SYNTHESIS OF SOME OLIGOSACCHARIDES CONTAINING THE *O*- α -L-FUCOPYRANOSYL-(1 \rightarrow 2)-D-GALACTOPYRANOSYL UNIT*

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ABSTRACT

A practical approach for the synthesis of oligosaccharides containing the O- α -L-fucopyranosyl-(1 \rightarrow 2)-D-galactopyranosyl unit has been developed by utilizing the readily accessible 3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- α -D-galactopyranosyl bromide (1). Thus, condensation of bromide 1 with 8-(methoxycarbonyl)octanol in benzene, in the presence of mercuric cyanide, afforded, after column-chromatographic separation, the anomers (3 and 4) of 8-(methoxycarbonyl)octyl 3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)-D-galactopyranoside in the ratio of 13:4. On the basis of their specific rotations, compounds 3 and 4 were tentatively assigned the α and β configuration, respectively. Similarly, bromide 1 was allowed to react with some other, appropriately protected sugars having a free alcohol group, to afford the corresponding α and β anomers of a galactopyranosyl residue having an α -L-fucopyranosyl substituent at O-2. Column-chromatographic separation of the anomeric mixtures, followed by systematic removal of the protecting groups, then provided the final oligosaccharides desired. In this manner, the synthesis of the following oligosaccharides has been accomplished: α -L-Fuc-(1 \rightarrow 2)- α -D-Gal-1 \rightarrow O(CH₂)₈CO₂Me (5), α -L-Fuc-(1 \rightarrow 2)- β -D-Gal-1 \rightarrow O(CH₂)₈CO₂Me (6), α -L-Fuc-(1 \rightarrow 2)- α -D-Gal- $(1\rightarrow 3)$ - β -D-GlcNAc-1 \rightarrow OC₆H₄NO₂-p (11), α -L-Fuc-(1 \rightarrow 2)- β -D-Gal-(1 \rightarrow 3)- β -D-GlcNAc-1 \rightarrow OC₆H₄NO₂-p (13). α -L-Fuc-(1 \rightarrow 2)- α -D-Gal-(1 \rightarrow 3)- α -D-GalNAc-1 \rightarrow OC₆H₄NO₂-o, and α -L-Fuc-(1 \rightarrow 2)- β -D-Gal-(1 \rightarrow 3)- α -D-GalNAc- $1 \rightarrow OC_6H_4NO_2$ -o (18). The structures assigned to compounds 5, 6, 11, 13, and 18 were all supported by their respective ¹³C-n.m.r. spectra.

INTRODUCTION

The disaccharide unit α -L-Fuc-(1 \rightarrow 2)- β -D-Gal has been found to occur as part of the carbohydrate moiety of various glycoconjugates, particularly the blood-

^{*}Synthetic Studies in Carbohydrates, Part XXXVII. For Part XXXVI, see ref. 1.

Antigenic determinant	Structure ^b
Lewis-d (H type 1)	α -Fuc-(1 \rightarrow 2)- β -Gal-(1 \rightarrow 3)- β -GlcNAc-
Lewis-b	α -Fuc-(1 \rightarrow 2)- β -Gal-(1 \rightarrow 3)-[α -Fuc- (1 \rightarrow 4)]- β -GlcNAc-
H type 2	α -Fuc-(1 \rightarrow 2)- β -Gal-(1 \rightarrow 4)- β -GlcNAc-
Y	α -Fuc- $(1\rightarrow 2)$ - β -Gal- $(1\rightarrow 4)$ - $[\alpha$ -Fuc- $(1\rightarrow 3)$]- β -GlcNAc-
AH type 2	β-GalÑAc-(1→3)-α-Fuc-(1→2)]-β- Gal-(1→4)]-β-GlcNAc-

TABLE I

STRUCTURE OF	SOME HEMANI	RLOOD-GROUP	DETERMINANTS ^a
anocionion	JOHD HOMAN	nroon-ouooi	DETERMINANTS

"Taken from ref. 2. ^bExcept for L-fucose, all of the sugars are members of the D series.

group-specific glycoproteins and glycolipids². The structures of some human, blood-group determinants containing this disaccharide unit are given in Table I.

