[2,3]Wittig Rearrangement of Enantiomerically-Defined α-(Allyloxy)stannanes: Solid Evidence for Inversion of Configuration at the Lithium-Bearing Migrating Terminus

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Abstract: The [2,3]Wittig rearrangement of enantiomerically-defined α -(allyloxy)stannanes, prepared from (S)-1-tributylstannyl-1-propanol, with butyllithium is shown to proceed with essentially complete inversion of configuration at the lithium-bearing migrating terminus. This inversion stereochemistry is further confirmed by the rearrangement of diastereomerically-defined α -(allyloxy)stannanes.

The [2,3]Wittig rearrangement has currently enjoyed widespread use in stereocontrolled synthesis.¹ However, the most fundamental question of [2,3]Wittig stereochemistry is the subject of much controversy that concerns the stereochemical course (inversion vs. retention) at the lithium-bearing terminus (Scheme 1). Conflicting conclusions might be drawn from the recent theoretical calculations on the transition state structures for the rearrangement of allyl lithiomethyl ether.², ³



More recently Verner and Cohen have reported the first experimental evidence for the inversion course in the reductive lithiation-based [2,3]Wittig process of a rather complicated, *diastereomerically-defined* tetrahydropyran system which considerably competes with the [1,2]Wittig process.⁴ We now disclose more

clear-cut evidence for complete inversion of configuration in the transmetalative [2,3]Wittig process using the *enantiomerically-defined* α -(allyloxy)stannanes 2 as the substrates (Scheme 2).⁵



(S)-1

Reagents: (a) MsCl, Et₃N, CH₂Cl₂ -23 °C; (b) RCH=CHCH₂OK, Et₂O, 20 °C; (c) *n*-BuLi (1.5 equiv), THF, -78 °C; (d) H₂, Pd-C,MeOH, 20 °C.

ŌН

(3S.4R)-(+)-4h

The optically active substrates (R)-2a and (R)-2b required for this study were successfully prepared in ca. 65% overall yield from (S)- α -stannyl alcohol 1⁶ via mesylation followed by S_N2 reaction with the potassium allylic alcoholate which proceeds with complete inversion of configuration.⁷, ⁸ Treatment of (R)-2a (62% ee)⁹ with *n*-BuLi at -78 °C was found to afford the rearrangement product (R)-3a in 62% ee (90% yield).¹⁰ A similar rearrangement of (R,E)-2b (88% ee, 100% E)⁹ gave rise to the (3S,4R)-anti isomer 3b in 87% ee and >99% de (95% yield).¹¹ Hydrogenation of 3b furnished an ant pheromone¹² (3S,4R)-4b: $[\alpha]_D^{22}$ +2.2° (CHCl₃); lit.¹² $[\alpha]_D^{22}$ +2.5° (CHCl₃). Since complete retention of configuration has been well established for the Sn/Li exchange process involved,¹³ these findings indicate clearly that the [2,3]Wittig rearrangement proceeds with essentially complete inversion of configuration at the Li-bearing terminus. This means that the lithium cation locates outside the pericyclic array in the transition state as predicted theoretically by Houk *et al.*² It should be noted that the extremely high $E \rightarrow anti$ diastereoselection observed here is also consistent with Houk's calculations.², ¹⁴

This inversion stereochemistry was further confirmed by experiments using the diastereomerically-defined stannane 5.¹⁵ The rearrangement of the *anti*-rich substrates, prepared from the corresponding syn-rich α -stannyl alcohols¹⁶ via the above-mentioned procedure, was found to afford the syn-rich products 6 (Scheme 3),17

Scheme 3



In summary, this work presents the strongest pieces of evidence in support of complete inversion of configuration at the Li-bearing migrating terminus in the [2,3]Wittig process via Sn/Li exchange. A key unanswered question is whether this inversion course is also the case for the synthetically more useful [2,3]Wittig variants induced by *direct lithiation*. In addition, this study opens a way to otherwise difficult preparations of enantiomerically-defined (α -alkoxy)alkylstannanes which would provide great opportunity to investigate the stereochemistry of various carbanion reactions. Further works along these lines are in progress.

References and Notes

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- 7. All attempts to prepare 2 via O-allylation of 1 with allyl halides or allyl tosylate failed.
- The completete inversion of configuration in this S_N2 reaction was proved by the transformation of (R)-2a (62% ee) to (R)-1 (62% ee) via the double bond migration (t-BuOK, DMSO, 50 °C) followed by hydrolysis of resulting enol ether (aq. HCl, acetone).
- The %ee refers to that of the starting stannyl alcohol employed, which was determined by ¹H NMR of the MTPA ester.

- 10. The %ee was determined by ¹H NMR of the MTPA ester and the (R)-configuration was assigned by ¹H NMR comparison of the MTPA ester of 1-*t*-butyldiphenylsiloxy-4-hexanol derived from **3a** (BH3•THF \rightarrow *t*-BuPh2SiCl) with an authentic sample prepared from (R)-(+)- γ -caprolactone.
- 11. The %ee was determined by HPLC analysis of its MTPA ester and the %de was determined by capillary GLC analysis. The *anti*-configuration was assigned by its GLC comparison with an authentic *syn*-rich sample prepared by the literature method: Yamamoto, Y.; Yatagai, H.; Naruta, Y.; Maruyama, K. J. Am. Chem. Soc. 1980, 102, 7107.
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- 14. Interestingly, the rearrangement of (Z)-2b (racemic) was found to provide syn-3b selectively (syn/anti = 90:10).
- Quite recently R. Brückner (Würzburg) has also obtained evidence for inversion of configuration in the rearrangement using a related, diastereomerically-defined α-(allyloxy)stannane (personal communication). We thank him for sending the manuscript prior to publication.
- 16. Prepared via reaction of 3-benzyloxy-2-methylpropanal with (n-Bu)3SnLi in THF (syn/anti = 66 : 34) and partially separated by column chromatography to afford the mixture of 82% syn. The syn-configuration of the major isomer was assigned by its transformation (with complete retention of configuration)¹³ to 1-benzyloxy-2-methyl-3-butanol (MOMCl / (i-Pr)2NEt → n-BuLi / Me2SO4 → HCl, MeOH) of which ¹H NMR data were correlated with those reported for an anti-rich mixture: Still, W. C.; Schneider, J. A. Tetrahedron Lett. 1980, 21, 1035.
- 17. The stereochemical assignment was made by ¹H NMR comparison with the litrature values: Heathcock, C. H.; Kiyooka, S.; Blumenkopf, T. A. J. Org. Chem. 1984, 49, 4214. The diastereomeric ratio was determined by ¹³C NMR assay. The reason for the discrepancy of %de between 5 and 6 is not clear at present, but might be due to the asymmetric induction involved.

(Received in Japan 3 July 1992)