

Original paper

Synthesis and biological evaluation of new C (4) heterofunctionalized monocyclic β -lactams derived from penicillin G

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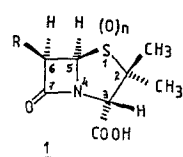
Summary — A simple and practical methodology has been developed for the preparation of monocyclic β -lactams **6** derived from penicillin G, possessing a thiocarbonyl (**6a**) or a sulfonyl (**6b**) substituent at C(4) as potentially good leaving groups. Two phthalidyl ester pro-drugs, **34f** (L = SO₂CH₂Ph) and **40f** (L = SO₂CH₂COOH), exhibited very weak anti-bacterial properties *in vitro*. However, no activity as β -lactamase inhibitors was detected for the two families of **6**.

Résumé — Synthèse et évaluation biologique de nouveaux β -lactames monocycliques hétérofonctionnalisés en C(4), dérivés de la pénicilline G. Une méthode simple et efficace a été développée pour la préparation de β -lactames monocycliques **6** dérivés de la pénicilline G, possédant en C(4) un substituant thioacétyle (**6a**) ou sulfonyle (**6b**) en tant que groupes nucléofuges potentiels. Deux esters biodégradables phthalidyles, **34f** (L = SO₂CH₂Ph) et **40f** (L = SO₂CH₂COOH), ont manifesté une faible activité antibiotique *in vitro*. Aucune inhibition de β -lactamases n'a été observée dans les deux familles de composés.

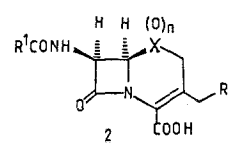
antibiotic / monocyclic β -lactam / penem / sulfone / suicide-inhibition

Introduction

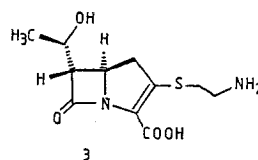
Penicillins **1a** and cephalosporins **2a**, the most popular drugs (for a review, see [1]) for the treatment of infections, are powerful inhibitors of the bacterial cell-wall D-D-peptidases (for a review, see [2]). They are characterized by a number of common structural features: a β -lactam ring bearing an acylamino side chain, fused with a sulfur containing heterocycle possessing an acidic residue. Recently, other related classes of interesting active bicyclic β -lactams have been discovered or synthesized, for instance, the carbapenems (thienamycin **3** [3]), the penems **4a** [4–11] and the 1-oxacephems **2d** [12, 13]. However, some monocyclic β -lactams also exhibit excellent bactericidal properties, as illustrated by the naturally occurring nocardicins **5a** [14] and monobactams **5b** [15] and the synthetic oxamazins **5c** [16]. This last discovery stimulated a growing interest in the preparation of new functionalized monocyclic β -



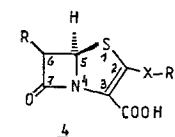
- 1 a** R = R¹CONH, n = 0
b R = R¹CONH, n = 1
c R = R¹CONH, n = 2
d R = H, n = 2



- 2 a** X = S, n = 0
b X = S, n = 1
c X = S, n = 2
d X = O, n = 0



(thienamycin)

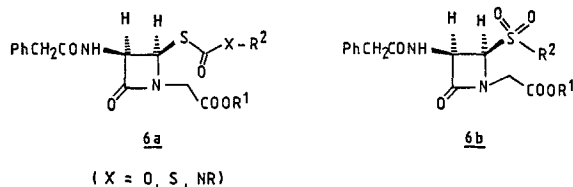
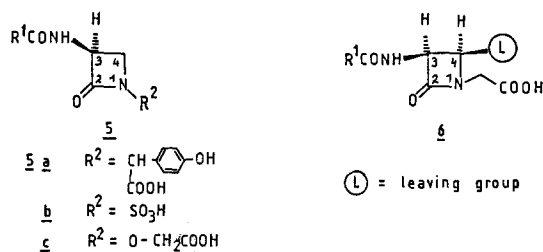


- 4 a** R = CH₃(OH)CH₂CH₂CH₂, X = S, O
b R = R¹CONH—, X = S, O, NR

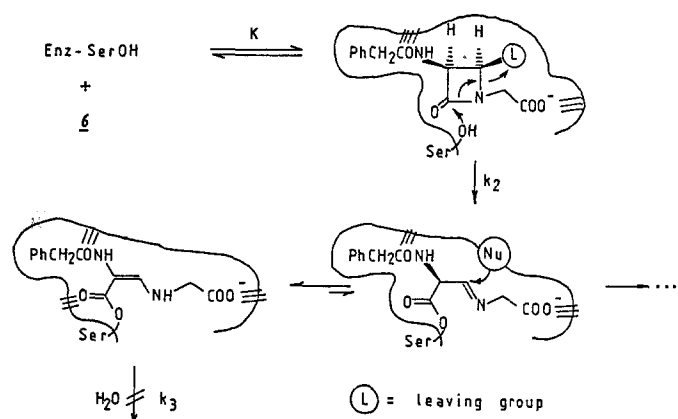
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Abbreviations: ALL: allyl; PNB: para-nitrobenzyl; TCE: trichloroethyl; PIV: pivaloyloxymethyl; TAL: phthalidyl; G: phenylacetamido side chain; DMF: dimethylformamide; MCPBA: meta-chloroperbenzoic acid; eq: equivalent; cpd: compound; sh: sharp; br: broad.

lactams as potential antibiotics. The chemical activation, necessary for biological activity [17], is usually obtained by placing different electron-withdrawing functional groups at the N(1) position [17–26]. Less attention has been devoted to the C(4) substituents. We therefore decided to investigate a series of chiral monocyclic β -lactams **6** bearing the appendages required for the enzymatic recognition at C(3) and N(1) and, moreover, possessing a heterofunctionalized group at C(4) in the *cis*-stereochemical relationship with regard to the C(3) side chain. This last functional group was designed to play an important role, not only as an electron-withdrawing—activating substituent, but moreover as a potential nucleofugal moiety. Indeed, the general structure **6** could represent the simplest model for a suicide-type inhibition of the bacterial D-D-peptidases (serine enzymes [2]), following the mechanism outlined in Scheme 1 and related to that of the classical β -lactamase inhibitors [27–28].



In this paper, we report the preparation and biological evaluation of two classes of compounds **6**, the thiocarbonyl-**6a** and the sulfonyl derivatives **6b**. The former can be considered as unstrained topological analogues of the C(2) heterosubstituted penems **4b** [29–32], in which the presence of an acylamino residue at C(6) dramatically enhances the chemical instability [33, 34], reducing their potential therapeutic applications. All the important pharmacophoric groups are maintained if one replaces the endocyclic (C2)—C(3) double bond in **4b** by an exocyclic carbonyl bond in **6a**, but the bicyclic strain is obviously lost. In the second class of compounds **6b**, the sulfur atom at C(4) has been oxidized. It is known that such an oxidation in penam or cephem derivatives can provide new biologically active materials. For instance, sulbactam **1d** [35] is a weak antibiotic but a powerful β -lactamase inhibitor. Some sulfones **1c** derived from penicillins are also active [36] against β -lactamases, whereas the (*R*)-sulfoxide **1b** of hetacillin [37] is reported to behave as an antibiotic. Similarly, the (*R*)- and (*S*)-sulfoxides **2b** [38, 39] and the sulfone



Scheme 1.

2c [39] derived from some cephalosporins exhibit good anti-bacterial properties.

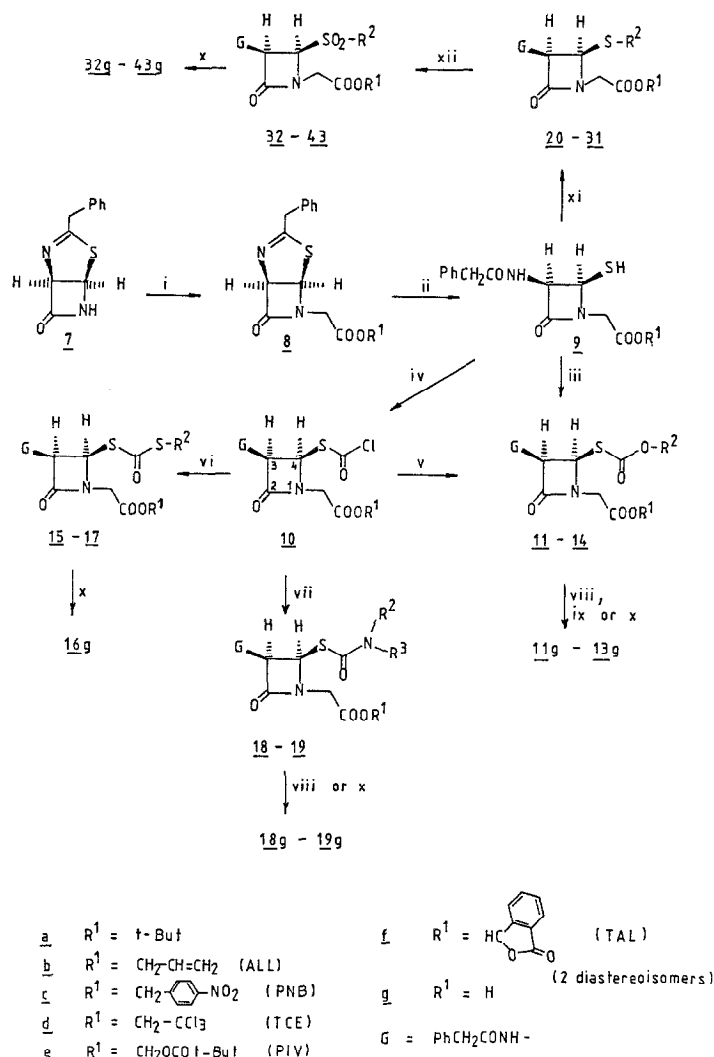
As a synthetic strategy, we planned to use penicillin G (**1a**, $R^1 = \text{PhCH}_2$) as a cheap and optically active starting material for the preparation of the β -lactams **6a, b**. Indeed, selective transformations of the thiazolidine ring can be achieved without destruction of the β -lactam moiety or alteration of the C(5)—C(6) stereochemistry.

Chemistry and Results

The thiazoline-azetidinone **7** (Scheme 2) is a useful chiral synthon derived from natural penicillin G by way of well-established methodologies [40–44]. Its *N*-alkylation with bromoacetic acid derivatives, leading to a variety of β -lactams **8**, has already been described [30]. Mild acid hydrolysis of **8** liberated the G side chain and the thiol group to furnish the compounds **9** as common precursors for the two families **6a** and **6b** (Scheme 2). Indeed, the thiol function could be readily acylated. For instance, reaction with ethylchloroformate [45], in the presence of triethylamine, gave the thiocarbonates **11** in good yield (Table I; **6a**, X = O). Attempts to deprotect the *t*-butyl ester **11a** under standard conditions [46] failed, giving rise to degradation of the small ring. The allyl ester **11b** could be cleaved [47] in the presence of Pd^0 —triphenylphosphine complex and 2-ethylhexanoic acid (or its K^+ salt). However, we were unable to extract pure acid **11g** from the reaction mixture. Hydrogenolysis of the *p*-nitrobenzyl ester **11c** with palladium on charcoal [33] gave the acid **11g** in moderate yield and required large amounts of catalyst. Finally, deprotection of the trichloroethyl ester **11d** with zinc dust in acetic acid and dimethylformamide [48] was found to be the best method for producing the expected acid **11g** with good yield and reasonable purity as ascertained from the ^1H NMR analysis (see Experimental protocols).

Similarly, acylation of the thiol **9** with benzylchloroformate produced the thiocarbonates **12** (Table I). Once again, the free acid was more conveniently prepared from the TCE ester **12d** than from the ALL ester **12b**.

The synthesis of more complex derivatives required the development of a new class of key-intermediates, the 4-

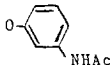


Scheme 2. Synthesis of the C(4)thiofunctionalized β -lactams **6**. Conditions and reagents: i: see [30]; ii: 1 N HCl, DMF, 0–20°C, 1 h; iii: ClCOOR² (1 eq), Et₃N (1 eq), CH₂Cl₂, –60–20°C; iv: COCl₂ (1 eq), Et₃N (1 eq), CH₂Cl₂, –60–20°C; v: R²OH (1 eq), Et₃N (1 eq), CH₂Cl₂, –60–20°C; vi: R²SH (1 eq), Et₃N (1 eq), CH₂Cl₂, –60–20°C; vii: R²R³NH (2 or 1 eq), Et₃N (0 or 1 eq), CH₂Cl₂, –60–20°C; viii: R¹ = ALL: Pd⁰ (Ph₃P)₄, 2-ethylhexanoic acid, 20°C; ix: R¹ = PNB: H₂, Pd-C, 20°C; x: R¹ = TCE: Zn, HOAc-DMF, 0°C; xi: R²-X (1–10 eq), Et₃N (1 eq), DMF, 0–20°C; xii: MCPBA (2–3 eq), CH₂Cl₂, 0–20°C.

thiochlorocarbonyl azetidinones **10**. These were easily prepared [49] by acylation of the thiols **9** with phosgene (Scheme 2). Reaction of **10** with *m*-acetamidophenol or 2-(α -pyridyl)ethanol afforded the thiocarbonates **13** and **14**, respectively, (Table I). From a practical point of view, it was not necessary to isolate the reactive intermediates **10**. Successive treatment of the thiols **9** with equimolar amounts of phosgene and triethylamine, followed by the appropriate alcohol and triethylamine, in a one-pot process, gave the 4-thiocarbonate azetidinones in moderate yield. Deprotection of the TCE esters furnished the acids **13g** and **14g**, of which the latter was found to be slightly stable.

The two other types of 4-thiofunctionalized β -lactams,

Table I. Yields of the C(4)thiocarbonyl β -lactams **6a**.

