

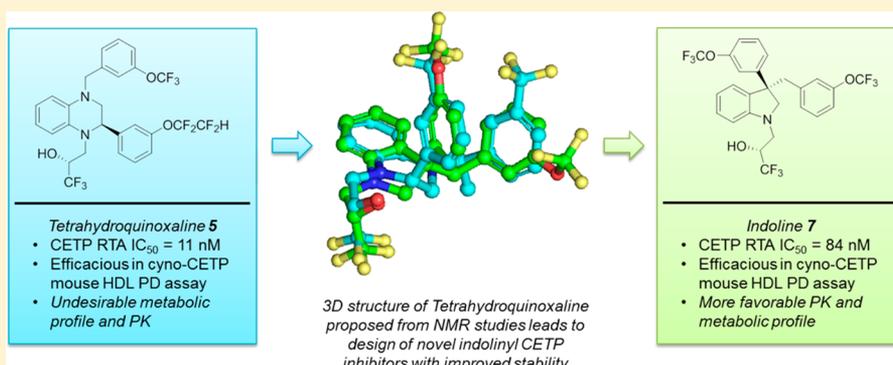
Discovery of Novel Indoline Cholesterol Ester Transfer Protein Inhibitors (CETP) through a Structure-Guided Approach

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S Supporting Information



ABSTRACT: Using the collective body of known (CETP) inhibitors as inspiration for design, a structurally novel series of tetrahydroquinoxaline CETP inhibitors were discovered. An exemplar from this series, compound 5, displayed potent in vitro CETP inhibition and was efficacious in a transgenic cynomolgus-CETP mouse HDL PD (pharmacodynamic) assay. However, an undesirable metabolic profile and chemical instability hampered further development of the series. A three-dimensional structure of tetrahydroquinoxaline inhibitor 6 was proposed from ¹H NMR structural studies, and this model was then used in silico for the design of a new class of compounds based upon an indoline scaffold. This work resulted in the discovery of compound 7, which displayed potent in vitro CETP inhibition, a favorable PK–PD profile relative to tetrahydroquinoxaline 5, and dose-dependent efficacy in the transgenic cynomolgus-CETP mouse HDL PD assay.

KEYWORDS: CETP inhibition, cholesterol ester transfer protein, HDL, indoline, tetrahydroquinoxaline

High levels of plasma low-density lipoprotein (LDL) are clinically associated with atherosclerosis and its consequences, which can include coronary heart disease, stroke, and peripheral vascular disease.^{1,2} Despite the successful development of therapies targeting cholesterol absorption, biosynthesis, and clearance, atherosclerosis remains a serious public health issue.³ As such the development of additional, novel therapies for cardiovascular disease is needed.

In contrast to LDL, high plasma levels of high-density lipoprotein (HDL) are correlated with a cardioprotective effect.⁴ This may be attributed to higher hepatic clearance of cholesterol via HDL relative to LDL through a process known as reverse cholesterol transport.^{5–7} However, HDL has antioxidant and anti-inflammatory properties that may also contribute to its antiatherogenic effects.^{8,9} The balance and interplay between LDL and HDL is influenced by cholesterol

ester transfer protein (CETP), a plasma protein that facilitates the exchange of cholesterol esters from HDL particles to LDL particles and triglycerides from LDL to HDL. Thus, inhibition of CETP would augment cholesterol excretion by increasing the HDL to LDL ratio and could hypothetically provide an opportunity for the treatment of cardiovascular disease via increased cholesterol efflux. Because of the prevalence of health conditions caused by cardiovascular disease this hypothesis is currently being tested in the clinic by multiple organizations.^{10–13}

Due to our continuing interest in new therapies for cardiovascular disease, including CETP inhibition, we initiated

Received: October 15, 2015

Accepted: January 4, 2016

an effort to identify structurally novel CETP inhibitors. We were inspired by the structural homology of the CETP inhibitors 1, 2, and 3 that had been described by Pfizer, Pharmacia, and Johnson and Johnson for two reasons.^{14–16} First, these three classes of compounds had all demonstrated efficacy *in vivo*. Second, we sought to investigate diverse chemical matter that would be complementary to Merck's established class of oxazolidinone CETP inhibitors, exemplified by the clinical compound, 4, anacetrapib (Figure 1).¹⁷ From

