

Use of *N,N*-Dimethylpropyleneurea (DMPU) as Solvent in the Efficient Preparation of Enantiomerically Pure Secondary Amines¹

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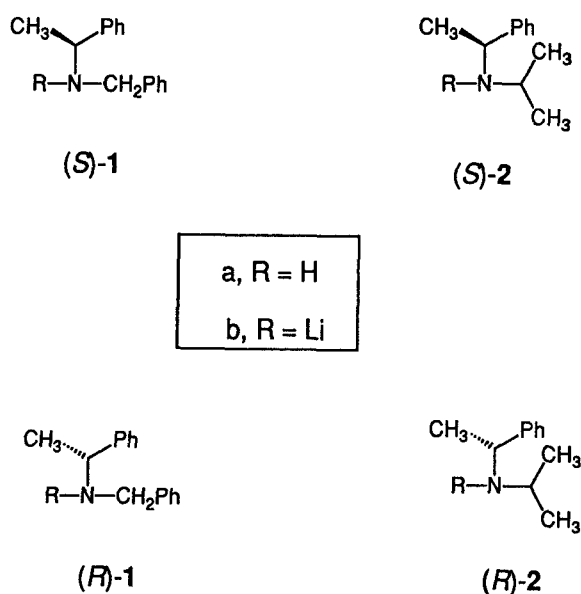
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The monoalkylation of (*R*)- and (*S*)-1-phenylethylamine proceeds readily with DMPU as the solvent. An efficient procedure was then developed for the high-yield preparation of the chiral secondary amines *N*-benzyl-, *N*-isopropyl-, *N*-(2-biphenyl)methyl-, *N*-(diphenyl)methyl-, and *N*-(2,2-dimethylpropyl)-1-phenylethylamine, in enantiomerically pure form.

The interest of organic chemists in the use of chiral lithium amides for the development of new asymmetric syntheses is growing rapidly.³ Pertinent examples are the enantioselective deprotonation of symmetric ketones,^{4–6} the diastereoselective 1,4-addition to nonchiral α,β -unsaturated esters,^{7,8} the kinetic resolution of 2-substituted cyclohexanones,⁹ and the enantiocontrolled protonation of enolates.¹⁰ Lithium organo(amido)cuprates, prepared from enantiomerically pure secondary amines, have also been found to undergo enantioselective conjugate addition to 2-cycloalkenones.¹¹ Furthermore, it has been demonstrated that achiral enolates afford enantioselective aldol additions, Michael additions to nitroolefins, carboxylation, and alkylation in the presence of chiral lithium amides or amines.¹²

Among the most useful chiral amines that have been examined in these reactions are the secondary amines **1a** and **2a**, which are readily metalated to **1b** and **2b** by means of butyllithium (Scheme 1).



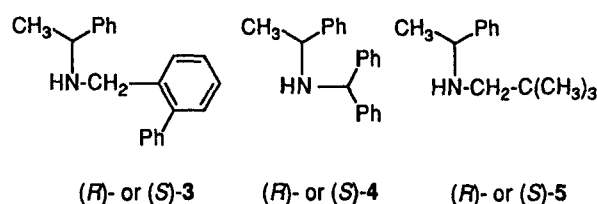
Scheme 1

The preparation of amines **1a** and **2a** was described by Simpkins et al.¹³ by reaction of (*R*)- or (*S*)-1-phenylethylamine with benzaldehyde or acetone and NaBH₃CN. This experimental procedure is fairly laborious, and careful control of the pH is required in order to prevent

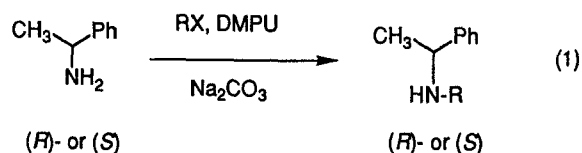
racemization. In fact, some racemization cannot be avoided in this procedure, so that crystallization of the HCl salts of the product is required. The alkylated products **1a** and **2a** are obtained in 56–64 % yield.¹³

The use of hexamethylphosphoramide (HMPA) as solvent has been recommended for the alkylation of unreactive amines,¹⁴ and several interesting applications have been described.¹⁵ On the other hand, DMPU¹ has been shown to be an excellent replacement for the carcinogenic HMPA.^{16,17}

We describe here the convenient and high-yield preparation of **1a** and **2a** by direct monoalkylation of 1-phenylethylamine in DMPU as solvent. In addition, the new chiral secondary amines **3–5** have been prepared according to the same method.



