

Efficient Synthesis of Some Novel Macrocyclic Diamides Using Fast Addition Method

Hashem Sharghi,* Abdolkarim Zare

Department of Chemistry, College of Sciences, Shiraz University, Shiraz 71454, Iran
Fax +98(711)2280926; E-mail: shashem@chem.susc.ac.ir

Received 15 September 2005

Abstract: Synthesis of some new macrocyclic diamides based on catechol scaffold by cyclization reactions between various diamines and 2-[2-(2-chloro-2-oxoethoxy)phenoxy]ethanoyl chloride using fast addition method has been described. The reactions were carried out in short reaction times and the expected macrocycles were obtained in good to high yields.

Key words: macrocyclic diamide, fast addition, catechol, diamine, dicarboxylic acid dichloride, cyclization

Macrocyclic compounds with amide units in the macro ring are of interest because they have important applications in selective noble metal complexing agents and metal ion-selective electrodes.¹ For example, some of these compounds were shown to be selective for the complexation with Hg²⁺,^{1a} Zn²⁺,^{1b} Cu²⁺,^{1c} and Sr²⁺ ions.^{1d} Furthermore, recently we have shown that these macrocycles can efficiently catalyze the regioselective ring opening of epoxides with elemental halogen^{2a,b} or ammonium thiocyanate.^{2c} Interestingly, some macrocyclic amides are receptors for anions such as OAc⁻, H₂PO₄⁻, F⁻ and Cl⁻.³ In this case, the electronegative atom from anions (oxygen, fluorine or chlorine) interact with NH groups of macrocycles.³ Lactams also have been used as starting materials for the preparation of polyaza-crowns and related compounds such as cryptands.⁴ Although macrocyclic diamides have many applications, only a few methods have been developed for their preparation. In these methods, carboxylic acid derivatives such as dicarboxylic acid diesters,⁵ labile dicarboxylic acid dichlorides⁶ or bis(α -chloroamide) compounds^{4a} were allowed to react with diamines under different conditions. The more common methods involve high dilution,⁷ template effect,⁸ high pressure approach,⁹ ESI-MS¹⁰ and ‘Crab-like’^{4d} techniques. Most of these methods are associated with several drawbacks such as long reaction times, low yields of products, harsh reaction conditions and difficulty in separation and purification.

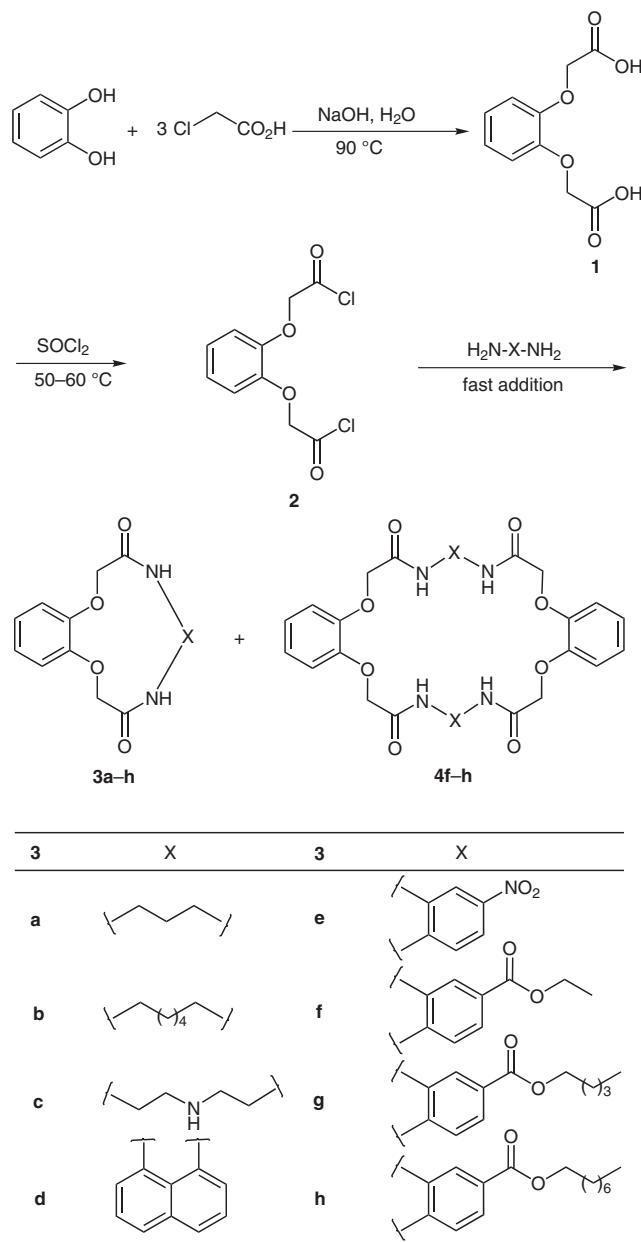
Recently, we provided a synthetic route toward dilactam derivatives.¹¹ In this method, the cyclization reaction does not require high dilution techniques or template effect. The advantages of this method (fast addition) are high yield, short reaction time, ease of purification, low vol-

ume of solvent, no side reactions and synthetic versatility (this method allows to provide various even or odd membered dilactams). In fact, because of high significance of these compounds and also in extension of the applicability of fast addition method, herein, we would like to report an efficient synthesis of some new macrocyclic diamides based on catechol structure using this method (Scheme 1).

