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Palladium-Catalyzed Synthesis of Indoles and Isoquinolines with *in situ* Generated Phosphinimine

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ABSTRACT. A palladium-catalyzed synthesis of poly-substituted indoles and isoquinolines through the coupling of arylbromides with 2-alkynyl arylazides or 2-alkynyl benzylazides has been developed. This method provides straightforward access to indoles and isoquinolines with high efficiency and excellent functional group compatibility. In this transformation, the iminophosphorane *in situ* generated from azides is served as the nucleophile that attacks the alkyne moiety in the cyclization process.

INTRODUCTION

Transition-metal-catalyzed carbene transfer reactions constitute one of the major domains of modern synthetic organic chemistry.¹ While diazo compounds are generally recognized as the most common precusors for the generation of metal carbene species, other metal carbene precusors, such as alkynes, cyclopropenes and enynones have also attracted significant attentions in recent years.² In particular, alkynes are known to serve as carbene precursors by reacting with nucleophiles bearing a leaving group under Au- or Pt-catalyzed conditions (Scheme 1a).³ Alkynes also serve as precursors for generating α -oxo and α -imino gold carbenes, which undergo typical carbene transformations, such as X-H bond insertion, 1,2-migration, cyclopropanation, ylide reaction and others.⁴ The azido moiety has been known as a useful nucleophile bearing an excellent leaving group (N₂). Consequently, the generation of metal carbene intermediate from azide and alkyne moiety has been recently established for the synthesis of nitrogen-containing heterocycles.⁵

Scheme 1. Alkynes as Metal Carbene Precursors



The Journal of Organic Chemistry

On the other hand, diazo compounds have been extensively explored as palladium carbene precursors for cross-coupling reactions.^{6,7} Recently, we have also demonstrated the use of ene-yne-ketones and allenyl ketones as the palladium carbene precursors in cross-coupling reactions.⁸ As a continuation of our interest in carbene-based coupling reactions, we have conceived that azido moiety may serve as nucleophiles for generating palladium carbenes from alkynes, similar to the coressponding Au- and Pt-catalyzed reactions (Scheme 1b).³ With 2-alkynyl arylazides as the substrates, the palladium carbene generated may be involved in cross-coupling reactions for the synthesis of poly-substituted indole,⁹ as proposed in Scheme 1c. Although generating palladium carbene from alkynes has been reported,¹⁰ the combination of such type of transformation with cross-coupling is not known in the literature. Similar strategy should also be applied to the synthesis of isoquinolines by employing 2-alkynyl benzylazides as the substrates instead of 2-alkynyl arylazides.¹¹

RESULTS AND DISCUSSIONS

To achive the proposed transformation, we have initially investigated the cross-coupling reaction by using alkynyl-substituted azides **1a** and aryl bromide **2a** as the substrates, and Et_3B was used as the reductant (Table 1, entry 1). The expected indole product **3a** could be isolated in 22% yield with 5 mol% Pd₂dba₃ and 20 mol% dpppe [1,5-bis(diphenylphosphino)pentane]. We suspected that the low yield might be attributed to the deactivation of the palladium catalyst. Indeed the reaction could be improved by increasing the loading of Pd₂dba₃ and dpppe (Table 1, entry 2). However, when the loading of dpppe was reduced, a sharp decrease of the yield was observed (Table 1, entry 3). On the contrast, only lowing the loading of Pd₂dba₃ does not lead to the reduction of the yield (Table 1, entries 4 and 5). It was thus speculated that dpppe not only acted as the ligand but also played other vital role in this transformation. Since dpppe may play the role as a reductant in the reaction went smoothly to afford the indole **3a** in an improved yield (Table 1, entry 6). When the concentration of the substrates was increased, the yield could be slightly improved with 1.5 equiv of base (Table 1, entry 7). The yield was diminished slightly

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when 1 equiv of water was added (Table 1, entry 8). Notably, dpppe could be replaced by PPh₃, but affording the product with slightly diminished yield (Table 1, entry 9). Moreover, base was found necessary for the reaction (Table 1, entry 10). In additon to LiO^{*t*}Bu, other types of bases could also make the reaction work smoothly (Table 1, entries 11-16).

Table 1. Optimization of the Palladium-Catalyzed Synthesis of Indole^a

1	ⁿ Bu + N ₃	Br Pd ₂ c LiO ^t Bu toluene, OMe 2a	dba₃, [P] <u>a, additive</u> 24 h,100 °C	OMe , nBu H 3a	
	Pd ₂ dba ₃		additive	,	
entry	(mol%)	base (equiv)	(equiv)	[P] (mol%) ^a	yield(%) ^b
1	5	$LiO^{t}Bu$ (2)	Et ₃ B (1.5)	dpppe (20)	22
2	10	$LiO^{t}Bu$ (2)	Et ₃ B (1.5)	dpppe (40)	58
3	10	$LiO^{t}Bu$ (2)	Et ₃ B (1.5)	dpppe (20)	25
4	5	LiO ^t Bu (2)	Et ₃ B (1.5)	dpppe (40)	68
5	2.5	LiO ^t Bu (2)	Et ₃ B (1.5)	dpppe (40)	69
6	2.5	LiO ^t Bu (2)	none	dpppe (50)	72
7^c	2.5	LiO ^t Bu (1.5)	none	dpppe (50)	75
8	2.5	LiO ^t Bu (1.5)	H ₂ O (1)	dpppe (50)	52
9^c	2.5	LiO ^t Bu (1.5)	none	PPh ₃ (100)	60

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The Journal of Organic Chemistry

10	2.5	none	none	PPh ₃ (100)	trace
11	2.5	NaO ^t Bu (1.5)	none	PPh ₃ (100)	33 ^e
12	2.5	NaOMe (1.5)	none	PPh ₃ (100)	67 ^e
13	2.5	KOMe (1.5)	none	PPh ₃ (100)	53 ^e
14	2.5	LiOMe (1.5)	none	PPh ₃ (100)	23 ^e
15	2.5	K ₂ CO ₃ (1.5)	none	PPh ₃ (100)	29 ^e
16	2.5	Cs ₂ CO ₃ (1.5)	none	PPh ₃ (100)	69 ^e

^{*a*}If not otherwise noted, the reaction was performed with **1a** (0.1 mmol), **2a** (0.15 mmol) in toluene (2 mL) at 100 °C. ^{*b*}Yields of isolated product are given. ^{*c*}The reaction was carried out with **1a** (0.2 mmol), **2a** (0.3 mmol) in toluene (2 mL) at 100 °C. ^{*d*}dpppe = 1,5-bis(diphenylphosphino)pentane. ^{*e*}NMR yields of the products are given.

With the optimized reaction conditions in hand, we then proceeded to study the substrate scope of the reaction with a series of 2-alkynyl arylazides and aryl bromides (Scheme 2). The substrates with *para*-substituted aryl bromides all afforded the corresponding products in the good yields (Scheme 2, **3a-j**), except the *p*-CO₂Me substituted bromide **2g**. With *meta*-substituted bromides, the reaction also went smoothly to give the corresponding products in good yields (Scheme 2, **3k-n**). The low yield in the case of **2p** may be attributed to steric effect of the *ortho*-substitution (Scheme 2, **3p**). In addition, The reaction with 4-bromo-1, 1'-biphenyl and 2-bromonaphthalene could deliever the corresponding indole products in the good yields (Scheme 2, **3q, r**). Notably, the reaction with the azide substrate bearing aryl substituent also afforded the corresponding products in moderate to good yields (Scheme 2, **3t-w**). The structure of indole product **3t** was further confirmed by X-ray crystallography.¹³

Scheme 2. Reaction Scope of the Aryl Bromides^a



^{*a*}If not otherwise noted, the reaction conditions are as following: 2-alkynyl arylazides **1a-f** (0.2 mmol), ary bromides **2a-r** (0.3 mmol), Pd₂dba₃ (2.5 mol%), dpppe (50 mol%), LiO^{*t*}Bu (1.5 equiv), toluene (2 mL), 100 °C, 24 h. ^{*b*}All the yields refer to the isolated indole products.

