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Letter

Organocatalytic Atroposelective Friedländer Quinoline Heteroannulation

You-Dong Shao, Meng-Meng Dong, You-An Wang, Pei-Ming Cheng, Tao Wang, and Dao-Juan Cheng*[©]

Department of Chemistry and Chemical Engineering, Heze University, Heze 274015, People's Republic of China

Supporting Information

ABSTRACT: An atroposelective Friedländer heteroannulation reaction of 2-aminoaryl ketones with α -methylene carbonyl derivatives has been developed for the first time with chiral phosphoric acid as an efficient organocatalyst. The desired enantioenriched axially chiral polysubstituted 4-arylquinoline architectures were prepared with good to high yields and enantioselectivities (up to 94% yield and up to 97% ee). Furthermore, the products can be readily derivatized to afford an



array of new quinoline-containing heteroatropisomers, which hold great potential in asymmetric catalysis and drug discovery.

Quinolines and their derivatives have been tagged as "Privileged Scaffolds" owing to their widespread appearance in natural and synthetic products that exhibit remarkable pharmacological or physical properties (Figure 1).^{1–3} While numerous synthetic methods have been



Figure 1. Selected privileged quinoline-containing natural product, bioactive molecules, and chiral ligand.

developed toward their synthesis, the Friedländer heteroannulation reaction, discovered over 135 years ago, is still one of the most straightforward approaches to access polysubstituted quinolines.⁴ This transformation is generally accomplished through a condensation/cyclodehydration sequence between 2-aminosubstituted aromatic carbonyl compounds and α -methylene carbonyl derivatives in the presence of acidic or basic catalysts.

In recent decades, organocatalysis has emerged as an important tool in organic synthesis.⁵ In this context, several

organocatalytic Friedländer reactions for the preparation of quinolines with remote stereogenic centers have been disclosed.^{6–8} On the other hand, axially chiral biaryl skeletons are frequently encountered in many natural products, bioactive compounds, and privileged chiral ligands and catalysts.⁹ Besides, atropisomerism plays an important role in controlling the pharmacological properties of biological systems.¹⁰ Accordingly, considerable efforts have been devoted to the search of efficient routes for the enantioselective construction of various atropisomeric aryl–aryl or aryl-heteroaryl structures.¹¹

Despite such significant advances, the enantioselective synthesis of axially chiral quinoline-containing biaryl backbones, especially those by organocatalysis, remains rare. In 2016, the Zhou group realized an efficient kinetic resolution of axially chiral 5- or 8-substituted quinoline derivatives via chiral phosphoric acid (CPA) promoted asymmetric transfer hydrogenation (Scheme 1a).¹² Later, Matsubara, Asano and coworkers developed a bifunctional organocatalytic aromatic electrophilic halogenation reaction of prochiral 3-(quinolin-8yl)phenols to access 8-arylquinoline atropisomers (Scheme 1b).¹³ Surprisingly, the atroposelective formation of enantioenriched 4-arylquinoline derivatives in an organocatalytic and atom-efficient fashion from simple achiral substrates is almost unexplored to date.¹⁴ Notably, 4-arylquinoline derivatives show a wide spectrum of bioactivities. For example, the HIV integrase inhibitor BI 224436 bearing a rotation inhibited axis is being investigated to target the LEDGF/p75 binding site¹⁵ and has recently been the first noncatalytic site integrase inhibitor (NCINI) to advance into a phase Ia clinical trial.¹⁶ The dibenzo [c, f] [2,7] naphthyridine derivative represents a potent phosphoinositide-dependent kinase-1 (PDK-1) inhib-

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Scheme 1. Synthesis of Quinoline-Containing Heterobiaryl Atropisomers



itor.¹⁷ Pitavastatin is a well tolerated and effective alternative treatment for patients with hypercholesterolemia.¹⁸ Even the structurally simple 4-phenyl chloroquinoline derivative **CPQE** was demonstrated to be a promising antioxidant and antidiabetic agent recently (Figure 1).¹⁹ Therefore, the design and atroposelective construction of a new family of 4-arylquinolines is highly appealing, yet still challenging. We posited that catalytic asymmetric Friedländer heteroannulation reaction might allow a straightforward access to these valuable heterobiaryl atropisomers.