Catechol was reacted with 2-chloroacetic acid in water in the presence of sodium hydroxide to give 2-[2-(2-hydroxy-2-oxoethoxy)phenoxy]acetic acid (**1**) as a white solid in 91% yield (Scheme 1). Treatment of **1** with thionyl chloride gave 85% yield of 2-[2-(2-chloro-2-oxoethoxy)phenoxy]ethanoyl chloride (**2**) (Scheme 1). The cyclization reactions between dicarboxylic acid dichloride **2** and different diamines were performed by fast addition procedure without high dilution technique. This step was carried out by fast addition of diamines in anhydrous solvent into a solution of dicarboxylic acid dichloride **2** in anhydrous solvent over a few seconds with vigorous stirring at room temperature.

Macrocycles **3a** and **3b** were synthesized to study the effect of alkyl chain length of aliphatic diamines on results of cyclization reactions. As shown in Table 1, 1,6-diaminohexane gave higher yield than 1,3-diaminopropane (Table 1, entries 1 and 2). This aspect indicates the dependency of macrocyclization yield to the ring size of the macrocycle. We have also extended the cyclization reaction to aromatic diamines. As Table 1 indicates, when aromatic diamines were used instead of aliphatic diamines, the reaction yields were higher. Moreover, the presence of CO₂R as functional group on the ring of aromatic diamines afforded the macrocyclic diamides in higher yields as well as shorter reaction times in comparison to the NO₂ (Table 1, entries 5, 6, 8 and 10). In the synthesis of compounds **3f–h**, two-to-two cyclization products **4f–h** were also obtained in low yields, and were easily separated from one-to-one adducts (Table 1, entries 7, 9 and 11).

We have also studied the effect of alkyl chain length of ester moiety (CO₂R) present on aromatic diamines on the yields and the solubility of products in organic solvents. The results have illustrated that increment of the length of alkyl chain decreased the yields of diamides and increased the yields of tetraamides (Table 1, entries 6–11). This aspect enhanced the solubility of dilactams and tetralactams in organic solvents. The useful solvents for synthesis of macrocycles **3a–h** and **4f–h** were ethyl acetate, dichlo-



Scheme 1

romethane, acetone, acetonitrile, THF, diethyl ether and ethyl acetate–dichloromethane (1:1). In all reactions, chloroform, DMSO and DMF were found to be unsuitable as solvents.

In conclusion, we have prepared some novel macrocyclic dilactams and tetralactams by cyclization reactions between dicarboxylic acid dichloride and different diamines (aliphatic and aromatic) using fast addition technique. Furthermore, these macrocycles with different ring sizes were obtained in high yields and short reaction times. The applications of these compounds in organic and analytical chemistry are currently under investigation.

All chemicals were obtained from Merck or Fluka. Solvents were purified and dried according to reported methods and stored over molecular sieves.¹² The progress of reactions was followed with

TLC using silica gel SILG/UV 254 plates. Chromatography was carried out on a column over silica gel 60, 0.063–0.200 mm (70–230 mesh ASTM). IR spectra were run on a Shimadzu FTIR-8300 spectrophotometer. The ¹H (250 MHz) and ¹³C NMR (62.5 MHz) were run on a Bruker Avance DPX-250, FT-NMR spectrometer (δ in ppm, J in Hz). Mass spectra were recorded on a Shimadzu GC MS-QP 1000 EX apparatus. Microanalyses were performed on a Perkin-Elmer 240-B microanalyzer. Melting points were recorded on a Büchi B-545 apparatus in open capillary tubes and are uncorrected.

2-[2-(2-Hydroxy-2-oxoethoxy)phenoxy]acetic Acid (1)

To a mixture of catechol (11.01 g, 100 mmol) and 2-chloroacetic acid (28.35 g, 300 mmol) at 90 °C on a water-bath, was added dropwise a solution of aq 33% (w/v) NaOH solution (48.02 g in 97.5 mL H₂O). The mixture was stirred at 90 °C for 2 h and after cooling to r.t., the solution was kept in ice bath. Then, concentrated HCl was added dropwise to the solution with stirring. The mixture was allowed to warm to r.t. and the solution was filtered and the white precipitate was washed with cold H₂O. Crystallization of the residue from hot water afforded **1** as white crystals; yield: 20.58 g (91%); mp 176–177 °C (Lit.¹³ mp 177–178 °C).