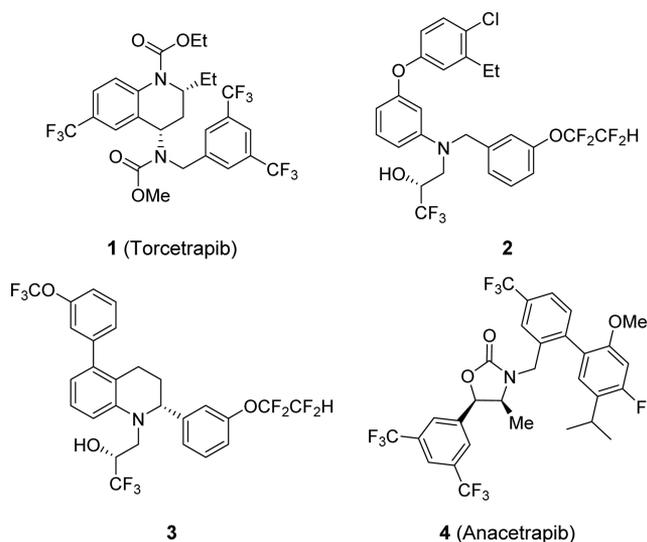


Figure 1. Structures used as inspiration for design of structurally novel CETP inhibitors.

this exercise we designed a novel series of substituted tetrahydroquinoxalines that bear elements of the four aforementioned classes of CETP inhibitors, while at the same time being distinct from each of them. This investigation led to the discovery of compound 5, which displayed promising *in vitro* CETP inhibition and reasonable rat PK (Figure 2).¹⁸ Interestingly, the optimal stereochemistry at C-2 of the tetrahydroquinoxaline is opposite to that found in tetrahydroquinoline 3.¹⁹

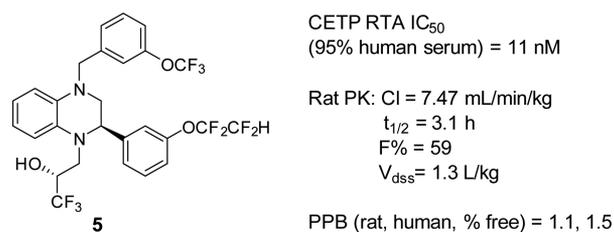


Figure 2. *In vitro* and PK properties of tetrahydroquinoxaline CETP inhibitor 5.

During the course of this work we discovered that compound 5, analogues of 5, and tetrahydroquinoxaline intermediates en route to 5 would gradually oxidize to the monocationic and aromatic dicationic quinoxalinium species when stored under ambient conditions over a timespan of days. These degradation products could be observed by LCMS. Additionally, metabolite identification studies following incubation of 5 in human liver microsomes showed extensive oxidative metabolism leading to dealkylation and the formation of possible reactive intermediates (Figure 3).²⁰ Although 5 was found to be efficacious *in vivo* in a transgenic cynomolgus-CETP mouse pharmacodynamic assay, it displayed highly nonlinear PK.^{21,22} These observations lead us to view the pharmacodynamic effects of 5 with skepticism.

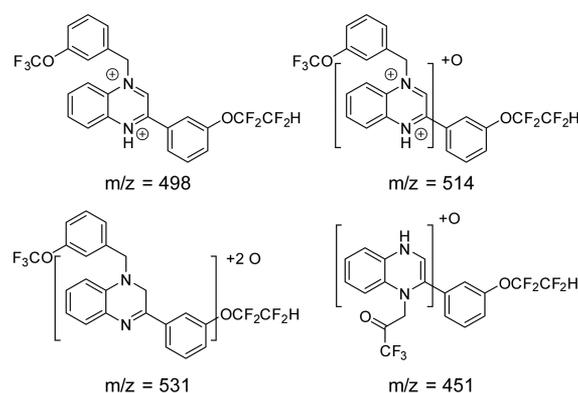


Figure 3. Proposed structures of oxidative metabolites observed after incubation of compound 5 with human liver microsomes.

in vivo in a transgenic cynomolgus-CETP mouse pharmacodynamic assay, it displayed highly nonlinear PK.^{21,22} These observations lead us to view the pharmacodynamic effects of 5 with skepticism.