For a typical Friedländer heteroannulation, the 2,3substituted 4-aryl quinolines produced from corresponding 2aminoaryl ketones do not possess axial chirality because of the low rotational barrier between the two aromatic rings (Scheme 1c). We envisioned that if Ar was an aryl bearing orthosubstituents that could hinder the free rotation of the newly formed arvl-quinoline bond, the resulting 4-arvlquinolines would be axially chiral. In order to reach this goal, the first task we encountered is selection of a suitable chiral catalyst prompt to efficiently promote the cascade process and induce axial enantiocontrol. Guided by our ongoing interest in Brønsted acid catalysis,²⁰ we designed a chiral phosphoric acid (CPA)²¹ catalyzed asymmetric Friedländer heteroannulation reaction, which provided an efficient route for the synthesis of atropoisomeric 4-arylquinoline structures with good to high yields and enantioselectivities (Scheme 1d).

To validate our hypothesis, we first conducted the Friedländer heteroannulation reaction of (2-aminophenyl)-(naphthalen-1-yl)methanone **1aa** with ethyl acetoacetate **2a** in the presence of 10 mol % of BINOL-derived CPA (R)-**C1** in chloroform (CHCl₃) at 70 °C. After a 48 h reaction time, the desired axially chiral trisubstituted 4-aryquinoline **3aa** was obtained in 54% yield with 5% ee (Table 1, entry 1). To improve the results, a series of chiral phosphoric acids with different substituents and backbones were assessed (Table 1, entries 2–10). Among them, SPINOL-derived catalyst (R)-**C6** with a phenanthryl group was found to be the best in terms of enantioselectivity (79% yield, 72% ee, Table 1, entry 6).



^{*a*}Unless otherwise stated, all reactions were carried out with 2aminoaryl ketones 1 (0.10 mmol), ethyl acetoacetate **2a** (0.30 mmol), catalyst CPA (10 mol %) and solvent (1.5 mL) in sealed tube at 70 °C for 48 h. ^{*b*}Isolated yield. ^cDetermined by chiral stationary phase HPLC analysis. ^{*d*}The reaction was run at 50 °C for 5 d. ^{*e*}The reaction was run at 100 °C. ^{*f*}5 Å molecular sieves (150 mg) were added. ^{*g*}Glycine *tert*-butyl ester (10 mol %) was added. ^{*h*}The reaction was run for 24 h.

Subsequently, various solvents were examined (Table 1, entries 11-14). Better results were observed with halogenated solvents, and reactions in CHCl₃ gave slightly higher enantioselectivity.

We also studied the temperature profile of this reaction and found that the atroposelectivity was slightly diminished at both higher and lower temperatures (Table 1, entries 15–16; for details on the catalyst and temperature screening, see Supplementary Table S1). Further investigations on the additives revealed that the addition of 10 mol % of a primary amine, namely glycine *tert*-butyl ester, together with 5 Å molecular sieves led to an evident acceleration on the reaction rate without any negative effect on the enantioselectivity (92%

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yield, 74% ee, Table 1, entry 17; for details on the additives screening, see Supplementary Table S2).

In view of the moderate result of the above model reaction of (2-aminophenyl)(naphthalen-1-yl)methanone **1aa** and the significant influence of the *ortho*-substituents of the 4-aryl group on the stereocontrol for the newly formed aryl-quinoline bond, we exchanged the Ar group of 2-aminoaryl ketone from 1-naphthyl (**1aa**) to 2-methylphenyl (**1ba**). Gratifyingly, the Friedländer heteroannulation of **1ba** with ethyl acetoacetate **2a** proceeded smoothly under the established optimal conditions and afforded the axially chiral 4-aryquinoline **3ba** in high yield (89%) with improved enantioselectivity (81% ee). Replacing the *o*-Me group with *i*Pr or CF₃ completely shut down the reaction, whereas the others (Ph, Cl, Br, and CO₂Et) led to poorer enantioselectivity, indicating that the functional group *ortho-* to the newly formed atropisomeric axis has a decisive effect on the reaction outcome.²²

