This article was downloaded by: [Florida State University] On: 19 May 2013, At: 14:20 Publisher: Taylor & Francis Informa Ltd Registered in England and Wales Registered Number: 1072954 Registered office: Mortimer House, 37-41 Mortimer Street, London W1T 3JH, UK



Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry

Publication details, including instructions for authors and subscription information: http://www.tandfonline.com/loi/lsyc20

Microwave-Assisted Synthesis of New Spiro[indoline-3,4'quinoline] Derivatives via a One-Pot Multicomponent Reaction

Song-Lei Zhu ^{a b} , Kai Zhao ^a , Xiao-Ming Su ^a & Shun-Jun Ji ^a

^a Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry and Chemical Engineering, Suzhou (Soochow) University, Suzhou, China

^b Department of Chemistry, Xuzhou Medical College, Xuzhou, China Published online: 19 Mar 2009.

To cite this article: Song-Lei Zhu , Kai Zhao , Xiao-Ming Su & Shun-Jun Ji (2009): Microwave-Assisted Synthesis of New Spiro[indoline-3,4'-quinoline] Derivatives via a One-Pot Multicomponent Reaction, Synthetic Communications: An International Journal for Rapid Communication of Synthetic Organic Chemistry, 39:8, 1355-1366

To link to this article: http://dx.doi.org/10.1080/00397910802527714

Full terms and conditions of use: <u>http://www.tandfonline.com/page/terms-and-conditions</u>

This article may be used for research, teaching, and private study purposes. Any substantial or systematic reproduction, redistribution, reselling, loan, sub-licensing, systematic supply, or distribution in any form to anyone is expressly forbidden.

The publisher does not give any warranty express or implied or make any representation that the contents will be complete or accurate or up to date. The accuracy of any instructions, formulae, and drug doses should be independently verified with primary sources. The publisher shall not be liable for any loss, actions, claims, proceedings, demand, or costs or damages whatsoever or howsoever caused arising directly or indirectly in connection with or arising out of the use of this material.





Microwave-Assisted Synthesis of New Spiro[indoline-3,4'-quinoline] Derivatives via a One-Pot Multicomponent Reaction

Song-Lei Zhu,^{1,2} Kai Zhao,¹ Xiao-Ming Su,¹ and Shun-Jun Ji¹

¹Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry and Chemical Engineering, Suzhou (Soochow) University, Suzhou, China
²Department of Chemistry, Xuzhou Medical College, Xuzhou, China

Abstract: A simple and efficient synthesis of spiro[indoline-3,4'-quinoline] derivatives by a one-pot reaction of isatin, malononitrile, and enaminone under microwave irradiation was investigated. This protocol has the advantages of short reaction time, high yields of products, broad substrate scope, and easy workup procedure.

Keywords: Microwave irradiation, multicomponent reactions, spiro[indoline-3,4'quinoline]

Multicomponent reactions (MCRs), in which multiple reactions are combined into one synthetic operation, have been used extensively in synthetic chemistry.^[1] Such reactions offer a wide range of possibilities for the efficient construction of highly complex molecules in a single procedural step, thus avoiding complicated purification operations and allowing savings of both solvents and reagents.^[2] Organic reactions accelerated by microwave irradiation (MWI) have also attracted considerable attention over past years for the rapid and efficient synthesis of a variety of organic compounds.^[3]

Received June 5, 2008.

Address correspondence to Shun-Jun Ji, Key Laboratory of Organic Synthesis of Jiangsu Province, College of Chemistry and Chemical Engineering, Suzhou (Soochow) University, Suzhou 215123, China. E-mail: chemjsj@suda.edu.cn

These technologies are particularly useful for the creation of diverse chemical libraries of druglike molecules for biological screening.^[4]

Isatins (1*H*-indole-2,3-diones) are synthetically versatile substrates that are utilized for the synthesis of a large variety of heterocyclic compounds.^[5] A number of spiro compounds derived from isatin are important pharmacophores and exhibit promising biological activities such as analgesic, fungicidal, antidepressant, antitumor, and antibiotic activities.^[6,7] Likewise, polysubstituted quinoline, an important heterocyclic molecule, has also been reported for use in the pharmaceutical industry and possesses a broad spectrum of biological potency including antiasthmatic, anti-inflammatory, and antimalarial activities.^[8] Therefore, the diverse biological activities reported for isatin and polysubstituted quinoline encouraged us to synthesize some new heterocyclic derivatives incorporating these moieties, which may lead to enhanced bioactivity.

