Chemistry Letters 1998

Stereospecific Reduction with Retention of Chiral Fluoral-derived 1,3-Oxazolidines with LiAlH₄: Asymmetric Synthesis of 1-Substituted 2,2,2-Trifluoroethylamines

Akihiro Ishii, ¹ Fumie Miyamoto, [†] Kimio Higashiyama, *[†] and Koichi Mikami*

Department of Chemical Technology, Tokyo Institute of Technology, Meguro-ku, Tokyo 152

[†] Faculty of Pharmaceutical Science, Hoshi University, Shinagawa-ku, Tokyo 142

(Received October 8, 1997; CL-970781)

The asymmetric synthesis of 1-substituted 2,2,2-trifluoroethylamines from chiral 1,3-oxazolidines having trifluoromethyl group derived from trifluoromethyl ketones and (R)-phenylglycinol is described. The stereospecific reduction of chiral 1,3-oxazolidines with LiAlH4 proceeds in retention fashion.

There is current interest in developing an efficient method for the asymmetric synthesis of organofluorine compounds.² However, the method established for nonfluorine substrates can not be always applied to organofluorine counterparts because of unique electronic and/or steric character of fluorine.³ As a part of a research program directed toward the asymmetric synthesis of organofluorine compounds, we report herein a practical route to the asymmetric synthesis of 1-substituted 2,2,2-trifluoroethylamines by the stereospecific reduction of chiral 1,3-oxazolidines with LiAlH4 in *retention* fashion. Particularly, 1-phenyl-2,2,2-trifluoroethylamine thus synthesized is useful as a chiral derivatizing agent for the chromatographic separation and ¹⁹F-NMR analysis.⁴

Ph., H
PPTS, toluene
reflux (-H₂O)
$$CF_3$$
 $R = Ph, PhCH_2$
 $(R) - 1$
 Ph , H
PPTS, toluene
reflux (-H₂O)
 CF_3
 $CF_$

Scheme 1

Table 1. Preparation of 1,3-oxazolidines

run	oxazolidine: R	$2S:2R^{a}$	time (h)	yield (%)b
1	2a: Ph	1.6: 1	50	97
2	2b : PhCH ₂	1.9:1	16	92

^a Determined by ¹H-NMR analysis. ^b Isolated yield.

The starting chiral 1,3-oxazolidine having trifluoromethyl group at 2-position was prepared as follows. The condensation of trifluoromethyl ketones with (R)-phenylglycinol (1) by heating in toluene with azeotropic removal of water gave the chiral 2-trifluoromethyl-1,3-oxazolidines in excellent yields as a diastereomeric mixture (2a (Rf (CH₂Cl₂: n-hexane = 1:2));

major (0.50): minor (0.56) = 1.6:1, **2b** (Rf); major (0.58): minor (0.62) = 1.9:1 by ¹H-NMR analysis) (Scheme 1, Table 1). The each diastereomer was separated by column chromatography. The absolute configuration of the minor diastereomer (2a) was determined to be 2R, 4R by X-ray analysis. 2-Benzyl-1,3-oxazolidine (2b) is also deduced to possess the same stereochemistry.

When each diastereomer was treated with LiAlH4, the opposite sense of diastereoselectivity was obtained, presumably through direct displacement of the 1,3-oxazolidines (Scheme 2, Table 2, run 1 vs 2, run 3 vs 4), in direct contrast to the nonfluorinated cases which give a single diastereomer by catalytic reduction through an equilibrium mixture with imine.⁵

Ph. H

$$CF_3$$
 CF_3
 CF_3

Scheme 2.

Table 2. Asymmetric reduction of 1,3-oxazolidine with LiAlH₄

run	oxazolidine	product	1R: 1S a	yield (%)b
1	(2S)- 2a	(1R)-3a	88: 12	93
2	(2R)- 2a	(1S)-3a	10:90	90
3	(2S)- 2b	(1R)-3b	94: 6	94
4	(2R)- 2b	(1S)- 3b	0:100	92

a Determined by ¹H-NMR analysis. b Isolated yield.

Therefore, asymmetric reduction of imine (4)⁶ having trifluoromethyl and benzyl group was next investigated with LiAlH₄ (Scheme 3). Because (1S)-3b was obtained in highly diastereoselective manner, it can be concluded that 1,3-oxazolidine does not react after transformation to the metalloimine⁷ through ring opening with LiAlH₄.

Scheme 3.

These results suggest that 1,3-oxazolidine having trifluoromethyl group at 2-position directly reacted with LiAlH4 in retention fashion (T) without formation of the imine intermediate.

These phenomenon may be attributed to retardation of ring opening to imine by introduction of highly electron-withdrawing group to destabilize the possible formation of the positive charge at the 2-position. To the best of our knowledge, this is the first example of stereospecific nucleophilic substitution in retention fashion at the carbon α to trifluoromethyl group. In this asymmetric reduction, both enantiomers of fluorinated amines can be prepared from the single enantiomeric source because the each diastereomer of 1,3-oxazolidines are transformed to the known4 chiral 1-substituted 2,2,2-trifluoroethylamines by oxidative cleavage or hydrogenolysis of the reduction products (3a, 3b) without epimerization.

Thus, we achieved the asymmetric synthetic route to 1-substituted 2,2,2-trifluoroethylamines by employing the stereospecific reduction of chiral 1,3-oxazolidines having trifluoromethyl group with LiAlH4 in retention fashion.

