



CLAUSENOL AND CLAUSENINE—TWO CARBAZOLE ALKALOIDS FROM *CLAUSENA ANISATA*

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Key Word Index—*Clausena anisata*; Rutaceae; stem bark; clausenol; clausenine; carbazole alkaloids; antimicrobial activity.

Abstract—Two new carbazole alkaloids, designated as clausenol and clausenine, were isolated from an alcoholic extract of the stem bark of *Clausena anisata*. Their structures were established as 1-hydroxy-6-methoxy-3-methylcarbazole and 1,6-dimethoxy-3-methyl carbazole, respectively, from physical and chemical evidence and synthesis. Clausenol was found to be active against Gram-positive and Gram-negative bacteria and fungi.

INTRODUCTION

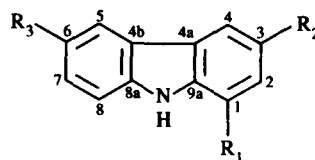
Previously, some carbazole alkaloids were reported from *Clausena anisata* [1, 2]. The present investigation reveals the presence of two new carbazole alkaloids, clausenol and clausenine, from the alcoholic extract of the stem bark of *C. anisata*.

RESULTS AND DISCUSSIONS

Clausenol 1, $C_{14}H_{13}NO_2$ ($[M]^+$ m/z 227) was found to be homogeneous by TLC and mass spectrometry. Zinc dust distillation of 1 led to the isolation of carbazole (3) and 3-methylcarbazole (4) indicating the presence of a 3-methylcarbazole residue in the molecule. A blue colouration with $FeCl_3$ indicated the presence of a phenolic hydroxyl group. Its UV spectrum λ_{max}^{EtOH} 228 (log ϵ 4.30), 253 (4.36), 303 (4.02), 356 nm (3.50) and IR spectrum ν_{max}^{KBr} 3395 (NH or OH) 1620, 1580 (aromatic system) 1204 ($ArOCH_3$) 975, 850, 820 cm^{-1} (substituted aromatic ring) suggested the presence of a carbazole nucleus. The 1H NMR spectrum (100 MHz, $DMSO-d_6$) showed signals at δ 10.56 (1H, s, OH proton, exchangeable with D_2O), 9.60 (1H, bs, NH proton, exchangeable with D_2O), 7.50 (1H, d, $J = 2$ Hz, H-4), 7.40 (1H, d, $J = 2$ Hz, H-5), 7.31 (1H, d, $J = 8$ Hz, H-8), 6.88 (1H, dd, $J = 8$ Hz, 2 Hz, H-7), 6.56 (1H, d, $J = 1.5$ Hz, H-2), 3.91 (3H, s, OCH_3) and 2.16 (3H, s, $Ar-CH_3$). Since the H-4 and H-2 protons were not *ortho*-coupled, positions 3 and 4 were substituted. The H-5 proton also was not *ortho*-substituted and slightly shielded, suggesting the position of the methoxyl group at C-6. The 1H NMR signal for the hydroxyl proton of clausenol was very similar to that of a 1-

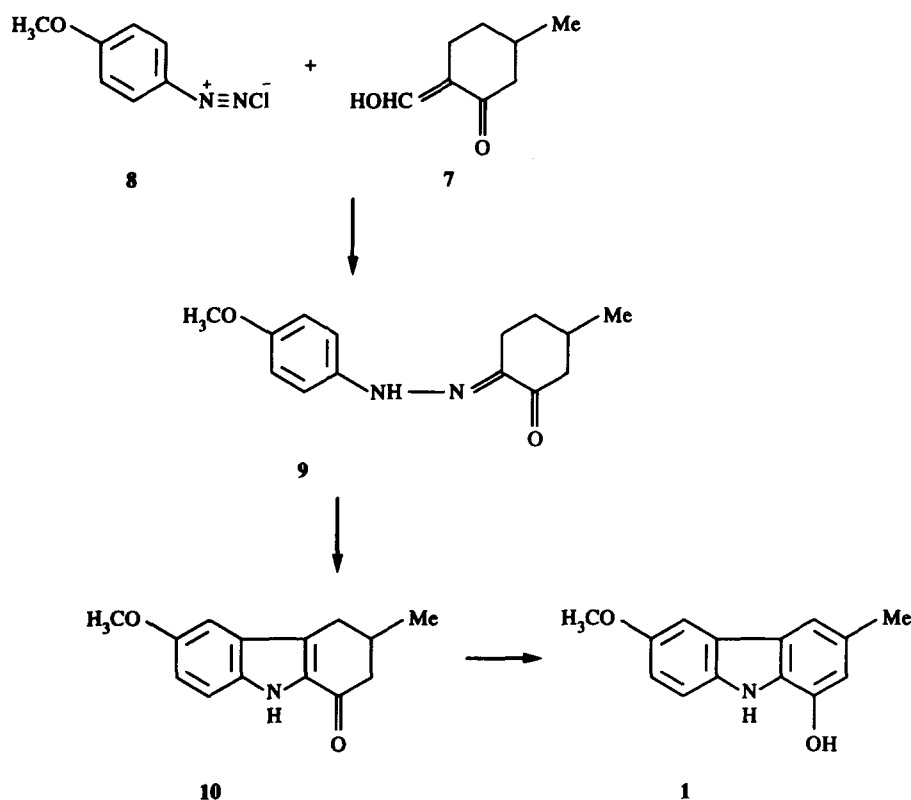
hydroxyl group [3]. In the ^{13}C NMR spectrum of 1 (25 MHz, $DMSO-d_6$), the upfield shifts of the C-2 and C-4 in comparison to carbazole is also suggestive of the presence of a hydroxyl group at position 1.