Recently, Abdel-Rahman et al. have reported the synthesis and evaluation of some new spiro indoline-based heterocycles as potentially active antimicrobial agents by the reaction of isatin, malononitrile or ethyl cyanoacetate, and oxazolone derivatives.^[7a] Dandia et al. have developed a microwave-assisted, one-pot, three-component, regioselective cyclocon-densation of isatin, 3-amino-1,2,4-triazole or 2-aminobenzimidazole and thioacids to afford of a series of novel spiro[indole-thiazolidinones].^[7b] In our previous work,^[9] we synthesized a series of indole derivatives, which have shown some potential biological activities, under different conditions. In this article, we investigated a simple and facile protocol for the synthesis of a series of new spiro[indoline-3,4'-quinoline] derivatives.

The target compounds 2'-amino-1'-aryl-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile **4** were synthesized by one-pot reactions of isatins **1**, malononitrile **2**, and enaminones **3** under MWI (Scheme 1).

Choosing an appropriate solvent is of crucial importance for a successful MW-assisted synthesis. In our initial study, various reaction conditions including solvent and temperature were tested in the one-pot, three-component synthesis of 4a under MW irradiation. Among different polar solvents, such as ethanol, acetic acid, glycol,



Scheme 1. Synthesis of 4 by reaction of isatins, malonitrile, and enaminones.

Entry	Solvent ^b	T (°C)	Time (min)	Yield ^c (%)
1	H ₂ O	80	15	Trace
2	EtOH	80	8	78
3	HOAc	80	10	50
4	DMF	80	10	42
5	Glycol	80	10	57
6	EtOH	60	10	65
7	EtOH	70	9	72
8	EtOH	90	8	76

Table 1. Optimization of reaction conditions of compound $4a^{a}$

^aThe reaction was carried out under MWI at the power of 150 W.

^bThe amount of solvent was 2 mL.

^cIsolated yields.

dimethylformamide (DMF), and water, the best yield was found when ethanol was employed. The most suitable temperature should be 80 °C. The results are summarized in Table 1.

Moreover, when **4a** was synthesized under both MWI and classical heating at 80 °C, we found that MWI efficiently promoted the reaction and led to both a dramatic reduction of the reaction time, from 6 h to 8 min, and a remarkable increase in yield from 50% to 78%.

To expand the scope of the present method, different substituted isatins and enaminones containing either electron-withdrawing or electron-donating groups were applied in this reaction. From Table 2, it can be seen that the method works with a wide variety of substrates, and a series of spiro[indoline-3,4'-quinoline] derivatives **4** were obtained in good yields under the same reaction conditions.

The products were characterized by melting point, infrared (IR), ¹H NMR, and high-resolution mass spectrometry (HRMS) spectral data. Furthermore, the structure of **4g** was further confirmed by x-ray diffraction (XRD) analysis (Fig. 1). Crystallographic data for the structure of **4g** reported in this article have been deposited at the Cambridge Crystallographic Data Centre with No. CCDC-680112.

The proposed mechanism for the synthesis of spiro derivative 4 is described in Scheme 2. The process represents a typical cascade reaction in which the isatin 1 first condenses with malononitrile 2 to afford isatylidene malononitrile 5. This step was regarded as a fast Knoevenagel condensation. Then, 5 is attacked via Michael addition with enaminone 3 to give the intermediate 6 and followed by the intramolecular cyclization to form the desired product 4.

Evidence supporting this proposed mechanism was provided by the observation that when 5 and 3a were subjected to the same conditions,

Entry	\mathbb{R}^1	\mathbb{R}^2	\mathbb{R}^3	\mathbb{R}^4	Time (min)	Yield ^a (%)	M.p (°C)
4 a	Н	Н	CH ₃	C ₆ H ₅	8	78	>300
4b	Н	Η	CH ₃	4-ClC ₆ H ₄	7	76	>300
4c	Н	Η	CH ₃	$4-CH_3C_6H_4$	7	75	>300
4d	Н	Η	CH_3	4-OCH ₃ C ₆ H ₄	8	77	>300
4 e	Н	Η	CH ₃	$4-BrC_6H_4$	7	85	>300
4f	Н	Η	CH ₃	1-Naphthyl	10	66	>300
4g	$-CH_3$	Η	CH_3	$4-ClC_6H_4$	9	81	>300
4h	$-CH_2Ph$	Η	CH_3	$4-ClC_6H_5$	10	70	>300
4i	$-CH_2Ph$	Η	CH_3	$4-CH_3C_6H_4$	9	72	>300
4j	Н	4-Br	CH_3	4-OCH ₃ C ₆ H ₄	10	67	>300
4k	Н	6-Br	CH_3	$4-ClC_6H_4$	8	82	>300
4 1	Н	6-Br	CH_3	$4-CH_3C_6H_4$	8	75	>300
4m	Н	Η	Η	$4-ClC_6H_4$	9	77	>300
4n	Н	Η	Η	$4-BrC_6H_4$	9	80	>300
40	Н	Н	Н	$4\text{-OCH}_3\text{C}_6\text{H}_4$	9	75	>300

Table 2. Synthesis of compound 4 under MWI

^aIsolated yields.

the expected product **4a** was obtained in a yield similar to that obtained in the one-pot reaction (Scheme 3).

