

MUKONICINE, A CARBAZOLE ALKALOID FROM LEAVES OF *MURRAYA KOENIGII*

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(Received 12 December 1982)

Key Word Index—*Murraya koenigii*; Rutaceae; mukonicine; carbazole alkaloid.

Abstract—A new carbazole alkaloid, mukonicine, was isolated from the leaves of *Murraya koenigii*. From physical methods coupled with chemical evidence its structure was determined as 1,2-[2:2-dimethyl- Δ^3 -pyrano]-3-methyl-6,8-dimethoxycarbazole.

In continuation of our investigations on the chemistry of carbazole alkaloids [1–4], we now wish to report the isolation and structural elucidation of a new carbazole alkaloid, mukonicine, from the leaves of *Murraya koenigii* Spreng.

Mukonicine (1), $C_{20}H_{21}NO_3$, mp 233–234° was isolated from an alcoholic extract of the leaves of *M. koenigii* by column chromatography over alumina. The homogeneity of mukonicine was confirmed by TLC using various solvent systems. Its UV spectrum in EtOH with λ_{max} at 226 (log ϵ 4.70), 240 (4.67) 300 (4.59) and 342 nm (4.26) was strikingly similar to that of koenimbine [5]. The IR spectrum of mukonicine showed absorption peaks at ν_{max}^{KBr} 3440 (–NH–), 1648 (–OMe), 1630, 1560, 1460 (aromatic residue), 1385 (C–Me) and 765, 745 cm^{-1} (substituted benzene derivative). The 1H NMR spectrum of mukonicine (90 MHz in $CDCl_3$) showed signals for one indolic proton (s, δ 7.81, confirmed by D_2O exchange), three aromatic protons at δ 7.56 (s), 7.40 (s) and 6.88 (s), two olefinic protons 6.55 and 5.65 (pair of doublets), six protons of two aromatic OMe (s, δ 3.9) and three protons of an aromatic C–Me group (s, δ 2.3). The appearance of one of the aromatic protons as a singlet (δ 7.56) clearly indicated that the position of the aromatic Me was at C-3 [5]. The signal for a C-5 proton at a higher field (δ 7.40, shielded) showed that one of the two OMe groups was at C-6, similar to that of koenimbine [5]. The position of the

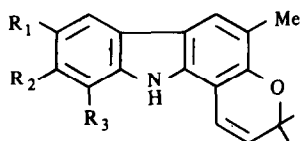
second OMe group was fixed at C-8 on the basis that the C-7 proton appeared at a much higher field (δ 6.88) than that of the C-5 proton. The signal for the six-proton singlet at δ 1.44 together with symmetrical doublets at δ 6.55 and 5.65 ($J = 9$ Hz each) suggested the presence of a 2:2-dimethyl- Δ^3 -pyran ring fused with a carbazole skeleton.

Mukonicine showed a $[M]^+$ at m/z 323. The most important peak appeared at m/z 308 $[M - 15]^+$, which is the base peak. The peak at m/z 308 was very similar to that of mahanimbine and girinimbine and can be represented by species 2. This also supports the presence of a 2:2-dimethyl- Δ^3 -pyran system in mukonicine. The 3-methyl carbazole skeleton of mukonicine was confirmed by the isolation of a 3-methyl carbazole after zinc dust distillation of mukonicine. Further proof for the presence of a 2:2-dimethyl- Δ^3 -pyran ring was provided by chromic acid oxidation, when acetone was obtained.

On the basis of the above evidence the structure of mukonicine can be represented by 1. Moreover, mukonicine was found to be quite different from koenigicine (3) [6, 7].