For the reaction mechanism, it was initially considered that the azide moiety of the substrate attack the arylpalladium(II)-activated triple bond to afford the cyclized intermediate. However, because the arylazide moiety can easily react with phosphine to generate phosphinimine under the current reaction conditions (Staudinger reaction),^{14,15} it was thus speculated that the phosphinimine might be first formed and then the nucleophilic nitrogen of the phosphinimine attacked the arylpalladium(II)-activated triple bond. To substantiate such speculation, mechanistic experiments were carried out. The arylazide **1a** was

The Journal of Organic Chemistry

first subjected to the reaction with triphenylphosphine in dichloromethane. After stirring at room temperature for 6 h, the arylazide **1a** was completely disappeared. The solvent was removed and the formation of phosphinimine **4** was confirmed by ¹H and ¹³C NMR spectra. The crude phosphinimine was then submitted to the Pd-catalyzed coupling reaction with arylbromide. The expected indole product **3a** was obtained in 56% yield. Similar experiment with dpppe afforded the indole product **3a** in 72% yield (Scheme 3a).

Scheme 3. Control Experiments



In another control experiment, 2-alkynyl azidobenzene **1a** was replaced with 2-(hex-1-yn-1yl)aniline **1a'** under the standard conditions. The formation of **3a** was not observed, as shown in Scheme 3b. The experiment rules out the mechanism in which the azide is first converted to the amine *in situ* and then the cyclization occurs. Notably, in Cacchi's aminopalladation/reductive elimination domino reaction for the construction of indole rings, a trifluoroacetyl substituent on the nitrogen is required.^{9d,e} The proposed reaction mechanism is shown in Scheme 4. There are two possible pathways. In path *a*, arylpalladium(II) species **B** is first formed through oxidative addition of aryl bromide **2** to Pd(0) catalyst, while iminophosphorane **C** is formed *in situ* from 2-alkynyl arylazide by Staudinger reaction. Then, the *5-endo-dig* cyclized intermediate **D** is generated through nucleophilic attack of the arylpalladium(II)activated triple bond by the nitrogen of the phosphinimine moiety. Reductive elimination of

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intermediate **D** leads to the intermediate **J**, which is further converted to the product by hydrolysis. The alternative mechanism involving the carbene process is shown in path b. From intermediate **D**, palladium(II) carbene species **F** is formed through back electron donating from palladium. Subsequent migratory insertion gives intermediate **G**, which is isomerized to intermediate **H**. Protonation of the intermediate **H** affords product **3** and Pd(II) species, which is reduced to Pd(0) catalyst by **E**.

Scheme 4. Proposed Reaction Mechanism



Encourged by the above results, we further conceived to extend the reaction to the synthesis of poly-substituted isoquinolines with 2-alkynyl benzylazide **5a** (Table 2). However, when the reaction was performed under the standard conditions for indole synthesis as described above, the desired product **6a** was only obtained in 29% yield (Table 2, entry 1). We then optimized the reaction conditions byscreening different bases, however, with K_2CO_3 and KOMe as the bases, only trace amount of the products were detected (Table 2, entries 2, 3). To our delight, when dpppe was replaced by PPh₃, the yield could be improved to 39% (Table 2, entry 4). The reaction with other solvents, such as dioxane,

Page 9 of 27

The Journal of Organic Chemistry

MeCN and DCE, did not show improved results (Table 2, entries 5-7). Finally, it was observed that the yield of **6a** could be significantly improved by increasing the loading of arylbromide **2a** (Table 2, entries 8-11).

 Table 2. Reaction Condition Optimization for Isoquinoline Synthesis^a

	ⁿ Bu N ₃ 5a	+ Br 2a	Pd ₂ dba ₃ (2.5 mol%) [P], base solvent, 24 h, 100 °C	OMe nBu 6a	
entry	2a (equiv)	base (equiv)	solvent	[P] (mol%)	yield $(\%)^b$
1	1.5	$LiO^{t}Bu$ (1.5)	toluene	dpppe (50)	29
2^c	1.0	K ₂ CO ₃ (1.5)	toluene	dpppe (50)	trace
3 ^{<i>c</i>}	1.0	KOMe (1.5)	toluene	dpppe (50)	trace
4 ^{<i>c</i>}	1.0	LiO ^t Bu (3.0)	toluene	PPh ₃ (100)	39
5 ^{<i>c</i>}	1.0	LiO ^t Bu (2.0)	dioxane	PPh ₃ (100)	22
6 ^{<i>c</i>}	1.0	LiO ^t Bu (2.0)	MeCN	PPh ₃ (100)	trace
7^c	1.0	LiO ^t Bu (2.0)	DCE	PPh ₃ (100)	trace
8	1.5	LiO ^t Bu (2.8)	toluene	PPh ₃ (150)	45
9	2.0	LiO ^t Bu (2.8)	toluene	PPh ₃ (150)	58
10	2.5	LiO ^t Bu (2.8)	toluene	PPh ₃ (150)	68
11	3.0	LiO ^t Bu (2.8)	toluene	PPh ₃ (150)	76

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^{*a*}If not otherwise noted, the reaction was performed with **5a** (0.2 mmol), in toluene (1 mL) at 100 °C. ^{*b*}Yields refer to the isolated product. ^{*c*}The reaction was carried out with **5a** (0.1 mmol) in toluene (1 mL) at 100 °C.

With the optimized reaction conditions (with 3 equivalents of arylbromide) (Table 2, entry 11), we proceeded to investigate the scope of the reaction (Scheme 5). Moderate to good yields could be obtained with a series of arylbromides (Scheme 5, **6a-i**). Besides, the reaction also worked well with 2-alkynyl benzylazides bearing various substituents (Scheme 5, **6j-l**).

Scheme 5. The Scope for the Synthesis of Isoquinoline^{*a*}



^{*a*}If not otherwise noted, the reaction conditions are as following: 2-alkynyl benzylazides **5a-e** (0.2 mmol), ary bromides (0.6 mmol), Pd₂dba₃ (2.5 mol%), PPh₃ (1.5 equiv), LiO^{*t*}Bu (2.8 equiv), toluene (1 mL), 100 °C, 24 h. ^{*b*}All the yields refer to the isolated products.

The Journal of Organic Chemistry

For the synthesis of isoquinolines, same control experiments as above-mentioned were carried out and similar results were obtained as shown in Scheme 6. Therefore, the mechanism for isoquinoline formation should be similar as that shown in Scheme 4^{12} .

Scheme 6. Control Experiments for the Reaction of Isoquinolines Synthesis.



CONCLUSION

In conclusion, we have developed an efficient method for the synthesis of poly-substituted indoles and isoquinolines *via* the iminophosphorane intermediates generated *in situ* through Staudinger reaction. The iminophosphoranes are versatile synthetic intermediates which have found many applications in organic chemistry. The reaction of iminophosphorane with palladium-activated alkyne described herein opens up new possibilities for the application of iminophosphorane in developing new reactions.

EXPERIMENTAL SECTION

General Methods. All the reactions of palladium-catalyzed synthesis of poly-substituted indoles and isoquinolines were performed under nitrogen atmosphere in a flame-dried reaction tube. All the solvents were distilled under nitrogen atmosphere prior to use. Toluene was dried over Na with benzophenone-ketyl intermediate as indicator. For chromatography, 200-300 mesh silica gel was employed. Chemical shifts for ¹H NMR (400 MHz) and ¹³C NMR (100 MHz) were reported relative to those of tetramethylsilane (TMS): chemical shifts (δ) were reported in ppm, and coupling constants (J) are in

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Hertz (Hz). IR spectra were performed neat on an FT-IR spectrophotometer and reported in wave numbers, cm⁻¹. For HRMS measurements, the mass analyzer is FT-ICR. 2-Alkynyl azidobenzenes and 2-alkynyl benzylazides were prepared according to the literature procedures.¹⁶⁻¹⁸ Other starting materials were obtained from commercial suppliers and were used without further purification. PE: petroleum ether; EA: ethyl acetate.

General procedure for the synthesis of *poly*-substituted indoles. Under N₂ atmosphere, Pd₂dba₃ (4.6 mg, 0.005 mmol, 2.5 mmol%), dpppe (44 mg, 0.1 mmol, 50 mmol%), LiO'Bu (24 mg, 0.3 mmol) were successively added to a flame-dried 10 mL Schlenk tube. The reaction tube was degassed three times with N₂ and dry toluene (2 mL) was added by syringe. Then, the 2-alkynyl azidobenzene (0.2 mmol) and the arylbromides (0.3 mmol) were added with syringe, respectively (Note: 2-alkynyl azidobenzenes or the arylbromides in solid form were added to the reaction tube prior to LiO'Bu). The reaction was heated at 100 °C with stirring for 24 h. Then the reaction mixture was cooled to room temperature and then it was filtered through a short plug of silica gel with ethyl acetate as eluents. The solvent was removed in *vacuo* to leave a crude mixture, which was purified by column chromatography on silica gel with petroleum ether/ethyl acetate (10:1) to afford the pure indole product **3a-w**.