In conclusion, we have developed a facile, one-pot, three-component reaction involving isatins, malononitrile, and enaminones for the synthesis of a series of spirooxindoles derivatives under MWI. Particularly valuable features of this method include broad substrate scope, shorter reaction time, high yields of the products, and straightforward procedure, which make it a useful and attractive process for the synthesis of these important compounds.

EXPERIMENTAL

Melting points were recorded on an Electrothermal digital melting-point apparatus and are uncorrected. IR spectra were recorded on a Nicolet FT-IR500 spectrophotometer using KBr optics. ¹H NMR spectra were recorded on a Varian Mercury 400-MHz spectrometer using dimethyl sulfoxide (DMSO)- d_6 as solvent and tetramethylsilane (TMS) as internal standard. HRMS were obtained using GCT-TOF instrument. X-ray diffraction data were made on a Rigaku Mercury CCD area detector with graphite monochromated Mo-Ka radiation. CHN analyses were recorded on a Carlo-Erba EA1110 CNNO-S analyzer.



Figure 1. X-ray crystal structure of 4g.



Scheme 2. Plausible mechanism for the reaction of isatin and malononitrile with enaminones.



Scheme 3. Synthesis of **4a** by the reaction of isatylidene malononitrile with enaminone.

Typical Experimental Procedure: Preparation of Compounds 4

The reaction was performed in a monomodal Emrys Creator from Personal Chemistry, Uppsala, Sweden. In a 10-mL Emrys reaction vial, isatin 1 (1 mmol), malononitrile 2 (1 mmol), enaminone 3 (1 mmol), and ethanol (2 mL) were mixed and then capped. The mixture was irradiated at 80° C and the maximum power of 150 W for the given time. Upon completion, monitored by thin-layer chromatography (TLC), the reaction mixture was allowed to cool to room temperature and then poured into cold water (100 mL). The solid product was filtered and washed with EtOH (95%). The product was purified by recrystallization from EtOH–DMF (1:1) to afford **4a–40**.

Data

2'-Amino-7',7'-dimethyl-2,5'-dioxo-1'-phenyl-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4a**)

White solid; mp > 300°C; IR (KBr): ν 3442, 3334, 3218, 2184, 1715, 1643, 1551, 1366, 1258, 1049, 733 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.17 (br s, 1H, NH), 7.58–7.63 (m, 3H, Ar-H), 7.46–7.48 (m, 2H, Ar-H), 7.16 (d, *J*=7.2 Hz, 1H, Ar-H), 7.12 (t, *J*=7.6 Hz, 1H, Ar-H), 6.91 (t, *J*=7.6 Hz, 1H, Ar-H), 6.76 (d, *J*=7.6 Hz, 1H, Ar-H), 5.30 (s, 2H, NH₂), 2.07–2.14 (m, 2H, CH₂), 1.80–1.95 (m, 2H, CH₂), 0.88 (s, 3H, CH₃), 0.82 (s, 3H, CH₃). HRMS found: *m*/*z* 410.1738 (M⁺); calcd. for C₂₅H₂₂N₄O₂: M, 410.1743. Anal. calcd. for C₂₅H₂₂N₄O₂: C, 73.15; H, 5.40; N, 13.65. Found: C, 73.27; H, 5.27; N, 13.72.

2'-Amino-1'-(4-chlorophenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4b**)

Yellow solid; mp > 300°C; IR (KBr): ν 3457, 3334, 3218, 2191, 1721, 1644, 1566, 1489, 1366, 1188, 1049, 756 cm⁻¹; ¹H NMR (400 MHz,