$X-R^2$	R^1	method (scheme 2)	cpd	yield(%)
$O-C_2H_5$	t-But	iii	<u>11a</u>	64 ^b
	ALL	iii	<u>11b</u>	73 ^a
	PNB	iii	<u>11c</u>	93 ^b
	TCE	iii	<u>11d</u>	82 ^b
	PIV	iii	<u>11e</u>	79 ^a
	TAL	iii	<u>11f</u>	71 ^b
	H	viii, ix, x	<u>11g</u>	72 ^c , 58 ^b , 83 ^b
$O-CH_2-$ 	ALL	iii	<u>12b</u>	79 ^a
	TCE	iii	<u>12d</u>	86 ^b
	PIV	iii	<u>12e</u>	77 ^a
	TAL	iii	<u>12f</u>	89 ^b
	H	viii, x	<u>12g</u>	74 ^c , 82 ^b
	TCE	iv+v	<u>13d</u>	31 ^a
	TAL	iv+v	<u>13f</u>	67 ^b
	H	x	<u>13g</u>	42 ^b
$O-(CH_2)_2-$ 	TCE	iv+v	<u>14d</u>	47 ^a
	TAL	iv+v	<u>14f</u>	30 ^a
	TCE	iv+vi	<u>15d</u>	38 ^a
	TAL	iv+vi	<u>15f</u>	40 ^a
	H	x	<u>15g</u>	28 ^b
$S-CH_2-$  $-OCH_3$	TCE	iv+vi	<u>16d</u>	59 ^a
	TAL	iv+vi	<u>16f</u>	66 ^b
	H	x	<u>16g</u>	65 ^b
$S-CH_2COOCH_3$	TAL	iv+vi	<u>17f</u>	26 ^a
$N(C_2H_5)_2$	ALL	iv+vi	<u>18b</u>	56 ^a
	TCE	iv+vi	<u>18d</u>	52 ^a
	TAL	iv+vi	<u>18f</u>	67 ^b
	H	viii, x	<u>18g</u>	64 ^c , 76 ^b
$NH(CH_2)_2-$ 	TCE	iv+vi	<u>19d</u>	33 ^a
	H	x	<u>19g</u>	56 ^b

^aAfter chromatography on silica gel.

^bAfter precipitation from ether or CH_2Cl_2 —ether.

^cImpure crude product.

the dithiocarbonates (Table I; **6a**, X = S) and the thiocarbonates (Table I; **6a**, X = NR), were prepared from the same key synthons **10**, using α -pyridylthiol (**15**), *p*-methoxybenzylthiol (**16**) or methyl thiolacetate (**17**) as nucleophiles, on the one hand, and diethylamine (**18**) or β -phenylethylamine (**19**), on the other. In all cases, reaction of the TCE esters with zinc was the best way to furnish the corresponding free acids.

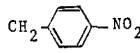
Several attempts to react isolated **10a** with bifunctional nucleophiles (*N*-(diethyl)ethylamine, ethanolamine, *o*-aminophenol, *m*-aminophenol, *p*-aminopyridine and 3-amino-1,2,4-triazole) gave untractable complex mixtures.

The direct precursors of the second class of target molecules, the sulfones **6b**, were obtained by alkylation of the thiols **9** (Scheme 2). Only a few examples of alkylation of a thiol function on β -lactam derivatives have been described in the literature [50–53]. We found that this reaction could be realized at 0°C or room temperature, with a variety of functionalized alkylhalides in dimethylformamide, as a polar solvent, and in the presence of triethylamine.

The yields of pure isolated products **20–31** were moderate to good (Table II). Treatment of the sulfide precursors with a slight excess (≥ 2 eq) of *m*-chloroperbenzoic acid in dichloromethane, at room temperature, afforded the corresponding sulfones **32–43** (Table II), which were separated from the *m*-chlorobenzoic acid by washing with bicarbonate solution and chromatography on silica gel or by fractional crystallization from ether. In one case (**40d**, $R^2 = \text{CH}_2\text{COOH}$), we were unable to achieve this last purification. All the protected sulfones ($R^1 \neq \text{H}$) are stable compounds, except the cyanomethyl derivative **43f** which decomposes slowly at room temperature.

The chemical cleavage of the trichloroethyl ester on

Table II. Yields of the C(4)thioalkyl precursors and the C(4)sulphonyl β -lactams **6b**.

R^2	R^1	sulphides		sulphones	
		cpd	%	cpd	%
CH_3	TCE	<u>20d</u>	63 ^a	<u>32d</u>	84 ^b
	TAL	<u>20f</u>	79 ^b	<u>32f</u>	86 ^b
	H			<u>32g</u>	57 ^b
$n\text{C}_3\text{H}_7$	TCE	<u>21d</u>	38 ^a	<u>33d</u>	59 ^a
	TAL	<u>21f</u>	10 ^a		
	H			<u>33g</u>	68 ^b
	TCE	<u>22d</u>	75 ^a	<u>34d</u>	89 ^a
	TAL	<u>22f</u>	77 ^b	<u>34f</u>	92 ^b
	H			<u>34g</u>	60 ^b
	TCE	<u>23d</u>	55 ^a	<u>35d</u>	78 ^b
	TAL	<u>23f</u>	75 ^b	<u>35f</u>	79 ^b
	H			<u>35g</u>	82 ^{b, d}
$\text{CH}_2=\text{CH}=\text{CH}_2$	TCE	<u>24d</u>	60 ^a	<u>36d</u>	56 ^a
	TAL	<u>24f</u>	43 ^a	<u>36f</u>	93 ^b
	H			<u>36g</u>	53 ^b
$\text{CH}_2=\text{C}\equiv\text{CH}$	TCE	<u>25d</u>	73 ^a	<u>37d</u>	76 ^a
	TAL	<u>25f</u>	77 ^b	<u>37f</u>	95 ^b
	H			<u>37g</u>	71 ^b
$\text{CH}_2-\text{COOC}_2\text{H}_5$	TCE	<u>26d</u>	65 ^a	<u>38d</u>	73 ^a
	TAL	<u>26f</u>	73 ^b	<u>38f</u>	86 ^b
	H			<u>38g</u>	63 ^b
$\text{CH}_2-\text{COOTCE}$	TCE	<u>27d</u>	29 ^a	<u>39d</u>	82 ^a
	H			<u>40g</u> ^c	82 ^b
CH_2-COOH	TCE	<u>28d</u>	81 ^b	<u>40d</u>	not isolated
	TAL	<u>28f</u>	90 ^b	<u>40f</u>	84 ^b
$\text{CH}_2-\text{CONH}(\text{CH}_2)_2$ 	TCE	<u>29d</u>	79 ^b	<u>41d</u>	83 ^b
	TAL	<u>29f</u>	85 ^b	<u>41f</u>	95 ^b
	H			<u>41g</u>	82 ^b
$\text{CH}_2-\text{CONH}_2$	TCE	<u>30d</u>	71 ^b	<u>42d</u>	81 ^a
	TAL	<u>30f</u>	80 ^b	<u>42f</u>	81 ^b
	H			<u>42g</u>	degradation
$\text{CH}_2-\text{C}\equiv\text{N}$	TCE	<u>31d</u>	82 ^b	<u>43d</u>	79 ^a
	TAL	<u>31f</u>	92 ^b	<u>43f</u>	83 ^{b, e}
	H			<u>43g</u>	67 ^b

^aAfter chromatography on silica gel.

^bAfter precipitation from ether.

^cDeprotection of the two TCE ester functions.

^dReduction of the nitro group; $R^2 = \text{CH}_2-\text{C}_6\text{H}_4-\text{NH}_2$.

^eImpure material.

the N(1) acetyl chain was achieved by treatment with zinc. Under these experimental conditions, the nitro group on the compound **35d** was also reduced and hence the *p*-aminobenzyl sulfone **35g'** ($R^2 = \text{CH}_2-\text{C}_6\text{H}_4-\text{NH}_2$) was recovered. In the case of **39d** ($R^2 = \text{CH}_2-\text{COOTCE}$), cleavage of the two ester functions furnished the acid **40g** ($R^2 = \text{CH}_2-\text{COOH}$) which was not directly accessible from **40d**. In one case (**42g**, $R^2 = \text{CH}_2-\text{CONH}_2$), the final deprotection led to some degradation of the β -lactam ring.

All the new compounds prepared were optically active materials possessing the penicillin's *cis*-relationship between the protons on the β -lactam ring as shown by the value of their coupling constant in the ^1H NMR spectra ($J = 4-5$ Hz, see Experimental protocols). Moreover, the high frequency stretching absorption of the β -lactam carbonyl, near 1785 cm^{-1} for **6a** and 1800 cm^{-1} for **6b**, in the IR spectra was similar to the values found in the penam or penem families.

The free acids and the biodegradable pivaloyloxymethyl (PIV) and phthalidyl (TAL) esters [54] were tested for biological activity against representative Gram-positive and Gram-negative bacterial strains. The acids **11g–13g**, **15g–16g**, **18g–19g**, **32g–43g** and the esters **11e–12e**, **11f–18f**, **32f**, **35f–38f**, **41f–43f**, did not show any significant bacteriostatic activity, even at high concentrations ($100-200\text{ }\mu\text{M}$). Two pro-drugs exhibited a very weak anti-bacterial effect: **34f** ($L = \text{SO}_2\text{CH}_2\text{Ph}$) was found to be active against *Staphylococcus aureus* ($\text{MIC}: 150\text{ }\mu\text{M}$; $\text{MBC}: 200\text{ }\mu\text{M}$) and *Streptococcus pyogenes* ($\text{MIC}: 75\text{ }\mu\text{M}$; $\text{MBC}: 150\text{ }\mu\text{M}$) and **40f** ($L = \text{SO}_2\text{CH}_2\text{COOH}$) against *S. aureus* ($\text{MIC}: 100\text{ }\mu\text{M}$; $\text{MBC}: 100\text{ }\mu\text{M}$). We have also established that their sulfide precursors **22f** ($L = \text{SCH}_2\text{Ph}$) and **28f** ($L = \text{SCH}_2\text{COOH}$) are devoid of anti-bacterial properties.

Since we had a series of new functionalized β -lactams on hand, it was of interest to test these compounds for activity as β -lactamase inhibitors. The acids **11g–13g**, **16g**, **18g–19g**, **34g–38g**, **40g–41g**, **43g** and the esters **11e–12e**, **34f**, **40f** were thus evaluated against three representative β -lactamases, one of each class A, B and C. No activity was detected at final concentrations of up to $100\text{ }\mu\text{M}$.

Conclusion

A simple and practical synthetic scheme has allowed the preparation of a series of monocyclic β -lactams **6**, functionalized at N(1) by an acetyl residue and at C(3) and C(4), respectively, by an acylamino side chain and a thiocarbonyl or sulfonyl moiety in a *cis*-stereochemical relationship. Thus, these compounds possess the minimum basic structural features of penicillins **1a** and cephalosporins **2a**, combined with the presence of a potentially good nucleofugal group at C(4) resulting from the acylation or oxidation of the sulfur atom. This latter substituent was designed to allow irreversible inhibition of the target enzymes of the classical bicyclic β -lactam antibiotics and, hence, to confer improved antibiotic properties on the monocyclic β -lactams **6**. Obviously, this assumption assumes that the enzymatic recognition process is similar for both **6** and the bicyclic β -lactams.

As expected, the compounds in the first class **6a** ($L = \text{SCOXR}^2$) are more stable than their corresponding bicyclic topological analogues, the penems **4b**. Indeed, we were able to deprotect the ester functional group on the N(1) acetyl chain in **6a** without destruction of the β -lactam ring, although this was not the case for the C(3) ester substituent in the C(2) heterosubstituted penems bearing a C(6) acylamino side chain [30]. Disappointingly, none of the new compounds **6a** prepared exhibited biological activity *in vitro*, neither as antibiotics nor as β -lactamase inhibitors. The β -lactams in the second class **6b** ($L = \text{SO}_2\text{-R}^2$) are also stable under the usual conditions of TCE ester deprotection. Two of them showed a definite but very low bactericidal activity *in vitro* against the Gram-positive strains. None of the compounds **6b** were found to be active against β -lactamases.

The biological activity of the penicillins is generally correlated with the chemical reactivity of their β -lactam function [1]. Examination of the IR frequency absorption [55] of the β -lactam carbonyl stretch in the new compounds **6** suggests that **6a** and particularly **6b** should possess the same chemical activation upon nucleophilic attack as the classic antibiotics, due to the presence of a strong electron-withdrawing substituent at C(4). We have thus compared the relative stability in the presence of hydroxylamine of one representative of each class of the β -lactams **6** and of one sulfide precursor as models of inactivated compounds (0.01 M solution of β -lactam, CH_3OH , excess H_2NOH , 20°C). **11a** ($L = \text{SCOOEt}$) and **34d** ($L = \text{SO}_2\text{-CH}_2\text{Ph}$) were found to react with half-lives of 285 and 2.5 min, respectively, whereas the sulfide **22d** ($L = \text{SCH}_2\text{Ph}$) was practically unreacted ($t_{1/2} \gg 10$ h). Under the same experimental conditions, the penicillin **1a** ($\text{R}^1 = \text{PhCH}_2$, Me ester) and the cephalosporin **2a** ($\text{R}^1 = \text{PhCH}_2$, $\text{R}^2 = \text{H}$, TCE ester [48]) had half-lives of 5.8 and 7.9 min, respectively. Hence, the thiocarbonyl derivatives **6a** can be considered as slightly activated β -lactams and the sulfones **6b** as strongly activated ones. Nevertheless, the loss of skeletal rigidity in **6**, as a result of opening the fused ring which is characteristic of the classical antibiotics, appears to dramatically handicap the biological properties. Variation of the acylamino chain at C(3) in the two weakly active compounds (**34f** and **40f**) is currently under investigation in order to confirm and eventually improve the bactericidal effect.

Experimental protocols

Chemistry

mps (Leitz microscope) are uncorrected. The rotations were determined ($\pm 1^\circ$) on a Perkin—Elmer 241 MC polarimeter and the IR spectra on a Perkin—Elmer 681 spectrometer (calibration with polystyrene) with CH_2Cl_2 as the solvent, unless otherwise stated. The ^1H NMR spectra (ppm) were recorded (CDCl_3 , unless otherwise stated, with tetramethylsilane (TMS) as the internal standard) on a Varian T60 (or XL200) instrument. Column chromatographies were performed with Merck silica gel 60 (70—230 mesh ASTM). CH_2Cl_2 , DMF and EtOAc were dried over P_2O_5 then distilled. Ether was dried over LiAlH_4 and Et_3N on KOH. Elemental analyses were carried out by the Continental Pharma Co. (Belgium). The results indicated by the symbols of the elements were within $\pm 0.4\%$ of the theoretical

values. The physical parameters of the compounds are summarized in Table III.

