

# Total Synthesis of (–)-Herbindoles A, B, and C via Transition-Metal-Catalyzed Intramolecular [2 + 2 + 2] Cyclization between Ynamide and Diynes

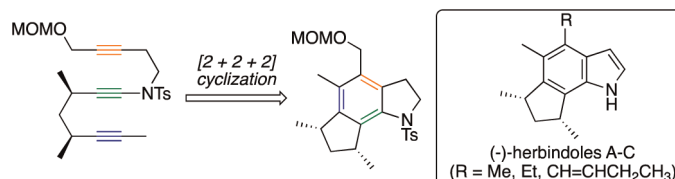
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## ABSTRACT



The total syntheses of (–)-herbindoles A, B, and C as naturally occurring forms were accomplished for the first time through transition-metal-catalyzed intramolecular [2 + 2 + 2] cyclization between ynamide and diynes. This strategy provided a highly efficient synthetic route to all three herbindoles from an identical indoline derivative as a common intermediate.

(–)-Herbindoles A, B, and C (Figure 1), belonging to polyalkylated cyclopent[gl]indole alkaloids, were isolated in 1990 from a Western Australian sponge, *Axinella* sp., and were shown to exert cytotoxicity against KB cells as well as general antifeedant activity against fishes.<sup>1,2</sup> In 1992, Natsume reported the first total synthesis of (+)-herbindoles A, B, and C through an indole cyclization of pyrrole derivatives, by which the absolute configurations of the naturally occurring herbindoles A, B, and C were unambiguously determined to be antipodes of their synthetic ones.<sup>3</sup> Although syntheses of racemic herbindoles A and B were independently reported by Kerr's group<sup>4a,b</sup> and Buszek's group,<sup>4c</sup> there have been no reports on the total synthesis of natural (–)-herbindoles.

(1) Herb, R.; Carroll, A. R.; Yoshida, W. Y.; Scheuer, P. J.; Paul, V. J. *Tetrahedron* **1990**, *46*, 3089.

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(3) For total synthesis of unnatural (+)-herbindoles A, B, and C and determination of their absolute configurations, see: (a) Muratake, H.; Mikawa, A.; Natsume, M. *Tetrahedron Lett.* **1992**, *33*, 4595. (b) Muratake, H.; Mikawa, A.; Seino, T.; Natsume, M. *Chem. Pharm. Bull.* **1994**, *42*, 854.

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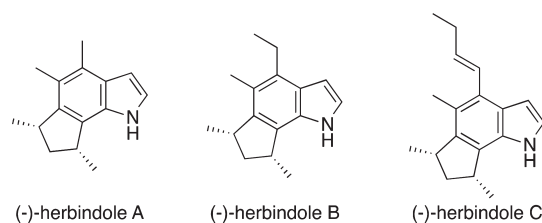
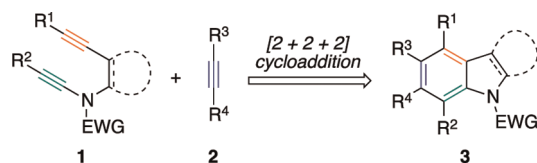


Figure 1. (–)-Herbindoles A, B, and C.

Transition-metal-catalyzed inter- and intramolecular [2 + 2 + 2] cycloaddition of three unsaturated bonds has been recognized as a useful and promising methodology for the synthesis of polycyclic compounds in recent organic synthesis.<sup>5,6</sup> In particular, the [2 + 2 + 2] cycloaddition of triynes is known to be an efficient synthetic protocol for the synthesis of various aromatic compounds. In this context, the [2 + 2 + 2] cycloaddition of three alkynes including an ynamide<sup>7–9</sup> (**1** and **2**) has been recently established as a new method for the construction of indole skeleton **3** (Scheme 1).<sup>10</sup>

With this as a background, we planned the total synthesis of (–)-herbindoles A, B, and C by intramolecular

**Scheme 1.** Construction of an Indole Skeleton via [2 + 2 + 2] Cycloaddition of Alkyne–Alkyne–Ynamide



[2 + 2 + 2] cyclization of ynamide-diyne using a transition metal catalyst. Our retrosynthetic analysis of them is shown in Scheme 2. All three herbindoles could potentially be synthesized from the identical cyclopentane-fused indoline derivative **4** as a key intermediate, whose aromatic

