This article was downloaded by: [Dalhousie University] On: 15 June 2013, At: 02:30 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Phosphorus, Sulfur, and Silicon and the Related Elements

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/gpss20

Studies of Thiazolopyridines, Part 6: Synthesis and Antimicrobial Evaluation of Some Novel Thiazolo[3,2-a] Pyridine and Thiazolo[2',3':6,1]-pyrido[2,3d]pyrimidine Derivatives

G. A. M. EI-Hag Ali $^{\rm a}$, A. K. Khalil $^{\rm b}$, R. Q. Lamphon $^{\rm c}$ & A. A. EI-Maghraby $^{\rm a}$

^a Department of Chemistry, Al-Azhar University, Nasr City, Cairo, Egypt

^b Department of Chemistry, Ain Shams University Cairo, Egypt

^c Department of Chemistry, King Abdel-Aziz University, Madine Munawarah, Saudia Published online: 18 Aug 2006.

To cite this article: G. A. M. El-Hag Ali , A. K. Khalil , R. Q. Lamphon & A. A. El-Maghraby (2005): Studies of Thiazolopyridines, Part 6: Synthesis and Antimicrobial Evaluation of Some Novel Thiazolo[3,2-a] Pyridine and Thiazolo[2',3':6,1]-pyrido[2,3d]pyrimidine Derivatives, Phosphorus, Sulfur, and Silicon and the Related Elements, 180:8, 1909-1919

To link to this article: <u>http://dx.doi.org/10.1080/104265090889620</u>

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.



Studies of Thiazolopyridines, Part 6: Synthesis and Antimicrobial Evaluation of Some Novel Thiazolo[3,2-a] Pyridine and Thiazolo[2',3':6,1]-pyrido[2,3-d]pyrimidine Derivatives

G. A. M. El-Hag Ali

Department of Chemistry, Al-Azhar University, Nasr City, Cairo, Egypt

A. K. Khalil

Department of Chemistry, Ain Shams University Cairo, Egypt

R. Q. Lamphon

Department of Chemistry, King Abdel-Aziz University, Madine Munawarah, Saudia

A. A. El-Maghraby

Department of Chemistry, Al-Azhar University, Nasr City, Cairo, Egypt

Condensation of thiazolinone 1 with aromatic aldehydes yielded the corresponding methylidene derivatives 2a–f. Cyclization of compounds 2a–f with arylidenemalononitrile 3 (1:1 molar ratio) in ethanol in the presence of piperidine furnished the novel thiazolo[3,2-a]pyridines 5a–v, via Michael adduct 4. Compounds 5p,r were cyclized with malononitrile in the presence of piperidine to yield thiazolo[3,2a][1,8]naphthryidines 7a,b. Thiazolo-[2',3':1,6]pyrido[2,3-d]pyrimidine 9a–c were obtained by cyclization of compounds 5c,p,r with formic acid. The structure of the synthesized compounds was established by analytical and spectral data. Also, some of the synthesized compounds were screened for antimicrobial activity in vitro.

Keywords Naphthryidine; thiazoline; thiazolo[3,2-a]pyridine;

Thiazolo[3,2-a] pyridines were reported to furnish various biological and pharmacological activities such as antimicrobial,¹ bactericide,² coronary dilator, antihypertensive, and muscle relaxant³ activities. In continuation with our work on the synthesis of thiazolo[3,2-a]pyridines, from readily available starting materials,^{4–7} we report here on the synthesis of thiazolo[3,2-a] pyridine, thiazolo[3,2-a][1,8]naphthyridine and

Received June 6, 2004; accepted September 7, 2004.

Address correspondence to A. A. El-Maghraby, Department of Chemistry, Al-Azhar University, Nasr City, Cairo, Egypt. E-mail: m_elgaby@hotmail.com

thiazolo[2',3':1,6]pyrido-[2,3-d]pyrimidine derivatives in order to investigate the antimicrobial activity of them.