2-[3-benzyl-6-oxo-2-thia-4,7-diaza-(1R, 5R)-bicyclo-[3,2,0]-hept-3-en-7-yl]-alkyl acetates **8**

8a, **8b**, **8c**, **8d** and **8g** were prepared according to reference [30]. Similarly, **8e** and **8f** were obtained from esterification of **8g** with chloromethyl-pivalate or 2-carboxybenzaldehyde.

8e: 65%; IR: 1780 (br) cm^{-1} ; ^1H NMR (200 MHz) δ : 1.22 (s, 9), 3.76 (ABd, 1, $J = 18.3$ Hz), 3.82 (ABd, 1, $J = 15.1$ Hz), 3.98 (ABd, 1, $J = 15.1$ Hz), 4.30 (ABd, 1, $J = 18.3$ Hz), 5.74 (d, 1, $J = 4$ Hz), 5.76 (s, 2), 6.05 (d, 1, $J = 4$ Hz), 7.30 (sh m, 5); Anal. $\text{C}_{19}\text{H}_{22}\text{O}_5\text{N}_2\text{S}$ (C, H, N).

8f: 66%; IR: 1785 (br) cm^{-1} ; ^1H NMR δ : 3.84 (two ABd, 1, $J = 18$ Hz), 3.90 (br s, 2), 4.39 (ABd, 1, $J = 18$ Hz), 5.77 (two d, 1, $J = 4$ Hz), 6.06 (sh m, 1), 7.30 (s, 5), 7.40 (s, 1), 7.50—8.06 (m, 4); Anal. $\text{C}_{21}\text{H}_{16}\text{O}_5\text{N}_2\text{S}$ (C, H, N).

2-[3-Phenylacetamido-2-thiol-(3R, 4R)-azetidin-2-on-1-yl]-alkyl acetates **9**

9 in DMF solution (1 g/20 ml) was treated at 0°C with 1.2 N aq HCl (10 ml/g of **9**). After 1 h of stirring at room temp., the mixture was poured into cold water. The precipitate of **9** was filtered off or extracted with EtOAc, then dried as usual.

9a: 84%; IR: 3420 (NH), 2550 (SH), 1775, 1738, 1687, 1510 cm^{-1} ; ^1H NMR δ : 1.46 (s, 9), 1.90 (d, 1, $J = 10$ Hz, SH), 3.65 (s, 2), 3.66 (ABd, 1, $J = 18$ Hz), 4.10 (ABd, 1, $J = 18$ Hz), 5.03—5.66 (m, 2), 7.10 (d, 1, $J = 8$ Hz, NH), 7.30 (s, 5).

9b: 73%; IR: 3420, 2560, 1777, 1748, 1682, 1605, 1510 cm^{-1} ; ^1H NMR δ : 1.93 (d, 1, $J = 10$ Hz), 3.66 (s, 2), 3.79 (ABd, 1, $J = 18$ Hz), 4.26 (ABd, 1, $J = 18$ Hz), 4.66 (br d, 2, $J = 5$ Hz), 5.06—6.33 (m, 6), 7.15 (br d, 1, $J = 8.5$ Hz), 7.33 (s, 5).

9c: 87%; IR (KBr): 3280, 2540, 1763, 1738, 1655, 1610, 1645—20 cm^{-1} ; ^1H NMR (CDCl_3 — $\text{DMSO}-d_6$, 5:1) δ : 2.78 (d, 1, $J = 9.5$ Hz), 3.55 (s, 2), 3.91 and 4.33 (two ABd, 2, $J = 18$ Hz), 4.86—5.63 and 5.30 (m + s, 4), 7.26 (s, 5), 7.60 (d, 2, $J = 9$ Hz), 8.16 (d, 2, $J = 9$ Hz), 8.82 (br d, 1, $J = 8$ Hz).

9d: 80%; IR: 3420, 2550, 1780, 1765, 1687, 1510 cm^{-1} ; ^1H NMR (200 MHz, CD_3COCD_3) δ : 2.82 (d, 1, $J = 8$ Hz), 3.65 (s, 2), 4.06 (ABd, 1, $J = 18$ Hz), 4.38 (ABd, 1, $J = 18$ Hz), 4.90 (sh m, 3), 5.45 (d $\times$ d, 1, $J = 4.8$ and 8.5 Hz), 7.30 (sh m, 5), 8.02 (br d, 1, $J = 8.5$ Hz).

9e: 76%; IR: 3420, 2555, 1775 (br), 1680, 1610, 1512 cm^{-1} ; ^1H NMR (CDCl_3 + 5% DMF) δ : 1.23 (s, 9), 1.90 (d, 1, $J = 10$ Hz), 3.63 (s, 2), 3.80 and 4.20 (two ABd, 2, $J = 18$ Hz), 4.90—5.63 (m, 2), 5.80 (s, 2), 7.06 (br d, 1, $J = 8$ Hz), 7.30 (s, 5).

9f: 92%; IR (KBr): 3275, 2535, 1795—65 (br), 1650, 1607, 1543 cm^{-1} ; ^1H NMR ($\text{DMSO}-d_6$) δ : 3.54 (s, 2), 4.04 (m, 1), 4.34 (br ABd, 1, $J = 18$ Hz), 5.38 (m, 1), 5.82 (m, 1), 7.30 (br s, 5), 7.50 (s, 1), 7.66—8.00 (m, 4).

2-[3-Phenylacetamido-2-thiochlorocarbonyl-(3R,4R)-azetidin-2-on-1-yl]-alkyl acetates **10**

To a solution of **9** (1 eq) in dry CH_2Cl_2 (1 g/20 ml), stirred at -60°C under argon, phosgene (20% in toluene, 1.1 eq) and then Et_3N (1.1 eq) were added dropwise with a syringe, through a rubber stopper. The mixture was allowed to reach room temp. slowly and stirring was continued for 2—3 h. Washing rapidly with cold 0.05 N HCl, extraction with CH_2Cl_2 , drying of the organic phases (CaCl_2), concentration under vacuum and precipitation with dry ether, afforded crude **10** which was directly used for further reactions or stored in dry ice.

10a: 88%; IR: 3420, 1788, 1758, 1740, 1687, 1510, 840 cm^{-1} ; ^1H NMR δ : 1.49 (s, 9), 3.60 (ABd, 1, $J = 18$ Hz), 3.65 (s, 2), 4.20 (ABd, 1, $J = 18$ Hz), 5.38 (d $\times$ d, 1, $J = 5$ and 8 Hz), 5.75 (d, 1, $J = 5$ Hz), 6.63 (br d, 1, $J = 8$ Hz), 7.33 (s, 5).

10b: 68%; IR: 3420, 1790, 1755, 1687, 1510, 840 cm^{-1} ; ^1H NMR δ : 3.66 (s, 2), 3.76 (ABd, 1, $J = 18$ Hz), 4.35 (ABd, 1, $J = 18$ Hz), 4.70 (br d, 2, $J = 5.5$ Hz), 5.16—5.57 (m, 3), 5.66—6.20 and 5.80 (m + d, 2, $J = 5$ Hz), 6.83 (br d, 1, $J = 8$ Hz), 7.35 (s, 5).

10d: 85%; IR: 3420, 1790, 1768, 1688, 1605, 1508, 840 cm^{-1} ; ^1H NMR δ : 3.63 (s, 2), 3.86 (ABd, 1, $J = 18.5$ Hz), 4.45 (ABd, 1, $J = 18.5$ Hz), 4.80 (s, 2), 5.33 (d $\times$ d, 1, $J = 5$ and 7.5 Hz), 5.78 (d, 1, $J = 5$ Hz), 6.70 (br d, 1, $J = 7.5$ Hz), 7.33 (s, 5).

10e: 71%; IR: 3420, 1790, 1763, 1687, 1510, 840 cm^{-1} ; ^1H NMR δ : 1.21 (s, 9), 3.63 (s, 2), 3.75 (ABd, 1, $J = 18$ Hz), 4.31 (ABd, 1,

Table III. Physical parameters of the β -lactams.

Cpd	m.p.(°C)	$[\alpha]_D^{25}$ (°) CHCl ₃ , c(%)	R _F CH ₂ Cl ₂ -EtOAc(v:v)
8e	85-6	-58.6 (0.435)	0.50 (70:30) ^a
8f	156-9	-48.0 (1.59)	0.41 (80:20)
11a	143-6	+31.8 (0.56)	0.30 (65:35) ^a
11b	155-6	+32.2 (0.345)	0.44 (75:25)
11c	144-7	+19.0 (0.185)	0.35 (70:30)
11d	133-5	+30.4 (0.44)	0.56 (70:30)
11e	70-3	+34.6 (0.625)	0.26 (70:30) ^a
11f	200-9	+24.4 (0.255)	
12b	127-8	+28.1 (0.265)	0.46 (75:25)
12d	125-6	+27.3 (0.62)	0.50 (70:30)
12e	71-2	+35.1 (0.24)	0.26 (80:20) ^a
12f	173-5	+21.2 (0.36)	
13d	74-8	+14.6 (0.185)	0.23 (60:40)
13f	167-9	+18.2 (0.245)	0.21 (60:40)
14d	25-8	+27.7 (0.435)	0.26 (60:40)
14f	72-6	+28.6 (0.44)	0.22 (40:60)
15d	135-9	+45.8 (0.12)	0.51 (60:40)
15f	113-8	+56.0 (0.36)	0.42 (40:60)
16d	116-9	+50.9 (0.925)	0.33 (80:20)
16f	178-81	+48.7 (0.19)	0.45 (60:40)
17f	107-13	+25.2 (0.235)	0.47 (60:40)
18b	134-6	+73.7 (0.24)	0.54 (40:60) ^a
18d	99-100	+57.4 (0.20)	0.70 (40:60)
18f	192-5	+56.4 (0.26)	0.34 (50:50)
19d	44-7	+47.7 (0.21)	0.52 (70:30)
20d	107-9	-43.2 (0.575)	0.54 (70:30)
20f	176-80	-35.8 (0.150)	0.56 (70:30)
22d	148-9	-109.4 (0.43)	0.45 (80:20)
22f	195-7	-46.9 (0.265) ^b	0.43 (60:40)
23d	136-8	-105.4 (0.255)	0.36 (80:20)
23f	198-202	-36.4 (0.30) ^b	
24d	(foam)	-108.2 (0.20)	0.43 (80:20)
24f	150-3	-39.2 (0.56)	0.32 (70:30)

25d	112-4	-53.0 (1.21)	0.33 (80:20)
25f	180-5	-21.6 (0.29) ^b	0.35 (60:40)
26d	141-3	-52.1 (0.555)	0.58 (80:20)
26f	172-8	-27.6 (0.655) ^b	0.35 (60:40)
27d	153-5	-44.0 (1.10)	0.39 (80:20)
28d	117-20	-34.5 (0.76) ^b	
28f	93-9	-25.4 (0.66) ^b	
29d	151-5	-32.2 (0.35) ^b	0.34 (60:40)
29f	172-6	-21.9 (0.38) ^b	0.22 (40:60)
30d	135-40	-29.5 (0.255) ^b	
30f	195-8	-27.7 (0.24) ^b	
31d	153-4	-32.5 (0.50) ^b	
31f	125-30	-27.6 (0.405) ^b	0.39 (40:60)
32d	129-30	-9.6 (0.25)	
32f	237-40	+15.3 (0.40) ^b	
34d	150-2	-92.2 (0.335)	0.54 (80:20)
34f	261-9	-16.9 (0.62) ^b	
35d	204-8	-68.6 (0.53)	
35f	189-94	-33.8 (0.51) ^b	
36d	49-54	-14.2 (0.085)	0.44 (80:20)
36f	194-9	+1.4 (0.53) ^b	
37d	88-93	-57.7 (0.28)	0.58 (70:30)
37f	158-68	+4.2 (0.365) ^b	
38d	75-103	-42.7 (0.135)	0.65 (80:20)
38f	193-9	-4.0 (0.715) ^b	
39d	46-9	-27.4 (0.395)	0.60 (80:20)
40f	110-9	-3.2 (0.72) ^b	
41d	166-8	-0.03 (0.21) ^b	
41f	208-17	+21.3 (0.27) ^b	
42d	117-21	-16.8 (0.255)	0.59 (0:100)
42f	205-10	+16.0 (0.79) ^b	
43d	155-8	-10.8 (0.695) ^b	0.51 (70:30)

^aBenzene-EtOAc.^bDMF.

$J = 18$ Hz), 5.35 (d×d, 1, $J = 5$ and 7.5 Hz), 5.73 (d, 1, $J = 5$ Hz), 5.80 (s, 2), 6.88 (br d, 1, $J = 7.5$ Hz), 7.33 (s, 5).

10f: 89%; IR: 3420, 1795-60 (br), 1686, 1605, 1510, 840 cm⁻¹; ¹H NMR δ : 3.62 (s, 2), 3.86 (br ABd, 1, $J = 18.5$ Hz), 4.43 (ABd, 1, $J = 18.5$ Hz), 5.13-5.50 (m, 1), 5.77 (br d, 1, $J = 5$ Hz), 7.10 (br d, 1), 7.30 (s, 5), 7.38 (br s, 1), 7.50-8.03 (m, 4).

2-[3-Phenylacetamido-2-ethylthiocarbonate-(3R,4R)-azetidin-2-on-1-yl]-alkyl acetates **11**

To a solution of **9** (1 eq) in dry CH₂Cl₂ (1 g/20 ml), stirred at -60°C under argon, ethylchloroformate (1.1 eq) and then Et₃N (1.1 eq) were added dropwise with a syringe. The mixture was allowed to reach room temp. slowly and stirring was continued for 2 h. Washing with 0.05 N HCl, extraction with CH₂Cl₂, drying (CaCl₂) and concentration gave a crude product which was purified by column chromatography on silica gel or (and) precipitation from dry ether.

