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Design and Application of Indole-Based Allylic Donors for Pd-Catalyzed Decarboxylative Allylation Reactions

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Summary of main observation and conclusion A new class of indole-based allylic donors have been designed and developed for palladium-catalyzed ecarboxylative allylations. In addition, the first application of these indole-based allylic donors in palladium-catalyzed decarboxylative [3+2] cycloaddition and allylic amination has been achieved by reacting with isocyanates and sulfonyl amines, respectively. This approach represents the first resign of indole-based allylic donors, which is helpful for settling the challenge of designing and developing new class of heterocycle-based allylic donors or Pd-catalyzed decarboxylative allylation reactions. Moreover, the application of this new class of allylic donors in cycloadditions and substitutions will add new contents to the research field of decarboxylative allylation.

Background and Originality Content

Transition-metal-catalyzed allylation reactions belong to a class f important transformations for forming new covalent bonds.^[1-2] Among them, palladium-catalyzed decarboxylative allylation (DcA) reaction has developed rapidly due to its unique characteristics such as generating organometallic intermediates via decarboxylation of allylic donors under mild conditions, which are easily intercepted by another reaction partner to construct useful wameworks.^[3]

The rapid development of this class of reactions relies on the design and synthesis of decarboxylative allylic donors with high eactivity. As illustrated in Scheme 1a, commonly used allylic donors for Pd-catalyzed DcA reactions mainly include vinyl xazinanones,^[4-6] vinyl oxazolidinones,^[7-8] vinylethylene carbonates,^[9-10] γ -methylidene- δ -valerolactones,^[11-12] 2-alkylidenetrimethylene carbonates,^[13] ξ -methylene-1,3-dioxolan-2-ones^[14] and so on.^[3] Up to date,

talyzed DcA reactions of these allylic donors have been well-established. The general reaction mode of Pd-catalyzed DcA reactions is summarized in Scheme 1b. Namely, in the presence of d (0) catalyst, allylic donors underwent decarboxylation to generate zwitterionic π -allylpalladium intermediate, which could urther undergo cycloadditions or substitutions with another eaction partner. $^{[3]}$

However, in spite of these progresses, there are still some challenges required to be solved in this research field. For example, the type of allylic donors for Pd-catalyzed DcA reactions very limited, and these commonly used allylic donors were designed and first synthesized almost ten years ago. So, *it has* become an urgent task to design and develop new class of allylic conors for Pd-catalyzed DcA reactions. Furthermore, heterocycle-based allylic donors have scarcely been designed and synthesized for Pd-catalyzed DcA reactions, which nevertheless will provide an easy access to biologically relevant heterocyclic frameworks. Therefore, *it is highly desired to design and develop new class of heterocycle-based allylic donors for Pd-catalyzed DcA reactions*.

Scheme 1 Profile of Pd-catalyzed decarboxylative allylation (DcA) reactions a) Commonly used allylic donors for Pd-catalyzed DcA reactions





Challenges:

Design and develop new class of allylic donors

Design and develop heterocycle-based allylic donors

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To fufill this task, we decided to incorporate the indole core to the structure of allylic donors based on the consideration that indole represents one of the most important heterocycles^[15] and the efficient synthesis of indole derivatives has become a lasting goal in the synthetic community.^[16] So, on the basis of our understanding of indole-based platform molecules,^[16f,17] we designed this new class of indole-based allylic donors for DcA reactions (Scheme 2). In the presence of Pd (0) catalyst, this class of indole-based allylic donors would transform into π -allyl-Pd zwitterion intermediate, which could principally undergo c cloaddition and substitution reactions with different reaction partners, thus providing an easy access to indole derivatives. Enlightened by this design, we synthesized this new class of mole-based allylic donors and realized two different types of DcA reactions. Herein, we reported the details of our investigation.

heme 2 Design of new class of indole-based allylic donors for DcA reactions



Sesults and Discussion

Based on our design of this new class of indole-based allylic donors, we developed a synthetic route for vinyl oxazoloindol-3-ones **1** (Scheme 3). Indole-2-carbaldehydes **S1** ere utilized as starting materials, which underwent an acylation reaction with benzoyl chloride to give *N*-Cbz-protected Ir dole-2-carbaldehydes **S2** in moderate to good yields. Then, the addition of vinylmagnesium bromide to the aldehyde group of **S2** resulted in allylic alcohols **S3** in moderate yields. Finally, an in tramolecular ester-exchange led to the cyclization and the ation of vinyl oxazoloindol-3-ones **1**.

