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A new total synthesis of an indolo[3,2-*j*]phenanthridine alkaloid calothrixin B

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Abstract—A new total synthesis of calothrixin B is described. The key step is an allene-mediated electrocyclic reaction involving the indole 2,3-bond for the construction of a suitable 4-oxygenated 2,3,4-trisubstituted carbazole ring system. © 2005 Elsevier Ltd. All rights reserved.

The novel pentacyclic metabolites, calothrixins A (1) and B (2) having an indolo[3,2-j]phenanthridine nucleus were isolated from *Calothrix* cyanobacteria in 1999 (Fig. 1).¹ They display potent inhibitory effects on the in vitro growth of both the human malarial parasite and human cancer cells¹ and inhibition of a bacterial RNA polymerase.² Synthetic efforts of these interesting targets have been reported by three groups of Kelly et al.,³ Chai and co-workers⁴ and Guingant and co-workers.⁵

We herein, describe a new total synthesis of calothrixin B (2) through a construction of a 4-oxygenated 2,3,4-trisubstituted carbazole ring, derived from a disconnection at the C-5 and C-6 bond, based on an allene-mediated electrocyclic reaction of 6π -electron system involving the indole 2,3-bond.⁶ For the synthesis of 4-oxygenated 2,3,4-trisubstituted carbazole 8, we chose 2-formyl-*N*-

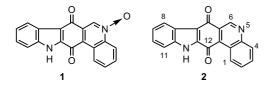


Figure 1.

phenylsulfonylindole $(3)^7$ as a starting material (Scheme 1).

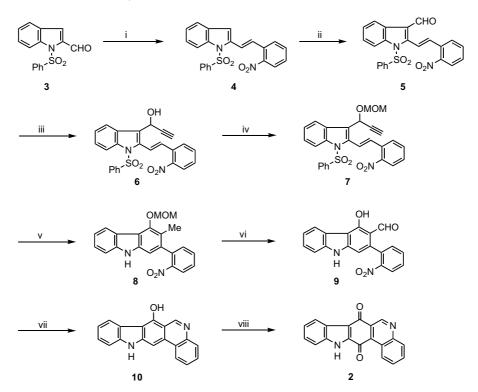
The Wittig reaction of 3 with 2-nitrobenzyltriphenylphosphorane gave the *trans*-2-(2-styryl)indole 4 (96%).⁸ Treatment of 4 with α, α -dichloromethyl methyl ether in the presence of $AlCl_3^9$ at -78 °C afforded the 3formylindole 5 (96%). The Grignard reaction of 5 with ethynyl magnesium bromide yielded the propargyl alcohol 6 (92%), which was protected with chloromethyl methyl ether in the presence of *i*-Pr₂NEt to produce the methoxymethyl ether 7 (93%). The propargyl ether 7 was subjected to an allene-mediated electrocyclic reaction⁶ in the presence of *t*-BuOK in *t*-BuOH and THF at 90 °C to yield the desired 4-oxygenated 2,3,4-trisubstituted carbazole 8 (29%). Subsequent oxidation of 8 with DDQ in the presence of lithium perchlorate in CH₂Cl₂ and H₂O gave the 3-formyl-4-hydroxycarbazole 9 with elimination of methoxymethyl group (70%). Reduction of the nitro group of 9 with 10% Pd–C and H₂, followed by the intramolecular condensation afforded the pentacyclic indolo[3,2-j]phenanthridine 10, which was oxidized to quinone with CAN (cerric ammonium nitrate) to provide calothrixin B (2) (67% yield from 9).¹⁰ The physical and spectroscopic data of synthetic sample 2 were identical with those of natural calothrixin B (2).¹

Thus, a synthetic route to calothrixin B (2) was newly completed with a 10.7% overall yield by the construction of an appropriate 4-oxygenated 2,3,4-trisubstituted carbazole ring 8 based on our methodology,⁶ followed by the connection between C-5 and C-6 to the pentacyclic

Keywords: Indolo[3,2-*j*]phenanthridine; Calothrixin B; Synthesis; Quinone.

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Scheme 1. Reagents and conditions: (i) 2-nitrobenzyl triphenylphosphonium bromide, *n*-BuLi, THF, 0 °C, 3 h (96%); (ii) Cl₂CHOMe, AlCl₃, CH₂Cl₂, -78 °C, 1 h (96%); (iii) ethynylmagnesium bromide, THF, 0.5 h (92%); (iv) MOMCl, *i*-Pr₂NEt, CH₂Cl₂, 50 °C, 12 h (93%); (v) *t*-BuOK, *t*-BuOH, THF, 0.5 h (29%); (vi) (1) DDQ, LiClO₄, CH₂Cl₂, H₂O, rt, 40 h, (2) 6 N HCl, ethylene glycol, THF, 60 °C, 3 h (70%); (vii) 10% Pd–C, H₂, EtOH, rt, 5 h; (viii) CAN, MeCN, H₂O, 0 °C, 1 h (67% from 9).

indolo[3,2-*j*]phenanthridine ring **10**. Although our route to **2** had the lower overall yield in the comparison with those of the earlier three groups,^{3–5} it would be characterized that our synthetic strategy facilitates a construction of a chemical library, in particular E-ring of calothrixins A (**1**) and B (**2**) for biological evaluations.

Acknowledgements

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- 10. Calothrixin B: mp >300 °C. ¹H-NMR (DMSO- d_6 , 300 MHz): δ 7.39 (1H, t, J = 7.7 Hz), 7.47 (1H, t, J = 7.7 Hz), 7.63 (1H, d, J = 7.7 Hz), 7.89 (1H, t, J = 7.7 Hz), 7.96 (1H, t, J = 7.7 Hz), 8.17 (1H, d, J = 7.7 Hz), 8.19 (1H, d, J = 7.7 Hz), 9.58 (1H, d, J = 7.7 Hz), 9.63 (1H, s), 13.18 (1H, br s).