11a: IR: 3420, 1780, 1740, 1725, 1695, 1510 cm⁻¹; ¹H NMR δ : 1.26 (t, 3, $J = 7$ Hz), 1.46 (s, 9), 3.54 (ABd, 1, $J = 18$ Hz), 3.60 (s, 2), 4.13 (ABd, 1, $J = 18$ Hz), 4.23 (q, 2, $J = 7$ Hz), 5.53-5.76 (ABXm, 2), 6.74 (d, 1, $J = 8$ Hz), 7.30 (s, 5) after D₂O exchange, 5.60 (sh ABq, 2, $J = 5$ Hz); Anal. C₂₀H₂₆O₆N₂S (C, H, N).

11b: IR: 3418, 1782, 1750, 1722, 1695, 1508 cm⁻¹; ¹H NMR δ : 1.28 (t, 3, $J = 7$ Hz), 3.64 (s, 2), 3.71 (ABd, 1, $J = 18$ Hz), 4.30 (ABd, 1, $J = 18$ Hz), 4.25 (q, 2, $J = 7$ Hz), 4.68 (br d, 2, $J = 5$ Hz), 5.13-6.00 (m, 5), 6.50 (br d, 1), 7.35 (s, 5); Anal. C₁₉H₂₂O₆N₂S (C, H, N).

11c: IR: 3420, 1783, 1758, 1720, 1695, 1610, 1528, 1510 cm⁻¹; ¹H NMR δ : 1.28 (t, 3, $J = 7$ Hz), 3.63 (s, 2), 3.80 (ABd, 1, $J = 18$ Hz), 4.23 (q, 2, $J = 7$ Hz), 4.33 (ABd, 1, $J = 18$ Hz), 5.30 (s, 2), 5.40-

5.76 (ABXm, 2), 6.46 (br d, 1, $J = 7.5$ Hz), 7.33 (s, 5), 7.55 (d, 2, $J = 8.5$ Hz), 8.28 (d, 2, $J = 8.5$ Hz); Anal. C₂₃H₂₃O₆N₂S (C, H, N).

11d: IR: 3420, 1786, 1770, 1720, 1695, 1505 cm⁻¹; ¹H NMR δ : 1.28 (t, 3, $J = 7$ Hz), 3.64 (s, 2), 3.85 (ABd, 1, $J = 18$ Hz), 4.25 (q, 2, $J = 7$ Hz), 4.43 (ABd, 1, $J = 18$ Hz), 4.81 (s, 2), 5.40-5.83 (ABXm, 2), 6.45 (br d, 1, $J = 7.5$ Hz), 7.33 (s, 5); Anal. C₁₈H₁₉O₆N₂SCl₃ (C, H, N).

11e: IR: 3420, 1785, 1675, 1722, 1695, 1508 cm⁻¹; ¹H NMR δ : 1.23 (s, 9), 1.30 (t, 3, $J = 7$ Hz), 3.64 (s, 2), 3.76 (ABd, 1, $J = 19$ Hz), 4.30 (ABd, 1, $J = 19$ Hz), 4.26 (q, 2, $J = 7$ Hz), 5.45-5.80 (ABXm, 2), 5.83 (s, 2), 6.60 (br d, 1, $J = 8$ Hz), 7.33 (s, 5); Anal. C₂₂H₂₈O₆N₂S (C, H, N).

11f: IR: 3410, 1788 (br), 1720, 1696, 1608, 1505 cm⁻¹; ¹H NMR (CD₃COCD₃) δ : 1.29 (t, 3, $J = 7$ Hz), 3.65 (s, 2), 3.80-4.50 (m, 4), 5.36-5.90 (ABXm, 2), 7.33 (s, 5), 7.54 (s, 1), 7.88 (m, 4); Anal. C₂₄H₂₂O₈N₂S (C, H, N).

11g: General procedure for TCE deprotection. The TCE ester in DMF-HOAc (3:1) solution (100 mg/ml) was treated at 0°C with zinc dust (large excess) for 4 h under vigorous stirring. After dilution with EtOAc, the mixture was filtered and the solid residue washed twice with EtOAc. The filtrates were washed with 0.05 N HCl and the aqueous phase extracted with EtOAc. The collected organic phases were washed with brine, dried (CaCl₂) and concentrated under vacuum at 25°C. Precipitation from ether at 0°C gave the acid as a very hygroscopic white solid; IR (KBr): 3265, 3100-2900, 2600-2300, 1775, 1720-1710, 1658, 1530, 1417 cm⁻¹; ¹H NMR (CDCl₃ + 5% DMSO-d₆) δ : 1.30 (t, 3, $J = 7$ Hz), 3.63 (s, 2), 3.69 (ABd, 1, $J = 18$ Hz), 4.23 (ABd, 1, $J = 18$ Hz), 4.25 (q, 2, $J = 7$ Hz), 5.36-5.80 (ABXm, 2), 7.00 (br d, 1, $J = 7.5$ Hz), 7.32 (s, 5), 8.10 (sh m, 1); ¹H NMR (CDCl₃-DMSO-d₆, 1:1) δ : 5.43 (d×d, 1, $J = 5$ and 8 Hz) and 5.70 (d, 1, $J = 5$ Hz).

2-[3-Phenylacetamido-2-benzylthiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **12**

Using benzylchloroformate as the reagent, the procedure described for **11** was applied.

12b: IR: 3418, 1783, 1750, 1715, 1687, 1508 cm⁻¹; ¹H NMR δ : 3.61 (s, 2), 3.70 (ABd, 1, J = 18 Hz), 4.30 (ABd, 1, J = 18 Hz), 4.67 (br d, 2, J = 5 Hz), 5.22 (s, 2), 5.10–6.00 (m, 5), 6.45 (br d, 1, J = 8 Hz), 7.33 (s, 5), 7.40 (s, 5); Anal. C₂₄H₂₄O₆N₂S (C, H, N).

12d: IR: 3420, 1784, 1770, 1715, 1690, 1508 cm⁻¹; ¹H NMR δ : 3.62 (s, 2), 3.85 (ABd, 1, J = 18 Hz), 4.43 (ABd, 1, J = 18 Hz), 4.79 (s, 2), 5.23 (s, 2), 5.43–5.86 (ABXm, 2), 6.50 (br d, 1, J = 8 Hz), 7.33 (s, 5), 7.41 (s, 5); Anal. C₂₈H₂₁O₆N₂SCl₃ (C, H, N).

12e: IR: 3420, 1785, 1765, 1714, 1690, 1505 cm⁻¹; ¹H NMR δ : 1.21 (s, 9), 3.58 (s, 2), 3.71 (ABd, 1, J = 18 Hz), 4.28 (ABd, 1, J = 18 Hz), 5.20 (s, 2), 5.40–5.75 (ABXm, 2), 5.78 (s, 2), 6.63 (br d, 1, J = 8 Hz), 7.28 and 7.35 (s, 5); Anal. C₂₇H₃₀O₈N₂S (C, H, N).

12f: IR: 3420, 1788 (br), 1712, 1687, 1605, 1503 cm⁻¹; ¹H NMR (CD₃COCD₃) δ : 3.63 (s, 2), 3.96 (br ABd, 1, J = 19 Hz), 4.43 (ABd, 1, J = 19 Hz), 5.28 (s, 2), 5.40–5.87 (ABXm, 2), 7.32 (s, 5), 7.40 (s, 5), 7.52 (s, 1), 7.83 (sh m, 4); Anal. C₂₉H₂₄O₈N₂S (C, H, N).

12g: IR: 3420, 1780, 1720, 1680, 1650, 1610, 1510 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMF) δ : 3.60 (s, 2), 3.70 (ABd, 1, J = 18 Hz), 4.25 (ABd, 1, J = 18 Hz), 5.17 (s, 2), 5.43–5.83 (ABXm, 2), 6.93 (br d, 1, J = 7.5 Hz), 7.30 (s, 5), 7.40 (s, 5), 8.08 (sh m, 1).

2-[3-Phenylacetamido-2-(*m*-acetamido)phenylthiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **13**

To a solution of **10** (1 eq) in dry CH₂Cl₂ (1 g/20 ml), stirred at –60°C under argon, solid *m*-acetamidophenol (1 eq) and then Et₃N (1 eq) were added successively. The mixture was allowed to reach 20°C slowly and stirring continued for 2 h. Washing with 0.01 N HCl, extraction with CH₂Cl₂, drying (CaCl₂) and concentration gave a crude product which was purified by column chromatography on silica gel and (or) precipitation from ether.

13d: IR: 3427, 3330, 1787, 1770, 1730, 1690 (br), 1611, 1602, 1520 (br), 1492 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-*d*₆) δ : 2.11 (s, 3), 3.63 (s, 2), 4.00 (ABd, 1, J = 18.5 Hz), 4.50 (ABd, 1, J = 18.5 Hz), 4.83 (s, 2), 5.51 (d × d, 1, J = 5 and 7.5 Hz), 5.83 (d, 1, J = 5 Hz), 6.83 (m, 1), 7.33 (br s, 7), 7.66 (sh m, 1), 8.76 (d, 1, J = 7.5 Hz), 9.66 (br s, 1); Anal. C₂₄H₂₂O₇SN₃Cl₃ (C:46.01, H, N).

13f: IR: 3420, 3360, 1785 (br), 1725, 1695 (br), 1605, 1520 cm⁻¹; ¹H NMR (CDCl₃ + 20% DMSO-*d*₆) δ : 2.03 and 2.11 (two s, 3), 3.63 (br s, 2), 3.80–4.60 (m, 2), 5.53 (m, 1), 5.80 (br d, 1, J = 4 Hz), 6.83 (m, 1), 7.10–8.10 (m), 7.33 (s, 5), 7.50 (s, 1), 9.03 (br d, 1, J = 8 Hz), 9.83 (br s, 1); Anal. C₃₀H₂₅O₈SN₃ (C, H, N).

13g: IR (KBr): 3340, 3280, 1775, 1730, 1660, 1604, 1540, 1487, 1420 cm⁻¹; ¹H NMR (CDCl₃—DMSO-*d*₆, 1:1) δ : 2.08 (s, 3), 3.61 (s, 2), 4.00 (br m, 2), 5.48 (m, 1), 5.80 (d, 1, J = 5 Hz), 6.83 (m, 1), 7.33 (sh m, 7), 7.63 (br s, 1), 7.94 (br d, 1, J = 8 Hz).

2-[3-Phenylacetamido-2-β(*α*-pyridyl)ethylthiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **14**

General one-pot procedure from 9. A solution of thiol **9** (1 eq) in CH₂Cl₂ was treated at –60°C with phosphine (1 eq) and Et₃N (1 eq) as described previously. After 2 h at 20°C, the crude mixture was cooled again at –60°C. β-Pyridylethanol (1 eq) and Et₃N (1 eq) were added successively. The next work-up was conducted as described for compounds **13**.

14d: IR: 3415, 1788, 1770, 1710, 1687, 1594, 1510 cm⁻¹; ¹H NMR δ : 3.13 (t, 2, J = 7 Hz), 3.63 (s, 2), 3.82 (ABd, 1, J = 18 Hz), 4.41 (ABd, 1, J = 18 Hz), 4.60 (t, 2, J = 7 Hz), 4.80 (s, 2), 5.36–5.80 (ABXm, 2), 6.46 (d, 1, J = 7 Hz), 7.30 and 7.03–8.06 (s + m, 8), 8.57 (m, 1); Anal. C₂₃H₂₂O₆SN₃Cl₃ (C: 47.57, H, N).

14f: IR: 3420, 1790 (br), 1710, 1688, 1600, 1503, 1410 cm⁻¹; ¹H NMR δ : 3.11 (t, 2, J = 7 Hz), 3.61 (s, 2), 3.76 (ABd, 1, J = 19 Hz), 4.36 (ABd, 1, J = 19 Hz), 4.58 (t, 2, J = 7 Hz), 5.30–5.80 (ABXm, 2), 6.53 (br d, 1), 6.96–8.10 (m), 7.28 (s, 5), 7.40 (s, 1), 8.50 (m, 1); Anal. C₂₉H₂₅O₈SN₃ (C, H, N).

2-[3-Phenylacetamido-2-(*α*-pyridyl)dithiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **15**

The one-pot procedure described for **14** was applied using *α*-pyridylthiol as the reagent.

15d: IR: 3415, 1788, 1770, 1688 (br), 1640, 1575, 1510, 1450 cm⁻¹; ¹H NMR (CDCl₃—DMSO-*d*₆, 1:1) δ : 3.56 (s, 2), 3.97 (ABd, 1, J

= 19 Hz), 4.46 (ABd, 1, J = 19 Hz), 4.86 (s, 2), 5.50 (d × d, 1, J = 5 and 7.5 Hz), 5.91 (d, 1, J = 5 Hz), 7.32 (s, 5), 7.43–8.10 (m, 3), 8.66 (m, 1), 9.00 (br d, 1, J = 7.5 Hz); Anal. C₂₁H₁₈O₅S₂N₃Cl₃ (C, H, N).

15f: IR: 3418, 1790 (br), 1689, 1640, 1607, 1575, 1508, 1415 cm⁻¹; ¹H NMR δ : 3.61 (s, 2), 3.78 (ABd, 1, J = 19 Hz), 4.36 (ABd, 1, J = 19 Hz), 5.43 (d × d, 1, J = 5 and 7 Hz), 5.83 (d, 1, J = 5 Hz), 6.60–7.30 (m, 4), 7.27 (s, 5), 7.36 (s, 1), 7.45–8.00 (m, 4), 8.54 (m, 1); Anal. C₂₇H₂₁O₇S₂N₃ (C, H, N).

15g: IR (KBr): 3400–3200, 1775, 1730, 1662, 1530, 1452, 1420 cm⁻¹; ¹H NMR (CDCl₃—DMSO-*d*₆, 5:1) δ : 3.60 (s, 2), 3.63 (ABd, 1, J = 18 Hz), 4.20 (ABd, 1, J = 18 Hz), 5.50 (d × d, 1, J = 5 and 8 Hz), 5.85 (d, 1, J = 5 Hz), 7.30 (sh m), 7.53–8.13 (m), 8.56 (br d, 1, J = 8 Hz).

2-[3-Phenylacetamido-2-(*p*-methoxybenzyl)dithiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **16**

The one-pot procedure described for **14** was applied using *p*-methoxybenzylthiol as the reagent.