We first sought to address the oxidative liabilities of the core by stabilizing the tetrahydroquinoxaline ring system through introduction of a heteroatom or an electron withdrawing substituent in the aromatic portion of the core. Unfortunately these strategies generally led to a reduction in potency and little or no change in the metabolic or chemical stability of the compounds.²³ Due to concerns about the generation of reactive intermediates, the potential difficulties of progressing a chemically unstable compound, and our inability to stabilize the tetrahydroquinoxaline core without a substantial loss of potency we chose to re-engineer this series.

Faced with this challenge, we turned our attention to elucidation of the three-dimensional structure of compounds like 5 with the hope that structural insight would spur the design of novel scaffolds that would mimic the three-dimensional chemical shape of the tetrahydroquinoxalines but would be devoid of the metabolic and chemical instability inherent to this core. Stereochemical analysis by ¹H NMR revealed a dependence of the molecular conformation on the relative stereochemistry of C-2 and the 1,1,1-trifluoropropanol side chain that allowed for the assignment of the absolute stereochemistry of compound 6. This led to the proposal of a U-shaped three-dimensional structure in which the C-2 aromatic group is in a pseudoaxial position and the *N*-benzyl group lies parallel to the C-2 group on the same face of the tetrahydroquinoxaline core (Figure 4).²⁴

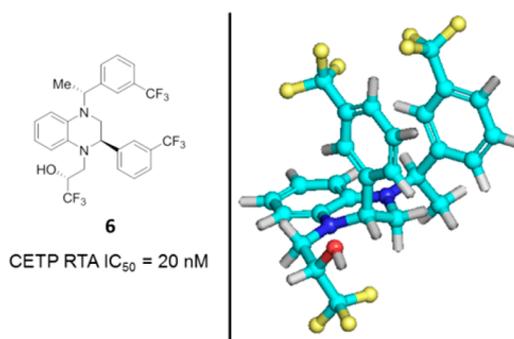


Figure 4. Proposed three-dimensional structure of 6.

Using the proposed solution-phase conformation of the tetrahydroquinoxalines as inspiration, we designed molecules that could adopt the U-shape that we speculated was critical for CETP inhibition and that would lack the metabolic and chemical instability of the tetrahydroquinoxaline core. Although we were well aware that the solution-phase conformation may not be representative of the active conformation of the tetrahydroquinoxaline inhibitors when bound to CETP, we nevertheless chose to pursue this strategy in the hope that we would uncover novel CETP inhibitor chemotypes. Using this strategy, we designed the 3,3-disubstituted indoline 7, which overlaid well with the tetrahydroquinoxaline inhibitor 6 fixed in the proposed U-shaped three-dimensional structure arrived at from NMR structural studies (Figure 5).²⁵ Compounds of this

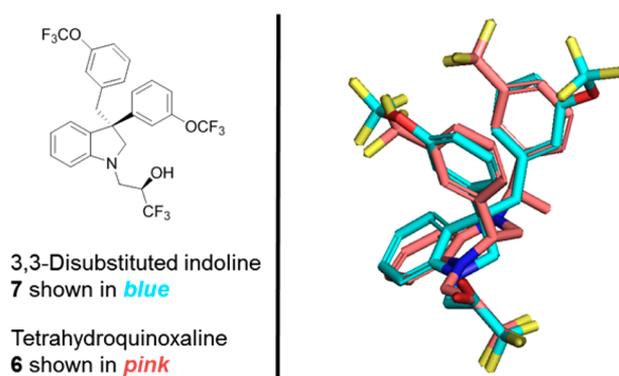


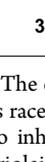
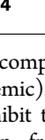
Figure 5. Overlay of tetrahydroquinoxaline 6 with 3,3-disubstituted indoline 7 (hydrogens removed for clarity).

type should not inherently be prone to the oxidative chemical instability associated with the tetrahydroquinoxaline core because the aromatization process that results from oxidation of 7 is not possible on an indoline scaffold in which the 3-position is a fully substituted quaternary carbon atom.