RESULTS AND DISCUSSION

The starting material 1 was synthesized by cyclocondensation of cyanoacetamide with thioglycolic acid.⁸ Thiazoline 1 was condensed with various aromatic aldehydes in ethanol in the presence of piperidine under reflux to furnish 2-acetamido-5-arylmethylidene-4,5-dihydro-4thiazolinones **2a-f** (Scheme 1). Cyclocondensation of compounds **2a-f** with benzylidenemalononitriles $\mathbf{3}$ (1:1 molar ratio) in ethanol at reflux temperature in the presence of piperidine yielded the novel thiazolo[3,2a]-pyridines 5a-v in high yields (Scheme 1). The structures of all the compounds 5a-c were established using microanalysis and spectroscopy. The infrared spectra of thiazolopyridines 5a-v exhibited the presence of amino, cyano, and carbonyl (thiazolinone and carbamoyl) function groups. Also, the ¹H-NMR spectra revealed a signal characteristic for the 4H-pyridine. The formation of thiazolopyridine 5 is assumed to proceed via Michael addition of methylene function group in compound 2 to the benzylidene moiety 3 to yield Michael adduct 4 followed by intramolecular cyclization⁷ at the cyano group to form thiazolopyridine 5 (Scheme 1). Also, the structure of thiazolopyridine 5 was established by another synthetic route by ternary condensation of compound 2, aromatic aldehyde, and malononitrile (1:1:1 molar ratio) in ethanol in the presence of piperidine under reflux.

The reactivity of thiazolopyridine **5** towards malononitrile was studied. Thus, when compound **5p**,**r** were allowed to react with malononitrile in ethanol containing piperidine under reflux, the condensed pyridine derivatives **7a**,**b** were obtained. On the basis of analytical and spectral data, the other possible structure **6** was ruled out. The infrared spectrum of compound **7a** showed $\nu_{C=O}$ at 1698 cm⁻¹ characteristic for thiazolinone and $\nu_{C=O}$ at 1674 cm⁻¹ (CONH₂). Formation of **7** is assumed to proceed through the addition of an amino group in **3** to the cyano group of malononitrile followed by intramolecular cyclization to furnish **7** (Scheme 2).

Cyclization of compounds **5c**,**p**,**r** with formic acid furnished the novel thiazolo[2',3':6,1]pyrido[2,3-d]pyrimidines **9a–c**, via intermediate amide formation⁹ **8** followed by intramolecular cyclization with formic acid. Also, cyclocondensation of compound **5c** with acetic anhydride under reflux yielded the corresponding thiazolopyridopyrimidine derivative **10** (Scheme 3). Compound **10** was characterized through the absence of cyano functional group in its infrared spectrum.

.CONH₂ CONH₂ ArCHO / piperidine EtOH / reflux Ar-HC (1) (2a-f) SH Ar'CH=0 ċн₂соон (3) CH2CONH2 EtOH / piperidine ĊΝ reflux CN -CN H₂N H₂N NH₂ Ar-HC Ar-HC (5a-v) (4) $2a; Ar = C_6H_4Cl-4$ **2b**; Ar= $C_6H_4CH_3$ -4 2c; Ar= C_6H_4Cl-2 2d; Ar= C_6H_4OH-4 2e; Ar= $C_6H_4OCH_3$ -4 $2f; Ar = C_6H_3(OCH_3)(OH)-3,4$ 5a; Ar= C_6H_4Cl-4 , Ar'= $C_6H_3(OCH_3)(OH)-3.4$ **5b**; Ar= C_6H_4Cl-4 , Ar'= C_6H_4Cl-2 5c; Ar= C_6H_4Cl-4 , Ar'= C_6H_4OH-4 5d; $Ar = C_6H_4CH_3-4$, $Ar' = C_6H_4OH-4$ 5e; Ar= $C_6H_4CH_3$ -4, Ar'= C_6H_4Cl -2 5f; Ar= $C_6H_4CH_3$ -4, Ar'= C_6H_4Cl -4 5g; Ar= C_6H_4Cl-2 , Ar'= C_6H_4OH-4 5h; Ar= C_6H_4Cl-2 , Ar'= $C_6H_3(OCH_3)(OH)-3,4$ 5i; Ar= C_6H_4Cl-2 , Ar'= C_6H_4Cl-4 5j; Ar= C_6H_4OH-4 , Ar'= $C_6H_4CH_3-4$ 5l; Ar= C₆H₄OH-4 , Ar'= C₆H₄Cl-4 5k; Ar= C_6H_4OH-4 , Ar'= C_6H_4Cl-2 $5m; Ar = C_6H_4OH-4, Ar' = C_6H_4OCH_3-4$ $5n; Ar = C_6H_4OH-4, Ar' = C_6H_3(OCH_3)(OH)-3,4$ 50; Ar= $C_6H_4OCH_3$ -4, Ar'= $C_6H_3(OCH_3)(OH)$ -3,4 5p; Ar= $C_6H_4OCH_3$ -4, Ar'= C_6H_4Cl -2 $5q; Ar = C_6H_4OCH_3-4, Ar' = C_6H_4Cl-4$ $5r; Ar = C_6H_4OCH_3-4, Ar' = C_6H_4CH_3-4$ 5s; $Ar = C_6H_4OCH_3-4$, $Ar' = C_6H_4OH-4$ 5t; Ar= $C_6H_3(OCH_3)(OH)$ -3,4, Ar'= C_6H_4OH -4 5u; Ar= $C_6H_3(OCH_3)(OH)-3,4$, Ar'= $C_6H_4CH_3-4$ 5v; Ar= $C_6H_3(OCH_3)(OH)-3,4$, Ar'= C_6H_4Cl-2 **SCHEME 1**