Continuing with the 4-(*o*-tolyl)quinoline scaffold, we sought out to explore a range of substitutions on the *o*-methyl phenyl ring (Scheme 2). In general, adding substitution with different





^{*a*}All reactions were carried out with 2-aminoaryl ketones **1b** (0.10 mmol), ethyl acetoacetate **2a** (0.30 mmol), catalyst (*R*)-C6 (10 mol %), glycine *tert*-butyl ester (10 mol %), and CHCl₃ (1.5 mL) in a sealed tube at 70 °C for 48 h. Yields refer to isolated pure compounds. The ee was determined by chiral stationary phase HPLC analysis.

electronic properties at the *ortho-*, *meta-*, and *para-* positions of the 2-methyl group had little effect on the reaction efficiency, giving **3ba-3bk** in good yields ranging from 70% to 89% and moderate to high ee values (71–91%). What is worthy of note is that incorporation of additional groups *para-* to the 2-methyl group resulted in an evident increase in the enantioselectivity

(3be-3bj), and best results were obtained with a trifluoromethyl (3bi, 90% ee) or ester substituent (3bj, 91% ee).

Next, we evaluated the use of various aromatic amines (Scheme 3) and found that the atroposelectivity of this





^{*a*}Unless otherwise stated, all reactions were carried out with 2aminoaryl ketones 1 (0.10 mmol), α -methylene carbonyl derivatives 2 (0.30 mmol), catalyst (*R*)-C6 (10 mol %), glycine *tert*-butyl ester (10 mol %), and CHCl₃ (1.5 mL) in sealed tube at 70 °C for 48 h. Yields refer to isolated pure compounds. The ee was determined by chiral stationary phase HPLC analysis. ^{*b*}The reaction was performed at 1.0 mmol scale. ^{*c*}The reaction was run at 90 °C for 96 h.

catalytic asymmetric Friedländer reaction was very sensitive to substitutions on the phenyl ring of the newly formed quinoline. Compared with other positions (6- and 8-), higher enantioselectivities were obtained with 7-chloro (3bo, 97% ee) followed by 7-methyl (3bm, 96% ee) and 7-methoxy (3bn, 94% ee) substituted 4-arylquinolines. 2-Aminonaphthyl ketone was also a competent substrate, furnishing axially chiral benzo[g]quinoline derivatives with high optical purity (3br, 93% ee and 3bs, 91% ee). At this stage, we deduced that installation of an appropriate substituent at the 7-position of the newly formed quinolines might be able to improve the enantioselectivity of other 4-arylquinolines. Pleasingly, adding a chloro-substituent to the initial model substrate 1aa afforded significantly improved enantioselectivity (3ab, 97% ee vs 3aa, 74% ee). A similar result was observed with 4-naphthylbenzo-[g]quinoline 3ac (94%, 89% ee). Notably, these selectivities were conserved when different o-halogenated substrates were used, as exemplified by 3c and 3d, which contain a handle for the postfunctionalization. It bears mentioning that the

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importance of the C7 position in 4-arylquinoline based drug candidates on the serum shifted potency has been established recently.¹⁶

Further efforts were focused toward testing the compatibility of various α -methylene carbonyl derivatives 2. The transformation proceeded equally well with different alkyl acetoacetates (3ad, 95% ee and 3ae, 96% ee). Variation of the R^2 group had a very strong influence on the reaction results. Ethyl propionylacetate smoothly underwent the reaction and provided product 3bt in good yield and selectivity (80% ee). However, the reaction turned out to be very sluggish for sterically demanding β -keto esters. An elevated reaction temperature was required to achieve a better conversion. For example, the 2-isopropyl 4-arylquinoline 3af was obtained in an acceptable yield with a high ee value of 90% after being heated at 90 °C for 4 days. However, utilizing ethyl benzoylacetate resulted in a dramatic drop in ee value to 37% (3ag). Besides, several 1,3-diones, including acetylacetone, heptane-3,5-dione, and cyclohexane-1,3-dione, were all tolerated in the catalytic process and afforded axially chiral quinolines 3bu-3bw with good to high stereoselectivities (76-92% ee). The absolute configuration of 3ab was assigned to be (aS) by single-crystal X-ray analysis,²³ and that of the remaining products were assigned by analogy. We also tested the configurational stability of the product by heating a solution of (aS)-3ab in toluene at 100 °C for 48 h. Chiral HPLC analysis showed that the ee was unaffected (see the Supporting Information for details).

