



Accepted Article

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Authors: Meng Yu, Xuefeng Yong, Weiwei Gao, Chun–Yu Ho*

This manuscript has been accepted and appears as an Accepted Article online.

This work may now be cited as: *Chin. J. Chem.* **2021**, *39*, 10.1002/cjoc.202000651.

The final Version of Record (VoR) of it with formal page numbers will soon be published online in Early View: http://dx.doi.org/10.1002/cjoc.202000651.

WILEY-VCH SIOC CCS

ISSN 1001-604X • CN 31-1547/O6 mc.manuscriptcentral.com/cjoc www.cjc.wiley-vch.de Cite this paper: Chin. J. Chem. 2021, 39, XXX-XXX. DOI: 10.1002/cjoc.202100XXX

Diastereodivergent Hydrosilylative Enyne Cyclization Catalyzed by NHC-Ni(o)

Meng Yu,^a Xuefeng Yong,^b Weiwei Gao,^b Chun–Yu Ho*^{a,b}

Guangdong Province Key Laboratory of Catalysis, Department of Chemistry, Southern University of Science and Technology (SUSTech), 1088 Xueyuan Avenue, Shenzhen, 518055, Guangdong, China.

Shenzhen Grubbs Institute, Department of Chemistry, Southern University of Science and Technology (SUSTech), 1088 Xueyuan Avenue, Shenzhen, 518055, Guangdong, China.

edicated to Department of Chemistry, SUSTech, on the occasion of her 10th founding anniversary.

Keywords

N-heterocyclic carbenes | nickel | diastereodivergent synthesis | enyne | silylative hydroalkenylation

Main observation and conclusion

Catalytic diastereodivergent hydrosilylative enyne cyclization with high generality and broad scope was achieved using electronic activated NHC-Ni(0) as a catalyst and R_3SiH as silane (IPr^{CI}, syn-:anti-selectivity from up to 98:2 to 7:93 by Z = O, NH vs NMs, R^1 = n-pentyl). Heterocycles bearing homoallylsilane rather than vinylsilane was obtained chemoselectively. The undesired yet highly competitive reactivity was suppressed, like direct hydrosilylation of alkene and alkyne concurrently. Optionally, the homoallylsilane products could be reduced further in 1-pot using IPr^{Me} as ligand and (EtO)₃SiH as silane under otherwise the same standard condition as the above, offers practical access to additional stereocenters and more diverse product structures from enynes.

Comprehensive Graphic Content

10.1002/cjoc.202000651



R^{1 =} linear/cyclic alkyls: alkenyls: aryls & heteroaryls R^{2 =} H and alkyl; Allow Contiguous & Skipped Stereocenters Synth. NHC Steric & Electronic Controlled Cyclization over Direct Hydrosilylation

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Chin. J.	Chem. 2021 , <i>39</i> , XXX—XXX	© 2021 SIOC, CAS, Shanghai, & WILEY-VCH GmbH	
	This article has been accepted	for publication and undergone full	peer review but has not been
	through the copyediting, type	esetting, pagination and proofreadi	ng process which may lead to
	differences between this ver	sion and the Version of Record. H	Please cite this article as doi:

Background and Originality Content

Transition metal-catalyzed hydrosilylative cyclization is a renowned strategy to prepare key intermediates for more complicated products.¹⁻⁴ Notable methodologies have been developed over the years to utilize those silylated products, in which Hiyama coupling⁵⁻⁹ and Fleming-Tamao oxidations are often exploited.¹⁰⁻¹² In particular, the catalytic hydrosilylative envne cyclizations have received a lot of attention over decades and efficiently construct various carbo-/hetero-cycles (Scheme 1a).¹³⁻¹⁵ Unlike the hydrosihative divne cyclization, most of the envne mechanism's first step is the corresponding metal hydride or metal silyl insertions to the kyne rather than metalacycle formation. That is because of the relatively lower oxidative cyclization reactivity of the enyne and the somewhat easier oxidative addition of silane.^{16,17} After subse-Jent insertion and reductive elimination steps, homoallyl-/vinyl-silanes products are formed, and the catalyst is regenered accordingly. Common complications are those related to premature reductive elimination after alkyne insertion, competi-/e intermolecular alkyne versus alkene insertions, as well as reductive cyclization without silyl group incorporation (Scheme 1a, reduction).^{18, 19} Therefore, many well-designed catalysts and rategies have been developed recently in order to cope with different enyne structural characteristics for high regioselectivity. r example, Rh₄(CO)₁₂,^{20, 21} Rh₂Co₂(CO)₁₂,^{20, 21} NHC-Rh,²² Rh-Co nanoparticle²³ and PBu₃-Ni²⁴ complexes for products with vinylsilane and allyl silane, while $\mathsf{Cp}^*{}_2\mathsf{Y}^{25\text{-}27},$ $\mathsf{IP}.\mathsf{CoCl}_2{}^{28}$ and a few π allyl-Pd²⁹ complexes for homoallylsilane derivatives (Scheme 1a-c). Asymmetric and diastereoselective versions have also been achieved using chiral Rh and Y catalysts.^{30, 31} Yet, diastereodiverent synthesis development has been limited often by the delicate balance of the above highly competitive reactivity, especially when nfavorable or ineffective steric repulsions are involved.³²⁻³⁴ In cases of 1,n-enynes without a cyclic template or quaternary center ir Ni(0) catalysis, non-selective oligomerization may lower the , eld and selectivity further sometimes (e.g. n = 7).³⁵⁻³⁷ Overall, there is a great demand to develop a new methodology to meet le new challenges in the catalytic hydrosilylative enyne cyclization and quickly provide products with more diverse substitution atterns.

e have recently achieved a diastereodivergent 1,n-azaenynes reductive hydroalkenylation³⁸ and hydroacylation³⁹ using IF r–Ni(0) as catalysts. Alcohols and aldehydes have been used as eductants and hydroacylating reagents, respectively. Our mechanistic study has shown that nickelacyclopentene formation is cruc al in those reactions. Electronic manipulations of the 1,n-enyne's .etero-substituent can work with 1,3-allylic strain and govern the heterocycles' C-Z-C bond angles as shown in our DFT calculation.³⁹ A a result, the product's preferred configurations can be con-.rolled logically.

Next, we envisioned a diastereodivergent hydrosilylative enyne cyclization that could be achieved by trapping the nickelacyclopentene with a suitable silane (Scheme 1d). Yet, the above complications involved silanes and enynes can be very significant. Key challenges ahead are the highly competitive NHC-Ni(0) catalyzed alkyne or alkene hydrosilylations through NHC-Ni(II)H(SiR₃) formation at r.t. (Scheme 2, pathway I).^{41, 42} Indeed, unless a gem-difluoromethylene was used as alkene terminus, the hydrosilylative enyne cyclization was unable to proceed with Ni(0) (Scheme 1c).²⁴ Moreover, the silane needs to be more reactive than the enyne in the nickelacyclopentene ring-opening step to avoid undesired enyne homo-reaction (Scheme 2, pathway II). At the same time, it has to be less reactive toward the Pathway I.



Scheme 1 Catalytic Hydrosilylative Enyne Cyclizations.

Herein, we report our continuous effort in diastereodivergent heterocycle synthesis, illustrating the scope and selectivity further by the practical catalytic synthesis of structurally more diverse heterocycles bearing homoallylsilanes (Scheme 1d, c.f. preliminary study: $R^1 = Ph$, $R^2 = H$, Z = NH and NMs^{38,40}). Interestingly, the optimization showed that the exocyclic olefin on the heterocycle could be hydrogenated using a suitable combination of catalyst and siliane in one-pot. That discovery offers an option to access an additional stereocenter for structurally more diverse product structures at high efficiency.



Scheme 2 Competitions in NHC-Ni(0) Catalyzed Enyne Hydrosilylation Cyclizations.