2-Butyl-3-(4-methoxyphenyl)-1H-indole (**3a**).^{5b} Yield: 42 mg (75%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.90 (s, 1H), 7.59 (d, J = 7.7 Hz, 1H), 7.41-7.38 (m, 2H), 7.28 (d, J = 7.9 Hz, 1H), 7.16-7.07 (m, 2H), 7.02-6.99 (m, 2H), 3.85 (s, 3H), 2.79 (t, J = 7.8 Hz, 2H), 1.67-1.59 (m, 2H), 1.39-1.30 (m, 2H), 0.87 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 157.9, 135.7, 135.1, 130.6, 128.1, 127.8, 121.4, 119.7, 118.8, 113.9, 110.3, 55.3, 32.0, 26.0, 22.5, 13.8; IR (film): 1330, 1459, 1487, 1557, 1613, 2931, 2957, 3408 cm⁻¹.

2-Butyl-3-(p-tolyl)-1H-indole (**3b**). Yield: 42 mg (79%), pale yellow liquid, $R_f = 0.5$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.86 (s, 1H), 7.62 (d, J = 7.8 Hz, 1H), 7.37 (d, J = 8.0, 2H), 7.29-7.25 (m, 3H), 7.17-7.07 (m, 2H), 2.81 (t, J = 7.8 Hz, 2H), 2.40 (s, 3H), 1.68-1.60 (m, 2H), 1.40-1.30 (m, 2H), 0.88 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 135.8, 135.4, 135.2, 132.4, 129.5, 129.2, 128.0, ACS Paragon Plus Environment

The Journal of Organic Chemistry

121.4, 119.7, 118.9, 114.2, 110.3, 32.0, 26.1, 22.5, 21.2, 13.8; HRMS (ESI, *m*/*z*): calcd for C₁₉H₂₂N [M+H]⁺ 264.1747, found 264.1750; IR (film): 1017, 1257, 1330, 1458, 1510, 1558, 2957, 3406 cm⁻¹.

2-Butyl-3-(4-(tert-butyl)phenyl)-1H-indole (3c). Yield:

48 mg (79%), pale yellow liquid, $R_f = 0.5$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.91 (s, 1H), 7.66 (d, J = 7.8 Hz, 1H), 7.48-7.41 (m, 4H), 7.30 (d, J = 7.9 Hz, 1H), 7.17-7.13 (m, 1H), 7.11-7.07 (m, 1H), 2.84 (t, J = 7.9 Hz, 2H), 1.71-1.63 (m, 2H), 1.43-1.33 (m, 11H), 0.90 (t, 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 148.5, 135.9, 135.2, 132.4, 129.1, 128.0, 125.3, 121.4, 119.7, 119.1, 114.2, 110.3, 34.5, 32.1, 31.4, 26.1, 22.6, 13.9; HRMS (ESI, m/z): calcd for C₂₂H₂₈N [M+H]⁺ 306.2216, found 306.2219; IR (film): 1268, 1330, 1363, 1459, 1509, 1563, 2960, 3408 cm⁻¹.

2-Butyl-3-(4-(trimethylsilyl)phenyl)-1H-indole (**3d**). Yield: 52 mg (81%), pale yellow liquid, $R_f = 0.5$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.89 (s, 1H), 7.66 (d, J = 7.8 Hz, 1H), 7.61 (d, J = 8.0 Hz, 2H), 7.48 (d, J = 8.0 Hz, 2H), 7.28 (d, J = 7.9 Hz, 1H), 7.18-7.07 (m, 2H), 2.83 (t, J = 7.9 Hz, 2H), 1.69-1.61 (m, 2H), 1.41-1.32 (m, 2H), 0.89 (t, J = 7.4 Hz, 3H), 0.32 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 137.3, 136.2, 135.9, 135.2, 133.5, 128.9, 127.8, 121.5, 119.8, 119.0, 114.3, 110.4, 32.1, 26.2, 22.6, 13.9, -1.0; HRMS (ESI, m/z): calcd for C₂₁H₂₈NSi [M+H]⁺ 322.1986, found 322.1988; IR (film): 839, 1108, 1248, 1329, 1459, 1600, 2956, 3409 cm⁻¹.

2-Butyl-3-(4-(trifluoromethyl)phenyl)-1H-indole (3e). Yield: 56 mg (88%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.05 (s, 1H), 7.70 (d, J = 8.2 Hz, 2H), 7.60 (t, J = 7.6 Hz, 3H), 7.35 (d, J = 8.0 Hz, 1H), 7.21-7.17 (m, 1H), 7.15-7.11 (m, 1H), 2.85 (t, J = 7.8 Hz, 2H), 1.72-1.64 (m, 2H), 1.42-1.33 (m, 2H), 0.90 (t, J = 7.3Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 139.4, 136.8, 135.2, 129.6, 127.8 (q, $J^2 = 32.3$ Hz), 127.5, 125.4 (q, $J^3 = 3.7$ Hz), 124.5 (q, $J^1 = 271.7$ Hz), 121.9, 120.5, 120.3, 118.6, 113.2, 110.5, 31.9, 26.1, 22.5, 13.8; HRMS (ESI, m/z): calcd for C₁₉H₁₉F₃N [M+H]⁺ 318.1464, found 318.1467; IR (film): 1321, 1407, 1440, 1459, 1560, 1615, 2958, 3400 cm⁻¹.

*4-(2-Butyl-1H-indol-3-yl)benzonitrile (***3***f)*. Yield: 47 mg (86%), yellow solid, mp: 120-121 °C; $R_f = 0.2$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.22 (s, 1H), 7.62 (d, J = 8.3 Hz, 2H), 7.50 (t, J = 8.7 Hz, 3H), 7.28 (d, J = 7.9 Hz, 1H), 7.14-7.09 (m, 1H), 7.07-7.03 (m, 1H), 2.76 (t, J = 7.8 Hz, 2H), 1.63-1.55 (m, 2H), 1.33-1.23 (m, 2H), 0.81 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 140.9, 137.3, 135.2, 132.3, 129.7, 127.0, 122.0, 120.4, 119.4, 118.3, 112.7, 110.7, 108.7, 31.8, 26.1, 22.4, 13.7; HRMS (ESI, m/z): calcd for C₁₉H₁₉N₂ [M+H]⁺ 275.1543, found 275.1545; IR (film): 1174, 1262, 1460, 1504, 1605, 2231, 2959, 3336 cm⁻¹.

*Methyl 4-(2-butyl-1H-indol-3-yl)benzoate (3g).*¹⁹ Yield: 24 mg (40%), pale yellow solid, mp: 121-122 ^oC; $R_f = 0.2$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.18 (s, 1H), 8.14-8.12 (m, 2H), 7.64 (d, J = 7.8 Hz, 1H), 7.57 (d, J = 8.4 Hz, 2H), 7.35 (d, J = 7.9 Hz, 1H), 7.21-7.11 (m, 2H), 3.95 (s, 3H), 2.86 (t, J = 7.8, 2H), 1.72-1.64 (m, 2H), 1.41-1.31 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 167.3, 140.7, 136.9, 135.2, 129.8, 129.2, 127.4, 127.2, 121.8, 120.2, 118.7, 113.5, 110.5, 52.0, 31.9, 26.2, 22.5, 13.8; IR (film): 1287, 1309, 1437, 1459, 1606, 1700, 2955, 3372 cm⁻¹.

2-Butyl-3-(4-fluorophenyl)-1H-indole (**3h**). Yield: 41 mg (77%), pale yellow solid, mp: 49-50 °C; R_f = 0.4 (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.91 (s, 1H), 7.55 (d, J = 7.8 Hz, 1H), 7.44-7.39 (m, 2H), 7.30 (d, J = 8.0 Hz, 1H), 7.18-7.08 (m, 4H), 2.79 (t, J = 7.8 Hz, 2H), 1.67-1.60 (m, 2H), 1.39-1.30 (m, 2H), 0.88 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 161.4 (d, J = 244.4 Hz), 136.0, 135.1, 131.3 (d, J = 3.3 Hz), 131.0 (d, 7.6 Hz), 127.9, 121.6, 119.9, 118.6, 115.3 (d, J = 21,2 Hz), 113.4, 110.4, 31.9, 26.0, 22.4, 13.8; HRMS (ESI, m/z): calcd for C₁₈H₁₉FN [M+H]⁺ 268.1496, found 268.1497; IR (film): 1330, 1459, 1506, 1560, 2861, 2929, 2957, 3401 cm⁻¹.