On acetylation, 1 furnished an acetate 5, mp 135°. The IR spectrum of 2 showed a strong peak at 1752 cm^{-1} for the acetoxy function and absence of the hydroxyl group. Reduction of the tosyl derivative of 1 with Raney nickel gave a compound 6, mp 181°, which was identified as glycozoline [4] by direct comparison with an authentic sample. Since the structure of glycozoline has already been established as 6-methoxy-3-methylcarbazole, 1, must have a methoxyl group at C-6 and a methyl group at the C-3 position. From all of this evidence, the structure of 1 was established as 1-hydroxy-6-methoxy-3-methyl carbazole, which was further supported by its ^{13}C NMR data. Finally, the structure of 1 was confirmed by synthesis (scheme 1).



- 1 $R_1 = OH$; $R_2 = Me$; $R_3 = OMe$
- 2 $R_1 = R_3$; OMe ; $R_2 = Me$
- 3 $R_1 = R_2$; $R_3 = H$
- 4 $R_1 = R_3$; H ; $R_2 = Me$
- 5 $R_1 = OCOMe$; $R_2 = Me$; $R_3 = OMe$
- 6 $R_1 = H$; $R_2 = Me$; $R_3 = OMe$

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Scheme 1. Synthesis of clausenol (1).

2-Hydroxymethylene-5-methylcyclohexanone (7) [5] on condensation with 4-methoxybenzene diazonium chloride (8) under Japp-Klingemann conditions furnished 4-methylcyclohexane-1,2-dione-1-(4-methoxy) phenyl hydrazone (9, 49.5%), which on indolization with concentrated HCl and acetic acid furnished 6-methoxy-3-methyl-1-oxo-1,2,3,4-tetrahydrocarbazole (10, 64.5%). Compound 10 was refluxed with 10% Pd/C in decalin and from the reaction product, the alkali-soluble fraction was separated and chromatographed over silica gel. Elution with CH_2Cl_2 furnished material (19.9%) identical with natural clausenol in all respects (mp, mmp, UV, IR, ^1H NMR).

Clausenine (2), $\text{C}_{15}\text{H}_{15}\text{NO}_2$ ($[\text{M}]^+ m/z$ 241) had a UV spectrum $\lambda_{\text{max}}^{\text{EtOH}}$ 226 (log ϵ 4.34), 242 (4.32), 299 (4.00), 340 (3.44), 354 nm (3.42), and IR spectrum $\nu_{\text{max}}^{\text{KBr}}$ 3400 ($-\text{NH}$), 1580 (aromatic), 1218 (aromatic ether), 940, 840, 820 cm^{-1} (aromatic substitution) characteristic of a carbazole nucleus. The ^1H NMR (100 MHz, $\text{DMSO}-d_6$) showed signals at δ 10.40 (1H, s, NH proton exchangeable with D_2O), 7.54 (1H, d, $J = 2\text{ Hz}$, H-4), 7.44 (1H, d, $J = 2\text{ Hz}$, H-5), 7.32 (1H, d, $J = 8\text{ Hz}$, H-8), 6.92 (1H, dd, $J = 8\text{ Hz}$, 2.5 Hz, H-7), 6.74 (1H, d, $J = 1.5\text{ Hz}$, H-2), 3.92 (3H, s, Ar- OCH_3), 3.80 (3H, s, Ar- OCH_3) and 2.42 (3H, s, Ar- CH_3). This spectrum was very similar to that of clausenol (1) except that instead of a hydroxyl proton, an additional methoxy signal (δ 3.80) was apparent. It was therefore suggested that clausenine may be the methyl

ether of clausenol. Methylation of 1 with diazomethane furnished a compound identical in all respects (mp, mmp, UV, IR, ^1H NMR) with clausenine. From this, the structure of 2, was established as 1,6-dimethoxy-3-methyl carbazole, which was further supported by its ^{13}C NMR spectrum (25 MHz, $\text{DMSO}-d_6$).