Results and Discussion

Our study commenced with enyne P1^{oxy}a as a substrate and a ethoxylsilane 2a as a hydrosilylating reagent (Scheme 3). Unfortunately, this initial attempt using IPr-Ni(0) catalyst did not offer good chemoselectivity and yield. Direct hydrosilylation of kyne^{41, 43} and non-selective [2+2+2] cycloaddition⁴⁴ occurred significantly as the literature implied. The desired hydrosilylative hyne cyclization product 1,3-*syn*-P3^{oxy}aa was formed only in a small amount, yet the diastereoselectivity and homoallylsilane formation selectivity were remarkably high (no 1,3-*anti*-P3^{oxy}aa and no vinylsilane by ¹H NMR). Interestingly, an unexpected exocyclic methylene reduction of 1,3-*syn*-P3^{oxy}aa was observed and resulted in a significant amount of 1,3-*syn*-P3^{oxy}aa-H product.

Notable advances in yield and selectivity came by screening ands and silanes.45,46 First, ligand screening was conducted using a less reactive triethylsilane 2b to avoid those undesired reactivities related to hydrosilane oxidative addition, like exocyclic ethylene reduction, as well as direct alkyne and/or alkene hydrosilylations. While very low reactivity and basically no imovement were observed using PCy₃, SIPr and IMes,²⁴ much better results were observed using sterically bulkier NHCs like IPr^{Me} nd IPr^{CI} than the IPr itself (Table 1, entry 1-4). Since the difference in V%_{bur} of IPr^{Me} and IPr^{CI} is relatively small,⁴⁵⁻⁴⁸ the better and higher catalyst reactivity obtained from IPr^{CI} was attributed mainly to the electronic property change (entry 3 and 4). That could result in a more selective nickelacyclopentene opening Jith **2b** over [2+2+2] reactivity. It should be noted that the IPr^{CI} catalyst has a lower alkyne hydrosilylation reactivity than IPr, as e idenced by a model study using a terminal alkyne for a compar-Jon (see SI), which may also help the desired pathway. To our delight, the use of IPr^{CI} also provided good results in the synthesis azacycle 1,3-syn-P3^{NH}ab (entry 5). Yet, the decisive improvement brought by the use of IPr^{CI} is even more apparent in the synthesis of azacycle 1,3-anti-P3^{NMs}ab (entry 6-9). With the key HC identified, other combinations of R₃SiH with various steric and electronic properties could be used (Table 2). Desired products could be obtained in reasonably good yield, high d.r. and homoallyl-:vinyl-silane selectivity in general (syn-:anti- > 95:5 for Z = O and NH; > 5:95 for Z = NMs).

Next, such an NHC and silane cooperation effect in preparing sterically more challenging and diversely substituted heterocycles was tested (Table 3). Even though the changes in enyne substituents could cause undesired steric repulsions and reactions, the desired selectivity and reactivity were maintained very well. First, highly consistent 1,3-syn-/anti-configuration preferences and high homoallyl-:vinyl-silane ratio were observed in general among

enynes with various R¹, like linear/cyclic alkyls, alkenyls, and (hetero-)aryls (Set 1a, 1,7-enyne **P1^z** with propargyl-Z). Moreover, the method was found equally effective in the catalytic preparation of 1,4-diastereomers using 1,7-enynes A1^z with allyl-Z (Set 1b, syn-:anti- > 5:95 for Z = O and NH; > 95: 5 for Z = NMs). This switch of the product's syn-/anti-configurations preference resembled well with our former diastereodivergent reductive hydroalkenylation and hydroacylation. Enynes with internal alkenes were found incompatible due to the more apparent terminal alkyne oligomerization. On the other hand, enynes with internal alkynes are decent substrates under our condition (Set 2a-b). That serves as a new method to synthesize stereodefined trisubstituted homoallylsilane on the ring. Finally, the method helps set up contiguous/skipped stereocenters on heterocycles (Set 3-4). The preferred outcomes in these hydrosilylative enyne cyclizations still follow the choice of Z selected as indicated above.



Scheme 3 Initial attempt in catalytic hydrosilylative oxaenyne cyclization (product structures are shown in relative configurations).