In order to rapidly survey a broad range of substituents at the C-3 position of the indoline core, compounds were evaluated for CETP inhibition first as a mixture of diastereomers at C-3, and the results are presented in Table 1. To our delight, indoline 8 afforded significant intrinsic CETP inhibitory potency with an IC_{50} of 69 nM. Truncation of the arene at R1 or R2 completely ablated CETP activity (compounds 9 and 10). Substitution at the 3-position of the R1 arene was preferred over unsubstituted, 2-substituted, and 4-substituted benzenes (compounds 11, 12, and 13). Multiple 3-substituted benzenes and heteroarenes were explored at R1. However, none of these substituents were found to be superior to the 3-trifluoromethoxy group (compounds 14 through 19). Attempts to transpose the 3,5-bis-trifluoromethylbenzene group of anacetrapib and torcetrapib onto the indoline core were unsuccessful (compounds 20 and 21). The 3-trifluoromethoxybenzene was again found to be optimal at R2 despite attempts to find suitable replacements (compounds 22 through 27). Finally, 4-, 5-, 6-, and 7-substituted indolines were explored, but the unsubstituted core was found to be superior to those that were surveyed with respect to CETP inhibition (compounds 28 through 34).

The diastereomers and enantiomers of compound 8 were synthesized and resolved to determine the impact of stereochemistry on CETP inhibitory potency (Table 2). The C-3 diastereomers, 7 and 35, show a substantial difference in CETP

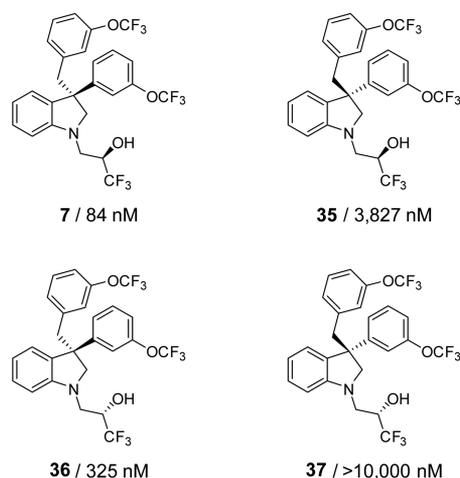
Table 1. 3,3-Disubstituted Indoline CETP Inhibitor SAR

compound ^a	R ¹ =	R ² =	R ³ =	CETP RTA IC_{50} (nM) ^b
8	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	69
9	3-OCF ₃ -C ₆ H ₄	CH ₃	H	>10,000
10	H	3-OCF ₃ -C ₆ H ₄	H	>10,000
11	C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	746
12	2-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	3,374
13	4-OMe-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	9,683
14	3-CN-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	917
15	3-CO ₂ Me-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	937
16	3-CO ₂ H-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	>10,000
17	3-SO ₂ Me-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	8,958
18	3-Br-C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	308
19		3-OCF ₃ -C ₆ H ₄	7-F	1996
20	3,5-bis-CF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	H	>10,000
21	3-OCF ₃ -C ₆ H ₄	3,5-bis-CF ₃ -C ₆ H ₄	H	>10,000
22	3-OCF ₃ -C ₆ H ₄	3-CF ₃ -C ₆ H ₄	H	1,218
23	3-OCF ₃ -C ₆ H ₄	3-OCF ₂ CF ₂ H-C ₆ H ₄	H	94
24	3-OCF ₃ -C ₆ H ₄	3-Cl	H	1,375
25	3-OCF ₃ -C ₆ H ₄	4-OCF ₃ -C ₆ H ₄	H	232
26	3-OCF ₃ -C ₆ H ₄		H	101
27	3-OCF ₃ -C ₆ H ₄		H	2,602
28	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	4-CH ₃	>10,000
29	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	5-F	763
30	C ₆ H ₅	3-OCF ₃ -C ₆ H ₄	5-OCF ₃	>10,000
31	3-OCF ₃ -C ₆ H ₄	C ₆ H ₅	5-OCF ₃	>10,000
32	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	6-CF ₃	>10,000
33	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	7-Me	491
34	3-OCF ₃ -C ₆ H ₄	3-OCF ₃ -C ₆ H ₄	7-F	222