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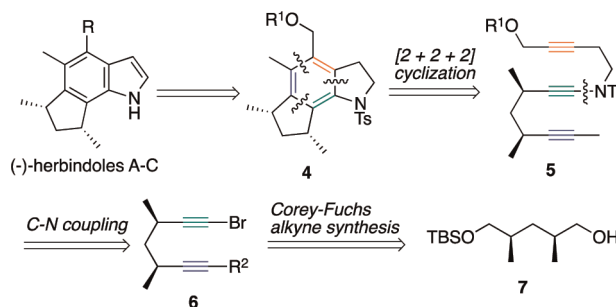
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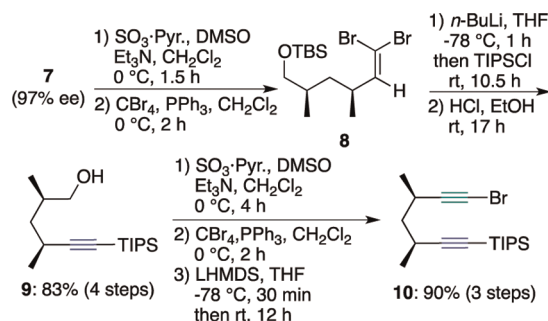
ring would be constructed by transition metal-catalyzed [2 + 2 + 2] cyclization of the dialkynylynamide **5**. Construction of the ynamide part of **5** could be performed by C–N coupling between bromoalkyne **6** and the corresponding nitrogen nucleophile. Furthermore, the bromoalkyne **6** could be easily synthesized from the known optically active alcohol **7**<sup>11</sup> via Corey–Fuchs alkyne synthesis.

**Scheme 2.** Retrosynthesis of (–)-Herbindoles A, B, and C



Preparation of the bromoalkyne unit is shown in Scheme 3. First, the known alcohol **7** (97% ee) was transformed into dibromoalkene **8**, from which alkyne formation followed by deprotection of the TBS group was conducted to give an alcohol **9**. Then dibromoalkene prepared from **9** via

**Scheme 3.** Preparation of Bromoalkyne Part



(9) For our reports on transition metal catalysis utilizing ynamide as a platform, see: (a) Saito, N.; Sato, Y.; Mori, M. *Org. Lett.* **2002**, *4*, 803. (b) Mori, M.; Wakamatsu, H.; Saito, N.; Sato, Y.; Narita, R.; Sato, Y.; Fujita, R. *Tetrahedron* **2006**, *62*, 3872. (c) Saito, N.; Katayama, T.; Sato, Y. *Org. Lett.* **2008**, *10*, 3829. (d) Saito, N.; Katayama, T.; Sato, Y. *Heterocycles* **2011**, *82*, 1181. (e) Saito, N.; Saito, K.; Shiro, M.; Sato, Y. *Org. Lett.* **2011**, *13*, 2718.

(10) For examples of transition-metal-catalyzed [2 + 2 + 2] cycloaddition of alkyne–alkyne–ynamide leading to an indole skeleton, see: (a) Witulski, B.; Stengel, T. *Angew. Chem., Int. Ed.* **1999**, *38*, 2426. (b) Witulski, B.; Stengel, T.; Fernández-Hernández, J. M. *Chem. Commun.* **2000**, 1965. (c) Witulski, B.; Alayrac, C. *Angew. Chem., Int. Ed.* **2002**, *41*, 3281. For related transition-metal-catalyzed [2 + 2 + 2] cycloaddition of alkyne–nitrile–ynamide leading to hetroaromatic compounds, see: (d) Alayrac, C.; Schollmeyer, D.; Witulski, B. *Chem. Commun.* **2009**, 1464. (e) Garcia, P.; Moulin, S.; Miclo, Y.; Leboeuf, D.; Gandon, V.; Aubert, C.; Malacria, M. *Chem.—Eur. J.* **2009**, *15*, 2129. (f) Nissen, F.; Detert, H. *Eur. J. Org. Chem.* **2011**, 2845. (g) Nissen, F.; Richard, V.; Alayrac, C.; Witulski, B. *Chem. Commun.* **2011**, 47, 6656. (h) Garcia, P.; Evanno, Y.; George, P.; Sevrin, M.; Ricci, G.; Malacria, M.; Aubert, C.; Gandon, V. *Org. Lett.* **2011**, *13*, 2030.