Studies on the antimicrobial properties of clausenol revealed that it was highly active against both Gram-positive and Gram-negative bacteria and fungi, while clausenine showed low activity against Gram-negative bacteria and fungi. The minimum inhibitory concentration (MIC) of clausenol and clausenine was determined by the agar dilution method. The MIC were studied up to $100\text{ }\mu\text{g ml}^{-1}$ and incubations were done at 37° . The MIC value of standard compounds, streptomycin ($100\text{ }\mu\text{g ml}^{-1}$), penicillin ($2\text{ }\mu\text{g ml}^{-1}$), nystatin and griseofulvin ($1\text{ }\mu\text{g ml}^{-1}$) were also studied. The MIC values are shown in Table 1.

EXPERIMENTAL

All mps are uncorr. UV and IR spectra were recorded in EtOH and as KBr pellets, respectively. ^1H and ^{13}C NMR were recorded at 100 and 25 MHz, respectively.

Isolation of alkaloids. Air-dried powdered stem bark (1 kg) of *C. anisata* was extracted with petrol in a Soxhlet for 36 hr. After extraction, the stem bark was dried and

Table 1. Minimum inhibitory concentrations of clausenol and clausenine

Sl. No.	Name of the organism	Compounds (MIC μ g ml)					
		Clausenol	Clausenine	P	S	G	N
1	<i>Escherichia coli</i> ST 203	7	40	—	5	—	—
2	<i>Becillus subtilis</i> ST 204	14	> 100	0.01	—	—	—
3	<i>Salmonella typhi</i> ST 288	12	> 100	—	10	—	—
4	<i>Pseudomonas aeruginosa</i> ST 243	14	16	—	6	—	—
5	<i>Staphylococcus aureus</i> MC 27927	1.3	> 100	0.03	—	—	—
6	<i>Candida albicans</i> ST 388	5	16	—	—	—	0.05
7	<i>Trychophyton rubrum</i> ST 389	2	5	—	—	0.02	—

Key: — = Not tested; P = penicillin; S = streptomycin, G = griseofulvin, N = nystatin.

re-extracted with EtOH for 30 hr. The extract was freed from the solvent and the residue taken up in CHCl_3 and sepd into acidic, basic and neutral frs. The acidic fr. was concd and chromatographed over silica gel (500 g). The column was eluted with petrol, petrol- CH_2Cl_2 (1:1), CHCl_3 and finally with a 2% CHCl_3 -MeOH mixt. From the CHCl_3 eluate, clausenol (1, 120 mg, 0.012%) was obtained, which was recrystallized from benzene, mp 139° . (Found: C, 74.23; H, 5.65; N, 6.34%. Calcd for $\text{C}_{14}\text{H}_{13}\text{NO}_2$: C, 73.99; H, 5.77; N, 6.16%). ^{13}C NMR: δ 145.5 (C-1, s), 108.0 (C-2, s), 124.1 (C-3, s), 121.5 (C-4, d), 115.5 (C-4a, s), 116.0 (C-4b, s), 102.9 (C-5, d), 154.3 (C-6, s), 115.4 (C-7, d), 112.3 (C-8, d), 135.0 (C-8a, s), 134.6 (C-9a, s), 21.0 (Me, q). On concn of the neutral fr. it was subjected to CC over silica gel (400 g). The column was eluted with the same solvents as described above. From the petrol- CH_2Cl_2 (1:1) eluate a light yellow semi-solid mass was obtained. This was subjected to prep. TLC on silica gel G (1 mm thick) eluting with benzene- CHCl_3 (9:1). A major band (Rf 0.32) was sepd and extracted with CHCl_3 . The residue obtained after removal of solvent was recrystallized from benzene-petrol, when crystals of **2** (170 mg, 0.042%), mp 151° were obtained. (Found: C, 74.99; H, 6.17; N, 5.95%. Calcd. for $\text{C}_{15}\text{H}_{15}\text{NO}_2$: C, 74.67; H, 6.27; N, 5.80%). ^{13}C NMR: δ 145.5 (C-1, s), 107.5 (C-2, d), 126.5 (C-3, s), 127.5 (C-4, s), 122.5 (C-4a, s), 123.6 (C-4b, s), 102.8 (C-5, d), 153.5 (C-6, s), 115.4 (C-7, d), 112.4 (C-8, d), 134.5 (C-8a, s), 135.1 (C-9a, s), 55.5 (1-OMe, q), 55.5 (6-OMe q).

Zinc dust distillation of 1. Compound **1** (100 mg) was thoroughly mixed with Zn dust (10 g) previously dried at 250° and heated in a sealed tube for 2 hr. The ether-sol. portion of the reaction product was dissolved in benzene and chromatographed on alumina (5 g). Elution with benzene-petrol furnished a solid (5 mg) which showed

the presence of two compounds by TLC on silica gel (petrol-HOAc, 10:1, Rfs 0.55 and 0.65, respectively). The two compounds were identified as, carbazole (mp 242°) and 3-methylcarbazole (207°), respectively by GC and HPLC [6, 7].

