, 58, 12190.

(18) Zhou, L.; Wu, X.; Yang, X.; Mou, C.; Song, R.; Yu, S.; Chai, H.; Pan, L.; Jin, Z.; Chi, Y. R. Angew. Chem., Int. Ed. 2020, 59, 1557.

(19) Yang, W.; Ling, B.; Hu, B.; Yin, H.; Mao, J.; Walsh, P. J. Angew. Chem., Int. Ed. 2020, 59, 161.

(20) (a) Chen, X. Y.; Liu, Q.; Chauhan, P.; Enders, D. Angew. Chem., Int. Ed. **2018**, 57, 3862. (b) Liu, L.; Guo, D.; Wang, J. Org. Lett. **2020**, 22, 7025.

(21) Fang, S.; Jin, S.; Ma, R.; Lu, T.; Du, D. Org. Lett. 2019, 21, 5211.

(22) (a) Rivinoja, D. J.; Gee, Y. S.; Gardiner, M. G.; Ryan, J. H.; Hyland, C. J. T. *ACS Catal.* **2017**, *7*, 1053. (b) Spielmann, K.; van der Lee, A.; de Figueiredo, R. M.; Campagne, J. M. *Org. Lett.* **2018**, *20*, 1444.

(23) Atbrecht, L.; Albrecht, A.; Janecki, T. In *Natural Lactones and Lactams: Synthesis, Occurrence and Biological Activity*, Vol. 147; Janecki, T., Ed.; Wiley–VCH, 2014.

(24) Zhu, G.-G.; Tong, X.-F.; Cheng, J.; Sun, Y.-H.; Li, D.; Zhang, Z.-G. J. Org. Chem. 2005, 70, 1712.