Recent progress in the field of carbohydrate chemistry provided various approaches for the chemical synthesis of the carbohydrate units of such blood-group antigens^{3,4}. However, it is always tempting, and often desirable, to improve on the existing methods with the aim of rendering them simpler and more amenable. With this in mind, we now describe an alternative approach for the synthesis of some oligosaccharides containing the α -L-Fuc-(1 \rightarrow 2)-D-Gal unit.

RESULTS AND DISCUSSION

Since it was first announced, almost two decades ago^5 , halide-ion-catalyzed glycosylation has been one of the most extensively utilized methods for the synthesis of α -linked fucopyranosides. However, the preparation of the fucosylating agent (*i.e.*, 2,3,4-tri-O-benzyl- α -L-fucopyranosyl bromide) from the commercially available L-fucose involves a lengthy, multistep procedure. An alternative method, advocated by Flowers and Dejter-Juszynski⁶ also suffers from a similar shortcoming, as it utilizes 2-O-benzyl-3,4-di-O-(p-nitrobenzoyl)- α -L-fucopyranosyl bromide, the preparation of which is far from simple.

In their first synthesis of the disaccharide α -L-Fuc-(1 \rightarrow 2)-Gal, Levy and coworkers⁷ condensed 2,3,4-tri-O-acetyl- α -L-fucopyranosyl bromide with benzyl 6-O-benzoyl-3,4-O-isopropylidene- β -D-galactopyranoside. This method appears to be attractive, in so far as the preparation of the glycosylating agent (*i.e.*, the acetylated fucosyl bromide) from L-fucose requires only two, relatively good-yielding, steps. In our own experience⁸, however, use of the readily accessible 1,3,4,6-tetra-O-acetyl- α -D-galactopyranose⁹, instead of benzyl 6-O-benzoyl-3,4-O-isopropylidene- β -D-galactopyranoside, under otherwise similar reaction-conditions, provided a yet even simpler (and straightforward) synthesis of the same disaccharide.

Also, we previously described the synthesis of 3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- α -D-galactopyranosyl bromide¹⁰ (1), and, by syn-

thesis of p-nitrophenyl 2-O- α -L-fucopyranosyl- β -D-galactopyranoside, demonstrated that it can be utilized as a glycosyl donor. We have now extended the use of bromide 1 for the synthesis of a variety of other oligosaccharides containing the α -L-Fuc-(1 \rightarrow 2)-Gal unit.

On condensation of bromide 1 with 8-(methoxycarbonyl)octanol (2) in benzene, in the presence of mercuric cyanide, examination of the reaction mixture by thin-layer chromatography (t.l.c.) revealed the presence of a major, slower-migrating, and a minor, faster-migrating product. Column-chromatographic separation of the mixture afforded compounds 3 and 4 in the ratio of 13:4. On the basis of their specific rotations, compounds 3 and 4 were tentatively designated as the α and β anomer, respectively; this was supported by the ¹³C-n.m.r. spectra of their Odeacetylated derivatives. Thus, O-deacetylation of 3 in methanol with Amberlyst A-26 (OH⁻) anion-exchange resin¹¹ furnished, quantitatively, 8-(methoxycarbonyl)octyl 2-O- α -L-fucopyranosyl- α -D-galactopyranoside (5), the ¹³C-n.m.r. spectrum of which was in accord with the structure assigned; the signals for C-1 (98.86 p.p.m.) and C-1' (102.59 p.p.m.) were in agreement with an α and a β configuration, respectively, at the two glycosidic linkages. Similarly, O-deacetylation of 4 afforded 8-(methoxycarbonyl)octyl 2- $O-\alpha$ -L-fucopyranosyl- β -D-galactopyranoside (6), the ¹³C-n.m.r. spectrum of which was, also, in agreement with the structure assigned; see Table II. It is noteworthy that such compounds as 6 are complete hap-



tens, as they possess the functional group (CO_2Me) that had been successfully employed by Lemieux *et al.*¹² for the preparation of artificial oligosaccharide antigens having a nine-carbon-atom linking-arm.