Acetylation of 1. Compound **1** (70 mg) was dissolved in dry pyridine (2 ml) and Ac_2O (2 ml) and refluxed at 100° for 1 hr. After reaction the mixt. was poured into ice- H_2O (25 g) when a solid was obtained. The solid was filtered, washed with dil. HCl and finally with H_2O . It was then dried and recrystallized from benzene-petrol to give needles (55 mg, 66.3%) of **5**, mp 132° . UV: $\lambda_{\text{max}}^{\text{EtOH}}$ 230.8 (log ϵ 4.40), 252.2 (4.13), 263.4 (3.98), 303.2 (4.18), 269 nm (3.45). IR: $\nu_{\text{max}}^{\text{KBr}}$ 3400 (-NH), 1752 cm^{-1} (-OCOCH₃). (Found: C, 71.75; H, 5.70; N, 5.39%. Calcd for $\text{C}_{16}\text{H}_{15}\text{NO}_3$: C, 71.36; H, 5.61; N, 5.20%).

Tosylation and reduction of tosyl derivative of 1. The tosyl derivative (50 mg) of **1**, prepd by the usual method, was dissolved in EtOH (6.5 ml) and refluxed in the presence of Raney Ni (0.2 g) for 2 hr. The solid obtained after work-up was purified by sublimation under vacuum followed by recrystallization from benzene-petrol when crystals of **6** (15 mg, 54.1%), mp 180° , were obtained. The compound was found to be identical with glycozoline in all respects (mp, mmp, UV, IR).

4-Methylcyclohexene - 1,2-dione-1-(4-methoxy)phenylhydrazone (9). 2-Hydroxymethylene-5-methylcyclohexanone (5.6 g) in MeOH (50 ml) was added to aq. NaOAc (8.10 g in 30 ml H_2O). To this soln was added an aq. acid soln of 4-methoxyphenyldiazonium chloride (prepd from 4.85 g of 4-methoxyaniline) during 30 min with mechanical agitation to afford a red ppt. of **9**. This ppt was filtered and washed with H_2O to remove acid. The product was recrystallized from EtOH when red needles of **9** (4.8 g, 49.8%), mp 191° , were obtained. (Found C, 68.66;

H, 7.57; N, 11.20%. Calcd. for $C_{14}H_{18}N_2O_2$: C, 68.27; H, 7.37; N, 11.37%. IR: ν_{\max}^{KBr} 3400 (–NH), 1650 ($>C=O$), 1610 cm^{-1} .

6-Methoxy-3-methyl-1-Oxo-1,2,3,4-tetrahydrocarbazole (10). Compound **9** (3 g) was added to boiling HOAc (25 ml) and conc. HCl (6 ml) added through the reflux condenser. The soln was boiled for 3 min and then dild with ice–H₂O (150 ml). The solid obtained was filtered, washed with H₂O, dried and recrystallized from benzene to afford light ash-coloured crystals of **10** (1.8 g, 64.5%), mp 192°. (Found: C, 73.44; H, 6.61; N, 6.08, Calcd. for $C_{14}H_{15}NO_2$ C, 73.34, H, 6.59, N, 6.11%). UV: λ_{\max}^{EtOH} 212.4 (log ϵ 4.21), 313 nm (435). IR: ν_{\max}^{KBr} 3300 (–NH), 1635 ($>C=O$).

1-Hydroxy-6-methoxy-3-methylcarbazole (1). Compound **10** (1.5 g) was heated under reflux with Pd/C (10%, 0.5 g) in decalin (20 ml) for 5 hr. After reaction, decalin was removed by dist. under vacuum. The residue was dissolved in $CHCl_3$ –MeOH (49:1) and filtered. The filtrate on evapn gave a brown residue (0.5 g), which was dissolved in Et₂O and extracted ($\times 3$) with aq. NaOH (2%, w/v). The aq. alkaline soln on acidification with HCl gave a brown solid, which was subjected to CC over silica gel (10 g). The column was eluted with petrol, CH_2Cl_2 and

$CHCl_3$, and the CH_2Cl_2 frs yielded a light brown solid which on recrystallization from benzene–petrol furnished light brown crystals, mp 138°, of **1** (0.35 g, 19.9%).

Methylation of 1 and formation of 2. A MeOH soln (50 ml) of **1** (80 mg) was kept at 4°C for 24 hr with CH_2N_2 . On removal of solvent, a residue was obtained which was washed with 2% aq. NaOH. The solid obtained was recrystallized from benzene–petrol to yield needles of **2** (65 mg, 90.6%), mp 150°.

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