16d: IR: 3420, 1785, 1770, 1687, 1642, 1610, 1511, 1495 cm⁻¹; ¹H NMR δ : 3.61 (s, 2), 3.80 (s, 3), 3.78 (ABd, 1, J = 19 Hz), 4.16 (s, 2), 4.43 (ABd, 1, J = 19 Hz), 4.81 (s, 2), 5.55 (d × d, 1, J = 5 and 8 Hz), 5.88 (d, 1, J = 5 Hz), 6.42 (d, 1, J = 8 Hz), 6.86 (d, 2, J = 9 Hz), 7.23 (d, 2, J = 9 Hz), 7.33 (s, 5); Anal. C₂₄H₂₃O₆S₂N₃Cl₃ (C, H, N).

16f: IR: 3418, 1787 (br), 1685, 1640, 1610, 1510 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-*d*₆) δ : 3.61 (s, 2), 3.80 (s, 3), 3.86 (br ABd, 1, J = 18.5 Hz), 4.20 (br s, 2), 4.46 (ABd, 1, J = 18.5 Hz), 5.53 (m, 1), 5.93 (br d, 1, J = 4.5 Hz), 6.86 (br d, 2, J = 9 Hz), 7.06–8.06 (m), 7.33 (s, 5), 8.33 (m, 1); Anal. C₃₀H₂₆O₈S₂N₃ (C, H, N).

16g: IR (KBr): 3600–2800, 1770, 1740, 1660, 1640, 1610, 1512, 1415 cm⁻¹; ¹H NMR (CDCl₃—DMSO-*d*₆, 5:1) δ : 3.58 (s, 2), 3.63 (ABd, 1, J = 18 Hz), 3.80 (s, 3), 4.18 (s, 2), 4.23 (ABd, 1, J = 18 Hz), 5.50 (d × d, 1, J = 5 and 8 Hz), 5.95 (d, 1, J = 5 Hz), 6.85 (d, 2, J = 9 Hz), 7.25 (d, 2, J = 9 Hz), 7.32 (s, 5), 8.56 (br d, 1, J = 8 Hz); Anal. C₂₂H₂₂O₆S₂N₃ (C, H, N).

2-[3-Phenylacetamido-2-(methylacetate)dithiocarbonate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetate **17**

The procedure described for **13** was applied using thiol-methylacetate as the reagent.

17f: IR: 3420, 1788 (br), 1745, 1688, 1655, 1500 cm⁻¹; ¹H NMR δ : 3.63 (s, 2), 3.76 (br s, 5), 3.80 (ABd, 1, J = 18 Hz), 4.40 (ABd, 1, J = 18 Hz), 5.50 (m, 1), 5.86 (br d, 1, J = 5 Hz), 6.50 (m, 1), 7.30 (s, 5), 7.40 (s, 1), 7.47–8.00 (m, 4); Anal. C₂₅H₂₂O₆S₂N₃ (C, H, N).

2-[3-Phenylacetamido-2-(*N*-diethyl)thiocarbamate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **18**

The procedure described for **13** was applied using diethylamine (2 eq) as the reagent and base.

18b: IR: 3420, 1778, 1750, 1685, 1650, 1510, 1412 cm⁻¹; ¹H NMR δ : 1.13 (t, 6, J = 7 Hz), 3.33 (br q, 4, J = 7 Hz), 3.65 (s, 2), 3.76 (ABd, 1, J = 18 Hz), 4.31 (ABd, 1, J = 18 Hz), 4.68 (br d, 2, J = 6 Hz), 5.10–6.00 (m, 5), 6.70 (m, 1), 7.33 (s, 5); Anal. C₂₁H₂₇O₅N₃S (C, H, N).

18d: IR: 3420, 1782, 1768, 1688, 1650, 1510, 1412 cm⁻¹; ¹H NMR δ : 1.13 (t, 6, J = 7 Hz), 3.30 (br q, 4, J = 7 Hz), 3.65 (s, 2), 3.86 (ABd, 1, J = 18 Hz), 4.43 (ABd, 1, J = 18 Hz), 4.81 (s, 2), 5.43–5.83 (m, 2), 6.50 (br d, 1, J = 8 Hz), 7.33 (s, 5); Anal. C₂₀H₂₄O₅N₃SCl₃ (C, H, N).

18f: IR: 3418, 1795–1770 (br), 1686, 1650, 1508, 1411, 980 cm⁻¹; Anal. C₂₆H₂₇O₇SN₃ (C, H, N).

18g: IR (KBr): 3280, 1775, 1743, 1714, 1660, 1623, 1528, 1415, 1260 cm⁻¹; ¹H NMR (CDCl₃ + 5% DMF) δ : 1.10 (m, 6), 3.26 (m, 4), 3.64 (s, 2), 3.76 (ABd, 1, J = 18 Hz), 4.33 (ABd, 1, J = 18 Hz), 5.43–5.96 (m, 2), 7.30 (br s, 6 + 1), 8.00 (m, 1).

2-[3-Phenylacetamido-2-*N*(β-phenylethyl)thiocarbamate-(3*R*,4*R*)-azetidin-2-on-1-yl]-alkyl acetates **19**

The procedure described for **13** was applied using β-phenylethylamine as the reagent.

19d: IR: 3420, 3340, 1783, 1768, 1685, 1500, 1412 cm⁻¹; ¹H NMR δ : 2.76 (br t, 2, J = 7 Hz), 3.46 (m, 2), 3.60 (s, 2), 3.80 (ABd, 1, J = 18 Hz), 4.38 (ABd, 1, J = 18 Hz), 4.78 (s, 2), 5.43–5.86 (m, 3), 6.66 (m, 1), 7.30 (br s, 10); Anal. C₂₄H₂₄O₅N₃SCl₃ (C, H, N).

19g: IR (KBr): 3280, 1765 (br), 1660 (br), 1605, 1525 (br), 1495, 1220, 1200 cm⁻¹; ¹H NMR (CDCl₃ + 5% DMF) δ : 2.7 (m, 2),

3.36 (m, 2), 3.50 (s, 2), 3.63 (ABd, 1, $J = 18$ Hz), 4.11 (ABd, 1, $J = 18$ Hz), 5.26–5.70 (m, 2), 6.43 (m, 1), 7.03 (br s, 10), 8.00 (m, 1).

2-[3-Phenylacetamido-2-thioalkyl-(3R,4R)-azetidin-2-on-1-yl]-alkyl acetates

A solution of the thiol **9** (1 eq) in dry DMF (100 mg/ml) was treated at 0°C with the alkylating reagent (1–10 eq) and then with triethylamine (dropwise addition, with a syringe, through a rubber stopper, 1 eq in all cases except 2 eq with iodoacetic acid). The mixture was stirred for 1 h at 0°C and 6–7 h at 20°C. After dilution with EtOAc, the solution was washed with 0.05 N HCl and the aqueous phase extracted twice with EtOAc. The organic layers were washed with brine, dried (MgSO₄) and concentrated under vacuum. The crude product was purified by column chromatography on silica gel or (and) precipitation from dry ether.

Methyliodide (10 eq)

20d: IR: 3420, 1780, 1765, 1684, 1510 cm⁻¹; ¹H NMR δ : 1.89 (s, 3), 3.66 (s, 2), 3.86 (ABd, 1, $J = 18$ Hz), 4.41 (ABd, 1, $J = 18$ Hz), 4.80 (s, 2), 5.03 (d, 1, $J = 4.5$ Hz), 5.56 (d $\times$ d, 1, $J = 4.5$ and 9 Hz), 6.50 (br d, 1, $J = 9$ Hz), 7.33 (s, 5); Anal. C₁₆H₁₇O₄N₃SCl₃ (C, H, N).

20f: IR: 3420, 1790–1770 (br), 1683, 1510, 1410 cm⁻¹; ¹H NMR δ : 1.90 (s, 3), 3.66 (s, 2), 3.83 (ABd, 1, $J = 18$ Hz), 4.40 (ABd, 1, $J = 18$ Hz), 5.06 (d, 1, $J = 4.5$ Hz), 5.60 (d $\times$ d, 1, $J = 4.5$ and 9 Hz), 6.56 (br d, 1, $J = 9$ Hz), 7.34 (s, 5), 7.56–8.10 (m, 4), 7.45 (s, 1); Anal. C₂₂H₂₀O₆N₃S (C, H, N).

n-Propyliodide (10 eq)

21d: IR: 3405, 1775 (br), 1680, 1510, 1495 cm⁻¹; ¹H NMR δ : 0.96 (br t, 3), 1.57 (m, 2), 2.40 (m, 2), 3.66 (br s, 2), 4.13 (m, 2), 4.80 (s, 2), 5.10 (d, 1, $J = 4.5$ Hz), 5.52 (d $\times$ d, 1, $J = 4.5$ and 8 Hz), 6.76 (br d, 1), 7.33 (s, 5).

Benzylbromide (10 eq)

22d: IR: 3420, 1780, 1767, 1685, 1605, 1510 cm⁻¹; ¹H NMR δ : 3.43 (ABd, 1, $J = 18.5$ Hz), 3.58 (s, 2), 3.63 (s, 2), 4.17 (ABd, 1, $J = 18.5$ Hz), 4.67 (s, 2), 5.03 (d, 1, $J = 4.5$ Hz), 5.51 (d $\times$ d, 1, $J = 4.5$ and 9 Hz), 6.33 (br d, 1, $J = 9$ Hz), 7.25 (s, 5), 7.30 (s, 5); Anal. C₂₂H₂₁O₄N₃SCl₃ (C, H, N).

22f: IR: 3420, 1790–1770 (br), 1686, 1607, 1512, 1497, 1410 cm⁻¹; ¹H NMR (DMSO-d₆) δ : 3.55 (s, 2), 3.76 (s, 2), 3.53 (ABd, 1, $J = 18.5$ Hz), 4.26 (ABd, 1, $J = 18.5$ Hz), 5.06–5.50 (ABXm, 2), 7.30 (br s, 10), 7.50 (s, 1), 7.86 (sh m, 4), 8.96 (br d, 1, $J = 7$ Hz); Anal. C₂₈H₂₄O₆N₃S (C, H, N).

p-Nitrobenzylbromide (1.2 eq)

23d: IR: 3420, 1780, 1765, 1683, 1605, 1520, 1347 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-d₆) δ : 3.70 (s, 4), 3.75 (ABd, 1, $J = 18$ Hz), 4.36 (ABd, 1, $J = 18$ Hz), 4.80 (s, 2), 5.08 (d, 1, $J = 4.3$ Hz), 5.50 (d $\times$ d, 1, $J = 4.3$ and 8 Hz), 7.40 (s, 5), 7.43 (d, 2, $J = 9$ Hz), 8.06 (br d, 1, $J = 8$ Hz), 8.20 (d, 2, $J = 9$ Hz); Anal. C₂₂H₂₀O₆N₃SCl₃ (C, H, N).

23f: IR (KBr): 3280, 1775 (br), 1660, 1605, 1540, 1520, 1346 cm⁻¹; ¹H NMR (DMSO-d₆) δ : 3.60 (s, 2), 3.93 (br s, 2), 4.20 (ABq, 2), 5.23 (ABXm, 2), 7.36 (s, 5), 7.53 (s, 1), 7.58 (d, 2, $J = 8$ Hz), 7.90 (br s, 4), 8.20 (d, 2, $J = 8$ Hz), 9.06 (br d, 1, $J = 7$ Hz); Anal. C₂₈H₂₈O₈N₃S (C, H, N).

Allylbromide (10 eq)

24d: IR: 3410, 1780, 1770, 1682, 1510, 1490 cm⁻¹; ¹H NMR δ : 3.02 (d, 2, $J = 6$ Hz), 3.66 (s, 2), 3.85 (ABd, 1, $J = 18$ Hz), 4.41 (ABd, 1, $J = 18$ Hz), 4.80 (s, 2), 4.93–5.76 (m, 5), 6.33 (br d, 1), 7.33 (s, 5); Anal. C₁₈H₁₉O₄N₃SCl₃ (C, H, N).

24f: IR: 3420, 1790–1770 (br), 1685, 1608, 1510, 1410 cm⁻¹; ¹H NMR δ : 3.02 and 3.05 (two d, 2, $J = 6$ Hz), 3.66 (s, 2), 3.76 and 3.85 (two ABd, 1, $J = 18$ Hz), 4.33 and 4.38 (two ABd, 1, $J = 18$ Hz), 4.88–5.33 (m, 3), 5.37–5.83 (m, 2), 6.26 (m, 1), 7.36 (s, 5), 7.41 and 7.43 (two s, 1), 7.56–8.13 (m, 4); Anal. C₂₄H₂₂O₆N₃S (C, H, N).

Propargylbromide (10 eq)

25d: IR: 3420, 3305, 1780, 1765, 1685, 1510, 1410 cm⁻¹; ¹H NMR δ : 2.26 (t, 1, $J = 2.7$ Hz), 3.16 (d, 2, $J = 2.7$ Hz), 3.66 (s, 2), 3.99 (ABd, 1, $J = 19$ Hz), 4.48 (ABd, 1, $J = 19$ Hz), 4.83 (s, 2), 5.33 (d, 1, $J = 4.5$ Hz), 5.60 (d $\times$ d, 1, $J = 4.5$ and 8 Hz), 6.83 (br d, 1, $J = 8$ Hz), 7.36 (s, 5); Anal. C₁₈H₁₇O₄N₃SCl₃ (C, H, N).

25f: IR: 3420, 3300, 1785 (br), 1685, 1605, 1510, 1410 cm⁻¹; ¹H NMR (CD₃CN + 15% DMSO-d₆) δ : 2.60 (t, 1, $J = 2.7$ Hz), 3.30 (d, 2, $J = 2.7$ Hz), 3.60 (s, 2), 4.25 (two sh ABq, 2, $J = 18$ Hz), 5.36 (ABXm, 2), 7.36 (s, 5), 7.51 (s, 1), 7.90 (sh m, 4), 8.40 (br d, 1); Anal. C₂₄H₂₀O₆N₃S (C, H, N).