Table 1 Screening of Ligands fo	· Hydrosilylative	Enyne Cyclization.[a]
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entry	/ 1	2	Ligand	P3 ^z	yield (%)	Syn:Anti ^[d]	P3 ^z -H
1 ^[b]			РСу₃		18	n.d.	< 5
2 ^[c]	P1 ^{0xy} a	2h	IPr	P3 ^{0xy} ah	53	> 95 :5	< 5
3 ^[c]	· 1 u	2.5	IPr ^{Me}	15 45	68	> 95 :5	<5
4			IPr ^{ci}		84	> 95 :5	< 5
5	P1 ^{NH} a	2b	IPr ^{ci}	P3 ^{NH} ab	71	97 :3 ^[d]	< 5
6			PCy ₃		32	13:87	< 5
7	P1 [№] sa	2b	IPr	P3 ^{™s} ab	35	10:90	< 5
8			IPr ^{Me}		56	8:92	< 5

9	lPr ^c	1	87	6: 94	< 5	
DIPP	X X X N N N DIPP	NHC ⁼ IPr IPr ^{Me} IPr ^{CI}	X ⁼ H Me Cl	V% _{Bur} ^{[e] =} 34.9 35.7 35.5	(Aver) ^υ co [f 2023.9 2021.3 2028.3	ŋ=

[a] See scheme 3 for the reaction equation, enyne and product structures. See Experimental Section for procedure except otherwise indicated. [b] 2 equiv. of PCy₃. [c] 20 mol% catalyst. [d] By GCMS. [e] From $[(NHC)AuCl_3]$.⁴⁷ ^[7] From $[IrCl(CO)_2(NHC)]$.⁵⁰

ble	2 Screening	of Silanes fo	or Hydrosil	vlative Env	ines Cv	clization ^[a]
NIC.		or Shanes n	or riyarosii	yiacive Lii		chiza cioni.

entr	y P1 ^z	Silane 2	P3 ^z	yield (%)	Syn:Anti
1		(EtO)₃SiH 2a	P3 ^{oxy} aa	66	> 95 :5
2		BnMe₂SiH 2c	P3 ^{oxy} ac	68	> 95 :5
3	P1 ^{oxy} a	PhMe₂SiH 2d	P3 ^{0xy} ad	71	> 95 :5
4		Ph₃SiH 2e	P3 ^{oxy} ae	67	> 95 :5
5		tBuMe₂SiH 2f	P3 ^{oxy} af	61	> 95 :5
6	Р1 ^{№н} а		P3 ^{NH} aa	68	98 :2 ^[b]
7	P1 ^{™s} a	(EtO)₃SiH 2a	РЗ ^{№м} аа	83	5: 95 ^[b]
8	P1 ^{№н} а		P3 ^{NH} ae	65	> 95 :5
9	P1 ^{№Ms} a	Ph₃SiH 2e	P3 [™] ae	82 ^[c]	8: 92

see scheme 1d for the reaction equation, enyne and product structures. See Experimental Section for procedure except otherwise indicated. [a] ¹⁷ .^{CI} was used as ligand. [b] Determined by GC-MS. [c] Major product structure was confirmed by XRD.

Overall, the product's syn-/anti-configurations control here is consistent with the related examples that we noted before. So, we prive that the diastereodivergent hydrosilylative enyne cyclization went through the same set of nickelacyclopentenes successfully when we employed a suitable choice of electronic activated bulky NHC and silane (Scheme 2, the pathway I vs II). Next, if might involve a regioselective silane metathesis at the nickelacyclopentene Csp³-Ni over the Csp²-Ni site according to the product structure we obtained. That high regioselectivity was nilar to the regioselective hydroacylation. That was presumably a result of the longer Csp³-Ni bond and the steric repulsion control based on NHC and silane substituents. The undesired oligomerizan and [2+2+2] reactivity was suppressed mainly by the NHC electronic and steric properties as shown in Table 1.

ble 3 Scope of the diastereodivergent hydrosilylative enyne cyclization by IPr^{CI}-Ni(0) catalyst and Et₃SiH **2b**.^[a]