When an equimolar mixture of (*R*)- or (*S*)-1-phenylethylamine and benzyl bromide was dissolved in DMPU and heated to 100 °C for 1.5 h, the desired, enantiomerically pure monoalkylated amine (*R*)- or (*S*)-**1** was obtained in over 95 % yield (eq 1; R = CH₂Ph). This reaction is carried out in the presence of two equivalents of Na₂CO₃, in order to neutralize the HX which is initially formed.



In similar fashion, (*R*)- and (*S*)-**2a** were obtained in 91 % yield of purified product from the reaction of the corresponding 1-phenylethylamine and isopropyl iodide (eq 1; R = *i*-Pr). In this case, the reaction temperature employed for this alkylation was slightly lower: 4 h at 80 °C, plus 0.5 h at 90 °C.

Having demonstrated the utility of DMPU in the preparation of the known chiral amines **1a** and **2a**, we explored the generality of the method for the synthesis of other secondary amines, where the steric or electronic requirements would be different. Of course, such structural and inductive properties may lead to substantial differences in

Table 1. Monoalkylation of (*R*)- and (*S*)-1-Phenylethylamine in the Preparation of Enantiomerically Pure Secondary Amines 1–5.

Product	Configuration	Reaction temp. (°C) Reaction time (h)	Yield (%) ^a	[α] _D ^b (°)
1a	<i>R</i>	100/1.5	> 95	+ 54.4
	<i>S</i>	100/1.5	> 95	– 52.8
2a	<i>R</i>	80/4 + 90/0.5	91	+ 60.1
	<i>S</i>	80/4 + 90/0.5	91	– 59.7
3	<i>R</i>	100/2.5	92	+ 41.6
	<i>S</i>	100/2.5	91	– 45.7
4	<i>R</i>	130/4.5	70	+ 38.9
	<i>S</i>	130/4.5	75	– 39.9
5	<i>R</i>	120/65	32	+ 63.8
	<i>S</i>	120/65	29	– 60.7

^a After flash chromatography.^b Solvent and concentration in the experimental section.

the enantioselectivities by the chiral amine (or its chiral lithium amide) in the processes of interest (protonation, deprotonation, addition, kinetic resolution).

In the event, (*R*)- and (*S*)-1-phenylethylamine were monoalkylated with 2-(bromomethyl)biphenyl to afford (*R*)- and (*S*)-3 in 92 and 91 % yield, respectively (eq 1; R = 2-biphenylmethyl).

Furthermore, diphenylmethyl bromide as the electrophile allowed the preparation of (*R*)- and (*S*)-4 in 70 and 75 % yield, respectively, according to the general procedure (eq 1; R = CHPh₂).

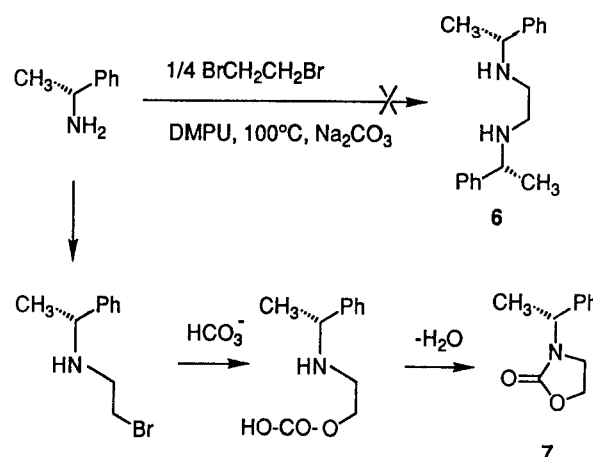
Interestingly, even neopentyl iodide reacted with (*R*)- and (*S*)-1-phenylethylamine under similar conditions (albeit in lower yield) to afford the highly hindered secondary amines (*R*)- and (*S*)-5 (eq 1; R = neopentyl) in 32 and 29 % yields, respectively.