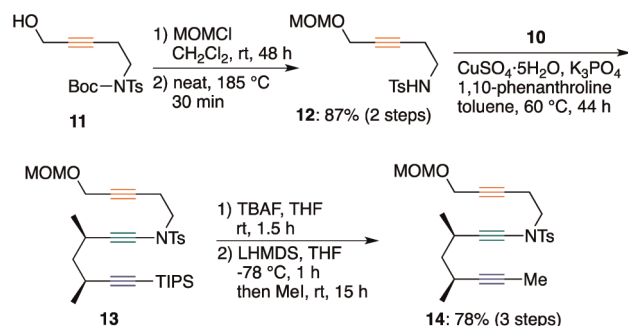
(11) Prusov, E.; Röhm, H.; Maier, M. E. *Org. Lett.* **2006**, *8*, 1025.

oxidation—dibromoalkenylation was treated with LHMDS<sup>12</sup> to afford the bromoalkyne unit **10**.

The hydroxy group of the known compound **11**<sup>13</sup> was protected by the MOM group, and deprotection of the Boc group by heating gave tosylamide **12** (Scheme 4). In the presence of a copper catalyst,<sup>14</sup> bromoalkyne **10** and amide **12** were coupled to give dialkynylnamide **13**, which was converted into the [2 + 2 + 2] cyclization precursor **14** in good yield.

Next, we examined the [2 + 2 + 2] cyclization of **14** using various transition metal catalysts, which have been employed previously in a variety of inter- and intramolecular [2 + 2 + 2] cycloadditions<sup>5</sup> (Table 1). The cyclization of **14** in the presence of a Cp\*RuCl(cod) catalyst proceeded at room temperature to give the expected indoline derivative **15** in 91% yield (run 1). Group 9 metal complexes, CpCo(CO)<sub>2</sub> and RhCl(PPh<sub>3</sub>)<sub>3</sub>, also showed good catalytic activity for the [2 + 2 + 2] cyclization of ynamide derivative **14**, and the cyclized product **15** was obtained in excellent yields (runs 2 and 3). The reaction of **14** by using an Ni(0)-PPh<sub>3</sub> catalyst also afforded **15** in good yield (run 4). On the other hand, a Pd(0)-PPh<sub>3</sub> catalyst did not promote the cyclization, and the starting **14** was recovered in almost quantitative yield (run 5). Thus, we decided to employ the Wilkinson's catalyst to synthesize herbindoles.

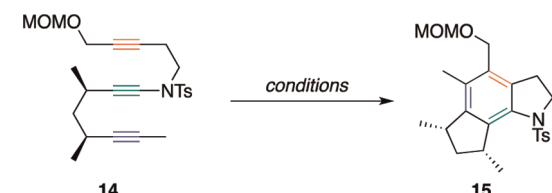
**Scheme 4.** Synthesis of Dialkynylnamide **14** as a [2 + 2 + 2] Cyclization Precursor



With the supposed common intermediate **15** in hand, we set out to conduct the transformation of **15** into three herbindoles. The cyclized product **15** was treated with BBr<sub>3</sub>, giving benzylic bromide derivative **16** which, after radical reduction, gave **17** in high yield (Scheme 5). Finally, deprotection of the tosyl group followed by aromatization in the presence of a cobalt(II) catalyst<sup>15</sup> produced (–)-herbindole A, whose spectral data were identical to those reported for naturally occurring herbindole A. The value of [α]<sub>D</sub> was also identical with the synthetic

(+)-herbindole A except for the sign of [α]<sub>D</sub>.<sup>3</sup> It is noteworthy that the overall yield of (–)-herbindole A from the known compound **7** was 49% yield in 15 steps (average ca. 95.3% yield in each step).

**Table 1.** Construction of a Cyclopent[*g*]indole Skeleton by Transition-Metal-Catalyzed [2 + 2 + 2] Cyclization of **14**