New Spiro[indoline-3,4'-quinoline] Derivatives

DMSO- d_6): δ 10.21 (br s, 1H, NH), 7.56 (d, J = 8.4 Hz, 2H, Ar-H), 7.45–7.53 (m, 2H, Ar-H), 7.17 (d, J = 7.2 Hz, 1H, Ar-H), 7.11 (t, J = 7.6 Hz, 1H, Ar-H), 6.89 (t, J = 7.6 Hz, 1H, Ar-H), 6.75 (d, J = 7.6 Hz, 1H, Ar-H), 5.55 (s, 2H, NH₂), 2.07–2.16 (m, 2H, CH₂), 1.79–1.93 (m, 2H, CH₂), 0.89 (s, 3H, CH₃), 0.81 (s, 3H, CH₃). HRMS found: m/z 444.1342 (M⁺); calcd. for C₂₅H₂₁ClN₄O₂: M, 444.1353. Anal. calcd. for C₂₅H₂₁ClN₄O₂: C, 67.49; H, 4.76; N, 12.59. Found: C, 67.38; H, 4.67; N, 12.65.

2'-Amino-7',7'-dimethyl-2,5'-dioxo-1'-p-tolyl-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4c**)

White solid; mp > 300°C; ν 3434, 3318, 3218, 2191, 1721, 1643, 1559, 1366, 1258, 1196, 1048, 748 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d₆*): δ 10.16 (br s, ¹H, NH), 7.41 (d, *J*=8.0 Hz, 2H, Ar-H), 7.30–7.41 (m, 2H, Ar-H), 7.09–7.15 (m, 2H, Ar-H), 6.90 (t, *J*=7.2 Hz, 1H, Ar-H), 6.75 (d, *J*=7.6 Hz, 1H, Ar-H), 5.27 (s, 2H, NH₂), 2.41 (s, 3H, CH₃), 2.06–2.14 (m, 2H, CH₂), 1.82–1.94 (m, 2H, CH₂), 0.88 (s, 3H, CH₃), 0.82 (s, 3H, CH₃). HRMS found: m/z 424.1892 (M⁺); calcd. for C₂₆H₂₄N₄O₂: M, 424.1899. Anal. calcd. for C₂₆H₂₄N₄O₂: C, 73.56; H, 5.70; N, 13.20. Found C, 73.62; H, 5.77; N, 13.12.

2'-Amino-1'-(4-methoxyphenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4d**)

White solid; mp > 300°C; IR (KBr): ν 3435, 3337, 3218, 2184, 1716, 1643, 1592, 1470, 1361, 1195, 1051, 725 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.20 (br s, 1H, NH), 7.32–7.41 (m, 2H, Ar-H), 7.09–7.16 (m, 4H, Ar-H), 6.89 (t, *J* = 7.2 Hz, 1H, Ar-H), 6.75 (d, *J* = 7.2 Hz, 1H, Ar-H), 5.37 (s, 2H, NH₂), 3.84 (s, 3H, OCH₃), 2.06–2.15 (m, 2H, CH₂), 1.82–1.93 (m, 2H, CH₂), 0.88 (s, 3H, CH₃), 0.81 (s, 3H, CH₃). HRMS found: m/z 440.1860 (M⁺); calcd. for C₂₆H₂₄N₄O₃: M, 440.1848. Anal. calcd. for C₂₆H₂₄N₄O₃: C, 70.89; H, 5.49; N, 12.72. Found: C, 70.80; H, 5.56; N, 12.78.

2'-Amino-1'-(4-bromophenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4e**)

Yellow solid; mp > 300°C; IR (KBr): ν 3417, 3340, 3291, 2190, 1745, 1657, 1485, 1364, 1261, 1193, 1012, 751 cm⁻¹; ¹H NMR (400 MHz, DMSO- d_6): δ 10.21 (br s, 1H, NH), 7.78 (d, J = 8.8 Hz, 2H, Ar-H),

7.40–7.48 (m, 2H, Ar-H), 7.17 (d, J = 7.6 Hz, 1H, Ar-H), 7.11 (t, J = 7.6 Hz, 1H, Ar-H), 6.89 (t, J = 7.6 Hz, 1H, Ar-H), 6.75 (d, J = 7.6 Hz, 1H, Ar-H), 5.56 (s, 2H, NH₂), 2.07–2.15 (m, 2H, CH₂), 1.79-1.93 (m, 2H, CH₂), 0.88 (s, 3H, CH₃), 0.81 (s, 3H, CH₃). HRMS found: m/z 488.0843 (M⁺); calcd. for C₂₅H₂₁BrN₄O₂: M, 488.0848. Anal. calcd. for C₂₅H₂₁BrN₄O₂: C, 61.36; H, 4.33; N, 11.45. Found: C, 61.43; H, 4.40; N, 11.52.