Table 1 contains pertinent data concerning the preparation of secondary amines (*R*)- and (*S*)-1–5 from (*R*)- and (*S*)-1-phenylethylamine.

Attempted preparation of the C₂-symmetric chiral diamine **6** by treatment of 1,2-dibromoethane with four equivalents of (*R*)-1-phenylethylamine afforded, under the same conditions, the chiral oxazolidin-2-one **7** in 82 % isolated yield. This unexpected product probably arises from the reaction of the initially formed β -amino bromide and HCO₃[–] (Scheme 2). Indeed, the preparation of **6** can be carried out in the absence of Na₂CO₃.¹⁸

On the other hand, in a related study,¹⁹ the *N*-alkylation of highly hindered diamines was only possible in solvent DMPU.

In summary, the alkylation of highly hindered and other unreactive amines is facilitated by the use of DMPU as solvent. This procedure has been used for the efficient preparation of the previously described chiral secondary amines **1a** and **2a**, as well as the new, enantiomerically pure amines **3–5**. The different steric and electronic characteristics present in **1–5** render these amines, and their corresponding lithium amides, as attractive chiral

**Scheme 2**

reagents for the development of enantioselective reactions,^{3,20} and as synthetic equivalents of “chiral NH₃”.^{8,21}

DMPU was stirred at 60 °C over CaH₂ for 1 h, and then distilled at reduced pressure (bp 71–72 °C/0.1 mmHg) before use.

TLC: Merck-DC-F₂₅₄ plates; detection by UV light. Flash column chromatography:²² Fluka silica gel (40–63 μ m), and a pressure of 0.2–0.6 bar.

Melting points: Büchi-510; in open capillary tubes. Optical rotations, [α]_D: Perkin-Elmer 241, 1 dm cells; concentration *c* in g/100 mL. ¹H NMR spectra: Varian-Gemini-200, in CDCl₃; chemical shifts (δ) are given in parts per million downfield from the internal TMS reference. MS: Hitachi-Perkin-Elmer RMU-6M instrument; *m/z* values with relative intensity in parenthesis. IR spectra: Perkin-Elmer 983.

Microanalyses were performed by the microanalytical laboratories at ETH-Zürich. All compounds **3–5** and **7** gave C, H, N \pm 0.3 %, except (*R*)-**7**, C – 0.40, N + 0.41.

Alkylation of 1-Phenylethylamine in DMPU;²³ General Procedure: (*R*)- or (*S*)-1-Phenylethylamine was dissolved under argon in DMPU, and treated with two equivalents of Na₂CO₃. To the resulting white suspension was then added the alkylating agent (1–1.2 equiv of alkyl halide; dropwise addition), and the reaction mixture was heated until no starting amine was observable by TLC. The white suspension was then cooled down in an ice bath, and treated with water until a solution was formed, which was extracted with three portions of Et₂O. The combined organic phases were washed three times with water, dried over anhyd. MgSO₄ and concentrated in a rotary evaporator. Final purification was achieved by flash chromatography or by distillation on a Kugelrohr apparatus.

(*R*)-*N*-Benzyl-1-phenylethylamine [(*R*)-1a]:

(*R*)-1-Phenylethylamine (1.21 g, 10 mmol) in 40 mL of DMPU was treated with 2.12 g (20 mmol) of Na₂CO₃ and 1.2 mL (10 mmol) of benzyl bromide, and heated to 100 °C for 1.5 h, according to the general procedure. Workup (30 mL of H₂O, 3 \times 50 mL of Et₂O, 50 mL of H₂O) and flash chromatography (hexane/EtOAc, 1:1) afforded 2.0 g (9.5 mmol, 95 % yield) of (*R*)-**1a** as a colorless oil with bp 135 °C/0.15 mmHg (lit.¹¹ bp 128–130 °C/0.2 mmHg) [α]_D = + 54.5° (*c* = 0.5, EtOH) [lit.¹¹ [α]_D = + 54.4° (*c* = 3.86, EtOH)].