S heme 3 Synthetic route for vinyl oxazoloindol-3-ones **1**



At first, the application of this new class of indole-based allylic donors was commenced with Pd-catalyzed decarboxylative [3+2] cycloaddition of vinyl oxazoloindol-3-one 1a with sulfonyl isocyanate 2a to testify our hypothesis (Table 1). Gratifyingly, in the presence of Pd(PPh₃)₄, this reaction in toluene at 25 °C smoothly occurred to give decarboxylative [3+2] cycloaddition product 3aa albeit with a low yield (entry 1). Nevertheless, this preliminary result demonstrated the feasibility of our design on this new class of indole-based allylic donors for DcA reactions. To improve the yield, a series of representative solvents were evaluated for this reaction (entries 1-6), and other solvents did not display better capability than toluene in promoting the reaction (entries 2-6 vs entry 1). Then, the reaction temperature was altered (entries 7-11). It was found that lowering the temperature would decrease the yield (entries 7-8 vs entry 1), while elevating the temperature could remarkably enhance the yield (entries 9-11 vs entry 1). Among different temperatures, 70 °C was revealed to be the most suitable reaction temperature with regard to the yield (entry 10). To further enhance the yield, the reagents ratio was modulated (entries 12-15), and it was discovered that suitably increasing the amount of sulfonyl isocyanate 2a could increase the yield to a high level of 80% (entry 13). So, these conditions as listed in entry 13 were selected as the optimal conditions for this Pd-catalyzed decarboxylative cycloaddition.

 Table 1 Conditions optimization for Pd-catalyzed decarboxylative [3+2]

 cycloaddition^a

L N 1a	$ + \begin{matrix} N \\ C \\$	5 mol% Pd(solvent, ⁻	T°C	Jaa
entry	solvent	T (°C)	1a:2a	yield (%) ^b
1	toluene	25	1:1.2	28
2	DCE	25	1:1.2	23
3	EtOAc	25	1:1.2	23
4	acetone	25	1:1.2	24
5	CH₃CN	25	1:1.2	trace

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6 1,4-dioxane 25 1:1.2 trace 7 toluene 0 1:1.2 22 8 toluene -10 1:1.2 11 9 toluene 50 1:1.2 49 10 toluene 70 1:1.2 70 11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51	_					
7 toluene 0 1:1.2 22 8 toluene -10 1:1.2 11 9 toluene 50 1:1.2 49 10 toluene 70 1:1.2 65 11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		6	1,4-dioxane	25	1:1.2	trace
8 toluene -10 1:1.2 11 9 toluene 50 1:1.2 49 10 toluene 70 1:1.2 70 11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		7	toluene	0	1:1.2	22
9 toluene 50 1:1.2 49 10 toluene 70 1:1.2 70 11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		8	toluene	-10	1:1.2	11
10 toluene 70 1:1.2 70 11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		9	toluene	50	1:1.2	49
11 toluene 90 1:1.2 65 12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		10	toluene	70	1:1.2	70
12 toluene 70 1:2 75 13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		11	toluene	90	1:1.2	65
13 toluene 70 1:3 80 14 toluene 70 1:4 63 15 toluene 70 2:1 51		12	toluene	70	1:2	75
14 toluene 70 1:4 63 15 toluene 70 2:1 51		13	toluene	70	1:3	80
15 toluene 70 2:1 51		14	toluene	70	1:4	63
		15	toluene	70	2:1	51

Jnless indicated otherwise, the reaction was carried out at 0.1 mmol scale in a solvent (0.5 mL) for 12 h. ^bIsolated yield. DCE = ,2-dichloroethane.