2-[2-(2-Chloro-2-oxoethoxy)phenoxy]ethanoyl Chloride (2)

The acid **1** (5.65 g, 25 mmol) was heated in SOCl₂ (50 mL) for 4 h at 50–60 °C. The SOCl₂ was evaporated under vacuo at low temperature (40 °C) and the residue was crystallized from petroleum ether (bp 60–80 °C) to give **2** as white crystals; yield: 5.58 g (85%); mp 48–49 °C.

IR (KBr): 3075, 2977, 1805 (s), 1501 cm⁻¹.

¹H NMR (CDCl₃, 250 MHz): δ = 4.96 (s, 4 H), 6.78–6.88 (complex, 4 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 73.3, 116.7, 123.5, 146.5, 170.6.

Ethyl, Pentyl and Octyl 3,4-Diaminobenzoate; General Procedure

The appropriate alcohol (2.0 mmol) and carboxylic acid (2.0 mmol) were added to a stirred mixture of Al₂O₃ (acidic type 540 C, 0.31 g, 3.0 mmol) and MeSO₃H (1 mL). The mixture was allowed to stir in an oil bath at 80 °C for 2–3 h, then poured into H₂O (20 mL), and extracted with CHCl₃ (50 mL). The organic layer was washed with sat. aq solution of NaHCO₃ (2 × 20 mL), dried (CaCl₂), and evaporated to give the corresponding ester.¹⁴

Ethyl 3,4-Diaminobenzoate

Pale yellow solid; yield: 0.32 g (89%); mp 79–80 °C.

IR (KBr): 3327, 3188, 3048, 2959, 1686 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 1.25 (t, 3 H, J = 6.9 Hz), 3.52 (s, 4 H), 4.24 (q, 2 H, J = 6.9 Hz), 6.60 (d, 1 H, J = 8.1 Hz), 7.33–7.41 (complex, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 14.8, 60.8, 115.2, 118.6, 121.7, 123.5, 133.5, 140.7, 167.3.

MS: *m/z* (%) = 181 (M⁺ + 1, 18.6), 180 (M⁺, 97.1), 152 (57.0), 151 (6.4), 135 (100), 107 (45.5), 75 (1.4), 73 (1.7), 45 (5.6), 43 (51.2).

Pentyl 3,4-Diaminobenzoate

Pale yellow solid; yield: 0.38 (85%); mp 65–67 °C.

IR (KBr): 3331, 3190, 3050, 2950, 1685 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 0.94–1.02 (complex, 5 H), 1.54–1.72 (complex, 4 H), 3.63 (s, 4 H), 4.28 (t, 2 H, J = 7.0 Hz), 6.66 (d, 1 H, J = 8.2 Hz), 7.40–7.47 (complex, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 16.9, 26.6, 30.1, 37.9, 63.5, 115.3, 118.7, 121.7, 123.6, 133.3, 140.8, 167.4.

Table 1 Macrocyclization Reactions between Dicarboxylic Acid Dichloride **2** and Various Diamines Using Fast Addition Method

Entry	Diamine	Product	Reaction time (min)	Solvent	Yield (%) ^a
1		3a	10	acetone	31
2		3b	5	acetone	70
3		3c	5	EtOAc	45
4		3d	3	EtOAc	75
5		3e	5	CH2Cl2	69
6		3f	3	CH2Cl2	89
7		4f	3	CH2Cl2	7
8		3g	3	CH2Cl2	82
9		4g	3	CH2Cl2	11
10		3h	3	CH2Cl2	74
11		4h	3	CH2Cl2	16

^a Isolated yields.

MS: *m/z* (%) = 223 ($M^+ + 1$, 15.2), 222 (M^+ , 41.5), 194 (28.3), 166 (14.1), 151 (2.5), 135 (100), 107 (25.6), 71 (4.9), 57 (38.5), 43 (46.1).

Octyl 3,4-Diaminobenzoate

Pale yellow solid; yield: 0.44 g (83%); mp 56–58 °C.

IR (KBr): 3329, 3191, 3054, 2955, 1685 cm^{-1} (s).

^1H NMR (CDCl_3 , 250 MHz): δ = 0.88 (t, 3 H, J = 6.8 Hz), 1.25–1.34 (complex, 10 H), 1.70 (m, 2 H), 3.59 (s, 4 H), 4.24 (t, 2 H, J = 7.0 Hz), 6.68 (d, 1 H, J = 8.3 Hz), 7.44–7.51 (complex, 2 H).

^{13}C NMR (CDCl_3 , 62.5 MHz): δ = 14.5, 23.0, 26.5, 29.2, 29.6, 29.7, 32.2, 65.1, 115.3, 118.8, 121.8, 123.7, 133.3, 140.7, 167.3.

MS: *m/z* (%) = 265 ($M^+ + 1$, 6.7), 264 (M^+ , 31.6), 165 (1.3), 151 (3.3), 135 (40.6), 107 (19.0), 84 (19.7), 70 (31.8), 56 (60.0), 43 (78.3), 41 (100).