¹H NMR: δ = 1.37 (d, *J* = 12.5 Hz, CH₃), 1.62 (s, NH), 3.59 and 3.68 (AB, *J* = 12.5 Hz, CH₂), 3.82 [q, *J* = 12.5 Hz, H-C(1)], 7.18–7.45 (m, 5 arom H). (*S*)-**1a**. Quantitative yield; [α]_D = – 52.8° (*c* = 0.60, EtOH) {Lit.¹³ [α]_D = – 53.6° (*c* = 3.80, EtOH)}.

(*R*)-*N*-Isopropyl-1-phenylethylamine [(*R*)-2a]:

(*R*)-1-Phenylethylamine (1.21 g, 10 mmol) in 25 mL of DMPU was treated with 2.12 g (20 mmol) of Na₂CO₃ and 1.2 mL (12 mmol) of

isopropyl iodide, and heated to 80 °C for 4 h plus an additional 0.5 h at 90 °C, according to the general procedure. Workup (50 mL of H₂O, 3 × 50 mL of Et₂O, and 3 × 50 mL of H₂O) and Kugelrohr distillation, bp 50–55 °C/0.1 mmHg (lit.¹³ bp 60–66 °C/0.5 mmHg) afforded 1.48 g (91 % yield) of (*R*)-**2a** as a colorless oil. $[\alpha]_D^{25} = +60.1^\circ$ ($c = 0.71$, CHCl₃) {lit.¹³ $[\alpha]_D^{25} = +61.4^\circ$ ($c = 2.23$, CHCl₃)}.

IR (film): $\nu = 3320$ (w), 2960 (s), 1945 (w), 1874 (w), 1806 (w), 1735 (w), 1602 (m), 1492 (s), 1451 (s), 1368 (s), 1169 (s), 749 (s), 687 (s).

¹H NMR: $\delta = 0.99$ (d, $J = 6.0$ Hz, CH₃-C(1')), 1.03 [d, $J = 6.0$ Hz, CH₃-C(1')], 1.25–1.40 (bs, NH), 1.34 [d, $J = 6.0$ Hz, CH₃-C(1')], 2.62 [m, H-C(1')], 3.89 [q, $J = 6.0$ Hz, H-C(1')], 7.18–7.40 (m, 5 arom H).

¹³C NMR: $\delta = 21.91$, 23.80, 24.60, 45.21, 54.77, 126.14, 126.44, 128.11, 145.82.

(*S*)-**2a**: Yield = 91 %; $[\alpha]_D^{25} = -59.7^\circ$ ($c = 1.01$, CHCl₃) {lit.¹³ $[\alpha]_D^{25} = +61.4^\circ$ ($c = 2.23$, CHCl₃)} for the (*R*)-enantiomer.

(*R*)-*N*-(2-Biphenyl)methyl-1-phenylethylamine [(*R*)-**3**]:

According to the general procedure, 1.21 g (10 mmol) of (*R*)-1-phenylethylamine in 20 mL of DMPU was treated with 2.12 g (20 mmol) of Na₂CO₃ and 1.9 mL (10 mmol) of 2-(bromomethyl)biphenyl, and heated to 100 °C for 2.5 h. Workup (50 mL of H₂O, 3 × 50 mL of Et₂O, 3 × 50 mL of H₂O) and flash chromatography (hexane/EtOAc, 4:1) afforded 1.33 g (92 % yield) of (*R*)-**3** as a colorless oil with bp 200–210 °C/0.15 mmHg. $[\alpha]_D^{25} = +41.6^\circ$ ($c = 0.76$, CHCl₃).

IR (film): $\nu = 3323$ (w), 3023 (w), 1953 (w), 1877 (w), 1811 (w), 1599 (w), 1478 (m), 1450 (m), 1010 (w), 751 (s), 701 (s).

¹H NMR: $\delta = 1.25$ (d, $J = 6.0$ Hz, CH₃), 1.48 (s, NH), 3.56 and 3.70 (AB, $J = 12.5$ Hz, CH₂), 3.66 [q, $J = 6.0$ Hz, H-C(1')], 7.16–7.50 (m, 14 arom H).