ANTIMICROBAL ACTIVITY

Some of the synthesized compounds were screened *in vitro* for their antimicrobal activities against three strains: *Staphylococcus aureus*



7a; Ar= C₆H₄OCH₃-4 , Ar'= C₆H₄Cl-2

7b; Ar= C₆H₄OCH₃-4, Ar'= C₆H₄CH₃-4

SCHEME 2



9a; Ar= C_6H_4Cl-4 , Ar'= C_6H_4OH-4 9b; Ar= $C_6H_4OCH_3-4$, Ar'= C_6H_4Cl-2 9c; Ar= $C_6H_4OCH_3-4$, Ar'= $C_6H_4CH_3-4$ 10; Ar= C_6H_4Cl-4 , Ar'= C_6H_4OH-4

SCHEME 3

Compound no.	Staphylococcus aureus (NCTC-7447)	Saracia maxima (ATCC-33910)	Aspergillus funigatus
5b	16	19	14
5d	12	12	18
5f	12	19	18
5h	14	17	14
5k	17	12	16
5p	19	18	15
5q	15	18	17
5 r	18	14	12
5 s	20	20	25
5t	12	13	13
5v	13	18	12
7a	20	21	20
9b	21	20	22

TABLE I Antimicrobial Activity of the Synthesized Compounds and Inhibition Zones (mm). Standard: for Gram Positive and Gram Negative Bacteria, Ampicillin 25 μ g mL⁻¹; for fungi: Mycostatine 30 μ g mL⁻¹

(NCTC-7447), Saracia maxima (ATCC-33910), and Aspergillus funigatus by the agar diffusion techniques.¹⁰ The tested compounds were dissolved in *N*, *N*-dimethylformamide (DMF) to get a solution of 1000 μ g mL⁻¹ concentration. The bacteria and fungi cultures were maintained on nutrient agar and Czapek's-Dox agar media, respectively. DMF showed no inhibition zones. The agar media were incubated with different microorganisms culture tested. After 24 h of incubation at 30°C for bacteria and 48 h of incubation at 28°C for fungi, the diameter of inhibition zone (mm) was measured (Table I). Ampicillin in a concentration 25 μ g mL⁻¹ and mycostatine in a concentration 30 μ g mL⁻¹ were used as references for antibacterial and antifungal activities, respectively. The results are illustrated in Table I. None of the tested compounds showed a superior activity than the reference.

EXPERIMENTAL

Melting points are uncorrected. IR spectra were recorded (KBr) on a Perkin Elmer 1650 spectrophotometer. ¹H NMR spectra were recorded on a Varian Gemini spectrometer 200 (200 MHz), using DMSO-d₆ as a solvent and TMS as an internal standard. Chemical shifts are expressed as δ ppm units. Microanalytical data were obtained from the Microanalytical Data Unit at Cairo University. Physical data for the synthesized compounds are given in Table II. Also, the spectral data are collected in Table III.