Condensation of bromide 1 with *p*-nitrophenyl 2-acetamido-2-deoxy-4,6-O-(*p*-methoxybenzylidene)- α -D-glucopyranoside (7) in 1:1 benzene-nitromethane, in the presence of mercuric cyanide, also provided an anomeric mixture that was separated by repeated column-chromatography on silica gel, to afford compounds 8 and 9. Cleavage of the acetal group of 8 and 9 in warm, 70% aqueous acetic acid gave compounds 10 and 12, respectively; these were separately purified by silica gel

Atoms	Compound			
	6	13	18	
C-1	102.45	98.83	96.39	
C-2	77.72	53.53	96.39	
C-3	74.82	79.51	75.37	
C-4	70.59	68.03	67.42	
C-5	75.89	76.39	72.46	
C-6	61.81	60.15	60.10	
NHCOCH ₃		22.79	22.17	
C=O	_	168.88	169.81	
OCH ₃	53.03	_		
C-1'	100.39	99.66	101.60	
C-2'	69.35	76.91	79.64	
C-3′	71.38	71.31	71.28	
C-4′	72.92	67.79	67.42	
C-5′	67.70	75.45	74.92	
C-6'	16.57	60.15	60.10	
C-1"		100.49	100.74	
C-2"		68.60	68.96	
C-3"		69.46	69.64	
C-4"		73.74	72.88	
C-5″		65.87	66.07	
C-6″		16.29	16.20	

TABLE II

¹³C-N.M.R. CHEMICAL SHIFTS^a

^aSolvent was Me₂SO- d_6 , except for D₂O for 6. The reference standard (Me₄Si) was internal for solutions in Me₂SO- d_6 , and external for solutions in D₂O.

chromatography and then deacetylated, to afford, respectively, crystalline trisaccharide α -L-fuc- $(1\rightarrow 2)$ - α -D-Gal- $(1\rightarrow 3)$ - β -D-GlcNAc-1 \rightarrow OC₆H₄NO₂-p (11) and amorphous α -L-fuc- $(1\rightarrow 2)$ - β -D-Gal- $(1\rightarrow 3)$ - β -D-GlcNAc-1 \rightarrow OC₆H₄NO₂-p (13).

Condensation of bromide 1 with *o*-nitrophenyl 2-acetamido-2-deoxy-4,6-O-(*p*-methoxybenzylidene)- α -D-galactopyranoside (14) under conditions similar to those already described also gave a mixture of anomers that was separated, and analogously processed, to furnish the final oligosaccharides 16 and 18.

It is noteworthy that our synthetic disaccharide β -Gal-(1 \rightarrow 3)- α -GalNAc-1 \rightarrow OC₆H₄NO₂-p acted as a chromogenic substrate for an endo- α -N-acetylgalactosaminidase¹³. It would thus be reasonable to expect compound **18** to act, also, as



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a chromogenic substrate for such endoenzymes. Moreover, compound 18 may also serve as a reference compound for the modification of an assay procedure for α -(1 \rightarrow 2)-L-fucosyltransferase, when the disaccharide β -Gal-(1 \rightarrow 3)- α -GalNAc-1 \rightarrow OC₆H₄NO₂-o is used as the acceptor¹⁴.

After reduction of the nitro group, and subsequent attachment to a solid support, oligosaccharides O-glycosylically linked to p-nitrophenol have frequently been utilized for the purification of lectins¹⁵, and also as synthetic antigens^{16,17}. It would seem plausible to designate compound **13** "a potential blood-group Lewis-d antigen", as the carbohydrate sequence α -Fuc-(1 \rightarrow 2)- β -Gal-(1 \rightarrow 3)- β -GlcNAc-1 \rightarrow (see Table I) occurs as part of the blood-group Lewis-d (H type 1) antigen.