Dilactams **3a–h** and **4f–h**; General Procedure

A solution of diamine (2.0 mmol) in anhyd solvent (50 mL) was added quickly to a vigorous stirring solution of dicarboxylic acid dichloride (2.0 mmol) in anhyd solvent (50 mL) at r.t. The mixture was stirred for a further 3–10 min (see Table 1 for solvent used and time of reaction). The organic layer was washed first with aq NaHCO_3 solution (2 × 50 mL), then with H_2O (2 × 50 mL), and dried (MgSO_4). The crude product obtained by evaporation of the solvent was purified by column chromatography over silica gel using *n*-hexane–EtOAc as eluent (15:1 to 10:1).

5,6,7,8-Tetrahydro-2*H*-1,11,4,8-benzodioxadiazacyclotri-decine-3,9(4*H*,10*H*)-dione (**3a**)

Colorless crystals; mp 178–179 °C.

IR (KBr): 3334, 3078, 2956, 1647 cm^{-1} (s).

¹H NMR (CDCl₃, 250 MHz): δ = 1.92 (t, 2 H, J = 5.9 Hz), 3.63 (t, 4 H, J = 5.9 Hz), 4.47 (s, 4 H), 6.98–7.09 (complex, 4 H), 7.41 (s, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 28.1, 40.1, 71.4, 115.4, 124.5, 149.2, 168.5.

MS: m/z (%) = 265 (M⁺ + 1, 11.7), 264 (M⁺, 11.9), 150 (9.9), 142 (3.9), 122 (12.5), 114 (4.6), 100 (3.7), 99 (8.9), 92 (8.4), 76 (7.2), 70 (9.1), 57 (17.3), 52 (45.5), 42 (100).

Anal. Calcd for C₁₃H₁₆N₂O₄ (264.28): C, 59.08; H, 6.10; N, 10.60. Found: C, 59.01; H, 6.19; N, 10.51.

4,5,6,7,8,9,10,11-Octahydro-1,14,4,11-benzodioxazacyclohexadecine-3,12(2H,13H)-dione (3b)

Colorless crystals; mp 199–201 °C.

IR (KBr): 3342, 3069, 2948, 1649 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 1.52–1.69 (complex, 8 H), 3.49 (t, 4 H, J = 5.7 Hz), 4.52 (s, 4 H), 6.86–7.04 (complex, 4 H), 7.54 (s, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 24.5, 30.1, 41.0, 70.7, 118.2, 124.6, 149.3, 168.8.

MS: m/z (%) = 307 (M⁺ + 1, 7.0), 306 (M⁺, 30.3), 248 (5.5), 236 (4.3), 184 (4.2), 150 (9.3), 137 (44.8), 121 (45.7), 114 (15.8), 107 (15.9), 97 (38.6), 43 (100).

Anal. Calcd for C₁₆H₂₂N₂O₄ (306.36): C, 62.73; H, 7.24; N, 9.14. Found: C, 62.66; H, 7.19; N, 9.02.

5,6,7,8,9,10-Hexahydro-2H-1,13,4,7,10-benzodioxatriazacyclopentadecine-3,11(4H,12H)-dione (3c)

Colorless crystals; mp 230 °C (dec.).

IR (KBr): 3360, 3244, 3080, 2979, 1676 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 1.75 (s, 1 H), 2.94 (t, 4 H, J = 5.3 Hz), 3.48 (t, 4 H, J = 5.3 Hz), 4.48 (s, 4 H), 6.87–6.98 (complex, 4 H), 7.81 (s, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 38.5, 47.8, 67.2, 112.6, 122.4, 146.5, 167.5.

MS: m/z (%) = 294 (M⁺ + 1, 73.3), 293 (M⁺, 12.0), 278 (1.1), 250 (3.7), 192 (1.3), 150 (4.1), 121 (19.8), 108 (1.2), 100 (8.2), 99 (14.1), 92 (5.1), 86 (8.9), 85 (32.0), 71 (20.3), 57 (29.5), 52 (19.9), 44 (85.5), 43 (100), 42 (84.4).

Anal. Calcd for C₁₄H₁₉N₃O₄ (293.32): C, 57.33; H, 6.53; N, 14.33. Found: C, 57.42, H, 6.69; N, 14.41.

7H,16H-Naphtho[1,8-ef][1,11,4,8]benzodioxazacyclotri-decine-8,17(9H,18H)-dione (3d)

Colorless crystals; mp 276 °C (dec.).

IR (KBr): 3294, 3053, 2922, 1661 cm⁻¹ (s).

¹H NMR (250 MHz, DMSO-*d*₆): δ = 4.68 (s, 4 H), 7.12 (d, 2 H, J = 7.5 Hz), 7.21 (dd, 2 H, J = 5.7, 7.5 Hz), 7.58 (dd, 2 H, J = 6.8, 7.9 Hz), 7.86 (d, 2 H, J = 6.8 Hz), 7.98 (d, 2 H, J = 7.9 Hz), 11.06 (s, 2 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 70.5, 118.2, 122.1, 123.0, 123.5, 125.7, 126.8, 132.4, 135.5, 148.4, 167.9.