¹³C NMR: $\delta = 24.13$, 49.40, 57.49, 126.37, 126.51, 126.67, 126.76, 127.21, 127.92, 128.09, 128.77, 129.41, 129.88, 137.61, 140.98, 141.57, 145.18.

MS: $m/z = 287$ (M⁺, 5), 272 (100), 210 (3), 182 (18), 167 (99), 152 (18), 136 (4), 128 (2), 120 (5), 105 (24), 77 (10), 51 (4), 39 (2), 28 (22).

(*S*)-**3**: Yield: 91 %; $[\alpha]_D^{25} = -45.7^\circ$ ($c = 0.82$, CHCl₃).

(*R*)-*N*-Diphenylmethylamine [(*R*)-**4**]:

According to the general procedure, 0.61 g (5 mmol) of (*R*)-1-phenylethylamine in 15 mL of DMPU was treated with 1.06 g (10 mmol) of Na₂CO₃ and 1.54 g (6 mmol) of diphenylmethyl bromide, and heated to 120–130 °C for 4.5 h. Workup (50 mL of H₂O, 3 × 50 mL of Et₂O, 3 × 50 mL of H₂O) and flash chromatography (hexane/EtOAc, 3:1) afforded 1.0 g (70 % yield) of (*R*)-**4** as a colorless oil with bp 200–210 °C/0.1 mmHg. $R_f = 0.46$ (hexane/EtOAc, 3:1); $[\alpha]_D^{25} = +38.5^\circ$ ($c = 0.74$, CHCl₃).

IR (film): $\nu = 3326$ (w), 3025 (s), 2961 (m), 2863 (w), 1950 (w), 1881 (w), 1808 (w), 1754 (w), 1599 (m), 1492 (s), 1451 (s), 1028 (m), 763 (s), 745 (s).

¹H NMR: $\delta = 1.36$ (d, $J = 6.0$ Hz, CH₃), 1.68 (bs, NH), 3.69 [q, $J = 6.0$ Hz, H-C(1')], 4.63 [s, H-C(1')], 7.12–7.49 (m, 15 arom H).

¹³C NMR: $\delta = 24.25$, 54.99, 63.57, 126.49, 126.65, 126.71, 127.17, 127.47, 128.15, 128.27, 143.45, 144.44, 145.38.

MS: $m/z = 287$ (M⁺, 2), 272 (35), 210 (14), 182 (26), 167 (100), 152 (11), 134 (3), 120 (5), 105 (45), 77 (13), 51 (5), 28 (23).

(*S*)-**4**: Yield: 75 %; $[\alpha]_D^{25} = 39.9^\circ$ ($c = 0.7$, CHCl₃).

(*R*)-*N*-(2,2-Dimethyl)propyl-1-phenylethylamine [(*R*)-**5**]:

According to the general procedure, 0.61 g (5 mmol) of (*R*)-1-phenylethylamine in 15 mL of DMPU was treated with 1.06 g (10 mmol) of Na₂CO₃ and 1.0 g (7.5 mmol) of (2,2-dimethyl)propyl iodide, and heated to 100 °C for 65 h. Workup (50 mL of H₂O, 3 × 50 mL of Et₂O, and 3 × 50 mL of H₂O), and flash chromatography (hexane/EtOAc, 1:1) afforded 0.31 g (32 % yield) of (*R*)-**5** as a colorless oil with bp 50–60 °C/0.1 mmHg. $R_f = 0.74$ (hexane/EtOAc, 1:1); $[\alpha]_D^{25} = +63.8^\circ$ ($c = 0.58$, CHCl₃).

IR (film): $\nu = 3300$ (w), 3025 (w), 2952 (s), 1945 (w), 1875 (w), 1807 (w), 1737 (w), 1689 (m), 1603 (m), 1452 (m), 1362 (m), 1126 (m), 761 (s).

¹H NMR: $\delta = 0.90$ (s, *t*-Bu), 1.34 [d, $J = 6.0$ Hz, CH₃-C(1')] 2.14 and 2.28 (AB, $J = 11.0$ Hz, CH₂), 3.70 [q, $J = 6.0$ Hz, H-C(1')], 7.15–7.35 (m, 5 arom H).