EXPERIMENTAL

General methods. — These were the same as those already described¹, except that the following solvent systems (v/v) were used for chromatography: A, 1:1 ethyl acetate-hexane; B, 3:1 chloroform-methanol; C, 2:1 chloroform-acetone; D, 4:1 chloroform-acetone; E, 6:1 chloroform-acetone; F, 9:1 chloroform-acetone; acetone; G, 15:1 chloroform-acetone; and H, 19:1 chloroform-acetone.

Condensation of 3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- α -D-galactopyranosyl bromide (1) with 8-(methoxycarbonyl)octanol (2). — A mixture of bromide 1 (1.1 g), mercuric cyanide (0.7 g), and 8-(methoxycarbonyl)octanol (2; 0.37 g) in dry benzene (50 mL) was stirred for two days at room temperature. T.l.c. (solvent A) then revealed the presence of a major, slower-migrating, and a minor, faster-migrating product. The solution was successively washed with aqueous sodium hydrogenearbonate and water, dried, and evaporated to a thick syrup which was dissolved in a small volume of ethyl acetate, and the solution applied to a column of silica gel. On elution with solvent A, evaporation of the fractions corresponding to the minor product afforded syrupy 8-(methoxycarbonyl)octyl 3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- β -D-galactopyranoside* (4) (0.2 g, 13%); $[\alpha]_D$ -88.8° (c 2.6, chloroform).

Continued elution with the same solvent-system, and evaporation of the fractions corresponding to the major product, afforded the α -D-galactopyranosyl* anomer 3 (0.65 g, 43%) as a syrup; $[\alpha]_D - 11.4^\circ$ (c 1.0, chloroform).

8-(Methoxycarbonyl)octyl 2-O-α-L-fucopyranosyl-α-D-galactopyranoside (5). — A solution of 3 (0.5 g) in methanol (20 mL), containing a catalytic amount of Amberlyst A-26 (OH⁻) anion-exchange resin¹¹, was stirred overnight at room temperature. The resin was then filtered off and washed with methanol, and the filtrate and washings were combined and evaporated, to furnish compound 5 (0.3 g, 90%), amorphous; $[\alpha]_D$ +1.1° (c 1.3, methanol); t.l.c. (solvent B): R_F 0.62; ¹H-n.m.r. data (D₂O): δ 1.60–2.20 [15 H, (CH₂)₆ + CMe], 2.84 (t, 2 H, J 7 Hz, CH₂CO), 4.16 (s, 3 H, OMe), and 5.52 (2 H, H-1,1'); ¹³C-n.m.r. data (D₂O): δ 16.67 (C-6'), 53.03 (OCH₃), 62.01 (C-6), 79.08 (C-2), 98.86 (C-1), and 102.59 (C-1').

Anal. Calc. for C₂₂H₄₀O₁₂: C, 53.21; H, 8.12. Found: C, 52.93; H, 8.31.

8-(Methoxycarbonyl)octyl 2-O-α-L-fucopyranosyl-β-D-galactopyranoside (6). — O-Deacetylation of 4 (90 mg), as described for 3, to give 5, furnished amorphous 6 (30 mg, 50%); $[\alpha]_D$ -81.5° (c 0.8, methanol); t.l.c. (solvent B): R_F 0.48; ¹Hn.m.r. data (D₂O): δ 1.60–2.20 [m, 15 H, (CH₂)₆ + CMe], 2.84 (t, 2 H, J 6.5 Hz, CH₂CO), 4.16 (s, 3 H, OMe), 4.93 (d, 1 H, $J_{1,2}$ 7 Hz, H-1), and 5.72 (1 H, H-1'); for ¹³C-n.m.r. data, see Table II.