Set 1a 1'3 Relations by propargyl Z R ¹ SiEt ₃ V'S' R ¹ SiEt ₃	∫SiEt₃	S ^e t 2 Internal Alk	yne				
P3 ^Z Syn: anti (yield) P3 ^Z Syn: anti (yield) P3 ^Z Syn. P3 ^{O×y} bb >95:5 (80%) P3 ^{O×y} cb 92:8 (73%) P3 ^{O×y} db 29:8 P3 ^{O×y} db 29:1 P3 ^{O×y} db 29:1 P3 ^{O×y} db 29:1 P3 ^{NH} db 29:1 P3 ^{NH} db 29:1 P3 ^{NMs} db 29:1 P3 ^{O×y} gb 29:1 P3 ^{NH} gb 29:1 P3 ^{NH} gb 29:1 P3 ^{NH} gb 29:1 P3 ^{NMs} gb 20:1 P3 ^{NMs} gb	: <i>anti</i> (yield) 95 :5 (79%) 96 :4 (91%) ^[4] 10: 90 (72%) 95 :5 (92%) 93 :7 (90%) ^[4] 3 : 97 (81%)	Z R ¹ p ₃ z syn: a P ₃ o ^x yhb 95 P ₃ ^{NMs} hb 63	Sil ⁿ ti (Y 5 (71 6 (51 94 (81	Et ₃ ^{V'S'} R ¹ i ^{eld})P3 ² %) ^{[^c]P3⁰ %)^[c]P3⁰ %)P3^N}	Z syn ^{xy} ib ^H ib ^{Ms} ib	95:5 93:7 8:92	SiEt ₃ (yi ^e ld) (70%) (52%) (94%)
Set 1b 1'4 Relations by Allyl Z R ¹ , SiEt ₃ v:s: R ¹ , SiEt ₃ v:	Et ₃		∕_Sii	Et ₃ v·s· R ¹		4	SiEt ₃
$\begin{array}{llllllllllllllllllllllllllllllllllll$	^{- an} ti (yi ^e ld) -5: 95 (80%) 	A4 ^z syn: a A4 ^{oxy} hb <5 A4 ^{NH} hb <5 A4 ^{NMs} hb 9	1 ⁱⁿ ti (Y ;: 95 (6 ;: 95 (6) 1 :9 (7	^{ield}) A4 ^z 58%) A4 ^o 51%) A4 ^N 70%) A4 ^N	syn ^{xy} ib ^H ib ^{Ms} ib	: an _{ti} <5: 95 6: 94 72 :28	(^{yield}) (87%) (62%) (94%)
Stereocenters Phr 1 SiEt3 V.S. Z 4 SiEt3 V.S. Z 4 SiEt3	A4 ^z syn: a A4 ^{NH} nb 1 [:] A4 ^{NMs} nb 87	ⁿ ti (yi ^e ld) 99 (71%) ^[^a] 7 [:] 13 (84%)	1a 1b 1¢	R ¹ Ph M ^e n pentyl	R ² H H H	R ³ H H H	R⁴ H H H
Stereocenters Ph 1 SIEt ₃ SIEt ₃ Ph 1 SIEt ₃ SIEt ₃ SIEt ₃	P3 ^Z syn: P3 ^{Oxy} jb >9 P3 ^{NH} jb 9 P3 ^{NMs} jb 2	anti (yi ^e ld) 95:5 (87%) 93:7 (62%) 4:76 (64%)	1d 1 ^e 1f	Cy ∕~₹ {∑≻+	н н н	н н н	н н н
Ph ⁺ SiEt ³ ^{V'S'} Ph ⁺ SiEt ₃ ^{V'S'}	P3 ^{Oxy} kb 9 P3 ^{NH} kb 9 P3 ^{NMs} kb 20	95:5 (68%) 95:5 (81%) 0:80 (92%)	•9 1h 1i 1j	n _p en _{tyl} Ph Ph	Me Me	Me H	H H
z vs' z vs' z $siEt_3$ $siEt_3$ $siEt_3$ $siEt_3$	P3 ^{NH} Ib 9 P3 ^{NMS} Ib 6	<mark>5</mark> :5 (78%) 9 4 (81%)	1k 1l 1n	Ph ि Ph	H H H	H We	Me H

e Experimental Section for general procedure, structures are shown in relative configuration. Syn-:Anti-selectivity and yield (in parenthesis) were determined by NMR and isolation except otherwise indicated. [a] By GCMS. [b] 20 mol% catalyst.