¹³C NMR: $\delta = 24.64$, 27.49, 31.07, 58.63, 59.82, 126.31, 127.97, 145.80.

MS: $m/z = 191$ (M⁺, 2), 176 (5), 134 (44), 118 (3), 105 (100), 91 (5), 77 (17), 72 (7), 56 (3), 51 (6), 41 (11), 30 (34).

(*S*)-**5**: Yield: 29 %; bp 60–70 °C/0.15 mmHg; $[\alpha]_D^{25} = -60.7^\circ$ ($c = 0.75$, CHCl₃).

3*N*-(*R*)-1-Phenylethylloxazolidin-2-one [(*R*)-**7**]:

According to the general procedure, 1.45 g (12 mmol) of (*R*)-1-phenylethylamine in 10 mL of DMPU was treated with 2.12 g (20 mmol) of Na₂CO₃ and 0.26 mL (3 mmol) of ethylene bromide, and heated to 100 °C for 36 h. The usual workup procedure and flash chromatography (hexane/EtOAc 1:1) afforded 0.36 g (82 % yield) of (*R*)-**7** as a yellowish oil with bp 200–210 °C/0.1 mmHg. $R_f = 0.29$ (hexane/EtOAc, 1:1); $[\alpha]_D^{25} = 95.6^\circ$ ($c = 0.66$, CHCl₃).

IR (CHCl₃): $\nu = 3008$ (m), 2922 (w), 1742 (s), 1603 (w), 1493 (m), 1483 (s), 1454 (m), 1424 (s), 1120 (w), 1067 (m), 1043 (m), 1020 (m), 639 (w).

¹H NMR: $\delta = 1.59$ [d, $J = 6.0$ Hz, CH₃-C(1')], 3.15 (distorted ddd, 2 H), 3.50 (distorted ddd, 2 H), 4.22 (distorted ddd, 2 H), 4.30 (distorted ddd, 2 H), 5.23 [q, $J = 6.0$ Hz, H-C(1')], 7.23–7.43 (m, 10 arom H).

¹³C NMR: $\delta = 1605$, 39.77, 51.18, 61.63, 126.75, 127.59, 128.41, 139.21, 157.9.

MS: $m/z = 191$ (M⁺, 97), 176 (80), 163 (8), 146 (5), 132 (30), 114 (7), 105 (100), 91 (8), 77 (24), 70 (9), 51 (9), 42 (5), 28 (6).

HRMS: m/z (M⁺) Found 191.0931. Calc. for C₁₁H₁₃NO₂, 191.0946.

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- (1) DMPU = *N,N'*-dimethylpropyleneurea = 1,3-dimethyl-2-oxo-hexahydropyrimidine.
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- (3) For a review, see: Cox, P. J.; Simpkins, N. S. *Tetrahedron: Asymmetry* **1991**, 2, 1.
For an overview on asymmetric reactions between achiral substrates and chiral reagents, see: Juaristi, E. *Introduction to Stereochemistry and Conformational Analysis*; Wiley: New York, 1991; Ch. 12.
- (4) Shirai, R.; Tanaka, M.; Koga, K. *J. Am. Chem. Soc.* **1986**, 108, 543.
See, also: Koga, K. *Enantioselective Reactions Using Chiral Chelated Lithium Amides*; In *Organic Synthesis in Japan – Past, Present, and Future*; Noyori, Ed.; Tokyo Kagaku Dozin 1992; pp 195–203.
- (5) Simpkins, N. S. *J. Chem. Soc., Chem. Commun.* **1986**, 88.
Bambridge, K.; Simpkins, N. S.; Clark, B. P. *Tetrahedron Lett.* **1992**, 33, 8141.
- (6) See, also: Majewski, M.; Gleave, D. M. *J. Org. Chem.* **1992**, 57, 3599.
Wild, H.; Born, L. *Angew. Chem., Int. Ed. Engl.* **1991**, 30, 1685.
- (7) Hawkins, J. M.; Fu, G. C. *J. Org. Chem.* **1986**, 51, 2820.
Hawkins, J. M.; Lewis, T. A. *J. Org. Chem.* **1992**, 57, 2114.
- (8) Davies, S. G.; Ichihara, O. *Tetrahedron: Asymmetry* **1991**, 2, 183.
- (9) Kim, H. D.; Kawasaki, H.; Nakajima, M.; Koga, K. *Tetrahedron Lett.* **1989**, 30, 6537.
- (10) Vedejs, E.; Lee, N. *J. Am. Chem. Soc.* **1991**, 113, 5483.
- (11) Rossiter, B. E.; Eguchi, M.; Miao, G.; Swingle, N. M.; Hernández, A. E.; Vickers, D.; Fluckiger, E.; Patterson, R. G.; Reddy, K. V. *Tetrahedron* **1993**, 49, 965.