(Ethyl)bromoacetate (5 eq)

26d: IR: 3420, 1780, 1765, 1735, 1687, 1610, 1415 cm⁻¹; ¹H NMR δ : 1.26 (t, 3, $J = 7$ Hz), 3.16 (s, 2), 3.66 (s, 2), 4.00 (ABd, 1, $J = 18$ Hz), 4.20 (q, 2, $J = 7$ Hz), 4.45 (ABd, 1, $J = 18$ Hz), 4.81 (s, 2), 5.28 (d, 1, $J = 4$ Hz), 5.53 (d $\times$ d, 1, $J = 4$ and 8.5 Hz), 6.56 (br d, 1, $J = 8.5$ Hz), 7.33 (s, 5); Anal. C₁₉H₂₁O₆N₃SCl₃ (C, H, N).

26f: IR: 3420, 1785 (br), 1733, 1685, 1607, 1510 cm⁻¹; ¹H NMR (CDCl₃ + 20% DMSO-d₆) δ : 1.26 (t, 3, $J = 7$ Hz), 3.31 (s, 2), 3.64 (s, 2), 3.86–4.73 (m, 4), 5.16–5.60 (ABXm, 2), 7.40 (s, 5), 7.53 (br s, 1), 7.73–8.10 (sh m, 4), 8.90 (br d, 1, $J = 7$ Hz); Anal. C₂₅H₂₄O₈N₃S (C, H, N).

(Trichloroethyl)iodoacetate (1.1 eq)

27d: IR: 3420, 1782, 1766, 1685, 1510, 1410 cm⁻¹; ¹H NMR δ : 3.23 (s, 2), 3.62 (s, 2), 3.92 (ABd, 1, $J = 18$ Hz), 4.36 (ABd, 1, $J = 18$ Hz), 4.70 (s, 2), 4.73 (s, 2), 5.23 (d, 1, $J = 4.5$ Hz), 5.43 (d $\times$ d, 1, $J = 4.5$ and 8 Hz), 6.10 (m, 1), 7.27 (s, 5); Anal. C₁₉H₁₈O₆N₃SCl₆ (C, H, N).

Iodoacetic acid (1.1 eq)

28d: IR (KBr): 3300, 1770, 1756, 1710, 1669, 1537, 1420 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ : 3.33 (s, 2), 3.53 (s, 2), 4.16 (ABd, 1, $J = 18.4$ Hz), 4.44 (ABd, 1, $J = 18.4$ Hz), 4.94 (sh ABq, 2, $J = 12$ Hz), 5.19 (d, 1, $J = 4.6$ Hz), 5.27 (d $\times$ d, 1, $J = 4.6$ and 8.2 Hz), 7.28 (sh m, 5), 9.02 (d, 1, $J = 8.2$ Hz); Anal. C₁₇H₁₇O₆N₃SCl₃ (C, H, N).

28f: IR (KBr): 3650–3150, 1780–1720 (br), 1663, 1540, 1410 cm⁻¹; ¹H NMR (CDCl₃–DMSO-d₆, 1:1) δ : 3.23 (s, 2), 3.58 (s, 2), 4.10 (ABd, 1, $J = 18$ Hz), 4.41 (ABd, 1, $J = 18$ Hz), 5.13–5.50 (sh m, 2), 7.30 (s, 5), 7.47 (s, 1), 7.80 (sh m, 4), 8.83 (br d, 1, $J = 7$ Hz); Anal. C₂₃H₂₀O₈N₃S (C: 55.87, H, N).

N-(Phenylethyl)iodoacetamide (1 eq)

29d: IR: 3420, 1782, 1768, 1675 (br), 1510, 1410 cm⁻¹; ¹H NMR (CDCl₃ + 20% DMSO-d₆) δ : 2.80 (br t, 2, $J = 7$ Hz), 3.12 (s, 2), 3.43 (m, 2), 3.63 (s, 2), 4.06 (ABd, 1, $J = 18.5$ Hz), 4.46 (ABd, 1, $J = 18.5$ Hz), 4.85 (s, 2), 5.16–5.53 (ABXm, 2), 7.27 (s, 5), 7.33 (s, 5), 7.83 (m, 1), 8.83 (br d, 1); Anal. C₂₅H₂₆O₅N₃SCl₃ (C, H, N).

29f: IR: 3420, 3300, 1785 (br), 1675 (br), 1605, 1510, 1410 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ : 2.70 (br t, 2, $J = 7.2$ Hz), 3.16 (br s, 2), 3.29 (sh m, 2), 3.54 (s, 2), 4.08 and 4.1 (two ABd, 1, $J = 18.2$ Hz), 4.40 (br ABd, 1, $J = 18.2$ Hz), 5.18–5.34 (ABXm, 2), 7.12–7.40 (m, 10), 7.50 and 7.52 (two s, 1), 7.70–7.96 (m, 4), 8.16 (m, 1), 9.00 (d, 1, $J = 8$ Hz); Anal. C₃₁H₂₉O₇N₃S (C, H, N).

Iodoacetamide (1 eq)

30d: IR (KBr): 3300, 3200, 1775 (br), 1660 (br), 1530 (br), 1410 cm⁻¹; ¹H NMR (CDCl₃–DMSO-d₆, 1:1) δ : 3.16 (s, 2), 3.60 (s, 2), 4.31 (sh ABq, 2), 4.93 (s, 2), 5.36 (ABXm, 2), 7.34 (s, 5), 7.53 (m, 2), 9.00 (br d, 1); Anal. C₁₇H₁₈O₅N₃SCl₃ (C: 41.67, H, N).

30f: IR (KBr): 3280, 3210, 1775 (br), 1665 (br), 1610, 1535, 1410 cm⁻¹; ¹H NMR (DMSO-d₆) δ : 3.20 (s, 2), 3.60 (s, 2), 4.33 (sh ABq, 2), 5.33 (ABXm, 2), 7.13 (m, 2), 7.36 (s, 5), 7.60 (s, 1), 7.96 (br s, 4), 9.10 (br d, 1); Anal. C₂₃H₂₁O₇N₃S (C, H, N).

Iodoacetonitrile (1 eq)

31d: IR (KBr): 3420, 2248 (w), 1785, 1765, 1684, 1510, 1410 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-d₆) δ : 3.26 (s, 2), 3.65 (s, 2), 4.06 (ABd, 1, $J = 19$ Hz), 4.53 (ABd, 1, $J = 19$ Hz), 4.86 (s, 2), 5.23–5.56 (ABXm, 2), 7.36 (s, 5), 8.70 (br d, 1); Anal. C₁₇H₁₆O₄N₃SCl₃ (C, H, N).

31f: IR: 3420, 2245 (w), 1785 (br), 1685, 1605, 1510, 1410 cm⁻¹; ¹H NMR (CD₃CN + 10% DMSO-d₆) δ : 3.46 (s, 2), 3.60 (s, 2), 4.05 (ABd, 1, $J = 18$ Hz), 4.43 (ABd, 1, $J = 18$ Hz), 5.30 (ABXm, 2), 7.34 (s, 5), 7.46 (s, 1), 7.83 (br s, 4), 8.40 (m, 1); Anal. C₂₃H₁₉O₆N₃S (C, H, N).

2-[3-Phenylacetamido-2-alkylsulfonyl-(3R,4R)-azetidin-2-on-1-yl]-alkyl acetates and acetic acids

A solution of the sulfide precursor (1 eq) in dry CH₂Cl₂ (100 mg/2–5 ml) was treated at 0°C with MCPBA (80% purity, 3 eq; addition

of the solid in small portions). The mixture was stirred for 1 h at 0°C and 2–3 h at 20°C. Work-up A: dilution with CH₂Cl₂, washing with 5% NaHCO₃, drying (CaCl₂) and concentration under vacuum gave the crude product which was purified by column chromatography on silica gel or (and) precipitation from ether. Work-up B: removal of the solvent under vacuum and precipitation from ether yielded the sulfone which was washed twice with ether.

Methylsulfones

32d: IR: 3410, 1800, 1768, 1693, 1510, 1412 cm⁻¹; ¹H NMR δ: 2.46 (s, 3), 3.65 (s, 2), 4.18 (ABd, 1, *J* = 18.5 Hz), 4.73 (ABd, 1, *J* = 18.5 Hz), 4.80 (s, 2), 5.10 (d, 1, *J* = 4.5 Hz), 5.89 (d×d, 1, *J* = 4.5 and 9.5 Hz), 7.10 (br d, 1, *J* = 9.5 Hz), 7.33 (s, 5); Anal. C₁₆H₁₇O₆N₂SCl₃ (C, H, N).

32f: IR (KBr): 3320, 1790–1770 (br), 1670, 1535, 1410 cm⁻¹; ¹H NMR (DMSO-d₆) δ: 2.93 (s, 3), 3.63 (s, 2), 4.46 (sh ABq, 2), 5.16–5.93 (m, 2), 7.33 (s, 5), 7.55 (s, 1), 7.86 (br s, 4); Anal. C₂₂H₂₀O₈N₂S (C, H, N).

32g: IR (KBr): 3305, 1790, 1745, 1676, 1533, 1390 cm⁻¹; ¹H NMR (CDCl₃–DMSO-d₆, 3:1) δ: 2.66 (s, 3), 3.60 (s, 2), 3.86 (ABd, 1, *J* = 18.5 Hz), 4.36 (ABd, 1, *J* = 18.5 Hz), 5.13 (d, 1, *J* = 4.5 Hz), 5.67 (d×d, 1, *J* = 4.5 and 8.5 Hz), 7.30 (s, 5).

n-Propylsulfones

33d: IR: 3410, 1795, 1767, 1690, 1510 cm⁻¹; ¹H NMR δ: 0.90 (t, 3, *J* = 7 Hz), 1.61 (m, 2), 2.50 (m, 2), 3.63 (s, 2), 4.20 (ABd, 1, *J* = 19 Hz), 4.70 (ABd, 1, *J* = 19 Hz), 4.78 (s, 2), 5.06 (d, 1, *J* = 5 Hz), 5.86 (d×d, 1, *J* = 5 and 9 Hz), 7.00 (br d, 1, *J* = 9 Hz), 7.33 (s, 5); Anal. C₁₈H₂₁O₆N₂SCl₃ (C, H, N).

33g: IR (KBr): 3320, 1789, 1735, 1670, 1532, 1390, 1320, 1288, 1255 cm⁻¹.

Benzylsulfones

34d: IR: 3410, 1800, 1770, 1695, 1605, 1510, 1380 cm⁻¹; ¹H NMR δ: 3.55 (ABd, 1, *J* = 14 Hz), 3.67 (s, 2), 3.91 (ABd, 1, *J* = 14 Hz), 4.08 (ABd, 1, *J* = 19 Hz), 4.65 (ABd, 1, *J* = 19 Hz), 4.71 (sh ABq, 2), 4.88 (d, 1, *J* = 5 Hz), 5.86 (d×d, 1, *J* = 5 and 9 Hz), 7.13 (d, 1, *J* = 9 Hz), 7.37 (sh m, 10); Anal. C₂₂H₂₁O₆N₂SCl₃ (C, H, N).

34f: IR (KBr): 3290, 1798, 1785, 1773, 1672, 1550, 1495 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 3.60 (br s, 2), 4.24 (br s, 2), 4.12 and 4.34 (two ABd, 1, *J* = 18 Hz), 4.34 and 4.53 (two ABd, 1, *J* = 18 Hz), 5.15 and 5.26 (two d, 1, *J* = 4.8 Hz), 5.63 and 5.66 (two d×d, 1, *J* = 4.8 and 8 Hz), 7.10–7.40 (m, 10), 7.46 and 7.48 (two s, 1), 7.70–7.96 (m, 4), 9.13 and 9.18 (two d, 1, *J* = 8 Hz); Anal. C₂₈H₂₄O₈N₂S (C, H, N).

34g: IR (KBr): 3300, 1790, 1740, 1670, 1530, 1495 cm⁻¹; ¹H NMR (DMSO-d₆) δ: 3.65 (s, 2), 3.85 (ABd, 1, *J* = 18 Hz), 4.31 (s, 2), 4.36 (ABd, 1, *J* = 18 Hz), 5.21 (d, 1, *J* = 5 Hz), 5.70 (d×d, 1, *J* = 5 and 8.5 Hz), 7.33 (br s, 10), 9.13 (d, 1, *J* = 8.5 Hz).

p-Nitrobenzylsulfones

35d: IR: 3410, 1800, 1768, 1690, 1609, 1529, 1510 cm⁻¹; ¹H NMR (200 MHz) δ: 3.62 (ABd, 1, *J* = 14 Hz), 3.70 (sh ABq, 2), 3.83 (ABd, 1, *J* = 14 Hz), 4.12 (ABd, 1, *J* = 18 Hz), 4.60 (ABd, 1, *J* = 18 Hz), 4.63 (ABd, 1, *J* = 12 Hz), 4.80 (ABd, 1, *J* = 12 Hz), 4.84 (d, 1, *J* = 4.5 Hz), 5.81 (d×d, 1, *J* = 4.5 and 8.7 Hz), 6.76 (d, 1, *J* = 8.7 Hz), 7.34 (s, 5), 7.32 (d, 2, *J* = 8 Hz), 8.22 (d, 2, *J* = 8 Hz); Anal. C₂₂H₂₀O₈N₃SCl₃ (C, H, N).

35f: IR (KBr): 3280, 1780 (br), 1660, 1610, 1522, 1350 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 3.50 (br s, 2), 3.86 and 3.90 (two s, 2), 4.19 (ABd, 1, *J* = 18 Hz), 4.38 and 4.54 (two ABd, 1, *J* = 18 Hz), 5.09 and 5.16 (two d, 1, *J* = 4.6 Hz), 5.27 (d×d, 1, *J* = 4.6 and 8 Hz), 7.26 (br s, 5), 7.46 and 7.49 (two s, 1), 7.47 and 7.50 (two d, 2, *J* = 8 Hz), 7.70–7.98 (m, 4), 8.08 and 8.10 (two d, 2, *J* = 8 Hz), 8.97 and 9.01 (two d, 1, *J* = 8 Hz); Anal. C₂₈H₂₃O₁₀N₃S (C, H, N).