MS: m/z (%) = 349 (M⁺ + 1, 25.3), 348 (M⁺, 100), 272 (2.6), 198 (13.7), 184 (60.5), 169 (58.2), 154 (8.9), 141 (4.5), 122 (8.9), 108 (1.9), 92 (6.9), 76 (15.4), 52 (54.9).

Anal. Calcd for C₂₀H₁₆N₂O₄ (348.35): C, 68.96; H, 4.63; N, 8.04. Found: C, 68.82; H, 4.50; N, 7.93.

10-Nitro-8,13-dihydribenzo[b,h][1,4,7,10]dioxadiazacyclododecene-7,14(6H,15H)-dione (3e)

Colorless crystals; mp 252–253 °C (dec.).

IR (KBr): 3348, 3122, 3063, 2960, 1697 (s), 1676 cm⁻¹ (s).

¹H NMR (DMSO-*d*₆, 250 MHz): δ = 4.82 (s, 2 H), 4.88 (s, 2 H), 7.04 (d, 2 H, J = 7.7 Hz), 7.32 (dd, 2 H, J = 6.1, 7.7 Hz), 8.10–8.18 (complex, 2 H), 8.35 (s, 1 H), 9.92 (s, 1 H), 10.17 (s, 1 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 70.9, 71.9, 117.3, 118.0, 120.4, 121.0, 122.1, 122.4, 123.3, 129.0, 137.2, 143.7, 148.4, 148.6, 167.3, 167.6.

MS: m/z (%) = 344 (M⁺ + 1, 11.8), 343 (M⁺, 25.8), 193 (5.3), 192 (14.0), 179 (8.0), 164 (100), 149 (8.4), 122 (26.4), 121 (55.8), 108 (7.0), 92 (11.4), 76 (14.1), 52 (55.2).

Anal. Calcd for C₁₆H₁₃N₃O₆ (343.29): C, 55.98; H, 3.82; N, 12.24. Found: C, 56.11; H, 3.76; N, 12.33.

Ethyl 7,14-Dioxo-6,7,8,13,14,15-hexahydrodibenzo[b,h][1,4,7,10]dioxadiazacyclododecene-10-carboxylate (3f)

Colorless crystals; mp 186–187 °C.

IR (KBr): 3364, 3242, 3050, 2980, 1721 (s), 1699 (s), 1653 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 1.35 (t, 3 H, J = 7.0 Hz), 4.36 (q, 2 H, J = 7.0 Hz), 4.60 (s, 2 H), 4.65 (s, 2 H), 6.90–7.00 (complex, 4 H), 7.79–7.99 (complex, 3 H), 8.26 (s, 1 H), 9.18 (s, 1 H).

¹³C NMR (CDCl₃, 62.5 MHz): δ = 14.7, 61.6, 71.9, 73.6, 117.5, 118.3, 122.1, 124.3, 124.5, 127.2, 127.4, 127.8, 128.4, 129.9, 149.3, 149.6, 165.8, 167.5, 168.6.

MS: m/z (%) = 371 (M⁺ + 1, 20.3), 370 (M⁺, 22.7), 325 (17.2), 248 (8.4), 207 (19.7), 206 (100), 192 (12.4), 178 (39.8), 163 (39.8), 136 (10.3), 122 (10.5), 108 (6.3), 92 (10.2), 52 (24.8).

Anal. Calcd for C₁₉H₁₈N₂O₆ (370.36): C, 61.62; H, 4.90; N, 7.56. Found: C, 61.52; H, 4.77; N, 7.39.

Diethyl 7,14,23,30-Tetraoxo-6,7,8,13,14,15,22,23,24,29,30,31-dodecahydrotetrabenzo[b,h,n,f][1,4,13,16,7,10,19,22]tetraoxatetraazacyclotetrasocine-10,27-dicarboxylate (4f)

Colorless crystals; mp 242–244 °C.

IR (KBr): 3356, 3269, 3070, 2953, 1720 (s), 1693 (s), 1657 cm⁻¹ (s).

¹H NMR (DMSO-*d*₆, 250 MHz): δ = 1.34 (t, 6 H, J = 6.9 Hz), 4.38 (q, 4 H, J = 6.9 Hz), 4.68–4.79 (complex, 8 H), 7.03–7.11 (complex, 8 H), 7.78–7.88 (complex, 6 H), 8.15 (s, 2 H), 9.31 (s, 2 H).

¹³C NMR (DMSO-*d*₆, 62.5 MHz): δ = 14.5, 61.3, 71.8, 73.2, 117.9, 118.6, 122.5, 124.1, 124.9, 127.7, 127.9, 128.1, 128.5, 129.8, 149.8, 150.5, 167.0, 167.15, 168.4.

MS: m/z (%) = 576 (3.3), 476 (2.6), 434 (1.8), 373 (1.1), 389 (4.3), 370 (7.3), 354 (11.4), 326 (15.4), 312 (20.1), 297 (4.8), 267 (8.2), 239 (3.7), 222 (2.5), 206 (23.7), 178 (6.9), 150 (41.8), 134 (3.4), 121 (100), 106 (4.2), 92 (7.6), 76 (12.1), 52 (44.3).