Then, we performed the application of indole-based allylic donors 1 in Pd-catalyzed decarboxylative [3+2] cycloaddition with a wide range of isocyanates 2 under the optimal reaction conditions (Table 2). Apart from sulfonyl isocyanate 2a (entry 1), a variety of phenyl isocyanates **2b-2l** could be utilized as suitable reaction partners to undergo decarboxylative [3+2] cycloadditions with vinyl oxazoloindol-3-one **1a** in generally high yields of 59% to 98% (entries 2-12). In detail, either para-substituted (entries 2-5), meta-substituted (entries 6-8) or ortho-substituted (entries 9-12) phenyl groups could serve as competent R¹ groups for isocyanates 2. Among them, para-chlorophenyl isocyanate 2e reacted with vinyl oxazoloindol-3-one 1a in the highest yield of 98% (entry 5). in addition, isocyanate 2m bearing an aliphatic substituent could moothly participate in the decarboxylative [3+2] cycloaddition with **1a** in a good yield of 70%. Moreover, cyclopentyl isocyanate 2n could undergo the desired [3+2] cycloaddition with 1a Ithough the yield was unsatisfying (entry 14). In addition to substrate 1a, methyl-substituted vinyl oxazoloindol-3-one 1b ould also serve as a suitable indole-based allylic donor, which successfully underwent decarboxylative [3+2] cycloadditions with a series of phenyl isocyanates 2 in overall high yields of 72%-92% (entries 15-19).

 Table 2 Application of indole-based allylic donors 1 in Pd-catalyzed

 cecarboxylative [3+2] cycloaddition^a

)	R J	N - +	N 1 5 mol% Pd(PPh ₃ 5 mol% Pd(PPh ₃ 1 1 1 1 1 1 1 1)₄ R	J-N N-R ¹ 3
	entry	R (1)	R ¹ (2)	3	yield (%) ^b
	1	H (1a)	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂ (2a)	3aa	80
	2	H (1a)	<i>p</i> -CH ₃ C ₆ H ₄ (2b)	3ab	90
	3	H (1a)	<i>p</i> -OMeC ₆ H ₄ (2c)	3ac	66
\neg	4	H (1a)	<i>p</i> -FC ₆ H ₄ (2d)	3ad	66

5	H (1 a)	<i>p</i> -ClC ₆ H ₄ (2e)	3ae	98
6	H (1a)	<i>m</i> -CH ₃ C ₆ H ₄ (2f)	3af	63
7	H (1a)	<i>m</i> -OCH ₃ C ₆ H ₄ (2g)	3ag	67
8	H (1a)	<i>m</i> -FC ₆ H ₄ (2h)	3ah	70
9	H (1a)	<i>o</i> -CH ₃ C ₆ H ₄ (2i)	3ai	79
10	H (1a)	<i>o</i> -OCH ₃ C ₆ H ₄ (2j)	3aj	80
11	H (1a)	<i>o</i> -FC ₆ H ₄ (2k)	3ak	59
12	H (1a)	o-CIC ₆ H ₄ (2I)	3al	81
13	H (1a)	$CICH_2CH_2CH_2$ (2m)	3am	70
14	H (1a)	cyclopentyl (2n)	3an	39
15	Me (1b)	<i>p</i> -CH ₃ C ₆ H ₄ SO ₂ (2a)	3ba	92
16	Me (1b)	<i>p</i> -CH ₃ C ₆ H ₄ (2b)	3bb	72
17	Me (1b)	<i>p</i> -ClC ₆ H ₄ (2e)	3be	80
18	Me (1b)	<i>m</i> -FC ₆ H ₄ (2h)	3bh	90
19	Me (1b)	<i>o</i> -OCH₃C ₆ H₄ (2j)	3bj	90

^{*a*}Unless indicated otherwise, the reaction was carried out at 0.1 mmol scale in toluene (0.5 mL) at 70 °C for 12 h, and the molar ratio of **1:2** was 1:3. ^{*b*}Isolated yield.

After establishing the decarboxylative [3+2] cycloaddition, we then performed the application of this new class of indole-based allylic donors in Pd-catalyzed decarboxylative substitution. As shown in Table 3, the decarboxylative allylic amination reaction of vinyl oxazoloindol-3-one 1a with sulfonyl amine 4a was employed as a model reaction to examine the possibility of our design. Indeed, in the presence of Pd(PPh₃)₄, the reaction in toluene at room temperature occurred to generate allylic amination product 5aa with a linear selectivity in a moderate yield of 56% (entry 1), which demonstrated the feasibility of our design on this new class of indole-based allylic donors in Pd-catalyzed decarboxylative substitution. Then, the evaluation of representative solvents revealed that toluene is much better than other solvents in terms of the yield (entry 1 vs entries 2-6). To further enhance the yield, the reaction temperature was changed by either lowering or elevating the reaction temperature (entries 7-10). It was found that the yield could be greatly enhanced to 90% when the reaction temperature was lowered to 0 °C (entry 7 vs entry 1). Finally, the modulation of reagents ratio (entries 11-12) discovered that properly increasing the amount of sulfonyl amine 4a could further improve the yield to 95% (entry 11). It should be noted that only (E)-isomer of 5aa was observed in all cases. So, these conditions as listed in entry 11 were chosen as the optimal reaction conditions for this allylic amination reaction.