Compound	Yield	Solvent	М.р.	Molecular	Elemental analysis Calcd./Found %		
no.	(%)	cryst.	(°C)	formula (Mol. Wt.)	С	Η	Ν
2a	62	Benzene/ ethanol	240-2	$\begin{array}{c} \mathrm{C_{12}H_9ClN_2O_2S}\\ (280) \end{array}$	$51.42 \\ 51.40$	$3.21 \\ 3.00$	$10.00 \\ 10.10$
2b	57	Ethanol	197–8	$\substack{C_{13}H_{12}N_2O_2S\\(260)}$	$\begin{array}{c} 60.10\\ 60.05 \end{array}$	$\begin{array}{c} 4.61 \\ 4.60 \end{array}$	$10.76 \\ 10.70$
2c	71	Benzene	241–3	$\begin{array}{c} {\rm C}_{12}{\rm H}_9{\rm ClN}_2{\rm O}_2{\rm S}\\ (280) \end{array}$	$51.42 \\ 51.30$	$3.21 \\ 3.20$	$\begin{array}{c} 10.00\\ 10.10\end{array}$
2d	83	Benzene/ ethanol	190–2	$\begin{array}{c} C_{12}H_{10}N_{2}O_{3}S\\ (262)\end{array}$	$54.96 \\ 54.80$	$3.81 \\ 3.90$	$\begin{array}{c} 10.68\\ 10.70 \end{array}$
2e	62	Benzene	230–2	$\begin{array}{c} C_{13}H_{12}N_{2}O_{3}S\\ (276) \end{array}$	$\begin{array}{c} 56.52 \\ 56.50 \end{array}$	$\begin{array}{c} 4.34\\ 4.30\end{array}$	$\begin{array}{c} 10.14\\ 10.10\end{array}$
2f	56	Benzene	220–2	$\begin{array}{c} C_{13}H_{13}N_{2}O_{4}S\\ (293) \end{array}$	$\begin{array}{c} 55.42 \\ 55.40 \end{array}$	$\begin{array}{c} 4.43 \\ 4.40 \end{array}$	$9.55 \\ 10.00$
5a	63	Dioxane	256-8	$\begin{array}{c} {\rm C}_{23}{\rm H}_{17}{\rm ClN}_{4}{\rm O}_{4}{\rm S}\\ (480.5)\end{array}$	$57.44 \\ 57.40$	$\begin{array}{c} 3.52 \\ 3.50 \end{array}$	$\begin{array}{c} 11.65\\ 11.60 \end{array}$
5b	77	Benzene	258–9	$\begin{array}{c} C_{22}H_{14}Cl_2N_4O_2S \\ (469) \end{array}$	$56.28 \\ 56.30$	$\begin{array}{c} 2.98\\ 3.00 \end{array}$	$\begin{array}{c} 11.94 \\ 11.90 \end{array}$
5c	81	Benzene	286–8	$C_{22}H_{15}ClN_4O_3S$ (450.5)	$58.60 \\ 58.60$	$3.32 \\ 3.30$	$12.43 \\ 12.40$
5d	69	Benzene	>300	$C_{23}H_{18}N_4O_3S$ (430)	64.18 64.20	$4.18 \\ 4.20 \\ 0.70$	13.02 13.10
5e	88	Benzene	220-2	$C_{23}H_{17}CIN_4O_2S$ (448.5)	61.53 61.50	3.79 3.80	12.48 12.50
51	79	Benzene	226-8	$C_{23}H_{17}CIN_4O_2S$ (448.5)	61.53 61.50	3.79 3.80	12.48 12.40
og sh	74 64	Disuana	271-3	(450.5)	58.60 58.60	3.32 3.30 2.59	12.43 12.40
5;	66	Bonzono	202-4	(480.5)	57.44 57.40	3.52 3.50	11.60 11.60
51	75	Bonzono	> 200	(469)	56.30	2.98 2.90	11.94 11.90
5]-	80	Bongono/	270.2	(430)	64.20 58.60	4.20	13.10 12.43
51	90	ethanol	210-2	(450.5)	58.60 58.60	3.30 3.32	12.40 12.40
51 5m	30 79	Bonzono	200-2	(450.5)	58.60 61.88	3.30	12.40 12.55
511	82	Bonzono	262-4	(446)	61.88 61.70	4.000	12.55 12.50 12.12
50	50	Bonzono	254-0	(462)	59.74 59.70	3.90	12.12 12.10
50 5n	59 77	Diovano	100 9	(476)	60.50 60.40	4.20 4.20 3.65	11.70 11.80
чh		Dioxaile	190-2	(464.5)	59.41 59.40	$3.00 \\ 3.50$	12.05 12.10

TABLE II Characteristics Data for the Prepared Compounds

(Continued on next page)