2'-Amino-7',7'-dimethyl-1'-(naphthalen-1-yl)-2,5'-dioxo-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4f**)

White solid; mp > 300°C; IR (KBr): ν 3446, 3334, 3221, 2182, 1728, 1661, 1557, 1418, 1242, 1192, 1054, 920, 805, 726 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.28 (br s, 1H, NH), 8.11–8.18 (m, 2H, Ar-H), 7.89–7.94 (m, 2H, Ar-H), 7.67–7.75 (m, 3H, Ar-H), 7.20 (d, *J*=9.6 Hz, 1H, Ar-H), 7.14 (t, *J*=8.8 Hz, 1H, Ar-H), 6.93 (t, *J*=10.0 Hz, 1H, Ar-H), 6.79 (d, *J*=9.6 Hz, 1H, Ar-H), 5.47 (s, 2H, NH₂), 2.12 (d, *J*=23.2 Hz, 1H, CH₂), 1.99 (s, 2H, CH₂), 1.55 (d, *J*=22.8 Hz, 1H, CH₂), 0.79 (s, 3H, CH₃), 0.69 (s, 3H, CH₃). HRMS found: m/z 460.1882 (M⁺); calcd. for C₂₉H₂₄N₄O₂: M, 460.1899. Anal. calcd. for C₂₉H₂₄N₄O₂: C, 75.63; H, 5.25; N, 12.17. Found: C, 75.53; H, 5.30; N, 12.25.

2'-Amino-1'-(4-chlorophenyl)-1,7',7'-trimethyl-2,5'-dioxo-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4g**)

White solid; mp > 300°C; IR (KBr): ν 3463, 3312, 2190, 1716, 1650, 1568, 1491, 1364, 1090, 1018, 915, 753 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.65 (d, *J*=8.4 Hz, 2H, Ar-H), 7.51–7.54 (m, 2H, Ar-H), 7.23 (t, *J*=8.0 Hz, 2H, Ar-H), 6.93–6.70 (m, 2H, Ar-H), 5.55 (s, 2H, NH₂), 3.13 (s, 3H, -NCH₃), 2.05–2.17 (m, 2H, CH₂), 1.81–1.93 (m, 2H, CH₂), 0.89 (s, 3H, CH₃), 0.81 (s, 3H, CH₃). HRMS found: m/z 458.1501 (M⁺); calcd. for C₂₅H₂₃ClN₄O₂: M, 458.1510. Anal. calcd. for C₂₅H₂₃ClN₄O₂: C, 68.04; H, 5.05; N, 12.21. Found: C, 68.13; H, 5.12; N, 12.30.

2'-Amino-1-benzyl-1'-(4-chlorophenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4h**)

White solid; mp > 300°C; IR (KBr): ν 3454, 3365, 2182, 1703, 1644, 1556, 1487, 1362, 1259, 1094, 917, 850, 755 cm⁻¹; ¹H NMR (400 MHz, DMSO*d*₆): δ 7.67 (d, *J*=11.6 Hz, 2H, Ar-H), 7.55 (d, *J*=9.2 Hz, 4H, Ar-H),

New Spiro[indoline-3,4'-quinoline] Derivatives

7.25–7.33 (m, 4H, Ar-H), 7.10 (t, J = 10.4 Hz, 1H, Ar-H), 6.97 (t, J = 9.6 Hz, 1H, Ar-H), 6.61 (d, J = 10.4 Hz, 1H, Ar-H), 5.67 (s, 2H, NH₂), 4.83–4.96 (m, 2H, CH₂), 2.11–2.22 (m, 2H, CH₂), 1.81–1.98 (m, 2H, CH₂), 0.90 (s, 3H, CH₃), 0.82 (s, 3H, CH₃). HRMS found: m/z 534.1831 (M⁺); calcd. for C₃₂H₂₇ClN₄O₂: M, 534.1823. Anal. calcd. for C₃₂H₂₇ClN₄O₂: C, 71.83; H, 5.09; N, 10.47. Found: C, 71.76; H, 5.17; N, 10.35.

2'-Amino-1-benzyl-7',7'-dimethyl-2,5'-dioxo-1'-p-tolyl-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4i**)