Anal. Calc. for C₂₂H₄₀O₁₂: C, 53.21; H, 8.12. Found: C, 53.45; H, 8.38.

p-Nitrophenyl 2-acetamido-2-deoxy-4,6-O-(p-methoxybenzylidene)-3-O-[3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- α -D-galactopyranosyl]- β -D-glucopyranoside (8) and p-nitrophenyl 2-acetamido-2-deoxy-4,6-O-(pmethoxybenzylidene)-3-O-[3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- β -D-galactopyranosyl]- β -D-glucopyranoside (9). — A solution of compound 7 (0.92 g, 2 mmol) in 1:1 (v/v) benzene-nitromethane (100 mL) was boiled until 20 mL of the solvent had distilled off. The temperature was then adjusted to ~55° (bath), mercuric cyanide (0.76 g, 3 mmol) and a solution of bromide 1 (1.92 g, 3 mmol) in 1:1 nitromethane-benzene (20 mL) were added, and the mixture was stirred for 2 days, allowed to cool to room temperature, diluted with benzene (50 mL), successively washed with aqueous sodium hydrogencarbonate, and water, dried, and evaporated; the crude mixture was subjected to column chromatog-

^{*}The assignments of these configurations are only tentative; but, see later.

raphy on silica gel. Elution with solvent *H*, and evaporation of the fractions corresponding to the major product gave the α anomer 8 (1.1 g). Further elution, with solvent *G*, afforded the minor, slower-migrating β anomer 9 (0.3 g). Both 8 and 9 were characterized after cleavage of their *p*-methoxybenzylidene group; see later.

p-Nitrophenyl 2-acetamido-2-deoxy-3-O-[3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-Oacetyl- α -L-fucopyranosyl)- α -D-galactopyranosyl]- β -D-glucopyranoside (10).Compound 8 (1.1 g) in 70% aqueous acetic acid (50 mL) was heated for 1 h at 55°. The acetic acid was evaporated under diminished pressure, the last traces being removed by co-evaporation with several portions of water, and then of toluene. The solid residue so obtained was applied to a column of silica gel and sequentially eluted with chloroform, with solvent F, and with solvent D. Evaporation of the fractions corresponding to the product gave 10 (0.8 g, 44% overall yield, based on)7); amorphous; $[\alpha]_D = -4.7^\circ$ (c 0.9, methanol); t.l.c. (solvent C): $R_F 0.49$; ¹H-n.m.r. data (Me₂SO-d₆): δ 1.05 (d, 1 H, J 6.5 Hz, CMe), 1.83-2.16 (cluster of singlets, 21 H, 6 OAc and NAc), 5.17 (d, 1 H, $J_{1'',2''}$ 3.5 Hz, H-1"), 5.65 (d, 1 H, $J_{1',2'}$ 3.5 Hz, H-1'), and 7.25 and 8.27 (2 m, 2 \times 2 H, aromatic); ¹³C-n.m.r. data (Me₂SO-d₆): δ 15.42 (C-"), 20.28 (OAc), 22.55 (NAc), 53.62 (C-2), 60.15 (C-6), 60.59 (C-6'), 77.00 (C-2'), 78.21 (C-3), 95.88 (C-1"), 96.95 (C-1'), 97.85 (C-1), and 169.03-169.98 (C=O).

Anal. Calc. for $C_{38}H_{50}N_2O_{23}$: C, 50.55; H, 5.58; N, 3.10. Found: C, 50.09; H, 5.63; N, 2.75.

p-Nitrophenyl 2-acetamido-2-deoxy-3-O-[3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl-α-L-fucopyranosyl)-β-D-galactopyranosyl]-β-D-glucopyranoside (12). — Cleavage of the acetal group of compound 9 (0.3 g), exactly as described for 8 (to give 10), afforded compound 12 (0.2 g, 11% overall yield, based on 7); amorphous; $[\alpha]_D - 42.6^\circ$ (c 1.2, methanol); t.l.c. (solvent C): $R_F 0.54$; ¹H-n.m.r. data (Me₂SO-d₆): δ 1.05 (d, 3 H, J 6 Hz, CMe), 1.90–2.13 (cluster of singlets, 21 H, 6 OAc and NAc), 5.10 (d, 1 H, $J_{1'',2''}$ 4 Hz, H-1''), 7.11 and 8.25 (2 m, 2 × 2 H, aromatic), and 8.09 (d, 1 H, J 9 Hz, NH); ¹³C-n.m.r. data (Me₂SO-d₆): δ 15.21 (C-6''), 20.28 (OAc), 22.50 (NAc), 53.37 (C-2), 60.23 (C-6), 60.81 (C-6'), 76.83 (C-2'), 79.56 (C-3), 95.96 (C-1), 98.60 (C-1''), 99.19 (C-1'), and 168.54–169.77 (C=O).