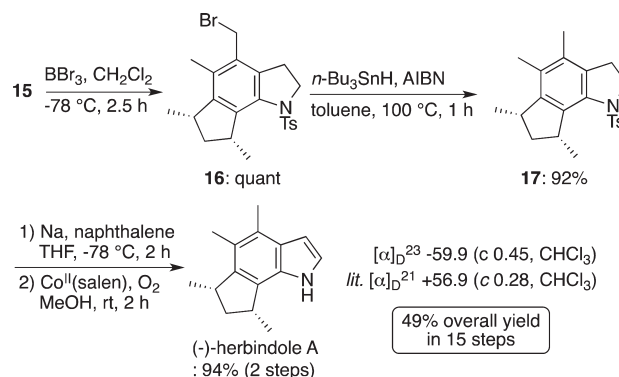


run	catalyst (mol %)	solvent	temp (°C)	time (h)	yield (%)
1	Cp*RuCl(cod) (5)	toluene	rt	48	91
2	CpCo(CO) <sub>2</sub> (10)	<i>p</i> -xylene	140	24	95 <sup>a</sup>
3	RhCl(PPh <sub>3</sub> ) <sub>3</sub> (4)	toluene	50	5	97
4	Ni(cod) <sub>2</sub> (5) PPh <sub>3</sub> (10)	THF	50	24	80
5	Pd <sub>2</sub> dba <sub>3</sub> ·CHCl <sub>3</sub> (2.5) PPh <sub>3</sub> (10)	toluene	50	24	(98) <sup>b</sup>

<sup>a</sup>NMR yield using 1,3,5-trimethoxybenzene as an internal standard. <sup>b</sup>The values in parentheses are the yields of starting dialkynylnamide **14**.

The synthesis of (–)-herbindole B was also achieved as shown in Scheme 6. Thus, after removal of the MOM group of **15**, oxidation of the corresponding alcohol **18** by Dess-Martin periodinane (DMP) followed by methylenation using Tebbe reagent and hydrogenation afforded indoline derivative **19**. Finally, (–)-herbindole B was synthesized through deprotection of **19** followed by aromatization in excellent yield (average ca. 94.2% yield in each step).

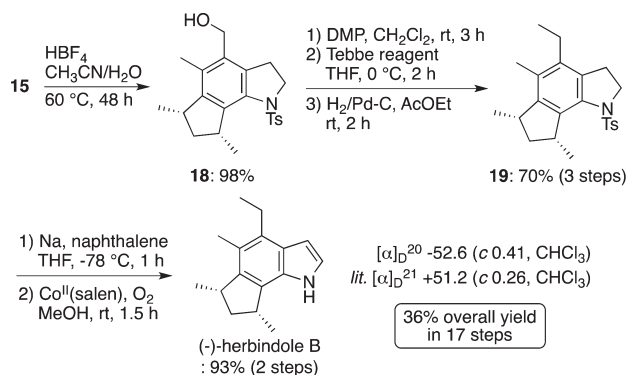
**Scheme 5.** Synthesis of (–)-Herbindole A from **15**



As shown in Scheme 7, the above alcohol **18** was easily converted into **20** in 91% yield through oxidation,

(12) Huang, Z.; Negishi, E. *J. Am. Chem. Soc.* **2007**, *129*, 14788.  
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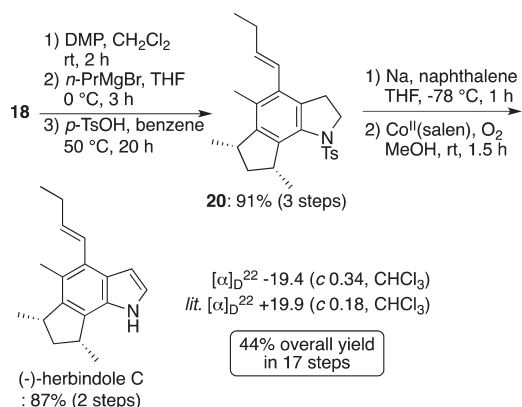
**Scheme 6.** Synthesis of (–)-Herbindole B from **15**



Grignard reaction and subsequent dehydration. The synthesis of (–)-herbindole C was also achieved by similar transformation from **20** as the above-described procedure (average ca. 95.3% yield in each step).

In summary we have achieved, for the first time, an efficient synthesis of (–)-herbindoles A, B, and C in naturally occurring forms. The key reaction of the total syntheses was transition metal-catalyzed [2 + 2 + 2] cycloaddition of an ynamide-diyne, with ruthenium, cobalt, and rhodium as well as nickel complexes catalyzing the cyclization to give the cyclopent[*g*]indole skeleton in high yield. This strategy provides a highly efficient synthetic route to all three herbindoles from an identical indoline derivative as a common intermediate.

**Scheme 7.** Synthesis of (–)-Herbindole C from **18**



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**Supporting Information Available.** Experimental procedure and spectral data. This material is available free of charge via the Internet at <http://pubs.acs.org>.

The authors declare no competing financial interest.