Anal. Calcd for C₃₈H₃₆N₄O₁₂ (740.71): C, 61.62; H, 4.90; N, 7.56. Found: C, 61.75; H, 4.82; N, 7.67.

Pentyl 7,14-Dioxo-6,7,8,13,14,15-hexahydrodibenzo[b,h][1,4,7,10]dioxadiazacyclododecene-10-carboxylate (3g)

Colorless crystals; mp 148–151 °C.

IR (KBr): 3360, 3250, 3080, 2959, 1719 (s), 1695 (s), 1651 cm⁻¹ (s).

¹H NMR (CDCl₃, 250 MHz): δ = 0.86–0.94 (complex, 5 H), 1.53–1.70 (complex, 4 H), 4.24 (t, 2 H, J = 6.8 Hz), 4.50 (s, 2 H), 4.55 (s, 2 H), 6.82–6.91 (complex, 4 H), 7.79–7.92 (complex, 3 H), 8.29 (s, 1 H), 9.11 (s, 1 H).

¹³C NMR (CDCl_3 , 62.5 MHz): δ = 16.9, 22.9, 26.5, 34.6, 64.3, 70.2, 71.9, 117.5, 118.4, 122.1, 124.3, 124.5, 127.3, 128.3, 129.8, 132.8, 136.2, 149.3, 149.6, 165.9, 167.5, 168.6.
 MS: m/z (%) = 413 ($M^+ + 1$, 11.1), 412 (M^+ , 14.9), 354 (1.1), 290 (11.8), 262 (12.7), 248 (35.4), 233 (24.7), 192 (8.4), 178 (100), 164 (18.8), 136 (10.6), 122 (9.0), 108 (5.5), 92 (9.0), 42 (11.4).
 Anal. Calcd for $\text{C}_{22}\text{H}_{24}\text{N}_2\text{O}_6$ (412.44): C, 64.07; H, 5.87; N, 6.79. Found: C, 63.95; H, 5.96; N, 6.71.

Dipentyl 7,14,23,30-Tetraoxo-6,7,8,13,14,15,22,23,24,29,30,31-dodecahydrotetrabenzo[b,h,n,f][1,4,13,16,7,10,19,22]tetraoxotetraazacyclotetrasine-10,27-dicarboxylate (4g)

Colorless crystals; mp 218–220 °C.

IR (KBr): 3360, 3240, 2935, 3055, 2961, 1719 (s), 1680 (s), 1655 cm^{-1} (s).
¹H NMR (CDCl_3 , 250 MHz): δ = 0.89–1.03 (complex, 10 H), 1.59–1.69 (complex, 8 H), 4.31 (t, 4 H, J = 7.0 Hz), 4.72 (s, 4 H), 4.77 (s, 4 H), 7.02–7.07 (complex, 8 H), 7.91–8.02 (complex, 6 H), 8.14 (s, 2 H), 9.21 (s, 2 H).
¹³C NMR (CDCl_3 , 62.5 MHz): δ = 16.7, 22.2, 27.1, 34.2, 63.4, 70.9, 72.4, 117.2, 119.1, 122.2, 124.8, 125.2, 127.7, 128.5, 130.1, 132.2, 135.6, 149.1, 149.8, 164.4, 167.9, 168.7.
 MS: m/z (%) = 577 (2.1), 466 (1.4), 452 (2.3), 428 (5.9), 412 (7.1), 382 (2.1), 354 (3.4), 341 (5.4), 339 (3.9), 297 (2.4), 296 (4.9), 220 (3.3), 203 (100), 150 (2.5), 122 (6.9), 108 (2.5), 92 (2.9), 76 (7.2), 52 (17.1).
 Anal. Calcd for $\text{C}_{44}\text{H}_{48}\text{N}_4\text{O}_{12}$ (824.87): C, 64.07; H, 5.87; N, 6.79. Found: C, 64.15; H, 5.98; N, 6.65.

Octyl 7,14-Dioxo-6,7,8,13,14,15-hexahydridibenzo[b,h][1,4,7,10]dioxadiazacyclododecine-10-carboxylate (3h)

Colorless crystals; mp 135–136 °C.

