- N. S. Fedotov, E. V. Grishina, and V. D. Sheludyakov, Zh. Obshch. Khim., <u>56</u>, No. 7, 1554– 1547 (1986).
- V. I. Shvedov, O. A. Safonova, I. Ya. Korsakova, et al., Khim.-farm. Zh., <u>14</u>, No. 2, 54-57 (1980).
- 10. I. Tabushi and R. Oda, J. Chem. Soc. Jpn. Pure Chem. Sect., 89, No. 8, 794-797 (1968).

SYNTHESIS AND BIOLOGICAL ACTIVITY OF BENZIMIDAZOLE-5-CARBOXAMIDES

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We have previously reported [4] that in 2-hydrazinobenzimidazoles, the greatest antiviral activity is shown by compounds bearing a carboxy or ethoxycarbonyl group in the 5-position. Continuing work on the synthesis and examination of the antiviral activity of benzimidazoles, we have synthesized some benzimidazole-5-carboxamides substituted in the 2-position with chloro-, hydrazine-, and isopropylidene (and isobutylidene)hydrazino-groups (III-XIX).



 $\begin{array}{l} R = Me \; (Ia - VIa, IXa - X'a, XVa), \; CH_2Ph \; (Ib - VIb, VII, `VIII, IXb - XIb, XII - XIV, XVb, XVI - XIX); \; R^1 = CONMe_2 \; (VII, XIII), \; CONEt_2 \; (III, IX, XV, XVIII), \; CONHCMe_3 \; (VI, XII, XVII), \; CONHCH_2Ph \; (IV, X, XVI, XIX), \; CONHC_{18}H_{37} \; (V, XI), \; CON(CH_2CHMe_2)_2 \; (VIII, \; XIV); \; R^2 = Cl(III - VIII) \end{array}$

These compounds have become accessible as a result of our observation of the replacement of the mercapto-group by chlorine on treatment of the thiones (Ia, b) with thionyl chloride [3], high yields of 1-methyl(and benzyl)-2-chlorobenzimidazole-5-carbonyl chlorides (IIa, b) being obtained. Treatment of the acid chlorides with amines (diethylamine, benzylamine, octadecylamine, tert-butylamine, and diisobutylamine) gave the 1-methyl (and benzyl)-2-chlorobenzmidazole-5-carboxamides (IIIa, b-VIa, b, VII, and VIII). Hydrazinolysis of the 5-amido-2chlorobenzimidazoles proceeded under milder conditions than in the case of the 5-unsubstituted 2-chlorobenzimidazole, namely on boiling (III-VIII) with an excess of hydrazine hydrate for 15-30 min. It should be noted that some of the hydrazino-compounds are unstable, darkening rapidly on keeping, and for this reason only (IXa, b-XIa, b, XII, and XIV) could be synthesized and submitted for testing for antiviral and antimicrobial activity. On brief boiling in acetone or ethyl methyl ketone, the corresponding hydrazones (XVa, b, XVI-XIX) were obtained. The structures of the hydrazino-compounds were confirmed by their PMR spectra, and their conversion into the hydrazones. The PMR spectra of the products showed a complex multiplet in the aromatic proton region (6.8-7.7 ppm), and a broadened signal for the NH proton at 3.5-4.5 ppm. The signals for the substituents in the amide group were in full agreement with the multiplicity and chemical shifts of the structures assigned to them.

The 2-chloro- and 2-hydrazino-derivatives and the hydrazones were examined for antiviral (against A and B influenza viruses) and antimicrobal activity. The testing results for those