White solid; mp > 300°C; IR (KBr): ν 3441, 3330, 2183, 1716, 1644, 1559, 1486, 1362, 1259, 1185, 1052, 917, 751 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 7.55 (d, *J* = 7.6 Hz, 2H, Ar-H), 7.35–7.43 (m, 4H, Ar-H), 7.31 (t, *J* = 7.2 Hz, 2H, Ar-H), 7.25 (d, *J* = 7.6 Hz, 2H, Ar-H), 7.09 (t, *J* = 7.6 Hz, 1H, Ar-H), 6.97 (t, *J* = 7.6 Hz, 1H, Ar-H), 6.61 (d, *J* = 8.0 Hz, 1H, Ar-H), 5.37 (s, 2H, NH₂), 4.85–4.95 (m, 2H, CH₂), 2.42 (s, 3H, CH₃), 2.11–2.19 (m, 2H, CH₂), 1.86–1.98 (m, 2H, CH₂), 0.90 (s, 3H, CH₃), 0.83 (s, 3H, CH₃). HRMS found: m/z 514.2391 (M⁺); calcd. for C₃₃H₃₀N₄O₂: M, 514.2369. Anal. calcd. for C₃₃H₃₀N₄O₂: C, 77.02; H, 5.88; N, 10.89. Found: C, 77.12; H, 5.79; N, 10.95.

2'-Amino-4-bromo-1'-(4-methoxyphenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4j**)

Yellow solid; mp > 300°C; IR (KBr): ν 3457, 3334, 3218, 2184, 1721, 1643, 1512, 1443, 1366, 1250, 1180, 1049, 910, 764 cm⁻¹; ¹H NMR (400 MHz, DMSO- d_6): δ 10.47 (br s, 1H, NH), 7.34–7.36 (m, 1H, Ar-H), 7.26–7.29 (m, 1H, Ar-H), 7.06–7.13 (m, 4H, Ar-H), 6.78–6.80 (m, 1H, Ar-H), 5.48 (s, 2H, NH₂), 3.83 (s, 3H, OCH₃), 2.07–2.17 (m, 2H, CH₂), 1.77–1.95 (m, 2H, CH₂), 0.89 (s, 3H, CH₃), 0.86 (s, 3H, CH₃). HRMS found: m/z 518.0954 (M⁺); calcd. for C₂₆H₂₃BrN₄O₃: M, 518.0930. Anal. calcd. for C₂₆H₂₃BrN₄O₃: C, 60.12; H, 4.46; N, 10.79. Found: C, 60.18; H, 4.51; N, 10.85.

2'-Amino-6-bromo-1'-(4-chlorophenyl)-7',7'-dimethyl-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4**k)

Yellow solid; mp > 300°C; IR (KBr): ν 3462, 3328, 3210, 2186, 1721, 1644, 1560, 1470, 1362, 1261, 1049, 782 cm⁻¹; ¹H NMR (400 MHz, DMSO- d_6): δ 10.40 (br s, 1H, NH), 7.59–7.61 (m, 3H, Ar-H), 7.42–7.54

(m, 2H, Ar-H), 7.08–7.17 (m, 2H, Ar-H), 6.90 (s, 1H, Ar-H), 5.47 (s, 2H, NH₂), 2.07–2.12 (m, 2H, CH₂), 1.78–1.96 (m, 2H, CH₂), 0.86 (s, 3H, CH₃), 0.80 (s, 3H, CH₃). HRMS found: m/z 522.0448 (M⁺); calcd. for $C_{25}H_{20}BrClN_4O_2$: M, 522.0458. Anal. calcd. for $C_{25}H_{20}BrClN_4O_2$: C, 57.32; H, 3.85; N, 10.70. Found: C, 57.28; H, 3.91; N, 10.78.

2'-Amino-6-bromo-7',7'-dimethyl-2,5'-dioxo-1'-p-tolyl-5',6',7',8'tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4**)

Yellow solid; mp > 300°C; IR (KBr): ν 3442, 3323, 3217, 2183, 1729, 1645, 1555, 1477, 1362, 1317, 1258, 1052, 919, 747 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.33 (br s, 1H, NH), 7.40–7.42 (m, 3H, Ar-H), 7.27–7.34 (m, 2H, Ar-H), 7.07–7.14 (m, 2H, Ar-H), 6.90 (s, 1H, Ar-H), 5.36 (s, 2H, NH₂), 2.41 (s, 3H, CH₃), 2.06–2.12 (m, 2H, CH₂), 1.83–1.95 (m, 2H, CH₂), 0.88 (s, 3H, CH₃), 0.82 (s, 3H, CH₃). HRMS found: m/z 502.1022 (M⁺); calcd. for C₂₆H₂₃BrN₄O₂: M, 502.1004. Anal. calcd. for C₂₆H₂₃BrN₄O₂: C, 62.03; H, 4.61; N, 11.13. Found: C, 62.13; H, 4.52; N, 11.21.