^aThe compounds are a 1:1 mixture of diastereomers (C-2 stereocenter is racemic). ^bThis in vitro assay measures the ability of the compounds to inhibit the CETP mediated transfer of [³H]-cholesterol or [³H]-triolein from LDL to HDL. The assay was performed using endogenous CETP in 95% human serum (HS), $n = 1$. Data for the same in vitro assay run in 2% human serum are available in the Supporting Information.

inhibition as would be expected from consideration of overlays with compound 6. The remaining two diastereomers, 36 and 37, were also considerably less potent. From this data it appears that the quaternary stereocenter is the primary driver for potency (7 and 36 versus 35 and 37), while the alcohol stereochemistry has a less of an impact on potency (7 versus 36). In the CETP RTA in vitro assay in 95% human serum compound 7 is superior to compounds 2 and 3 (2, $IC_{50} = 197$ nM; 3, $IC_{50} = 295$ nM) and comparable with the clinical

Table 2. Effect of Stereochemistry on CETP Potency

compound / CETP RTA IC₅₀ value^a

^aThis in vitro assay measures the ability of the compounds to inhibit the CETP mediated transfer of [³H]-cholesterol oleate or [³H]-triolein from LDL to HDL. The assay was performed using endogenous CETP in 95% human serum ($n = 1$). Data for the same assay run in 2% human serum are available in the [Supporting Information](#).

CETP inhibitors 1 and 4 (1, IC₅₀ = 45 nM; 4, IC₅₀ = 53 nM).^{15,16}

The CETP inhibitory action of the indoline series was further confirmed by the ability of 7 to increase HDL-cholesterol in an exposure-dependent manner when orally administered to cynomolgus-CETP transgenic mice ([Figure 6](#)).^{21,26} This

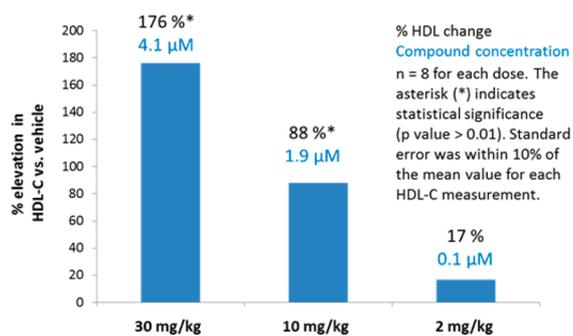


Figure 6. Elevation of HDL-C 4 h postdose in cynomolgus-CETP transgenic mice orally administered compound 7.

exposure-dependent PD effect of 7, which displays good pharmacokinetic properties in a rat (dose 0.15 mg/kg; Cl = 36.0 mL/min/kg; $t_{1/2}$ = 4.76 h; V_{dss} = 6.70 L/kg) is in stark contrast to 5, which exhibited an unusual exposure–PD relationship. This difference is most likely due to the increased chemical and metabolic stability of 7 relative to 5. Unfortunately, the in vivo efficacy of 7 can not be directly compared with compounds 1–4 because HDL measurements from the cyno-CETP mouse PD assay were recorded at different time points for the respective sets of compounds.²⁷ However, the PK profile, promising in vivo efficacy, and structural novelty of the indoline series render them an attractive starting point for the development of future CETP inhibitors.

In summary, a novel class of CETP inhibitors based on an indoline scaffold has been discovered.^{28,29} The indoline series was designed from in silico overlays with tetrahydroquinoxaline CETP inhibitor 6, of which a three-dimensional structure was proposed from structural information derived from ¹H NMR studies. Indoline 7 is a potent inhibitor of CETP in vitro, displays improved metabolic and chemical stability relative to its progenitor, tetrahydroquinoxaline 5, and shows dose-dependent efficacy in the cynomolgus-CETP mouse HDL pharmacodynamic assay. Furthermore, the three-dimensional structures of 6 and 7 proposed herein may be a useful template from which to design additional novel classes of CETP inhibitors.

■ ASSOCIATED CONTENT

Supporting Information

The Supporting Information is available free of charge on the ACS Publications website at DOI: [10.1021/acsmchemlett.5b00404](https://doi.org/10.1021/acsmchemlett.5b00404).