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(J-N-	+ HN–Ts Boc	10 mol% Pd(PPr	¹³⁾⁴ ►	
	0 1a	4a	,		H Boc 5aa
	entry	solvent	T (°C)	1a:4a	yield (%) ^b
	1	toluene	25	1:1	56
	2	DCE	25	1:1	30
_	3	EtOAc	25	1:1	40
	4	acetone	25	1:1	30
	5	CH ₃ CN	25	1:1	40
-	6	1,4-dioxane	25	1:1	46
	7	toluene	0	1:1	90
	8	toluene	-10	1:1	54
	9	toluene	-30	1:1	45
	10	toluene	50	1:1	26
	11	toluene	0	1:2	95

 Table 3 Conditions optimization for Pd-catalyzed decarboxylative substitution^a

onless indicated otherwise, the reaction was carried out at 0.1 mmol scale in a solvent (0.5 mL) for 12 h. bIsolated yield and only (*E*)-isomer of **5**aa was observed in all cases.

With the optimal reaction conditions in hand, we carried out the investigation on the application of indole-based allylic donors in Pd-catalyzed decarboxylative substitution (Table 4). As shown entries 1-10, vinyl oxazoloindol-3-one 1a could undergo decarboxylative allylic amination with sulfonyl amines 4a-4j bearing various *para-*, *meta-*, and *ortho*-substituted phenyl groups, and these reactions successfully generated allylic amination products **5aa-5aj** with uniformly linear selectivity in overall high elds (61%-98%). In addition, methyl-substituted vinyl oxazoloindol-3-one 1b could also be utilized as an allylic donor to ndergo decarboxylative allylic amination with different sulfonyl arnines 4 with exclusively linear selectivity in good yields 90%, entries 11-16). Notably, this reaction has an excellent (*E/Z*)-selectivity, and only (*E*)-isomers of **5** were observed in all c ses.

 Table 4 Application of indole-based allylic donors 1 in Pd-catalyzed

 d carboxylative substitution^a

R	-N 0 +	H Boc SO ₂ R ¹ 10 mol% r toluene,	Pd(PPh ₃) ₄	5	N-SO ₂ R ¹ Boc ¹
entry	R (1)	R ¹ (4)	5	E/Z ^b	yield (%) ^c
1	H (1a)	<i>p</i> -CH ₃ C ₆ H ₄ (4a)	5aa	>95:5	95
2	H (1a)	<i>p</i> -FC ₆ H ₄ (4b)	5ab	>95:5	75

3	H (1a)	<i>p</i> -BrC ₆ H ₄ (4c)	5ac	>95:5	70
4	H (1a)	<i>p</i> -CF ₃ C ₆ H ₄ (4d)	5ad	>95:5	61
5	H (1a)	<i>m</i> -FC ₆ H ₄ (4e)	5ae	>95:5	85
6	H (1a)	<i>m</i> -BrC ₆ H ₄ (4f)	5af	>95:5	88
7	H (1a)	<i>o</i> -CH₃C ₆ H₄ (4g)	5ag	>95:5	95
8	H (1a)	<i>o</i> -FC ₆ H ₄ (4h)	5ah	>95:5	98
9	H (1a)	<i>o</i> -ClC ₆ H ₄ (4i)	5ai	>95:5	91
10	H (1a)	Ph (4j)	5aj	>95:5	76
11	Me (1b)	<i>p</i> -CH ₃ C ₆ H ₄ (4a)	5ba	>95:5	67
12	Me (1b)	<i>p</i> -FC ₆ H ₄ (4b)	5bb	>95:5	87
13	Me (1b)	<i>m</i> -BrC ₆ H ₄ (4f)	5bf	>95:5	84
14	Me (1b)	<i>o</i> -CH ₃ C ₆ H ₄ (4g)	5bg	>95:5	89
15	Me (1b)	<i>o</i> -FC ₆ H ₄ (4h)	5bh	>95:5	64
16	Me (1b)	o-CIC ₆ H ₄ (4i)	5bi	>95:5	90

^oUnless indicated otherwise, the reaction was carried out at 0.1 mmol scale in toluene (0.5 mL) at 0 $^{\circ}$ C for 12 h, and the molar ratio of **1:4** was 1:2. ^bDetermined by ¹H NMR. ^cIsolated yield.