Table 4 Diastereoselective reduction and effect of R¹ by using IPr^{Me}-Ni(0) catalyst and (EtO)₃SiH 2a.

entry	R ¹	starting materials	product	yield (%)	3,4- syn:ant
1	Ph	P1 ^{0xya}		61%	88:12[a]
2		1,3-syn P3 ^{Oxyaa}	1,3-syn ⁻ P3 ^{Oxyaa-} H	69%	87:13 ^[b]
3		1,3-syn P3 ^{Oxyaa}		67%	67:33[c]
4	Ph	A1 ^{Oxya}		71%	83:17[a]
5		1,4-anti A4 ^{0xy} aa	1,4-anti A4 ^{0xyaa} H	87%	83:17[b]
6		1,4-anti A4 ^{0xy} aa		64%	12: 88 [c]
7	n-pentyl	P1 ^{Oxy} C	1,3-syn P3 ^{Oxyca} H	82%	90:10[a]
8		A1 ^{Oxy}	1,4-anti A40xyca H	76%	75:25[a]
9	Су	P1 ^{Oxy} d	1,3-syn P3 ^{Oxy} da H	77%	90:10[a]
10		A1 ^{Oxy} d	1,4-anti A4 ^{0xy} da H	68%	80:20[a]
	entry 1 2 3 4 5 6 7 8 9 10	entry R1 1 Ph 2 3 4 Ph 5 6 7 n-pentyl 8 9 9 Cy 10 10	entry R1 starting materials 1 Ph P1 ⁰ xva 2 1,3-syn P3 ⁰ xyaa 3 1,3-syn P3 ⁰ xyaa 4 Ph A1 ⁰ xya 5 1,4-anti/A4 ⁰ xyaa 6 1,4-anti/A4 ⁰ xyaa 7 n-pentyl 9 Cy 9 Cy 10 A1 ⁰ xyd	entry R1 starting materials product 1 Ph P1 ^{oxya} 1,3-syn [*] P3 ^{oxyaa} 1,4-anti A4 ^{oxy} aa	entry R1 starting materials product yield (%) 1 Ph P1 ^{oxya} 61% 2 1,3-syn P3 ^{oxyaa} 1,3-syn P3 ^{oxyaa} 61% 3 1,3-syn P3 ^{oxyaa} 1,3-syn P3 ^{oxyaa} 69% 4 Ph A1 ^{oxya} 71% 5 1,4-antf A4 ^{oxyaa} 1,4-antf A4 ^{oxyaa} 87% 6 1,4-antf A4 ^{oxyaa} 64% 7 n-pentyl P1 ^{oxyc} 1,3-syn P3 ^{oxyca} H 82% 8 A1 ^{oxyd} 1,4-antf A4 ^{oxyaa} 76% 9 Cy P1 ^{oxyd} 1,3-syn P3 ^{oxyda} H 77% 10 A1 ^{oxyd} 1,4-antf A4 ^{oxyda} 1,4-antf A4 ^{oxyda} H 68%

Structures are shown in relative configuration, selectivity was determined by NMR. [a] one-pot reaction by using P1^z or A1^z and 2a as substrates. [b] Rec iction by using isolated cyclization products P3^z or A4^z. [c] Reduction of the isolated cyclization products P3^z or A4^z by using cat. Pd/C H₂.

Given the above proposed mechanism, a one-pot reduction was examined to provide an additional stereocenter from the exocyclic olefin (Table 4). Again, **P1**^{Oxy}a and **A1**^{Oxy}a were selected as model substrates for such a development, but at this time (EtO)₃SiH **2a** and IPr^{Me} were used to provide the right balance in enyne oxidative cyclization and the silane oxidative addition reactivities for later on reduction. To our delight, that could be done easily under otherwise the same condition except for 24 hrs reaction time (entry 1 and 4).⁴⁹ This discovery offered an new option to prepare vicinal side chains at 3,4-positions with high syn-diastereoselectivity. That selectivity was presumably a direct result of minimizing undesired steric repulsions between the silylated side chain and the NHC-Ni catalyst. Additional control experiments using isolated 1,3-syn-P3^{Oxy}aa and 1,4-anti-A4^{Oxy}aa as substrates showed highly comparable reduction selectivities as the one-pot hydrosilylative enyne cyclization-reduction (Table 4, entry 1, 2, 4, 5), indicating that the reduction might occur independently and after the primary catalytic cycle. Interestingly, our reduction showed a much stronger 3,4-syn-selectivity preference than the traditional Pd/C catalyzed hydrogenation, showing a strong catalyst control character of our reduction (entry 3 and 6). Finally, enynes with different R¹ were examined briefly in the catalytic one-pot diastereoselective reduction (entry 1, 4, 7-9). The reactions proceeded smoothly, and the 3,4-*syn*-selectivity in **P1**^{Oxy} was higher than **A1**^{Oxy} in general.