35g: IR (KBr): 3300, 1775 (br), 1740, 1668 (br), 1613, 1520 (br), 1382 cm⁻¹; ¹H NMR (250 MHz, DMSO-d₆) δ: 3.64 (sh ABq, 2, *J* = 14 Hz), 3.82 (ABd, 1, *J* = 18.5 Hz), 4.16 (sh ABq, 2, *J* = 14 Hz), 4.28 (ABd, 1, *J* = 18.5 Hz), 5.15 (d, 1, *J* = 5 Hz), 5.64 (d×d, 1, *J* = 5 and 8.5 Hz), 6.84 (d, 2, *J* = 8 Hz), 7.03 (d, 2, *J* = 8 Hz), 7.30 (sh m, 5), 9.17 (d, 1, *J* = 8.5 Hz).

Allylsulfones

36d: IR: 3412, 1798, 1769, 1690, 1638, 1605, 1510 cm⁻¹; ¹H NMR δ: 3.26 (sh m, 2), 3.65 (s, 2), 4.17 (ABd, 1, *J* = 19 Hz), 4.71 (ABd,

1, *J* = 19 Hz), 4.80 (sh ABq, 2), 5.20 (d, 1, *J* = 4.5 Hz), 5.30–6.10 and 5.85 (m + d×d, 4, *J* = 4.5 and 9.5 Hz), 7.13 (d, 1, *J* = 9.5 Hz), 7.36 (s, 5); Anal. C₁₈H₁₉O₆N₂SCl₃ (C, H, N).

36f: IR (KBr): 3320, 1805, 1784, 1773, 1672, 1630, 1605, 1535 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 3.57 (sh ABq, 2, *J* = 13.8 Hz), 3.69 and 3.71 (two d, 2, *J* = 6 Hz), 4.21 and 4.30 (two ABd, 1, *J* = 18 Hz), 4.50 and 4.59 (two ABd, 1, *J* = 18 Hz), 5.22 and 5.33 (two d, 1, *J* = 4.8 Hz), 5.34 (br d×d, 2, *J* = 14.9 Hz), 5.55–5.78 (m, 2), 7.30 (sh m, 5), 7.50 and 7.52 (two s, 1), 7.70–8.0 (m, 4), 9.06 and 9.08 (two d, 1, *J* = 8 Hz); Anal. C₂₄H₂₂O₈N₂S (C, H, N).

36g: IR (KBr): 3300, 1790, 1740, 1670, 1640, 1525, 1500 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-d₆) δ: 3.35 (br d, 2, *J* = 6 Hz), 3.67 (s, 2), 3.93 (ABd, 1, *J* = 19 Hz), 4.48 (ABd, 1, *J* = 19 Hz), 5.20 (d, 1, *J* = 4.5 Hz), 5.26–6.0 and 5.78 (m + d×d, 4, *J* = 4.5 and 9 Hz), 7.40 (s, 5).

Propargylsulfones

37d: IR: 3410, 3300, 1800, 1765, 1692, 1510, 1413 cm⁻¹; ¹H NMR δ: 2.50 (sh m, 1), 3.46 (sh m, 2), 3.65 (s, 2), 4.18 (ABd, 1, *J* = 19 Hz), 4.70 (ABd, 1, *J* = 19 Hz), 4.81 (sh ABq, 2), 5.46 (d, 1, *J* = 5 Hz), 5.97 (d×d, 1, *J* = 5 and 9.5 Hz), 6.90 (br d, 1, *J* = 9.5 Hz), 7.36 (s, 5); Anal. C₁₈H₁₇O₆N₂SCl₃ (C, H, N).

37f: IR (KBr): 3320, 3280, 1890 (br), 1673, 1610, 1533, 1408 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 3.64 (br s, 2), 4.23 (sh m, 2), 4.30–4.67 (m, two ABq, 2, *J* = 18 Hz), 5.37 and 5.44 (two d, 1, *J* = 4.5 Hz), 5.69 (m, 1), 7.30 (br s, 5), 7.53 (br s, 1), 7.72–8.00 (m, 4), 9.09 (br d, 1, *J* = 8 Hz); Anal. C₂₄H₂₀O₈N₂S (C, H, N).

37g: IR (KBr): 3400, 3290, 1787 (br), 1740, 1675, 1525, 1390 cm⁻¹; ¹H NMR (CDCl₃ + 10% DMSO-d₆) δ: 2.76 (br s, 1), 3.60 (s, 2), 3.73 (br s, 2), 3.90 (ABd, 1, *J* = 18 Hz), 4.34 (ABd, 1, *J* = 18 Hz), 5.33 (d, 1, *J* = 4 Hz), 5.70 (d×d, 1, *J* = 4 and 9 Hz), 7.21 (s, 5), 8.53 (d, 1, *J* = 9 Hz).

Ethoxycarbonyl-methylsulfones

38d: IR: 3410, 1800, 1768, 1745, 1695, 1510, 1415 cm⁻¹; ¹H NMR δ: 1.3 (t, 3, *J* = 7 Hz), 3.53 (s, 2), 3.66 (s, 2), 4.17 (ABd, 1, *J* = 18 Hz), 4.26 (q, 2, *J* = 7 Hz), 4.67 (ABd, 1, *J* = 18 Hz), 4.81 (s, 2), 5.60 (d, 1, *J* = 4.5 Hz), 5.93 (d×d, 1, *J* = 4.5 and 9 Hz), 6.90 (br d, 1, *J* = 9 Hz), 7.34 (s, 5); Anal. C₁₉H₂₁O₈N₂SCl₃ (C, H, N).

38f: IR: 3408, 1798, 1775, 1741, 1690, 1605, 1510 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 1.21 (t, 3, *J* = 7 Hz), 3.57 (s, 2), 4.10–4.42 (q + sh ABq + two ABd, 5), 4.53 and 4.62 (two ABd, 1, *J* = 17.6 Hz), 5.42 and 5.48 (two d, 1, *J* = 4.7 Hz), 5.70 (m, 1), 7.30 (br s, 5), 7.53 (br s, 1), 7.72–8.0 (m, 4), 9.08 (br d, 1, *J* = 8 Hz); Anal. C₂₅H₂₄O₁₀N₂S (C, H, N).

38g: IR (KBr): 3320, 1793, 1743, 1674, 1530, 1395 cm⁻¹; ¹H NMR (CDCl₃ + 20% DMSO-d₆) δ: 1.30 (t, 3, *J* = 7 Hz), 3.63 (s, 2), 3.82 (br s, 2), 3.9–4.54 (q + two ABd, 4), 5.06–6.0 (m, 4), 7.30 (s, 5).

Trichloroethoxycarbonyl-methylsulfone

39d: IR: 3415, 1801, 1768, 1695, 1510, 1380 cm⁻¹; ¹H NMR (200 MHz) δ: 3.49 (ABd, 1, *J* = 15.2 Hz), 3.59 (ABd, 1, *J* = 14.6 Hz), 3.61 (ABd, 1, *J* = 15.2 Hz), 3.70 (ABd, 1, *J* = 14.6 Hz), 4.18 (ABd, 1, *J* = 18.8 Hz), 4.63 (ABd, 1, *J* = 18.8 Hz), 4.71 (ABd, 1, *J* = 12 Hz), 4.75 (ABd, 1, *J* = 12 Hz), 4.83 (ABd, 1, *J* = 12 Hz), 4.86 (ABd, 1, *J* = 12 Hz), 5.59 (d, 1, *J* = 4.6 Hz), 5.92 (d×d, 1, *J* = 4.6 and 9.6 Hz), 6.70 (d, 1, *J* = 9.6 Hz), 7.32 (sh m, 5); Anal. C₁₉H₁₈O₈SN₂Cl₆ (C, H, N).

Hydroxycarbonyl-methylsulfones

40d: ¹H NMR δ: 3.63 (s, 2), 3.83 (sh ABq, 2), 4.16 (ABd, 1, *J* = 18 Hz), 4.61 (ABd, 1, *J* = 18 Hz), 4.73 (sh ABq, 2), 5.63 (d, 1, *J* = 4.5 Hz), 5.92 (d×d, 1, *J* = 4.5 and 8 Hz).

40f: IR (KBr): 3500–3200, 1785 (br), 1740, 1672, 1530, 1415 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 3.56 (br s, 2), 4.13 and 4.16 (two s, 2), 4.25 and 4.31 (two ABd, 1, *J* = 18.5 Hz), 4.52 and 4.58 (two ABd, 1, *J* = 18.5 Hz), 5.42 and 5.48 (two d, 1, *J* = 4.8 Hz), 5.68 (m, 1), 7.30 (br s, 5), 7.52 (s, 1), 7.70–8.00 (m, 4), 9.03 (d, 1, *J* = 9 Hz); Anal. C₂₃H₂₀O₁₀SN₂ (C: 52.10, H, N).

40g (from **39d**): IR (KBr): 3360, 3100–2400 (br), 1810, 1732 (br), 1635 (br), 1533, 1418 cm⁻¹; ¹H NMR (200 MHz, DMSO-d₆) δ: 2.73 (s, 3), 2.89 (s, 3), 3.56 (br s, 2), 3.94 (ABd, 1, *J* = 18.4 Hz), 4.13 (s, 2), 4.34 (ABd, 1, *J* = 18.4 Hz), 5.40 (d, 1, *J* = 4.8 Hz), 5.64 (d×d, 1, *J* = 4.8 and 8.8 Hz), 7.28 (br s, 5), 7.95 (s, 1), 9.02 (d, 1, *J* = 8.8 Hz) (1:1 complex with DMF).

(β -Phenylethyl)aminocarbonyl-methylsulfones

41d: IR: 3420, 1800, 1768, 1688 (br), 1605, 1510 (br), 1380 cm^{-1} ; ^1H NMR (DMSO- d_6) δ : 2.90 (m, 2), 3.33 (m, 2), 3.60 (s, 2), 3.99 (sh ABq, 2), 4.30 (ABd, 1, $J = 18$ Hz), 4.69 (ABd, 1, $J = 18$ Hz), 4.96 (s, 2), 5.50 (d, 1, $J = 5$ Hz), 5.78 (d $\times$ d, 1, $J = 5$ and 9 Hz), 7.30 (br s, 10), 8.53 (m, 1), 8.90 (br d, 1, $J = 9$ Hz); Anal. $\text{C}_{25}\text{H}_{26}\text{O}_7\text{N}_3\text{S}$ (C, H, N).

41f: IR (KBr): 3400–3300, 1790, 1775, 1670, 1650, 1607, 1530 (br), 1325 cm^{-1} ; ^1H NMR (250 MHz, DMSO- d_6) δ : 2.74 (t, 2, $J = 7$ Hz), 3.35 (t, 2, $J = 7$ Hz), 3.57 (sh ABq, 2, $J = 14.5$ Hz), 3.85 and 3.88 (two ABd, 1, $J = 15$ Hz), 4.10 (ABd, 1, $J = 15$ Hz), 4.31 and 4.35 (two ABd, 1, $J = 18.5$ Hz), 4.59 and 4.63 (two ABd, 1, $J = 18.5$ Hz), 5.47 and 5.53 (two d, 1, $J = 5$ Hz), 5.74 (two d $\times$ d, 1, $J = 5$ and 9 Hz), 7.30 (sh m, 10), 7.56 and 7.58 (two s, 1), 7.76–8.00 (m, 4), 8.57 (m, 1), 8.99 and 9.01 (two d, 1, $J = 9$ Hz); Anal. $\text{C}_{31}\text{H}_{29}\text{O}_9\text{N}_3\text{S}$ (C, H, N).

41g: IR (KBr): 3500–3200, 1788, 1735, 1660 (br), 1535 (br), 1325 cm^{-1} ; ^1H NMR (DMSO- d_6) δ : 2.85 (m, 2), 3.30 (m, 2), 3.60 (s, 2), 3.96 (sh ABq, 2), 3.94 (ABd, 1, $J = 18$ Hz), 4.40 (ABd, 1, $J = 18$ Hz), 5.45 (d, 1, $J = 5$ Hz), 5.71 (d $\times$ d, 1, $J = 5$ and 9 Hz), 7.30 (br s, 10), 8.50 (m, 1), 8.90 (d, 1, $J = 9$ Hz).

Aminocarbonyl-methylsulfones

42d: IR: 3500, 3400, 1800, 1768, 1695 (br), 1600, 1510 cm^{-1} ; ^1H NMR (200 MHz) δ : 3.48 (sh ABq, 2, $J = 15$ Hz), 3.63 (sh ABq, 2, $J = 14$ Hz), 4.20 (ABd, 1, $J = 19$ Hz), 4.60 (ABd, 1, $J = 19$ Hz), 4.78 (sh ABq, 2, $J = 12$ Hz), 5.45 (d, 1, $J = 5$ Hz), 5.75 (br s, 1, CONH $_2$), 5.92 (d $\times$ d, 1, $J = 5$ and 9 Hz), 6.35 (br s, 1, CONH $_2$), 6.83 (d, 1, $J = 9$ Hz), 7.30 (sh m, 5); Anal. $\text{C}_{17}\text{H}_{18}\text{O}_7\text{N}_3\text{S}$ (C, H, N).

42f: IR (KBr): 3320, 3200, 1783 (br), 1675 (br), 1610, 1528, 1410 cm^{-1} ; ^1H NMR (200 MHz, DMSO- d_6) δ : 3.52 (br s, 2), 3.74 and 3.78 (two ABd, 1, $J = 14$ Hz), 4.03 (ABd, 1, $J = 14$ Hz), 4.24 and 4.30 (two ABd, 1, $J = 18$ Hz), 4.52 and 4.58 (two ABd, 1, $J = 18$ Hz), 5.44 and 5.48 (two d, 1, $J = 5$ Hz), 5.68 and 5.70 (two d $\times$ d, 1, $J = 5$ and 9 Hz), 7.26 (br s, 5), 7.50 (br s, 1), 7.60–7.96 (m, 4), 8.92 (br d, 1, $J = 9$ Hz); Anal. $\text{C}_{23}\text{H}_{21}\text{O}_9\text{N}_3\text{S}$ (C: 52.84, H, N).