The structures of products **3** and **5** were unambiguously determined by NMR, IR and HR MS. In addition, the structures of products **3aa** and **5bi** were confirmed by their single-crystal X-ray diffraction analysis (Figure 1).^[18]



Figure 1 Single-crystal structures of products 3aa and 5bi

In addition, to examine the practicability of the Pd-catalyzed decarboxylative [3+2] cycloaddition and allylic amination, we performed two one-mmol-scale reactions (Scheme 4). Compared with small-scale reactions (Table 2, entry1 and Table 4, entry 1), these one-mmol-scale reactions smoothly occurred to give products **3aa** and **5aa** in nearly retained high yields, which demonstrated that these reactions could be scaled up.

Scheme 4 One-mmol-scale reactions

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Based on the experimental results, we suggested possible eaction pathways to explain the chemistry and regioselectivity of the two decarboxylative allylation reactions (Scheme 5). As xemplified by the formation of products 3aa and 5aa, the two eactions commenced with the generation of zwitterionic π -allylpalladium intermediate **A** via decarboxylation in the resence of Pd (0). In the [3+2] cycloaddition (Scheme 5a), this intermediate A first attacked sulfonyl isocyanate 2a to undergo an za-Mannich reaction, producing another intermediate B. Then, an intramolecular allylic amination of intermediate B occurred with a branched selectivity to form a five-membered ring. If the tramolecular allylic amination occurred with a linear selectivity, a seven-membered ring would be constructed. Perhaps because ve-membered ring is easier to form than seven-membered ring, the second step of intramolecular allylic amination occurred with a branched selectivity, thus accomplishing the observed [3+2] vcloaddition. Instead, the linear selectivity of intermolecular allylic amination of 1a with 4a could be explained by the absence of hydrogen-bond direction (Scheme 5b), which was reported to e a crucial factor to control the regioselectivity in Pd-catalyzed allylic reactions.^[3b] Namely, π -allylpalladium intermediate **A** could cransform into intermediate C via deprotonation of sulfonyl amine 4a. There was no hydrogen-bond direction between the NH group f intermediate C and the deprotonated amine anion, thus leading to the linear selectivity of the allylic amination.

Notably, the allylic amination reaction (Scheme 5b) proceeded well at 0 °C, while the [3+2] cycloaddition reaction (Scheme 5a) occurred at a higher temperature of 70 °C. The rate difference between the two reactions might stem from the first step after the formation of π -allylpalladium intermediate **A**. Specially, in me [3+2] cycloaddition reaction (Scheme 5a), the first step was the nucleophilic addition of intermediate A to the C=N bond of sulfonyl isocyanate 2a, which displayed the nucleophilicity of intermediate A. While in the allylic amination reaction (Scheme 5b), the first step was the deprotonation of sulfonyl amine 4a by intermediate A, which exhibited the basicity of intermediate A. Because the basicity of intermediate A (nitrogen anion) was very strong, the deprotonation of sulfonyl amine 4a by intermediate A could occur easily. However, the nucleophilic addition of intermediate A to the C=N bond was relatively difficult compared with the deprotonation of sulfonyl amine 4a by intermediate A. These reasons might result in the observed rate difference between the two reactions.

Scheme 5 Suggested reaction pathways



Furthermore, to examine the utility of vinyl oxazoloindol-3-ones in [3+2] cycloadditions, we utilized benzylidene malononitrile **6**, an electron-deficient alkene, as a dipolarophile^[4b,5b,19] to react with oxazoloindol-3-one **1a** under the catalysis of palladium (Scheme 6). This [3+2] cycloaddition smoothly occurred to generate product **7** in a high yield of 94% albeit with a low diastereoselectivity of 56:44 dr. Nevertheless, this result indicated that oxazoloindol-3-ones could be utilized in palladium-catalyzed 1,3-dipolar [3+2] cycloadditions.