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<u></u>	Yield, %		Found, %				Calculated, %		
Com- pound		mp, °C (solvent)	C	н́.	N	Empirical formula	С	н	N
IIIa IIIb IVa IVb Vb Vla Vb Vla Vlb Vl1 Vl1 Vl1 Vl1 Vl1 Vl1 Xb Xb Xla Xb Xla Xlb Xl1 XIV XVa XVb XVb	54 70 75 68 70 65 75 68 64 60 80 75 68 75 68 75 68 75 68 75 68 75 68 75 68 75 68 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 68 75 75 75 68 75 75 75 75 75 75 75 75 75 75 75 75 75	203-5 (acetonitrile) 108-9 (30°_0} ethanol) 158-60 (50°_0} ethanol) 168-9 (isopropanol) 122-5 (ethyl acetate) 125-7 (ethyl acetate) 160-1 (ethanol) 166-7 (DMF) 118-20 (benzene) 103-5 (ethyl acetone) 160-1 (benzene) 160-1 (benzene) 160-1 (benzene) 160-1 (benzene) 160-1 (benzene) 160-1 (benzene) 160-1 (benzene) 160-1 (ethyl acetate) 102-1 (ethyl acetate) 118-9 (ethanol) 198-9 (ethanol) 198-9 (ethanol) 130-2 (ethyl acetate) 172-3 (acetone) 167-8 (acetone)	60,5 66,64 69,9 73,68 665,64 69,9,9 73,68 665,61 69,7,7 67,7,7 67,7,7 69,9 67,7,7 69,9,9 65,69,1,7 70,78 69,9,9 65,69,9 65,9,9 7,9	6,32 6,4,89 9,01 5,57,7,6,8,60 10,79 6,29 10,77 6,8,60 10,79 6,29 7,79 6,80 10,77 6,80 10,77 6,80 10,77 6,80 10,77 7,76 8,777 7,76 8,777 10,7777 10,7777 10,7777 10,7777 10,7777 10,7777 10,7777 10,7777 10,77777 10,77777 10,77777 10,7777777777	$\begin{array}{c} 15,3\\12,4\\13,9\\11,2\\9,1\\7,8\\16,1\\12,5\\13,4\\10,2\\27,0\\21,0\\23,85\\15,3\\12,1\\20,5\\15,3\\12,1\\20,5\\22,4\\23,1\\17,4\\23,1\\18,6\\2\end{array}$	$\begin{array}{c} C_{14}H_{18}N_{3}CIO\\ C_{19}H_{20}N_{3}CIO\\ C_{19}H_{14}N_{3}CIO\\ C_{22}H_{14}N_{3}CIO\\ C_{22}H_{14}N_{3}CIO\\ C_{23}H_{48}N_{3}CIO\\ C_{33}H_{48}N_{3}CIO\\ C_{19}H_{20}N_{3}CIO\\ C_{19}H_{20}N_{3}CIO\\ C_{19}H_{20}N_{3}CIO\\ C_{19}H_{20}N_{3}CIO\\ C_{12}H_{19}N_{5}O\\ C_{13}H_{19}N_{5}O\\ C_{13}H_{19}N_{5}O\\ C_{19}H_{23}N_{5}O\\ C_{12}H_{11}N_{5}O\\ C_{33}H_{52}N_{5}O\\ C_{33}H_{52}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{17}H_{20}N_{5}O\\ C_{18}H_{21}N_{5}O\\ C_{18}H_{22}N_{5}O\\ C_{18}H_{22}N_{5}O\\ C_{18}H_{22}N_{5}O\\ C_{18}H_{28}N_{5}O\\ C_{18}H_{28}N_{5}$	60.1 66.8 64.1 70.3 73.4 58.8 65.1 69.4 59.8 65.0 71.1 69.5 67.6 65.9 65.5 67.6 65.2 63.7 69.8 70.2 70.2 71.8 65.1 70.8 65.1 70.8 65.1 70.8 70.2 70.2 70.2 70.2 70.2 70.2 70.2 70.2	6,59 4,86819 4,86819 5,57 7,9887 5,57 7,9887 109,96 6,97 7,4	$\begin{array}{c} 15.0\\ 12.3\\ 14.0\\ 9.1\\ 7.4\\ 15.8\\ 12.3\\ 13.4\\ 10.7\\ 27.5\\ 20.8\\ 23.7\\ 18.9\\ 15.3\\ 12.3\\ 20.8\\ 22.6\\ 17.8\\ 23.3\\ 18.5\\ 17.8\\ 23.3\\ 18.5\\ 17.8\\ 23.3\\ 18.5\\ 17.8\\ 23.3\\ 18.5\\ 17.8\\ 17.8\\ 18.5\\ 17.8\\ 18.5\\ 17.8\\ 18.5\\ 17.8\\ 18.5\\ 17.8\\ 18.5\\ 18.5\\ 17.8\\ 18.5\\ 1$
XVI XVII XVIII XIX	70 70 72 68	1220-1 (acetone) 178-9 (ethanol) 168-9 (30% ethanol) 194-5 (ethanol	70,4 70,5 73,4	0,3 7,2 7,5 6,8	17,3 18,7 17,9 16,2	$\begin{array}{c} C_{25}H_{25}N_{5}O\\ C_{22}H_{27}N_{5}O\\ C_{23}H_{29}N_{5}O\\ C_{32}H_{29}N_{5}O\\ C_{32}H_{30}N_{5}O\\ C_{32}H_{30}N_{5}O\\ C_{32}H_{30}N_{5}O\\ C_{33}H_{30}N_{5}O\\ C_{33}H_{30}N_{5}O\\$	72,9 70,0 70,5 73,2	6,1 7,2 7,5 6,6	17,0 18,6 17,2 16,4