2-Butyl-3-(4-chlorophenyl)-1H-indole (**3i**).¹⁹ Yield: 48 mg (85%), pale yellow solid, mp: 41-43 °C; R_f = 0.4 (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.94 (s, 1H), 7.57 (d, J = 7.8 Hz, 1H), 7.42-7.38 (m, 4H), 7.31 (d, J = 7.9 Hz, 1H), 7.21-7.08 (m, 2H), 2.79 (t, J = 7.8 Hz, 2H), 1.67-1.60 (m, 2H), 1.39-1.30 (m, 2H), 0.88 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 136.2, 135.1, 134.0, 131.6, 130.8,

The Journal of Organic Chemistry

128.6, 127.7, 121.7, 120.04, 118.6, 113.2, 110.4, 31.9, 26.0, 22.5, 13.8; IR (film): 1186, 1258, 1329, 1458, 1491, 1555, 2957, 3404 cm⁻¹.

3-(4-Bromophenyl)-2-butyl-1H-indole (3j). Yield: 48 mg (73%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.97 (s, 1H), 7.58-7.55 (m, 3H), 7.35-7.31 (m, 3H), 7.17 (t, J = 7.5 Hz, 1H), 7.10 (t, J = 7.4 Hz, 1H), 2.80 (t, J = 7.8 Hz, 2H), 1.68-1.61 (m, 2H), 1.40-1.30 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 136.2, 135.1, 134.4, 131.6, 131.2, 127.6, 121.7, 120.1, 119.7, 118.6, 113.3, 110.4, 31.9, 26.0, 22.5, 13.8; HRMS (ESI, m/z): calcd for C₁₈H₁₉BrN [M+H]⁺ 328.0695, found 328.0697; IR (film): 1186, 1257, 1329, 1458, 1488, 1552, 2958, 3406 cm⁻¹.

2-Butyl-3-(3-methoxyphenyl)-1H-indole (**3**k). Yield: 44 mg (78%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.98 (s, 1H), 7.65 (d, J = 7.8 Hz, 1H), 7.38-7.31 (m, 2H), 7.18-7.14 (m, 1H), 7.12-7.05 (m, 3H), 6.88-6.85 (m, 1H), 3.85 (s, 3H), 2.86(t, J = 7.8 Hz, 2H), 1.71-1.63 (m, 2H), 1.42-1.33 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 159.7, 136.9, 136.1, 135.1, 129.4, 127.8, 122.1, 121.5, 119.9, 118.9, 115.1, 114.3, 111.5, 110.3, 55.2, 32.0, 26.1, 22.5, 13.8; HRMS (ESI, *m/z*): calcd for C₁₉H₂₂NO [M+H]⁺ 280.1696, found 280.1696; IR (film): 1262, 1329, 1419, 1490, 1571, 1603, 2956, 3406 cm⁻¹.

3-(3-(1,3-Dioxolan-2-yl)phenyl)-2-butyl-1H-indole (3l). Yield: 35 mg (55%), pale yellow liquid, $R_f = 0.2$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.02 (s, 1H), 7.62-7.60 (m, 2H), 7.49-7.46 (m, 2H), 7.44-7.41 (m, 1H), 7.31 (d, J = 7.9 Hz, 1H), 7.17-7.13 (m, 1H), 7.11-7.07 (m, 1H), 5.89 (s, 1H), 4.16-4.13 (m, 2H), 4.06-4.03 (m, 2H), 2.82 (t, J = 7.8 Hz, 2H), 1.70-1.62 (m, 2H), 1.40-1.31 (m, 2H), 0.88 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 138.1, 136.2, 135.6, 135.2, 130.4, 128.5, 127.9, 127.7, 123.9, 121.5, 119.9, 118.8, 114.1, 110.3, 103.9, 65.3, 32.0, 26.0, 22.5, 13.8; HRMS (ESI, m/z): calcd for C₂₁H₂₄NO₂ [M+H]⁺ 322.1802, found 322.1801; IR (film): 1327, 1378, 1459, 1491, 1601, 1668, 2956, 3396 cm⁻¹.

2-Butyl-3-(3-(trifluoromethyl)phenyl)-1H-indole (**3m**).²⁰ Yield: 50 mg (78%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.00 (s, 1 H), 7.75 (s, 1H), 7.67-7.64 (m, 1H), 7.60-7.54 (m, 3H), 7.33 (d, J = 7.9 Hz, 1H), 7.21-7.11 (m, 2H), 2.82 (t, J = 7.8 Hz, 2H), 1.70-1.63 (m, 2H), 1.41-1.32 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 136.6, 136.3, 135.2, 132.7, 130.8 (q, J = 31.9 Hz), 128.9, 127.5, 126.1 (q, J = 3.8 Hz), 124.3 (q, J = 272.4 Hz), 122.5 (q, J = 3.8 Hz), 121.9, 120.3, 118.4, 113.1, 110.5, 31.9, 26.0, 22.4, 13.7; IR (film): 1322, 1460, 1492, 1558, 1612, 2931, 2958, 3396 cm⁻¹.

2-Butyl-3-(3-chlorophenyl)-1H-indole (**3n**). Yield: 48 mg (84%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.95 (s, 1H), 7.60 (d, J = 7.8 Hz, 1H), 7.47-7.46 (m, 1H), 7.36-7.35 (m, 2H), 7.31 (d, J = 7.9 Hz, 1H), 7.28-7.25 (m, 1H), 7.21-7.15 (m, 1H), 7.13-7.09 (m, 1H), 2.81 (t, J = 7.8 Hz, 2H), 1.68-1.61 (m, 2H), 1.40-1.31 (m, 2H), 0.89 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 137.4, 136.4, 135.1, 134.2, 129.7, 129.4, 127.7, 127.6, 125.9, 121.7, 120.1, 118.6, 113.1, 110.4, 31.9, 26.0, 22.4, 13.8; HRMS (ESI, *m*/*z*): calcd for C₁₈H₁₉ClN [M+H]⁺ 284.1201, found 284.1204; IR (film): 1253, 1327, 1404, 1458, 1561, 1596, 2957, 3402 cm⁻¹.

2-(2-Butyl-1H-indol-3-yl)benzonitrile (**3o**). Yield: 41 mg (75%), pale yellow liquid, $R_f = 0.2$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.30 (s, 1H), 7.79-7.77 (m, 1H), 7.66-7.61 (m, 1H), 7.56-7.54 (m, 1H), 7.42-7.38 (m, 2H), 7.31 (d, J = 7.9 Hz, 1H), 7.18-7.08 (m, 2H), 2.83-2.67 (m, 2H), 1.69-1.48 (m, 2H), 1.30-1.19 (m, 2H), 0.79 (t, J = 7.4 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 139.7, 137.8, 135.2, 133.3, 132.5, 131.9, 127.8, 126.7, 121.8, 120.1, 119.0, 118.3, 113.5, 111.0, 110.6, 31.2, 26.4, 22.2, 13.6; HRMS (ESI, m/z): calcd for C₁₉H₁₉N₂ [M+H]⁺ 275.1543, found 275.1545; IR (film): 1017, 1327, 1459, 1491, 1597, 2226, 2957, 3390 cm⁻¹.

3-(2-Bromophenyl)-2-butyl-1H-indole (3p). Yield: 27 mg (42%), pale yellow liquid, $R_f = 0.5$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.99 (s, 1H), 7.71 (d, J = 7.9Hz, 1H), 7.36-7.33 (m, 3H), 7.28 (d, J = 7.8 Hz, 1H), 7.24-7.19 (m, 1H), 7.18-7.14 (m, 1H), 7.07 (t, J = 7.5 Hz, 1H), 2.76-2.60 (m, 2H), 1.66-1.53 (m, 2H), 1.34-1.26 (m, 2H), 0.84 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 136.7, ACS Paragon Plus Environment

136.3, 134.8, 133.1, 132.8, 128.4, 128.2, 127.0, 125.7, 121.4, 119.7, 119.2, 114.1, 110.3, 31.4, 26.4, 22.3, 13.7; HRMS (ESI, m/z): calcd for C₁₈H₁₉BrN [M+H]⁺ 328.0695, found 328.0697; IR (film): 1025, 1056, 1256, 1330, 1458, 1474, 2955, 3407 cm⁻¹.