2'-Amino-1'-(4-chlorophenyl)-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4m**)

Light yellow solid; mp > 300°C; IR (KBr): ν 3442, 3334, 2190, 1719, 1655, 1561, 1491, 1357, 1200, 1087, 1008, 743 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.21 (br s, 1H, NH), 7.64 (d, *J*=8.8 Hz, 2H, Ar-H), 7.50–7.54 (m, 2H, Ar-H), 7.18 (d, *J*=7.2 Hz, 1H, Ar-H), 7.11 (t, *J*=7.2 Hz, 1H, Ar-H), 6.89 (t, *J*=7.6 Hz, 1H, Ar-H), 6.74 (d, *J*=7.6 Hz, 1H, Ar-H), 5.57 (s, 2H, NH₂), 2.11–2.17 (m, 2H, CH₂), 1.95–2.07 (m, 2H, CH₂), 1.68–1.77 (m, 2H, CH₂). HRMS found: m/z 416.1024 (M⁺); calcd. for C₂₃H₁₇ClN₄O₂: M, 416.1040. Anal. calcd. for C₂₃H₁₇ClN₄O₂: C, 66.27; H, 4.11; N, 13.44. Found: C, 66.38; H, 4.20; N, 13.27.

2'-Amino-1'-(4-bromophenyl)-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4n**)

Yellow solid; mp > 300°C; IR (KBr): ν 3446, 3337, 3288, 2190, 1718, 1653, 1560, 1488, 1357, 1302, 1200, 1013, 743 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.16 (br s, 1H, NH), 7.77 (d, *J* = 8.4 Hz, 2H, Ar-H), 7.45 (d, *J* = 8.0 Hz, 2H, Ar-H), 7.17 (d, *J* = 7.2 Hz, 1H, Ar-H), 7.11 (t, *J* = 7.6 Hz, 1H, Ar-H), 6.89 (t, *J* = 7.2 Hz, 1H, Ar-H), 6.74

(d, J=7.6 Hz, 1H, Ar-H), 5.48 (s, 2H, NH₂), 1.96–2.18 (m, 4H, CH₂), 1.62–1.80 (m, 2H, CH₂). HRMS found: m/z 460.0523 (M⁺); calcd. for C₂₃H₁₇BrN₄O₂: M, 460.0535. Anal. calcd. for C₂₃H₁₇BrN₄O₂: C, 59.88; H, 3.71; N, 12.15. Found: C, 59.98; H, 3.79; N, 12.23.

2'-Amino-1'-(4-methoxyphenyl)-2,5'-dioxo-5',6',7',8'-tetrahydro-1'*H*-spiro[indoline-3,4'-quinoline]-3'-carbonitrile (**4o**)

White solid; mp > 300°C; IR (KBr): ν 3469, 3327, 3206, 2179, 1637, 1510, 1421, 1359, 1252, 1195, 1009, 952, 755 cm⁻¹; ¹H NMR (400 MHz, DMSO-*d*₆): δ 10.15 (br s, 1H, NH), 7.36–7.41 (m, 2H, Ar-H), 7.09–7.17 (m, 4H, Ar-H), 6.89 (t, *J* = 7.6 Hz, 1H, Ar-H), 6.74 (d, *J* = 7.6 Hz, 1H, Ar-H), 5.31 (s, 2H, NH₂), 3.84 (s, 3H, OCH₃), 1.99–2.19 (m, 4H, CH₂), 1.69–1.79 (m, 2H, CH₂). HRMS found: m/z 412.1536 (M⁺); calcd. for C₂₄H₂₀N₄O₃: M, 412.1535. Anal. calcd. for C₂₄H₂₀N₄O₃: C, 69.89; H, 4.89; N, 13.58. Found: C, 69.75; H, 4.79; N, 13.65.

Preparation of Compound 5

A mixture of isatin 1 (2 mmol) and malononitrile 2 (2 mmol) in absolute ethanol (5 mL) was irritated for 2 min at a maximum power of 100 W and 80 °C. Upon completion, monitored by TLC, the reaction mixture was allowed to cool to room temperature and washed with EtOH (95%) to give pure deep red product 5.

2-Oxo-(3H)-indol-3-ylidene Malononitrile (5)

Mp 239–240 °C (lit.^[6c] 239 °C).

ACKNOWLEDGMENTS

This work was partially supported by the Key Laboratory of Organic Synthesis of Jiangsu Province at Suzhou University (No. S8109108), the Natural Science Foundation of Jiangsu Province (No. BK2006048), and the National Science Foundation of China (Nos. 20472062 and 20672079), Nature Science Key Basic Research of Jiangsu Province for Higher Education (Nos. 06KJA15007 and 05KJB150116), Jiangsu Provincial Key Laboratory of Fine Petrochemical Technology (No. KF0402), and a research grant from the Innovation Project for Graduate Students of Jiangsu Province.