Compound	Yield (%)	Solvent cryst.	M.p. (°C)	Molecular	Elemental analysis Calcd./Found %		
no.				formula (Mol. Wt.)	С	Η	Ν
5q	67	Ethanol	160–2	$C_{23}H_{17}ClN_4O_3S$ (464.5)	$59.41 \\ 59.40$	$3.65 \\ 3.40$	$12.05 \\ 12.00$
5r	87	Dioxane	258–9	$C_{24}H_{20}N_4O_3S$ (444)	$64.86 \\ 64.80$	$4.50 \\ 4.40$	$12.61 \\ 12.60$
5s	67	Benzene	240-2	$C_{23}H_{18}N_4O_4S$ (446)	$61.88 \\ 61.70$	$4.03 \\ 4.10$	$12.55 \\ 12.50$
5t	72	Dioxane	>300	$C_{23}H_{18}N_4O_5S$ (462)	$59.74 \\ 59.70$	$3.89 \\ 3.90$	$12.12 \\ 12.10$
5u	68	Benzene/ ethanol	238–9	$C_{24}H_{20}N_4O_4S$ (460)	$62.60 \\ 62.60$	$4.34 \\ 4.30$	$12.17 \\ 12.20$
5v	85	Dioxane	270–2	$C_{23}H_{17}ClN_4O_4S$ (480.5)	$57.44 \\ 57.40$	$3.53 \\ 3.50$	$11.65 \\ 11.60$
7a	69	Benzene/ ethanol	288–9	$C_{26}H_{19}ClN_6O_3S$ (530.5)	$58.81 \\ 58.80$	$3.58 \\ 3.60$	$15.83 \\ 15.80$
7b	59	Benzene/ ethanol	218–9	$C_{27}H_{22}N_6O_3S$ (510)	$63.52 \\ 63.50$	$\begin{array}{c} 4.31 \\ 4.20 \end{array}$	$\begin{array}{c} 16.74 \\ 16.70 \end{array}$
9a	70	Dioxane	290	$C_{23}H_{15}ClN_4O_4S$ (478.5)	$57.68 \\ 57.70$	$2.92 \\ 2.90$	$11.70 \\ 11.70$
9b	81	Benzene	210–2	$C_{24}H_{17}ClN_4O_3S$ (492.5)	$58.47 \\ 58.50$	$3.45 \\ 3.40$	$11.37 \\ 11.40$
9c	63	Dioxane	>300	$C_{25}H_{20}N_4O_4S$ (472)	63.55 63.50	4.233 4 20	11.86 11.50
10	71	Dioxane	>300	$C_{24}H_{17}ClN_4O_4S$ (492.5)	$58.48 \\ 58.50$	3.45 3.60	11.30 11.37 11.30

TABLE II Characteristics Data for the Prepared Compounds (Continued)

2-(Acetamido-2-yl)-5-arylmethylidene-4,5-dihydro-4thiazolinones (2a–f)

A mixture of compound 1 (0.01 mole), aromatic aldehyde (0.01 mole), and piperidine (0.5 mL) in ethanol (40 mL) was heated under reflux for 3 h. The product obtained was recrystallized from suitable solvent to give 2a-f.

5-Amino–7-aryl-8-carbamoyl-3-oxo-2-arylmethylidene-2,3dihydro-*H*-thiazolo[3,2-a]pyridine-6-carbonitriles (5a–v)

Method (A):

A mixture of compound 2 (0.01 mole), benzylidene-malononitrile 3 (0.01 mole), and piperidine (0.5 mL) in ethanol (30 mL) was heated