IR (KBr): 3355, 3240, 3060, 2924, 1721 (s), 1693 (s), 1657 cm^{-1} (s).
¹H NMR (CDCl_3 , 250 MHz): δ = 0.92 (t, 3 H, J = 6.8 Hz), 1.26–1.37 (complex, 10 H), 1.76 (m, 2 H), 4.30 (t, 2 H, J = 7.1 Hz), 4.69 (s, 2 H), 4.74 (s, 2 H), 7.01–7.09 (complex, 4 H), 7.97 (d, 1 H, J = 7.8 Hz), 8.01–8.12 (complex, 3 H), 9.23 (s, 1 H).
¹³C NMR (CDCl_3 , 62.5 MHz): δ = 14.5, 23.1, 26.4, 29.1, 29.58, 29.6, 32.2, 65.8, 72.0, 73.6, 117.5, 118.4, 121.9, 124.5, 124.6, 127.9, 128.6, 130.1, 132.4, 137.2, 149.3, 149.6, 167.4, 168.6.
 MS: m/z (%) = 455 ($M^+ + 1$, 13.7), 454 (M^+ , 16.7), 332 (9.9), 325 (14.0), 290 (47.1), 275 (15.0), 192 (6.3), 179 (18.7), 178 (100), 164 (15.9), 136 (9.9), 122 (7.4), 108 (4.4), 92 (6.7), 52 (9.0).
 Anal. Calcd for $\text{C}_{25}\text{H}_{30}\text{N}_2\text{O}_6$ (454.52): C, 66.06; H, 6.65; N, 6.16. Found: C, 66.21; H, 6.51; N, 6.03.

Diethyl 7,14,23,30-Tetraoxo-6,7,8,13,14,15,22,23,24,29,30,31-dodecahydrotetrabenzo[b,h,n,f][1,4,13,16,7,10,19,22]tetraoxotetraazacyclotetrasine-10,27-dicarboxylate (4h)

Colorless crystals; mp 201–204 °C.

IR (KBr): 3362, 3244, 3070, 2926, 1715 (s), 1678 (s), 1661 cm^{-1} (s).
¹H NMR (CDCl_3 , 250 MHz): δ = 0.89 (t, 6 H, J = 6.7 Hz), 1.24–1.35 (complex, 20 H), 1.72 (m, 4 H), 4.26 (t, 4 H, J = 7.1 Hz), 4.54 (s, 4 H), 4.65 (s, 4 H), 6.98–7.07 (complex, 8 H), 7.55 (d, 2 H, J = 7.9 Hz), 7.68–7.79 (complex, 4 H), 8.04 (s, 2 H), 9.05 (s, 2 H).
¹³C NMR (CDCl_3 , 62.5 MHz): δ = 14.2, 22.8, 26.1, 29.3, 29.9, 30.2, 32.4, 66.2, 71.4, 73.2, 117.1, 118.3, 122.2, 124.8, 125.1, 127.6, 129.2, 129.9, 133.1, 137.5, 149.8, 150.3, 166.5, 167.7, 168.3.
 MS: m/z (%) = 576 (1.5), 454 (2.9), 439 (3.7), 438 (2.5), 424 (2.1), 411 (2.1), 397 (12.8), 396 (21.4), 381 (2.4), 378 (1.7), 348 (2.0), 332

(3.2), 290 (18.7), 275 (9.7), 192 (78.0), 178 (100), 164 (11.6), 150 (40.7), 129 (4.0), 122 (33.5), 108 (2.7), 92 (11.4), 52 (26.6).

Anal. Calcd for $\text{C}_{50}\text{H}_{60}\text{N}_4\text{O}_{12}$ (909.03): C, 66.06; H, 6.65; N, 6.16. Found: C, 65.94; H, 6.54; N, 5.98.

Acknowledgment

We thank the Shiraz University council for partial support of this work. We are also grateful to Dr. A. Khalafi-Nezhad for helpful discussion and to Mr. H. Sajedian Fard for recording the NMR spectra.