Anal. Calc. for $C_{38}H_{50}N_2O_{23} \cdot 0.5 H_2O$: C, 50.05; H, 5.64; N, 3.07. Found: C, 49.94; H, 5.66; N, 2.79.

p-Nitrophenyl 2-acetamido-2-deoxy-3-O-(2-O- α -L-fucopyranosyl- α -D-galactopyranosyl)- β -D-glucopyranoside (11). — A solution of compound 10 (0.6 g) in a mixture of methanol (18 mL), triethylamine (6 mL), and water (5 mL) was kept overnight at ~4°, whereupon the deacetylation product crystallized out. The crystalline material was filtered off, thoroughly washed with ethanol, and dried, to afford compound 11 (0.23 g, 53%); m.p. 262-264°, [α]_D +14.3° (c 0.5, Me₂SO).

Anal. Calc. for C₂₆H₃₈N₂O₁₇: C, 48.00; H, 5.89; N, 4.31. Found: C, 48.01; H, 6.00; N, 4.13.

p-Nitrophenyl 2-acetamido-2-deoxy-3-O-(2-O- α -L-fucopyranosyl- β -D-galactopyranosyl)- β -D-glucopyranoside (13). — O-Deacetylation of 12 (0.15 g), as described for 10 (to give 11), furnished trisaccharide 13 (0.1 g, 92%); amorphous; $[\alpha]_D -52.5^\circ$ (c 0.5, Me₂SO); ¹H-n.m.r. data (Me₂SO-d₆): δ 1.09 (d, 3 H, J 6.5 Hz, CMe), 1.90 (s, 3 H, NAc), 5.14 (d, 1 H, J_{1,2} 7 Hz, H-1), 7.18 and 8.25 (2 m, 2 × 2 H, aromatic), and 7.88 (d, 1 H, J 9 Hz, NH); for ¹³C-n.m.r. data, see Table II.

Anal. Calc. for $C_{26}H_{38}N_2O_{17} \cdot 0.5 H_2O$: C, 47.34; H, 5.96; N, 4.25. Found: C, 47.12; H, 6.27; N, 4.31.

o-Nitrophenyl 2-acetamido-2-deoxy-3-O-[3,4,6-tri-O-acetyl-2-O-(2,3,4-tri-O-acetyl- α -L-fucopyranosyl)- α -D-(15) and - β -D-(17)-galactopyranosyl]- α -D-gluco-pyranoside. — Bromide 1 (1.92 g, 3 mmol) was condensed with compound 14 (0.92 g, 2 mmol) in 1:1 benzene-nitromethane, in the presence of mercuric cyanide (0.72 g, 3 mmol), exactly as described for 7 (to give 8 and 9). After processing in the usual manner, t.l.c. (solvent D) revealed the presence of two products, both faster-migrating than 14; a little unchanged 14; and some faster-migrating contaminants (possibly due to the decomposition of 1). The crude product was applied to a column of silica gel, and sequentially eluted with solvents H, G, and F. Evaporation of the fractions containing the faster-migrating product yielded 1.15 g of material that was contaminated (t.l.c., solvent D) with some faster-migrating impurities. On evaporation, the fractions containing the slower-migrating product yielded 1.30 g of material that was contaminated (t.l.c., solvent D) with some unchanged 14.