- (12) Seebach, D. *Proc. Robert A. Welch Found. Conf. Chem. Res.* 27: *Stereospecificity in Chemistry and Biochemistry* (November 7–9, 1983), Houston, TX 1984; p. 93–145.
Seebach, D. *Angew. Chem., Int. Ed. Engl.* **1988**, 27, 1624.
Hansen, J. *Doctoral Dissertation No. 7863*: ETH-Zürich 1985.
Mukhopadhyay, T. Unpublished results.
Juaristi, E.; Beck, A. K.; Hansen, J.; Matt, T.; Mukhopadhyay, T.; Simson, M.; Seebach, D. *Synthesis* **1993**, 1271.
Yasukata, T.; Koga, K. *Tetrahedron: Asymmetry* **1993**, 4, 35.
- (13) Cain, C. M.; Cousins, R. P. C.; Coumbarides, G.; Simpkins, N. S. *Tetrahedron* **1990**, 46, 523.
- (14) Juaristi, E.; Reyna, d. *Rev. Soc. Qhim. Mex.* **1982**, 26, 295.
Albright, J. D.; DeVries, V. G.; Largis, E. E.; Miner, T. G.; Reich, M. F.; Schaffer, S. A.; Shepherd, R. G.; Upeslakis, J. *J. Med. Chem.* **1983**, 26, 1378.
Juaristi, E.; Reyna, D. *Tetrahedron Lett.* **1984**, 25, 3521.
Juaristi, E.; Madrigal, D. *Tetrahedron* **1989**, 45, 629.
- (15) See, for example: Satoh, T.; Yaneko, Y.; Sakata, K.; Yamakawa, K. *Chem. Lett.* **1985**, 585.
Jackman, L. M.; Scarmoutzos, L. M. *J. Am. Chem. Soc.* **1987**, 109, 5348.
Fouquey, C.; Lehn, J.-M.; Malthete, J. *J. Chem. Soc., Chem. Commun.* **1987**, 1424.
- Hocart, C. H.; Letham, D. S.; Parker, C. W. *Phytochem.* **1991**, 30, 2477.
- (16) Mukhopadhyay, T.; Seebach, D. *Helv. Chem. Acta* **1982**, 65, 385.
Seebach, D. *Chem. Br.* **1985**, 21, 632.
- (17) DMPU has been shown to be non-mutagenic and safe: *Chimia* **1985**, 39, 147.
- (18) Boog, A. unpublished results.
Horner, L.; Dickerhof, K. *Liebigs Ann. Chem.* **1984**, 1240.
- (19) Seebach, D.; Hayakawa, M.; Sakaki, J.-i.; Schweizer, W. B. *Tetrahedron* **1993**, 49, 1711.
- (20) See Table 9 in: Seebach, D. *Angew. Chem., Int. Ed. Engl.* **1988**, 27, 1624.
- (21) Estermann, H.; Seebach, D. *Helv. Chim. Acta* **1988**, 71, 1824.
Juaristi, E.; Escalante, J.; Lamatsch, B.; Seebach, D. *J. Org. Chem.* **1992**, 57, 2396.
- (22) Still, W. C.; Kahn, M.; Mitra, A. *J. Org. Chem.* **1978**, 43, 2923.
- (23) DMPU could be recovered in 60–70 % yield by distillation. We are grateful to Mr. A. Beck for suggesting this possibility to us. Of course, recovery of the DMPU is important in large-scale preparations.