3-([1,1'-Biphenyl]-4-yl)-2-butyl-1H-indole (**3***q*). Yield: 54 mg (82%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.93 (s, 1H), 7.70-7.66 (m, 5H), 7.57-7.55 (m, 2H), 7.45 (t, J = 7.6 Hz, 2H), 7.35-7.30 (m, 2H), 7.19-7.10 (m, 2H), 2.86 (t, J = 7.8 Hz, 2H), 1.71-1.63 (m, 2H), 1.42-1.33 (m, 2H), 0.90 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 141.0, 138.5, 136.2, 135.2, 134.6, 129.9, 128.8, 127.9, 127.1, 127.0, 127.0, 121.5, 119.9, 118.9, 113.9, 110.4, 32.0, 26.2, 22.5, 13.8; HRMS (ESI, *m*/*z*): calcd for C₂₄H₂₄N [M+H]⁺ 326.1903, found 326.1903; IR (film): 1331, 1458, 1488, 1561, 1610, 2927, 2956, 3414 cm⁻¹.

2-Butyl-3-(naphthalen-2-yl)-1H-indole (**3r**).^{5b} Yield: 45 mg (75%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.94-7.84 (m, 5H), 7.70-7.64 (m, 2H), 7.50-7.43 (m, 2H), 7.32 (t, J = 7.9 Hz, 1H), 7.20-7.10 (m, 2H), 2.87 (t, J = 7.8 Hz, 2H), 1.70-1.62 (m, 2H), 1.39-1.30 (m, 2H), 0.87 (t, J = 7.4 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 136.4, 135.2, 133.8, 133.1, 131.9, 128.4, 128.1, 127.9, 127.8, 127.7, 125.9, 125.3, 121.6, 120.0, 118.9, 114.3, 110.4, 32.0, 26.1, 22.5, 13.8; IR (film): 1324, 1458, 1503, 1601, 1629, 2928, 2956, 3415 cm⁻¹.

2-*Cyclopropyl-3-(4-methoxyphenyl)-1H-indole (3s).*^{5b} Yield: 33 mg (62%), pale yellow liquid, $R_f = 0.3$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 7.72 (s, 1H), 7.63 (d, J = 7.8 Hz, 1H), 7.56-7.53 (m, 2H), 7.27 (d, J = 7.7 Hz, 1H), 7.16-7.07 (m, 2H), 7.03-7.00 (m, 2H), 3.86 (s, 3H), 2.22-2.15 (m, 1H), 0.98-0.93 (m, 2H), 0.73-0.69 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 157.7, 135.8, 134.7, 130.6, 128.2, 127.7, 121.5, 119.9, 118.6, 114.7, 113.8, 110.4, 55.3, 8.1, 7.5; IR (film): 1241, 1333, 1461, 1512, 1558, 1612, 1655, 3403 cm⁻¹.

*3-(4-Methoxyphenyl)-2-phenyl-1H-indole (3t).*²¹ Yield: 43 mg (72%), yellow solid, mp: 179-181 °C; $R_f = 0.3 \text{ (PE : EA = 10 : 1); }^{1}\text{H NMR} (400 \text{ MHz, CDCl}_3) \delta 8.19 (s, 1H), 7.64 (d, J = 7.9\text{Hz}, 1H), 7.43-$ 7.40 (m, 3H), 7.37-7.36 (m, 1H), 7.35-7.32 (m, 2H), 7.30-7.28 (m, 1H), 7.25-7.21 (m, 2H), 7.16-7.12 (m, 1H), 6.95-6.91 (m, 2H), 3.84 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.1, 135.8, 133.7, 132.8, 131.2, 129.0, 128.6, 128.0, 127.5, 127.3, 122.6, 120.3, 119.7, 114.7, 114.0, 110.8, 55.2; IR (film): 746, 1036, 1177, 1245, 1284, 1456, 1513, 2969 cm⁻¹.

3-(4-Methoxyphenyl)-2-(p-tolyl)-1H-indole (*3u*).²² Yield: 48 mg (77 %), yellow solid, mp: 109-110 °C; $R_f = 0.3$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.16 (s, 1H), 7.63 (d, J = 7.9Hz, 1H), 7.40-7.30 (m, 5H), 7.21-7.19 (m, 1H), 7.15-7.11 (m, 3H), 6.95-6.91 (m, 2H), 3.84 (s, 3H), 2.34 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.1, 137.4, 135.7, 133.8, 131.1, 129.9, 129.4, 129.0, 127.9, 127.5, 122.4, 120.2, 119.5, 114.2, 114.0, 110.7, 55.2, 21.2; IR (film): 1175, 1244, 1285, 1329, 1455, 1520, 2959, 3408 cm⁻¹.

3-(4-Methoxyphenyl)-2-(4-(trifluoromethyl)phenyl)-1H-indole (3v). Yield: 61 mg (83 %), pale yellow liquid, $R_f = 0.3$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.23 (s, 1H), 7.63 (d, J = 8.0 Hz, 1H), 7.52 (q, J = 8.4 Hz, 4H), 7.41 (d, J = 8.1 Hz, 1H), 7.34-7.30 (m, 2H), 7.26-7.23 (m, 1H), 7.17-7.13 (m, 1H), 6.97-6.93 (m, 2H), 3.85 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.5, 136.3, 136.1, 131.9, 131.2, 129.1 (q, J = 32.5 Hz), 128.9, 128.0, 126.7, 125.6 (q, J = 3.7 Hz), 124.1 (q, J = 272.0 Hz), 123.3, 120.6, 120.0, 116.3, 114.3, 111.0, 55.23; HRMS (ESI, m/z): calcd for C₂₂H₁₇F₃NO [M+H]⁺ 368.1257, found 368.1258; IR (film): 734, 843, 1064, 1125, 1244, 1323, 1523, 1616 cm⁻¹.

2-(4-Fluorophenyl)-3-(4-methoxyphenyl)-1H-indole (**3***w*).²² Yield: 54 mg (84 %), pale yellow liquid, $R_f = 0.2$ (PE : EA = 10 : 1); ¹H NMR (400 MHz, CDCl₃) δ 8.14 (s, 1H), 7.63 (d, J = 7.9 Hz, 1H), 7.37-7.30 (m, 5H), 7.21-7.19 (m, 1H), 7.15-7.11 (m, 1H), 7.01-6.95 (m, 2H), 6.94-6.91 (m, 2H), 3.82 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.2 (d, J = 247.9 Hz), 158.1, 135.8, 132.8, 131.1, 129.8 (d, J = 8.0Hz), 128.9 (d, J = 3.4 Hz), 128.8, 127.1, 122.7, 120.4, 119.6, 115.7 (d, J = 21.7 Hz), 114.6, 114.1, 110.8, 55.2; IR (film): 1035, 1159, 1244, 1454, 1496, 1519, 1598, 2971 cm⁻¹.

The Journal of Organic Chemistry

Experiments for mechanistic investigations (eq 1). To a solution of PPh₃ (52.4 mg, 0.2 mmol) in dried dichloromethane (1 mL) was added the 2-alkynyl azidobenzene **1a** (40 mg, 0.2 mmol) at room temperature and the mixture was stirred for 6 h. 2-Alkynyl azidobenzene was completely disappeared as judged by TLC (thin-layer chromatography). The solvent was then removed under vacuum to afford phosphinimine **4** as the crude product. The structure of phosphinimine **4** was confirmed by comparing the ¹H and ¹³C NMR spectra with standard spectra.²⁴ Then Pd₂dba₃ (4.6 mg, 0.005 mmol, 2.5 mol%), LiO'Bu (24 mg, 0.3 mmol), dry toluene (2 mL), arylbromide **2a** (56 mg, 0.3 mmol) and the phosphinimine **4** were added to reaction tube under N₂ gas and the solution was stirred at 100 °C for 24 h. The product **3a** could be isolated in 56% yield. Similar reaction with dpppe afforded **3a** in 72% yield.

