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## Low-valent Titanium Induced One Pot Syntheses of Imidazolidines

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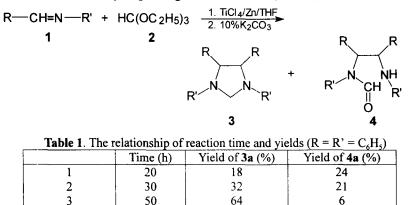
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**Abstract:** Under the action of a low-valent titanium reagent, imidazolidine derivatives were synthesized from imines and triethyl orthoformate in moderate yields. NMR spectroscopy was used to assign the configuration of the products. © 1999 Published by Elsevier Science Ltd. All rights reserved.

Low-valent titanium reagents are versatile reagents in organic synthesis, with high ability for reductive coupling of many functional groups.<sup>1-3</sup> Their application has allowed the synthesis of heterocycles from simple molecules in one step, for example, the reductive cyclization of nitriles to symmetrically substituted tetraalkylpyrazines,<sup>2</sup> and the three molecule reductive cyclization of isothiocyanates to substituted indole-2-carbothioamides.<sup>3</sup> In connection with our interest in investigating new reductive coupling reactions induced by low-valent titanium, we report herein a cross-coupling reaction of imines and triethyl orthoformate, which leads to the formation of imidazolidines.

When a mixture of imine 1 and triethyl orthoformate 2 was treated with low-valent titanium in THF for 50 hours, the cross-coupling product imidazolidine 3 was formed in moderate yield. The reaction intermediate, compound 4, was also isolated.<sup>4</sup> It was found that the amount of the intermediate, which was isolated, could be reduced by lengthening the reaction time. (Table 1)



The structures of the products 3 were determined from their IR, <sup>1</sup>H-NMR and MS spectra and elemental analysis.<sup>4</sup> Different imines led to different isomer distributions of the products in Table 2.

When one of the isomers is predominant, the pure isomer (3-meso or 3-dl) can be obtained by recrystallization. We recorded the <sup>1</sup>H-NMR spectra of the unrecrystallized products, which were obtained directly by chromatography, to measure the ratio of the two isomers (Table 2).

	R	R'	Yields(%)	meso:dl
a	C <sub>6</sub> H,	C <sub>6</sub> H,	64	20:80
b	C <sub>6</sub> H <sub>5</sub>	p-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	53	85:15
с	C <sub>6</sub> H <sub>5</sub>	p-ClC <sub>6</sub> H <sub>4</sub>	68	10:90
d	C <sub>6</sub> H <sub>5</sub>	m-Cl C <sub>6</sub> H <sub>4</sub>	47	5:95
e	C <sub>6</sub> H <sub>5</sub>	o-Cl C <sub>6</sub> H₄	55	85:15
f	p-CH <sub>3</sub> C <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	51	95:5
g	p-CH <sub>3</sub> OC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H,	48	85:15
h	<i>p</i> -ClC <sub>6</sub> H <sub>4</sub>	C <sub>6</sub> H <sub>5</sub>	45	95:5

Table 2. The yields of product 3 and the isomer distributions

In conclusion, we have described a new and efficient one-pot coupling and cyclization reaction that furnished imidazolidine derivatives from simple starting materials. Results related to the stereochemistry and complete data will be reported as a full paper.

## **REFERENCES AND NOTES**

- 1. a) McMurry, J. E. Chem. Rev., 1989, 89, 1513;
  - b) Furstner, A., Bogdanovic, B. Angew. Chem. Int. Ed. Engl., 1996, 35, 2442.
- 2. Chen W.-H., Zhang J.-H., Hu M.-Y. and Wang X.-C., Synthesis, 1990, 701.
- 3. Li J., Shi D.-Q., Chen W.-X., Heterocycles 1997, 45, 2381.
- 4. A general procedure is as follows: A dry 100mL flask was charged with zinc dust (2.60g, 40mmol), TiCl<sub>4</sub> (2.20mL, 20mmol) and THF (35mL). The mixture was refluxed for 2 h under argon, then cooled to r. t.. During that time black slurry was formed. A mixed solution of triethyl orthoformate (~3.0g, 20mmol) and imine (1, 10mmol) in THF (5mL) was added to the reaction mixture using a syringe, stirred for another 2 h at r.t., then refluxed for 50 h. After removing the THF, the mixture was quenched with 10% K<sub>2</sub>CO<sub>3</sub> and extracted with CHCl<sub>3</sub> (6×50mL). The organic layer was dried (Na<sub>2</sub>SO<sub>4</sub>) and evaporated. The crude product was purified by flash chromatography on silica gel (petroleum ether (60-90°C)) to give imidazolidines.

Imidazolidine **3b** (*meso*): Mp 141-143°C;  $v_{max}$  (KBr) cm<sup>-1</sup>: 1610, 1510, 1385, 1270, 810, 790, 690; <sup>1</sup>H NMR (500MHz, CDCl<sub>3</sub>)  $\delta$ : 6.98-6.53 (m, 18H), 5.62 (d, J=3.0Hz, 1H), 5.10(s, 2H), 4.75 (d, J=3.0Hz, 1H), 2.22(s, 6H); <sup>13</sup>C NMR (500MHz, CDCl<sub>3</sub>)  $\delta$ : 143.4, 138.3, 129.5, 127.6, 127.4, 126.7, 114.5, 96.1, 70.9, 67.9, 20.4; EI-MS, m/z (%): 404.2 (7.6%), 208.1 (100%), 209.1 (93.4%).

Imidazolidine **3c** (*dl*): Mp 181-183°C;  $v_{max}$  (KBr) cm<sup>-1</sup> 1600, 1490, 1450, 1390, 1338, 805, 760, 700; <sup>1</sup>H-NMR (60MHz, CDCl<sub>3</sub>)  $\delta$ : 7.30-6.45 (m, 18H), 5.34 (s, 2H), 4.82 (s, 2H); EI-MS, m/z(%): 446.2 (9.4%), 444.2 (14.1%), 229.1 (100%), 228.0 (99.0%).

The intermediate **4a**: Mp 177-178°C;  $v_{max}$  (KBr) cm<sup>-1</sup>: 3300, 3280, 1660, 1595, 1490, 1280, 745, 700; <sup>1</sup>H-NMR (500MHz, CDCl<sub>3</sub>)  $\delta$ : 8.40 (s, 1H), 7.45-6.55 (m, 20H), 6.20 (s, 1H), 6.05 (s, 1H), 5.20 (br, 1H, disappeared when D<sub>2</sub>O was added); EI-MS, m/z (%): 392.3 (0.8%), 182.1 (100%).