Compound no.	IR/. ν_{max} (cm ⁻¹)	¹ H NMR(δ /ppm)(DMSO- d_6)
2a	3379, 3147 (NH ₂), 2977 (CH-aliph), 1720 (C=O; thiazolinone), 1666 (C=O, amide).	3.15 (s, 3H, CH ₂), 5.40 (s, 2H, NH ₂), 6.40–7.62 (m, 5H, Ar-H and methine-H).
2b	3334, 3127 (NH ₂), 2977 (CH-alph), 1710 (C=O; thiazolinone), 1658 (C=O; amide).	2.20 (s, 3H, CH ₃), 3.10 (s, 2H, CH ₂), 5.20 (s, 2H, NH ₂), 6.30–7.70 (m, 5H, Ar-H and methine-H).
2e	3300, 3180 (NH ₂), 2980 (CH-aliph), 1712 (C=O; thiazolinone), 1660 (C=O; amide).	 3.20 (s, 2H, CH₂), 3.60 (s, 3H, OCH₃), 5.30 (s, 2H, NH₂), 6.25–7.82 (m, 5H, Ar-H and methine-H).
2f	3379, 3147 (NH ₂), 2930 (CH-aliph), 1720 (C=O; thiazolinone), 1666 (C=O; amide).	3.21 (s, 2H, CH ₂), 3.85 (s, 3H, OCH ₃), 5.76 (s, 2H, NH ₂), 6.90–7.46 (m, 4H, Ar-H and methine-H), 10.80 (s, 1H, OH).
5a	3471, 3433 (NH ₂), 2180 (C≡N), 1710 (C=O; thiazolinone), 1651 (C=O; amide).	3.81 (s, 3H, OCH ₃), 3.99 (s, 1H, pyridine-H), 5.65 (s, 2H, NH ₂), 6.66–7.25 (m, 8H, Ar-H and methine-H), 7.39 (s, 3H, NH ₂), 10.10 (br, 1H, OH).
5b	3386, 3170 (NH ₂); 2183 (C=N), 1697 (C=O; thiazolinone), 1651 (C=O; amide).	3.99 (s, 1H, pyridine-H), 5.76 (s, 2H, NH ₂), 6.73–7.50 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂).
5c	3433, 3332 (NH ₂); 2183 (C≡N); 1700 (C=O; thiazolinone), 1658 (C=O; amide).	4.08 (s, 1H, pyridine-H), 5.73 (s, 2H, NH ₂), 6.74–7.43 (m, 9H, Ar-H and methine-H), 7.47 (s, 1H, NH ₂), 10.90 (br, 1H, OH).
5d	3433, 3147 (NH ₂), 2109 (C=N); 1720 (C=O; thiazolinone), 1666 (C=O; amide).	2.19 (s, 3H, CH ₃), 4.15 (s, 1H, pyridine-H), 5.46 (s, 2H, NH ₂), 5.87–6.40 (m, 9H, Ar-H and methine-H), 6.42 (s, 2H, NH ₂), 10.30 (br, 1H, OH).
5e	3359, 3225 (NH ₂), 2191 (C≡N), 1681 (C=O; thiazolinone), 1651 (C=O; amide).	2.20 (s, 3H, CH ₃), 4.00 (s, 1H, pyridine-H), 5.10 (s, 2H, NH ₂), 6.74–7.51 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂).
5g	3433, 3320 (NH ₂), 2214 (C=N), 1720 (C=O; thiazoline), 1666 (C=O; amide).	4.09 (s, 1H, pyridine-H), 5.73 (s, 2H, NH ₂), 6.74–7.42 (m, 9H, Ar-H and methine-H), 7.54 (s, 2H, NH ₂), 10.90 (br, 1H, OH).
5h	3440, 3350 (NH ₂), 2191 (C=N), 1681 (C=O; thiazolinone), 1651 (C=O; amide).	3.90 (s, 3H, OCH ₃), 4.08 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.75–7.48 (m, 8H, Ar-H and methine-H), 8.04 (s, 2H, NH ₂), 10.90 (br, 1H, OH). (Continued on next page)