General procedure for the synthesis of *poly*-substituted isoquinolines. Under N₂ gas, Pd₂dba₃ (4.6 mg, 0.005 mmol, 2.5 mol%), PPh₃ (78.6 mg, 0.3 mmol), LiO^tBu (44.8 mg, 0.56 mmol) are successively added to a flame-dried 10 mL Schlenk tube. The reaction tube was degassed three times with N₂ and dry toluene (1 mL) was added by syringe. Then, 2-alkynyl benzylazide (0.2 mmol) and arylbromide (0.6 mmol) were added by syringe, respectively. Note that arylbromide in solid form was added to the reaction tube prior to LiO^tBu. The reaction is heated at 100 °C with stirring for 24 h, then it was cooled to room temperature and filtered through a short plug of silica gel with ethyl acetate as eluents. The solvent was removed in *vacuo* to leave a crude mixture, which was purified to afford the pure product **6a-l**.

3-Butyl-4-(4-methoxyphenyl)isoquinoline (6a). Yield: 45 mg (76%), pale yellow liquid, $R_f = 0.3$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.24 (s, 1H), 7.96-7.94 (m, 1H), 7.55-7.48 (m, 2H), 7.41-7.39 (m, 1H), 7.21 (d, J = 8.6 Hz, 2H), 7.04 (d, J = 8.6 Hz, 2H), 3.90 (s, 3H), 2.74 (t, J = 7.9 Hz, 2H), 1.71-1.63 (m, 2H), 1.32-1.22 (m, 2H), 0.82 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158. 9, 153.4, 151.3, 136.3, 131.2, 130.1, 130.0, 129.5, 127.3, 126.7, 125.9, 125.2, 113.8, 55.2, 35.3, 32.5, 22.7, 13.9; HRMS (ESI, m/z): calcd for C₂₀H₂₂NO [M+H]⁺ 292.1696, found 292.1689; IR (film): 1235, 1286, 1377, 1463, 1514, 1573, 1610, 2929, 2957 cm⁻¹.

3-Butyl-4-phenylisoquinoline (*6b*). Yield: 36 mg (70%), pale yellow liquid, $R_f = 0.5$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.25 (s, 1H), 7.97-7.95 (m, 1H), 7.54-7.43 (m, 5H), 7.37-7.34 (m, 1H), 7.30-7.28 (m, 2H), 2.72 (t, J = 7.9 Hz, 2H), 1.71-1.63 (m, 2H), 1.30-1.21 (m, 2H), 0.81 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.0, 151.5, 137.5, 135.9, 130.4, 130.2, 130.1, 128.4, 127.4, 127.3, 126.6, 126.0, 125.2, 35.3, 32.5, 22.7, 13.9; HRMS (ESI, m/z): calcd for C₁₉H₂₀N [M+H]⁺ 262.1590, found 262.1585; IR (film): 1249, 1376, 1443, 1500, 1573, 1619, 2928, 2957 cm⁻¹.

3-Butyl-4-(4-(trimethylsilyl)phenyl)isoquinoline (6c). Yield: 39 mg (59%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.25 (s, 1H), 7.96-7.94 (m, 1H), 7.65 (d, J = 8.0 Hz, 2H), 7.53-7.48 (m, 2H), 7.39-7.36 (m, 1H), 7.27 (d, J = 8.0 Hz, 2H), 2.72 (t, J = 7.9 Hz, 2H), 1.72-1.64 (m, 2H), 1.31-1.22 (m, 2H), 0.81 (t, J = 7.3Hz, 3H), 0.36 (s, 9H); ¹³C NMR (100 MHz, CDCl₃) δ 153.0, 151.5, 139.5, 137.8, 135.9, 133.3, 130.5, 130.0, 129.5, 127.3, 126.6, 126.0, 125.3, 35.3, 32.5, 22.6, 13.9, -1.0; HRMS (ESI, *m/z*): calcd for C₂₂H₂₈NSi [M+H]⁺ 334.1986, found 334.1988; IR (film): 818, 857, 967, 1105, 1249, 1377, 1572, 2955 cm⁻¹.

3-Butyl-4-(4-fluorophenyl)isoquinoline (6d). Yield: 36 mg (65%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.25 (s, 1H), 7.98-7.96 (m, 1H), 7.57-7.50 (m, 2H), 7.34-7.32 (m, 1H), 7.28-7.19 (m, 4H), 2.70 (t, J = 7.9 Hz, 2H), 1.70-1.62 (m, 2H), 1.31-1.22 (m, 2H), 0.82 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 162.2 (d, J = 246.5 Hz), 153.2, 151.7, 136.0, 133.3 (d, J = 3.6 Hz), 131.8 (d, J = 8.0 Hz), 130.3, 129.4, 127.4, 126.6, 126.1, 124.9, 115.5 (d, J = 21.3 Hz), 35.3, 32.4, 22.7, 13.8; HRMS (ESI, m/z): calcd for C₁₉H₁₉FN [M+H]⁺ 280.1496, found 280.1493; IR (film): 1222, 1378, 1507, 1573, 1604, 1619, 2929, 2958 cm⁻¹.

3-Butyl-4-(4-chlorophenyl)isoquinoline (6e). Yield: 38 mg (65%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.25 (s, 1H), 7.98-7.96 (m, 1H), 7.57-7.52 (m, 2H), 7.49 (d, J = 8.3 Hz, 2H), 7.34-7.31 (m, 1H), 7.25-7.23 (m, 2H), 2.70 (t, J = 7.9 Hz, 2H), 1.70-1.62 (m, 2H), 1.31-1.22 (m, 2H), 0.82 (t, J = 7.4 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.0, 151.9, 135.9, 135.7, 133.6, 131.6, 130.3, 129.2, 128.7, 127.5, 126.6, 126.2, 124.8, 35.3, 32.4, 22.7, 13.9; HRMS (ESI, m/z): calcd ACS Paragon Plus Environment

The Journal of Organic Chemistry

for C₁₉H₁₉ClN [M+H]⁺ 296.1201, found 296.1193; IR (film): 1376, 1455, 1487, 1498, 1572, 1619, 2926, 2956 cm⁻¹.

4-(4-bromophenyl)-3-butylisoquinoline (6f). Yield: 40 mg (59%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.27 (s, 1H), 7.99-7.96 (m, 1H), 7.65 (d, J = 8.3 Hz, 2H), 7.57-7.51 (m, 2H), 7.34-7.31 (m, 1H), 7.18 (d, J = 8.3 Hz, 2H), 2.70 (t, J = 7.9 Hz, 2H), 1.70-1.62 (m, 2H), 1.31-1.22 (m, 2H), 0.83 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 152.9, 151.8, 136.4, 135.7, 131.9, 131.7, 130.4, 129.2, 127.5, 126.6, 126.2, 124.8, 121.7, 35.3, 32.5, 22.7, 13.9; HRMS (ESI, m/z): calcd for C₁₉H₁₉BrN [M+H]⁺ 340.0695, found 340.0689; IR (film): 1377, 1455, 1486, 1497, 1572, 1619, 2927, 2955 cm⁻¹.

3-Butyl-4-(3-chlorophenyl)isoquinoline (6g). Yield: 36 mg (61%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.26 (s, 1H), 7.98-7.96 (m, 1H), 7.58-7.51 (m, 2H), 7.46-7.45 (m, 2H), 7.34-7.31 (m, 2H), 7.21-7.18 (m, 1H), 2.71 (t, J = 7.9 Hz, 2H), 1.71-1.64 (m, 2H), 1.32-1.23 (m, 2H), 0.83 (t, J = 7.4 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.0, 152.0, 139.4, 135.6, 134.4, 130.4, 130.2, 129.7, 129.0, 128.5, 127.8, 127.5, 126.6, 126.12, 124.8, 35.3, 32.4, 22.6, 13.8; HRMS (ESI, m/z): calcd for C₁₉H₁₉ClN [M+H]⁺ 296.1201, found 296.1194; IR (film): 1078, 1374, 1496, 1571, 1594, 1619, 2927, 2957 cm⁻¹.