Experimental procedures and characterization for compounds 5 through 37, a description of the in vitro assay, and data for compounds 5–37 in the CETP RTA assay run in 2% human serum ([PDF](#))

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Notes

The authors declare no competing financial interest.

■ ACKNOWLEDGMENTS

The authors gratefully acknowledge Dr. Christopher Sinz, Dr. Christopher Plummer, and Eric Meade for reviewing the article and providing comments.

■ ABBREVIATIONS

CETP, cholesterol ester transfer protein; HDL, high density lipoprotein; PPB, plasma protein binding; HS, human serum; LDL, low density lipoprotein; PK, pharmacokinetic; PD, pharmacodynamic

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(19) The remaining three diastereomers were prepared and the CETP RTA IC₅₀ values for those compounds are as follows (alcohol stereochemistry is listed first followed by the C-2 stereochemistry): (R,S)-isomer = 3273 nM; (S,S)-isomer = 716 nM; (S,R)-isomer = 60 nM.

(20) Compound **5** (10 μ mol) was incubated with C57 mouse liver microsomes (1 mg/mL protein) for 60 min at 37 °C. Structure assignments are tentative, based on LC–MS/MSE data. The elemental compositions of the metabolites were confirmed by high-resolution mass analysis (QTOF).

(21) Compounds or noncompound-containing vehicle were orally administered to transgenic mice that overexpress the cynomolgus cholesteryl ester transfer protein. HDL-cholesterol was measured from plasma collected at 4 h postdose using a commercially available kit (Wako Diagnostics). The effect of the test compound on HDL-cholesterol is expressed as the percent elevation compared to that of

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(22) Compound **5** did not show the expected dose–exposure relationship when dosed orally at 1, 3, and 10 mg/kg in the transgenic cynomolgus-mouse PD assay. The exposures were 1.3, 1.8, and 1.9 μ M, respectively.

(23) Both isomers of the 1,2,3,4-tetrahydropyrido[2,3-*b*]pyrazine core were prepared in addition to one isomer of the 1,2,3,4-tetrahydropyrido[3,4-*b*]pyrazine core. Only one isomer of 1,2,3,4-tetrahydropyrido[2,3-*b*]pyrazine showed CETP activity, but none of the compounds showed improved chemical or metabolic stability. Fluoro- and trifluoromethyl-quinoxalines were also explored but were not found to have improved stability.

(24) The details of the NMR studies can be found in the [Supporting Information](#).

(25) Ensemble conformations for both compounds **6** and **7** were generated using omega (openeye scientific software); Hawkins, P. C. D.; Skillman, A. G.; Warren, G. L.; Ellingson, B. A.; Stahl, M. T. *J. Chem. Inf. Model.* **2010**, *50*, 572–584. and overlaid using rocs (openeye scientific software); Hawkins, P. C. D.; Skillman, A. G.; Nicholls, A. *J. Med. Chem.* **2007**, *50*, 74. The best overlaid pair was then refined with the flexible alignment module of the MOE package (Molecular Operating Environment, 2014.09; Chemical Computing Group Inc., 1010 Sherbooke St. West, Suite #910, Montreal, QC, Canada, H3A 2R7, 2015). [Figures 4](#) and [5](#) were rendered using PyMOL (The PyMOL Molecular Graphics System, Version 1.7.4 Schrödinger, LLC).

(26) The data in [Figure 6](#) for the 2 mg/kg dose and the 10 and 30 mg/kg doses were collected on different days. A positive control, structurally related to anacetrapib (CETP RTA IC₅₀ in 95% human serum = 43 nM) was run in both instances. The positive control increased HDL by 67% on the day the 2 mg/kg dose was administered. On the day of the 10 and 30 mg/kg dosing, the positive control increased HDL by 191%.

(27) HDL measurements in the cyno-CETP mouse PD assay conducted at Merck for compound **7** were measured at 4 h postdose, while measurements for compounds **2** and **3** were taken at 18 h postdose (compound **2**, 30 mpk PO dose, –20% change in HDL 18 h postdose relative to vehicle; compound **3**, 30 mpk PO dose, +4.2% change in HDL 18 h postdose relative to vehicle).

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