TABLE III Spectral Data of the Synthesized Compounds

Compound no.	IR/. v_{max} (cm ⁻¹)	¹ H NMR(δ/ppm)(DMSO-d ₆)
5i	3379, 3309 (NH ₂), 2190 (C≡N), 1720 (C=O; thiazolinone),	4.08 (s, 1H, pyridine-H), 5.70 (s, 2H, NH ₂), 6.82–7.33 (m, 9H, Ar-H and
	1666 (C=O; amide).	methine-H), 7.52 (s, 2H, NH ₂).
5k	3425, 3340 (NH ₂), 2180 (C≡N),	4.09 (s, 1H, pyridine-H), 5.73 (s,
	1700 (C=O; thiazolinone), 1666 (C=O; amide).	2H, NH ₂), 6.79–7.52 (m, 9H, Ar-H and methine-H), 7.62 (s, 2H, NH ₂), 10.72 (br. 1H, OH).
51	3386, 3170 (NH ₂), 2183 (C ≡ N);	4.24 (s, 1H, pyridine-H), 5.70 (s,
	1690 (C=O; thiazolinone), 1658 (C=O; amide).	2H, NH ₂), 6.74–7.58 (m, 9H, Ar-H and methine-H), 7.62 (s, 2H, NH ₂) 10 00 (br 1H OH)
5n	3433, 3355 (NH ₂), 2191 (C≡N), 1700 (C=O; thiazolinone), 1658 (C=O; amide).	3.74 (s, 3H, OCH ₃), 4.32 (s, 1H, pyridine-H), 5.71 (s, 2H, NH ₂), 6.73–7.46 (m, 8H, Ar-H and
_		methine-H), 10.0 (br, 2H, 2OH).
50	$3433, 3355 (NH_2), 2214 (C N),$	$3.76, 3.89 (2s, 6H, 20CH_3), 4.07 (s, 10, 10, 10, 10, 10, 10, 10, 10, 10, 10$
	1700 (C=O; thiazolinone), 1651 (C=O; amide).	1H, pyridine-H), 5.50 (s, 2H, NH ₂), 6.45–7.59 (m, 8H, Ar-H and methine-H), 8.03 (s, 2H, NH ₂), 11.20 (br. 1H. OH).
5р	3440, 3209 (NH ₂), 2214 (C≡N),	3.70 (s, 3H, OCH ₃), 4.12 (s, 1H,
-	1700 (C=O; thiazolinone),	pyridine-H), 5.71 (s, 2H, NH ₂),
	1651 (C=O; amide).	6.48–7.45 (m, 9H, Ar-H and
		methine-H), 7.51 (s, 2H, NH ₂).
5q	3355, 3232 (NH ₂), 2214 (C=N),	3.74 (s, 3H, OCH ₃), 3.99 (s, 1H,
	1720 (C=O; thiazolinone),	pyridine-H), 5.47 (s, 2H, NH ₂),
	1657 (C=O; amide).	6.03–7.42 (m, 9H, Ar-H and
_		methine-H), 11.70 (s, 2H, NH_2).
5r	$3463, 3348 (NH_2), 2214 (C=N),$	2.2 (s, 3H, CH_3), 3.67 (s, 3H,
	1689 (C=O; thiazolinone), 1651 (C=O; amide).	OCH_3), 3.99 (s, 1H, pyridine-H), 5.67 (s, 2H, NH ₂), 6.68–7.42 (m,
		9H, AF-H and methine-H), 7.78 (s, $9H$ NH)
59	3379 3147 (NH₂) 2191 (C≡N)	$3.80 (g 3H OCH_{2}) / 11 (g 1H$
03	1720 (C=0: thiazolinone)	$nvridine-H) 5.70 (s. 2H NH_s)$
	1666 (C=O; amide).	6.85–7.42 (m, 9H, Ar-H and
		methine-H), 7.41 (s, 2H, NH ₂), 10.56 (s, 1H, OH).
5t	3379, 3147 (NH₂), 2221 (C≡N).	$3.78 (s, 3H, OCH_3), 4.09 (s, 1H.$
	1720 (C=O; thiazolinone),	pyridine-H), 5.55 (s, 2H, NH ₂),
	1658 (C=O; amide).	6.75–7.90 (m, 8H, Ar-H and
		methine-H), 7.90 (s, 2H, NH ₂),
		9.01, 13.00 (2br, 2H, 2OH).
		(Continued on next page)

 TABLE III Spectral Data of the Synthesized Compounds (Continued)

Compound		
no.	IR/. v_{max} (cm ⁻¹)	1 H NMR(δ /ppm)(DMSO- d_{6})
5v	3463, 3348 (NH ₂), 2214 (C≡N), 1689 (C=O; thiazolinone), 1651 (C=O; amide).	3.76 (s, 3H, OCH ₃), 4.09 (s, 1H, pyridine-H), 5.55 (s, 2H, NH ₂), 6.47–7.35 (m, 9H, Ar-H and methine-H), 7.90 (s, 2H, NH ₂), 11.30 (br. 1H, OH).
7a	3376, 3318 (NH ₂), 2190 (C≡N), 1698 (C=O; thiazolinone), 1674 (C=O; amide).	3.60 (s, 3H, OCH ₃), 4.10 (s, 1H, pyridine-H), 5.62 (s, 2H, NH ₂), 6.48–7.45 (m, 9H, Ar-H and methine-H), 7.53, 8.62 (s, 2H, NH ₂).
7b	3318, 3200 (NH ₂), 2188 (C≡N), 1700 (C=O; thiazolinone), 1676 (C=O; amide).	-
9c	3358, 3176 (NH ₂), 1702 (C=O; thiazolinone), 1656 (C=O; amide).	2.30 (s, 3H, CH ₃), 3.69 (s, 3H, OCH ₃), 4.21 (s, 1H, pyridine-H), 5.67 (s, 2H, NH ₂), 6.75–7.50 (m, 9H, Ar-H and methine-H), 7.78 (s, 1H, NH), 8.90 (s, 1H, pyrimidine-H).
10	3318, 3200 (NH ₂), 1700 (C=O; thiazolinone), 1676 (C=O; amide).	3.10 (s, 3H, CH ₃), 4.32 (s, 1H, pyridine-H), 5.55 (s, 2H, NH ₂), 6.74–7.43 (m, 9H, Ar-H and methine-H), 8.54 (s, 1H, NH), 10.72 (br, 1H, OH).