4-([1,1'-Biphenyl]-4-yl)-3-butylisoquinoline (**6**h). Yield: 40 mg (59%), pale yellow liquid, $R_f = 0.3$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.27 (s, 1H), 7.99-7.96 (m, 1H), 7.76-7.71 (m, 4H), 7.55-7.44 (m, 5H), 7.41-7.36 (m, 3H), 2.77 (t, J = 7.9 Hz, 2H), 1.74-1.67 (m, 2H), 1.33-1.24 (m, 2H), 0.83 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.1, 151.6, 140.6, 140.2, 136.5, 135.9, 130.7, 130.1, 128.8, 127.4, 127.4, 127.1, 126.7, 126.0, 125.2, 35.4, 32.5, 22.7, 13.9; HRMS (ESI, m/z): calcd for C₂₅H₂₄N [M+H]⁺ 338.1903, found 338.1897; IR (film): 733, 765, 1377, 1485, 1573, 1619, 2927, 2956 cm⁻¹. *3-Butyl-4-(naphthalen-2-yl)isoquinoline (6i).* Yield: 29 mg (47%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.30 (s, 1H), 7.99-7.95 (m, 3H), 7.89-7.86 (m, 1H), 7.78 (s, 1H), 7.58-7.53 (m, 2H), 7.52-7.47 (m, 2H), 7.43-7.40 (m, 1H), 7.37-7.35 (m, 1H), 2.78-2.73 (m, 2H), 1.74-1.66 (m, 2H), 1.28-1.19 (m, 2H), 0.78 (t, J = 7.3 Hz, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 153.2, 151.7, 136.0, 135.0, 133.3, 132.6, 130.3, 130.1, 129.1, 128.4, 128.1, 128.0, 127.8, 127.4, 126.7, 126.4, 126.2, 126.0, 125.2, 35.4, 32.5, 22.6, 13.9; HRMS (ESI, m/z): calcd for C₂₃H₂₂N [M+H]⁺ 312.1747, found 312.1749; IR (film): 1242, 1377, 1465, 1494, 1572, 1618, 2927, 2957 cm⁻¹.

4-(4-methoxyphenyl)-3-pentylisoquinoline (6j). Yield: 46 mg (76%), pale yellow liquid, $R_f = 0.4$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.24 (s, 1H), 7.96-7.94 (m, 1H), 7.54-7.47 (m, 2H), 7.41-7.39 (m, 1H), 7.22-7.19 (m, 2H), 7.06-7.03 (m, 2H), 3.90 (s, 3H), 2.75-2.71 (m, 2H), 1.72-1.65 (m, 2H), 1.24-1.21 (m, 4H), 0.84-0.80 (m, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.9, 153.4, 151.4, 136.3, 131.2, 130.1, 130.0, 129.6, 127.3, 126.7, 125.9, 125.2, 113.8, 55.2, 35.6, 31.8, 30.0, 22.4, 13.9; HRMS (ESI, m/z): calcd for C₂₁H₂₄NO [M+H]⁺ 306.1852, found 306.1849; IR (film): 1228, 1286, 1377, 1514, 1573, 1610, 2928, 2956 cm⁻¹.

3-Cyclopropyl-4-(4-methoxyphenyl)isoquinoline (6k). Yield: 47 mg (85%), yellow solid, mp: 135-136 $^{\circ}$ C; $R_f = 0.5$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.12 (s, 1H), 7.91-7.88 (m, 1H), 7.52-7.48 (m, 1H), 7.46-7.42 (m, 2H), 7.33-7.30 (m, 2H), 7.08-7.04 (m, 2H), 3.89 (s, 3H), 1.99-1.92 (m, 1H), 1.18-1.15 (m, 2H), 0.86-0.82 (m, 2H); ¹³C NMR (100 MHz, CDCl₃) δ 158.9, 152.9, 151.4, 135.8, 131.7, 130.0, 129.4, 129.4, 127.3, 126.4, 125.4, 124.7, 113.9, 55.2, 14.4, 9.7; HRMS (ESI, *m/z*): calcd for C₁₉H₁₈NO [M+H]⁺ 276.1383, found 276.1382; IR (film): 1033, 1176, 1228, 1287, 1453, 1513, 1573, 1610 cm⁻¹.

3, 4-Bis(4-methoxyphenyl)isoquinoline (61).²³ Yield: 42 mg (62%), pale yellow solid, mp: 157-159 °C; $R_f = 0.2$ (PE : EA = 5 : 1); ¹H NMR (400 MHz, CDCl₃) δ 9.33 (s, 1H), 8.03-8.01 (m, 1H), 7.70-7.68 (m, 1H), 7.62-7.55 (m, 2H), 7.35-7.32 (m, 2H), 7.18-7.15 (m, 2H), 6.95-6.91 (m, 2H), 6.78-6.75 (m, 2H), 3.86 (s, 3H), 3.78 (s, 3H); ¹³C NMR (100 MHz, CDCl₃) δ 158.7, 158.6, 151.5, 150.4, 136.3, 133.4, ACS Paragon P²us Environment 132.3, 131.5, 130.3, 129.7, 129.6, 127.5, 127.2, 126.5, 125.5, 113.9, 113.1, 55.2, 55.1; IR (film): 829, 1033, 1176, 1244, 1513, 1608, 2834, 2930 cm⁻¹.

ASSOCIATED CONTENT

Supporting Information

Copies of ¹H and ¹³C spectra for all products and X-ray crystallographic data (CIF) for product **3t**. This material is available free of charge *via* the Internet at <u>http://pubs.acs.org</u>.

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Notes

The authors declare no competing financial interest.

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REFERENCES

(1) For selected reviews, see: (a) Ye, T.; McKervey, M. A. Chem. Rev. 1994, 94, 1091-1160. (b) Doyle,

M. P.; McKervey, M. A.; Ye, T. Modern Catalytic Methods for Organic Synthesis with Diazo *Compounds*; Wiley, New York, 1998. (c) Davies, H. M. L.; Beckwith, R. E. J. Chem. Rev. 2003, 103, 2861-2903. (d) Zhang, Z.; Wang, J. Tetrahedron 2008, 64, 6577-6605. (e) Davies, H. M. L.; Denton,

J. R. *Chem. Soc. Rev.* 2009, *38*, 3061-3071. (f) Doyle, M. P.; Duffy, R.; Ratnikov, M.; Zhou, L. *Chem. Rev.* 2010, *110*, 704-724. (g) Zhu, S. F.; Zhou, Q. L. *Acc. Chem. Res.* 2012, *45*, 1365-1377. (h) Ford, A.; Miel, H.; Ring, A.; Slaterry, C. N.; Maguire, A. R.; McKervey, M. A. *Chem. Rev.* 2015, *115*, 9981-10080.

- (2) For a recent comprehensive review on metal carbene from non-diazo sources, see: Jia, M.; Ma, S. *Angew. Chem. Int. Ed.* **2016**, *55*, 9135-9166.
- (3) For reviews, see: (a) Xiao, J.; Li, X. Angew. Chem. Int. Ed. 2011, 50, 7226-7236. (b) Yeom, H.-S.;
 Shin, S. Acc. Chem. Res. 2014, 47, 966-977. (c) Davies, P. W.; Garzón, M. Asian J. Org. Chem. 2015, 4, 694-708.
- (4) Zhang, L. Acc. Chem. Res. 2014, 47, 877-888.
- (5) (a) Gorin, D. J.; Davis, N. R.; Toste, F. D. J. Am. Chem. Soc. 2005, 127, 11260-11261. (b) Lu, B.;
 Luo, Y.; Liu, L.; Ye, L.; Wang, Y.; Zhang, L. Angew. Chem. Int. Ed. 2011, 50, 8358-8362. (c) Yan,
 Z.-Y.; Xiao, Y.; Zhang, L. Angew. Chem. Int. Ed. 2012, 51, 8624-8627. (d) Wetzel, A.; Gagosz, F.
 Angew. Chem. Int. Ed. 2011, 50, 7354-7358. (e) Gronnier, C.; Boissonnat, G.; Gagosz, F. Org. Lett.
 2013, 15, 4234-4237. (f) Hiroya, K.; Matsumoto, S.; Ashikawa, M.; Ogiwara, K.; Sakamoto, T. Org.
 Lett. 2006, 8, 5349-5352. (g) Suzuki, T.; Tokimizu, Y.; Sakano, Y.; Katoono, R.; Fujiwara, K.; Naoe,
 S.; Fujii, N.; Ohno, H. Chem. Lett. 2013, 42, 1001-1003. (h) Suzuki, T.; Sakano, Y.; Tokimizu, Y.;
 Miura, Y.; Katoono, R.; Fujiwara, K.; Yoshioka, N.; Fujii, N.; Ohno, H. Chem. Asian J. 2014, 9, 1841-1846. (i) Tokimizu, Y.; Oishi, S.; Fujii, N.; Ohno, H. Org. Lett. 2014, 16, 3138-3141.
- (6) For reviews, see: (a) Zhang, Y.; Wang, J. Eur. J. Org. Chem. 2011, 1015-1026. (b) Barluenga, J.; Vald és, C. Angew. Chem. Int. Ed. 2011, 50, 7486-7500. (c) Shao, Z.; Zhang, H. Chem. Soc. Rev. 2012, 41, 560-572. (d) Xiao, Q.; Zhang, Y.; Wang, J. Acc. Chem. Res. 2013, 46, 236-247. (e) Liu, Z.; Wang, J. J. Org. Chem. 2013, 78, 10024-10030. (f) Xia, Y.; Zhang, Y.; Wang, J. ACS Catal. 2013, 3, 2586-2598. (g) Hu, F.; Xia, Y.; Ma, C.; Zhang, Y.; Wang, J. Chem. Commun. 2015, 51, 7986-7995.
 (7) For recent examples, see: (a) Chan, W.-W.; Lo, S.-F.; Zhou, Z.; Yu, W.-Y. J. Am. Chem. Soc. 2012, 2012, 2012, 2013, 2013, 2013, 2014, 2014, 2015, 2014, 2014, 2014, 2014, 2015, 2014, 2
 - , 13565-13568. (b) Hyster, T. K.; Ruhl, K. E.; Rovis, T. *J. Am. Chem. Soc.* **2013**, *135*, 5364-5367. ACS Paragon Plus Environment