References

- (a) Shamsipur, M.; Fakhari, A. R.; Sharghi, H.; Eshghi, H.; Ganjali, M. R. *Polish J. Chem.* **2002**, *76*, 1665; *Chem. Abstr.* **2003**, *138*, 244761. (b) Shamsipur, M.; Rouhani, S.; Ganjali, M. R.; Sharghi, H.; Eshghi, H. *Sensors and Actuators B* **1999**, *59*, 30. (c) Shamsipur, M.; Rouhani, S.; Ganjali, M. R.; Eshghi, H.; Sharghi, H. *Microchem. J.* **1999**, *63*, 202. (d) Shamsipur, M.; Rouhani, S.; Sharghi, H.; Ganjali, M. R.; Eshghi, H. *Anal. Chem.* **1999**, *71*, 4938. (e) Ostaszewski, R.; Stevens, T. W.; Verboom, W.; Reinhoudt, D. N. *Recl. Trav. Chim. Pays-Bas* **1991**, *110*, 198. (f) Bartsch, R. A.; Chapateau, E.; Czech, B. P.; Krzykawski, J.; Kumar, A.; Robinson, T. W. *J. Org. Chem.* **1994**, *59*, 616.
- (a) Sharghi, H.; Massah, A. R.; Eshghi, H.; Niknam, K. *J. Org. Chem.* **1998**, *63*, 1455. (b) Sharghi, H.; Niknam, K.; Pooyan, M. *Tetrahedron* **2001**, *57*, 6057. (c) Sharghi, H.; Nasseri, M. A.; Niknam, K. *J. Org. Chem.* **2001**, *66*, 7287.
- (a) Bondy, C. R.; Loeb, S. *J. Coord. Chem. Rev.* **2003**, *240*, 77. (b) Wisner, J. A.; Beer, P. D.; Drew, M. G. B.; Sambrook, M. R. *J. Am. Chem. Soc.* **2002**, *124*, 12469.
- (a) Krakowiak, K. E.; Bradshaw, J. S.; Izatt, R. M. *J. Org. Chem.* **1990**, *55*, 3364. (b) Kimura, E.; Lin, Y.; Machida, R.; Zenda, H. *J. Chem. Soc., Chem. Commun.* **1986**, 1021. (c) Singh, H.; Kumar, S.; Hundal, M. S.; Hundal, G.; Kaurand, N. *Tetrahedron* **1997**, *53*, 1084. (d) Krakowiak, K. E.; Bradshaw, J. S.; Izatt, R. M. *J. Heterocycl. Chem.* **1990**, *27*, 1585. (e) Zeng, F.; Zimmerman, S. C. *Chem. Rev.* **1997**, *97*, 1681. (f) Izatt, R. M.; Christensen, J. J. *Synthetic Multidentate Macrocyclic Compounds*; Academic Press: New York, **1978**. (g) Grayson, S. M.; Frechet, J. M. *J. Chem. Rev.* **2001**, *101*, 3819. (h) Lindoy, L. F. *J. Iranian Chem. Soc.* **2004**, *1*, 1; <http://www.ics-ir.org/jics>.
- (a) Gryko, D. T.; Piętek, P.; Jurczak, J. *Tetrahedron* **1997**, *53*, 7957. (b) Jurczak, J.; Stankiewicz, T.; Kasprzyk, S.; Lipkowski, P. *Tetrahedron* **1993**, *49*, 1478.
- (a) Petranek, J.; Ryba, O. *Collect. Czech. Chem. Commun.* **1980**, *45*, 1567. (b) Petranek, J.; Ryba, O. *Collect. Czech. Chem. Commun.* **1983**, *48*, 1945.
- (a) Overman, L. E. *J. Org. Chem.* **1972**, *37*, 4214. (b) Knops, P.; Sendhoff, N.; Mekelburger, H. B.; Vogtle, F. *Top. Curr. Chem.* **1992**, *161*, 1. (c) Bradshaw, J. S.; Krakowiak, K. E.; Izatt, R. M. *Chemistry of Heterocyclic Compounds*, Vol. 51; Wiley: New York, **1993**. (d) Gunnlaugsson, T.; Gunaratne, H. Q. N.; Nieuwenhuizen, M.; Leonard, J. P. *J. Chem. Soc., Perkin Trans. I* **2002**, 1954. (e) Kleefisch, G.; Kreutz, C.; Bargon, J.; Silva, G.; Schalley, C. A. *Sensors* **2004**, *4*, 136.
- (a) Alexander, V. *Chem. Rev.* **1995**, *95*, 275. (b) Thompson, M. C.; Busch, D. H. *J. Am. Chem. Soc.* **1964**, *86*, 3651.
- (a) Jurczak, J.; Pietraszkiewicz, M. *Top. Curr. Chem.* **1985**, *130*, 183. (b) Jurczak, J.; Gryko, D. T.; Lipkowski, P.; Sałński, P. *Rev. High Pressure Technol.* **1998**, *7*, 1236.

- (10) (a) Tabushi, I.; Okino, H.; Kuroda, Y. *Tetrahedron Lett.* **1976**, *17*, 4339. (b) Tabushi, I.; Taniguchi, Y.; Kato, H. *Tetrahedron Lett.* **1977**, *18*, 1049. (c) Gryko, D. T.; Piatek, P.; Saiański, P.; Jurczak, J. *Tetrahedron: Asymmetry* **1998**, *9*, 1771. (d) Gryko, D. T.; Piatek, P.; Jurczak, J. *Synthesis* **1999**, *336*. (e) Szczepańska, A.; Saiański, P.; Jurczak, J. *Heterocycles* **2000**, *52*, 537. (f) Achmatowicz, M.; Szczepańska, A.; Gryko, D. T.; Saiański, P.; Jurczak, J. *Supramol. Chem.* **2000**, *12*, 93. (g) Szczepańska, A.; Saiański, P.; Jurczak, J. *Tetrahedron* **2003**, *59*, 4775.
- (11) (a) Sharghi, H.; Eshghi, H. *Tetrahedron* **1995**, *51*, 913. (b) Sharghi, H.; Niknam, K.; Massah, A. H. *J. Heterocycl. Chem.* **1999**, *36*, 601. (c) Sharghi, H.; Paziraee, Z. *Synthesis* **2004**, *600*.
- (12) Vogel, A. I. *Practical Organic Chemistry*; Longman, Green and Co.: London, **1954**, 161.
- (13) Wang, X.; Wei, T.; Chen, J.; Li, J. *Synth. Commun.* **1996**, *26*, 2765.
- (14) Sharghi, H.; Hosseini, S. M. *Synthesis* **2003**, *879*.