Conclusions

In summary, highly efficient 1,3- and 1,4-diastereodivergent hydrosilvlative envne cyclizations were first established. The elecbnic effect of the NHC was found to be very crucial for the chemoselectivity (vs. alkene and alkyne direct hydrosilylation, and nyne [2+2+2] cycloadditions). Through the steric and electronic manipulations of the NHC and silane, the silane direct oxidative a Idition to Ni(0) was mitigated (Pathway I), and the enyne oxidave cyclization was preferred (Pathway II, NHC = IPr^{CI}). Such changes in reactivity preference first united our highly diastereovergent nickelacyclopentene formation strategy and the hydrosilvlative enyne cyclization (syn-:anti- up to 98:2 and 6:94). Also, ey formed the heterocycles bearing homoallylsilane (Pathway II) rather than vinylsilane (Pathway I) by Ni(0) catalyst for the first time (Homoallyl-:Vinyl-silane >95:5). Moreover, that change in activity preference has dramatically expanded the scope of the Ni(0) catalyzed hydrosilylative enyne cyclization, which was priarily limited to those with activated difluroalkene side chain and 5-member ring formation before. Finally, a one-pot reduction could occur using suitable NHC and silane (IPr^{Me} and (EtO)₃SiH), offering an option for adding additional stereocenter on the ring through NHC-Ni catalyst control. Further exploration along this line is now underway.

F (perimental

General procedure for the catalytic hydrosilylative enyne cyzations: 0.05 mmol IPr^{CI}-Ni(0) catalyst was generated in situ in a glove box from Ni(cod)₂ and IPr^{CI} in 2 mL toluene at r.t.. 0.5 mmol heteroenyne P1^z or A1^z and 4 mmol triethylsilane 2b in 1 mL toluene was added dropwise to the catalyst in 1 hr and stirred for 3 hrs. The reaction was worked up by following the general workup p ocedure. Yield and selectivity were determined by NMR except the vise indicated. The products were obtained and separated by column chromatography. Enyne homo-dimerization, oligomerization, alkyne hydrosilylation, and a small amount of isomerizaon were obtained in some ineffective cases. No vinylsilane was observed on the heterocycle in general.

General procedure for the catalytic 1-pot reduction of the hydrosilylative enyne cyclizations: Same as the above except IPr^{Me}-Ni(0) was used as catalyst, triethoxylsilane 2a was used as ane and extended reaction time to 24 hrs.

upporting Information

The supporting information for this article is available on the WWW under https://doi.org/10.1002/cjoc.2021xxxxx. X-Ray crystallographic data: CCDC 1972918 contain the data for this paper, and can be obtained from Cambridge Crystallographic Data Centre via www.ccdc.cam.ac.uk/data_request/cif.

Acknowledgement

This work was supported by Guangdong Provincial Key Laboratory of Catalysis (2020B121201002), NSFC (22071096) and SUS-Tech (Y01501808, Y01506014).

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(The following will be filled in by the editorial staff)

Manuscript received: XXXX, 2021

Manuscript revised: XXXX, 2021

Manuscript accepted: XXXX, 2021

Accepted manuscript online: XXXX, 2021

Version of record online: XXXX, 2021

Entry for the Table of Contents

Diastereodivergent Hydrosilylative Enyne Cyclization Catalyzed by NHC-Ni(o) Meng Yu,^a Xuefeng Yong,^b Weiwei Gao,^b Chun–Yu Ho^{*a,b} *Chin. J. Chem.* 2021, *39*, XXX—XXX. DOI: 10.1002/cjoc.202100XXX



R^{1 =} linear/cyclic alkyls, alkenyls, aryls & heteroaryls

R² = H and alkyl; Allow Contiguous & Skipped Stereocenters Synth. NHC Steric & Electronic Controlled Cyclization over Direct Hydrosilylation Accepted Article