The Journal of Organic Chemistry

(c) Shi, Z.; Koester, D. C.; Boultadakis-Arapinis, M.; Glorius, F. J. Am. Chem. Soc. 2013, 135, 12204-12207. (d) Hu, M.; He, Z.; Gao, B.; Li, L.; Ni, C.; Hu, J. J. Am. Chem. Soc. 2013, 135, 17302-17305. (e) Hu, F.; Xia, Y.; Ye, F.; Liu, Z.; Ma, C.; Zhang, Y.; Wang, J. Angew. Chem. Int. Ed. 2014, 53, 1364-1367. (f) Xu, S.; Wu, G.; Ye, F.; Wang, X.; Li, H.; Zhao, X.; Zhang, Y.; Wang, J. Angew. Chem. Int. Ed. 2015, 54, 4669-4672. (g) Xia, Y.; Feng, S.; Liu, Z.; Zhang, Y.; Wang, J. Angew. Chem. Int. Ed. 2015, 54, 7891-7894. (h) Luo, H.; Wu, G.; Zhang, Y.; Wang, J. Angew. Chem. Int. Ed. 2015, 54, 7891-7894. (h) Luo, H.; Wu, G.; Zhang, Y.; Wang, J. Angew. Chem. Int. Ed. 2015, 54, 14503-14507. (i) Liu, Z.; Tan, H.; Fu, T.; Xia, Y.; Qiu, D.; Zhang, Y. Wang, J. J. Am. Chem. Soc. 2015, 137, 12800-12803.

- (8) (a) Xia, Y.; Qu, S.; Xiao, Q.; Wang, Z.-X.; Qu, P.; Chen, L.; Liu, Z.; Tian, L.; Huang, Z.; Zhang, Y.; Wang, J. *J. Am. Chem. Soc.* 2013, *135*, 13502-13511. (b) Xia, Y.; Xia, Y.; Ge, R.; Liu, Z.; Xiao, Q.; Zhang, Y.; Wang, J. *Angew. Chem. Int. Ed.* 2014, *53*, 3917-3921. (c) Xia, Y.; Liu, Z.; Ge, R.; Xiao, Q.; Q.; Zhang, Y.; Wang, J. *Chem. Commun.* 2015, *51*, 11233-11235.
- (9) For selected reports on poly-substituted indole synthesis, see: (a) Larock, R. C.; Yum, E. K. J. Am. Chem. Soc. 1991, 113, 6689-6690. (b) Larock, R. C.; Yum, E. K.; Refvik, M. D. J. Org. Chem. 1998, 63, 7652-7662. (c) Battistuzzi, G.; Cacchi, S.; Fabrizi, G. Eur. J. Org. Chem. 2002, 2671-2681. (d) Lu, B. Z.; Zhao, W.; Wei, H.-X.; Dufour, M.; Farina, V.; Senanayake, C. H. Org. Lett. 2006, 8, 3271-3274. e) Würtz, S.; Rakshit, S.; Neumann, J. J.; Dröge, T.; Glorius, F. Angew. Chem. Int. Ed. 2008, 47, 7230-7233. (f) Nallagonda, R.; Rehan, M.; Ghorai, P. Org. Lett. 2014, 16, 4786-4789. (g) Sagadevan, A.; Ragupathi, A.; Hwang, K. C. Angew. Chem. Int. Ed. 2015, 54, 13896-13901 and references cited therein.
- (10) (a) Nakamura, I.; Bajracharya, G. B.; Mizushima, Y.; Yamamoto, Y. Angew. Chem. Int. Ed.
 2002, 41, 4328-4331. (b) Monteiro, N.; Gore, J.; Balme, G. Tetrahedron 1992, 48, 10103. (c) Monteiro, N.; Balme, G. J. Org. Chem. 2000, 65, 3223-3226.

(11) For selected examples on poly-substituted isoquinolines synthesis: (a) Roesch, K. R.; Zhang, H.;
Larock, R. C. J. Org. Chem. 2001, 66, 8042-8051. (b) Dai, G.; Larock, R. C. J. Org. Chem. 2002, 67, 7042-7047. (c) Fischer, D.; Tomeba, H.; Pahadi, N. K.; Patil, N. T.; Yamamoto, Y. Angew. Chem.
ACS Paragon Pfus Environment

Int. Ed. 2007, 46, 4764-4766. (d) Fischer, D.; Tomeba, H.; Pahadi, N. K.; Patil, N. T.; Huo, Z.;
Yamamoto, Y. J. Am. Chem. Soc. 2008, 130, 15720-15725. (e) Han, W.; Zhang, G.; Li, G.; Huang,
H. Org. Lett. 2014, 16, 3534-3535. (f) Zhao, D.; Lied, F.; Glorius, F. Chem. Sci. 2014, 5, 2869-2873.
(g) Jiang, H.; Cheng, Y.; Wang, R.; Zhang, Y.; Yu, S. Chem. Commun. 2014, 50, 6164-6167 and
references cited therein.

(12) For the mechanism for synthesis of isoquinoline., see *Supporting Information*.

(13) CCDC 1484689 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge from The Cambridge Crystallographic Data Centre *via* <u>www.ccdc.cam.ac.uk/data request/cif</u>. Also see *Supporting Information*.

(14) Staudinger, H.; Meyer, J. *Helv. Chim. Acta* **1919**, *2*, 635-646.

(15) For reviews, see: (a) Gololobov, Y. G.; Zhmurova, I. N.; Kasukhin, L. F. *Tetrahedron* 1981, *37*, 437-472. (b) Gololobov, Y. G.; Kasukhin, L. F. *Tetrahedron* 1992, *48*, 1353-1406. (c) Molina, P.; Vilaplana, M. J. *Synthesis* 1994, 1197-1218. (d) Köhn, M.; Breinbauer, R. *Angew. Chem. Int. Ed.* 2004, *43*, 3106-3116. (e) Palacios, F.; Alonso, C.; Aparicio, D.; Rubiales, G.; Santos, J. *Tetrahedron* 2007, *63*, 523-575.

- (16) Shen, Z.; Lu, X. Adv. Synth. Catal. 2009, 351, 3107-3112.
- (17) Wetzel, A.; Gagosz, F. Angew. Chem. Int. Ed. 2011, 50, 7354-7358.
- (18) Thompson, A.; Humphrey, G. R.; DeMarrrco, A. M.; Mathre, D. J.; Grabowski, E. J. J. J. Org.
 Chem. 1993, 58, 5886-5888.
- (19) Arcadi, A.; Cacchi, S.; Carnicelli, V.; Marinelli, F. Tetrahedron 1994, 50, 437-452.
- (20) Ackermann, L.; Sandmann, R.; Villar, A.; T. Kaspar, L. Tetrahedron 2008, 64, 769-777.
- (21) Cacchi, S.; Fabrizi, G.; Goggiamani, A.; Perboni, A.; Sferrazza, A.; Stabile, P. Org. Lett. 2010, 12, 3279-3281.
- (22) Zhang, H.-C.; Ye, H.; White, K. B.; Maryanoff, B. E. *Tetrahedron Lett.* **2001**, *42*, 4751-4754.
- (23) Dai, G.; Larock, R. J. Org. Chem. 2003, 68, 920-928.
- (24) Saito, T.; Nihei, H.; Otani, T.; Suyama, T.; Furukawa, N.; Saito, M. *Chem. Commun.* 2008, 172-ACS Paragon Plus Environment

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