Cyanomethylsulfones

43d: IR: 3410, 2260 (w), 1803, 1768, 1690, 1510, 1380 cm^{-1} ; ^1H NMR (CD_3CN) δ : 3.60 (s, 2), 4.12 (s, 2), 4.17 (ABd, 1, $J = 19$ Hz), 4.55 (ABd, 1, $J = 19$ Hz), 4.85 (s, 2), 5.35 (d, 1, $J = 5$ Hz), 5.83 (d $\times$ d, 1, $J = 5$ and 9 Hz), 7.23 (br d, 1), 7.30 (s, 5); Anal. $\text{C}_{17}\text{H}_{16}\text{O}_6\text{N}_3\text{SCl}_3$ (C, H, N).

43g: IR (KBr): 3300, 2260 (w), 1785, 1735, 1670, 1525, 1410 cm^{-1} ; ^1H NMR (250 MHz, DMSO- d_6) δ : 3.57 (s, 2), 3.96 (ABd, 1, $J = 18.5$ Hz), 4.32 (ABd, 1, $J = 18.5$ Hz), 4.98 (br m, 2), 5.51 (d, 1, $J = 5$ Hz), 5.73 (d $\times$ d, 1, $J = 5$ and 8.5 Hz), 7.30 (sh m, 5), 9.23 (d, 1, $J = 8.5$ Hz).

Reaction with hydroxylamine

Methanolic solutions of the β -lactam (0.02 M) and of hydroxylamine (0.25 M) were mixed at 20°C (equal volumes). The disappearance of the β -lactam was monitored by high performance liquid chromatography (instruments: Waters Associates, Inc., model 6000A and Absorbance Detector, model 440 (254 nm); Chromatointegrator Merck–Hitachi, model D-2000; column: 150 $\times$ 4.6 mm, packed with Rosil C18 (5 μm); eluant: 60–65% CH_3OH and 40–35% H_2O with a flow-rate of 1.8 ml/min).

Biological methods

The MIC of the monocyclic β -lactams (esters and free acids) were measured by microdilution tests in 96-microwell plates (Nunc, Roskilde, Denmark) as described by Thrupp [56], using a final inoculum of 10^5 CFU/ml and Mueller–Hinton broth (BBL, Microbiological Systems, Cockeysville, PA, U.S.A.). To allow the growth of *Streptococcus*, 5% horse blood (Gibco, Ghent, Belgium) was added to the culture medium. The range of concentrations obtained by 2-fold dilutions were from 200 to 0.006 μM . For water insoluble products, dimethyl sulfoxide was used as solvent, but its concentration never exceeded 2%. Ampicillin and penicillin G TAL or PIV esters (when ester compounds were tested) were included in each experiment as active reference antibiotics. The bacterial strains were obtained from

Institut Pasteur (Paris, France): *Escherichia coli* ATCC 25422, *Klebsiella pneumoniae* ATCC 10.031, *Pseudomonas aeruginosa* ATCC 27853, *Proteus mirabilis* NCTC 8309, *Serratia marcescens* ATCC 4003, *Salmonella typhimurium* LT2 60.62, *Staphylococcus aureus* ATCC 25923, and *Streptococcus pyogenes* ATCC 8668. *Proteus vulgaris*, a sensitive laboratory strain, was provided by Prof. J. M. Ghuysen (University of Liège, Belgium). Hydrolysis of the TAL and PIV esters (20 mM) was performed by incubation for 120 min at 37°C in the presence of human plasma. Then the solutions were diluted adequately and tested as above [56].

Three β -lactamases, one each of class A, B and C as defined by Ambler [57] and Jaurin and Grundström [58] were chosen for the study. The β -lactamase from *Bacillus licheniformis* 749 (active-serine enzyme, class A) was purified as described by Thatcher [59] and assayed at 30°C using nitrocefin (Glaxo) as the substrate. The enzyme was added to 500 μl of 100 μM nitrocefin in 50 mM sodium phosphate buffer, pH 7.0. The absorbance was monitored at 482 nm during 10 min by means of a Beckman DU-8 spectrophotometer. With 2 ng of enzyme a ΔA of 0.3 was obtained after 10 min of incubation. The β -lactamase from *Bacillus cereus* 5/B/6 (Zn^{++} enzyme, class B) was purified by chromatography on Amberlite CG-50 and CM-Sephadex. The activity was determined using 500 μl of 100 μM cephaloridin (Eli Lilly & Co.) in 100 mM Hepes buffer, pH 7.0, containing 0.5 M NaCl and 20 μM ZnCl_2 . At 30°C, 0.2 μg of enzyme produced a ΔA of -0.3 at 260 nm in 10 min. The β -lactamase from *Enterobacter cloacae* P99 (active-serine enzyme, class C) was purified as described by Ross [60]. The activity was determined using 500 μl of 100 μM cephaloridin in 50 mM sodium phosphate buffer, pH 7.0. At 30°C, 0.03 μg of enzyme produced a ΔA of -0.3 at 260 nm in 10 min. Before the assays, the stock enzyme solutions (0.5–1.0 $\text{mg}\cdot\text{ml}^{-1}$) were diluted in the buffer used for preparing the substrate solutions, supplemented with 0.1 mg/ml of bovine serum albumin. All potential inhibitors and inactivators were tested at final concentrations up to 100 μM .

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References

- 1 Dürckheimer W., Blumbach J., Lattrell R. & Scheunemann K. H. (1985) *Angew. Chem. Int. Ed. Engl.* 24, 180–202
- 2 Frère J. M. & Joris B. (1985) *CRC Crit. Rev. Microbiol.* 11, 299–396
- 3 Albers-Schonberg G., Arison B. H., Hensens O. D., Hirshfield J., Hoogsteen K., Kaczka E. A., Rhodes R. E., Kahan J. S., Kahan F. M., Ratcliffe R. W., Walton E., Ruswinkle L. J., Morin R. B. & Christensen B. G. (1978) *J. Am. Chem. Soc.* 100, 6491
- 4 Cooke M. D., Moore K. W., Ross B. C. & Turner S. E. (1983) *Tetrahedron Lett.* 24, 3373
- 5 Yoshida A., Hayashi T., Takeda N., Oida S. & Ohki E. (1983) *Chem. Pharm. Bull.* 31, 768
- 6 Girijavallabhan V. M., Ganguly A. K., Pinto P. & Versace R. (1983) *J. Chem. Soc. Chem. Commun.* 908
- 7 Leanza W. J., DiNinno F., Muthard D. A., Wilkening R. R., Wildonger K. J., Ratcliffe R. W. & Christensen B. G. (1983) *Tetrahedron* 39, 2505
- 8 Wasserman H. H. & Han W. T. (1985) *J. Am. Chem. Soc.* 107, 1444
- 9 Emmer G., Kneussel P., Hildebrandt J., Turnowsky F., Haselberger A., Wenzel A. & Stuetz P. (1985) *J. Antibiot.* 38, 1371
- 10 Girijavallabhan V. M., Ganguly A. K., Liu Y. T., Pinto P. A., Patel N., Hare R. H. & Miller G. H. (1986) *J. Antibiot.* 39, 1187
- 11 Lang M., Schneider P., Scartazzini R., Tosch W., Konopk E. A. & Zak O. (1987) *J. Antibiot.* 40, 217
- 12 Sammes P. G. (1980) in: *Topics in Antibiotic Chemistry*, Vols. 3 & 4 John Wiley & Sons, New York, pp. 000

- 13 Morin R. B. & Gorman M. (1982) in: *Chemistry and Biology of β -Lactam Antibiotics*, Vol. 2 Academic Press, New York, pp. 000
- 14 Aoki H., Sakai H., Koshaka M., Kamani J., Hosada J., Kubachi Y., Iguchi E. & Imanaka H. (1976) *J. Antibiot.* 29, 492
- 15 Iamada A., Kitano K., Kintaka K., Muroi M. & Asai M. (1981) *Nature* 289, 590
- 16 Woulfe S. R. & Miller M. J. (1985) *J. Med. Chem.* 28, 1447
- 17 Cimarusti C. M. & Sykes R. B. (1984) *Med. Res. Rev.* 4, 1—24
- 18 Slusarchyk W. A., Dejneka T., Gordon E. M., Weaver E. R. & Koster W. H. (1984) *Heterocycles* 21, 191
- 19 Woulfe S. R. & Miller M. J. (1984) *Tetrahedron Lett.* 25, 3293
- 20 Noguchi N., Masuya H., Saguwara T., Kawano Y., Matsuo T. & Ochiai M. (1985) *J. Antibiot.* 38, 1387
- 21 Breuer H., Straub H., Treuner U. D., Drossard J. M., Höhn H. & Lindner K. R. (1985) *J. Antibiot.* 38, 813
- 22 Woulfe S. R., Iwagami H. & Miller M. J. (1985) *Tetrahedron Lett.* 26, 3891
- 23 Woulfe S. R. & Miller M. J. (1985) *J. Med. Chem.* 28, 1447
- 24 Woulfe S. R. & Miller M. J. (1986) *J. Org. Chem.* 51, 3133
- 25 Yoshida C., Tanaka K., Nakano J., Todo Y., Yamafuji T., Hattori R., Fukuoka Y. & Saikawa I. (1986) *J. Antibiot.* 39, 76
- 26 Boyd D. B., Eigenbrot C., Indelicato J. M., Miller M. J., Pasini C. E. & Woulfe S. R. (1987) *J. Med. Chem.* 30, 528
- 27 Cartwright S. J. & Waley S. G. (1983) *Med. Res. Rev.* 3, 341—382
- 28 Knowles J. R. (1985) *Acc. Chem. Res.* 18, 97—104
- 29 Marchand-Brynaert J., Ghosez L. & Cossement E. (1980) *Tetrahedron Lett.*, 21 3085
- 30 Ghosez L., Marchand-Brynaert J., Vekemans J., Bogdan S. & Cossement E. (1983) *Tetrahedron* 39, 2493
- 31 Cossement M., Marchand-Brynaert J., Bogdan S. & Ghosez L. (1983) *Tetrahedron Lett.* 24, 2563
- 32 Declercq J. P., Piccini-Leopardi C. & Marchand-Brynaert J. (1987) *New J. Chem.* 11, 499
- 33 Ernest I., Gosteli J., Greengrass C. W., Holick W., Jackman D.-E., Pfaendler H. R. & Woodward R. B. (1978) *J. Am. Chem. Soc.* 100, 8214
- 34 Lang M., Prasad K., Holick W., Gosteli J., Ernest I. & Woodward R. B. (1979) *J. Am. Chem. Soc.* 101, 6296 and following papers
- 35 English A. R., Retsema J. A., Girard A. E., Lynch J. E. & Barth W. E. (1978) *Antimicrob. Agents Chemother.* 14, 414
- 36 Morin R. B. & Gorman M. (1982) in: *Chemistry and Biology of β -Lactam Antibiotics* Vol. 3, Academic Press, New York, pp. 195—199
- 37 Gottstein W. J., Kim C. U., Shih K. M. & McGregor D. N. (1978) *J. Med. Chem.* 21, 240
- 38 De Koning J. J., Marx A. F., Poot M. M., Smid P. M. & Verweij J. (1977) in: *Recent Advances in the Chemistry of β -Lactam Antibiotics* (Elks J., ed.), Chem. Soc. Special Publication, 28, Chapt. 17
- 39 Dürckheimer W., Klesel N., Limbert M., Schrinner E., Seeger K. & Seliger H. (1981) in: *Recent Advances in the Chemistry of β -Lactam Antibiotics* (Gregory G. I., ed.), Chem. Soc. Special Publication, 38, Chapt. 4
- 40 Cooper R. D. G. & José F. L. (1970) *J. Am. Chem. Soc.* 92, 2575
- 41 Cooper R. D. G. & José F. L. (1972) *J. Am. Chem. Soc.* 94, 1021
- 42 Brain E. G., Eglington A. J., Nayler J. H. C., Pearson M. J. & Southgate R. (1972) *J. Chem. Soc. Chem. Commun.* 229
- 43 Brain E. G., Eglington A. J., Nayler J. H. C., Pearson M. J. & Southgate R. (1976) *J. Chem. Soc. Perkin I* 447
- 44 Barton D. H. R., Underwood G. W., Stoke P., Looker E. B. & Hewitt G. (Glaxo) (1972) D.O.S. 2138319; (1972) *Chem. Abstr.* 77, 48447
- 45 Aratani M. & Hashimoto M. (1981) in: *Recent Advances in the Chemistry of β -Lactam Antibiotics* (Gregory G. I., ed.), Chem. Soc. Special Publication 38, pp. 97—108
- 46 McOmie J. F. W. (1973) in: *Protective Groups in Organic Chemistry* Plenum Press, London, pp. 000
- 47 Jeffrey P. D. & McCombie S. W. (1982) *J. Org. Chem.* 47, 587
- 48 Chauvette J., Pennington P. A., Ryan C. W., Cooper R. D. G., José F. L., Wright I. G., Van Heyningen E. M. & Huffman G. W. (1971) *J. Org. Chem.* 36, 1259
- 49 Marchand-Brynaert J., Vekemans J., Bogdan S., Cossement M., Ghosez L. & Cossement E. (1981) in: *Recent Advances in the Chemistry of β -Lactam Antibiotics* (Gregory G. I., ed.), Chem. Soc. Special Publication 38, pp. 269—278
- 50 Glaxo (1972) D.O.S. 2138319; (1972) *Chem. Abstr.* 77, 48447
- 51 Lattrell R. & Lohaus G. (1974) *Liebigs Ann. Chem.* 921
- 52 Lattrell R. (1974) *Liebigs Ann. Chem.* 1361
- 53 Osborne N. F. (1980) *J. Chem. Soc. Perkin I* 146
- 54 Ferres H. (1980) *Chem. Ind.* 11, 435
- 55 Flynn E. H. (1972) in: *Cephalosporins and Penicillins, Chemistry and Biology* Academic Press, New York, pp. 312—366
- 56 Thrupp L. D. (1980) in: *Antibiotics in Laboratory Medicine* (Lorian V., ed.), Williams and Wilkins, Baltimore, pp. 73—113
- 57 Ambler R. P. (1980) *Phil. Trans. R. Soc. London B* 289, 321—331
- 58 Jaurin B. & Grundström T. (1981) *Proc. Natl. Acad. Sci. USA* 78, 4897—4901
- 59 Thatcher D. R. (1975) *Methods Enzymol.* 43, 653—664
- 60 Ross G. W. (1975) *Methods Enzymol.* 43, 678—687