We thank Emeritus Prof. Hiroshi Takahashi for his encouragement and useful discussion.

References and Notes

- 1 On leave from Chemical Research Center, Central Glass Co. Ltd., Kawagoe, Saitama 350-11
- 2 Reviews: "Enantiocontrolled Synthesis of Fluoro-Organic Compounds: Stereochemical Challenges and Biomedical Targets," ed by V. A. Soloshonok, Wiley, Chichester (1998); G. Resnati and V. A. Soloshonok, Tetrahedron, 52, 1 (1996); K. Iseki and Y. Kobayashi, J. Synth. Org. Chem., Jpn., 54, 122 (1996); T. Hayashi and V. A. Soloshonok, Tetrahedron: Asymmetry, 5, 955 (1994); G. Resnati, Tetrahedron, 49, 9385 (1993); T. Kitazume, T. Ishihara, and T. Taguchi, "Fluorine Chemistry," Kodansha Scientific, Tokyo (1993); G. A. Olah, R. D. Chambers, and G. K. S. Prakash, "Synthetic Fluorine Chemistry," Wiley, N.Y. (1992).

- 3 In the reaction of fluorinated compounds, lower degree or opposite sense of stereoselectivity were observed in comparison with nonfluorinated counterparts: a) N. Sayo, T. Saito, S. Akutagawa, R. Noyori, and H. Takaya, E. Patent 297752A2 (1989). b) M. Saburi, L. Shao, T. Sakurai, and Y. Uchida, Tetrahedron Lett., 33, 1631 (1992). c) K. E. Koeing, G. L. Bachman, and B. D. Vineyard, J.Org. Chem., 45, 2362 (1980). d) K. Iseki, S. Oishi, T. Taguchi, and Y. Kobayashi, Tetrahedron Lett., 34, 8147 (1993). Also see: e) E. J. Corey, X-M. Cheng, A. Karlene, and S. Sarshar, Tetrahedron Lett., 32, 6835 (1991). f) P. V. Ramachandran, B. Gong, A. V. Teodorovic, and H. C. Brown, Tetrahedron: Asymmetry, 5, 1061 (1994).
- Previous reports on the preparation of chiral 1-substituted 2,2,2-trifluoroethylamines: a) Asymmetric reduction of chiral imine: W. H. Pirkle and J. R. Hauske, J. Org. Chem., 42, 2436 (1977). b) Optical resolution: Y. Wang and H. S. Mosher, Tetrahedron Lett., 32, 987 (1991). c) 1,3-proton shift: V. A. Soloshonok and T. Ono, J. Org. Chem., 62, 3030 (1997).
 C. K. Miao, R. Sorcek, and P.-J. Jones, Tetrahedron Lett., 34, 2259
- C. K. Miao, R. Sorcek, and P.-J. Jones, *Tetrahedron Lett.*, 34, 2259 (1993).
- 6 Benzyl trifluoromethyl ketone (940 mg, 5 mmol) was added to a solution of (R)-phenylglycinol (1) (343 mg, 2.5 mmol) in benzene (50 ml), and the reaction mixture was refluxed for 12 h using a Dean-Stark trap. After cooling, the mixture was concentrated under reduced pressure to give the crude product, which was subjected to column chromatography on silica gel with CH2Cl2 to give benzyl imine product (4) (43% yield). The geometry of chiral imine (4) was deduced to be E from the result of X-ray analysis of a similar (R)-1-trifluoromethyl-N-(2-hydroxy-1-phenylethyl)benzylideneamine.
- H. Takahashi, Y. Suzuki, and T. Hori, Chem. Pharm. Bull., 31, 2183 (1983).
- a) F. Fülöp, G. Bernáth, J. Mattinen, and K. Pihlaja, *Tetrahedron*, 45, 4317 (1989).
 b) R. H. Premchandran, M. L. Ogletree, and J. Fried, *J. Org. Chem.*, 58, 5724 (1993).
 c) Y. Takagi, K. Nakai, T. Tsuchiya, and T. Takeuchi, *J. Med. Chem.*, 39, 1582 (1996).
- 9 Substitution in inversion fashion at the carbon α to trifluoromethyl group: a) T. Hagiwara, K. Tanaka, and T. Fuchikami, Tetrahedron Lett., 37, 8187 (1996). b) H. Matsutani, J. Yaruva, H. Poras, T. Kusumoto, and T. Hiyama, Abstr., B109 in the 44th Symposium on Organometallic Chemistry of the Kinki Chemical Society, Osaka, Japan, 1997. H. Matsutani, T. Kusumoto, and T. Hiyama, Abstr., 1G201 in the 72th Annual Meeting of the Chemical Society of Japan, Tokyo, 1997. For the mechanism on the Lewis acid-catalyzed nucleophilic substitution of nonfluorinated acetals, see: c) T. Sammakia and R. S. Smith, J. Am. Chem., 116, 7915 (1994). d) K. Ishihara, N. Hanaki, and H. Yamamoto, J. Am. Chem., 115, 10695 (1993). e) S. E. Denmark and N. G. Almstead, J. Am. Chem., 113, 8089 (1991). f) A. Alexakis and P. Mangeney, Tetrahedron: Asymmetry, 1, 477 (1990).
- 10 a) Oxidative cleavage by Pb(OAc)4: H. Takahashi, B. C. Hsieh, and K. Higashiyama, *Chem. Pharm.* Bull., 38, 2429 (1990). b) Hydrogenolysis by Pd-C: Ref. 4a.