Furthermore, the products can be readily derivatized to afford an array of new quinoline-containing heteroatropisomers (Scheme 4). The ester group in the chiral product **3ab** could



be readily reduced by LiAlH_4 at -20 °C or hydrolyzed with NaOH in refluxed ethanol to produce the corresponding axially chiral primary alcohol 4 and carboxylic acid 5 in high yields, respectively, without the loss of enantiopurity. Treatment of **3bi** with *m*-CPBA led to a quantitative formation of the quinoline *N*-oxide 6 with retained optical integrity.

Two fundamentally different mechanisms exist for the classical Friedländer reaction. While most of the evidence favored the pathway involving the initial formation of the Schiff base intermediate,²⁴ a recent report has summarized that Friedländer reaction was initiated with an intermolecular aldol reaction under the typically used acidic or basic conditions.²⁵ Fortunately, an evident byproduct was observed in the reaction of utilizing 1,3-cyclohexanedione **2i** as a substrate, which was later demonstrated to be the enamine structure 7**bw**.²⁶ When 7**bw** was subjected to the standard protocol (without glycine

tert-butyl ester additive), the axially chiral benzo[g]quinoline **3bw** could be isolated in quantitative yield with 83% ee after 36 h (Scheme 5).

Scheme 5. Isolation of Intermediate and Control Experiment



On the basis of the above experimental results, we propose the following catalytic process (Scheme 6). The first step is

Scheme 6. Proposed Catalytic Cycle



condensation of the primary amine group in 2-aminoaryl ketones 1 and 2 under promotion of CPA, leading to imine intermediates 8, which then tautomerize to enamines 7. Then, the intramolecular aldol reaction causes ring closure and gives compounds 9, which finally lose water and furnish expected quinoline products 3. It is obvious that CPA played an important role in the asymmetric induction by creating a suitable chiral environment in the cyclization step through cooperative hydrogen bonding interactions. The observation that adding a catalytic amount of achiral primary amine, such as glycine *tert*-butyl ester, could accelerate the reaction to a certain extent might be ascribed to the formation of imines 10,²⁶ which was beneficial for the initial condensation to take place.

In summary, we have developed the first organocatalytic atroposelective Friedländer heteroannulation reaction to assemble stereochemically stable enantioenriched polysubstituted 4-arylquinolines in high yields with good to excellent enantioselectivities (up to 97% ee). The current study not only expands the existing state of the art of *N*-heteroatropisomers construction but also provides a useful avenue to access axially chiral polysubstituted 4-arylquinoline scaffolds which might be of significance for future biological and medicinal studies.

ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: 10.1021/acs.or-glett.9b01731.

Experimental procedures, NMR spectra, HPLC traces, X-ray and analytical data (PDF)

Accession Codes

CCDC 1906249 contains the supplementary crystallographic data for this paper. These data can be obtained free of charge via www.ccdc.cam.ac.uk/data_request/cif, or by emailing data_request@ccdc.cam.ac.uk, or by contacting The Cambridge Crystallographic Data Centre, 12 Union Road, Cambridge CB2 1EZ, UK; fax: +44 1223 336033.

AUTHOR INFORMATION

Corresponding Author

*E-mail: chengdaojuan0614@163.com.

ORCID ©

Dao-Juan Cheng: 0000-0001-7834-7563

Notes

The authors declare no competing financial interest.

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(22) Besides, different alkyl acetoacetates were tested and best results were obtained with the ethyl group (for details, see Supplementary Table S3).

(23) See the Supporting Information for details.

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(26) Additionally, ESI-mass spectroscopic analysis of the reaction mixture also allowed us to identify tentatively similar key acyclic intermediate. For details on the mechanistic studies, please see the Supporting Information.