TABLE 1. Properties of 1,2,5-Substituted Benzimidazoles

TABLE 2. Antiviral Activity of Benzimidazoles*

Componed	Influenza	a virus A	Influenza virus B				
Componing	in ovo	in vivo	in ovo	in vivo			
IX b XI a XIV XVa XVI XVI XVII XIX			 ++++ ++++	+ : : : : : : : : : : : : : : : : : : :			

*Index of effectiveness: -) less than 30%, +) 30-39%, ++) 40-59%, +++) 60-79%.

compounds which showed activity are given in Tables 2 and 3. As will be seen from Table 2, the hydrazinobenzimidazoles (IXb), (XIa), and (XIV), and the hydrazones (XV-XVII) and (XIX) were active against both A and B influenza viruses. In tests on developing chick embryos, the index of effectiveness of these drugs was 33-71%. In animal experiments, the index was no greater than 54%, for the most part against influenze virus A. It is interesting that in some instances the activity in mice was greater than in chick embryos (compounds (XIV), (XVI), and (XIX)). 2-Chlorobenzimidazoles failed to show any activity.

Antibacterial and antifungal activity was determined in Gram-positive and Gram-negative representative pathogenic microorganisms, yeasts, and phytopathogenic fungi. The test results are shown in Table 3. 1-Methyl(and benzyl)-2-chlorobenzimidazoles carrying a tert-butylamino-carbonyl substituent in the 5-position showed high antifungal activity, inhibiting the growth of <u>Verticillium dahliae</u> at concentrations of 20-50 μ g/ml. No antibacterial activity was shown by these compounds. The hydrazino-compounds (IXa, b) and (XII), and the hydrazone (XVb) inhibited the growth of both Gram-positive and Gram-negative microorganisms. The bacteriostatic dose was 50-400 μ g/ml.

These results of tests for antiviral, antibacterial, and antifungal activity therefore suggest that a further search for new antifungal and antimicrobial compounds in these benzimidazoles holds promise.

TABLE 3. Antimicrobial Activity of Benzimidazoles

	Bacteriostatic and Fungistatic doses, µg/ml									
Com- pound	Staphylococ- cus aureus	Staphylococ- cus albus	Bacillus ce- reus	Corynebacte rium deve- ricatum	Mycobacte- rium Bs	E. coli	Proteus vul- garis	Pseudomonas aeruginosa	Candida albi- cans	Verticilitum dahliae,
VIa VIb IXa IXb XII XV b		200 50 400 100							100 400 400 200 400	20 50 400 400 400 400

Note. None of the test compounds showed antimicrobial activity.

EXPERIMENTAL CHEMICAL

<u>1-Methyl(and benzyl)-2-chloro-5-aminocarbonylbenzimidazoles (III-VIII).</u> The acid chloride (IIa, b) (0.01 mole) was dissolved in benzene, and a benzene solution of the appropriate amine added. The mixture was kept at 20-30°C for 3 h, evaporated to dryness, and the residue crystallized from a suitable solvent. The properties of the compounds are given in Table 1.

<u>1-Methyl(and benzyl)-5-aminocarbonyl-2-hydrazinobenzimidazoles</u> were obtained by boiling 0.01 mole of the 2-chlorobenzimidazole in 10-15 ml of hydrazine hydrate for 30 min. The solid which separated on cooling was filtered off and crystallized. If no solid separated on cooling the reaction mixture, it was precipitated with water.

Isopropylidene(and isobutylidene)hydrazones (XVa, b) and (XVI-XIX) were obtained by boiling the hydrazino-compounds in acetone or ethyl methyl ketone for 30 min, followed by evaporation to dryness. The properties of the compounds are given in Table 1.

EXPERIMENTAL BIOLOGICAL

Antimicrobial activity was examined by serial dilution of the drug in liquid nutrient media which were optimum for the growth of the test cultures used, against Gram-positive and Gram-negative representative pathogenic microorganisms, dermatophytes, yeasts, and phytopathogenic fungi [1]. The strains of <u>Staphylococcus</u> <u>aureus</u>, <u>S. Albus</u>, <u>Bacillus</u> <u>cereus</u>, <u>Mycobacterium</u> <u>B_s</u>, <u>Escherichia</u> <u>coli</u>, <u>Proteus</u> <u>vulgaris</u>, <u>Pseudomonas</u> <u>aeruglinisa</u> were grown on meatpeptone agar and broth for 18-24 h at 37°C, the strain of <u>Corynebacterium</u> <u>devericatum</u> under the same conditions at 28°C, the strain of yeast <u>Candida</u> <u>albicans</u> at 37°C for 18-24 h on agarized must, and the pathogenic fungi strain <u>Verticillium</u> <u>dahliae</u> on the same medium for 72-168 h at 28°C. The microbial loading was 200,000 microbial bodies per ml of medium. The antimicrobial activity of the compound was expressed as the minimum bacteriostatic dose giving total inhibition of the growth of the test culture.