Scheme 6 Using electron-deficient alkene 6 as a dipolarophile



Finally, we performed an investigation on the catalytic asymmetric version of the [3+2] cycloaddition between 1a and 2a (Table 5). As listed in entries 1-13, a variety of BINOL, H₈-BINOL, and SPINOL-derived chiral phosphoramidites L1-L13 were utilized as chiral ligands for this palladium-catalyzed allylation reaction. However, in most cases, very weak enantio-control of product 3aa was observed. Among these ligands, L5, L10 and L12 could deliver the [3+2] cycloaddition with some extent of enantioselectivity (around 20% ee). In addition, several chiral bidentate phosphine ligands L14-L18 were tested for this reaction (entries 14-18). Nevertheless, these chiral ligands could hardly control the enantioselectivity of the [3+2] cycloaddition (< 8% ee). So, it seemed that L10 was a promising chiral ligand for this reaction considering the relatively higher yield and better

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enantioselectivity of the [3+2] cycloaddition in the presence of **L10** (entry 10). However, it is still a challenging task to realize a highly enantioselective [3+2] cycloaddition between **1a** and **2a**, which might require exquisite elaboration of reaction conditions and structural modifications of chiral ligands.





ol% Pd2dba3 CHCl3

10 mol% L*

	1a d	8 2a	louene, n.	Ö 3aa	
	entry	L*	yield (%) ^b	ee (%) ^c	
	1	L1	15	<5	
-	2	L2	20	<5	
	3	L3	15	8	
2	4	L4	11	<5	
	5	L5	11	20	
	6	L6	20	<5	
_	7	L7	16	11	
	8	L8	19	<5	
	9	L9	18	-8	
	10	L10	66	-20	
	11	L11	23	<5	
	12	L12	12	22	
	13	L13	40	6	
	14	L14	18	<5	

15	L15	23	<5
16	L16	33	<5
17	L17	33	<5
18	L18	12	8

^{*a*}Unless indicated otherwise, the reaction was carried out at 0.05 mmol scale in toluene (1 mL) for 12 h, and the molar ratio of **1a:2a** was 1:1.2. ^{*b*}Isolated yield. ^{*c*}The ee value was determined by HPLC.

Conclusions

In summary, we have designed and developed a new class of donors palladium-catalyzed indole-based allylic for decarboxylative allylations. In addition, we have realized their first application in palladium-catalyzed decarboxylative [3+2] cycloaddition and allylic amination, respectively, by reacting with isocyanates and sulfonyl amines. This approach represents the first design of indole-based allylic donors, which is helpful for settling the challenge of designing and developing new class of heterocycle-based allylic donors for Pd-catalyzed decarboxylative allylation reactions. Moreover, the application of this new class of allylic donors in cycloadditions and substitutions will add new contents to the research field of decarboxylative allylation.

Experimental

General Procedure for the synthesis of products 3:

A flame-dried 10 mL flask tube was charged with indole-based allylic donors 1 (0.1 mmol) and Pd(PPh₃)₄ (0.005 mmol), sealed with a septum, and evacuated and backfilled with argon for three times. Then, isocyanates 2 (0.3 mmol) and toluene (0.5 mL) was added via syringe. The reaction mixture was stirred at 70 °C for 12 h. After the completion of the reaction which was indicated by TLC, the reaction mixture was directly purified through preparative thin layer chromatography on silica gel to afford pure products **3**.

General Procedure for the synthesis of products 5:

A flame-dried 10 mL flask tube was charged with indole-based allylic donors 1 (0.1 mmol), sulfonyl amines 4 (0.2 mmol) and Pd(PPh₃)₄ (0.01 mmol), sealed with a septum, and evacuated and backfilled with argon for three times. Then, toluene (0.5 mL) was added via syringe. The reaction mixture was stirred at 0 °C for 12 h. After the completion of the reaction which was indicated by TLC, the reaction mixture was directly purified through preparative thin layer chromatography on silica gel to afford pure products **5**.

Supporting Information

The supporting information for this article is available on the WWW under https://doi.org/10.1002/cjoc.2018xxxxx.

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Entry for the Table of Contents

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Design and Application of Indole-Based Allylic Donors for Pd-Catalyzed Decarboxylative Allylation Reactions



ng-Qing Hang, Si-Jia Liu, Lei Yu, Ting-Ting Sun,
 Yu-Chen Zhang,* Guang-Jian Mei* and Feng Shi*

A new class of indole-based allylic donors have been designed and applied in palladium-catalyzed decarboxylative [3+2] cycloaddition and allylic amination.