REFERENCES

- Zhu, J.; Bienaymé, H. Multicomponent Reactions; Wiley-VCH: Weinheim, 2005.
- (a) Tietze, L. F. Domino reactions in organic-synthesis. *Chem. Rev.* 1996, 96, 115–136;
 (b) Dömling, A.; Ugi, I. Multicomponent reactions with isocyanides. *Angew. Chem., Int. Ed. Engl.* 2000, 39, 3168–3210.
- (a) Loupy, A. *Microwaves in Organic Synthesis*; Wiley-VCH: Weinheim, 2002;
 (b) Kappe, C. O. Controlled microwave heating in modern organic synthesis. *Angew. Chem., Int. Ed. Engl.* 2004, 43, 6250–6284; (c) Dallinger, D.; Kappe, C. O. Microwave-assisted synthesis in water as solvent. Chem. Rev. 2007, 107, 2563–2592.
- (a) Weber, L. Multi-component reactions and evolutionary chemistry. *Drug Discovery Today* 2002, 7, 143–147; (b) Dömling, A. Recent advances in isocyanide-based multicomponent chemistry. *Curr. Opin. Chem. Biol.* 2002, 6, 306–313.
- Da-Silva, J. F. M.; Garden, S. J.; Pinto, A. C. The chemistry of isatins: A review from 1975 to 1999. J. Braz. Chem. Soc. 2001, 12, 273–324.
- (a) Joshi, K. C.; Chand, P. Biologically active indole derivatives. *Pharmazie* 1982, 37, 1–12; (b) Joshi, K. C.; Dandia, A.; Baweja, S.; Joshi, A. Facile microwave-assisted one-pot solid phase synthesis of spiro[3H-indole-3,4'-pyrazolo[3,4-b]pyridines]. *J. Heterocycl. Chem.* 1989, 26, 1097–1099; (c) Jones, G.; Rae, W. J. Knoevenagel condensation products from some cyclic ketones: Structure and stereochemistry. *Tetrahedron* 1966, 22, 3021–3026.
- (a) Abdel-Rahman, A. H.; Keshk, E. M.; Hanna, M. A.; El-Bady, S. M. Synthesis and evaluation of some new spiro indoline-based heterocycles as potentially active antimicrobial agents. *Bioorg. Med. Chem.* 2004, *12*, 2483–2488; (b) Dandia, A.; Singh, R.; Khaturia, S.; Merienne, C.; Morgant, G.; Loupy, A. Efficient microwave enhanced regioselective synthesis of a series of benzimidazolyl/triazolyl spiro[indole-thiazolidinones] as potent antifungal agents and crystal structure of spiro[3*H*-indole-3,2'-thiazolidine]-3'(1,2,4triazol-3-yl)-2,4'(1H)-dione. *Bioorg. Med. Chem.* 2006, *14*, 2409–2417.
- (a) Ma, Z.; Hano, Y.; Nomura, T.; Chen, Y. Novel quinazoline-quinoline alkaloids with cytotoxic and DNA topoisomerase II inhibitory activities. *Bioorg. Med. Chem. Lett.* 2004, 14, 1193–1196; (b) Denton, T. T.; Zhang, X.; Cashman, J. R. 5-Substituted, 6-substituted, and unsubstituted 3-heteroaromatic pyridine analogues of nicotine as selective inhibitors of cytochrome P-450 2A6. J. Med. Chem. 2005, 48, 224–239.
- 9. (a) Wang, S. Y.; Ji, S. J. Facile synthesis of 3,3-di(heteroaryl)indolin-2-one derivatives catalyzed by ceric ammonium nitrate (CAN) under ultrasound irradiation. *Tetrahedron* 2006, *62*, 1527–1535; (b) Zhu, S. L.; Ji, S. J.; Zhang, Y. A simple and clean procedure for three-component synthesis of spirooxindoles in aqueous medium. *Tetrahedron* 2007, *63*, 9365–9372; (c) Zhu, S. L.; Ji, S. J.; Su, X. M.; Sun, C.; Liu, Y. Facile and efficient synthesis of a new class of bis(3'-indolyl)pyridine derivatives via one-pot multicomponent reactions. *Tetrahedron Lett.* 2008, *49*, 1777–1781; (d) Gu, D. G.; Ji, S. J.; Wang, H. X.; Xu, Q. Y. Acidic ionic liquid-catalyzed highly efficient reaction of indoles to α,β-unsaturated ketones. *Synth. Commun.* 2008, *38*, 1212–1223.