Examination for antiviral activity was carried out in developing chick embryos and in white mice infected with influenza viruses A and B. In each test for activity in developing chick embryos and white mice three groups were used: embryos (mice) treated with the test compound, embryos (mice) treated with a compound active against the test virus (remantadine in the case of the A group virus and adapromine in testing for activity with type B virus), and embryos (mice) treated with a placebo (the solvent, usually physiological saline or distilled water). In tests on mice, five doses of the compound were given, at 24 h and 1 h before infection, and 24, 48, and 72 h after infection. This mode of testing comprises simultaneously tests for prophylactic and therapeutic activity. The examinations and assessment of the activity of the compounds was carried out as described in the methodological instructions approved by the Ministry of Health of the USSR [2].

Examinations for antimicrobial activity were carried out in the D. K. Zabolotnii Institute of Microbiology and Virology of the Academy of Sciences of the Ukrainian SSR (Kiev), and for antiviral activity at the All-Union Research Institute for Influenza, Ministry of Health of the USSR (Leningrad).

LITERATURE CITED

- 1. N. S. Egorov, Scientific Fundamentals of Antibiotics [in Russian], 3rd edition, Moscow (1979), pp. 46-47.
- 2. V. I. Il'enko, Methods of Testing and Assessment of the Antiviral Activity of Chemical Compounds against Influenze Virus. Methodology [in Russian], Leningrad (1977).
- 3. G. A. Mokrushina, S. K. Kotovskaya, and G. A. Yurchenko, Conference on the Chemistry and Technology of Organic Sulfur Compounds and Sulfur-Containing Petroleum Oils. No. 16: Abstracts of Papers [in Russian], Riga, 1984, p. 187.
- 4. G. A. Mokrushina, S. K. Kotovskaya, G. N. Tyurenkova, et al., Khim.-farm. Zh., No. 2, 195 (1988).

SYNTHESIS AND ANTI-INFLUENZA ACTIVITY

OF N-SUBSTITUTED-2-BENZAZOL-2-YLHYDRAZINOCARBOTHIOAMIDES

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The search for drugs with antiviral activity is currently being carried out amongst various types of compounds. Many compounds with antiviral activity have been found in heterocycles, in particular benzimidazole and benzothiazole [1, 5, 6]. A number of ureido, thioureido, and hydrazinocarbothiamides (thiosemicarbazides) are known to display antirival activity [1, 2, 4]. In order to study anti-influenza activity, some compounds have been synthesized containing two potentially active groupings, benzazole and hydrazinocarbothioamide, namely N-substituted-2-(benzazol-2-y1)hydrazinocarbothioamides of general formula:



$$\begin{split} & (I-XXVII X = NH (I, II), NCH_3 (III-VI, XI, XII, XVII, XVIII, XXII, XXIII), NCH_2C_9H_5 (VII-X, XV, XVI), NC_4H_9 (XIII, XIV), NC_6H_5 (XIX), NCH_2CH_9CH (XX, XXI), S (XXIV, XXV), O (XXVI, XXVII); R = CH_3 (III, VII), C_9H_5 (V, IX, XII, XV, XVII), P^-CIC_6H_4 (II, VI, X, XII, XIV, XVI, XVII, XIV, XVII, XIV), P^-CIC_6H_4 (II, VI, X, XII, XIV, XVI, XVII, XVII, XXII, XXVI), P^-C2H_5OC_6H_4 (I), allyl·(IV, VIII, XI, XXIV); R^1 = H (I-X, XIX-XXI, XXIV-XXVII), CO(NH)_2C(=S)NHC_6H_5 (XXII), CO(NH)_2C(=S)NHC_6H_4CI P (XXIII); R^2 = H (I-X, XIII-XXVII), CH_3 (XI, XII). \end{split}$$

Compounds (I-XXVII) were obtained from the appropriate 2-hydrazinobenzazoles and alkyl (or aryl) isothicyanates at ambient temperature or with gentle heating. The starting 2-hydrazinobenzazoles were obtained by hydrazinolysis of 2-chloro- or 2-methylsulfonyl-compounds, or of the benzazole-2-sulfonic acids:



(for (XVII), (XVIII), and (XX-XXII) as in [1]);



(for (XI-XVI) and (XIX) as in [2, 4]);

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