The slower-migrating, crude product (1.3 g) in 70% aqueous acetic acid (50 mL) was heated for 1 h at ~55°. After processing in the usual manner, the product was purified in a column of silica gel by sequential elution with solvents *E*, *D*, and *C*, to afford **15** (0.63 g; 51.4% yield based on **14**); amorphous; $[\alpha]_D$ +95.9° (*c* 2.3, methanol); ¹H-n.m.r. data (Me₂SO-d₆): δ 1.13 (d, 3 H, *J* 6 Hz, CMe), 1.90–2.14 (cluster of singlets, 21 H, 6 AcO and NAc), 5.92 (d, 1 H, J_{1,2} 4 Hz, H-1), and 7.17–7.98 (m, 5 H, NH and aromatic); ¹³C-n.m.r. data (Me₂SO-d₆): δ 15.72 (C-6"), 20.31 (OAc), 22.52 (NAc), 47.26 (C-2), 60.44 (C-6'), 60.82 (C-6), 72.75 (C-2'), 74.70 (C-3), 95.34 (C-1"), 95.83 (C-1'), 97.10 (C-1), and 168.95–169.87 (C=O).

Anal. Calc. for $C_{38}H_{50}N_2O_{23}$: C, 50.55; H, 5.58; N, 3.10. Found: C, 50.28; H, 5.79; N, 2.95.

Cleavage of the acetal group of the faster-migrating compound afforded, after processing and column-chromatographic purification by using solvents *E*, *D*, and *C* as the eluants (as described for the slower-migrating compound to give **15**), the β anomer **17** (0.23 g; 19% yield, based on **14**); $[\alpha]_D +55.4^\circ$ (*c* 1.5, methanol); ¹H-n.m.r. data (Me₂SO-*d*₆): δ 1.09 (d, 3 H, *J* 6 Hz, CMe), 1.92–2.14 (cluster of singlets, 21 H, 6 AcO and NAc), 5.66 (d, 1 H, *J*_{1,2} 4 Hz, H-1), 7.20–8.00 (m, 4 H, aromatic, *J* 9 Hz, NH); ¹³C-n.m.r. data (Me₂SO-*d*₆): δ 15.23 (C-6"), 20.27 (OAc), 22.67 (NAc), 47.61 (C-2), 60.02 (C-6), 61.32 (C-6'), 73.56 (C-2'), 73.85 (C-3), 95.78 (C-1), 97.82 (C-1"), 100.69 (C-1'), and 168.80–169.75 (C=O).

Anal. Calc. for C₃₈H₅₀N₂O₂₃: C, 50.55; H, 5.58; N, 3.10. Found: C, 50.36; H, 5.74; N, 3.17.

o-Nitrophenyl 2-acetamido-2-deoxy-3-O-(2-O- α -L-fucopyranosyl- α -D-galactopyranosyl)- α -D-galactopyranoside (16). — A solution of compound 15 (0.48 g) in methanol (20 mL) containing 0.5M sodium methoxide in methanol (1 mL) was refrigerated overnight, whereupon crystallization occurred. The crystalline material was filtered off, thoroughly washed with cold methanol, and dried, to give compound 16 (0.3 g, 87%); m.p. 290–291° (from methanol), $[\alpha]_D$ +161.9° (c 0.5, Me₂SO).

Anal. Calc. for C₂₆H₃₈N₂O₁₇: C, 48.00; H, 5.89; N, 4.31. Found: C, 47.55; H, 6.04; N, 4.26.

o-Nitrophenyl 2-acetamido-2-deoxy-3-O-(2-O- α -L-fucopyranosyl- β -D-galactopyranosyl)- α -D-galactopyranoside (18). — O-Deacetylation of compound 17 (0.17 g), as described for 15 (to give 16), afforded the trisaccharide 18 (90 mg, 74%); [α]_D +116.1° (c 0.5, Me₂SO); ¹H-n.m.r. data (Me₂SO-d₆): δ 1.09 (d, 3 H, J 6 Hz, CMe), 1.82 (s, 3 H, NAc), 4.64 (1 H, H-1'), 4.92 (1 H, H-1"), 5.85 (d, 1 H, J 3.5 Hz, H-1), and 7.16–7.96 (m, 5 H, NH and aromatic); for ¹³C-n.m.r. data, see Table II.

Anal. Calc. for $C_{26}H_{38}N_2O_{17} \cdot 0.5 H_2O$: C, 47.34; H, 5.96; N, 4.25. Found: C, 47.42; H, 6.06; N, 3.91.

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