TABLE III Spectral Data of the Synthesized Compounds (Continued)

under reflux for 6 h; the solid product which produced on heating was collected to give **5**.

Method (B):

A mixture of compound 2 (0.01 mole), aromatic aldehyde (0.01 mole), and malononitrile (0.01 mole) in ethanol (40 mL) was heated under reflux for 6 h; the solid product which produced on heating was collected to give 5.

5-Aryl-8-arylmethylidene-6-carbamoyl-2,4-diamino-8,9dihydro-*H*-thiazolo[3,2-a][1,8]naphthyridine-3-carbonitriles (7a,b)

General Procedure

A mixture of compound 5 (0.01 mole), malononitrile (0.01 mole) in ethanol (30 mL) was heated under reflux for 3 h. The solid product which precipitated upon heating was collected by filtration and recrystallized to give 7.

5-Aryl-8-arylmethylidene-4-oxo-3,4,8,9-tetrahydro-5*H*-thiazolo-[2',3':1,6]pyrido[2,3-d]pyrimidine-6-carbamoyl (9a–c)

General Procedure

A mixture of compound 5 (0.01 mole) and formic acid (10 mL) was heated under reflux for 24 h. The reaction mixture was concentrated in vacuo and the precipitate was collected by filtration, washed with water, and recrystallized from a proper solvent to give 9.

Formation of thiazolo[2',3':1,6]pyrido[2,3-d]pyrimidine derivative (10)

A solution of compound 5c (0.01 mole) in acetic anhydride (5 mL) was refluxed for 24 h. The solid product, thus formed after cooling, was collected and recrystallized to give 10.

REFERENCES

- U. Olthoff, K. Matthey, and B. Ditscher, *Ger (East)* 84, 850 (Cl. C07D), 05 Oct 1971, Appl. WP C07D/148314, 16 Jun 1970; *Chem. Abstr.*, 78, 72121y (1973).
- K. Sato and U. Nagai, Jpn. Kokai Tokkyo Koho JP 6222, 798 [8722, 798] (Cl. C07K7/00), 30 Jan 1987, Appl. 85/163, 306, 24 Jul 1985; Chem. Abstr., 107, 115975s (1987).
- [3] H. Meyer, F. Bossert, W. Vater, and K. Stoepel, Ger. Offen., 2, 210, 633 (Cl. C07D), 20 Sep 1973, Appl. P. 2210, 633, 06 Mar 1972; Chem. Abstr., 79, 146519d (1973).
- [4] G. A. M. El-Hag Ali, A. Khalil, A. H. A. Ahmed, and M. S. A. El-Gaby, Acta Chim. Slov., 49, 365 (2002).
- [5] A. A. El-Maghraby, G. A. M. El-Hag Ali, A. H. Ahmed, and M. S. A. El-Gaby, *Phosphorus, Sulfur, and Silicon*, 177, 293 (2002).
- [6] G. A. M. El-Hag Ali, Phosphorus, Sulfur, and Silicon, 178, 711 (2003).
- [7] M. S. A. El-Gaby, M. M. Khafagy, G. A. M. El-Hag Ali, H. A. Eyada, A. A. El-Maghraby, and M. H. Helal, *Phosphorus, Sulfur, and Silicon*, **178**, 1688 (2003).
- [8] G. L. Isidor and R. L. Mckee, J. Org. Chem., 38, 3615 (1973).
- [9] C. G. Dave and R. D. Shah, J. Heterocyclic Chem., 35, 1295 (1998).
- [10] L. P. Carrod and F. D. Grady, Antibiotic and Chemotherapy. (3rd ed.) Edinburgh: Churchill Livingstone (1972).