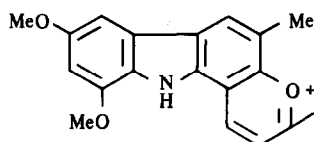
EXPERIMENTAL

All mps are uncorr. UV and IR spectra were recorded in EtOH and as KBr pellets, respectively. 1H NMR were measured at



1 $R_1 = R_3 = -OMe$, $R_2 = H$

3 $R_1 = R_2 = -OMe$, $R_3 = H$



2

90 MHz in CDCl_3 with TMS as int. standard.

Isolation of mukonicine. Defatted leaves were extracted with EtOH. The conc. extract was digested with 10% HCl for 3 hr, the soln filtered, the residue washed with H_2O until acid-free and then dried. The dark green residue was extracted with C_6H_6 and the extract concd and chromatographed over Al_2O_3 . The column was successively eluted with petrol, C_6H_6 and CHCl_3 . Mukonicine was obtained from the C_6H_6 eluate and crystallized from $\text{C}_6\text{H}_6\text{-CHCl}_3$, mp 233–234°. TLC on Si gel G and $\text{C}_6\text{H}_6\text{-CHCl}_3$ (1:1), R_f 0.46 (Found: C, 74.18; H, 6.62; N, 4.62. Calc. for $\text{C}_{20}\text{H}_{21}\text{NO}_3$: C, 74.23; H, 6.55; N, 4.33%.)

Zn dust distillation. The product was obtained from mukonicine on distillation with Zn dust and subsequent crystallization from petrol- C_6H_6 (1:1) mp 206–207°. It was identical with 3-methyl carbazole in all respects (mp, mmp., UV and TLC). (Found: C, 86.06; H, 6.21; N, 7.68. Calc. for $\text{C}_{13}\text{H}_{11}\text{N}$: C, 86.15; H, 6.12; N, 7.73%.)

Chromic acid oxidation. Mukonicine (25 mg) dissolved in HOAc (10 ml) was treated with CrO_3 in HOAc (0.1 g in 5 ml) with constant stirring. The reaction mixture was kept at room temp. for 72 hr. It was then neutralized with 50% aq. NaOH soln and steam-distilled. The distillate was collected in a flask containing a freshly prepared aq. soln of 2,4-dinitrophenyl hydrazine sulphate, when a flocculent ppt. appeared. It was then extracted with CCl_4 , the extract washed with H_2O and dried (Na_2SO_4). On removing the solvent, yellow crystals were obtained which were dissolved in C_6H_6 and chromatographed over

Al_2O_3 . The column was successively eluted with petrol, petrol- C_6H_6 (1:1) and C_6H_6 . In the petrol- C_6H_6 eluate, a yellow solid was obtained which was crystallized from MeOH, mp 125–126°. This compound was found to be identical with an authentic sample of the 2,4-dinitrophenyl hydrazone of Me_2CO .

Acknowledgements—The NMR and MS spectra were obtained through the courtesy of Mr. E. Vonbank of Organisch-Chemisches Institut, Zurich and Professor Beckman of Wayne State University, Detroit. The financial assistance to one of us (S.M.) from the C.S.I.R., India is gratefully acknowledged.

REFERENCES

1. Chakraborty, D. P., Chatterjee, D. and Ganguly, S. N. (1969) *Chem. Ind. (London)* 1662.
2. Bordner, J., Chakraborty, D. P., Chowdhury, B. K., Ganguly, S. N., Das, K. C. and Weinstein, B. (1972) *Experientia* **28**, 1406.
3. Chakraborty, D. P., Ganguly, S. N., Maji, P. N., Mitra, A. R., Das, K. C. and Weinstein, B. (1973) *Chem. Ind. (London)* 322.
4. Ganguly, S. N. and Sarkar, A. (1978) *Phytochemistry* **17**, 1816.
5. Narashimhan, N. S., Paradhkar, M. V. and Chitguppi, V. P. (1968) *Tetrahedron Letters* 5501.
6. Narashimhan, N. S., Paradhkar, M. V. and Kelkar, S. L. (1970) *Indian J. Chem.* **8**, 473.
7. Kureel, S. P., Kapil, R. S. and Popli, S. P. (1969) *Experientia* **25**, 790.